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# 1-Methyl-3-(2-methylphenyl)-4-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidine

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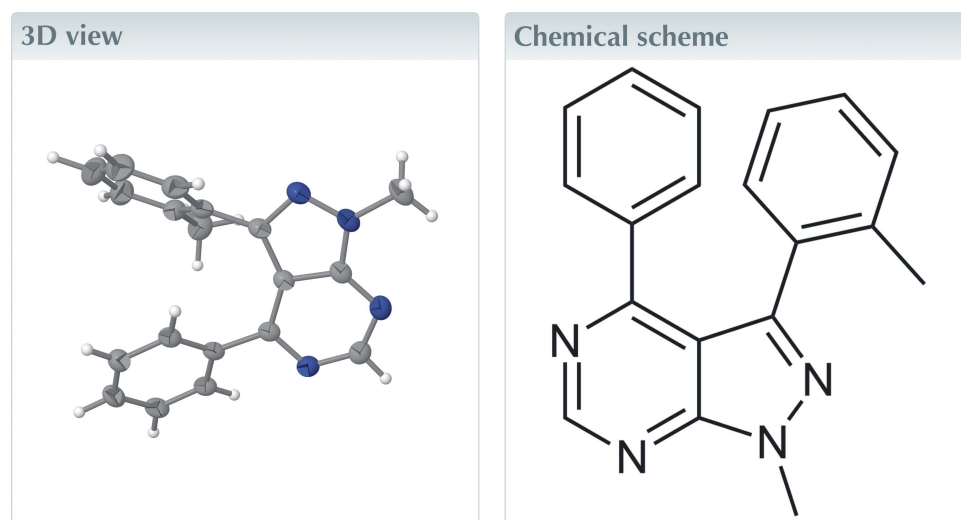
Keywords: crystal structure; pyrimidine;  $\pi$ -stacking; pyrazolo[3,4-*d*]pyrimidine.

CCDC reference: 1576254

Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

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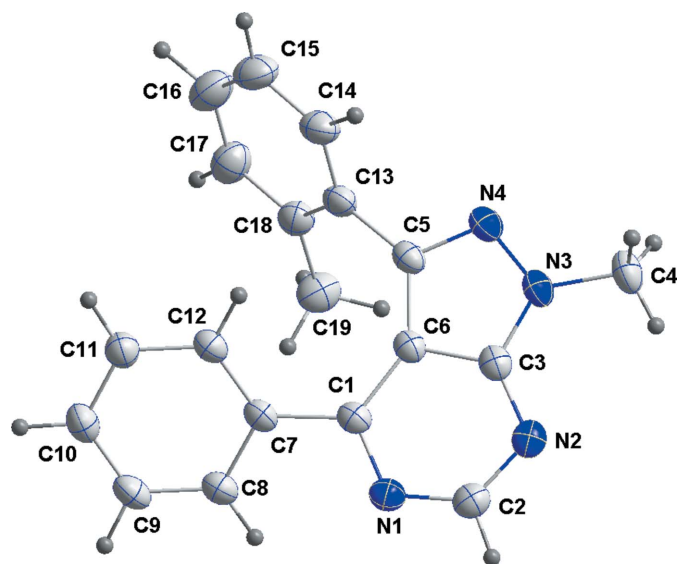
In the title compound, C<sub>19</sub>H<sub>16</sub>N<sub>4</sub>, the pyrazolopyrimidine unit is slightly twisted. A combination of  $\pi$ -stacking and offset  $\pi$ -stacking interactions forms columns along the *b*-axis direction.



## Structure description

Among the various classes of nitrogen containing heterocyclic compounds, pyrazolo[3,4-*d*]pyrimidine derivatives display a broad spectrum of biological activities, because of their structural resemblance to purine nucleobases. In recent years, researchers have reported the use of purine derivatives of pyrazolo[3,4-*d*]pyrimidine as kinase inhibitors (Diner *et al.*, 2012), antiviral agents (Bektemirov *et al.*, 2010) and antitubercular agents (Trivedi *et al.*, 2012). The present paper is a continuation of our research work devoted to the development of pyrazolo[3,4-*d*]pyrimidine derivatives with potential pharmacological activities (El Fal *et al.*, 2013).

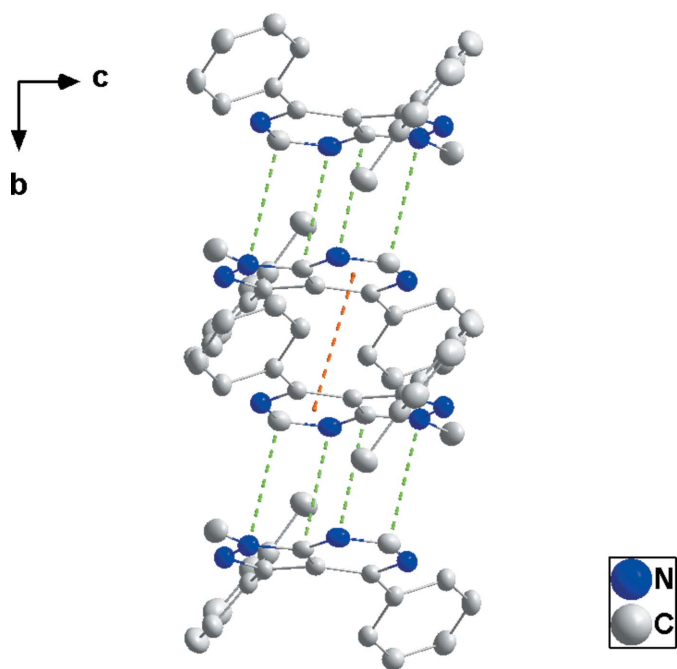
In the title compound, the pyrazolopyrimidine moiety is slightly twisted, as indicated by the dihedral angle of 3.19 (6)° between the mean planes through the five- and six-membered rings (Fig. 1). The *o*-tolyl ring is inclined to the pyrazole ring by 57.26 (6)°, while the phenyl ring is inclined to the pyrimidine ring by 33.04 (6)°. In the crystal,  $\pi$ -stacking interactions between pyrimidine rings form centrosymmetric dimers [centroid-centroid = 3.5178 (6) Å], which are formed into stacks along the *b*-axis direction by offset  $\pi$ -stacking interactions between the C2/N2/C3/N3 portions of the centrosymmetrically related dimers [interplanar spacing = 3.1850 (4) Å; Figs. 2 and 3].



**Figure 1**  
The molecular structure of the title molecule, with the atom-labelling scheme and displacement ellipsoids drawn at the 50% probability level.

### Synthesis and crystallization

Under an atmosphere of argon, a mixture of 1-methyl-4-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidine (0.1 g, 0.47 mmol), 2-iodotoluene (0.12 ml, 0.94 mmol), Cs<sub>2</sub>CO<sub>3</sub> (0.46 g, 1.42 mmol), K<sub>3</sub>PO<sub>4</sub> (0.25 g, 1.18 mmol), 1,10-phenanthroline (0.034 g, 0.19 mmol) and Pd(OAc)<sub>2</sub> (0.021 g, 0.094 mmol) in DMA (3 ml) was flushed with argon and heated to 438 K for



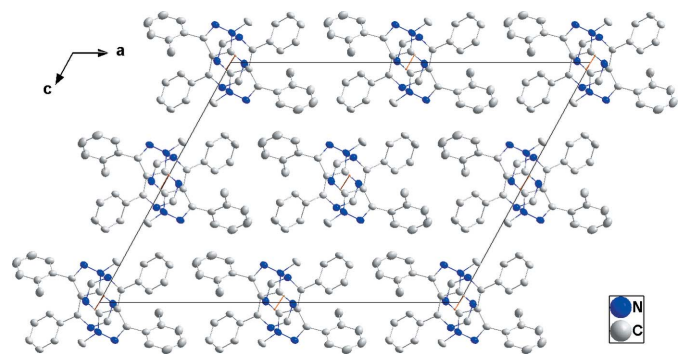
**Figure 2**  
Detail of the  $\pi$ -stacking (orange dashed line) and the offset  $\pi$ -stacking (green dashed lines) viewed along the *a*-axis direction. H atoms have been omitted for clarity.

**Table 1**  
Experimental details.

<b>Crystal data</b>	
Chemical formula	C <sub>19</sub> H <sub>16</sub> N <sub>4</sub>
<i>M<sub>r</sub></i>	300.36
Crystal system, space group	Monoclinic, <i>C2/c</i>
Temperature (K)	150
<i>a</i> , <i>b</i> , <i>c</i> (Å)	25.8546 (6), 6.9306 (2), 19.6783 (5)
$\beta$ (°)	118.658 (1)
<i>V</i> (Å <sup>3</sup> )	3094.16 (14)
<i>Z</i>	8
Radiation type	Cu <i>K</i> $\alpha$
$\mu$ (mm <sup>-1</sup> )	0.62
Crystal size (mm)	0.29 × 0.15 × 0.11
<b>Data collection</b>	
Diffractometer	Bruker D8 VENTURE PHOTON 100 CMOS
Absorption correction	Multi-scan ( <i>SADABS</i> ; Bruker, 2016)
<i>T<sub>min</sub></i> , <i>T<sub>max</sub></i>	0.85, 0.94
No. of measured, independent and observed [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )] reflections	11463, 3124, 2874
<i>R<sub>int</sub></i>	0.028
( <i>sin</i> $\theta$ / $\lambda$ ) <sub>max</sub> (Å <sup>-1</sup> )	0.625
<b>Refinement</b>	
<i>R</i> [ <i>F</i> <sup>2</sup> > 2 $\sigma$ ( <i>F</i> <sup>2</sup> )], <i>wR</i> ( <i>F</i> <sup>2</sup> ), <i>S</i>	0.037, 0.100, 1.04
No. of reflections	3124
No. of parameters	273
H-atom treatment	All H-atom parameters refined
$\Delta\rho_{\max}$ , $\Delta\rho_{\min}$ (e Å <sup>-3</sup> )	0.20, -0.20

Computer programs: *APEX3* and *SAINT* (Bruker, 2016), *SHELXT* (Sheldrick, 2015a), *SHELXL2016* (Sheldrick, 2015b), *DIAMOND* (Brandenburg & Putz, 2012) and *SHELXTL* (Sheldrick, 2008).

48 h. After completion of the reaction, the mixture was allowed to cool to room temperature and the solvent was removed under reduced pressure. Water (15 ml) was added and the resulting aqueous phase was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 15 ml). The combined organic layers were dried with MgSO<sub>4</sub> and concentrated under vacuum. The residue was purified by column chromatography on silica gel (EtOAc/petroleum ether). The title compound was recrystallized from ethanol at room temperature, giving colourless crystals (yield 63%; m.p. 405–407 K).



**Figure 3**  
Packing viewed along the *b*-axis direction, with the  $\pi$ -stacking interactions shown as orange dashed lines. H atoms have been omitted for clarity.

## Refinement

Crystal data, data collection and structure refinement details are presented in Table 1.

## Acknowledgements

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## full crystallographic data

*IUCrData* (2017). **2**, x171370 [https://doi.org/10.1107/S2414314617013700]

1-Methyl-3-(2-methylphenyl)-4-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidine

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1-Methyl-3-(2-methylphenyl)-4-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidine*Crystal data*

$C_{19}H_{16}N_4$

$M_r = 300.36$

Monoclinic,  $C2/c$

$a = 25.8546$  (6) Å

$b = 6.9306$  (2) Å

$c = 19.6783$  (5) Å

$\beta = 118.658$  (1)°

$V = 3094.16$  (14) Å<sup>3</sup>

$Z = 8$

$F(000) = 1264$

$D_x = 1.290$  Mg m<sup>-3</sup>

Cu  $K\alpha$  radiation,  $\lambda = 1.54178$  Å

Cell parameters from 9566 reflections

$\theta = 5.1$ – $74.5$ °

$\mu = 0.62$  mm<sup>-1</sup>

$T = 150$  K

Column, colourless

$0.29 \times 0.15 \times 0.11$  mm

*Data collection*

Bruker D8 VENTURE PHOTON 100 CMOS  
diffractometer

Radiation source: INCOATEC I $\mu$ S micro-focus  
source

Mirror monochromator

Detector resolution: 10.4167 pixels mm<sup>-1</sup>

$\omega$  scans

Absorption correction: multi-scan  
(*SADABS*; Bruker, 2016)

$T_{\min} = 0.85$ ,  $T_{\max} = 0.94$

11463 measured reflections

3124 independent reflections

2874 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$

$\theta_{\max} = 74.5$ °,  $\theta_{\min} = 5.1$ °

$h = -31 \rightarrow 31$

$k = -8 \rightarrow 8$

$l = -24 \rightarrow 22$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.037$

$wR(F^2) = 0.100$

$S = 1.04$

3124 reflections

273 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map

Hydrogen site location: difference Fourier map

All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0497P)^2 + 1.7251P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.20$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.20$  e Å<sup>-3</sup>

Extinction correction: *SHELXL2016* (Sheldrick,  
2015*b*),  $F_c^* = kF_c[1 + 0.001xF_c^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.00168 (14)

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.51451 (4)	0.28267 (14)	0.60439 (5)	0.0309 (2)
N2	0.54018 (4)	0.19678 (13)	0.50584 (6)	0.0315 (2)
N3	0.45894 (4)	0.22510 (14)	0.37751 (6)	0.0312 (2)
N4	0.40026 (4)	0.26735 (14)	0.34325 (6)	0.0319 (2)
C1	0.45863 (5)	0.32694 (15)	0.55272 (6)	0.0264 (2)
C2	0.55098 (5)	0.21902 (17)	0.57827 (7)	0.0320 (3)
C3	0.48400 (5)	0.23908 (15)	0.45530 (6)	0.0277 (2)
H3	0.5903 (6)	0.187 (2)	0.6178 (8)	0.031 (3)*
C4	0.48669 (7)	0.1722 (2)	0.33122 (8)	0.0384 (3)
H4A	0.4883 (9)	0.281 (3)	0.3029 (12)	0.076 (6)*
H4B	0.5266 (8)	0.130 (3)	0.3651 (10)	0.053 (5)*
H4C	0.4639 (8)	0.067 (3)	0.2960 (11)	0.062 (5)*
C5	0.38752 (5)	0.31035 (16)	0.39930 (6)	0.0281 (2)
C6	0.43977 (5)	0.29888 (15)	0.47323 (6)	0.0262 (2)
C7	0.42235 (5)	0.40674 (16)	0.58573 (6)	0.0270 (2)
C8	0.43410 (5)	0.34806 (18)	0.65990 (6)	0.0315 (3)
H8	0.4642 (6)	0.250 (2)	0.6866 (8)	0.039 (4)*
C9	0.40480 (6)	0.43019 (19)	0.69569 (7)	0.0361 (3)
H9	0.4154 (7)	0.390 (2)	0.7493 (9)	0.046 (4)*
C10	0.36296 (6)	0.5724 (2)	0.65802 (7)	0.0372 (3)
H10	0.3424 (7)	0.628 (2)	0.6850 (9)	0.047 (4)*
C11	0.35024 (5)	0.63067 (18)	0.58415 (7)	0.0337 (3)
H11	0.3201 (7)	0.728 (2)	0.5559 (9)	0.046 (4)*
C12	0.37979 (5)	0.54851 (16)	0.54796 (6)	0.0291 (3)
H12	0.3720 (6)	0.593 (2)	0.4963 (8)	0.031 (3)*
C13	0.32570 (5)	0.35885 (17)	0.37772 (6)	0.0306 (3)
C14	0.29848 (6)	0.51143 (19)	0.32611 (7)	0.0387 (3)
H14	0.3218 (6)	0.577 (2)	0.3046 (8)	0.039 (4)*
C15	0.24060 (7)	0.5620 (2)	0.30375 (8)	0.0486 (4)
H15	0.2227 (8)	0.669 (3)	0.2667 (10)	0.060 (5)*
C16	0.20979 (6)	0.4615 (2)	0.33326 (9)	0.0512 (4)
H16	0.1656 (8)	0.497 (3)	0.3175 (10)	0.062 (5)*
C17	0.23610 (6)	0.3093 (2)	0.38378 (9)	0.0443 (3)
H17	0.2145 (7)	0.234 (3)	0.4087 (10)	0.056 (5)*
C18	0.29390 (5)	0.25321 (18)	0.40601 (7)	0.0340 (3)

C19	0.31878 (6)	0.0783 (2)	0.45680 (9)	0.0425 (3)
H19A	0.3415 (12)	0.112 (4)	0.5138 (16)	0.110 (8)*
H19B	0.3482 (10)	0.005 (4)	0.4467 (13)	0.098 (7)*
H19C	0.2857 (10)	-0.014 (4)	0.4487 (13)	0.098 (7)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0322 (5)	0.0299 (5)	0.0291 (5)	-0.0016 (4)	0.0136 (4)	0.0013 (4)
N2	0.0344 (5)	0.0257 (5)	0.0387 (5)	-0.0015 (4)	0.0211 (4)	0.0020 (4)
N3	0.0407 (5)	0.0301 (5)	0.0301 (5)	-0.0013 (4)	0.0228 (4)	-0.0008 (4)
N4	0.0387 (5)	0.0320 (5)	0.0278 (5)	-0.0010 (4)	0.0183 (4)	0.0003 (4)
C1	0.0314 (5)	0.0228 (5)	0.0253 (5)	-0.0030 (4)	0.0137 (4)	0.0008 (4)
C2	0.0314 (6)	0.0288 (6)	0.0354 (6)	-0.0009 (4)	0.0157 (5)	0.0031 (5)
C3	0.0358 (6)	0.0213 (5)	0.0307 (6)	-0.0023 (4)	0.0196 (5)	0.0007 (4)
C4	0.0526 (8)	0.0377 (7)	0.0385 (7)	-0.0001 (6)	0.0329 (6)	-0.0021 (6)
C5	0.0356 (6)	0.0262 (5)	0.0239 (5)	-0.0015 (4)	0.0154 (4)	0.0008 (4)
C6	0.0323 (5)	0.0225 (5)	0.0261 (5)	-0.0012 (4)	0.0160 (4)	0.0010 (4)
C7	0.0308 (5)	0.0272 (5)	0.0231 (5)	-0.0043 (4)	0.0130 (4)	-0.0022 (4)
C8	0.0351 (6)	0.0337 (6)	0.0242 (5)	-0.0016 (5)	0.0129 (4)	0.0004 (4)
C9	0.0418 (6)	0.0436 (7)	0.0250 (5)	-0.0040 (5)	0.0177 (5)	-0.0004 (5)
C10	0.0423 (7)	0.0422 (7)	0.0337 (6)	-0.0018 (5)	0.0236 (5)	-0.0053 (5)
C11	0.0368 (6)	0.0332 (6)	0.0333 (6)	0.0011 (5)	0.0186 (5)	-0.0003 (5)
C12	0.0340 (6)	0.0290 (5)	0.0251 (5)	-0.0030 (4)	0.0148 (4)	-0.0002 (4)
C13	0.0338 (6)	0.0311 (6)	0.0234 (5)	-0.0003 (4)	0.0110 (4)	-0.0028 (4)
C14	0.0435 (7)	0.0380 (7)	0.0291 (6)	0.0029 (5)	0.0129 (5)	0.0030 (5)
C15	0.0474 (8)	0.0447 (8)	0.0405 (7)	0.0126 (6)	0.0106 (6)	0.0076 (6)
C16	0.0358 (7)	0.0515 (8)	0.0560 (9)	0.0066 (6)	0.0137 (6)	-0.0006 (7)
C17	0.0358 (7)	0.0452 (7)	0.0501 (8)	-0.0010 (6)	0.0192 (6)	-0.0023 (6)
C18	0.0327 (6)	0.0351 (6)	0.0319 (6)	-0.0017 (5)	0.0137 (5)	-0.0025 (5)
C19	0.0400 (7)	0.0402 (7)	0.0482 (8)	-0.0028 (6)	0.0218 (6)	0.0081 (6)

*Geometric parameters (Å, °)*

N1—C1	1.3440 (14)	C9—C10	1.3861 (19)
N1—C2	1.3462 (16)	C9—H9	0.993 (16)
N2—C2	1.3235 (16)	C10—C11	1.3879 (17)
N2—C3	1.3412 (15)	C10—H10	0.991 (16)
N3—C3	1.3496 (15)	C11—C12	1.3919 (16)
N3—N4	1.3642 (14)	C11—H11	0.978 (16)
N3—C4	1.4510 (15)	C12—H12	0.987 (14)
N4—C5	1.3277 (14)	C13—C14	1.3998 (17)
C1—C6	1.4109 (15)	C13—C18	1.4004 (17)
C1—C7	1.4806 (15)	C14—C15	1.387 (2)
C2—H3	0.964 (14)	C14—H14	0.998 (15)
C3—C6	1.4111 (15)	C15—C16	1.379 (2)
C4—H4A	0.95 (2)	C15—H15	0.984 (19)
C4—H4B	0.967 (18)	C16—C17	1.384 (2)

C4—H4C	0.982 (19)	C16—H16	1.061 (17)
C5—C6	1.4363 (15)	C17—C18	1.3961 (18)
C5—C13	1.4820 (16)	C17—H17	1.043 (18)
C7—C12	1.3943 (16)	C18—C19	1.5053 (18)
C7—C8	1.4009 (15)	C19—H19A	1.01 (3)
C8—C9	1.3805 (17)	C19—H19B	1.01 (3)
C8—H8	0.977 (15)	C19—H19C	1.02 (3)
C1—N1—C2	118.79 (10)	C10—C9—H9	120.8 (9)
C2—N2—C3	111.54 (10)	C9—C10—C11	119.95 (11)
C3—N3—N4	111.12 (9)	C9—C10—H10	118.2 (9)
C3—N3—C4	128.11 (11)	C11—C10—H10	121.9 (9)
N4—N3—C4	120.76 (10)	C10—C11—C12	120.29 (12)
C5—N4—N3	107.33 (9)	C10—C11—H11	121.3 (9)
N1—C1—C6	118.80 (10)	C12—C11—H11	118.4 (9)
N1—C1—C7	115.52 (9)	C11—C12—C7	120.13 (10)
C6—C1—C7	125.67 (10)	C11—C12—H12	120.5 (8)
N2—C2—N1	128.65 (11)	C7—C12—H12	119.3 (8)
N2—C2—H3	116.0 (8)	C14—C13—C18	119.76 (11)
N1—C2—H3	115.4 (8)	C14—C13—C5	118.51 (11)
N2—C3—N3	125.83 (10)	C18—C13—C5	121.71 (10)
N2—C3—C6	126.73 (10)	C15—C14—C13	120.78 (13)
N3—C3—C6	107.40 (10)	C15—C14—H14	122.0 (8)
N3—C4—H4A	109.8 (12)	C13—C14—H14	117.2 (8)
N3—C4—H4B	109.2 (10)	C16—C15—C14	119.47 (13)
H4A—C4—H4B	108.1 (16)	C16—C15—H15	122.0 (11)
N3—C4—H4C	108.7 (10)	C14—C15—H15	118.5 (11)
H4A—C4—H4C	110.8 (16)	C15—C16—C17	120.23 (13)
H4B—C4—H4C	110.1 (15)	C15—C16—H16	121.3 (10)
N4—C5—C6	109.96 (10)	C17—C16—H16	118.5 (10)
N4—C5—C13	118.42 (10)	C16—C17—C18	121.40 (14)
C6—C5—C13	131.61 (10)	C16—C17—H17	122.0 (10)
C1—C6—C3	115.28 (10)	C18—C17—H17	116.5 (9)
C1—C6—C5	140.57 (10)	C17—C18—C13	118.31 (12)
C3—C6—C5	104.15 (9)	C17—C18—C19	119.00 (12)
C12—C7—C8	118.78 (10)	C13—C18—C19	122.66 (11)
C12—C7—C1	122.23 (10)	C18—C19—H19A	112.7 (16)
C8—C7—C1	118.83 (10)	C18—C19—H19B	113.5 (14)
C9—C8—C7	120.92 (11)	H19A—C19—H19B	104.1 (19)
C9—C8—H8	120.2 (8)	C18—C19—H19C	110.2 (13)
C7—C8—H8	118.8 (8)	H19A—C19—H19C	108.4 (19)
C8—C9—C10	119.92 (11)	H19B—C19—H19C	108 (2)
C8—C9—H9	119.2 (9)		
C3—N3—N4—C5	0.49 (13)	C6—C1—C7—C12	-34.01 (16)
C4—N3—N4—C5	-179.90 (10)	N1—C1—C7—C8	-30.76 (15)
C2—N1—C1—C6	2.10 (15)	C6—C1—C7—C8	150.72 (11)
C2—N1—C1—C7	-176.53 (10)	C12—C7—C8—C9	-0.82 (17)

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C3—N2—C2—N1	-2.75 (17)	C1—C7—C8—C9	174.62 (11)
C1—N1—C2—N2	1.98 (18)	C7—C8—C9—C10	0.21 (19)
C2—N2—C3—N3	-177.90 (10)	C8—C9—C10—C11	0.56 (19)
C2—N2—C3—C6	-0.49 (16)	C9—C10—C11—C12	-0.70 (19)
N4—N3—C3—N2	176.18 (10)	C10—C11—C12—C7	0.09 (18)
C4—N3—C3—N2	-3.39 (19)	C8—C7—C12—C11	0.66 (17)
N4—N3—C3—C6	-1.64 (12)	C1—C7—C12—C11	-174.61 (10)
C4—N3—C3—C6	178.79 (11)	N4—C5—C13—C14	-56.95 (15)
N3—N4—C5—C6	0.86 (12)	C6—C5—C13—C14	123.35 (13)
N3—N4—C5—C13	-178.91 (9)	N4—C5—C13—C18	121.40 (12)
N1—C1—C6—C3	-4.64 (15)	C6—C5—C13—C18	-58.30 (17)
C7—C1—C6—C3	173.83 (10)	C18—C13—C14—C15	1.29 (19)
N1—C1—C6—C5	175.06 (13)	C5—C13—C14—C15	179.68 (12)
C7—C1—C6—C5	-6.5 (2)	C13—C14—C15—C16	0.5 (2)
N2—C3—C6—C1	4.03 (16)	C14—C15—C16—C17	-1.1 (2)
N3—C3—C6—C1	-178.17 (9)	C15—C16—C17—C18	-0.1 (2)
N2—C3—C6—C5	-175.77 (10)	C16—C17—C18—C13	1.9 (2)
N3—C3—C6—C5	2.02 (11)	C16—C17—C18—C19	-175.90 (13)
N4—C5—C6—C1	178.48 (13)	C14—C13—C18—C17	-2.47 (18)
C13—C5—C6—C1	-1.8 (2)	C5—C13—C18—C17	179.20 (11)
N4—C5—C6—C3	-1.79 (12)	C14—C13—C18—C19	175.27 (12)
C13—C5—C6—C3	177.93 (11)	C5—C13—C18—C19	-3.06 (18)
N1—C1—C7—C12	144.51 (11)		

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