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3-Benzoyl-4-methyl-2-oxo-2H-chromen-7-yl acetate

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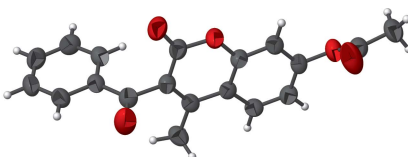
Keywords: crystal structure; 3-benzoyl coumarin.

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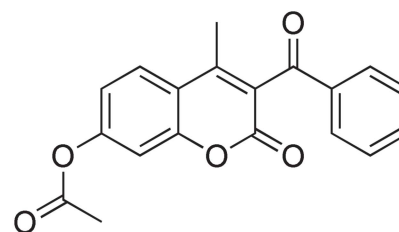
Structural data: full structural data are available from iucrdata.iucr.org

In the title compound, C₁₉H₁₄O₅, the dihedral angle between the coumarin ring system (r.m.s. deviation = 0.026 Å) and the pendant benzoyl group is 81.91 (7)°. In the crystal, weak C—H···O interactions link the molecules into a three-dimensional network.

3D view



Chemical scheme



Structure description

The coumarin nucleus occurs in many natural products (Yu *et al.*, 2003) and it is also widely used in materials chemistry (Swanson *et al.*, 2003). As part of our studies of 3-aryl coumarins, we report here the synthesis and crystal structure of the title compound (Fig. 1).

The C7—O1, C8—O2 and C18—O5 bond lengths are 1.216 (2), 1.208 (2) and 1.186 (2) Å, respectively. They are shorter than the standard C=O bond length [1.231 (2) Å; Gao *et al.*, 2014]. The C14—O4 [1.393 (2) Å], C18—O4 [1.367 (2) Å], C12—O3 [1.378 (2) Å] and C8—O3 [1.370 (2) Å] bond lengths are obviously longer than the C=O bond length, indicating that they are single bonds. As expected, the coumarin ring is nearly planar (r.m.s. deviation = 0.026 Å) and subtends dihedral angles of 81.91 (7) and 65.35 (9)° with the benzoyl and acetate substituents, respectively.

In the crystal, weak C—H···O interactions (Table 1, Fig. 2) link the molecules into a three-dimensional network. Weak aromatic π – π stacking between inversion-related pairs of C11–C16 rings is also observed [centroid–centroid separation = 3.7482 (10), slippage = 0.98 Å].

Synthesis and crystallization

A 25 ml Schlenk tube equipped with a magnetic stirring bar was charged with AgNO₃ (0.25 mmol, 42.5 mg), potassium persulfate (1.0 mmol, 270 mg), 4-methyl-2-oxo-2H-

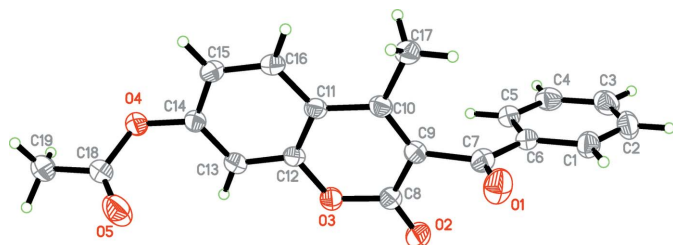


Figure 1
The molecular structure showing 50% probability displacement ellipsoids

chromen-7-yl acetate (0.25 mmol, 54.5 mg), 2-oxo-2-phenylacetic acid (0.5 mmol, 75 mg), and 2 ml of CH₃CN/H₂O (*v*₁/*v*₂ = 1:1) was then added. The reaction mixture was heated in an oil bath at 90°C for 10 h (monitored by TLC). After completion of the reaction, the resulting solution was cooled to room temperature, and the solvent was removed with the aid of a rotary evaporator. The residue was purified by column chromatography on silica gel using ethylacetate/petroleum ether (1:4) as eluant to provide the desired product. Yellow blocks of (I) were recrystallized from trichloromethane solution, m.p. 114–115°C. ¹H NMR (CDCl₃) δ: 7.91 (*dd*, *J*_{H–H} = 8.0 Hz, *J*_{H–H} = 0.8 Hz, 2H), 7.71 (*d*, *J*_{H–H} = 8.7 Hz, 1H), 7.61 (*td*, *J*_{H–H} = 7.4 Hz, *J*_{H–H} = 1.2 Hz, 1H), 7.47 (*t*, *J*_{H–H} = 8.0 Hz, 2H), 7.18 (*d*, *J*_{H–H} = 2.2 Hz, 1H), 7.14 (*dd*, *J*_{H–H} = 8.7 Hz, *J*_{H–H} = 2.2 Hz, 1H), 2.35 (*s*, 3H), 2.33 (*s*, 3H). ¹³C NMR (CDCl₃) δ: 192.9 (C=O), 168.6, 158.4 (C=O), 153.7, 153.6, 149.7, 136.0, 134.3 (CH), 129.3 (CH), 129.0 (CH), 126.2 (CH), 125.3, 118.7 (CH), 117.4, 110.5 (CH), 21.1 (CH₃), 16.0 (CH₃). IR (KBr) *ν* (cm⁻¹): 3066, 2925, 1768, 1720, 1670, 1614, 1450, 1186. HR MS

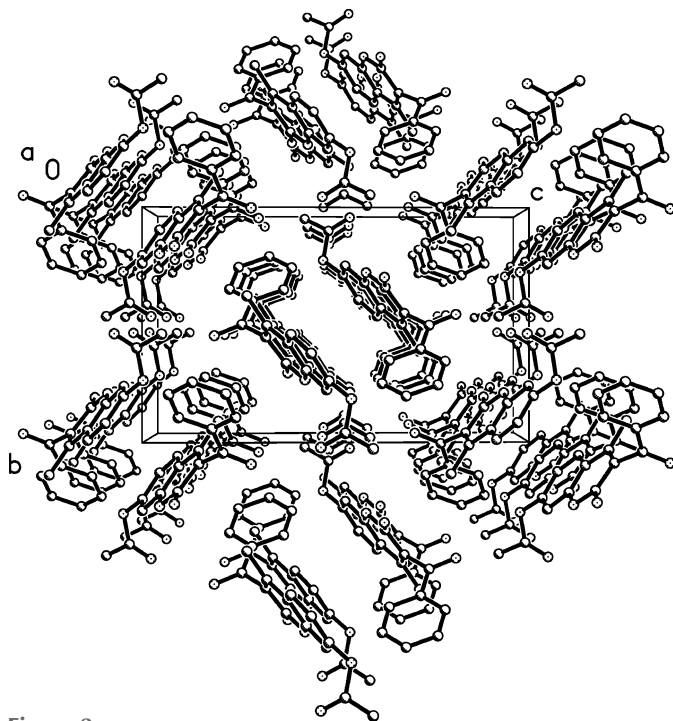


Figure 2
The crystal packing viewed down [100].

Table 1
Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C2–H2···O2 ⁱ	0.93	2.57	3.475 (3)	164
C4–H4···O3 ⁱⁱ	0.93	2.56	3.361 (3)	144
C19–H19C···O3 ⁱⁱⁱ	0.96	2.47	3.409 (2)	165

Symmetry codes: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $-x + 1, -y + 1, -z + 1$; (iii) $-x, -y, -z + 1$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₉ H ₁₄ O ₅
<i>M</i> _r	322.30
Crystal system, space group	Monoclinic, <i>P</i> ₂ / <i>1</i> / <i>n</i>
Temperature (K)	291
<i>a</i> , <i>b</i> , <i>c</i> (Å)	9.1617 (2), 10.2307 (3), 16.8581 (5)
β (°)	92.812 (3)
<i>V</i> (Å ³)	1578.20 (7)
<i>Z</i>	4
Radiation type	Cu <i>K</i> α
μ (mm ⁻¹)	0.82
Crystal size (mm)	0.25 × 0.2 × 0.18
Data collection	
Diffractometer	Agilent Xcalibur, Eos, Gemini
Absorption correction	Multi-scan (<i>CrysAlis PRO</i> ; Agilent, 2013)
<i>T</i> _{min} , <i>T</i> _{max}	0.544, 1.000
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	5738, 2817, 2327
<i>R</i> _{int}	0.037
(sin θ / λ) _{max} (Å ⁻¹)	0.597
Refinement	
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.048, 0.138, 1.04
No. of reflections	2817
No. of parameters	220
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{max}$, $\Delta\rho_{min}$ (e Å ⁻³)	0.26, -0.18

Computer programs: *CrysAlis PRO* (Agilent, 2013), *SHELXS* (Sheldrick, 2008), *SHELXL* (Sheldrick, 2015) and *OLEX2* (Dolomanov *et al.*, 2009).

(ESI) *m/z* 323.0916 [*M* + H]⁺ (calculated for C₁₉H₁₅O₅⁺ 323.0914).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Funding information

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full crystallographic data

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3-Benzoyl-4-methyl-2-oxo-2H-chromen-7-yl acetate

Crystal data

$C_{19}H_{14}O_5$	$F(000) = 672$
$M_r = 322.30$	$D_x = 1.356 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$
$a = 9.1617 (2) \text{ \AA}$	Cell parameters from 2227 reflections
$b = 10.2307 (3) \text{ \AA}$	$\theta = 5.4\text{--}72.2^\circ$
$c = 16.8581 (5) \text{ \AA}$	$\mu = 0.82 \text{ mm}^{-1}$
$\beta = 92.812 (3)^\circ$	$T = 291 \text{ K}$
$V = 1578.20 (7) \text{ \AA}^3$	Block, yellow
$Z = 4$	$0.25 \times 0.2 \times 0.18 \text{ mm}$

Data collection

Agilent Xcalibur, Eos, Gemini diffractometer	5738 measured reflections
Radiation source: Enhance (Cu) X-ray Source	2817 independent reflections
Graphite monochromator	2327 reflections with $I > 2\sigma(I)$
Detector resolution: $16.2312 \text{ pixels mm}^{-1}$	$R_{\text{int}} = 0.037$
ω scans	$\theta_{\text{max}} = 67.0^\circ$, $\theta_{\text{min}} = 5.1^\circ$
Absorption correction: multi-scan (CrysAlisPro; Agilent, 2013)	$h = -10 \rightarrow 7$
$T_{\text{min}} = 0.544$, $T_{\text{max}} = 1.000$	$k = -12 \rightarrow 12$
	$l = -20 \rightarrow 20$

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.048$	$w = 1/[\sigma^2(F_o^2) + (0.0713P)^2 + 0.2573P]$
$wR(F^2) = 0.138$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.04$	$(\Delta/\sigma)_{\text{max}} < 0.001$
2817 reflections	$\Delta\rho_{\text{max}} = 0.26 \text{ e \AA}^{-3}$
220 parameters	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$
0 restraints	Extinction correction: SHELXL-2014/7 (Sheldrick 2014,
Primary atom site location: structure-invariant direct methods	$Fc^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
	Extinction coefficient: 0.0398 (18)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.42067 (18)	0.45664 (18)	0.79826 (9)	0.0787 (5)
O2	0.47734 (14)	0.30569 (16)	0.61932 (10)	0.0695 (5)
O3	0.24925 (13)	0.28252 (13)	0.57529 (8)	0.0549 (4)
O4	-0.23097 (14)	0.21970 (13)	0.46535 (8)	0.0549 (4)
O5	-0.2083 (2)	0.03492 (17)	0.53544 (11)	0.0900 (6)
C1	0.6308 (2)	0.6465 (2)	0.76609 (12)	0.0563 (5)
H1	0.6289	0.6163	0.8181	0.068*
C2	0.7325 (2)	0.7391 (2)	0.74621 (14)	0.0655 (6)
H2	0.7983	0.7717	0.7851	0.079*
C3	0.7372 (2)	0.7834 (2)	0.66899 (14)	0.0650 (6)
H3	0.8058	0.8458	0.6560	0.078*
C4	0.6407 (2)	0.7353 (2)	0.61166 (13)	0.0605 (5)
H4	0.6446	0.7646	0.5596	0.073*
C5	0.5374 (2)	0.64341 (18)	0.63057 (11)	0.0501 (5)
H5	0.4720	0.6115	0.5913	0.060*
C6	0.53109 (19)	0.59858 (17)	0.70822 (10)	0.0452 (4)
C7	0.4219 (2)	0.50110 (19)	0.73141 (11)	0.0505 (5)
C8	0.35343 (19)	0.34511 (19)	0.62240 (11)	0.0502 (5)
C9	0.30657 (19)	0.45207 (17)	0.67102 (10)	0.0459 (4)
C10	0.16730 (19)	0.49705 (16)	0.66636 (10)	0.0446 (4)
C11	0.06022 (18)	0.42967 (16)	0.61466 (10)	0.0418 (4)
C12	0.10533 (18)	0.32206 (16)	0.57183 (10)	0.0435 (4)
C13	0.0120 (2)	0.24892 (18)	0.52289 (11)	0.0482 (4)
H13	0.0457	0.1777	0.4948	0.058*
C14	-0.13290 (19)	0.28508 (17)	0.51713 (11)	0.0454 (4)
C15	-0.18368 (19)	0.39151 (18)	0.55818 (11)	0.0492 (5)
H15	-0.2819	0.4145	0.5534	0.059*
C16	-0.08818 (19)	0.46300 (18)	0.60608 (11)	0.0481 (5)
H16	-0.1225	0.5349	0.6333	0.058*
C17	0.1211 (2)	0.6136 (2)	0.71267 (12)	0.0598 (5)
H17A	0.0810	0.6787	0.6768	0.090*
H17B	0.2042	0.6490	0.7422	0.090*
H17C	0.0485	0.5878	0.7487	0.090*
C18	-0.2599 (2)	0.09088 (19)	0.47921 (12)	0.0528 (5)
C19	-0.3583 (3)	0.0359 (2)	0.41586 (14)	0.0674 (6)
H19A	-0.3360	0.0740	0.3658	0.101*
H19B	-0.4578	0.0551	0.4270	0.101*
H19C	-0.3451	-0.0571	0.4136	0.101*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0800 (11)	0.0931 (13)	0.0617 (9)	-0.0215 (9)	-0.0092 (7)	0.0224 (8)
O2	0.0404 (8)	0.0679 (10)	0.0998 (11)	0.0070 (7)	-0.0007 (7)	-0.0154 (8)
O3	0.0417 (7)	0.0473 (8)	0.0755 (9)	0.0053 (5)	0.0001 (6)	-0.0161 (6)

O4	0.0555 (8)	0.0433 (7)	0.0641 (8)	-0.0033 (6)	-0.0143 (6)	0.0026 (6)
O5	0.1241 (16)	0.0584 (10)	0.0840 (11)	-0.0260 (10)	-0.0321 (10)	0.0174 (9)
C1	0.0539 (11)	0.0613 (13)	0.0530 (10)	0.0007 (9)	-0.0040 (8)	-0.0083 (9)
C2	0.0516 (11)	0.0637 (14)	0.0804 (14)	-0.0075 (10)	-0.0044 (10)	-0.0229 (11)
C3	0.0547 (12)	0.0505 (12)	0.0905 (16)	-0.0055 (9)	0.0108 (10)	-0.0049 (11)
C4	0.0622 (12)	0.0525 (12)	0.0673 (12)	0.0034 (9)	0.0084 (9)	0.0093 (10)
C5	0.0509 (10)	0.0444 (10)	0.0541 (10)	0.0033 (8)	-0.0057 (8)	-0.0002 (8)
C6	0.0441 (9)	0.0403 (9)	0.0506 (9)	0.0041 (7)	-0.0030 (7)	-0.0021 (7)
C7	0.0500 (10)	0.0494 (11)	0.0515 (10)	0.0017 (8)	-0.0034 (8)	0.0040 (8)
C8	0.0418 (10)	0.0437 (10)	0.0648 (11)	0.0008 (8)	0.0002 (8)	-0.0009 (8)
C9	0.0456 (9)	0.0401 (9)	0.0519 (10)	-0.0031 (7)	-0.0001 (7)	0.0032 (7)
C10	0.0506 (10)	0.0356 (9)	0.0474 (9)	0.0006 (7)	0.0005 (7)	0.0024 (7)
C11	0.0425 (9)	0.0348 (9)	0.0482 (9)	0.0016 (7)	0.0028 (7)	0.0034 (7)
C12	0.0405 (9)	0.0347 (9)	0.0551 (9)	0.0027 (7)	0.0019 (7)	0.0018 (7)
C13	0.0496 (10)	0.0366 (9)	0.0583 (10)	0.0020 (7)	0.0013 (8)	-0.0039 (8)
C14	0.0465 (9)	0.0382 (9)	0.0508 (9)	-0.0027 (7)	-0.0053 (7)	0.0061 (7)
C15	0.0419 (9)	0.0467 (10)	0.0585 (10)	0.0069 (8)	-0.0016 (7)	0.0044 (8)
C16	0.0470 (10)	0.0415 (10)	0.0558 (10)	0.0091 (8)	0.0013 (8)	-0.0025 (8)
C17	0.0626 (12)	0.0530 (12)	0.0625 (12)	0.0124 (9)	-0.0096 (9)	-0.0134 (9)
C18	0.0544 (11)	0.0425 (10)	0.0612 (11)	-0.0008 (8)	-0.0013 (8)	-0.0005 (9)
C19	0.0685 (14)	0.0523 (12)	0.0799 (14)	-0.0022 (10)	-0.0106 (11)	-0.0145 (11)

Geometric parameters (Å, °)

O1—C7	1.216 (2)	C8—C9	1.445 (3)
O2—C8	1.208 (2)	C9—C10	1.355 (2)
O3—C8	1.370 (2)	C10—C11	1.454 (2)
O3—C12	1.378 (2)	C10—C17	1.497 (2)
O4—C14	1.393 (2)	C11—C12	1.390 (2)
O4—C18	1.367 (2)	C11—C16	1.402 (2)
O5—C18	1.186 (2)	C12—C13	1.379 (2)
C1—H1	0.9300	C13—H13	0.9300
C1—C2	1.381 (3)	C13—C14	1.377 (3)
C1—C6	1.392 (2)	C14—C15	1.383 (3)
C2—H2	0.9300	C15—H15	0.9300
C2—C3	1.381 (3)	C15—C16	1.372 (2)
C3—H3	0.9300	C16—H16	0.9300
C3—C4	1.369 (3)	C17—H17A	0.9600
C4—H4	0.9300	C17—H17B	0.9600
C4—C5	1.382 (3)	C17—H17C	0.9600
C5—H5	0.9300	C18—C19	1.475 (3)
C5—C6	1.391 (3)	C19—H19A	0.9600
C6—C7	1.479 (3)	C19—H19B	0.9600
C7—C9	1.516 (2)	C19—H19C	0.9600
C8—O3—C12	121.69 (14)	C12—C11—C16	116.77 (15)
C18—O4—C14	118.72 (14)	C16—C11—C10	124.67 (16)
C2—C1—H1	120.0	O3—C12—C11	121.25 (15)

C2—C1—C6	119.91 (19)	O3—C12—C13	115.48 (15)
C6—C1—H1	120.0	C13—C12—C11	123.27 (16)
C1—C2—H2	119.8	C12—C13—H13	121.2
C3—C2—C1	120.39 (19)	C14—C13—C12	117.64 (17)
C3—C2—H2	119.8	C14—C13—H13	121.2
C2—C3—H3	120.0	C13—C14—O4	120.35 (17)
C4—C3—C2	119.9 (2)	C13—C14—C15	121.53 (16)
C4—C3—H3	120.0	C15—C14—O4	117.99 (16)
C3—C4—H4	119.8	C14—C15—H15	120.2
C3—C4—C5	120.5 (2)	C16—C15—C14	119.59 (16)
C5—C4—H4	119.8	C16—C15—H15	120.2
C4—C5—H5	119.9	C11—C16—H16	119.4
C4—C5—C6	120.12 (18)	C15—C16—C11	121.20 (17)
C6—C5—H5	119.9	C15—C16—H16	119.4
C1—C6—C7	118.77 (17)	C10—C17—H17A	109.5
C5—C6—C1	119.16 (18)	C10—C17—H17B	109.5
C5—C6—C7	122.08 (16)	C10—C17—H17C	109.5
O1—C7—C6	122.32 (17)	H17A—C17—H17B	109.5
O1—C7—C9	117.43 (18)	H17A—C17—H17C	109.5
C6—C7—C9	120.25 (15)	H17B—C17—H17C	109.5
O2—C8—O3	116.64 (17)	O4—C18—C19	111.14 (17)
O2—C8—C9	125.67 (18)	O5—C18—O4	121.96 (18)
O3—C8—C9	117.69 (15)	O5—C18—C19	126.90 (19)
C8—C9—C7	114.50 (16)	C18—C19—H19A	109.5
C10—C9—C7	123.35 (17)	C18—C19—H19B	109.5
C10—C9—C8	121.97 (16)	C18—C19—H19C	109.5
C9—C10—C11	118.65 (16)	H19A—C19—H19B	109.5
C9—C10—C17	121.98 (16)	H19A—C19—H19C	109.5
C11—C10—C17	119.37 (15)	H19B—C19—H19C	109.5
C12—C11—C10	118.54 (15)		
O1—C7—C9—C8	-94.8 (2)	C8—O3—C12—C13	-179.03 (16)
O1—C7—C9—C10	80.4 (3)	C8—C9—C10—C11	4.1 (3)
O2—C8—C9—C7	-10.1 (3)	C8—C9—C10—C17	-175.79 (17)
O2—C8—C9—C10	174.6 (2)	C9—C10—C11—C12	-0.3 (2)
O3—C8—C9—C7	170.11 (16)	C9—C10—C11—C16	177.90 (17)
O3—C8—C9—C10	-5.1 (3)	C10—C11—C12—O3	-2.7 (3)
O3—C12—C13—C14	-179.65 (16)	C10—C11—C12—C13	178.08 (16)
O4—C14—C15—C16	-176.12 (16)	C10—C11—C16—C15	-177.57 (16)
C1—C2—C3—C4	-0.1 (3)	C11—C12—C13—C14	-0.3 (3)
C1—C6—C7—O1	-3.7 (3)	C12—O3—C8—O2	-177.65 (17)
C1—C6—C7—C9	176.75 (17)	C12—O3—C8—C9	2.2 (3)
C2—C1—C6—C5	1.1 (3)	C12—C11—C16—C15	0.6 (3)
C2—C1—C6—C7	-179.06 (19)	C12—C13—C14—O4	176.41 (15)
C2—C3—C4—C5	0.6 (3)	C12—C13—C14—C15	0.6 (3)
C3—C4—C5—C6	-0.3 (3)	C13—C14—C15—C16	-0.2 (3)
C4—C5—C6—C1	-0.6 (3)	C14—O4—C18—O5	2.9 (3)
C4—C5—C6—C7	179.57 (18)	C14—O4—C18—C19	-176.37 (17)

C5—C6—C7—O1	176.2 (2)	C14—C15—C16—C11	-0.4 (3)
C5—C6—C7—C9	-3.4 (3)	C16—C11—C12—O3	179.04 (16)
C6—C1—C2—C3	-0.8 (3)	C16—C11—C12—C13	-0.2 (3)
C6—C7—C9—C8	84.8 (2)	C17—C10—C11—C12	179.68 (17)
C6—C7—C9—C10	-100.1 (2)	C17—C10—C11—C16	-2.2 (3)
C7—C9—C10—C11	-170.69 (15)	C18—O4—C14—C13	65.0 (2)
C7—C9—C10—C17	9.4 (3)	C18—O4—C14—C15	-118.98 (19)
C8—O3—C12—C11	1.6 (3)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C2—H2 \cdots O2 ⁱ	0.93	2.57	3.475 (3)	164
C4—H4 \cdots O3 ⁱⁱ	0.93	2.56	3.361 (3)	144
C19—H19C \cdots O3 ⁱⁱⁱ	0.96	2.47	3.409 (2)	165

Symmetry codes: (i) $-x+3/2, y+1/2, -z+3/2$; (ii) $-x+1, -y+1, -z+1$; (iii) $-x, -y, -z+1$.