

Received 29 August 2023  
Accepted 6 September 2023

Edited by W. T. A. Harrison, University of Aberdeen, United Kingdom

**Keywords:** crystal structure; hydrogen bonding; C—H···π interactions.

**CCDC reference:** 2290063

**Structural data:** full structural data are available from iucrdata.iucr.org

## (2,4-Dichlorobenzylidene)[2-(1*H*-indol-3-yl)ethyl]amine

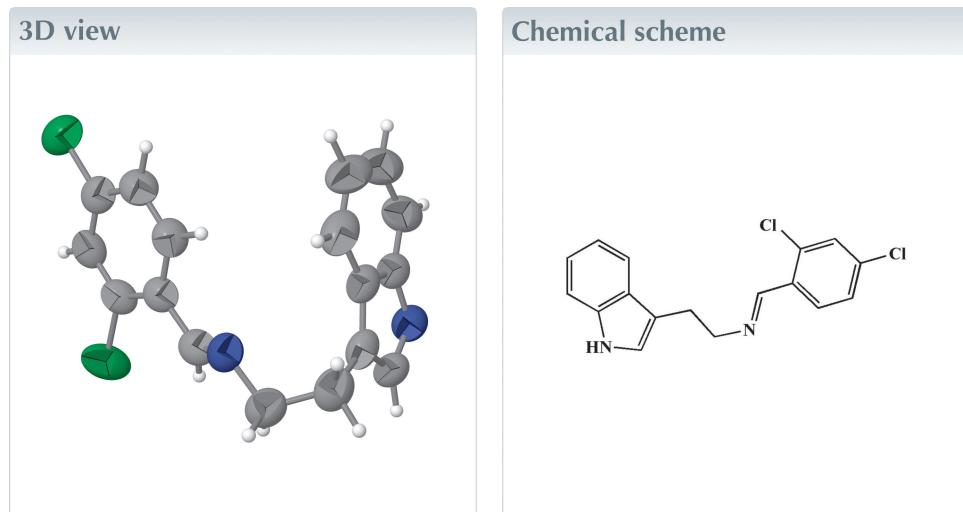
Suganya Murugan,<sup>a</sup> Anaglit Catherine Paul,<sup>b</sup> Themmila Khamrang,<sup>c</sup> Savaridasan Jose Kavitha,<sup>b</sup> Venkatachalam Rajakannan<sup>d</sup> and Madhukar Hemamalini<sup>b\*</sup>

<sup>a</sup>Department of Chemistry, Government Arts and Science College for Women, Kodaikanal, Tamil Nadu, India,

<sup>b</sup>Department of Chemistry, Mother Teresa Women's University, Kodaikanal, Tamil Nadu, India, <sup>c</sup>Assistant Professor, Department of Chemistry, DM College of Science, Dhanamanjuri University, Imphal, Manipur-795 001, India, and

<sup>d</sup>Department of Crystallography and Biophysics, University of Madras, Guindy Campus, Chennai-600 025, Tamil Nadu, India. \*Correspondence e-mail: hemamalini2k3@yahoo.com

In the title compound,  $C_{17}H_{14}Cl_2N_2$ , the molecule exists in an *E* configuration with respect to the C≡N bond of the Schiff base fragment. The dihedral angle between the indole ring system and the benzene ring is  $80.86(12)^\circ$ . In the crystal, molecules are connected by N—H···N hydrogen bonds, generating a *C*(7) chain extending along the *a*-axis direction. No aromatic π–π stacking occurs but weak C—H···π interactions are observed.



### Structure description

Schiff bases are widely used as catalysts, corrosion inhibitors and intermediates in organic synthesis, and also play a potential role in the development of coordination chemistry (Muralisankar *et al.*, 2016). Indole and its derivatives are useful starting compounds to derive pharmaceutical (Nalli *et al.*, 2020) and biological (Arumugam *et al.*, 2021) materials. In the present study, the hydrogen-bonding interactions and C—H···π interactions of the title compound are investigated.

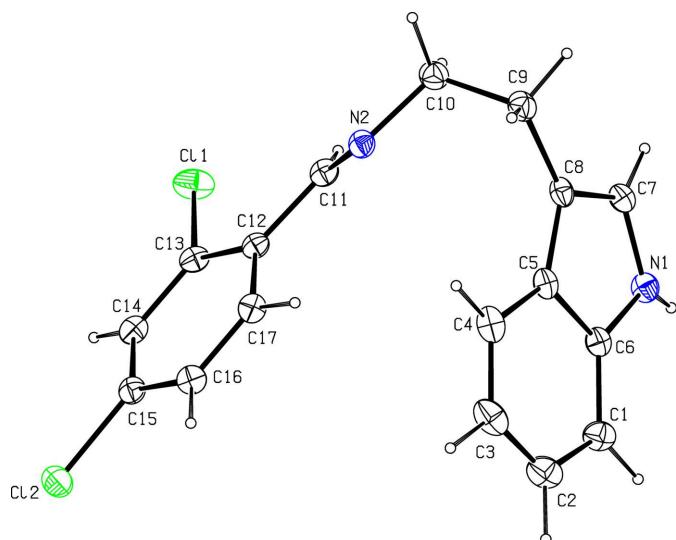
The asymmetric unit of the title compound is shown in Fig. 1. The C≡N double bond adopts an *E* configuration. The bond lengths and angles in the title molecule are normal and agree with those in other indole-imine compounds (*e.g.*, Suresh *et al.*, 2016; Ho *et al.*, 2006). The dihedral angle between the C1—C8/N1 indole ring system and the C12—C17 benzene ring is  $80.86(10)^\circ$ .

In the extended structure, the N1—H5 group is a hydrogen-bond donor to atom N2 of the imino group (Table 1). These hydrogen bonds generate a *C*(7) chain extending along the *a*-axis direction, as shown in Fig. 2. There are no π–π interactions in this crystal structure but weak C—H···π interactions occur.



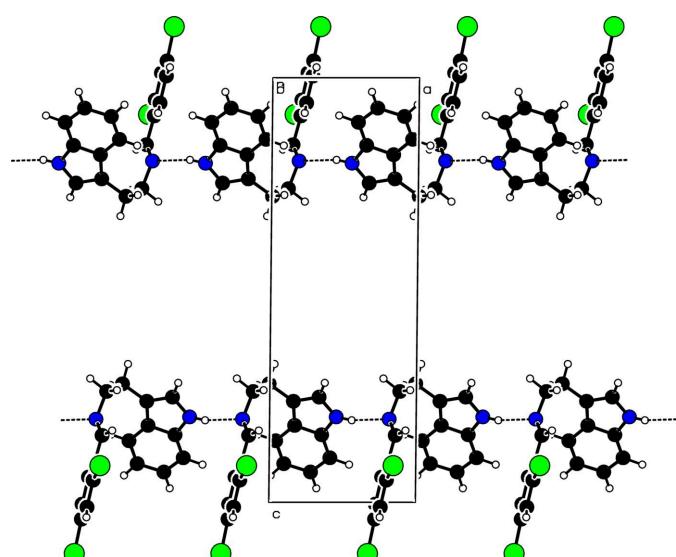
OPEN ACCESS

Published under a CC BY 4.0 licence

**Figure 1**

The molecular structure of the title compound showing 50% displacement ellipsoids.

A search of the Cambridge Structural Database (Version 5.43, update November 2022; Groom *et al.*, 2016) for the benzylidene)-[2-(1*H*-indol-3-yl)-ethyl]-amine skeleton yielded the hits 1-(anthracen-9-yl)-*N*-[2-(1*H*-indol-3-yl)ethyl]methanimine (CSD refcode TEGJIB; Faizi *et al.*, 2017), 2-[2-(1*H*-indol-3-ylethyliminomethyl]-5-methylphenol (PEVXEW; Brink *et al.*, 2018), *rac*-4-[(*E*)-[1-cyano-1-cyclohexyl-2-(1*H*-indol-3-yl)ethyl]iminomethyl] benzonitrile (OCEWIE; Letessier *et al.*, 2011), 1*H*-indole-3-ethylenesalicylaldimine (FAJVIV; Rodriguez *et al.*, 1987) and 1-(4-chlorophenyl)-2-[(2-(1*H*-indol-3-yl)ethyl)imino]-2-(4-methoxyphenyl)ethan-1-one (AZUYUS; Li *et al.*, 2021).

**Figure 2**

Partial packing diagram for the title compound showing the formation of [100] hydrogen-bonded chains.

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D\cdots H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N1-H5\cdots N2^i$	0.83 (3)	2.17 (3)	2.971 (3)	163 (2)

Symmetry code: (i)  $x - 1, y, z$ .

**Table 2**  
Experimental details.

Crystal data	$C_{17}H_{14}Cl_2N_2$
Chemical formula	$C_{17}H_{14}Cl_2N_2$
$M_r$	317.20
Crystal system, space group	Monoclinic, $P2_1/n$
Temperature (K)	296
$a, b, c$ ( $\text{\AA}$ )	7.2107 (8), 10.2179 (13), 20.863 (3)
$\beta$ ( $^\circ$ )	90.562 (4)
$V$ ( $\text{\AA}^3$ )	1537.1 (3)
$Z$	4
Radiation type	Mo $K\alpha$
$\mu$ ( $\text{mm}^{-1}$ )	0.42
Crystal size (mm)	0.52 $\times$ 0.34 $\times$ 0.13
Data collection	Agilent Xcalibur, Atlas, Gemini Multi-scan (SADABS; Krause <i>et al.</i> , 2015)
Diffractometer	Agilent Xcalibur, Atlas, Gemini Multi-scan (SADABS; Krause <i>et al.</i> , 2015)
Absorption correction	68672, 3872, 1946
$T_{\min}, T_{\max}$	0.631, 0.746
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	3872
$R_{\text{int}}$	0.091
$(\sin \theta/\lambda)_{\max}$ ( $\text{\AA}^{-1}$ )	0.671
Refinement	0.048, 0.134, 1.01
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	3872
No. of reflections	246
No. of parameters	All H-atom parameters refined
H-atom treatment	$\Delta\rho_{\max}, \Delta\rho_{\min}$ ( $e \text{\AA}^{-3}$ )
$\Delta\rho_{\max}, \Delta\rho_{\min}$ ( $e \text{\AA}^{-3}$ )	0.20, -0.27

Computer programs: *CrysAlis PRO* and *CrysAlis RED* (Agilent, 2012), *SHELXT2018/2* (Sheldrick, 2015a), *SHELXL2018/3* (Sheldrick, 2015b) and *PLATON* (Spek, 2020).

## Synthesis and crystallization

The title compound was synthesized by condensing tryptamine, 2-(1*H*-indol-3-yl)ethan-1-amine (0.01 mmol) and 2,4-dichlorobenzaldehyde (0.01 mmol), which were taken separately, dissolved in 40 ml of ethanol, then mixed, and heated on a water bath for one h, then kept for crystallization. After a few days, colourless plate-shaped crystals were obtained.

## Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

## Acknowledgements

MH thanks SERB-IRE for financial support (Ref. No. SIR/2022/000011]. SJK thanks TANSCHE for financial support (File No. RGP/2019–20/MTWU/ HECP-0080).

## Funding information

Funding for this research was provided by: Department of Science and Technology, Ministry of Science and Technology,

India, Science and Engineering Research Board (grant No. SIR/2022/000011); Tamil Nadu State Council for Higher Education (grant No. RGP/2019-20/MTWU/HECP-0080).

## References

- Agilent (2012). *CrysAlis PRO* and *CrysAlis RED*. Agilent Technologies Ltd, Yarnton, England.
- Arumugam, N., Almansour, A. I., Kumar, R. S., Yeswanthkumar, S., Padmanaban, R., Arun, Y., Kansiz, S., Dege, N., Manohar, T. S. & Venketesh, S. (2021). *J. Mol. Struct.* **1225**, 129165–129166.
- Brink, A., Kroon, R. E., Visser, H. G., van Rensburg, C. E. J. & Roodt, A. (2018). *New J. Chem.* **42**, 5193–5203.
- Faizi, M. S. H., Dege, N., Malinkin, S. & Sliva, T. Y. (2017). *Acta Cryst. E* **73**, 1329–1332.
- Groom, C. R., Bruno, I. J., Lightfoot, M. P. & Ward, S. C. (2016). *Acta Cryst. B* **72**, 171–179.
- Ho, J. J., Black, D. St C., Messerle, B. A., Clegg, J. K. & Turner, P. T. (2006). *Organometallics*, **25**, 5800–5810.
- Krause, L., Herbst-Irmer, R., Sheldrick, G. M. & Stalke, D. (2015). *J. Appl. Cryst.* **48**, 3–10.
- Letessier, J., Schollmeyer, D., Detert, H. & Opitz, T. (2011). *Acta Cryst. E* **67**, o3435.
- Li, L., Zhang, S., Deng, X., Li, G., Tang, Z. & Zhao, G. (2021). *Org. Lett.* **23**, 6819–6824.
- Muralisankar, M., Haribabu, J., Bhuvanesh, N. S. P., Karvembu, R. & Sreekanth, A. (2016). *Inorg. Chim. Acta*, **449**, 82–95.
- Nalli, M., Armijos Rivera, J. I., Masci, D., Coluccia, A., Badia, R., Riveira-Muñoz, E., Brambilla, A., Cinquina, E., Turriziani, O., Falasca, F., Catalano, M., Limatola, C., Esté, J. A., Maga, G., Silvestri, R., Crespan, E. & La Regina, G. (2020). *Eur. J. Med. Chem.* **208**, 112696.
- Rodriguez, M. L., Medina de la Rosa, E., Gili, P., Zarza, P. M., Reyes, M. G. M., Medina, A. & Díaz González, M. C. (1987). *Acta Cryst. C* **43**, 134–136.
- Sheldrick, G. M. (2015a). *Acta Cryst. A* **71**, 3–8.
- Sheldrick, G. M. (2015b). *Acta Cryst. C* **71**, 3–8.
- Spek, A. L. (2020). *Acta Cryst. E* **76**, 1–11.
- Suresh, D., Ferreira, B., Lopes, P. S., Gomes, C. S. B., Krishnamoorthy, P., Charas, A., Vila-Viçosa, D., Morgado, J., Calhorda, M. J., Maçanita, A. L. & Gomes, P. T. (2016). *Dalton Trans.* **45**, 15603–15620.

# full crystallographic data

*IUCrData* (2023). **8**, x230780 [https://doi.org/10.1107/S2414314623007800]

## (2,4-Dichlorobenzylidene)[2-(1*H*-indol-3-yl)ethyl]amine

Suganya Murugan, Anaglit Catherine Paul, Themmila Khamrang, Savaridasan Jose Kavitha, Venkatachalam Rajakannan and Madhukar Hemamalini

### (2,4-Dichlorobenzylidene)[2-(1*H*-indol-3-yl)ethyl]amine

#### Crystal data

$C_{17}H_{14}Cl_2N_2$   
 $M_r = 317.20$   
Monoclinic,  $P2_1/n$   
 $a = 7.2107 (8)$  Å  
 $b = 10.2179 (13)$  Å  
 $c = 20.863 (3)$  Å  
 $\beta = 90.562 (4)^\circ$   
 $V = 1537.1 (3)$  Å<sup>3</sup>  
 $Z = 4$

$F(000) = 656$   
 $D_x = 1.371$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å  
Cell parameters from 3778 reflections  
 $\theta = 2.6\text{--}29.9^\circ$   
 $\mu = 0.42$  mm<sup>-1</sup>  
 $T = 296$  K  
Plate, colourless  
0.52 × 0.34 × 0.13 mm

#### Data collection

Agilent Xcalibur, Atlas, Gemini  
diffractometer  
Radiation source: fine-focus sealed tube  
 $\omega$  scans  
Absorption correction: multi-scan  
(SADABS; Krause *et al.*, 2015)  
 $T_{\min} = 0.631$ ,  $T_{\max} = 0.746$   
68672 measured reflections

3872 independent reflections  
1946 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.091$   
 $\theta_{\max} = 28.5^\circ$ ,  $\theta_{\min} = 2.0^\circ$   
 $h = -9 \rightarrow 9$   
 $k = -13 \rightarrow 13$   
 $l = -27 \rightarrow 27$

#### Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.134$   
 $S = 1.01$   
3872 reflections  
246 parameters  
0 restraints

Primary atom site location: dual  
Hydrogen site location: difference Fourier map  
All H-atom parameters refined  
 $w = 1/[\sigma^2(F_o^2) + (0.0498P)^2 + 0.4436P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.20$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.26$  e Å<sup>-3</sup>

#### Special details

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** All the H atoms were located in a difference Fourier map and allowed to refine freely (C—H = 0.93–0.96 and N—H = 0.83 Å).

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl2	0.82313 (10)	0.46440 (8)	0.37941 (3)	0.0896 (3)
Cl1	0.65404 (13)	0.74406 (7)	0.58638 (4)	0.1034 (3)
N2	0.6843 (2)	0.3930 (2)	0.69593 (9)	0.0656 (5)
N1	0.0458 (3)	0.2537 (2)	0.70118 (10)	0.0701 (6)
C5	0.3287 (3)	0.1772 (2)	0.68119 (11)	0.0628 (6)
C8	0.3389 (3)	0.2540 (2)	0.73804 (11)	0.0648 (6)
C6	0.1440 (3)	0.1792 (2)	0.65910 (11)	0.0631 (6)
C12	0.6934 (3)	0.4814 (2)	0.58989 (12)	0.0593 (6)
C7	0.1639 (3)	0.2993 (3)	0.74787 (13)	0.0680 (7)
C13	0.6992 (3)	0.5929 (2)	0.55222 (12)	0.0641 (6)
C14	0.7393 (3)	0.5892 (3)	0.48855 (14)	0.0683 (7)
C15	0.7767 (3)	0.4709 (3)	0.46026 (12)	0.0648 (6)
C17	0.7315 (3)	0.3640 (3)	0.55950 (14)	0.0675 (7)
C11	0.6416 (3)	0.4833 (3)	0.65769 (13)	0.0661 (7)
C16	0.7732 (3)	0.3571 (3)	0.49571 (14)	0.0707 (7)
C4	0.4574 (4)	0.1057 (3)	0.64506 (16)	0.0818 (8)
C10	0.6134 (4)	0.4015 (3)	0.76114 (14)	0.0785 (8)
C1	0.0861 (5)	0.1121 (3)	0.60482 (14)	0.0836 (8)
C3	0.4003 (6)	0.0415 (3)	0.59108 (17)	0.0979 (11)
C9	0.5055 (4)	0.2801 (3)	0.77920 (14)	0.0796 (8)
C2	0.2174 (6)	0.0439 (3)	0.57140 (17)	0.0972 (10)
H12	0.720 (3)	0.289 (3)	0.5830 (12)	0.082 (8)*
H11	0.573 (3)	0.556 (2)	0.6692 (10)	0.065 (7)*
H10	0.543 (4)	0.476 (3)	0.7641 (12)	0.081 (9)*
H9	0.726 (4)	0.407 (3)	0.7916 (13)	0.099 (9)*
H8	0.586 (4)	0.201 (3)	0.7761 (12)	0.091 (9)*
H4	0.579 (4)	0.105 (3)	0.6598 (12)	0.083 (9)*
H13	0.797 (3)	0.275 (3)	0.4752 (12)	0.076 (8)*
H14	0.743 (3)	0.663 (3)	0.4649 (12)	0.082 (8)*
H7	0.470 (3)	0.290 (2)	0.8258 (12)	0.078 (7)*
H6	0.126 (3)	0.357 (2)	0.7814 (10)	0.068 (7)*
H1	-0.045 (4)	0.108 (3)	0.5888 (13)	0.106 (10)*
H3	0.489 (4)	-0.005 (3)	0.5634 (15)	0.119 (11)*
H2	0.177 (5)	-0.006 (4)	0.5331 (17)	0.132 (13)*
H5	-0.060 (4)	0.282 (3)	0.6933 (12)	0.082 (9)*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl2	0.0795 (5)	0.1146 (6)	0.0746 (5)	0.0090 (4)	0.0029 (3)	0.0035 (4)
Cl1	0.1558 (8)	0.0600 (4)	0.0939 (6)	0.0151 (4)	-0.0242 (5)	-0.0064 (4)
N2	0.0525 (11)	0.0758 (14)	0.0685 (13)	0.0038 (10)	-0.0010 (9)	0.0048 (11)
N1	0.0601 (13)	0.0747 (15)	0.0757 (15)	0.0114 (12)	0.0050 (11)	0.0011 (11)
C5	0.0652 (15)	0.0551 (14)	0.0681 (15)	0.0121 (11)	0.0125 (11)	0.0153 (12)
C8	0.0605 (14)	0.0739 (16)	0.0602 (14)	0.0068 (12)	0.0087 (11)	0.0181 (13)

C6	0.0695 (15)	0.0528 (14)	0.0671 (15)	0.0071 (12)	0.0078 (12)	0.0093 (12)
C12	0.0467 (12)	0.0621 (15)	0.0690 (15)	0.0059 (11)	-0.0094 (10)	-0.0004 (12)
C7	0.0707 (16)	0.0718 (17)	0.0617 (15)	0.0074 (13)	0.0119 (13)	0.0002 (13)
C13	0.0644 (14)	0.0555 (15)	0.0721 (16)	0.0053 (11)	-0.0151 (12)	0.0018 (13)
C14	0.0659 (16)	0.0587 (16)	0.0800 (19)	-0.0008 (12)	-0.0136 (13)	0.0104 (15)
C15	0.0478 (13)	0.0780 (18)	0.0684 (15)	0.0042 (12)	-0.0071 (10)	0.0065 (14)
C17	0.0659 (15)	0.0596 (16)	0.0769 (18)	0.0066 (12)	-0.0027 (12)	0.0089 (14)
C11	0.0534 (14)	0.0663 (17)	0.0783 (18)	0.0082 (12)	-0.0057 (12)	-0.0042 (14)
C16	0.0676 (16)	0.0640 (17)	0.0803 (19)	0.0123 (13)	-0.0014 (13)	-0.0037 (15)
C4	0.079 (2)	0.0709 (18)	0.096 (2)	0.0228 (15)	0.0201 (17)	0.0209 (17)
C10	0.0702 (18)	0.096 (2)	0.0690 (18)	0.0042 (17)	-0.0010 (14)	-0.0028 (16)
C1	0.102 (2)	0.0655 (17)	0.083 (2)	0.0028 (17)	-0.0044 (17)	-0.0025 (16)
C3	0.138 (3)	0.0652 (19)	0.091 (2)	0.031 (2)	0.032 (2)	-0.0007 (17)
C9	0.0741 (18)	0.098 (2)	0.0663 (18)	0.0031 (16)	-0.0018 (14)	0.0154 (16)
C2	0.137 (3)	0.0653 (19)	0.090 (2)	0.010 (2)	0.006 (2)	-0.0077 (17)

Geometric parameters ( $\text{\AA}$ , °)

Cl2—C15	1.724 (3)	C14—H14	0.90 (3)
Cl1—C13	1.733 (2)	C15—C16	1.378 (4)
N2—C11	1.256 (3)	C17—C16	1.369 (4)
N2—C10	1.461 (3)	C17—H12	0.92 (3)
N1—C6	1.365 (3)	C11—H11	0.92 (2)
N1—C7	1.369 (3)	C16—H13	0.95 (3)
N1—H5	0.83 (3)	C4—C3	1.364 (5)
C5—C6	1.405 (3)	C4—H4	0.93 (3)
C5—C4	1.406 (4)	C10—C9	1.514 (4)
C5—C8	1.424 (3)	C10—H10	0.92 (3)
C8—C7	1.362 (3)	C10—H9	1.03 (3)
C8—C9	1.494 (4)	C1—C2	1.371 (4)
C6—C1	1.385 (4)	C1—H1	1.00 (3)
C12—C13	1.385 (3)	C3—C2	1.377 (5)
C12—C17	1.386 (3)	C3—H3	0.99 (3)
C12—C11	1.467 (3)	C9—H8	1.00 (3)
C7—H6	0.96 (2)	C9—H7	1.01 (2)
C13—C14	1.363 (4)	C2—H2	0.99 (4)
C14—C15	1.373 (4)		
C11—N2—C10	117.5 (2)	N2—C11—C12	122.7 (2)
C6—N1—C7	108.9 (2)	N2—C11—H11	123.5 (14)
C6—N1—H5	123.4 (18)	C12—C11—H11	113.8 (14)
C7—N1—H5	125.7 (19)	C17—C16—C15	118.9 (3)
C6—C5—C4	117.4 (3)	C17—C16—H13	121.4 (15)
C6—C5—C8	107.8 (2)	C15—C16—H13	119.6 (15)
C4—C5—C8	134.7 (3)	C3—C4—C5	119.8 (3)
C7—C8—C5	105.8 (2)	C3—C4—H4	123.2 (17)
C7—C8—C9	126.5 (3)	C5—C4—H4	117.0 (17)
C5—C8—C9	127.7 (2)	N2—C10—C9	111.6 (3)

N1—C6—C1	130.4 (3)	N2—C10—H10	108.1 (16)
N1—C6—C5	107.1 (2)	C9—C10—H10	112.3 (17)
C1—C6—C5	122.5 (2)	N2—C10—H9	107.3 (15)
C13—C12—C17	116.5 (2)	C9—C10—H9	107.4 (16)
C13—C12—C11	123.1 (2)	H10—C10—H9	110 (2)
C17—C12—C11	120.4 (2)	C2—C1—C6	117.6 (3)
C8—C7—N1	110.4 (2)	C2—C1—H1	117.6 (17)
C8—C7—H6	126.3 (14)	C6—C1—H1	124.7 (17)
N1—C7—H6	123.3 (13)	C4—C3—C2	121.2 (3)
C14—C13—C12	122.5 (2)	C4—C3—H3	121.5 (19)
C14—C13—C11	117.9 (2)	C2—C3—H3	117.2 (19)
C12—C13—C11	119.5 (2)	C8—C9—C10	114.6 (2)
C13—C14—C15	119.2 (3)	C8—C9—H8	106.5 (16)
C13—C14—H14	121.4 (17)	C10—C9—H8	110.3 (15)
C15—C14—H14	119.4 (17)	C8—C9—H7	111.0 (13)
C14—C15—C16	120.5 (3)	C10—C9—H7	107.0 (14)
C14—C15—Cl2	119.7 (2)	H8—C9—H7	107 (2)
C16—C15—Cl2	119.8 (2)	C1—C2—C3	121.4 (3)
C16—C17—C12	122.4 (3)	C1—C2—H2	118 (2)
C16—C17—H12	120.0 (17)	C3—C2—H2	121 (2)
C12—C17—H12	117.5 (16)		
C6—C5—C8—C7	-0.3 (3)	C13—C14—C15—Cl2	178.70 (17)
C4—C5—C8—C7	179.1 (3)	C13—C12—C17—C16	0.2 (3)
C6—C5—C8—C9	179.4 (2)	C11—C12—C17—C16	177.4 (2)
C4—C5—C8—C9	-1.1 (4)	C10—N2—C11—C12	-175.7 (2)
C7—N1—C6—C1	179.5 (3)	C13—C12—C11—N2	-159.7 (2)
C7—N1—C6—C5	0.9 (3)	C17—C12—C11—N2	23.3 (4)
C4—C5—C6—N1	-179.9 (2)	C12—C17—C16—C15	-0.3 (4)
C8—C5—C6—N1	-0.4 (3)	C14—C15—C16—C17	0.1 (4)
C4—C5—C6—C1	1.4 (4)	Cl2—C15—C16—C17	-178.35 (18)
C8—C5—C6—C1	-179.1 (2)	C6—C5—C4—C3	-0.3 (4)
C5—C8—C7—N1	0.9 (3)	C8—C5—C4—C3	-179.7 (3)
C9—C8—C7—N1	-178.9 (2)	C11—N2—C10—C9	124.4 (3)
C6—N1—C7—C8	-1.1 (3)	N1—C6—C1—C2	-179.7 (3)
C17—C12—C13—C14	0.2 (3)	C5—C6—C1—C2	-1.4 (4)
C11—C12—C13—C14	-176.9 (2)	C5—C4—C3—C2	-0.7 (5)
C17—C12—C13—Cl1	-179.82 (17)	C7—C8—C9—C10	-89.3 (3)
C11—C12—C13—Cl1	3.1 (3)	C5—C8—C9—C10	91.0 (3)
C12—C13—C14—C15	-0.4 (4)	N2—C10—C9—C8	-60.8 (4)
Cl1—C13—C14—C15	179.62 (17)	C6—C1—C2—C3	0.3 (5)
C13—C14—C15—C16	0.2 (4)	C4—C3—C2—C1	0.8 (5)

---

*Hydrogen-bond geometry (Å, °)*

D—H···A	D—H	H···A	D···A	D—H···A

---

N1—H5···N2 <sup>i</sup>	0.83 (3)	2.17 (3)	2.971 (3)	163 (2)
-------------------------	----------	----------	-----------	---------

Symmetry code: (i)  $x-1, y, z$ .