

rac-Ethyl *rel*-(2*R*,3*R*,4*S*)-4-hydroxy-1,2-dimethyl-5-oxopyrrolidine-3-carboxylate

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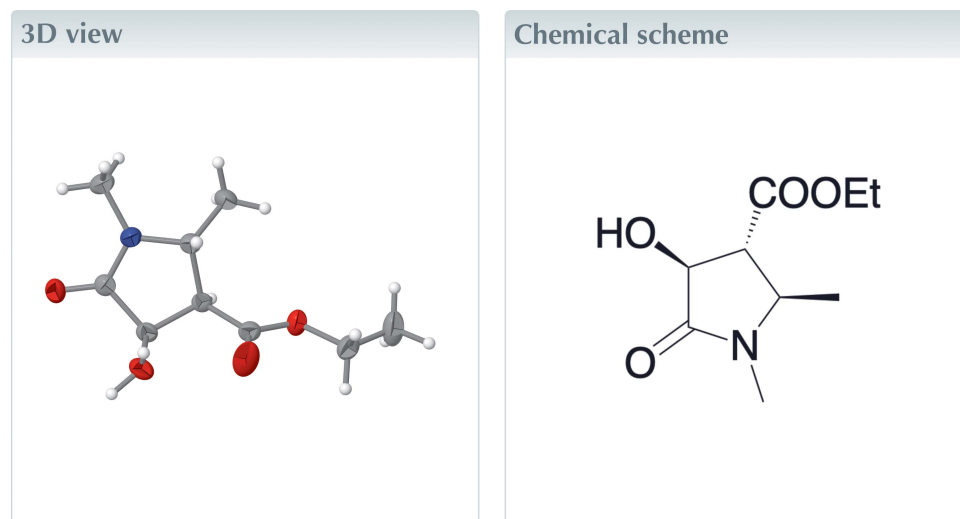
Keywords: crystal structure; oxopyrrolidine; ring conformation; hydrogen bonds.

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The asymmetric unit of the title compound, C₉H₁₅NO₄, consists of a functionalized pyrrolidine ring having an envelope conformation, synthesized as an ethyl ester. The molecule has three chiral centres and crystallized as a racemic mixture. In the crystal, molecules are linked by pairwise O—H···O bonds, generating dimers with twofold rotational symmetry.



Structure description

The heterocyclic compound 2-oxopyrrolidine and its derivatives have generated a lot of interest because of their practical significance (Pandya & Desai, 2020). These compounds have shown to be effective analgesics, anti-inflammatory (Salgın-Gökşen *et al.*, 2007), antiviral (Tian *et al.*, 2009), antimicrobial (Özkay *et al.*, 2010; Salgın-Gökşen *et al.*, 2007), antitumor (Abdel-Aziz *et al.*, 2021), anticonvulsant (Angelova *et al.*, 2016), antidepressant (Kulandasamy *et al.*, 2009), cardioprotective (Ghazouani *et al.*, 2019) and antiplatelet agents (Mashayekhi *et al.*, 2013; Ghazouani *et al.*, 2019).

During the course of our study towards pyrrolidine-based iminosugars, we have synthesized the title compound by reduction of 2,3-dioxopyrrolidine (Bacho *et al.*, 2020; Abdul Rashid *et al.*, 2020). The starting material, 2,3-dioxopyrrolidine, was initially prepared *via* a multicomponent reaction, according to a previously reported procedure (Mohammad *et al.*, 2009, 2011).

The title compound crystallizes in the monoclinic crystal system, space group *C*2/*c*, with one molecule in the asymmetric unit (Fig. 1). The pyrrolidine ring (C1–C4/N1) adopts an envelope conformation, with atom C4 deviating by 0.180 (1) Å from the mean

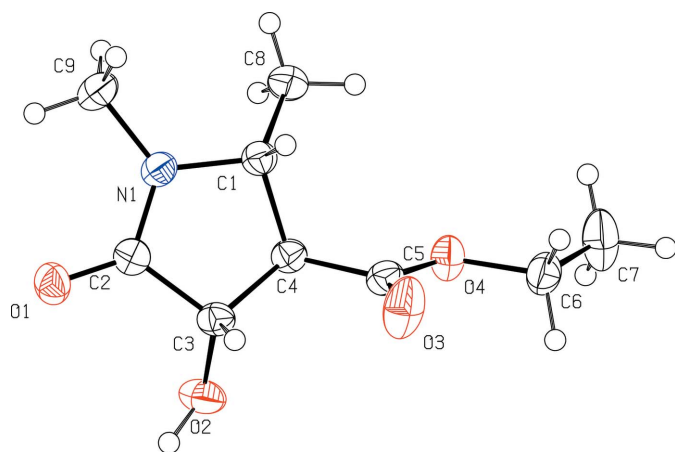


Figure 1
Crystal structure of the title compound, showing the atom-labelling scheme and displacement ellipsoids at the 50% probability level.

plane. There are three chiral centres within the ring, at C4, with a C1–C4–C5–O4 torsion angle of $-94.04(11)^\circ$. The methyl and hydroxyl groups, attached to C1 and C3, respectively, are orientated away from the mean plane with C2–N1–C1–C8 and N1–C2–C3–O2 torsion angles of $142.07(10)$ and $-135.48(10)^\circ$, respectively. Meanwhile, the

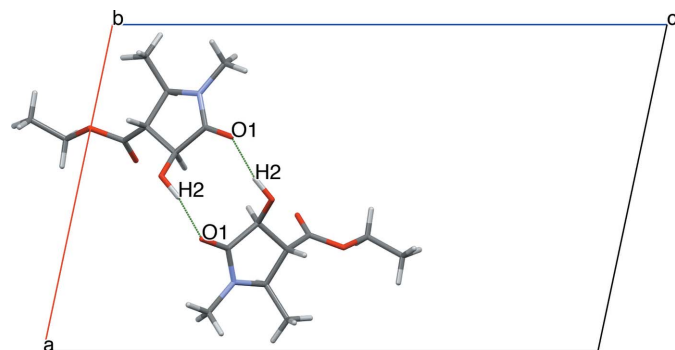


Figure 2
The O–H...O hydrogen bonds, indicated by green dashed bonds, forming $R_2^2(10)$ motifs in the crystal.

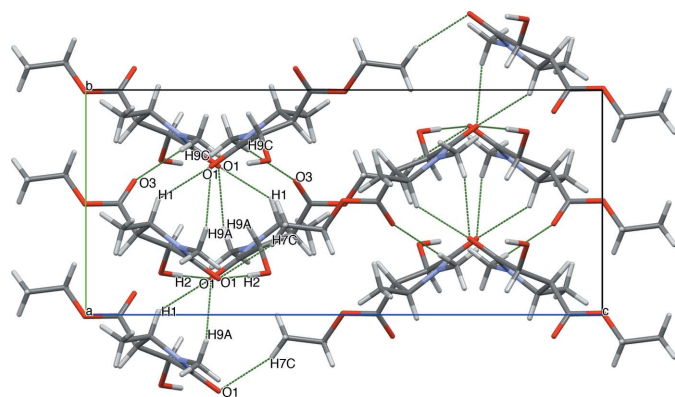


Figure 3
The molecular packing of title compound, viewed down the *a* axis. Intermolecular hydrogen bonds are indicated by green dashed lines.

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
O2–H2...O1 ⁱ	0.97 (1)	1.78 (1)	2.7405 (12)	170 (2)
C1–H1...O1 ⁱⁱⁱ	1.00	2.62	3.3953 (14)	134
C9–H9A...O1 ⁱⁱⁱ	0.98	2.51	3.3355 (16)	142
C9–H9C...O3 ⁱⁱⁱ	0.98	2.54	3.5086 (15)	169
C7–H7C...O1 ^{iv}	0.98	2.58	3.5134 (17)	160

Symmetry codes: (i) $-x + 1, y, -z + \frac{1}{2}$; (ii) $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$; (iii) $x + \frac{1}{2}, y - \frac{1}{2}, z$; (iv) $x, -y + 1, z + \frac{1}{2}$.

ethyl ester group (O3/C5/O4/C6/C7) occupies the equatorial position on the pyrrolidine ring at C1, C3, and C4. All bond lengths (Allen *et al.*, 1987) and angles in the molecule show normal values.

In the crystal, the molecules are linked by pairwise O–H...O hydrogen bonds, involving the carbonyl and hydroxy groups, forming centrosymmetric $R_2^2(10)$ ring motifs (Table 1, entry 1; Fig. 2). The packing also features C–H...O hydrogen bonds (Table 1), forming zigzag motifs propagating along the *c*-axis direction (Fig. 3).

Synthesis and crystallization

A solution of 2,3-dioxopyrrolidine (2.00 g, 10.04 mmol) together with Pd–C (10% wt; 1.39 g, 1.31 mmol) and acetic acid (4.59 ml, 80.32 mmol) was stirred in ethanol. The reaction was stirred vigorously under a hydrogen atmosphere to completion (24 h) and then filtered through Celite. After removal of the solvent, the crude product was purified by flash column chromatography on silica gel using ethyl acetate/petroleum ether (9/1), to afford two compounds; *trans*-hydroxyester **1** as a white solid and *cis*-hydroxyester **2** as a colourless oil. The white solid of *trans*-hydroxyester **1** was recrystallized from methanol solution to give single crystals of the title compound **1** (0.24 g, 12%).

trans-hydroxyester **1**: $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 4.57 (*d*, $J = 8.5$ Hz, 1H), 4.22 (*q*, $J = 6.9$ Hz, 2H), 3.63 (*s*, 1H), 2.82 (*s*, 3H), 2.67 (*t*, $J = 8.4$ Hz, 1H), 1.37 (*d*, $J = 3.7$ Hz, 3H), 1.29 (*t*, $J = 6.9$ Hz, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 173.00 (C=O), 171.54 (C=O), 72.26 (CHOH), 61.71 (OCH₂), 54.35 (CH), 31.23 (CHCH₃), 27.33 (CH₃N), 19.31 (CH₃), 14.27 (CH₃); GCMS *m/z* (EI, +ve): found: 201.10 ($[M]^+$), calculated for C₉H₁₅NO₄: 201.10.

cis-hydroxyester **2**: (0.50 g, 25%). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 4.44 (*d*, $J = 7.3$ Hz, 1H), 4.19 (*td*, $J = 7.2, 4.9$ Hz, 2H), 3.74 (*t*, $J = 6.6$ Hz, 1H), 3.38 (*t*, $J = 6.6$ Hz, 1H), 2.82 (*s*, 3H), 1.32–1.23 (*m*, 6H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 172.82 (C=O), 169.59 (C=O), 70.88 (CHOH), 61.11 (OCH₂), 53.06 (CH), 49.21 (CHCH₃), 27.13 (CH₃N), 15.28 (CH₃), 14.37 (CH₃); GCMS *m/z* (EI, +ve): found: 201.10 ($[M]^+$), calculated for C₉H₁₅NO₄: 201.10.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Table 2

Experimental details.

Crystal data	
Chemical formula	C ₉ H ₁₅ NO ₄
<i>M_r</i>	201.22
Crystal system, space group	Monoclinic, <i>C2/c</i>
Temperature (K)	173
<i>a</i> , <i>b</i> , <i>c</i> (Å)	12.1599 (15), 8.6065 (8), 20.217 (2)
β (°)	101.960 (3)
<i>V</i> (Å ³)	2069.9 (4)
<i>Z</i>	8
Radiation type	Mo <i>K</i> α
μ (mm ⁻¹)	0.10
Crystal size (mm)	0.2 × 0.2 × 0.1
Data collection	
Diffractometer	Rigaku XtaLAB P200
Absorption correction	Multi-scan (<i>REQAB</i> ; Rigaku, 1998)
<i>T_{min}</i> , <i>T_{max}</i>	0.879, 0.990
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	11140, 1874, 1769
<i>R_{int}</i>	0.019
(<i>sin</i> θ / λ) _{max} (Å ⁻¹)	0.603
Refinement	
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.033, 0.088, 1.04
No. of reflections	1874
No. of parameters	134
No. of restraints	1
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{\text{max}}$, $\Delta\rho_{\text{min}}$ (e Å ⁻³)	0.24, -0.18

Computer programs: *CrystalClear-SM Expert* (Rigaku, 2015), *SHELXT2018/2* (Sheldrick, 2015a), *SHELXL2018/3* (Sheldrick, 2015b) and *OLEX2* (Dolomanov *et al.*, 2009).

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full crystallographic data

IUCrData (2023). **8**, x230075 [https://doi.org/10.1107/S2414314623000755]

***rac*-Ethyl *rel*-(2*R*,3*R*,4*S*)-4-hydroxy-1,2-dimethyl-5-oxopyrrolidine-3-carboxylate**

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rac*-Ethyl *rel*-(2*R*,3*R*,4*S*)-4-hydroxy-1,2-dimethyl-5-oxopyrrolidine-3-carboxylateCrystal data*

$C_9H_{15}NO_4$

$M_r = 201.22$

Monoclinic, *C*2/*c*

$a = 12.1599$ (15) Å

$b = 8.6065$ (8) Å

$c = 20.217$ (2) Å

$\beta = 101.960$ (3)°

$V = 2069.9$ (4) Å³

$Z = 8$

$F(000) = 864$

$D_x = 1.291$ Mg m⁻³

Mo *K*α radiation, $\lambda = 0.71075$ Å

Cell parameters from 3484 reflections

$\theta = 2.1$ – 27.5 °

$\mu = 0.10$ mm⁻¹

$T = 173$ K

Prism, colorless

$0.2 \times 0.2 \times 0.1$ mm

Data collection

Rigaku XtaLAB P200

diffractometer

Detector resolution: 5.814 pixels mm⁻¹

ω scans

Absorption correction: multi-scan

(*REQAB*; Rigaku, 1998)

$T_{\min} = 0.879$, $T_{\max} = 0.990$

11140 measured reflections

1874 independent reflections

1769 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$

$\theta_{\max} = 25.4$ °, $\theta_{\min} = 2.1$ °

$h = -14$ → 14

$k = -10$ → 10

$l = -24$ → 24

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.033$

$wR(F^2) = 0.088$

$S = 1.04$

1874 reflections

134 parameters

1 restraint

Primary atom site location: dual

Secondary atom site location: difference Fourier map

Hydrogen site location: mixed

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0452P)^2 + 1.1669P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.24$ e Å⁻³

$\Delta\rho_{\min} = -0.18$ e Å⁻³

Special details

Refinement. The hydroxyl H atom (H2) was refined with free coordinates and isotropic displacement parameter.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.65691 (7)	0.16087 (10)	0.24315 (4)	0.0360 (2)
O2	0.53577 (7)	0.18679 (11)	0.35435 (4)	0.0388 (2)
O3	0.58426 (9)	0.59491 (13)	0.40623 (5)	0.0556 (3)
O4	0.68278 (7)	0.49308 (10)	0.50163 (4)	0.0348 (2)
N1	0.78865 (8)	0.29976 (11)	0.31723 (5)	0.0282 (2)
C2	0.68560 (9)	0.24451 (13)	0.29333 (5)	0.0275 (3)
C3	0.60731 (9)	0.30292 (13)	0.33799 (5)	0.0278 (3)
H3	0.561557	0.391982	0.315420	0.033*
C4	0.68834 (9)	0.35965 (13)	0.40091 (5)	0.0259 (3)
H4	0.706096	0.271938	0.433837	0.031*
C1	0.79582 (9)	0.40440 (13)	0.37550 (5)	0.0275 (3)
H1	0.789576	0.514521	0.359380	0.033*
C9	0.88156 (10)	0.28015 (15)	0.28261 (6)	0.0359 (3)
H9A	0.902826	0.381524	0.267070	0.043*
H9B	0.858357	0.211119	0.243670	0.043*
H9C	0.946000	0.234634	0.313807	0.043*
C5	0.64421 (9)	0.49526 (13)	0.43501 (6)	0.0294 (3)
C6	0.64892 (11)	0.62159 (15)	0.54030 (6)	0.0364 (3)
H6A	0.566573	0.619870	0.537083	0.044*
H6B	0.669524	0.722225	0.522480	0.044*
C7	0.70872 (15)	0.60126 (19)	0.61170 (7)	0.0567 (4)
H7A	0.790035	0.600863	0.614059	0.068*
H7B	0.686220	0.502495	0.629065	0.068*
H7C	0.689349	0.687011	0.639073	0.068*
C8	0.90310 (10)	0.38489 (16)	0.42851 (6)	0.0374 (3)
H8A	0.967321	0.419643	0.410005	0.045*
H8B	0.912839	0.275194	0.441383	0.045*
H8C	0.898462	0.447241	0.468402	0.045*
H2	0.4729 (10)	0.174 (2)	0.3162 (6)	0.063 (5)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0353 (5)	0.0407 (5)	0.0292 (4)	0.0041 (4)	0.0005 (3)	-0.0061 (4)
O2	0.0307 (4)	0.0512 (5)	0.0333 (5)	-0.0129 (4)	0.0038 (4)	0.0043 (4)
O3	0.0666 (7)	0.0579 (7)	0.0377 (5)	0.0363 (5)	0.0007 (5)	-0.0005 (4)
O4	0.0418 (5)	0.0345 (5)	0.0271 (4)	0.0084 (4)	0.0048 (3)	-0.0033 (3)
N1	0.0256 (5)	0.0321 (5)	0.0274 (5)	0.0024 (4)	0.0067 (4)	0.0004 (4)
C2	0.0280 (5)	0.0283 (6)	0.0247 (5)	0.0033 (4)	0.0022 (4)	0.0041 (4)
C3	0.0239 (5)	0.0324 (6)	0.0262 (6)	0.0001 (4)	0.0031 (4)	0.0030 (4)
C4	0.0238 (5)	0.0286 (6)	0.0247 (5)	0.0024 (4)	0.0034 (4)	0.0025 (4)
C1	0.0257 (5)	0.0273 (5)	0.0296 (6)	0.0003 (4)	0.0061 (4)	-0.0003 (4)
C9	0.0297 (6)	0.0449 (7)	0.0353 (6)	0.0061 (5)	0.0118 (5)	0.0014 (5)
C5	0.0248 (5)	0.0342 (6)	0.0291 (6)	0.0029 (5)	0.0054 (4)	0.0014 (5)
C6	0.0399 (7)	0.0341 (6)	0.0370 (7)	0.0036 (5)	0.0126 (5)	-0.0060 (5)

C7	0.0757 (10)	0.0528 (9)	0.0374 (7)	0.0168 (8)	0.0023 (7)	-0.0136 (6)
C8	0.0252 (6)	0.0478 (7)	0.0372 (6)	-0.0010 (5)	0.0021 (5)	-0.0065 (5)

Geometric parameters (Å, °)

O1—C2	1.2340 (14)	C1—H1	1.0000
O2—C3	1.4087 (14)	C1—C8	1.5162 (16)
O2—H2	0.973 (5)	C9—H9A	0.9800
O3—C5	1.1957 (14)	C9—H9B	0.9800
O4—C5	1.3313 (14)	C9—H9C	0.9800
O4—C6	1.4620 (14)	C6—H6A	0.9900
N1—C2	1.3337 (15)	C6—H6B	0.9900
N1—C1	1.4709 (14)	C6—C7	1.4860 (19)
N1—C9	1.4569 (14)	C7—H7A	0.9800
C2—C3	1.5266 (15)	C7—H7B	0.9800
C3—H3	1.0000	C7—H7C	0.9800
C3—C4	1.5192 (15)	C8—H8A	0.9800
C4—H4	1.0000	C8—H8B	0.9800
C4—C1	1.5488 (14)	C8—H8C	0.9800
C4—C5	1.5095 (15)		
C3—O2—H2	108.5 (10)	N1—C9—H9B	109.5
C5—O4—C6	116.81 (9)	N1—C9—H9C	109.5
C2—N1—C1	113.81 (9)	H9A—C9—H9B	109.5
C2—N1—C9	123.27 (10)	H9A—C9—H9C	109.5
C9—N1—C1	122.17 (9)	H9B—C9—H9C	109.5
O1—C2—N1	126.17 (10)	O3—C5—O4	123.63 (11)
O1—C2—C3	125.01 (10)	O3—C5—C4	124.81 (10)
N1—C2—C3	108.82 (9)	O4—C5—C4	111.53 (9)
O2—C3—C2	113.35 (10)	O4—C6—H6A	110.3
O2—C3—H3	109.8	O4—C6—H6B	110.3
O2—C3—C4	110.88 (9)	O4—C6—C7	107.16 (10)
C2—C3—H3	109.8	H6A—C6—H6B	108.5
C4—C3—C2	103.02 (8)	C7—C6—H6A	110.3
C4—C3—H3	109.8	C7—C6—H6B	110.3
C3—C4—H4	109.1	C6—C7—H7A	109.5
C3—C4—C1	104.33 (8)	C6—C7—H7B	109.5
C1—C4—H4	109.1	C6—C7—H7C	109.5
C5—C4—C3	113.62 (9)	H7A—C7—H7B	109.5
C5—C4—H4	109.1	H7A—C7—H7C	109.5
C5—C4—C1	111.32 (9)	H7B—C7—H7C	109.5
N1—C1—C4	101.50 (8)	C1—C8—H8A	109.5
N1—C1—H1	109.4	C1—C8—H8B	109.5
N1—C1—C8	113.44 (9)	C1—C8—H8C	109.5
C4—C1—H1	109.4	H8A—C8—H8B	109.5
C8—C1—C4	113.55 (9)	H8A—C8—H8C	109.5
C8—C1—H1	109.4	H8B—C8—H8C	109.5
N1—C9—H9A	109.5		

O1—C2—C3—O2	45.04 (15)	C1—N1—C2—O1	176.42 (11)
O1—C2—C3—C4	164.92 (11)	C1—N1—C2—C3	-3.06 (12)
O2—C3—C4—C1	148.36 (9)	C1—C4—C5—O3	84.29 (14)
O2—C3—C4—C5	-90.22 (11)	C1—C4—C5—O4	-94.04 (11)
N1—C2—C3—O2	-135.48 (10)	C9—N1—C2—O1	6.15 (18)
N1—C2—C3—C4	-15.60 (12)	C9—N1—C2—C3	-173.33 (10)
C2—N1—C1—C4	19.89 (12)	C9—N1—C1—C4	-169.72 (9)
C2—N1—C1—C8	142.07 (10)	C9—N1—C1—C8	-47.54 (14)
C2—C3—C4—C1	26.79 (11)	C5—O4—C6—C7	-175.57 (11)
C2—C3—C4—C5	148.21 (9)	C5—C4—C1—N1	-151.06 (9)
C3—C4—C1—N1	-28.13 (10)	C5—C4—C1—C8	86.83 (11)
C3—C4—C1—C8	-150.24 (10)	C6—O4—C5—O3	-0.29 (17)
C3—C4—C5—O3	-33.14 (16)	C6—O4—C5—C4	178.06 (9)
C3—C4—C5—O4	148.53 (9)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O2—H2 \cdots O1 ⁱ	0.97 (1)	1.78 (1)	2.7405 (12)	170 (2)
C1—H1 \cdots O1 ⁱⁱ	1.00	2.62	3.3953 (14)	134
C9—H9 <i>A</i> \cdots O1 ⁱⁱ	0.98	2.51	3.3355 (16)	142
C9—H9 <i>C</i> \cdots O3 ⁱⁱⁱ	0.98	2.54	3.5086 (15)	169
C7—H7 <i>C</i> \cdots O1 ^{iv}	0.98	2.58	3.5134 (17)	160

Symmetry codes: (i) $-x+1, y, -z+1/2$; (ii) $-x+3/2, y+1/2, -z+1/2$; (iii) $x+1/2, y-1/2, z$; (iv) $x, -y+1, z+1/2$.