

6-[4-(*tert*-Butyldimethylsilyloxy)phenyl]-1-oxaspiro[2.5]heptane

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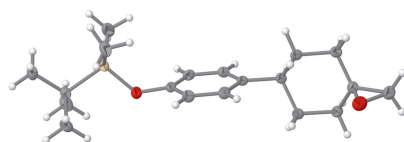
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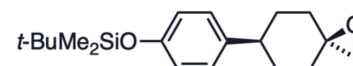
Structural data: full structural data are available from iucrdata.iucr.org

The title compound, $C_{19}H_{30}O_2Si$, has triclinic ($P\bar{1}$) symmetry at 100 K. The O atom of the epoxide group has a pseudoaxial orientation and the dihedral angle between the cyclohexyl and benzene rings is $85.80(8)^\circ$. The C—O—Si—C_t (*t* = *tert*-butyl) torsion angle is $-177.40(14)^\circ$. In the crystal, pairwise C—H \cdots O links connect the molecules into inversion dimers featuring $R_2^2(8)$ loops.

3D view



Chemical scheme



Structure description

Fumagillin and ovalicin, isolated from *Aspergillus fumigatus* and *Pseudorotium ovalis*, respectively, are sesquiterpene epoxides that exhibit anti-angiogenic activity. The key structural feature of both is a 1-oxospiro[2.5]heptane moiety. The structure of fumagillin was initially deduced by X-ray crystallographic analysis of its hydrolysis product fumagillol (McCorkindale & Sime, 1961) and the X-ray crystal structure of fumagillin was eventually reported (Halasz *et al.*, 2000). Preparation of the 1-oxospiro[2.5]heptane system by reaction of dimethylsulfoxonium methylide with substituted cyclohexanones generally proceeds with the formation of the exocyclic epoxide in which the oxygen atom has an axial orientation (Corey & Chaykovsky, 1965; Carlson & Behn, 1967). In connection with our studies on estrogen receptor beta-selective agonists (Hanson *et al.*, 2018; Wetzel *et al.*, 2022), we had the opportunity to prepare the title compound, $C_{19}H_{30}O_2Si$, and we now present its synthesis and crystal structure.

The title compound (Fig. 1) has an extended conformation with a *transoid* *t*-Bu—Si—O—Ar moiety. The cyclohexane ring has a chair conformation with the aryl substituent in an equatorial position and the epoxide oxygen atom in a pseudoaxial orientation. The C7—O1—Si1—C3 torsion angle is $-177.40(14)^\circ$. In the crystal, a weak C—H \cdots O link (Fig. 2, Table 1) connects the molecules into inversion dimers featuring $R_2^2(8)$ loops.

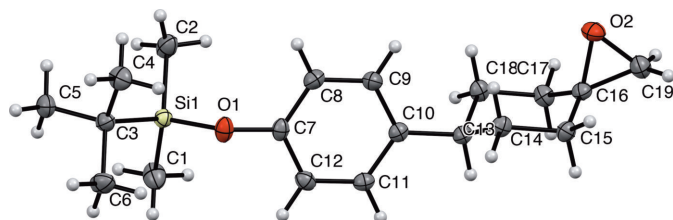


Figure 1
The molecular structure of the title compound showing 50% displacement ellipsoids.

Synthesis and crystallization

The reaction scheme is shown in Fig. 3. To a solution of potassium *tert*-butoxide (2.317 g, 20.65 mmol) in dimethylsulfoxide (DMSO) (10 ml) was added trimethylsulfoxonium iodide (5.00 g, 22.72 mmol). The solution was stirred for 30 min, and then a solution of 4-[(4-*t*-butyldimethylsilyloxy)phenyl]cyclohexanone (6.288 g, 20.65 mmol) in DMSO (50 ml) was added dropwise. The mixture was stirred for 24 h, and then partitioned between ethyl acetate and water. The aqueous layer was extracted with ethyl acetate. The combined organic extracts were washed with brine, dried (MgSO₄), and concentrated. Upon standing at room temperature overnight, colorless crystals formed. The crystals were filtered off to afford the title compound (5.262 g, 80%). Recrystallization from the mixed solvents of ethyl acetate/hexanes gave colorless flat prisms. ¹H NMR (400 MHz, CDCl₃) δ 7.08 (*d*, *J* = 8.0 Hz, 2H), 6.76 (*d*, *J* = 8.0 Hz, 2H), 2.68 (*s*, 2H), 2.54 (*t*, *J* = 8.2 Hz, 1H), 2.10–1.97 (*m*, 2H), 1.92–1.74 (*m*, 4H), 1.35 (*d*, *J* = Hz, 2H), 0.97 (*s*, 9H), 0.18 (*s*, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 153.7, 139.3, 127.6, 119.8, 57.9, 53.9, 42.4, 33.2, 31.7, 25.6, 18.1, –4.4 p.p.m..

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

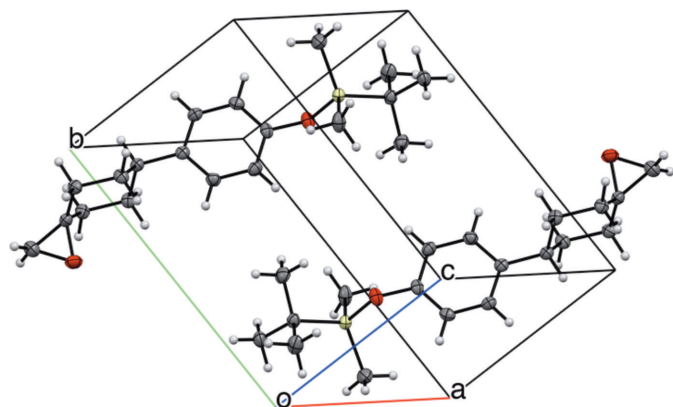


Figure 2
Unit-cell packing of the title compound.

Table 1
Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C15–H15A···O2 ⁱ	0.99	2.54	3.487 (2)	159

Symmetry code: (i) $-x + 3, -y + 1, -z + 2$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₉ H ₃₀ O ₂ Si
<i>M</i> _r	318.52
Crystal system, space group	Triclinic, <i>P</i> $\bar{1}$
Temperature (K)	100
<i>a</i> , <i>b</i> , <i>c</i> (Å)	7.3673 (3), 11.2285 (4), 12.5778 (5)
α , β , γ (°)	104.974 (4), 101.964 (4), 103.460 (4)
<i>V</i> (Å ³)	937.09 (7)
<i>Z</i>	2
Radiation type	Cu <i>K</i> α
μ (mm ⁻¹)	1.13
Crystal size (mm)	0.74 × 0.43 × 0.07
Data collection	
Diffractometer	Rigaku Oxford Diffraction SuperNova, Dual, Cu at home/near, Atlas
Absorption correction	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2018)
<i>T</i> _{min} , <i>T</i> _{max}	0.184, 1.000
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	16772, 3539, 3276
<i>R</i> _{int}	0.046
(sin θ / λ) _{max} (Å ⁻¹)	0.612
Refinement	
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.046, 0.131, 1.10
No. of reflections	3539
No. of parameters	212
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	0.70, –0.35

Computer programs: *CrysAlis PRO* (Rigaku OD, 2018), *OLEX2.solve* (Bourhis *et al.*, 2015), *SHELXL* (Sheldrick, 2015), and *OLEX2* (Dolomanov *et al.*, 2009).

Funding information

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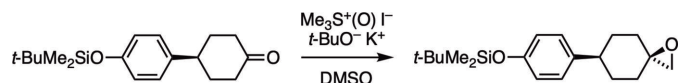


Figure 3
Reaction scheme.

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full crystallographic data

IUCrData (2024). **9**, x240590 [https://doi.org/10.1107/S241431462400590X]

6-[4-(*tert*-Butyldimethylsilyloxy)phenyl]-1-oxaspiro[2.5]heptane

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6-[4-(*tert*-Butyldimethylsilyloxy)phenyl]-1-oxaspiro[2.5]heptane*Crystal data*

$C_{19}H_{30}O_2Si$	$Z = 2$
$M_r = 318.52$	$F(000) = 348$
Triclinic, $P\bar{1}$	$D_x = 1.129 \text{ Mg m}^{-3}$
$a = 7.3673 (3) \text{ \AA}$	Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$
$b = 11.2285 (4) \text{ \AA}$	Cell parameters from 9134 reflections
$c = 12.5778 (5) \text{ \AA}$	$\theta = 4.3\text{--}70.5^\circ$
$\alpha = 104.974 (4)^\circ$	$\mu = 1.13 \text{ mm}^{-1}$
$\beta = 101.964 (4)^\circ$	$T = 100 \text{ K}$
$\gamma = 103.460 (4)^\circ$	Plate, colourless
$V = 937.09 (7) \text{ \AA}^3$	$0.74 \times 0.43 \times 0.07 \text{ mm}$

Data collection

Rigaku Oxford Diffraction SuperNova, Dual, Cu at home/near, Atlas diffractometer	$T_{\min} = 0.184$, $T_{\max} = 1.000$
Radiation source: micro-focus sealed X-ray tube, SuperNova (Cu) X-ray Source	16772 measured reflections
Mirror monochromator	3539 independent reflections
Detector resolution: $10.3756 \text{ pixels mm}^{-1}$	3276 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.046$
Absorption correction: gaussian (CrysAlisPro; Rigaku OD, 2018)	$\theta_{\max} = 70.6^\circ$, $\theta_{\min} = 3.8^\circ$
	$h = -8 \rightarrow 8$
	$k = -13 \rightarrow 13$
	$l = -15 \rightarrow 15$

Refinement

Refinement on F^2	Hydrogen site location: mixed
Least-squares matrix: full	H atoms treated by a mixture of independent and constrained refinement
$R[F^2 > 2\sigma(F^2)] = 0.046$	$w = 1/[\sigma^2(F_o^2) + (0.0862P)^2 + 0.2247P]$
$wR(F^2) = 0.131$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.10$	$(\Delta/\sigma)_{\max} < 0.001$
3539 reflections	$\Delta\rho_{\max} = 0.70 \text{ e \AA}^{-3}$
212 parameters	$\Delta\rho_{\min} = -0.35 \text{ e \AA}^{-3}$
0 restraints	
Primary atom site location: iterative	

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Si1	0.40119 (5)	0.19395 (4)	0.24021 (3)	0.01921 (16)
O1	0.47620 (15)	0.22061 (11)	0.38036 (9)	0.0263 (3)
O2	1.64517 (17)	0.47505 (10)	0.88309 (10)	0.0292 (3)
C1	0.3887 (3)	0.02593 (17)	0.16096 (17)	0.0356 (4)
H1A	0.518022	0.014659	0.180830	0.053*
H1B	0.344470	0.010513	0.078168	0.053*
H1C	0.296937	-0.035756	0.182195	0.053*
C2	0.5762 (2)	0.31022 (18)	0.20005 (16)	0.0315 (4)
H2A	0.556202	0.395409	0.222952	0.047*
H2B	0.555172	0.279906	0.116802	0.047*
H2C	0.709501	0.316690	0.239220	0.047*
C3	0.1549 (2)	0.21993 (14)	0.21951 (13)	0.0220 (3)
C4	0.1760 (2)	0.35451 (16)	0.29980 (15)	0.0290 (4)
H4A	0.228450	0.359582	0.379644	0.044*
H4B	0.048251	0.368942	0.289519	0.044*
H4C	0.264834	0.420774	0.281404	0.044*
C5	0.0722 (2)	0.21241 (16)	0.09442 (15)	0.0287 (4)
H5A	0.156028	0.282682	0.077753	0.043*
H5B	-0.059390	0.220714	0.083054	0.043*
H5C	0.067239	0.129025	0.042720	0.043*
C6	0.0151 (2)	0.11714 (17)	0.24914 (17)	0.0327 (4)
H6A	0.000713	0.030972	0.198313	0.049*
H6B	-0.112221	0.132184	0.238917	0.049*
H6C	0.067749	0.122664	0.329095	0.049*
C7	0.6508 (2)	0.21980 (15)	0.44469 (12)	0.0208 (3)
C8	0.8018 (2)	0.33469 (14)	0.50020 (13)	0.0237 (3)
H8	0.788337	0.412995	0.488878	0.028*
C9	0.9723 (2)	0.33456 (14)	0.57228 (13)	0.0226 (3)
H9	1.075023	0.413377	0.609708	0.027*
C10	0.9962 (2)	0.22092 (14)	0.59095 (12)	0.0196 (3)
C11	0.8430 (2)	0.10703 (14)	0.53377 (13)	0.0212 (3)
H11	0.855594	0.028445	0.544784	0.025*
C12	0.6725 (2)	0.10593 (14)	0.46116 (13)	0.0222 (3)
H12	0.570384	0.027005	0.422645	0.027*
C13	1.1790 (2)	0.22081 (13)	0.67256 (12)	0.0200 (3)
H13	1.167369	0.128847	0.667431	0.024*
C14	1.1985 (2)	0.29503 (15)	0.79754 (13)	0.0228 (3)
H14A	1.206850	0.386201	0.804842	0.027*
H14B	1.080957	0.256789	0.818228	0.027*
C15	1.3792 (2)	0.29093 (15)	0.88093 (13)	0.0230 (3)
H15A	1.394933	0.347476	0.959145	0.028*
H15B	1.361595	0.201524	0.883051	0.028*
C16	1.5596 (2)	0.33507 (13)	0.84517 (13)	0.0214 (3)
C17	1.5456 (2)	0.26890 (15)	0.72234 (13)	0.0232 (3)
H17A	1.539907	0.177436	0.711453	0.028*

H17B	1.663369	0.311268	0.703953	0.028*
C18	1.3644 (2)	0.27461 (14)	0.64068 (13)	0.0225 (3)
H18A	1.352443	0.223818	0.561100	0.027*
H18B	1.378788	0.365426	0.643765	0.027*
C19	1.7490 (2)	0.39772 (16)	0.93104 (15)	0.0285 (4)
H19A	1.865 (3)	0.394 (2)	0.9095 (18)	0.031 (5)*
H19B	1.758 (3)	0.4120 (18)	1.0091 (18)	0.024 (4)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Si1	0.0177 (2)	0.0210 (2)	0.0203 (3)	0.00703 (16)	0.00510 (16)	0.00808 (17)
O1	0.0212 (5)	0.0382 (6)	0.0223 (6)	0.0115 (5)	0.0069 (4)	0.0116 (5)
O2	0.0337 (6)	0.0184 (5)	0.0308 (6)	0.0041 (4)	0.0067 (5)	0.0055 (4)
C1	0.0330 (9)	0.0308 (9)	0.0370 (10)	0.0150 (7)	0.0022 (7)	0.0023 (7)
C2	0.0219 (8)	0.0424 (9)	0.0390 (10)	0.0118 (7)	0.0103 (7)	0.0245 (8)
C3	0.0176 (7)	0.0226 (7)	0.0274 (8)	0.0061 (5)	0.0066 (6)	0.0105 (6)
C4	0.0250 (8)	0.0288 (8)	0.0356 (9)	0.0123 (6)	0.0108 (7)	0.0089 (7)
C5	0.0227 (8)	0.0300 (8)	0.0316 (9)	0.0079 (6)	0.0020 (6)	0.0114 (7)
C6	0.0223 (8)	0.0334 (9)	0.0483 (11)	0.0068 (6)	0.0133 (7)	0.0220 (8)
C7	0.0196 (7)	0.0285 (7)	0.0164 (7)	0.0093 (6)	0.0062 (5)	0.0082 (6)
C8	0.0279 (8)	0.0219 (7)	0.0247 (8)	0.0096 (6)	0.0083 (6)	0.0105 (6)
C9	0.0226 (7)	0.0199 (7)	0.0234 (8)	0.0034 (5)	0.0053 (6)	0.0072 (6)
C10	0.0213 (7)	0.0212 (7)	0.0179 (7)	0.0072 (5)	0.0082 (6)	0.0064 (6)
C11	0.0242 (7)	0.0186 (7)	0.0226 (7)	0.0071 (5)	0.0089 (6)	0.0072 (6)
C12	0.0217 (7)	0.0220 (7)	0.0201 (7)	0.0032 (5)	0.0070 (6)	0.0043 (6)
C13	0.0211 (7)	0.0194 (7)	0.0206 (7)	0.0069 (5)	0.0066 (6)	0.0071 (6)
C14	0.0231 (7)	0.0267 (7)	0.0216 (8)	0.0092 (6)	0.0090 (6)	0.0090 (6)
C15	0.0252 (8)	0.0254 (7)	0.0200 (7)	0.0082 (6)	0.0072 (6)	0.0088 (6)
C16	0.0237 (7)	0.0183 (7)	0.0225 (8)	0.0072 (5)	0.0053 (6)	0.0073 (6)
C17	0.0209 (7)	0.0249 (7)	0.0230 (8)	0.0072 (6)	0.0073 (6)	0.0054 (6)
C18	0.0219 (7)	0.0270 (7)	0.0183 (7)	0.0065 (6)	0.0075 (6)	0.0063 (6)
C19	0.0250 (8)	0.0306 (8)	0.0254 (9)	0.0058 (6)	0.0040 (6)	0.0068 (7)

Geometric parameters (Å, °)

Si1—O1	1.6580 (11)	C8—C9	1.389 (2)
Si1—C1	1.8629 (17)	C9—C10	1.399 (2)
Si1—C2	1.8570 (16)	C10—C11	1.394 (2)
Si1—C3	1.8804 (15)	C10—C13	1.515 (2)
O1—C7	1.3735 (18)	C11—C12	1.387 (2)
O2—C16	1.4567 (17)	C13—C14	1.536 (2)
O2—C19	1.446 (2)	C13—C18	1.536 (2)
C3—C4	1.538 (2)	C14—C15	1.533 (2)
C3—C5	1.537 (2)	C15—C16	1.507 (2)
C3—C6	1.536 (2)	C16—C17	1.502 (2)
C7—C8	1.391 (2)	C16—C19	1.461 (2)
C7—C12	1.386 (2)	C17—C18	1.532 (2)

O1—Si1—C1	109.67 (8)	C9—C10—C13	121.75 (13)
O1—Si1—C2	109.22 (7)	C11—C10—C9	117.59 (14)
O1—Si1—C3	103.07 (6)	C11—C10—C13	120.64 (13)
C1—Si1—C3	112.59 (8)	C12—C11—C10	121.44 (14)
C2—Si1—C1	109.20 (9)	C7—C12—C11	120.11 (14)
C2—Si1—C3	112.88 (7)	C10—C13—C14	111.57 (12)
C7—O1—Si1	128.73 (9)	C10—C13—C18	112.72 (12)
C19—O2—C16	60.42 (10)	C18—C13—C14	110.02 (12)
C4—C3—Si1	108.90 (10)	C15—C14—C13	111.77 (12)
C5—C3—Si1	109.90 (10)	C16—C15—C14	111.09 (12)
C5—C3—C4	109.25 (13)	O2—C16—C15	113.99 (12)
C6—C3—Si1	110.17 (10)	O2—C16—C17	114.06 (12)
C6—C3—C4	108.91 (13)	O2—C16—C19	59.43 (10)
C6—C3—C5	109.69 (13)	C17—C16—C15	115.08 (12)
O1—C7—C8	120.15 (13)	C19—C16—C15	120.42 (14)
O1—C7—C12	120.07 (13)	C19—C16—C17	120.66 (13)
C12—C7—C8	119.63 (14)	C16—C17—C18	111.04 (12)
C9—C8—C7	119.77 (14)	C17—C18—C13	111.41 (12)
C8—C9—C10	121.45 (14)	O2—C19—C16	60.15 (10)
O1—Si1—C3—C4	54.6 (1)	C8—C7—C12—C11	0.8 (2)
O1—Si1—C3—C5	174.2 (1)	C7—C12—C11—C10	-0.5 (2)
O1—Si1—C3—C6	-64.8 (1)	C9—C10—C13—C14	66.7 (2)
O1—C7—C8—C9	175.0 (1)	C9—C10—C13—C18	-57.7 (2)
O1—C7—C12—C11	-174.6 (1)	C11—C10—C13—C14	-111.7 (2)
C1—Si1—C3—C4	172.7 (1)	C11—C10—C13—C18	124.0 (2)
C1—Si1—C3—C5	-67.7 (1)	C8—C9—C10—C13	-177.9 (1)
C1—Si1—C3—C6	53.3 (1)	C12—C11—C10—C13	178.2 (1)
C2—Si1—C3—C4	-63.1 (1)	C13—C14—C15—C16	52.9 (2)
C2—Si1—C3—C5	56.5 (1)	C14—C15—C16—C17	-51.4 (2)
C2—Si1—C3—C6	177.5 (1)	C15—C16—C17—C18	52.0 (2)
C1—Si1—O1—C7	62.5 (1)	C16—C17—C18—C13	-54.1 (2)
C2—Si1—O1—C7	-57.2 (1)	C17—C18—C13—C14	56.7 (2)
C3—Si1—O1—C7	-177.40 (14)	C18—C13—C14—C15	-56.2 (2)
Si1—O1—C7—C8	95.0 (2)	O2—C16—C17—C18	-82.4 (2)
Si1—O1—C7—C12	-89.6 (2)	O2—C16—C15—C14	83.1 (2)
C7—C8—C9—C10	-0.2 (2)	C19—C16—C17—C18	-149.9 (1)
C8—C9—C10—C11	0.5 (2)	C19—C16—C15—C14	150.5 (1)
C9—C10—C11—C12	-0.2 (2)	C10—C13—C14—C15	177.9 (1)
C9—C8—C7—C12	-0.5 (2)	C10—C13—C18—C17	-178.07 (13)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C15—H15A \cdots O2 ⁱ	0.99	2.54	3.487 (2)	159

Symmetry code: (i) $-x+3, -y+1, -z+2$.