

# Ethyl 10-cyano-7-hydroxy-6-oxo-3-phenyl-8,9,10,10a-tetrahydro-6H-benzo[c]chromene-10-carboxylate

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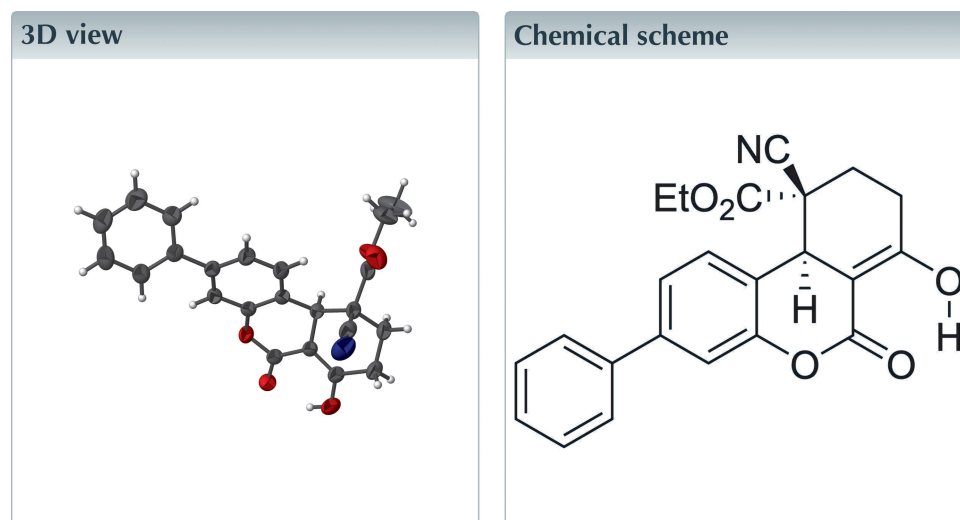
Edited by W. T. A. Harrison, University of Aberdeen, Scotland

Keywords: dibenzopyran; C—H... $\pi$  interaction; C—H...O interaction; crystal structure.

CCDC reference: 2153368

Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

In the title compound, C<sub>23</sub>H<sub>19</sub>NO<sub>5</sub>, the cyano group adopts an axial orientation and the ester group an equatorial orientation. The dihedral angle between the pendant phenyl group and the benzene ring of the fused-ring system is 25.97 (8)°. Intramolecular O—H...O and C—H...O hydrogen bonds are observed and the packing is consolidated by C—H...O and C—H... $\pi$  interactions.



## Structure description

Dibenzopyran-6-ones (also called 6H-benzo[c]chromen-6-ones or 3,4,5,6-dibenzo- $\alpha$ -pyranones) form an important group of biologically active natural products that occur in bacteria, fungi, lichens, higher plants and animal waste (Bialonska *et al.*, 2009). Elsamitricin, a dibenzopyran-6-one derived drug, is an efficient topoisomerase II inhibitor (Fiocchi *et al.*, 2011). As well as their biological activities, some dibenzopyran-6-ones have served as intermediates in the synthesis of more complex organic compounds (*see*, for example, Coghlan *et al.*, 2001). As a part of our ongoing studies in this area, we now describe the synthesis and crystal structure of the title compound.

The title compound has a dibenzopyran moiety decorated by several substituents, as shown in Fig. 1. There are two stereogenic centres: in the arbitrarily chosen asymmetric molecule, C15 and C20 have *S* and *R* configurations, respectively, but crystal symmetry generates a racemic mixture. The nitrile group attached to C20 occupies an axial position and is *anti* to the hydrogen atom attached to C19. The dihedral angle between the pendant C1–C6 phenyl group and the C7–C12 benzene ring of the fused-ring system is 25.97 (8)°. The Cremer–Pople puckering parameters of the O1/C9/C10/C13–C15 and

**Table 1**  
Hydrogen-bond geometry (Å, °).

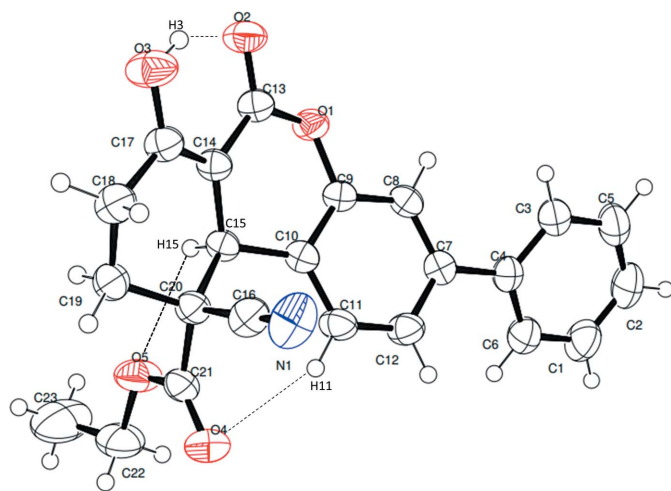
Cg2 and Cg3 are the centroids of the C1–C6 and C7–C12 rings, respectively.

D–H...A	D–H	H...A	D...A	D–H...A
O3–H3...O2	0.82	1.86	2.5702 (16)	145
C11–H11...O4	0.93	2.54	3.399 (2)	154
C18–H18B...O1 <sup>i</sup>	0.97	2.60	3.4289 (19)	144
C19–H19B...O2 <sup>i</sup>	0.97	2.60	3.285 (2)	128
C15–H15...Cg2 <sup>ii</sup>	0.98	2.95	3.7685 (17)	142
C23–H23B...Cg3 <sup>iii</sup>	0.96	2.82	3.686 (3)	151

Symmetry codes: (i)  $x - \frac{1}{2}, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (ii)  $-x + 1, -y, -z + 2$ ; (iii)  $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$ .

C14/C15/C17–C20 rings indicate half-chair conformations in each case with puckering amplitudes  $Q = 0.359$  Å;  $\theta = 104.52^\circ$ ;  $\varphi = 9.27^\circ$  and  $Q = 0.49$  Å;  $\theta = 134.17^\circ$ ;  $\varphi = 327.35^\circ$ , respectively. The O atom attached to C17 is stabilized in its enol (hydroxy) form, presumably as a result of forming a strong intramolecular hydrogen bond to O2. The packing is consolidated by weak C–H...O hydrogen bonds and C–H... $\pi$  interactions (Table 1) and an intramolecular C–H...O interaction is also observed (Fig. 2).

From a Cambridge Structural Database search (Groom *et al.*, 2016), we found compounds identified by refcodes OKEYUB (Xiao *et al.*, 2021), QABVEY (Wang *et al.*, 2021), ALTENU (McPhail *et al.*, 1973), AMUYIS (Alzaydi *et al.*, 2016), ANOPEG (Sosnovskikh *et al.*, 2016), ANOVIK (Sosnovskikh *et al.*, 2016), BUWJEK (Parveen *et al.*, 2015), BUXLOW (Fatunsin *et al.*, 2010), DIPTUR (Casiraghi *et al.*, 1986), DISJAS (Lee *et al.*, 2013), SEDFEN (Appel *et al.*, 2006), SIVQIZ (Poudel & Lee, 2014), SIJZER (Hussain *et al.*, 2007), TUPJOE (Siegel *et al.*, 2010), ZAQHIK (Dasari *et al.*, 2012) and IZACIY (Duan *et al.*, 2021) to be similar to the title compound.



**Figure 1**  
The molecular structure of the title compound with the atom-numbering scheme and displacement ellipsoids drawn at the 50% probability level. Intramolecular hydrogen bonds are shown as dashed lines.

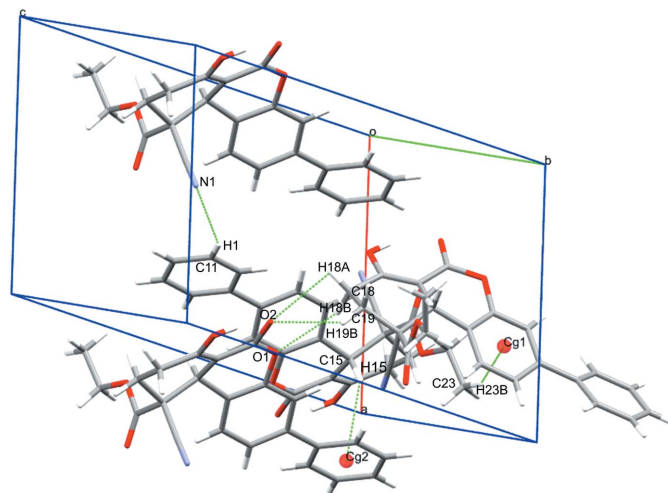
**Table 2**  
Experimental details.

Crystal data	C <sub>23</sub> H <sub>19</sub> NO <sub>5</sub>
Chemical formula	389.39
$M_r$	Monoclinic, $P2_1/n$
Crystal system, space group	293
Temperature (K)	9.7089 (8), 14.3510 (12), 14.2749 (15)
$a, b, c$ (Å)	106.946 (10)
$\beta$ (°)	1902.6 (3)
$V$ (Å <sup>3</sup> )	4
$Z$	Mo $K\alpha$
Radiation type	0.10
$\mu$ (mm <sup>-1</sup> )	0.75 × 0.44 × 0.42
Crystal size (mm)	
Data collection	Xcalibur, Eos
Diffractometer	Multi-scan ( <i>CrysAlis PRO</i> ; Agilent, 2014)
Absorption correction	0.932, 1.000
$T_{\min}, T_{\max}$	10689, 4413, 3119
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	0.026
$R_{\text{int}}$	0.686
( $\sin \theta/\lambda$ ) <sub>max</sub> (Å <sup>-1</sup> )	
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.046, 0.157, 0.95
No. of reflections	4413
No. of parameters	264
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å <sup>-3</sup> )	0.19, -0.20

Computer programs: *CrysAlis PRO* (Agilent, 2014), *SHELXT* (Sheldrick, 2015a), *SHELXL2018* (Sheldrick, 2015b), *ORTEP-3 for Windows* (Farrugia, 2012), *PLATON* (Spek, 2020) and *Mercury* (Macrae *et al.*, 2020).

## Synthesis and crystallization

A mixture of ethyl 10-cyano-7-hydroxy-6-oxo-3-[[trifluoromethyl)sulfonyl]oxy]-8,9,10,10a-tetrahydro-6H-benzo[*c*]chromene-10-carboxylate (100 mg, 0.22 mmol), phenylboronic acid (34 mg, 0.28 mmol, 1.3 equiv.), K<sub>3</sub>PO<sub>4</sub> (73 mg, 0.34 mmol, 1.6 equiv.) and Pd(PPh<sub>3</sub>)<sub>4</sub> (3 mg, 3 mol%) in degassed 1,4-dioxane (10 mL) was stirred at 100° C for 12 h under nitrogen. After completion of the coupling reaction (TLC), the mixture



**Figure 2**  
Intermolecular interactions in the title compound.

was cooled to room temperature, diluted with dichloromethane (DCM, 10 mL) and decanted. The residue was extracted with DCM (10 mL  $\times$  2) twice. The solvent was removed from the combined DCM layers and the residue was subjected to column chromatography on silica gel (100–200 mesh) by using increasing amounts of ethyl acetate in hexane (5% to 15%) as eluent to afford the title compound as a light-yellow solid in 90% yield (84 mg);  $R_f$  = 0.4 (hexanes:ethyl acetate, 7:3); m.p. 155–158° C. A sample suitable for single-crystal X-ray analysis was obtained by recrystallization the 50 mg of the solid from a mixture of 1 mL of distilled chloroform and 0.5 mL of distilled methanol.

### Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

### Acknowledgements

The authors thank the DST–FIST Single Crystal XRD facility at the Department of Chemistry, Pondicherry University, for the diffraction data and Dr Clara Gomes (FCT–UNL, Portugal) for the CSD database survey. MP thanks the Department of Chemistry for facilities. JM thanks Dr Amit Kumar Singh (Sharda University, India) for support.

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## full crystallographic data

*IUCrData* (2022). 7, x220199 [https://doi.org/10.1107/S2414314622001997]

## Ethyl 10-cyano-7-hydroxy-6-oxo-3-phenyl-8,9,10,10a-tetrahydro-6H-benzo[c]chromene-10-carboxylate

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Ethyl 10-cyano-7-hydroxy-6-oxo-3-phenyl-8,9,10,10a-tetrahydro-6H-benzo[c]chromene-10-carboxylate

### Crystal data

$C_{23}H_{19}NO_5$

$M_r = 389.39$

Monoclinic,  $P2_1/n$

$a = 9.7089$  (8) Å

$b = 14.3510$  (12) Å

$c = 14.2749$  (15) Å

$\beta = 106.946$  (10)°

$V = 1902.6$  (3) Å<sup>3</sup>

$Z = 4$

$F(000) = 816$

$D_x = 1.359$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 2883 reflections

$\theta = 3.0$ – $29.1$ °

$\mu = 0.10$  mm<sup>-1</sup>

$T = 293$  K

Block, colorless

$0.75 \times 0.44 \times 0.42$  mm

### Data collection

Xcalibur, Eos  
diffractometer

Radiation source: Enhance (Mo) X-ray Source

Detector resolution: 15.9821 pixels mm<sup>-1</sup>

$\omega$  scans

Absorption correction: multi-scan  
(CrysAlisPro; Agilent, 2014)

$T_{\min} = 0.932$ ,  $T_{\max} = 1.000$

10689 measured reflections

4413 independent reflections

3119 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.026$

$\theta_{\max} = 29.2$ °,  $\theta_{\min} = 3.0$ °

$h = -13 \rightarrow 13$

$k = -18 \rightarrow 17$

$l = -18 \rightarrow 18$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.046$

$wR(F^2) = 0.157$

$S = 0.95$

4413 reflections

264 parameters

0 restraints

Primary atom site location: iterative

Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.010$

$\Delta\rho_{\max} = 0.19$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.20$  e Å<sup>-3</sup>

### Special details

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

The hydrogen atoms in title compound were placed in calculated positions, with C—H = 0.93–0.97 Å and refined using a riding model with  $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$  or  $1.5 U_{\text{eq}}(\text{C-methyl})$ .

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.53859 (10)	0.16733 (8)	0.92065 (7)	0.0410 (3)
O2	0.63028 (12)	0.27465 (8)	0.84841 (8)	0.0509 (3)
C14	0.44382 (14)	0.18379 (10)	0.74486 (11)	0.0347 (3)
O5	0.27716 (13)	−0.06937 (8)	0.61784 (10)	0.0595 (4)
C15	0.36436 (14)	0.09212 (10)	0.73953 (10)	0.0332 (3)
H15	0.431514	0.042889	0.734017	0.040*
O3	0.50815 (13)	0.31695 (9)	0.66833 (9)	0.0588 (4)
H3	0.561932	0.325224	0.723781	0.088*
C9	0.42029 (14)	0.11197 (10)	0.92053 (11)	0.0337 (3)
C10	0.32907 (15)	0.07558 (10)	0.83485 (11)	0.0347 (3)
C7	0.29134 (15)	0.04148 (11)	1.02203 (11)	0.0366 (4)
C4	0.27309 (15)	0.02080 (11)	1.11980 (11)	0.0382 (4)
C19	0.28311 (17)	0.11903 (13)	0.55664 (11)	0.0447 (4)
H19A	0.360887	0.079214	0.550748	0.054*
H19B	0.203779	0.112652	0.497179	0.054*
C16	0.11816 (17)	0.14945 (12)	0.65498 (12)	0.0450 (4)
C13	0.54193 (15)	0.21194 (11)	0.83729 (11)	0.0375 (4)
C20	0.23367 (15)	0.08697 (11)	0.64512 (11)	0.0377 (4)
C8	0.40422 (15)	0.09584 (10)	1.01183 (11)	0.0366 (4)
H8	0.469369	0.121527	1.066950	0.044*
C21	0.17320 (17)	−0.01170 (12)	0.62229 (11)	0.0437 (4)
O4	0.05006 (13)	−0.03278 (10)	0.60776 (10)	0.0649 (4)
C12	0.19570 (17)	0.00624 (12)	0.93635 (12)	0.0443 (4)
H12	0.117331	−0.029036	0.940506	0.053*
C18	0.33369 (17)	0.21905 (12)	0.56740 (12)	0.0467 (4)
H18A	0.381800	0.232559	0.518170	0.056*
H18B	0.250752	0.259856	0.555766	0.056*
C17	0.43402 (16)	0.23902 (11)	0.66616 (12)	0.0410 (4)
C11	0.21487 (17)	0.02259 (12)	0.84563 (12)	0.0443 (4)
H11	0.149569	−0.002549	0.790253	0.053*
C6	0.20395 (18)	−0.05906 (13)	1.13558 (13)	0.0507 (4)
H6	0.166517	−0.099657	1.083600	0.061*
C3	0.32765 (18)	0.08032 (12)	1.19927 (12)	0.0466 (4)
H3A	0.375026	0.134713	1.191218	0.056*
C2	0.2424 (2)	−0.02057 (16)	1.30377 (15)	0.0627 (5)
H2	0.231601	−0.034154	1.364935	0.075*
C1	0.1887 (2)	−0.08045 (15)	1.22642 (15)	0.0637 (5)

H1	0.142442	-0.135050	1.235243	0.076*
C5	0.3116 (2)	0.05875 (15)	1.29030 (14)	0.0579 (5)
H5	0.348412	0.098916	1.342756	0.070*
C22	0.2376 (2)	-0.16616 (14)	0.59282 (18)	0.0714 (6)
H22A	0.150569	-0.169084	0.538150	0.086*
H22B	0.220030	-0.197877	0.648226	0.086*
C23	0.3565 (3)	-0.21069 (18)	0.5666 (2)	0.1099 (10)
H23A	0.372605	-0.179133	0.511451	0.165*
H23B	0.333040	-0.274743	0.550071	0.165*
H23C	0.442123	-0.207456	0.621170	0.165*
N1	0.03398 (17)	0.20167 (13)	0.66178 (13)	0.0658 (5)

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0360 (5)	0.0501 (7)	0.0321 (6)	-0.0093 (5)	0.0023 (4)	0.0025 (5)
O2	0.0501 (7)	0.0508 (7)	0.0448 (7)	-0.0184 (6)	0.0029 (5)	0.0017 (5)
C14	0.0322 (7)	0.0351 (8)	0.0334 (8)	-0.0002 (6)	0.0042 (6)	0.0018 (6)
O5	0.0520 (7)	0.0435 (7)	0.0799 (10)	-0.0124 (6)	0.0143 (6)	-0.0140 (6)
C15	0.0313 (7)	0.0349 (8)	0.0296 (8)	0.0014 (6)	0.0027 (6)	0.0016 (6)
O3	0.0606 (8)	0.0537 (8)	0.0516 (8)	-0.0181 (6)	-0.0003 (6)	0.0162 (6)
C9	0.0306 (7)	0.0323 (8)	0.0347 (8)	0.0019 (6)	0.0041 (6)	0.0024 (6)
C10	0.0364 (7)	0.0323 (8)	0.0318 (8)	0.0017 (6)	0.0040 (6)	0.0029 (6)
C7	0.0385 (7)	0.0357 (8)	0.0338 (9)	0.0056 (6)	0.0078 (6)	0.0026 (6)
C4	0.0365 (7)	0.0410 (9)	0.0375 (9)	0.0092 (6)	0.0115 (6)	0.0034 (7)
C19	0.0431 (8)	0.0569 (11)	0.0311 (9)	-0.0059 (8)	0.0060 (6)	0.0023 (7)
C16	0.0375 (8)	0.0543 (10)	0.0396 (10)	0.0015 (8)	0.0056 (6)	0.0104 (8)
C13	0.0334 (7)	0.0373 (8)	0.0383 (9)	-0.0003 (6)	0.0051 (6)	0.0039 (7)
C20	0.0341 (7)	0.0448 (9)	0.0307 (8)	-0.0029 (6)	0.0039 (6)	0.0012 (7)
C8	0.0379 (7)	0.0349 (8)	0.0324 (8)	0.0026 (6)	0.0028 (6)	-0.0005 (6)
C21	0.0431 (9)	0.0534 (10)	0.0307 (9)	-0.0102 (8)	0.0049 (6)	-0.0038 (7)
O4	0.0483 (7)	0.0789 (10)	0.0652 (9)	-0.0248 (7)	0.0132 (6)	-0.0154 (7)
C12	0.0437 (8)	0.0481 (10)	0.0389 (9)	-0.0103 (7)	0.0086 (7)	0.0044 (7)
C18	0.0430 (9)	0.0564 (11)	0.0361 (9)	-0.0032 (7)	0.0045 (7)	0.0121 (8)
C17	0.0377 (8)	0.0423 (9)	0.0399 (9)	-0.0027 (7)	0.0064 (6)	0.0052 (7)
C11	0.0444 (8)	0.0481 (10)	0.0344 (9)	-0.0110 (7)	0.0022 (7)	0.0010 (7)
C6	0.0548 (10)	0.0529 (11)	0.0485 (11)	-0.0025 (8)	0.0214 (8)	0.0004 (8)
C3	0.0476 (9)	0.0504 (10)	0.0426 (10)	0.0034 (8)	0.0144 (7)	-0.0021 (8)
C2	0.0683 (12)	0.0819 (15)	0.0468 (12)	0.0071 (11)	0.0307 (10)	0.0081 (10)
C1	0.0711 (12)	0.0699 (14)	0.0602 (13)	-0.0059 (11)	0.0348 (10)	0.0087 (10)
C5	0.0591 (11)	0.0746 (13)	0.0424 (11)	0.0072 (10)	0.0185 (8)	-0.0082 (9)
C22	0.0747 (13)	0.0481 (12)	0.0924 (17)	-0.0228 (10)	0.0258 (11)	-0.0196 (11)
C23	0.100 (2)	0.0564 (15)	0.188 (3)	-0.0203 (13)	0.065 (2)	-0.0405 (17)
N1	0.0558 (9)	0.0742 (11)	0.0700 (12)	0.0198 (9)	0.0222 (8)	0.0222 (9)

*Geometric parameters (Å, °)*

O1—C13	1.3599 (18)	C7—C4	1.487 (2)
O1—C9	1.3961 (17)	C4—C6	1.379 (2)
O2—C13	1.2212 (18)	C4—C3	1.395 (2)
C14—C17	1.355 (2)	C19—C18	1.510 (2)
C14—C13	1.442 (2)	C19—C20	1.546 (2)
C14—C15	1.516 (2)	C16—N1	1.134 (2)
O5—C21	1.321 (2)	C16—C20	1.474 (2)
O5—C22	1.457 (2)	C20—C21	1.531 (2)
C15—C10	1.516 (2)	C21—O4	1.1913 (18)
C15—C20	1.5604 (19)	C12—C11	1.381 (2)
O3—C17	1.3255 (19)	C18—C17	1.489 (2)
C9—C8	1.377 (2)	C6—C1	1.382 (2)
C9—C10	1.386 (2)	C3—C5	1.388 (2)
C10—C11	1.390 (2)	C2—C5	1.363 (3)
C7—C8	1.387 (2)	C2—C1	1.376 (3)
C7—C12	1.397 (2)	C22—C23	1.460 (3)
C13—O1—C9	119.76 (11)	O1—C13—C14	119.40 (13)
C17—C14—C13	117.57 (14)	C16—C20—C21	109.13 (13)
C17—C14—C15	123.65 (13)	C16—C20—C19	108.83 (12)
C13—C14—C15	118.68 (13)	C21—C20—C19	107.00 (13)
C21—O5—C22	117.30 (14)	C16—C20—C15	109.78 (13)
C14—C15—C10	109.62 (12)	C21—C20—C15	113.11 (12)
C14—C15—C20	111.00 (12)	C19—C20—C15	108.88 (12)
C10—C15—C20	115.36 (12)	C9—C8—C7	120.33 (13)
C8—C9—C10	123.53 (14)	O4—C21—O5	125.08 (16)
C8—C9—O1	114.50 (12)	O4—C21—C20	125.09 (16)
C10—C9—O1	121.96 (13)	O5—C21—C20	109.79 (13)
C9—C10—C11	115.70 (14)	C11—C12—C7	121.50 (15)
C9—C10—C15	118.66 (13)	C17—C18—C19	112.50 (13)
C11—C10—C15	125.57 (13)	O3—C17—C14	124.59 (14)
C8—C7—C12	117.11 (14)	O3—C17—C18	112.66 (13)
C8—C7—C4	121.65 (13)	C14—C17—C18	122.73 (14)
C12—C7—C4	121.25 (14)	C12—C11—C10	121.79 (14)
C6—C4—C3	117.61 (15)	C4—C6—C1	121.87 (18)
C6—C4—C7	121.05 (15)	C5—C3—C4	120.32 (17)
C3—C4—C7	121.33 (15)	C5—C2—C1	119.68 (18)
C18—C19—C20	111.65 (14)	C2—C1—C6	119.68 (19)
N1—C16—C20	176.08 (18)	C2—C5—C3	120.84 (18)
O2—C13—O1	115.36 (13)	O5—C22—C23	107.97 (17)
O2—C13—C14	125.24 (14)		

*Hydrogen-bond geometry (Å, °)*

*Cg2* and *Cg3* are the centroids of the C1–C6 and C7–C12 rings, respectively.

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O3—H3···O2	0.82	1.86	2.5702 (16)	145
C11—H11···O4	0.93	2.54	3.399 (2)	154
C18—H18 <i>B</i> ···O1 <sup>i</sup>	0.97	2.60	3.4289 (19)	144
C19—H19 <i>B</i> ···O2 <sup>i</sup>	0.97	2.60	3.285 (2)	128
C15—H15··· <i>Cg2</i> <sup>ii</sup>	0.98	2.95	3.7685 (17)	142
C23—H23 <i>B</i> ··· <i>Cg3</i> <sup>iii</sup>	0.96	2.82	3.686 (3)	151

Symmetry codes: (i)  $x-1/2, -y+1/2, z-1/2$ ; (ii)  $-x+1, -y, -z+2$ ; (iii)  $-x+1/2, y-1/2, -z+3/2$ .