

# 3,5-Bis(trifluoromethyl)phenyl 4-methylbenzene-sulfonate

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Received 30 June 2017

Accepted 3 July 2017

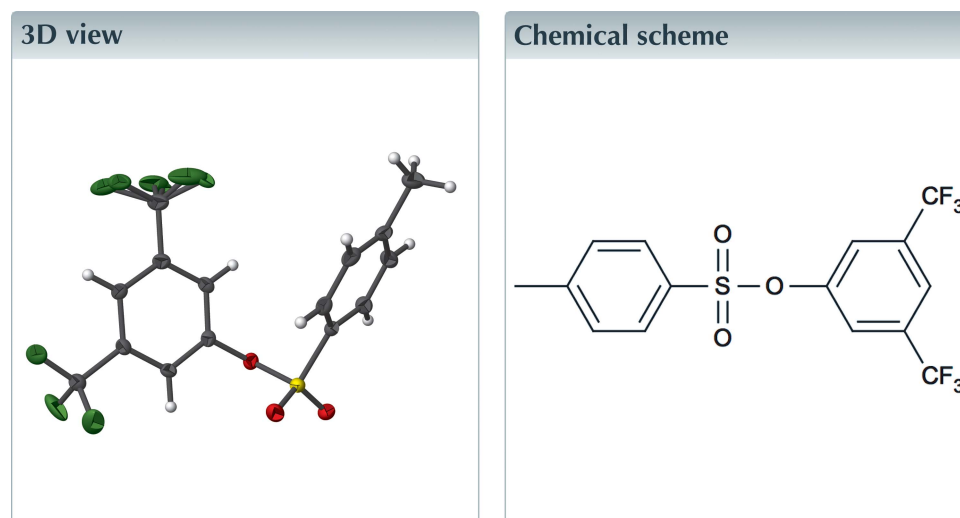
Edited by M. Bolte, Goethe-Universität Frankfurt, Germany

Keywords: crystal structure; tosylates; cross-coupling reactions.

CCDC reference: 1560197

Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

Molecules of the title compound,  $C_{15}H_{10}F_6O_3S$ , are composed of 3,5-bis(trifluoromethyl)phenyl substituted with a toluene-4-sulfonate group. The dihedral angle between two aromatic moieties is  $45.10(5)^\circ$ . In the crystal, molecules are connected by weak  $C-H \cdots O$  and  $C-H \cdots F$  contacts. One of the trifluoromethyl groups is disordered.

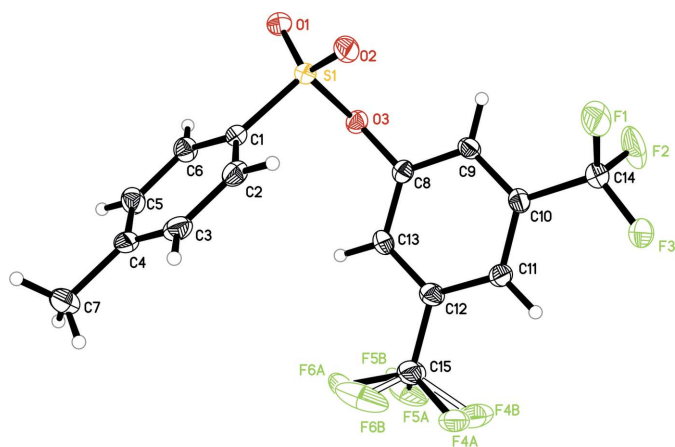


## Structure description

Transition-metal-catalyzed cross-coupling reactions can be applied to aryl tosylates (Ackermann *et al.*, 2006; Zhou & Fu, 2003). We apply 4-methylbenzenesulfonates to study the reaction progress and performance using first-row metal catalysts. The high natural abundance of these non-precious transition metal elements makes them an interesting target of study (Torborg & Beller, 2009).

In the asymmetric unit of the title compound, there is one independent molecule. The molecular structure is shown in Fig. 1. In the molecular structure, the bond lengths and angles are within normal ranges (Allen *et al.*, 2002). One of the trifluoromethyl groups in the molecule is disordered with occupation factors of 0.65 and 0.35. The dihedral angle between the aromatic moieties is  $45.10(5)^\circ$ .

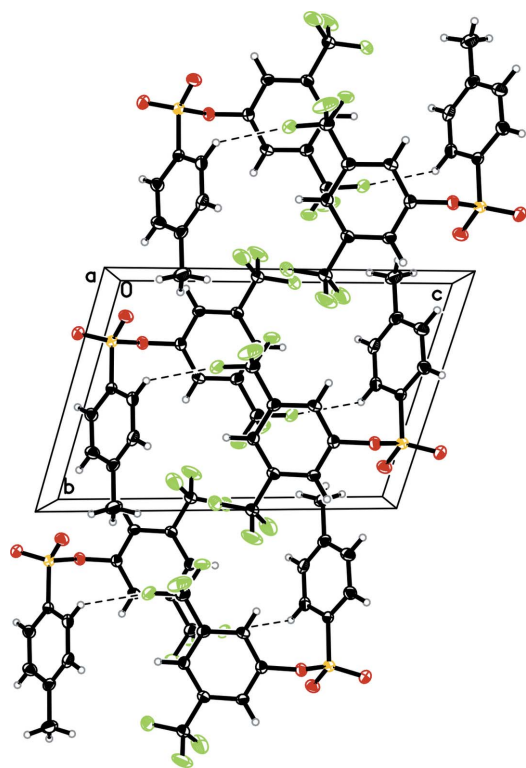
In the crystal (Fig. 2), there are two hydrogen bonds (Table 1), both of which connect the molecules into dimers. There are also two  $F \cdots \pi$  T-shape halogen bonds to a 3,5-di(trifluoromethyl)phenyl moiety. The distances of  $F5A^i$  and  $F5B^i$  from the centre of a plane defined by atoms C9–C13 of the 3,5-di(trifluoromethyl)phenyl ring are 3.348(12) and 3.256(19) Å [symmetry code: (i)  $-x, -y + 1, -z + 1$ ].



**Figure 1**  
The molecular structure of the title compound, with displacement ellipsoids drawn at the 50% probability level.

### Synthesis and crystallization

3,5-Di(trifluoromethyl)phenyl 4-methylbenzenesulfonate was synthesized according to a procedure described by Murai and co-workers (Murai *et al.*, 2012). The crystallization was performed in a diethyl ether solution. Diethyl ether (0.6 ml) was placed in storage reaction vials (8 ml) with silicone septa. The title compound was placed in small portions until a saturated solution was obtained. The solution was warmed, then left to stand in refrigerator ( $-20^{\circ}\text{C}$ ).



**Figure 2**  
The crystal packing of the title compound, viewed along the *a* axis. Disordered fluorine atoms have been omitted for clarity. Hydrogen bonds are shown as dashed lines

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^{\circ}$ ).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C2—H2...F4A <sup>i</sup>	0.93	2.55	3.308 (3)	139
C9—H9...O1 <sup>ii</sup>	0.93	2.40	3.2959 (17)	162

Symmetry codes: (i)  $-x + 1, -y + 1, -z + 1$ ; (ii)  $-x, -y, -z$ .

**Table 2**  
Experimental details.

Crystal data	
Chemical formula	$\text{C}_{15}\text{H}_{10}\text{F}_6\text{O}_3\text{S}$
$M_r$	384.29
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	100
<i>a</i> , <i>b</i> , <i>c</i> ( $\text{\AA}$ )	8.2805 (3), 8.6053 (3), 12.2396 (4)
$\alpha$ , $\beta$ , $\gamma$ ( $^{\circ}$ )	103.519 (3), 99.935 (3), 105.188 (3)
<i>V</i> ( $\text{\AA}^3$ )	792.56 (5)
<i>Z</i>	2
Radiation type	Mo $K\alpha$
$\mu$ ( $\text{mm}^{-1}$ )	0.28
Crystal size (mm)	$0.4 \times 0.25 \times 0.24$
Data collection	
Diffractometer	Oxford Diffraction Xcalibur
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	5437, 3070, 2614
$R_{\text{int}}$	0.013
$(\sin \theta/\lambda)_{\text{max}}$ ( $\text{\AA}^{-1}$ )	0.616
Refinement	
$R[F^2 > 2\sigma(F^2)]$ , $wR(F^2)$ , <i>S</i>	0.030, 0.080, 1.06
No. of reflections	3070
No. of parameters	254
No. of restraints	36
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}$ , $\Delta\rho_{\text{min}}$ ( $\text{e \AA}^{-3}$ )	0.27, $-0.37$

Computer programs: *CrysAlis CCD* and *CrysAlis RED* (Oxford Diffraction, 2008), *SHELXS2014* (Sheldrick, 2008), *SHELXL2014* (Sheldrick, 2015) and *SHELXTL* (Sheldrick, 2008).

### Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

### Funding information

Funding for this research was provided by: Narodowe Centrum Nauki (grant No. 2014/15/D/ST5/02731).

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## full crystallographic data

*IUCrData* (2017). 2, x170981 [https://doi.org/10.1107/S2414314617009816]

## 3,5-Bis(trifluoromethyl)phenyl 4-methylbenzenesulfonate

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*Crystal data*

$C_{15}H_{10}F_6O_3S$

$M_r = 384.29$

Triclinic,  $P\bar{1}$

$a = 8.2805$  (3) Å

$b = 8.6053$  (3) Å

$c = 12.2396$  (4) Å

$\alpha = 103.519$  (3)°

$\beta = 99.935$  (3)°

$\gamma = 105.188$  (3)°

$V = 792.56$  (5) Å<sup>3</sup>

$Z = 2$

$F(000) = 388$

$D_x = 1.610$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 5437 reflections

$\theta = 3.5$ – $26.0$ °

$\mu = 0.28$  mm<sup>-1</sup>

$T = 100$  K

Irregular, colourless

$0.4 \times 0.25 \times 0.24$  mm

*Data collection*

Oxford Diffraction Xcalibur  
diffractometer

Radiation source: fine-focus sealed tube

Detector resolution: 1024 x 1024 with blocks 2  
x 2 pixels mm<sup>-1</sup>

$\omega$ -scan

5437 measured reflections

3070 independent reflections

2614 reflections with  $I > 2\sigma(I)$

$R_{int} = 0.013$

$\theta_{max} = 26.0$ °,  $\theta_{min} = 3.5$ °

$h = -10 \rightarrow 10$

$k = -6 \rightarrow 10$

$l = -15 \rightarrow 15$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.030$

$wR(F^2) = 0.080$

$S = 1.06$

3070 reflections

254 parameters

36 restraints

Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0445P)^2 + 0.1137P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = 0.002$

$\Delta\rho_{max} = 0.27$  e Å<sup>-3</sup>

$\Delta\rho_{min} = -0.37$  e Å<sup>-3</sup>

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** All H atoms were found in a difference map but set to idealized positions and treated as riding with  $C_{Ar}-H = 0.93$  Å and  $U_{iso}(H) = 1.2U_{eq}(C)$  and with  $C_{methyl}-H = 0.96$  Å and  $U_{iso}(H) = 1.5U_{eq}(C)$ .

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.18701 (5)	0.27629 (4)	0.03515 (3)	0.01913 (11)	
O1	0.07383 (14)	0.23781 (13)	-0.07602 (9)	0.0260 (3)	
O2	0.28474 (14)	0.16929 (13)	0.05977 (9)	0.0253 (3)	
O3	0.05670 (13)	0.27937 (12)	0.11983 (8)	0.0194 (2)	
F1	0.27606 (16)	-0.09183 (13)	0.34224 (9)	0.0447 (3)	
F2	0.03104 (14)	-0.12166 (12)	0.38253 (10)	0.0457 (3)	
F3	0.26421 (14)	0.01544 (11)	0.51597 (8)	0.0371 (3)	
F4A	0.3770 (3)	0.6057 (3)	0.62880 (19)	0.0405 (6)	0.65
F5A	0.1573 (14)	0.6652 (12)	0.5462 (8)	0.0432 (17)	0.65
F6A	0.3974 (7)	0.7323 (6)	0.4990 (3)	0.0254 (6)	0.65
F5B	0.134 (2)	0.669 (2)	0.5219 (14)	0.0266 (17)	0.35
F4B	0.2762 (7)	0.5775 (6)	0.6351 (4)	0.0582 (13)	0.35
F6B	0.3987 (15)	0.7169 (12)	0.5299 (7)	0.055 (2)	0.35
C1	0.32014 (19)	0.48591 (18)	0.07914 (12)	0.0179 (3)	
C2	0.48144 (19)	0.53300 (19)	0.15499 (13)	0.0215 (3)	
H2	0.5222	0.4533	0.1802	0.026*	
C3	0.5807 (2)	0.7017 (2)	0.19247 (13)	0.0246 (3)	
H3	0.6894	0.7346	0.2429	0.030*	
C4	0.5209 (2)	0.82248 (19)	0.15611 (13)	0.0234 (3)	
C5	0.3587 (2)	0.77115 (19)	0.07913 (14)	0.0242 (3)	
H5	0.3177	0.8507	0.0538	0.029*	
C6	0.2580 (2)	0.60379 (19)	0.03999 (13)	0.0216 (3)	
H6	0.1503	0.5704	-0.0117	0.026*	
C7	0.6294 (2)	1.0054 (2)	0.19699 (15)	0.0336 (4)	
H7A	0.6946	1.0293	0.1418	0.050*	
H7B	0.7071	1.0288	0.2710	0.050*	
H7C	0.5556	1.0747	0.2042	0.050*	
C8	0.12143 (18)	0.28625 (18)	0.23573 (12)	0.0172 (3)	
C9	0.12658 (18)	0.13922 (17)	0.26128 (13)	0.0183 (3)	
H9	0.0925	0.0376	0.2028	0.022*	
C10	0.18379 (19)	0.14705 (18)	0.37620 (13)	0.0183 (3)	
C11	0.23417 (19)	0.29824 (18)	0.46388 (13)	0.0193 (3)	
H11	0.2714	0.3021	0.5409	0.023*	
C12	0.22819 (19)	0.44313 (18)	0.43511 (13)	0.0212 (3)	
C13	0.17096 (19)	0.43894 (18)	0.32017 (13)	0.0196 (3)	
H13	0.1663	0.5364	0.3010	0.024*	
C14	0.1894 (2)	-0.01242 (19)	0.40458 (14)	0.0247 (3)	
C15	0.2802 (3)	0.6082 (2)	0.52824 (14)	0.0308 (4)	

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0247 (2)	0.01547 (19)	0.01623 (19)	0.00521 (15)	0.00510 (15)	0.00420 (14)
O1	0.0326 (6)	0.0220 (6)	0.0168 (5)	0.0023 (5)	0.0022 (5)	0.0034 (4)
O2	0.0344 (6)	0.0198 (5)	0.0274 (6)	0.0126 (5)	0.0131 (5)	0.0088 (5)

O3	0.0201 (5)	0.0210 (5)	0.0170 (5)	0.0058 (4)	0.0030 (4)	0.0070 (4)
F1	0.0808 (9)	0.0347 (6)	0.0450 (7)	0.0411 (6)	0.0343 (6)	0.0212 (5)
F2	0.0434 (6)	0.0265 (5)	0.0610 (8)	-0.0035 (5)	0.0023 (5)	0.0253 (5)
F3	0.0649 (7)	0.0238 (5)	0.0235 (5)	0.0183 (5)	0.0016 (5)	0.0105 (4)
F4A	0.0712 (15)	0.0204 (9)	0.0191 (9)	0.0125 (11)	-0.0070 (11)	0.0012 (7)
F5A	0.038 (3)	0.027 (2)	0.056 (4)	0.0084 (15)	0.022 (3)	-0.008 (2)
F6A	0.0257 (11)	0.0137 (9)	0.0306 (14)	0.0016 (8)	0.0007 (11)	0.0049 (10)
F5B	0.034 (4)	0.011 (2)	0.039 (4)	0.012 (2)	0.017 (3)	0.006 (3)
F4B	0.116 (4)	0.033 (2)	0.0211 (18)	0.039 (3)	-0.005 (3)	-0.0005 (15)
F6B	0.032 (3)	0.037 (4)	0.070 (5)	0.004 (3)	0.009 (4)	-0.023 (3)
C1	0.0217 (7)	0.0169 (7)	0.0149 (7)	0.0049 (6)	0.0063 (6)	0.0042 (6)
C2	0.0247 (8)	0.0249 (8)	0.0169 (7)	0.0095 (7)	0.0054 (6)	0.0077 (6)
C3	0.0227 (8)	0.0302 (8)	0.0152 (7)	0.0036 (7)	0.0027 (6)	0.0027 (6)
C4	0.0279 (8)	0.0203 (8)	0.0187 (8)	0.0031 (7)	0.0113 (6)	0.0007 (6)
C5	0.0280 (8)	0.0200 (8)	0.0286 (9)	0.0102 (7)	0.0108 (7)	0.0092 (7)
C6	0.0213 (8)	0.0217 (8)	0.0221 (8)	0.0070 (6)	0.0043 (6)	0.0076 (6)
C7	0.0381 (10)	0.0219 (8)	0.0314 (10)	0.0005 (7)	0.0106 (8)	-0.0008 (7)
C8	0.0147 (7)	0.0199 (7)	0.0176 (7)	0.0048 (6)	0.0047 (6)	0.0066 (6)
C9	0.0188 (7)	0.0140 (7)	0.0195 (7)	0.0026 (6)	0.0050 (6)	0.0028 (6)
C10	0.0193 (7)	0.0162 (7)	0.0213 (8)	0.0059 (6)	0.0074 (6)	0.0071 (6)
C11	0.0225 (8)	0.0195 (7)	0.0166 (7)	0.0081 (6)	0.0050 (6)	0.0051 (6)
C12	0.0248 (8)	0.0177 (7)	0.0213 (8)	0.0088 (6)	0.0056 (6)	0.0039 (6)
C13	0.0220 (7)	0.0153 (7)	0.0235 (8)	0.0072 (6)	0.0058 (6)	0.0075 (6)
C14	0.0346 (9)	0.0183 (7)	0.0221 (8)	0.0086 (7)	0.0076 (7)	0.0072 (6)
C15	0.0468 (11)	0.0209 (9)	0.0233 (9)	0.0151 (9)	0.0010 (8)	0.0042 (7)

*Geometric parameters (Å, °)*

S1—O1	1.4234 (11)	C4—C5	1.395 (2)
S1—O2	1.4246 (11)	C4—C7	1.507 (2)
S1—O3	1.6208 (11)	C5—C6	1.382 (2)
S1—C1	1.7488 (15)	C5—H5	0.9300
O3—C8	1.4089 (17)	C6—H6	0.9300
F1—C14	1.3293 (19)	C7—H7A	0.9600
F2—C14	1.3374 (19)	C7—H7B	0.9600
F3—C14	1.3334 (18)	C7—H7C	0.9600
F4A—C15	1.356 (3)	C8—C13	1.378 (2)
F5A—C15	1.270 (11)	C8—C9	1.382 (2)
F6A—C15	1.402 (5)	C9—C10	1.384 (2)
F5B—C15	1.432 (19)	C9—H9	0.9300
F4B—C15	1.398 (5)	C10—C11	1.388 (2)
F6B—C15	1.158 (10)	C10—C14	1.502 (2)
C1—C2	1.387 (2)	C11—C12	1.383 (2)
C1—C6	1.393 (2)	C11—H11	0.9300
C2—C3	1.388 (2)	C12—C13	1.394 (2)
C2—H2	0.9300	C12—C15	1.502 (2)
C3—C4	1.391 (2)	C13—H13	0.9300
C3—H3	0.9300		

O1—S1—O2	121.19 (7)	C8—C9—C10	118.28 (13)
O1—S1—O3	102.31 (6)	C8—C9—H9	120.9
O2—S1—O3	108.14 (6)	C10—C9—H9	120.9
O1—S1—C1	110.29 (7)	C9—C10—C11	121.09 (13)
O2—S1—C1	110.36 (7)	C9—C10—C14	118.55 (13)
O3—S1—C1	102.61 (6)	C11—C10—C14	120.37 (13)
C8—O3—S1	117.42 (9)	C12—C11—C10	119.08 (14)
C2—C1—C6	121.31 (14)	C12—C11—H11	120.5
C2—C1—S1	119.98 (12)	C10—C11—H11	120.5
C6—C1—S1	118.66 (11)	C11—C12—C13	121.06 (14)
C1—C2—C3	118.59 (14)	C11—C12—C15	120.26 (14)
C1—C2—H2	120.7	C13—C12—C15	118.67 (13)
C3—C2—H2	120.7	C8—C13—C12	118.07 (13)
C2—C3—C4	121.34 (14)	C8—C13—H13	121.0
C2—C3—H3	119.3	C12—C13—H13	121.0
C4—C3—H3	119.3	F1—C14—F3	107.05 (13)
C3—C4—C5	118.76 (14)	F1—C14—F2	106.36 (13)
C3—C4—C7	121.10 (15)	F3—C14—F2	106.91 (12)
C5—C4—C7	120.14 (15)	F1—C14—C10	111.94 (13)
C6—C5—C4	120.95 (15)	F3—C14—C10	112.39 (13)
C6—C5—H5	119.5	F2—C14—C10	111.81 (13)
C4—C5—H5	119.5	F5A—C15—F4A	111.0 (4)
C5—C6—C1	119.04 (14)	F6B—C15—F4B	114.9 (5)
C5—C6—H6	120.5	F5A—C15—F6A	106.2 (5)
C1—C6—H6	120.5	F4A—C15—F6A	101.2 (2)
C4—C7—H7A	109.5	F6B—C15—F5B	107.7 (8)
C4—C7—H7B	109.5	F4B—C15—F5B	94.9 (7)
H7A—C7—H7B	109.5	F6B—C15—C12	119.4 (5)
C4—C7—H7C	109.5	F5A—C15—C12	115.2 (5)
H7A—C7—H7C	109.5	F4A—C15—C12	112.22 (16)
H7B—C7—H7C	109.5	F4B—C15—C12	109.2 (2)
C13—C8—C9	122.41 (13)	F6A—C15—C12	109.9 (2)
C13—C8—O3	118.51 (12)	F5B—C15—C12	107.7 (7)
C9—C8—O3	119.03 (12)		
O1—S1—O3—C8	169.55 (10)	C14—C10—C11—C12	179.80 (14)
O2—S1—O3—C8	40.56 (11)	C10—C11—C12—C13	0.7 (2)
C1—S1—O3—C8	-76.08 (11)	C10—C11—C12—C15	179.37 (15)
O1—S1—C1—C2	-152.88 (12)	C9—C8—C13—C12	0.1 (2)
O2—S1—C1—C2	-16.32 (15)	O3—C8—C13—C12	177.43 (12)
O3—S1—C1—C2	98.72 (13)	C11—C12—C13—C8	-0.4 (2)
O1—S1—C1—C6	29.51 (14)	C15—C12—C13—C8	-179.14 (15)
O2—S1—C1—C6	166.07 (11)	C9—C10—C14—F1	52.61 (19)
O3—S1—C1—C6	-78.89 (12)	C11—C10—C14—F1	-127.77 (16)
C6—C1—C2—C3	0.5 (2)	C9—C10—C14—F3	173.13 (13)
S1—C1—C2—C3	-177.05 (11)	C11—C10—C14—F3	-7.2 (2)
C1—C2—C3—C4	0.4 (2)	C9—C10—C14—F2	-66.61 (18)

C2—C3—C4—C5	-0.9 (2)	C11—C10—C14—F2	113.01 (16)
C2—C3—C4—C7	-179.91 (14)	C11—C12—C15—F6B	116.5 (6)
C3—C4—C5—C6	0.5 (2)	C13—C12—C15—F6B	-64.8 (6)
C7—C4—C5—C6	179.48 (14)	C11—C12—C15—F5A	-109.7 (5)
C4—C5—C6—C1	0.4 (2)	C13—C12—C15—F5A	69.1 (5)
C2—C1—C6—C5	-0.9 (2)	C11—C12—C15—F4A	18.7 (3)
S1—C1—C6—C5	176.65 (12)	C13—C12—C15—F4A	-162.53 (17)
S1—O3—C8—C13	101.10 (13)	C11—C12—C15—F4B	-18.6 (3)
S1—O3—C8—C9	-81.44 (14)	C13—C12—C15—F4B	160.1 (3)
C13—C8—C9—C10	0.0 (2)	C11—C12—C15—F6A	130.5 (3)
O3—C8—C9—C10	-177.35 (12)	C13—C12—C15—F6A	-50.8 (3)
C8—C9—C10—C11	0.3 (2)	C11—C12—C15—F5B	-120.5 (7)
C8—C9—C10—C14	179.88 (13)	C13—C12—C15—F5B	58.2 (7)
C9—C10—C11—C12	-0.6 (2)		

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C2—H2...F4A <sup>i</sup>	0.93	2.55	3.308 (3)	139
C9—H9...O1 <sup>ii</sup>	0.93	2.40	3.2959 (17)	162

Symmetry codes: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x, -y, -z$ .