

4-(Diphenylmethoxy)-3-ethoxybenzaldehyde

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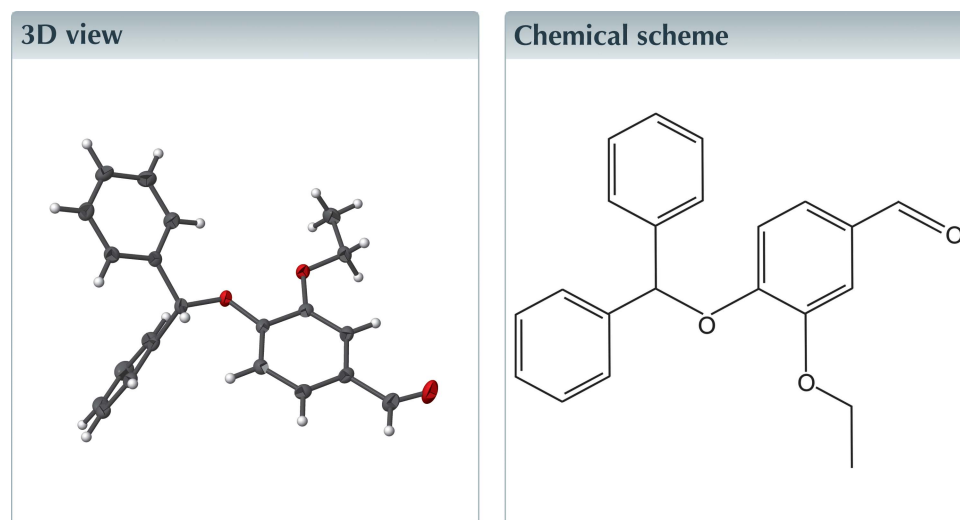
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Structural data: full structural data are available from iucrdata.iucr.org

In the title compound, $C_{22}H_{20}O_3$, the dihedral angle between the aromatic rings linked by the methine group is $81.265(4)^\circ$ and the ethoxy side chain adopts an extended conformation [$C-O-C = 177.24(10)^\circ$]. In the crystal, weak $C-H \cdots \pi$ and $C-H \cdots O$ interactions link the molecules into sheets.



Structure description

The preparation of Schiff bases (Omidi & Kakanejadifard, 2020) is nowadays an interesting topic, because of their various application and properties (e.g., Kizilkaya *et al.*, 2020). As part of our studies in this area, the title aldehyde, $C_{22}H_{20}O_3$, was prepared as a precursor to new Schiff bases and we now describe its crystal structure (Fig. 1).

As expected, the $C2-O2$, $C20-O2$, $C1-O1$ and $C7-O1$ bond lengths reveal single bond character while $C22=O3$ is a double bond. The dihedral angle between the $C1-C6$ and $C8-C13$ aromatic rings connected by the $C7$ methine group is $81.265(4)^\circ$. In the crystal, weak $C-H \cdots O$ (Table 1) and $C-H \cdots \pi$ interactions [$C20^{iii}-H1c20^{iii} \cdots Cg3 = 3.05 \text{ \AA}$ and $C5^{iv}-H1c5^{iv} \cdots Cg3 = 2.89 \text{ \AA}$; symmetry codes: (iii) $\frac{1}{2} - x, \frac{1}{2} + y, z$; (iv) $x - \frac{1}{2}, y, \frac{1}{2} - z$; $Cg3$ is the centroid of the $C14-C19$ ring] are observed. These link the molecules into sheets lying perpendicular to the c -axis direction (Fig. 2).

Synthesis and crystallization

3-Ethoxy-4-hydroxy benzaldehyde (0.20 mmol) and potassium carbonate (0.40 mmol) were mixed in dimethylformamide (25 ml) and stirred for 0.5 h. A solution of diphenylbromomethane (0.2 mmol) in ethanol (20 ml) was added dropwise and the mixture was stirred at 80°C for 24 h. After that, the solution was concentrated under reduced pressure. The cream precipitate of the title compound formed by adding cold water (250 ml) was filtered off and washed several times with cold ethanol. Colourless slabs were recrystallized from the mixed solvents of chloroform and ethanol (1:1).

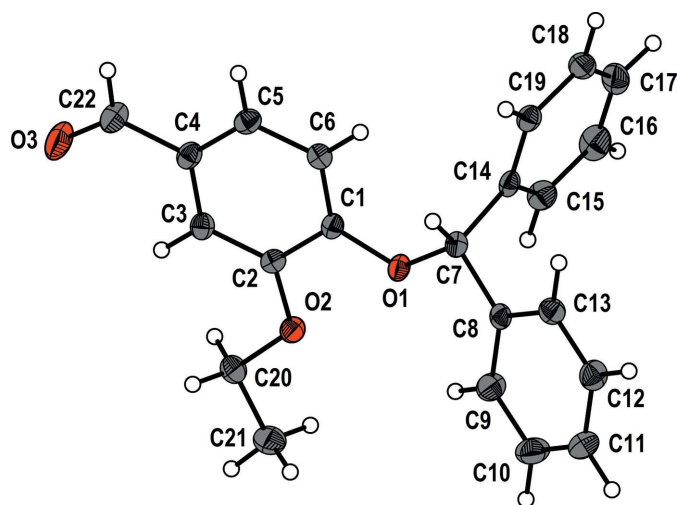


Figure 1
The molecular structure of the title compound with displacement ellipsoids drawn at the 50% probability level.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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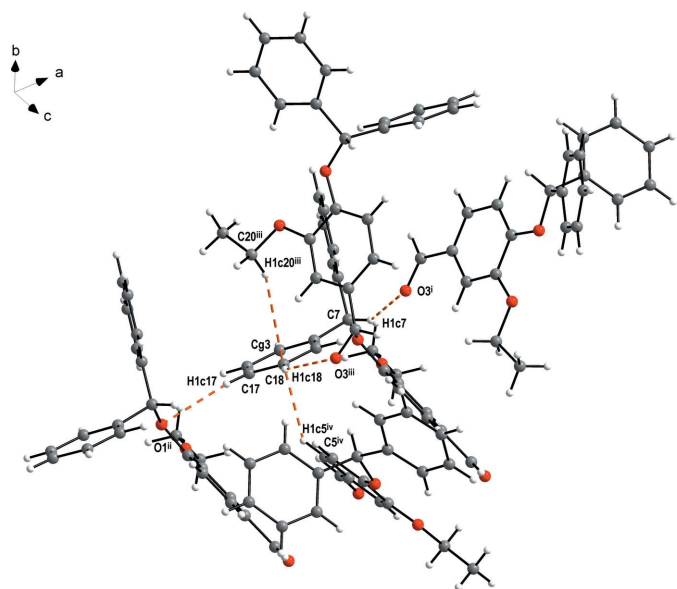


Figure 2
Partial packing diagram showing the hydrogen bonds in the title compound. Hydrogen atoms not involved in hydrogen bonding were omitted for clarity. Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + 1$; (ii) $x - 1, y, z$; (iii) $-x + \frac{1}{2}, y + \frac{1}{2}, z$.

Table 1
Hydrogen-bond geometry ($\text{\AA}, ^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C7-H1c7\cdots O3^i$	0.96	2.52	3.3794 (15)	148
$C17-H1c17\cdots O1^{ii}$	0.96	2.57	3.4440 (15)	152
$C18-H1c18\cdots O3^{iii}$	0.96	2.53	3.2906 (16)	136

Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x - 1, y, z$; (iii) $-x + \frac{1}{2}, y + \frac{1}{2}, z$.

Table 2
Experimental details.

Crystal data	
Chemical formula	$C_{22}H_{20}O_3$
M_r	332.4
Crystal system, space group	Orthorhombic, $Pbca$
Temperature (K)	120
a, b, c (\AA)	8.1123 (4), 15.8713 (9), 27.6155 (14)
V (\AA^3)	3555.6 (3)
Z	8
Radiation type	Mo $K\alpha$
μ (mm^{-1})	0.08
Crystal size (mm)	$0.83 \times 0.32 \times 0.26$
Data collection	
Diffractometer	Rigaku Oxford Diffraction Xcalibur, AtlasS2, Gemini ultra Analytical (<i>CrysAlis PRO</i> ; Rigaku OD, 2015)
Absorption correction	
T_{\min}, T_{\max}	0.958, 0.982
No. of measured, independent and observed [$I > 3\sigma(I)$] reflections	16215, 4454, 3284
R_{int}	0.028
$(\sin \theta/\lambda)_{\text{max}}$ (\AA^{-1})	0.695
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.040, 0.107, 1.44
No. of reflections	4454
No. of parameters	227
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ ($e \text{\AA}^{-3}$)	0.22, -0.20

Computer programs: *CrysAlis PRO* (Rigaku OD, 2015), *SUPERFLIP* (Palatinus & Chapuis, (2007)), *JANA2006* (Petříček *et al.*, 2014) and *DIAMOND* (Brandenburg, 1999).

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full crystallographic data

IUCrData (2021). 6, x210356 [https://doi.org/10.1107/S2414314621003564]

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Crystal data

$C_{22}H_{20}O_3$	$F(000) = 1408$
$M_r = 332.4$	$D_x = 1.242 \text{ Mg m}^{-3}$
Orthorhombic, <i>Pbca</i>	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: -P -2xab; -2ybc; -2zac	Cell parameters from 4539 reflections
$a = 8.1123 (4) \text{ \AA}$	$\theta = 3.9\text{--}29.2^\circ$
$b = 15.8713 (9) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$c = 27.6155 (14) \text{ \AA}$	$T = 120 \text{ K}$
$V = 3555.6 (3) \text{ \AA}^3$	Plate, colourless
$Z = 8$	$0.83 \times 0.32 \times 0.26 \text{ mm}$

Data collection

Rigaku Oxford Diffraction Xcalibur, AtlasS2, Gemini ultra diffractometer	$T_{\min} = 0.958, T_{\max} = 0.982$
Radiation source: X-ray tube	16215 measured reflections
Graphite monochromator	4454 independent reflections
Detector resolution: $5.1783 \text{ pixels mm}^{-1}$	3284 reflections with $I > 3\sigma(I)$
ω scans	$R_{\text{int}} = 0.028$
Absorption correction: analytical (CrysAlisPro; Rigaku OD, 2015)	$\theta_{\max} = 29.6^\circ, \theta_{\min} = 3.6^\circ$
	$h = -11 \rightarrow 8$
	$k = -20 \rightarrow 20$
	$l = -37 \rightarrow 37$

Refinement

Refinement on F^2	Weighting scheme based on measured s.u.'s $w =$
$R[F^2 > 2\sigma(F^2)] = 0.040$	$1/(\sigma^2(I) + 0.0016I^2)$
$wR(F^2) = 0.107$	$(\Delta/\sigma)_{\max} = 0.0004$
$S = 1.44$	$\Delta\rho_{\max} = 0.22 \text{ e \AA}^{-3}$
4454 reflections	$\Delta\rho_{\min} = -0.20 \text{ e \AA}^{-3}$
227 parameters	Extinction correction: B-C type 1 Gaussian
0 restraints	isotropic [Becker, P. J. & Coppens, P. (1974).
80 constraints	<i>Acta Cryst. A</i> 30 , 129–147]
H-atom parameters constrained	Extinction coefficient: 13000 (2000)

Special details

Refinement. All hydrogen atoms were discernible in difference Fourier maps and could be refined to reasonable geometry, but according to common practice they were refined as riding atoms with C—H = 0.96 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.34686 (10)	0.62710 (5)	0.13059 (3)	0.0193 (2)
O2	0.52510 (10)	0.49847 (5)	0.11100 (3)	0.0203 (2)
O3	0.64458 (13)	0.36127 (6)	0.27937 (3)	0.0385 (3)
C1	0.38844 (13)	0.57930 (7)	0.16975 (4)	0.0170 (3)
C2	0.48965 (13)	0.50922 (7)	0.15898 (4)	0.0170 (3)
C3	0.54565 (14)	0.45894 (7)	0.19596 (4)	0.0195 (3)
C4	0.50097 (14)	0.47681 (7)	0.24410 (4)	0.0205 (3)
C5	0.40159 (14)	0.54531 (8)	0.25426 (4)	0.0206 (3)
C6	0.34409 (14)	0.59647 (8)	0.21712 (4)	0.0198 (3)
C7	0.27297 (14)	0.70813 (7)	0.13914 (4)	0.0171 (3)
C8	0.32021 (13)	0.76454 (7)	0.09687 (4)	0.0178 (3)
C9	0.39897 (15)	0.73419 (8)	0.05577 (4)	0.0261 (4)
C10	0.43962 (18)	0.78854 (9)	0.01835 (5)	0.0326 (4)
C11	0.40310 (16)	0.87355 (9)	0.02180 (5)	0.0297 (4)
C12	0.32477 (15)	0.90430 (9)	0.06254 (5)	0.0261 (4)
C13	0.28374 (14)	0.85001 (8)	0.10003 (4)	0.0218 (4)
C14	0.08714 (14)	0.70028 (7)	0.14420 (4)	0.0177 (3)
C15	-0.00177 (15)	0.65479 (8)	0.11019 (5)	0.0256 (4)
C16	-0.17290 (16)	0.65229 (9)	0.11241 (5)	0.0322 (4)
C17	-0.25477 (16)	0.69522 (8)	0.14866 (5)	0.0298 (4)
C18	-0.16734 (15)	0.73910 (8)	0.18332 (5)	0.0250 (4)
C19	0.00414 (14)	0.74156 (7)	0.18104 (4)	0.0199 (3)
C20	0.63310 (15)	0.43012 (8)	0.09834 (4)	0.0245 (4)
C21	0.64587 (17)	0.42893 (9)	0.04403 (5)	0.0316 (4)
C22	0.55991 (16)	0.42412 (8)	0.28389 (5)	0.0275 (4)
H1c3	0.615307	0.411587	0.188954	0.0233*
H1c5	0.37214	0.557585	0.287186	0.0248*
H1c6	0.273891	0.643544	0.22428	0.0237*
H1c7	0.312056	0.732024	0.168939	0.0205*
H1c9	0.425323	0.675383	0.053266	0.0313*
H1c10	0.493341	0.767159	-0.010079	0.0391*
H1c11	0.432179	0.911071	-0.004095	0.0357*
H1c12	0.298754	0.963157	0.06492	0.0313*
H1c13	0.22977	0.871597	0.128373	0.0261*
H1c15	0.055107	0.624903	0.085003	0.0307*
H1c16	-0.23405	0.620772	0.088795	0.0386*
H1c17	-0.373047	0.694527	0.149741	0.0357*
H1c18	-0.224467	0.767824	0.208906	0.0301*
H1c19	0.065108	0.772039	0.205141	0.0239*
H1c20	0.586452	0.377955	0.10939	0.0294*
H2c20	0.740136	0.439896	0.112063	0.0294*
H1c21	0.719041	0.384571	0.034222	0.0379*
H2c21	0.538733	0.419448	0.030313	0.0379*
H3c21	0.687979	0.482033	0.032892	0.0379*
H1c22	0.529438	0.440345	0.316162	0.033*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0248 (4)	0.0172 (4)	0.0158 (4)	0.0056 (3)	-0.0024 (3)	0.0000 (3)
O2	0.0249 (4)	0.0180 (4)	0.0180 (4)	0.0037 (3)	0.0006 (3)	-0.0005 (3)
O3	0.0567 (6)	0.0310 (6)	0.0277 (5)	0.0151 (5)	-0.0056 (4)	0.0064 (4)
C1	0.0164 (5)	0.0162 (6)	0.0185 (6)	-0.0025 (5)	-0.0032 (4)	0.0020 (4)
C2	0.0181 (5)	0.0164 (6)	0.0165 (5)	-0.0037 (5)	-0.0010 (4)	-0.0009 (4)
C3	0.0207 (6)	0.0160 (6)	0.0217 (6)	-0.0001 (5)	-0.0018 (5)	0.0006 (4)
C4	0.0219 (6)	0.0190 (6)	0.0205 (6)	-0.0036 (5)	-0.0029 (5)	0.0032 (5)
C5	0.0220 (6)	0.0219 (6)	0.0180 (6)	-0.0037 (5)	-0.0007 (5)	0.0006 (5)
C6	0.0188 (5)	0.0197 (6)	0.0208 (6)	0.0006 (5)	0.0000 (4)	-0.0005 (5)
C7	0.0197 (6)	0.0151 (6)	0.0164 (5)	0.0019 (5)	-0.0028 (4)	-0.0022 (4)
C8	0.0143 (5)	0.0208 (6)	0.0182 (5)	-0.0004 (5)	-0.0042 (4)	0.0004 (4)
C9	0.0328 (7)	0.0224 (7)	0.0230 (6)	0.0036 (6)	0.0031 (5)	0.0003 (5)
C10	0.0405 (8)	0.0327 (8)	0.0245 (7)	0.0045 (6)	0.0101 (6)	0.0017 (6)
C11	0.0326 (7)	0.0299 (7)	0.0267 (7)	-0.0012 (6)	0.0039 (5)	0.0093 (6)
C12	0.0280 (6)	0.0200 (7)	0.0304 (7)	0.0006 (5)	-0.0002 (5)	0.0033 (5)
C13	0.0218 (6)	0.0211 (6)	0.0225 (6)	0.0012 (5)	0.0008 (5)	-0.0010 (5)
C14	0.0190 (5)	0.0145 (6)	0.0195 (6)	-0.0008 (5)	-0.0033 (4)	0.0050 (4)
C15	0.0268 (7)	0.0270 (7)	0.0230 (7)	-0.0037 (5)	-0.0032 (5)	-0.0009 (5)
C16	0.0279 (7)	0.0352 (8)	0.0335 (8)	-0.0112 (6)	-0.0109 (6)	0.0019 (6)
C17	0.0178 (6)	0.0313 (7)	0.0402 (7)	-0.0048 (6)	-0.0016 (5)	0.0120 (6)
C18	0.0234 (6)	0.0210 (6)	0.0307 (7)	0.0010 (5)	0.0053 (5)	0.0076 (5)
C19	0.0222 (6)	0.0167 (6)	0.0208 (6)	-0.0010 (5)	-0.0011 (5)	0.0029 (5)
C20	0.0256 (6)	0.0223 (6)	0.0257 (6)	0.0064 (5)	0.0022 (5)	-0.0009 (5)
C21	0.0347 (7)	0.0333 (8)	0.0267 (7)	0.0066 (6)	0.0065 (6)	-0.0034 (6)
C22	0.0347 (7)	0.0257 (7)	0.0221 (7)	0.0013 (6)	-0.0025 (5)	0.0035 (5)

Geometric parameters (Å, °)

O1—C1	1.3632 (13)	C11—C12	1.3813 (18)
O1—C7	1.4380 (13)	C11—H1c11	0.96
O2—C2	1.3662 (13)	C12—C13	1.3873 (18)
O2—C20	1.4377 (14)	C12—H1c12	0.96
O3—C22	1.2174 (16)	C13—H1c13	0.96
C1—C2	1.4142 (15)	C14—C15	1.3870 (17)
C1—C6	1.3838 (16)	C14—C19	1.3845 (16)
C2—C3	1.3736 (16)	C15—C16	1.3902 (18)
C3—C4	1.4067 (16)	C15—H1c15	0.96
C3—H1c3	0.96	C16—C17	1.3809 (19)
C4—C5	1.3821 (16)	C16—H1c16	0.96
C4—C22	1.4614 (17)	C17—C18	1.3807 (18)
C5—C6	1.3892 (16)	C17—H1c17	0.96
C5—H1c5	0.96	C18—C19	1.3931 (17)
C6—H1c6	0.96	C18—H1c18	0.96
C7—C8	1.5202 (16)	C19—H1c19	0.96
C7—C14	1.5192 (16)	C20—C21	1.5034 (17)

C7—H1c7	0.96	C20—H1c20	0.96
C8—C9	1.3885 (17)	C20—H2c20	0.96
C8—C13	1.3911 (17)	C21—H1c21	0.96
C9—C10	1.3865 (18)	C21—H2c21	0.96
C9—H1c9	0.96	C21—H3c21	0.96
C10—C11	1.3843 (19)	C22—H1c22	0.96
C10—H1c10	0.96		
C1—O1—C7	118.07 (8)	C11—C12—H1c12	120.04
C2—O2—C20	117.29 (9)	C13—C12—H1c12	120.04
O1—C1—C2	114.48 (9)	C8—C13—C12	120.52 (11)
O1—C1—C6	125.17 (10)	C8—C13—H1c13	119.74
C2—C1—C6	120.32 (10)	C12—C13—H1c13	119.74
O2—C2—C1	115.12 (9)	C7—C14—C15	119.76 (10)
O2—C2—C3	125.37 (10)	C7—C14—C19	120.76 (10)
C1—C2—C3	119.51 (10)	C15—C14—C19	119.40 (10)
C2—C3—C4	120.02 (10)	C14—C15—C16	120.29 (12)
C2—C3—H1c3	119.99	C14—C15—H1c15	119.86
C4—C3—H1c3	119.99	C16—C15—H1c15	119.86
C3—C4—C5	120.07 (11)	C15—C16—C17	119.88 (12)
C3—C4—C22	120.72 (11)	C15—C16—H1c16	120.06
C5—C4—C22	119.22 (11)	C17—C16—H1c16	120.06
C4—C5—C6	120.37 (11)	C16—C17—C18	120.30 (12)
C4—C5—H1c5	119.81	C16—C17—H1c17	119.85
C6—C5—H1c5	119.81	C18—C17—H1c17	119.85
C1—C6—C5	119.71 (11)	C17—C18—C19	119.70 (11)
C1—C6—H1c6	120.14	C17—C18—H1c18	120.15
C5—C6—H1c6	120.14	C19—C18—H1c18	120.15
O1—C7—C8	107.20 (8)	C14—C19—C18	120.39 (11)
O1—C7—C14	110.82 (9)	C14—C19—H1c19	119.8
O1—C7—H1c7	110.86	C18—C19—H1c19	119.8
C8—C7—C14	111.65 (9)	O2—C20—C21	107.10 (10)
C8—C7—H1c7	110.01	O2—C20—H1c20	109.47
C14—C7—H1c7	106.34	O2—C20—H2c20	109.47
C7—C8—C9	122.61 (10)	C21—C20—H1c20	109.47
C7—C8—C13	118.21 (10)	C21—C20—H2c20	109.47
C9—C8—C13	119.18 (11)	H1c20—C20—H2c20	111.74
C8—C9—C10	120.17 (12)	C20—C21—H1c21	109.47
C8—C9—H1c9	119.91	C20—C21—H2c21	109.47
C10—C9—H1c9	119.91	C20—C21—H3c21	109.47
C9—C10—C11	120.28 (12)	H1c21—C21—H2c21	109.47
C9—C10—H1c10	119.86	H1c21—C21—H3c21	109.47
C11—C10—H1c10	119.86	H2c21—C21—H3c21	109.47
C10—C11—C12	119.93 (12)	O3—C22—C4	125.21 (12)
C10—C11—H1c11	120.04	O3—C22—H1c22	117.39
C12—C11—H1c11	120.04	C4—C22—H1c22	117.4
C11—C12—C13	119.91 (12)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C7—H1c7 \cdots O3 ⁱ	0.96	2.52	3.3794 (15)	148
C17—H1c17 \cdots O1 ⁱⁱ	0.96	2.57	3.4440 (15)	152
C18—H1c18 \cdots O3 ⁱⁱⁱ	0.96	2.53	3.2906 (16)	136

Symmetry codes: (i) $-x+1, y+1/2, -z+1/2$; (ii) $x-1, y, z$; (iii) $-x+1/2, y+1/2, z$.