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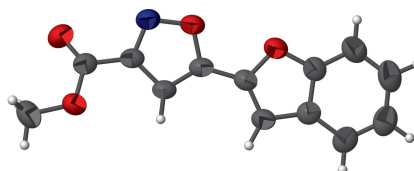
Methyl 5-(1-benzofuran-2-yl)isoxazole-3-carboxylate

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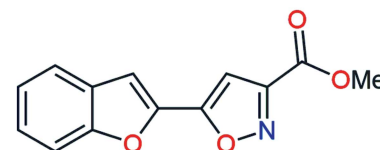
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The title compound, C₁₃H₉NO₄, is almost planar (r.m.s. deviation = 0.071 Å). In the crystal, weak C—H···O and C—H···N interactions connect the molecules into ribbons propagating parallel to the *a*-axis direction.

3D view



Chemical scheme



Structure description

Compounds containing a benzofuran ring system show interesting pharmaceutical applications (Khanam & Shamsuzzaman, 2015; Dawood, 2013). As part of our studies in this area, we now describe the crystal structure of the title compound.

Apart from the methyl hydrogen atoms, the molecule (Fig. 1) is almost planar [dihedral angle between the ring systems = 0.27 (9)°; r.m.s. deviation for the non-hydrogen atoms = 0.071 Å]. In the crystal, weak C—H···O and extremely weak C—H···N interactions form ribbons running parallel to [100] (Table 1, Fig. 2) with adjacent molecules in the chain related by *a*-glide symmetry.

Synthesis and crystallization

The title compound was synthesized based on a literature procedure (Siddiqui *et al.*, 2013) and recrystallized from dimethylformamide solution to yield colourless block-shaped crystals.

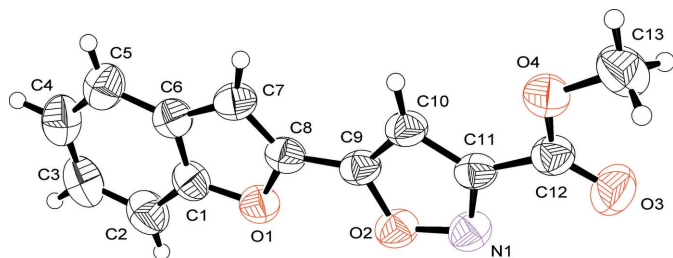


Figure 1
The molecular structure showing 50% displacement ellipsoids.

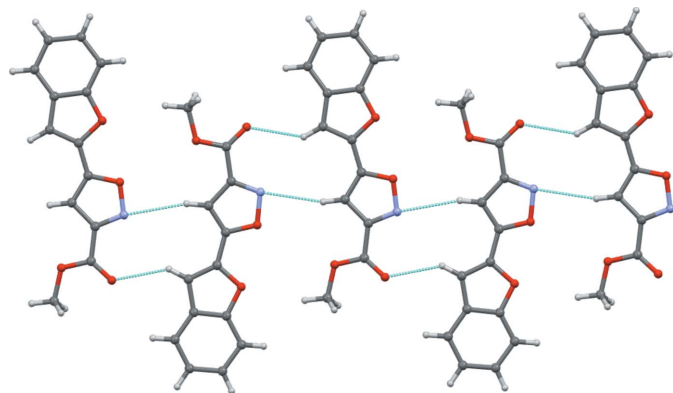


Figure 2
Intermolecular interactions forming a ribbon along the *a*-axis.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Funding information

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Table 1
Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C7–H7···O3 ⁱ	0.93	2.47	3.248 (3)	142
C10–H10···N1 ⁱ	0.93	2.69	3.600 (2)	165

Symmetry code: (i) $x - \frac{1}{2}, -y + \frac{3}{2}, -z + 1$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₃ H ₉ NO ₄
<i>M_r</i>	243.21
Crystal system, space group	Orthorhombic, <i>Pbca</i>
Temperature (K)	296
<i>a</i> , <i>b</i> , <i>c</i> (Å)	11.5170 (6), 8.7431 (4), 22.2595 (13)
<i>V</i> (Å ³)	2241.4 (2)
<i>Z</i>	8
Radiation type	Mo <i>K</i> α
μ (mm ⁻¹)	0.11
Crystal size (mm)	0.29 × 0.24 × 0.20
Data collection	
Diffractometer	Agilent SuperNova, Dual, Cu at zero, Atlas
Absorption correction	Gaussian (<i>CrysAlis PRO</i> ; Agilent, 2014)
<i>T_{min}</i> , <i>T_{max}</i>	0.993, 0.995
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	7841, 2754, 1729
<i>R_{int}</i>	0.023
(sin θ/λ) _{max} (Å ⁻¹)	0.693
Refinement	
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.051, 0.128, 1.04
No. of reflections	2754
No. of parameters	164
H-atom treatment	H-atom parameters constrained
Δρ _{max} , Δρ _{min} (e Å ⁻³)	0.16, −0.18

Computer programs: *CrysAlis PRO* (Agilent, 2014), *SHELXS2013* (Sheldrick, 2008), *SHELXL2013* (Sheldrick, 2015), *ORTEP-3 for Windows* and *WinGX* (Farrugia, 2012) and *CHEMDRAW Ultra* (Cambridge Soft, 2001).

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full crystallographic data

IUCrData (2017). **2**, x171820 [<https://doi.org/10.1107/S241431461701820X>]

Methyl 5-(1-benzofuran-2-yl)isoxazole-3-carboxylate

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Methyl 5-(1-benzofuran-2-yl)isoxazole-3-carboxylate

Crystal data

$C_{13}H_9NO_4$

$M_r = 243.21$

Orthorhombic, *Pbca*

$a = 11.5170$ (6) Å

$b = 8.7431$ (4) Å

$c = 22.2595$ (13) Å

$V = 2241.4$ (2) Å³

$Z = 8$

$F(000) = 1008$

$D_x = 1.441$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2289 reflections

$\theta = 4.7\text{--}27.2^\circ$

$\mu = 0.11$ mm⁻¹

$T = 296$ K

Block, colourless

$0.29 \times 0.24 \times 0.20$ mm

Data collection

Agilent SuperNova, Dual, Cu at zero, Atlas diffractometer

ω scans

Absorption correction: gaussian
(CrysAlis PRO; Agilent, 2014)

$T_{\min} = 0.993$, $T_{\max} = 0.995$

7841 measured reflections

2754 independent reflections

1729 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.023$

$\theta_{\max} = 29.5^\circ$, $\theta_{\min} = 3.5^\circ$

$h = -15 \rightarrow 11$

$k = -8 \rightarrow 11$

$l = -30 \rightarrow 30$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.051$

$wR(F^2) = 0.128$

$S = 1.04$

2754 reflections

164 parameters

0 restraints

Primary atom site location: structure-invariant direct methods

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0435P)^2 + 0.7049P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.16$ e Å⁻³

$\Delta\rho_{\min} = -0.18$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. All hydrogen atoms were placed in calculated positions and refined using a riding model. Methyl C—H bonds were fixed at 0.96 Å, with displacement parameters 1.5 times $U_{\text{eq}}(\text{C})$, and were allowed to spin about the C—C bond. Aromatic C—H distances were set to 0.93 Å and their $U(\text{iso})$ set to 1.2 times the U_{eq} for the atoms to which they are bonded.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.67167 (18)	0.3721 (2)	0.65576 (8)	0.0537 (5)
C2	0.7252 (2)	0.2758 (3)	0.69597 (10)	0.0708 (6)
H2	0.8055	0.2638	0.6967	0.085*
C3	0.6540 (3)	0.1981 (3)	0.73512 (10)	0.0758 (7)
H3	0.6866	0.1318	0.7631	0.091*
C4	0.5339 (2)	0.2169 (3)	0.73359 (11)	0.0730 (6)
H4	0.4880	0.1625	0.7605	0.088*
C5	0.4817 (2)	0.3140 (2)	0.69324 (10)	0.0655 (6)
H5	0.4015	0.3260	0.6927	0.079*
C6	0.55252 (17)	0.3945 (2)	0.65298 (9)	0.0521 (5)
C7	0.53436 (16)	0.5028 (2)	0.60614 (9)	0.0527 (5)
H7	0.4634	0.5416	0.5932	0.063*
C8	0.63964 (16)	0.5379 (2)	0.58428 (8)	0.0490 (4)
C9	0.67308 (15)	0.6425 (2)	0.53750 (8)	0.0480 (4)
C10	0.61160 (16)	0.7336 (2)	0.50057 (8)	0.0501 (4)
H10	0.5314	0.7451	0.4988	0.060*
C11	0.69573 (15)	0.8071 (2)	0.46541 (8)	0.0475 (4)
C12	0.67921 (17)	0.9195 (2)	0.41622 (9)	0.0527 (5)
C13	0.5415 (2)	1.0774 (3)	0.36919 (10)	0.0745 (6)
H13A	0.5791	1.1724	0.3785	0.112*
H13B	0.4589	1.0925	0.3682	0.112*
H13C	0.5674	1.0416	0.3307	0.112*
N1	0.80140 (13)	0.76549 (19)	0.47946 (8)	0.0568 (4)
O1	0.72689 (11)	0.45958 (15)	0.61324 (6)	0.0565 (4)
O2	0.78819 (10)	0.65920 (15)	0.52603 (6)	0.0570 (4)
O3	0.75434 (14)	0.96214 (19)	0.38300 (7)	0.0768 (5)
O4	0.56996 (11)	0.96551 (16)	0.41459 (6)	0.0610 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0607 (12)	0.0534 (11)	0.0471 (10)	0.0046 (9)	−0.0011 (9)	−0.0067 (9)
C2	0.0727 (15)	0.0787 (15)	0.0612 (13)	0.0129 (12)	−0.0069 (11)	0.0009 (12)
C3	0.105 (2)	0.0684 (14)	0.0541 (12)	0.0145 (14)	−0.0020 (13)	0.0021 (11)
C4	0.0961 (19)	0.0607 (13)	0.0621 (13)	−0.0018 (13)	0.0144 (13)	−0.0011 (11)
C5	0.0709 (14)	0.0580 (11)	0.0675 (13)	−0.0009 (11)	0.0119 (11)	−0.0039 (11)
C6	0.0546 (12)	0.0479 (10)	0.0538 (11)	0.0014 (9)	0.0025 (9)	−0.0097 (9)
C7	0.0442 (10)	0.0515 (10)	0.0625 (12)	0.0028 (8)	−0.0015 (9)	−0.0038 (9)
C8	0.0461 (10)	0.0496 (10)	0.0512 (10)	0.0041 (8)	−0.0043 (8)	−0.0075 (9)
C9	0.0387 (9)	0.0517 (10)	0.0537 (10)	−0.0005 (8)	0.0020 (8)	−0.0108 (9)

C10	0.0369 (9)	0.0585 (11)	0.0548 (10)	0.0005 (8)	-0.0005 (8)	-0.0049 (9)
C11	0.0399 (10)	0.0503 (10)	0.0522 (10)	0.0007 (8)	0.0006 (8)	-0.0104 (9)
C12	0.0486 (11)	0.0568 (11)	0.0526 (11)	-0.0025 (9)	0.0015 (9)	-0.0069 (9)
C13	0.0792 (16)	0.0744 (14)	0.0698 (14)	0.0081 (12)	-0.0176 (12)	0.0078 (12)
N1	0.0444 (9)	0.0608 (10)	0.0651 (10)	0.0012 (8)	0.0038 (8)	0.0006 (9)
O1	0.0463 (8)	0.0653 (8)	0.0579 (8)	0.0054 (6)	-0.0023 (6)	0.0010 (7)
O2	0.0407 (7)	0.0623 (8)	0.0680 (9)	0.0049 (6)	0.0003 (6)	0.0019 (7)
O3	0.0590 (9)	0.0931 (12)	0.0782 (10)	-0.0027 (8)	0.0141 (8)	0.0173 (9)
O4	0.0492 (8)	0.0714 (9)	0.0625 (8)	0.0041 (7)	-0.0041 (6)	0.0062 (7)

Geometric parameters (Å, °)

C1—O1	1.373 (2)	C8—C9	1.438 (3)
C1—C2	1.375 (3)	C9—C10	1.346 (3)
C1—C6	1.388 (3)	C9—O2	1.358 (2)
C2—C3	1.377 (3)	C10—C11	1.402 (3)
C2—H2	0.9300	C10—H10	0.9300
C3—C4	1.393 (4)	C11—N1	1.308 (2)
C3—H3	0.9300	C11—C12	1.483 (3)
C4—C5	1.375 (3)	C12—O3	1.198 (2)
C4—H4	0.9300	C12—O4	1.322 (2)
C5—C6	1.401 (3)	C13—O4	1.444 (2)
C5—H5	0.9300	C13—H13A	0.9600
C6—C7	1.424 (3)	C13—H13B	0.9600
C7—C8	1.342 (3)	C13—H13C	0.9600
C7—H7	0.9300	N1—O2	1.401 (2)
C8—O1	1.376 (2)		
O1—C1—C2	125.59 (19)	O1—C8—C9	117.38 (16)
O1—C1—C6	110.40 (16)	C10—C9—O2	109.58 (16)
C2—C1—C6	124.0 (2)	C10—C9—C8	132.65 (17)
C1—C2—C3	116.5 (2)	O2—C9—C8	117.77 (16)
C1—C2—H2	121.7	C9—C10—C11	104.40 (16)
C3—C2—H2	121.7	C9—C10—H10	127.8
C2—C3—C4	121.2 (2)	C11—C10—H10	127.8
C2—C3—H3	119.4	N1—C11—C10	112.46 (17)
C4—C3—H3	119.4	N1—C11—C12	118.69 (17)
C5—C4—C3	121.6 (2)	C10—C11—C12	128.85 (17)
C5—C4—H4	119.2	O3—C12—O4	125.17 (19)
C3—C4—H4	119.2	O3—C12—C11	124.70 (19)
C4—C5—C6	118.3 (2)	O4—C12—C11	110.13 (16)
C4—C5—H5	120.9	O4—C13—H13A	109.5
C6—C5—H5	120.9	O4—C13—H13B	109.5
C1—C6—C5	118.43 (19)	H13A—C13—H13B	109.5
C1—C6—C7	105.78 (17)	O4—C13—H13C	109.5
C5—C6—C7	135.8 (2)	H13A—C13—H13C	109.5
C8—C7—C6	106.55 (17)	H13B—C13—H13C	109.5
C8—C7—H7	126.7	C11—N1—O2	105.10 (14)

C6—C7—H7	126.7	C1—O1—C8	105.19 (14)
C7—C8—O1	112.08 (17)	C9—O2—N1	108.46 (13)
C7—C8—C9	130.53 (17)	C12—O4—C13	116.21 (17)

Hydrogen-bond geometry (Å, °)

<i>D—H</i> ⋯ <i>A</i>	<i>D—H</i>	<i>H</i> ⋯ <i>A</i>	<i>D</i> ⋯ <i>A</i>	<i>D—H</i> ⋯ <i>A</i>
C7—H7⋯O3 ⁱ	0.93	2.47	3.248 (3)	142
C10—H10⋯N1 ⁱ	0.93	2.69	3.600 (2)	165

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