

(2*S*,3'*S*,3*a*'*R*,5'*R*,7*a*'*R*)-5'-[(*E*)-5-(Furan-3-yl)-2-methylpent-1-en-1-yl]-3-hydroxy-3',4,7'-trimethyl-1',2',3',3*a*',5',7*a*'-hexahydro-5*H*-spiro[furan-2,4'-inden]-5-one

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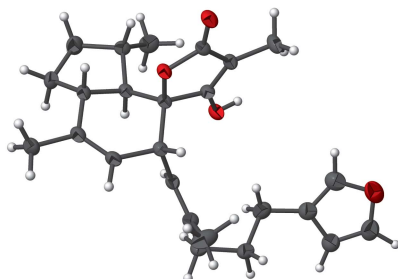
CCDC reference: 2047802

Structural data: full structural data are available from iucrdata.iucr.org

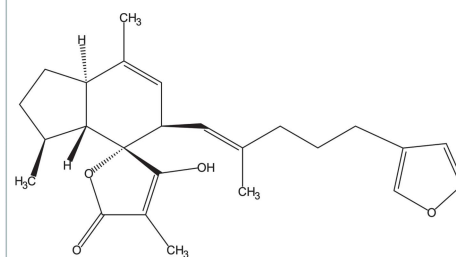
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The title compound, ircinianin, C₂₅H₃₂O₄, belongs to the sesterterpene tetronic acid compound family and was isolated from the marine sponge *Ircinia wistarii*. These chemical scaffolds are pharmacologically relevant, since they represent a new class of glycine receptor modulators. The furan ring makes a dihedral angle of 35.14 (12)° to the 4-hydroxy-3-methylfuran-2(5*H*)-one ring. The crystal packing is characterized by intermolecular O—H...O hydrogen bonds, which generate [010] chains.

3D view



Chemical scheme



Structure description

The genus *Ircinia* of the sea sponge family Irciniidae is a prolific source of natural products with a huge variety of different natural product classes like macrolides, alkaloids, steroids, peptides and terpenes (Coll *et al.*, 1997; Kondo *et al.*, 1992; Kobayashi *et al.*, 1995; Mau *et al.*, 1996; Chevallerier *et al.*, 2006). Particularly, regarding the latter compound class, *Ircinia* spp. are known to produce unusual and rare terpenoids, especially sesterterpene tetronic acids in a linear and cyclic form, like ircinianin and its structural congeners (Hofheinz & Schönholzer, 1977; Barrow *et al.*, 1988; Coll *et al.*, 1997; Höller *et al.*, 1997; Balansa *et al.*, 2013; Balansa *et al.*, 2010).

Balansa *et al.* (2013) showed that these analogues exhibit a significant isoform-selective potentiation of glycine-gated chloride channel receptors (GlyRs). The compounds have

Table 1
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O29-H29\cdots O27^i$	0.84 (3)	1.80 (3)	2.6207 (18)	166 (3)

Symmetry code: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$.

therefore the potential to be developed either as molecular tools to probe GlyR function or can serve as lead structures to treat GlyR-mediated neural disorders.

The title compound (Fig. 1) is a polycyclic sesterterpene tetrionic acid with a furan moiety. The furan ring makes a dihedral angle of $35.14 (12)^\circ$ to the 4-hydroxy-3-methylfuran-2(5*H*)-one ring. In the crystal, the molecules are linked by $O-H\cdots O$ hydrogen bonds (Table 1, Fig. 2), forming chains parallel to the *b* axis. The crystal structure of the title compound has already been reported in 1977 by researchers from the pharmaceutical company Hoffmann La Roche (Hofheinz & Schönholzer, 1977; CCDC reference: 1180878). However, in this study the hydrogen atoms were not refined, and only the relative stereochemistry could be deduced. The absolute structure was so far solely determined by asymmetric total synthesis in 1997 (Uenishi *et al.*, 1997).

Synthesis and crystallization

The title compound $C_{25}H_{32}O_4$ was isolated from the marine sponge *Ircinia wistarii*. The sample (voucher number HER6) was collected from Wistarii Reef, Heron Island, Great Barrier Reef, Australia in July 1998 from a depth of 20 m. After collection, the material was stored in EtOH and kept frozen at 253 K until use.

The sponge material (800 g, wet weight) was cut into smaller pieces (2×2 cm) and was extracted with a solvent mixture of $CHCl_3/MeOH$ (1:1, *v/v*; 2 l of volume per extraction step) for three times (after 4, 8 and 20 h). The extraction

solvent of each step was collected and combined. After filtration and evaporation to dryness, 25.46 g crude extract was obtained. The crude extract was redissolved in MeOH and fractionated by preparative reversed phase open column chromatography [Polygoprep 60–50 C_{18} (Macherey-Nagel) as stationary phase] using gravity and stepwise MeOH/ H_2O gradients with increasing lipophilicity and DCM. In total, eleven fractions were gained, and the ircinianin-enriched fraction (MeOH/ H_2O – 90:10) was identified by LC–MS. This fraction was then purified by reversed phase HPLC [Luna Omega 5 μm Polar C18 100 Å column, 250×4.6 mm, at 1.2 ml min^{-1} and UV detection at 215 nm with a 3 min gradient elution, from 20:80 to 55:45 ACN/ H_2O + 0.1% TFA, followed by ramping over 27 min to 90:10], yielding 140 mg of ircinianin, judged as pure based on total ion current profiles, ESI–MS and NMR spectrometry. Suitable crystals were prepared by slow evaporation at room temperature from a ACN/ H_2O (65:35) solution under atmospheric pressure.

Spectroscopic data of the title compound were in accordance with literature data (Balansa *et al.*, 2013). For ease of comparison with related compounds, the title compound was given in the NMR section the same numbering scheme as previously used in the literature (Balansa *et al.*, 2013):

1H NMR (400 MHz, MeOH- d_4): δ 7.38 (H-1, *t*, 1.6), 7.26 (H-4, *m*), 6.30 (H-2, *m*), 5.11 (H-10, *dd*, 10.3, 1.1), 5.03 (H-12, *m*), 3.08 (H-11, *dm*, 10.3), 2.42 (H-15, *m*)^A, 2.41 (H-5 *br t*, 7.5)^A, 2.04 (H-7, *m*), 2.00 (H-17*a*, *m*), 1.89 (H-16*a*, *m*), 1.71 (H-14, *m*), 1.68 (H-6, *m*), 1.65 (H-18, *m*), 1.64 (H-25, *s*), 1.60 (H-20, *m*), 1.57 (H-9, *d*, 1.3), 1.33 (H-16*b*, *m*), 1.31 (H-17*b*, *m*), 0.92 (H-19, *d*, 6.3).

^{13}C NMR (100 MHz, MeOH- d_4): δ 179.2 (C-22, *s*)^B, 177.7 (C-24, *s*)^B, 144.0 (C-1, *d*), 140.3 (C-4, *d*), 137.1 (C-13, *s*), 136.6 (C-8, *s*), 126.5 (C-3, *s*), 125.0 (C-10, *d*), 123.6 (C-12, *d*), 112.1

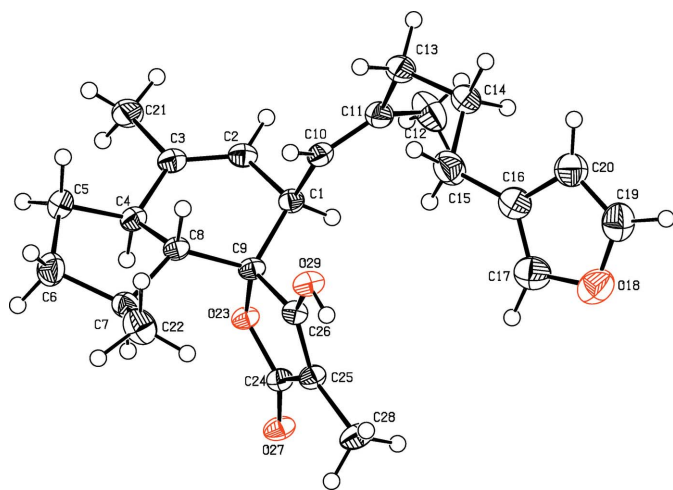


Figure 1
Perspective view of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

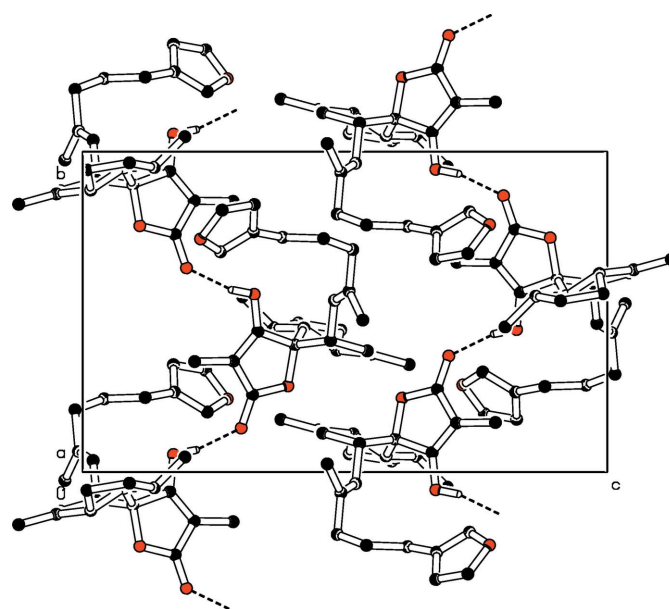


Figure 2
Partial packing diagram of the title compound. View along the *a*-axis.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₂₅ H ₃₂ O ₄
<i>M_r</i>	396.50
Crystal system, space group	Orthorhombic, <i>P</i> 2 ₁ 2 ₁ 2 ₁
Temperature (K)	120
<i>a</i> , <i>b</i> , <i>c</i> (Å)	10.8217 (2), 11.1644 (2), 18.2804 (5)
<i>V</i> (Å ³)	2208.60 (8)
<i>Z</i>	4
Radiation type	Cu <i>K</i> α
μ (mm ⁻¹)	0.63
Crystal size (mm)	0.91 × 0.08 × 0.08
Data collection	
Diffraction	Stoe <i>IPDS</i> 2T
Absorption correction	Integration
<i>T_{min}</i> , <i>T_{max}</i>	0.914, 0.990
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	18913, 3945, 3849
<i>R_{int}</i>	0.018
(sin θ/λ) _{max} (Å ⁻¹)	0.600
Refinement	
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.032, 0.086, 1.08
No. of reflections	3945
No. of parameters	377
H-atom treatment	All H-atom parameters refined
Δρ _{max} , Δρ _{min} (e Å ⁻³)	0.19, -0.20
Absolute structure	Flack <i>x</i> determined using 1639 quotients [(<i>I</i> ⁺) - (<i>I</i> ⁻)] / [(<i>I</i> ⁺) + (<i>I</i> ⁻)] (Parsons <i>et al.</i> , 2013)
Absolute structure parameter	0.03 (9)

Computer programs: *X-RED32* and *X-AREA* (Stoe & Cie, 2019), *SIR2004* (Burla *et al.*, 2005), *SHELXL2018/3* (Sheldrick, 2015) and *PLATON* (Spek, 2020).

(C-2, *d*), 97.5 (C-23, *s*), 86.9 (C-21, *s*), 52.0 (C-20, *d*), 48.7 (C-11, *d*), 46.2 (C-15, *d*), 40.5 (C-7, *t*), 33.6 (C-17, *t*), 33.2 (C-18, *d*), 29.5 (C-6, *t*), 27.3 (C-16, *t*), 25.3 (C-5, *t*), 20.8 (C-14, *q*), 20.7 (C-19, *q*), 16.3 (C-9, *q*), 6.1 (C-25, *q*). [^A Overlapping signals; ^B assignments interchangeable; ^C implied multiplicities determined by DEPT (*qC* = *s*; *CH* = *d*; *CH*₂ = *t*; *CH*₃ = *q*).]

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2. All hydrogen atoms were located

in difference Fourier maps and were refined with isotropic displacement parameters.

Acknowledgements

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full crystallographic data

IUCrData (2020). 5, x201578 [https://doi.org/10.1107/S2414314620015783]

(2*S*,3'*S*,3*a*'*R*,5'*R*,7*a*'*R*)-5'-[(*E*)-5-(Furan-3-yl)-2-methylpent-1-en-1-yl]-3-hydroxy-3',4,7'-trimethyl-1',2',3',3*a*',5',7*a*'-hexahydro-5*H*-spiro[furan-2,4'-inden]-5-one

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(2*S*,3'*S*,3*a*'*R*,5'*R*,7*a*'*R*)-5'-[(*E*)-5-(Furan-3-yl)-2-methylpent-1-en-1-yl]-3-hydroxy-3',4,7'-trimethyl-1',2',3',3*a*',5',7*a*'-hexahydro-5*H*-spiro[furan-2,4'-inden]-5-one

Crystal data

$C_{25}H_{32}O_4$

$M_r = 396.50$

Orthorhombic, $P2_12_12_1$

$a = 10.8217$ (2) Å

$b = 11.1644$ (2) Å

$c = 18.2804$ (5) Å

$V = 2208.60$ (8) Å³

$Z = 4$

$F(000) = 856$

$D_x = 1.192$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.54186$ Å

Cell parameters from 58680 reflections

$\theta = 2.4$ – 68.0°

$\mu = 0.63$ mm⁻¹

$T = 120$ K

Colourless

$0.91 \times 0.08 \times 0.08$ mm

Data collection

Stoe IPDS 2T
diffractometer

Radiation source: Incoatec microSource Cu

Detector resolution: 6.67 pixels mm⁻¹

rotation method, ω scans

Absorption correction: integration

$T_{\min} = 0.914$, $T_{\max} = 0.990$

18913 measured reflections

3945 independent reflections

3849 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.018$

$\theta_{\max} = 67.8^\circ$, $\theta_{\min} = 4.6^\circ$

$h = -12 \rightarrow 12$

$k = -13 \rightarrow 13$

$l = -19 \rightarrow 21$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.032$

$wR(F^2) = 0.086$

$S = 1.08$

3945 reflections

377 parameters

0 restraints

Primary atom site location: dual

Hydrogen site location: difference Fourier map

All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0583P)^2 + 0.3312P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.19$ e Å⁻³

$\Delta\rho_{\min} = -0.20$ e Å⁻³

Absolute structure: Flack x determined using

1639 quotients $[(I^-)-(I)]/[(I^+)+(I)]$ (Parsons *et al.*, 2013)

Absolute structure parameter: 0.03 (9)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.49694 (18)	0.41060 (15)	0.47235 (10)	0.0219 (4)
H1	0.428 (2)	0.357 (2)	0.4651 (13)	0.030 (6)*
C2	0.56472 (18)	0.37841 (17)	0.54252 (10)	0.0253 (4)
H2	0.513 (2)	0.368 (2)	0.5861 (13)	0.025 (5)*
C3	0.68627 (19)	0.36830 (17)	0.55010 (10)	0.0259 (4)
C4	0.76937 (17)	0.38055 (17)	0.48472 (10)	0.0236 (4)
H4	0.786 (2)	0.301 (2)	0.4652 (13)	0.030 (6)*
C5	0.89455 (18)	0.4432 (2)	0.49168 (11)	0.0301 (4)
H5A	0.954 (2)	0.393 (2)	0.5169 (13)	0.029 (4)*
H5AB	0.883 (2)	0.521 (2)	0.5200 (13)	0.029 (4)*
C6	0.9310 (2)	0.4654 (2)	0.41093 (12)	0.0380 (5)
H6A	0.993 (3)	0.406 (3)	0.3941 (16)	0.052 (5)*
H6AB	0.967 (3)	0.552 (3)	0.4041 (16)	0.052 (5)*
C7	0.81215 (17)	0.45092 (17)	0.36372 (10)	0.0259 (4)
H7	0.815 (3)	0.373 (2)	0.3362 (15)	0.042 (7)*
C8	0.70955 (17)	0.44944 (16)	0.42160 (10)	0.0214 (4)
H8	0.694 (2)	0.535 (2)	0.4385 (12)	0.024 (5)*
C9	0.58241 (17)	0.39703 (15)	0.40444 (10)	0.0205 (4)
C10	0.44429 (17)	0.53607 (16)	0.48011 (10)	0.0235 (4)
H10	0.500 (2)	0.602 (2)	0.4651 (13)	0.030 (6)*
C11	0.33472 (19)	0.56473 (18)	0.50782 (10)	0.0284 (4)
C12	0.2416 (2)	0.4744 (2)	0.53387 (18)	0.0485 (6)
H12A	0.181 (4)	0.508 (5)	0.553 (3)	0.111 (9)*
H12B	0.277 (5)	0.402 (4)	0.551 (3)	0.111 (9)*
H12C	0.208 (5)	0.437 (4)	0.482 (3)	0.111 (9)*
C13	0.2955 (2)	0.69443 (19)	0.51459 (12)	0.0333 (5)
H13A	0.368 (3)	0.746 (3)	0.5061 (17)	0.053 (6)*
H13B	0.260 (3)	0.702 (3)	0.5669 (18)	0.053 (6)*
C14	0.1950 (2)	0.7288 (2)	0.45910 (12)	0.0347 (5)
H14A	0.122 (3)	0.665 (3)	0.4642 (16)	0.054 (6)*
H14B	0.158 (3)	0.819 (3)	0.4699 (17)	0.054 (6)*
C15	0.2427 (2)	0.7328 (2)	0.38087 (13)	0.0426 (5)
H15A	0.304 (3)	0.814 (3)	0.3804 (18)	0.066 (6)*
H15B	0.280 (3)	0.652 (3)	0.3716 (18)	0.066 (6)*
C16	0.1423 (2)	0.7499 (2)	0.32518 (12)	0.0363 (5)
C17	0.1271 (2)	0.6877 (2)	0.26257 (13)	0.0422 (5)
H17	0.169 (2)	0.617 (2)	0.2377 (13)	0.033 (6)*
O18	0.02589 (17)	0.72790 (16)	0.22460 (9)	0.0467 (4)
C19	-0.0236 (2)	0.8178 (2)	0.26580 (14)	0.0444 (6)

H19	-0.103 (3)	0.863 (3)	0.2478 (16)	0.049 (8)*
C20	0.0435 (2)	0.8351 (2)	0.32708 (13)	0.0404 (5)
H20	0.029 (3)	0.894 (3)	0.3654 (16)	0.044 (7)*
C21	0.7463 (2)	0.3359 (2)	0.62195 (12)	0.0371 (5)
H21A	0.810 (3)	0.393 (3)	0.6351 (16)	0.052 (5)*
H21B	0.791 (3)	0.266 (3)	0.6173 (16)	0.052 (5)*
H21C	0.686 (3)	0.331 (3)	0.6612 (17)	0.052 (5)*
C22	0.8012 (2)	0.5468 (2)	0.30527 (13)	0.0395 (5)
H22A	0.727 (3)	0.533 (3)	0.2718 (17)	0.053 (4)*
H22B	0.881 (3)	0.544 (3)	0.2759 (17)	0.053 (4)*
H22C	0.792 (3)	0.626 (3)	0.3312 (17)	0.053 (4)*
O23	0.59733 (12)	0.26878 (10)	0.39111 (7)	0.0211 (3)
C24	0.55392 (17)	0.24167 (15)	0.32376 (9)	0.0217 (4)
C25	0.50545 (18)	0.34673 (15)	0.28756 (9)	0.0228 (4)
C26	0.52443 (16)	0.43886 (16)	0.33396 (9)	0.0209 (4)
O27	0.56032 (13)	0.13752 (11)	0.30220 (7)	0.0271 (3)
C28	0.4465 (2)	0.34514 (19)	0.21335 (11)	0.0321 (5)
H28A	0.434 (4)	0.417 (4)	0.195 (2)	0.077 (6)*
H28B	0.375 (4)	0.296 (3)	0.215 (2)	0.077 (6)*
H28C	0.499 (4)	0.307 (3)	0.178 (2)	0.077 (6)*
O29	0.50180 (13)	0.55511 (11)	0.32690 (7)	0.0258 (3)
H29	0.481 (3)	0.569 (3)	0.2837 (18)	0.048 (8)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0240 (8)	0.0201 (8)	0.0215 (9)	-0.0012 (7)	0.0001 (7)	-0.0006 (6)
C2	0.0310 (10)	0.0253 (9)	0.0196 (8)	0.0011 (8)	0.0014 (8)	-0.0005 (7)
C3	0.0316 (10)	0.0257 (9)	0.0205 (9)	0.0010 (8)	-0.0017 (8)	-0.0013 (7)
C4	0.0266 (9)	0.0231 (9)	0.0212 (9)	0.0005 (7)	-0.0032 (7)	-0.0007 (7)
C5	0.0255 (10)	0.0339 (10)	0.0310 (10)	-0.0016 (8)	-0.0049 (8)	-0.0019 (9)
C6	0.0274 (11)	0.0511 (14)	0.0356 (11)	-0.0073 (10)	0.0008 (9)	0.0015 (10)
C7	0.0269 (9)	0.0252 (9)	0.0255 (9)	-0.0016 (7)	0.0036 (7)	-0.0008 (7)
C8	0.0256 (9)	0.0180 (8)	0.0206 (8)	-0.0008 (7)	-0.0006 (7)	-0.0005 (7)
C9	0.0273 (9)	0.0147 (8)	0.0195 (8)	0.0010 (6)	-0.0015 (7)	-0.0012 (6)
C10	0.0263 (9)	0.0218 (8)	0.0224 (8)	-0.0006 (7)	-0.0033 (7)	-0.0026 (7)
C11	0.0295 (10)	0.0281 (9)	0.0276 (9)	0.0021 (8)	-0.0004 (7)	-0.0030 (8)
C12	0.0376 (12)	0.0383 (13)	0.0695 (18)	0.0016 (10)	0.0184 (12)	0.0015 (12)
C13	0.0331 (10)	0.0309 (11)	0.0358 (11)	0.0066 (9)	-0.0017 (9)	-0.0085 (8)
C14	0.0321 (10)	0.0346 (11)	0.0374 (11)	0.0067 (9)	-0.0001 (9)	-0.0071 (9)
C15	0.0372 (12)	0.0515 (14)	0.0393 (12)	0.0121 (11)	0.0019 (10)	0.0029 (11)
C16	0.0388 (12)	0.0351 (10)	0.0351 (11)	-0.0006 (9)	0.0030 (9)	0.0030 (9)
C17	0.0490 (13)	0.0400 (12)	0.0377 (12)	-0.0029 (11)	0.0054 (10)	0.0032 (10)
O18	0.0504 (10)	0.0526 (10)	0.0371 (8)	-0.0140 (8)	-0.0016 (7)	0.0023 (7)
C19	0.0394 (13)	0.0498 (14)	0.0439 (13)	-0.0018 (10)	-0.0021 (10)	0.0103 (11)
C20	0.0414 (12)	0.0399 (12)	0.0398 (12)	0.0038 (10)	-0.0014 (10)	0.0006 (10)
C21	0.0370 (12)	0.0530 (14)	0.0214 (10)	0.0089 (11)	-0.0025 (9)	0.0007 (9)
C22	0.0391 (12)	0.0416 (12)	0.0378 (12)	-0.0005 (10)	0.0090 (9)	0.0124 (10)

O23	0.0291 (7)	0.0148 (6)	0.0192 (6)	0.0009 (5)	-0.0029 (5)	-0.0005 (4)
C24	0.0255 (9)	0.0200 (8)	0.0196 (8)	-0.0018 (7)	0.0004 (7)	-0.0013 (6)
C25	0.0275 (9)	0.0202 (9)	0.0207 (8)	0.0004 (7)	-0.0027 (7)	0.0003 (7)
C26	0.0240 (8)	0.0180 (8)	0.0208 (8)	0.0007 (7)	0.0008 (7)	0.0011 (7)
O27	0.0388 (7)	0.0178 (6)	0.0247 (6)	0.0005 (5)	-0.0051 (6)	-0.0038 (5)
C28	0.0453 (12)	0.0258 (10)	0.0251 (9)	0.0035 (9)	-0.0114 (9)	-0.0027 (8)
O29	0.0379 (7)	0.0177 (6)	0.0217 (6)	0.0035 (5)	-0.0040 (6)	0.0013 (5)

Geometric parameters (Å, °)

C1—C10	1.519 (2)	C13—H13A	0.98 (3)
C1—C2	1.521 (2)	C13—H13B	1.03 (3)
C1—C9	1.555 (2)	C14—C15	1.521 (3)
C1—H1	0.96 (2)	C14—H14A	1.07 (3)
C2—C3	1.327 (3)	C14—H14B	1.10 (3)
C2—H2	0.98 (2)	C15—C16	1.502 (3)
C3—C4	1.502 (3)	C15—H15A	1.12 (3)
C3—C21	1.509 (3)	C15—H15B	1.01 (3)
C4—C5	1.530 (3)	C16—C17	1.349 (3)
C4—C8	1.530 (2)	C16—C20	1.431 (3)
C4—H4	0.97 (2)	C17—O18	1.372 (3)
C5—C6	1.548 (3)	C17—H17	1.01 (3)
C5—H5A	0.97 (2)	O18—C19	1.365 (3)
C5—H5AB	1.02 (3)	C19—C20	1.349 (3)
C6—C7	1.557 (3)	C19—H19	1.05 (3)
C6—H6A	0.99 (3)	C20—H20	0.98 (3)
C6—H6AB	1.05 (3)	C21—H21A	0.97 (3)
C7—C22	1.517 (3)	C21—H21B	0.92 (3)
C7—C8	1.534 (2)	C21—H21C	0.97 (3)
C7—H7	1.00 (3)	C22—H22A	1.02 (3)
C8—C9	1.528 (3)	C22—H22B	1.02 (3)
C8—H8	1.02 (2)	C22—H22C	1.01 (3)
C9—O23	1.4613 (19)	O23—C24	1.352 (2)
C9—C26	1.507 (2)	C24—O27	1.230 (2)
C10—C11	1.329 (3)	C24—C25	1.445 (2)
C10—H10	0.99 (3)	C25—C26	1.349 (3)
C11—C12	1.503 (3)	C25—C28	1.499 (2)
C11—C13	1.514 (3)	C26—O29	1.327 (2)
C12—H12A	0.83 (5)	C28—H28A	0.88 (4)
C12—H12B	0.95 (5)	C28—H28B	0.95 (4)
C12—H12C	1.10 (5)	C28—H28C	0.96 (4)
C13—C14	1.536 (3)	O29—H29	0.84 (3)
C10—C1—C2	108.67 (15)	C11—C13—C14	112.56 (17)
C10—C1—C9	112.79 (14)	C11—C13—H13A	108.7 (19)
C2—C1—C9	111.31 (15)	C14—C13—H13A	108.3 (19)
C10—C1—H1	107.0 (14)	C11—C13—H13B	105.1 (17)
C2—C1—H1	110.1 (14)	C14—C13—H13B	108.9 (17)

C9—C1—H1	106.9 (14)	H13A—C13—H13B	113 (3)
C3—C2—C1	125.88 (17)	C15—C14—C13	112.82 (19)
C3—C2—H2	117.9 (14)	C15—C14—H14A	110.7 (16)
C1—C2—H2	116.2 (14)	C13—C14—H14A	107.4 (17)
C2—C3—C4	120.16 (17)	C15—C14—H14B	105.3 (16)
C2—C3—C21	122.54 (19)	C13—C14—H14B	111.4 (16)
C4—C3—C21	117.16 (17)	H14A—C14—H14B	109 (2)
C3—C4—C5	120.37 (16)	C16—C15—C14	113.3 (2)
C3—C4—C8	113.11 (16)	C16—C15—H15A	108.9 (17)
C5—C4—C8	101.99 (15)	C14—C15—H15A	103.4 (17)
C3—C4—H4	108.6 (14)	C16—C15—H15B	107 (2)
C5—C4—H4	106.4 (14)	C14—C15—H15B	105.6 (19)
C8—C4—H4	105.2 (14)	H15A—C15—H15B	119 (3)
C4—C5—C6	102.69 (16)	C17—C16—C20	105.8 (2)
C4—C5—H5A	111.3 (14)	C17—C16—C15	126.7 (2)
C6—C5—H5A	112.2 (14)	C20—C16—C15	127.5 (2)
C4—C5—H5AB	108.8 (13)	C16—C17—O18	111.0 (2)
C6—C5—H5AB	112.4 (13)	C16—C17—H17	136.5 (14)
H5A—C5—H5AB	109.3 (19)	O18—C17—H17	112.5 (14)
C5—C6—C7	107.52 (16)	C19—O18—C17	105.96 (18)
C5—C6—H6A	111.3 (17)	C20—C19—O18	110.6 (2)
C7—C6—H6A	108.5 (18)	C20—C19—H19	129.1 (17)
C5—C6—H6AB	110.9 (16)	O18—C19—H19	120.3 (16)
C7—C6—H6AB	109.8 (16)	C19—C20—C16	106.7 (2)
H6A—C6—H6AB	109 (2)	C19—C20—H20	127.2 (17)
C22—C7—C8	115.92 (17)	C16—C20—H20	126.1 (17)
C22—C7—C6	112.42 (18)	C3—C21—H21A	111.3 (18)
C8—C7—C6	102.52 (15)	C3—C21—H21B	110.5 (19)
C22—C7—H7	104.9 (16)	H21A—C21—H21B	102 (3)
C8—C7—H7	111.4 (16)	C3—C21—H21C	111.6 (18)
C6—C7—H7	109.8 (16)	H21A—C21—H21C	109 (2)
C9—C8—C4	110.08 (15)	H21B—C21—H21C	112 (3)
C9—C8—C7	120.96 (15)	C7—C22—H22A	111.9 (17)
C4—C8—C7	102.67 (15)	C7—C22—H22B	106.6 (17)
C9—C8—H8	105.9 (12)	H22A—C22—H22B	110 (2)
C4—C8—H8	108.2 (12)	C7—C22—H22C	107.2 (18)
C7—C8—H8	108.6 (12)	H22A—C22—H22C	110 (2)
O23—C9—C26	101.95 (13)	H22B—C22—H22C	111 (3)
O23—C9—C8	108.06 (14)	C24—O23—C9	109.44 (13)
C26—C9—C8	115.58 (15)	O27—C24—O23	118.94 (16)
O23—C9—C1	107.11 (13)	O27—C24—C25	129.87 (17)
C26—C9—C1	113.87 (14)	O23—C24—C25	111.19 (14)
C8—C9—C1	109.53 (14)	C26—C25—C24	106.00 (15)
C11—C10—C1	126.34 (18)	C26—C25—C28	130.00 (17)
C11—C10—H10	118.1 (14)	C24—C25—C28	124.01 (16)
C1—C10—H10	115.5 (14)	O29—C26—C25	131.03 (16)
C10—C11—C12	123.88 (19)	O29—C26—C9	117.58 (15)
C10—C11—C13	120.76 (19)	C25—C26—C9	111.38 (15)

C12—C11—C13	115.36 (19)	C25—C28—H28A	114 (3)
C11—C12—H12A	111 (3)	C25—C28—H28B	109 (2)
C11—C12—H12B	114 (3)	H28A—C28—H28B	114 (3)
H12A—C12—H12B	125 (4)	C25—C28—H28C	111 (2)
C11—C12—H12C	102 (3)	H28A—C28—H28C	104 (3)
H12A—C12—H12C	106 (4)	H28B—C28—H28C	103 (3)
H12B—C12—H12C	95 (3)	C26—O29—H29	109 (2)
C10—C1—C2—C3	-109.2 (2)	C1—C10—C11—C12	-2.5 (3)
C9—C1—C2—C3	15.6 (3)	C1—C10—C11—C13	177.95 (18)
C1—C2—C3—C4	-4.4 (3)	C10—C11—C13—C14	107.9 (2)
C1—C2—C3—C21	180.00 (19)	C12—C11—C13—C14	-71.7 (3)
C2—C3—C4—C5	142.7 (2)	C11—C13—C14—C15	-69.9 (3)
C21—C3—C4—C5	-41.5 (3)	C13—C14—C15—C16	172.2 (2)
C2—C3—C4—C8	21.9 (3)	C14—C15—C16—C17	-133.3 (2)
C21—C3—C4—C8	-162.28 (18)	C14—C15—C16—C20	46.9 (3)
C3—C4—C5—C6	-165.52 (18)	C20—C16—C17—O18	0.3 (3)
C8—C4—C5—C6	-39.4 (2)	C15—C16—C17—O18	-179.6 (2)
C4—C5—C6—C7	16.7 (2)	C16—C17—O18—C19	-0.5 (3)
C5—C6—C7—C22	137.51 (19)	C17—O18—C19—C20	0.5 (3)
C5—C6—C7—C8	12.3 (2)	O18—C19—C20—C16	-0.4 (3)
C3—C4—C8—C9	-50.8 (2)	C17—C16—C20—C19	0.1 (3)
C5—C4—C8—C9	178.42 (14)	C15—C16—C20—C19	179.9 (2)
C3—C4—C8—C7	179.09 (15)	C26—C9—O23—C24	-0.34 (18)
C5—C4—C8—C7	48.36 (17)	C8—C9—O23—C24	-122.55 (15)
C22—C7—C8—C9	77.3 (2)	C1—C9—O23—C24	119.52 (15)
C6—C7—C8—C9	-159.85 (17)	C9—O23—C24—O27	178.94 (16)
C22—C7—C8—C4	-159.65 (18)	C9—O23—C24—C25	-0.9 (2)
C6—C7—C8—C4	-36.81 (18)	O27—C24—C25—C26	-177.91 (19)
C4—C8—C9—O23	-54.40 (18)	O23—C24—C25—C26	2.0 (2)
C7—C8—C9—O23	65.0 (2)	O27—C24—C25—C28	2.5 (3)
C4—C8—C9—C26	-167.82 (14)	O23—C24—C25—C28	-177.65 (18)
C7—C8—C9—C26	-48.4 (2)	C24—C25—C26—O29	177.21 (18)
C4—C8—C9—C1	61.96 (18)	C28—C25—C26—O29	-3.2 (4)
C7—C8—C9—C1	-178.59 (15)	C24—C25—C26—C9	-2.2 (2)
C10—C1—C9—O23	-163.92 (14)	C28—C25—C26—C9	177.4 (2)
C2—C1—C9—O23	73.62 (17)	O23—C9—C26—O29	-177.86 (14)
C10—C1—C9—C26	-52.0 (2)	C8—C9—C26—O29	-61.0 (2)
C2—C1—C9—C26	-174.48 (14)	C1—C9—C26—O29	67.2 (2)
C10—C1—C9—C8	79.12 (18)	O23—C9—C26—C25	1.60 (19)
C2—C1—C9—C8	-43.34 (18)	C8—C9—C26—C25	118.51 (17)
C2—C1—C10—C11	-88.2 (2)	C1—C9—C26—C25	-113.39 (17)
C9—C1—C10—C11	147.88 (19)		

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
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O29—H29···O27 ⁱ	0.84 (3)	1.80 (3)	2.6207 (18)	166 (3)
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Symmetry code: (i) $-x+1, y+1/2, -z+1/2$.