



IUCrData

ISSN 2414-3146

4-[(E)-2-(4-{(E)-2-[4-((E)-2-{4-[(E)-2-(4-Cyano-phenyl)ethenyl]phenyl}ethenyl)-2,5-bis(octyloxy)-phenyl]ethenyl}phenyl)ethenyl]benzonitrile

Heiner Detert* and Dieter Schollmeyer

University of Mainz, Department of Chemistry, Duesbergweg 10-14, 55099 Mainz, Germany. *Correspondence e-mail: detert@uni-mainz.de

Received 27 August 2026

Accepted 1 September 2026

Edited by W. T. A. Harrison, University of Aberdeen, United Kingdom

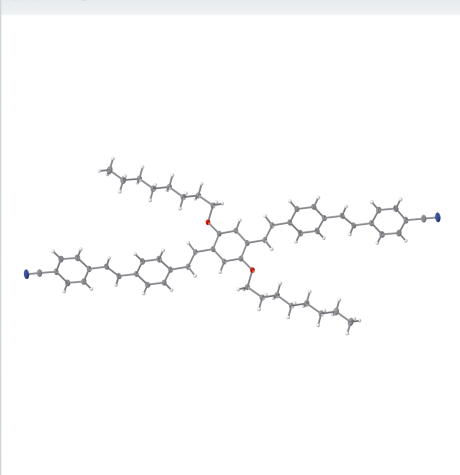
Keywords: crystal structure; phenylene vinylene; stilbene; oligomer.

CCDC reference: 2584677

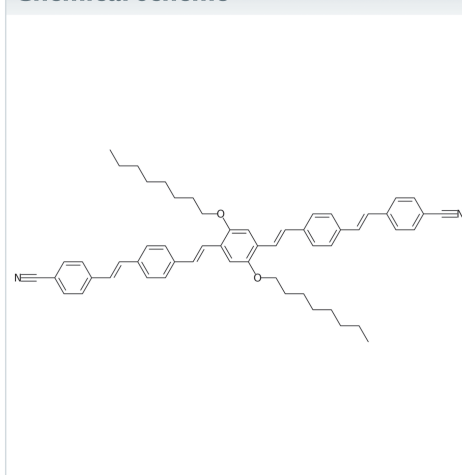
Structural data: full structural data are available from iucrdata.iucr.org

The complete molecule of the title compound, $C_{56}H_{60}N_2O_2$, which was prepared via a Horner olefination, is generated by a crystallographic centre of symmetry. Whereas the terminal cyanostilbene units are almost planar, the central diocetoxybenzene unit is substantially tilted [dihedral angle = $51.68(8)^\circ$], which limits conjugation. van der Waals forces between the alkyl chains connect the molecules, arranged in layers lying parallel to the *ac* plane.

3D view



Chemical scheme



Structure description

The title compound, $C_{56}H_{60}N_2O_2$ (**1**), is a linear *para*-oligophenylene-vinylene (OPV). It was prepared as part of a project on organic fluorophors with increased electron affinity (Detert & Sugiono, 2000; Stalmach & Detert, 2000; Detert & Schmitt, 2004). The structure of three related cyano-oligo-phenylenevinylenes with five benzene rings have been reported earlier (Gill *et al.* 1996; Detert *et al.* 2001) and the crystal structure of the unsubstituted 5-ring OPV has also been reported (van Hutten *et al.* 1999). Two molecules of (**1**) fill the monoclinic unit cell. The centrosymmetric molecules are composed of two terminal and almost planar stilbene units (ring 1 and 2) connected to a twisted divinylbenzene moiety (ring 3) (Fig. 1). The terminal cyano groups and central octyloxy chains result formally in an electronic donor–acceptor–donor (DAD) structure. The key torsion angles of the stilbene unit are $C9-C12-C13-C14 = -178.79(15)^\circ$, $C8-C9-C12-C13 = 179.36(16)^\circ$ and $C19-C14-C13-C12 = 178.27(16)^\circ$ and the dihedral angle between the $C6-C12$ and the $C14-C19$ aromatic rings is $6.03(8)^\circ$. Ring 2 ($C6-C11$) and the vinylene unit subtend a torsion angle $C7-C6-C5-C4$ of $-151.99(17)^\circ$ and the torsion angle between this vinylene unit and the central ring 3 ($C5-C4-C3-C1^1$) [symmetry code: (i) $1 - x, 1 - y, 1 - z$] amounts to $156.00(16)^\circ$. The dihedral angle between the central ring and the $C6-C12$ ring is $51.68(8)^\circ$. Whereas the first methylene groups show a *gauche* conformation about the $C21-C22$ bond



OPEN ACCESS

Published under a CC BY 4.0 licence

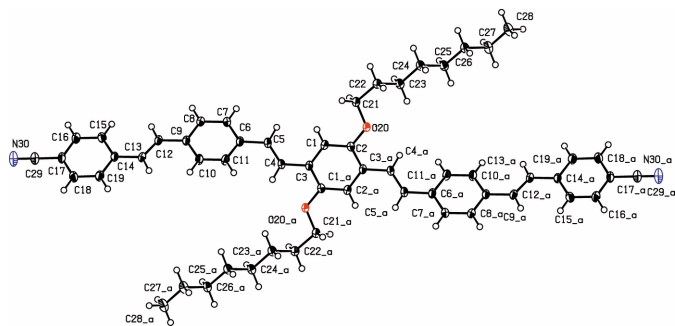


Figure 1

The molecular structure of (**I**) with displacement ellipsoids drawn at the 50% probability level. Symmetry code: (a) $1 - x, 1 - y, 1 - z$.

[O20—C21—C22—C23 = $60.83(19)^\circ$], the rest of the octyl chain is perfectly *all-anti* configured.

In the extended structure of (**I**), the molecules are arranged in alternating layers lying parallel to the *ac* plane (Fig. 2). The rings 1 and 2 are almost parallel to the *ac*-plane whereas the central ring is tilted: the dihedral angle between the central rings 3 and 3' of the alternating layers amounts to $71.50(7)^\circ$. The alkyl chains connect parallel layers of molecules, along the *b* axis, one chain interacts with the chains of molecules above and one with those below, forming aliphatic layers between the conjugated segments, also lying roughly parallel to the *ac* plane.

Synthesis and crystallization

A solution of potassium-*t*-butylate (336 mg, 3 mmol) and 60 mg 18-crown-6 in 30 ml of anhydrous tetrahydrofuran (THF) was purged with nitrogen and a solution of 594 mg (1 mmol) 1,4-bis(4-formylphenylethynyl)-2,5-dioctyloxybenzene (Detert *et al.* 2001). 2 mmol (506 mg) *p*-cyanobenzyl phosphonate in 15 ml of anhydrous THF was added dropwise over 15 min. Stirring was continued for 16 h before the reaction mixture was poured into 150 ml of 0.5 N hydrochloric acid. The product was isolated by filtration, the residue was washed with 100 ml of water/methanol (2/1) dried in vacuum and recrystallized from chloroform solution by adding methanol to the hot solution in 0.66 g (83%) yield. Recrystallization for single crystals was from the mixed solvents of

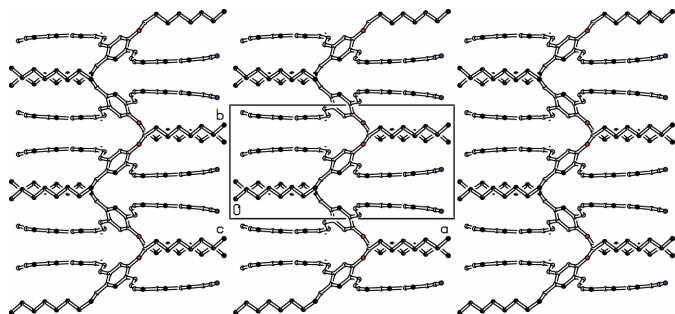


Figure 2

Part of the packing diagram of (**I**) viewed along the *c*-axis direction. Hydrogen atoms omitted for clarity.

Table 1

Experimental details.

Crystal data	
Chemical formula	$C_{56}H_{60}N_2O_2$
M_r	793.06
Crystal system, space group	Monoclinic, $P2_1/c$
Temperature (K)	120
a, b, c (Å)	22.0613 (7), 11.1029 (3), 9.1280 (3)
β ($^\circ$)	96.004 (3)
V (Å ³)	2223.59 (12)
Z	2
Radiation type	Mo $K\alpha$
μ (mm ⁻¹)	0.07
Crystal size (mm)	$0.26 \times 0.15 \times 0.05$
Data collection	
Diffractometer	Stoe <i>IPDS</i> 2T
Absorption correction	—
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	14627, 5254, 3801
R_{int}	0.060
$(\sin \theta/\lambda)_{max}$ (Å ⁻¹)	0.658
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.056, 0.150, 1.10
No. of reflections	5254
No. of parameters	272
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{max}, \Delta\rho_{min}$ (e Å ⁻³)	0.32, -0.23

Computer programs: *X-AREA WinXpose*, *X-AREA Recipe* and *X-AREA Integrate* (Stoe & Cie, 2020), *SHELXT2014* (Sheldrick, 2015a), *SHELXL2019/2* (Sheldrick, 2015b) and *PLATON* (Spek, 2009).

chloroform and 2-propanol. Bright yellow crystals with m.p. = 481 K. MS (FD): 793 (100) [M^+], 396.5 (1) [M^{2+}]; IR (KBr): ν = 3010, 2908, 2840, 2220, 1590, 1582, 1502, 1480, 1453, 1408, 1248, 1195, 1019, 957, 850, 830, 820 cm⁻¹; ¹H-NMR (CDCl₃, 400 MHz): δ = 0.88 (*t*, J = 6.4 Hz, 6 H, CH₃), 1.25–1.45 (*m*, 20 H, CH₂), 1.56 (*qui*, J = 7.5 Hz, 4 H, CH₂), 1.87 (*qui*, J = 7.5 Hz, 4 H, CH₂), 4.07 (*t*, J = 6 Hz, 4 H, OCH₂), 7.06 (*d*, J = 16.4 Hz, 2 H, *vin*), 7.11 (*s*, 2 H), 7.13 (*d*, J = 16 Hz, 2 H, *vin*), 7.19 (*d*, J = 16.4 Hz, 2 H, *vin*), 7.53 (*d*, J = 16 Hz, 2 H, *vin*), 7.52 (*AA'BB'*, 8 H), 7.59 (*AA'BB'*, 8 H); ¹³C-NMR (CDCl₃, 100 MHz): δ = 14.1 (CH₃), 22.7, 26.3, 29.3, 29.4, 29.5, 31.8 (CH₂), 69.6 (OCH₂), 110.5, 110.7, 119.0 (CN), 124.1, 126.4 (CH *vin*), 126.8, 127.0, 127.3 (CH *ar*), 138.3, 132.0 (CH *vin*), 132.5 (CH *ar*), 135.4, 138.5, 141.9, 151.3 (Cq). UV-Vis λ_{max} = 434 nm (dichloromethane, log ϵ = 4.93), fluorescence: λ_{max} = 485 nm (cyclohexane, Φ = 0.53), λ_{max} = 521 (acetonitrile).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 1.

References

- Detert, H. & Schmitt, V. (2004). *J. Phys. Org. Chem.* **17**, 105–108.
 Detert, H., Schollmeier, D. & Sugiono, E. (2001). *Eur. J. Org. Chem.* pp. 2927–2938.
 Detert, H. & Sugiono, E. (2000). *J. Phys. Org. Chem.* **13**, 587–590.
 Gill, R. E., van Hutten, P., Meetsma, A. & Hadziioannou, G. (1996). *Chem. Mater.* **8**, 1341–1346.
 Sheldrick, G. M. (2015a). *Acta Cryst.* **A71**, 3–8.

Sheldrick, G. M. (2015*b*). *Acta Cryst.* **C71**, 3–8.

Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

Stalmach, U. & Detert, H. (2000). *J. Prakt. Chem.* **342**, 10–16.

Stoe & Cie (2020). *X-AREA WinXpose*, *X-AREA Recipe* and *X-AREA Integrate*. Stoe & Cie, Darmstadt, Germany.

van Hutten, P. F., Wildeman, J., Meetsma, A. & Hadziioannou, G. (1999). *J. Am. Chem. Soc.* **121**, 5910–5918.

full crystallographic data

IUCrData (2026). **11**, x260912 [<https://doi.org/10.1107/S2414314626009120>]

4-[(*E*)-2-(4-{(*E*)-2-[4-((*E*)-2-{4-[(*E*)-2-(4-Cyanophenyl)ethenyl]phenyl}ethenyl)-2,5-bis(octyloxy)phenyl]ethenyl}phenyl)ethenyl]benzonitrile

Heiner Detert and Dieter Schollmeyer

4-[(*E*)-2-(4-{(*E*)-2-[4-((*E*)-2-{4-[(*E*)-2-(4-Cyanophenyl)ethenyl]phenyl}ethenyl)-2,5-bis(octyloxy)phenyl]ethenyl}phenyl)ethenyl]benzonitrile

Crystal data

$C_{56}H_{60}N_2O_2$
 $M_r = 793.06$
 Monoclinic, $P2_1/c$
 $a = 22.0613$ (7) Å
 $b = 11.1029$ (3) Å
 $c = 9.1280$ (3) Å
 $\beta = 96.004$ (3)°
 $V = 2223.59$ (12) Å³
 $Z = 2$

$F(000) = 852$
 $D_x = 1.184$ Mg m⁻³
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
 Cell parameters from 10876 reflections
 $\theta = 2.9$ – 28.4 °
 $\mu = 0.07$ mm⁻¹
 $T = 120$ K
 Plate, yellow
 $0.26 \times 0.15 \times 0.05$ mm

Data collection

Stoe IPDS 2T
 diffractometer
 Detector resolution: 6.67 pixels mm⁻¹
 rotation method, ω scans
 14627 measured reflections
 5254 independent reflections

3801 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.060$
 $\theta_{max} = 27.9$ °, $\theta_{min} = 2.9$ °
 $h = -28 \rightarrow 28$
 $k = -14 \rightarrow 14$
 $l = -11 \rightarrow 10$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.056$
 $wR(F^2) = 0.150$
 $S = 1.10$
 5254 reflections
 272 parameters
 0 restraints
 Primary atom site location: dual

Hydrogen site location: inferred from neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0575P)^2 + 0.8682P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} = 0.001$
 $\Delta\rho_{max} = 0.32$ e Å⁻³
 $\Delta\rho_{min} = -0.23$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Hydrogen atoms attached to carbon atoms were placed at calculated positions and were refined in the riding-model approximation with C—H = 0.95–0.99 Å, and with $U_{iso}(H) = 1.2 U_{eq}(C)$.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.54781 (7)	0.57938 (15)	0.53929 (17)	0.0239 (3)
C2	0.51135 (7)	0.59457 (15)	0.40744 (17)	0.0250 (3)
H2	0.519388	0.659476	0.344480	0.030*
C3	0.46279 (7)	0.51606 (14)	0.36499 (17)	0.0231 (3)
C4	0.42353 (7)	0.53272 (14)	0.22687 (17)	0.0234 (3)
H4	0.383510	0.500323	0.221817	0.028*
C5	0.43916 (7)	0.58960 (15)	0.10723 (17)	0.0243 (3)
H5	0.479187	0.621900	0.111866	0.029*
C6	0.39939 (7)	0.60616 (14)	−0.03081 (17)	0.0227 (3)
C7	0.42449 (7)	0.61702 (15)	−0.16410 (18)	0.0253 (3)
H7	0.467510	0.621205	−0.163772	0.030*
C8	0.38796 (7)	0.62187 (15)	−0.29725 (17)	0.0249 (3)
H8	0.406367	0.628526	−0.386451	0.030*
C9	0.32444 (7)	0.61709 (13)	−0.30210 (16)	0.0212 (3)
C10	0.29919 (7)	0.61295 (15)	−0.16739 (17)	0.0253 (3)
H10	0.256141	0.613619	−0.167190	0.030*
C11	0.33561 (7)	0.60793 (15)	−0.03533 (17)	0.0255 (3)
H11	0.317221	0.605628	0.054207	0.031*
C12	0.28738 (7)	0.61536 (14)	−0.44523 (17)	0.0227 (3)
H12	0.308816	0.618492	−0.530154	0.027*
C13	0.22659 (7)	0.60985 (14)	−0.46871 (16)	0.0232 (3)
H13	0.204763	0.608294	−0.384361	0.028*
C14	0.19060 (7)	0.60598 (14)	−0.61326 (17)	0.0222 (3)
C15	0.21729 (8)	0.60958 (17)	−0.74565 (18)	0.0305 (4)
H15	0.260282	0.617067	−0.743152	0.037*
C16	0.18224 (8)	0.60242 (17)	−0.87991 (18)	0.0314 (4)
H16	0.201187	0.605442	−0.968637	0.038*
C17	0.11929 (7)	0.59082 (15)	−0.88546 (18)	0.0257 (3)
C18	0.09147 (7)	0.58874 (16)	−0.75575 (19)	0.0296 (4)
H18	0.048438	0.581748	−0.758965	0.036*
C19	0.12712 (7)	0.59699 (16)	−0.62128 (18)	0.0282 (4)
H19	0.107929	0.596498	−0.532842	0.034*
O20	0.59453 (5)	0.65600 (11)	0.58728 (12)	0.0294 (3)
C21	0.61954 (8)	0.72790 (17)	0.47691 (19)	0.0312 (4)
H21A	0.588076	0.782783	0.429117	0.037*
H21B	0.634359	0.675574	0.400389	0.037*
C22	0.67179 (8)	0.79956 (16)	0.55458 (19)	0.0297 (4)
H22A	0.655559	0.851893	0.629142	0.036*
H22B	0.688589	0.852632	0.481586	0.036*
C23	0.72361 (7)	0.72408 (15)	0.63057 (19)	0.0280 (4)
H23A	0.741249	0.673462	0.556451	0.034*
H23B	0.707294	0.669872	0.703115	0.034*
C24	0.77349 (7)	0.80304 (16)	0.70892 (18)	0.0275 (4)
H24A	0.790550	0.854940	0.635149	0.033*
H24B	0.755105	0.856283	0.779200	0.033*

C25	0.82516 (7)	0.73201 (15)	0.79204 (19)	0.0276 (4)
H25A	0.844567	0.680762	0.721451	0.033*
H25B	0.808084	0.678287	0.863937	0.033*
C26	0.87357 (7)	0.81255 (15)	0.87348 (18)	0.0271 (3)
H26A	0.889982	0.867307	0.801669	0.033*
H26B	0.854171	0.862822	0.945071	0.033*
C27	0.92614 (8)	0.74288 (17)	0.9551 (2)	0.0321 (4)
H27A	0.909921	0.688541	1.027633	0.039*
H27B	0.945598	0.692386	0.883831	0.039*
C28	0.97400 (8)	0.82510 (19)	1.0348 (2)	0.0376 (4)
H28A	0.989332	0.880827	0.964013	0.056*
H28B	1.007783	0.776397	1.081085	0.056*
H28C	0.955712	0.871022	1.110644	0.056*
C29	0.08398 (8)	0.57799 (16)	−1.02704 (19)	0.0302 (4)
N30	0.05706 (8)	0.56627 (16)	−1.14068 (18)	0.0424 (4)

Atomic displacement parameters ($\text{\AA}^2$)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0224 (7)	0.0294 (8)	0.0190 (7)	−0.0037 (6)	−0.0026 (6)	0.0007 (6)
C2	0.0266 (8)	0.0299 (8)	0.0175 (7)	−0.0036 (6)	−0.0029 (6)	0.0040 (6)
C3	0.0227 (7)	0.0296 (8)	0.0161 (7)	−0.0010 (6)	−0.0023 (6)	0.0017 (6)
C4	0.0223 (7)	0.0277 (8)	0.0189 (7)	−0.0030 (6)	−0.0048 (6)	0.0006 (6)
C5	0.0238 (7)	0.0286 (8)	0.0188 (7)	−0.0017 (6)	−0.0047 (6)	0.0003 (6)
C6	0.0266 (8)	0.0228 (7)	0.0173 (7)	−0.0009 (6)	−0.0038 (6)	0.0010 (6)
C7	0.0213 (7)	0.0323 (8)	0.0212 (8)	−0.0025 (6)	−0.0034 (6)	0.0036 (6)
C8	0.0255 (8)	0.0307 (8)	0.0181 (7)	−0.0019 (6)	−0.0001 (6)	0.0023 (6)
C9	0.0243 (7)	0.0212 (7)	0.0170 (7)	−0.0002 (6)	−0.0030 (6)	0.0005 (6)
C10	0.0219 (7)	0.0345 (9)	0.0186 (7)	0.0039 (6)	−0.0014 (6)	0.0017 (6)
C11	0.0256 (8)	0.0338 (9)	0.0166 (7)	0.0039 (6)	0.0006 (6)	0.0000 (6)
C12	0.0271 (8)	0.0247 (7)	0.0157 (7)	0.0003 (6)	−0.0013 (6)	0.0007 (6)
C13	0.0261 (8)	0.0280 (8)	0.0149 (7)	−0.0003 (6)	−0.0010 (6)	0.0006 (6)
C14	0.0240 (7)	0.0237 (7)	0.0178 (7)	−0.0002 (6)	−0.0032 (6)	0.0011 (6)
C15	0.0233 (8)	0.0467 (10)	0.0210 (8)	−0.0053 (7)	−0.0002 (6)	−0.0013 (7)
C16	0.0306 (9)	0.0463 (10)	0.0168 (8)	−0.0042 (8)	−0.0001 (6)	0.0001 (7)
C17	0.0283 (8)	0.0266 (8)	0.0202 (8)	0.0004 (6)	−0.0071 (6)	0.0009 (6)
C18	0.0212 (7)	0.0400 (9)	0.0264 (8)	0.0013 (7)	−0.0037 (6)	0.0022 (7)
C19	0.0255 (8)	0.0381 (9)	0.0204 (8)	0.0015 (7)	0.0006 (6)	0.0017 (7)
O20	0.0315 (6)	0.0366 (7)	0.0183 (5)	−0.0139 (5)	−0.0067 (5)	0.0048 (5)
C21	0.0318 (9)	0.0371 (9)	0.0224 (8)	−0.0114 (7)	−0.0071 (7)	0.0078 (7)
C22	0.0305 (8)	0.0323 (9)	0.0250 (8)	−0.0084 (7)	−0.0037 (7)	0.0055 (7)
C23	0.0272 (8)	0.0297 (8)	0.0257 (8)	−0.0050 (7)	−0.0035 (7)	0.0001 (7)
C24	0.0268 (8)	0.0308 (8)	0.0236 (8)	−0.0057 (7)	−0.0034 (6)	−0.0006 (7)
C25	0.0255 (8)	0.0300 (8)	0.0262 (8)	−0.0021 (7)	−0.0019 (6)	−0.0018 (7)
C26	0.0265 (8)	0.0304 (8)	0.0234 (8)	−0.0025 (7)	−0.0025 (6)	−0.0009 (7)
C27	0.0272 (8)	0.0358 (9)	0.0316 (9)	0.0034 (7)	−0.0046 (7)	−0.0047 (7)
C28	0.0276 (8)	0.0458 (11)	0.0373 (10)	0.0022 (8)	−0.0061 (7)	−0.0069 (8)
C29	0.0296 (8)	0.0333 (9)	0.0257 (9)	0.0041 (7)	−0.0066 (7)	−0.0008 (7)

N30 0.0448 (9) 0.0499 (10) 0.0288 (8) 0.0090 (8) −0.0138 (7) −0.0058 (7)

Geometric parameters (Å, °)

C1—O20	1.3726 (18)	C17—C18	1.389 (2)
C1—C2	1.386 (2)	C17—C29	1.445 (2)
C1—C3 ⁱ	1.408 (2)	C18—C19	1.390 (2)
C2—C3	1.404 (2)	C18—H18	0.9500
C2—H2	0.9500	C19—H19	0.9500
C3—C4	1.465 (2)	O20—C21	1.4398 (19)
C4—C5	1.337 (2)	C21—C22	1.514 (2)
C4—H4	0.9500	C21—H21A	0.9900
C5—C6	1.470 (2)	C21—H21B	0.9900
C5—H5	0.9500	C22—C23	1.525 (2)
C6—C7	1.394 (2)	C22—H22A	0.9900
C6—C11	1.404 (2)	C22—H22B	0.9900
C7—C8	1.387 (2)	C23—C24	1.525 (2)
C7—H7	0.9500	C23—H23A	0.9900
C8—C9	1.398 (2)	C23—H23B	0.9900
C8—H8	0.9500	C24—C25	1.522 (2)
C9—C10	1.403 (2)	C24—H24A	0.9900
C9—C12	1.467 (2)	C24—H24B	0.9900
C10—C11	1.378 (2)	C25—C26	1.525 (2)
C10—H10	0.9500	C25—H25A	0.9900
C11—H11	0.9500	C25—H25B	0.9900
C12—C13	1.337 (2)	C26—C27	1.522 (2)
C12—H12	0.9500	C26—H26A	0.9900
C13—C14	1.468 (2)	C26—H26B	0.9900
C13—H13	0.9500	C27—C28	1.522 (2)
C14—C19	1.398 (2)	C27—H27A	0.9900
C14—C15	1.399 (2)	C27—H27B	0.9900
C15—C16	1.381 (2)	C28—H28A	0.9800
C15—H15	0.9500	C28—H28B	0.9800
C16—C17	1.391 (2)	C28—H28C	0.9800
C16—H16	0.9500	C29—N30	1.148 (2)
O20—C1—C2	123.36 (14)	C18—C19—C14	121.47 (15)
O20—C1—C3 ⁱ	115.85 (13)	C18—C19—H19	119.3
C2—C1—C3 ⁱ	120.76 (14)	C14—C19—H19	119.3
C1—C2—C3	121.39 (15)	C1—O20—C21	116.76 (12)
C1—C2—H2	119.3	O20—C21—C22	107.00 (13)
C3—C2—H2	119.3	O20—C21—H21A	110.3
C2—C3—C1 ⁱ	117.86 (14)	C22—C21—H21A	110.3
C2—C3—C4	121.64 (14)	O20—C21—H21B	110.3
C1 ⁱ —C3—C4	120.50 (14)	C22—C21—H21B	110.3
C5—C4—C3	125.64 (14)	H21A—C21—H21B	108.6
C5—C4—H4	117.2	C21—C22—C23	114.96 (15)
C3—C4—H4	117.2	C21—C22—H22A	108.5

C4—C5—C6	125.26 (14)	C23—C22—H22A	108.5
C4—C5—H5	117.4	C21—C22—H22B	108.5
C6—C5—H5	117.4	C23—C22—H22B	108.5
C7—C6—C11	117.50 (14)	H22A—C22—H22B	107.5
C7—C6—C5	120.22 (14)	C22—C23—C24	111.55 (14)
C11—C6—C5	122.25 (14)	C22—C23—H23A	109.3
C8—C7—C6	121.38 (15)	C24—C23—H23A	109.3
C8—C7—H7	119.3	C22—C23—H23B	109.3
C6—C7—H7	119.3	C24—C23—H23B	109.3
C7—C8—C9	120.97 (15)	H23A—C23—H23B	108.0
C7—C8—H8	119.5	C25—C24—C23	113.70 (14)
C9—C8—H8	119.5	C25—C24—H24A	108.8
C8—C9—C10	117.51 (14)	C23—C24—H24A	108.8
C8—C9—C12	119.47 (14)	C25—C24—H24B	108.8
C10—C9—C12	123.02 (14)	C23—C24—H24B	108.8
C11—C10—C9	121.30 (15)	H24A—C24—H24B	107.7
C11—C10—H10	119.3	C24—C25—C26	112.89 (14)
C9—C10—H10	119.3	C24—C25—H25A	109.0
C10—C11—C6	121.14 (15)	C26—C25—H25A	109.0
C10—C11—H11	119.4	C24—C25—H25B	109.0
C6—C11—H11	119.4	C26—C25—H25B	109.0
C13—C12—C9	126.83 (14)	H25A—C25—H25B	107.8
C13—C12—H12	116.6	C27—C26—C25	113.53 (15)
C9—C12—H12	116.6	C27—C26—H26A	108.9
C12—C13—C14	125.77 (14)	C25—C26—H26A	108.9
C12—C13—H13	117.1	C27—C26—H26B	108.9
C14—C13—H13	117.1	C25—C26—H26B	108.9
C19—C14—C15	117.81 (14)	H26A—C26—H26B	107.7
C19—C14—C13	119.59 (14)	C28—C27—C26	112.59 (16)
C15—C14—C13	122.60 (14)	C28—C27—H27A	109.1
C16—C15—C14	121.17 (15)	C26—C27—H27A	109.1
C16—C15—H15	119.4	C28—C27—H27B	109.1
C14—C15—H15	119.4	C26—C27—H27B	109.1
C15—C16—C17	120.13 (15)	H27A—C27—H27B	107.8
C15—C16—H16	119.9	C27—C28—H28A	109.5
C17—C16—H16	119.9	C27—C28—H28B	109.5
C18—C17—C16	119.93 (15)	H28A—C28—H28B	109.5
C18—C17—C29	121.06 (15)	C27—C28—H28C	109.5
C16—C17—C29	118.99 (15)	H28A—C28—H28C	109.5
C17—C18—C19	119.47 (15)	H28B—C28—H28C	109.5
C17—C18—H18	120.3	N30—C29—C17	178.4 (2)
C19—C18—H18	120.3		
O20—C1—C2—C3	−177.93 (15)	C12—C13—C14—C19	178.27 (16)
C3 ⁱ —C1—C2—C3	0.1 (3)	C12—C13—C14—C15	−1.0 (3)
C1—C2—C3—C1 ⁱ	−0.1 (3)	C19—C14—C15—C16	−1.1 (3)
C1—C2—C3—C4	179.06 (15)	C13—C14—C15—C16	178.19 (17)
C2—C3—C4—C5	24.8 (3)	C14—C15—C16—C17	−0.3 (3)

C1 ⁱ —C3—C4—C5	−156.00 (16)	C15—C16—C17—C18	1.2 (3)
C3—C4—C5—C6	−179.83 (15)	C15—C16—C17—C29	−177.20 (16)
C4—C5—C6—C7	−151.99 (17)	C16—C17—C18—C19	−0.7 (3)
C4—C5—C6—C11	26.0 (3)	C29—C17—C18—C19	177.69 (16)
C11—C6—C7—C8	−3.9 (2)	C17—C18—C19—C14	−0.7 (3)
C5—C6—C7—C8	174.14 (15)	C15—C14—C19—C18	1.6 (3)
C6—C7—C8—C9	0.6 (3)	C13—C14—C19—C18	−177.68 (15)
C7—C8—C9—C10	2.9 (2)	C2—C1—O20—C21	−21.7 (2)
C7—C8—C9—C12	−176.47 (15)	C3 ⁱ —C1—O20—C21	160.09 (15)
C8—C9—C10—C11	−3.0 (2)	C1—O20—C21—C22	−177.28 (14)
C12—C9—C10—C11	176.31 (15)	O20—C21—C22—C23	60.83 (19)
C9—C10—C11—C6	−0.3 (3)	C21—C22—C23—C24	−178.65 (14)
C7—C6—C11—C10	3.8 (2)	C22—C23—C24—C25	177.60 (14)
C5—C6—C11—C10	−174.23 (15)	C23—C24—C25—C26	−178.22 (14)
C8—C9—C12—C13	179.35 (16)	C24—C25—C26—C27	−179.01 (14)
C10—C9—C12—C13	0.0 (3)	C25—C26—C27—C28	179.70 (15)
C9—C12—C13—C14	−178.79 (15)		

Symmetry code: (i) $-x+1, -y+1, -z+1$.