

# 3,5-Dimethoxyphenyl 4-methylbenzenesulfonate

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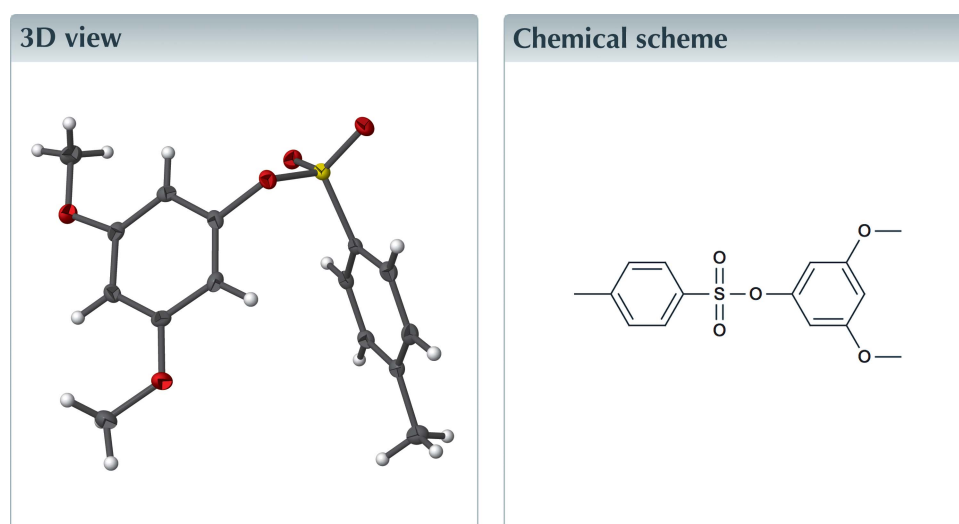
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CCDC reference: 1559861

Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

Molecules of the title compound,  $C_{15}H_{16}O_5S$ , are composed of a 3,5-dimethoxyphenyl moiety substituted with a toluene-4-sulfonate group. The dihedral angle between two aromatic rings is  $57.23(4)^\circ$ . In the crystal, molecules are connected by weak  $C-H \cdots O$  hydrogen bonds and  $S \cdots O$  van der Waals interactions.



## Structure description

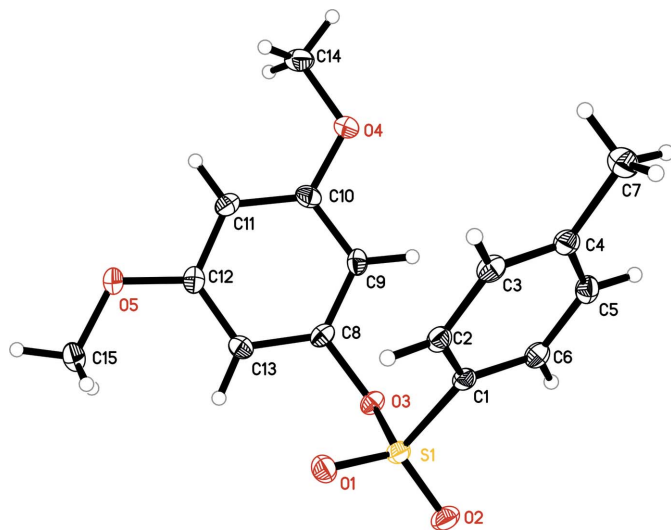
Aryl tosylates attract considerable attention as electrophiles in transition-metal-catalyzed cross-coupling reactions (Chen *et al.*, 2015; Nguyen *et al.*, 2003). 4-Methylbenzenesulfonate derivatives emerged as substrates for first-row transition metal catalysts. These non-precious transition metal elements are suitable alternatives to execute challenging cross-coupling with higher efficiency (Ananikov, 2015).

In the asymmetric unit of the title compound there is one independent molecule. The molecular structure is shown in Fig. 1. In the molecular structure, the bond lengths and angles are within normal ranges (Allen *et al.*, 2002). The dihedral angle between two aromatic moieties is  $57.23(4)^\circ$ .

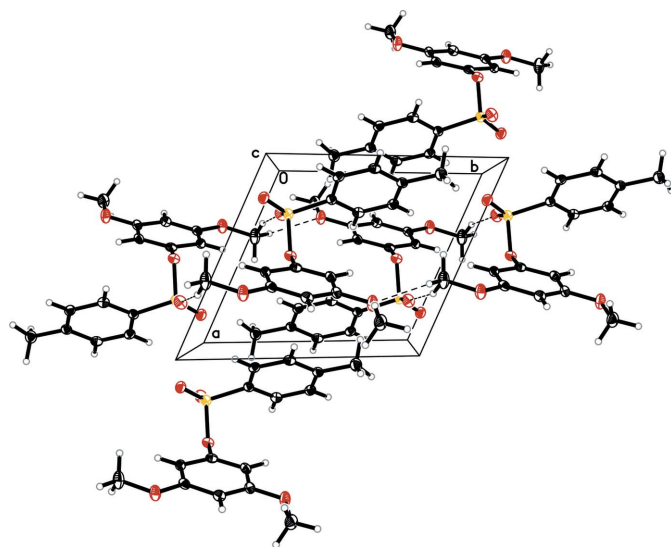
The crystal structure (Fig. 2) features weak  $C-H \cdots O$  hydrogen bonds (Table 1) and  $S1 \cdots O1^i$  van der Waals interactions [symmetry code: (i)  $-x + 2, -y + 2, -z + 1$ . The  $S1 \cdots O1^i$  distance is  $3.2929(10)$  Å]. The  $S1 \cdots O1^i$  contacts connect the molecules into dimers, while  $C-H \cdots O$  bonds arrange the molecules into chains along  $b$  axis.

## Synthesis and crystallization

3,5-Dimethoxyphenyl 4-methylbenzenesulfonate was synthesized according to the procedure described by Murai and co-workers (Murai *et al.*, 2012). The crystallization was performed in a diethyl ether solution. Diethyl ether (0.6 ml) was placed in storage reaction vials (8 ml) with silicone septa. The title compound was placed in small portions



**Figure 1**  
The molecular structure of the title compound, with displacement ellipsoids drawn at the 50% probability level.



**Figure 2**  
The crystal packing of the title compound, viewed along the *c* axis.

until a saturated solution was obtained. The solution was warmed, then left to stand in a refrigerator ( $-20^{\circ}\text{C}$ ).

### Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^{\circ}$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C15}-\text{H15A}\cdots\text{O4}^{\text{i}}$	0.96	2.35	3.2051 (18)	147
$\text{C15}-\text{H15C}\cdots\text{O2}^{\text{ii}}$	0.96	2.51	3.4063 (19)	155

Symmetry codes: (i)  $x, y + 1, z$ ; (ii)  $-x + 1, -y + 2, -z + 1$ .

**Table 2**  
Experimental details.

Crystal data	
Chemical formula	$\text{C}_{15}\text{H}_{16}\text{O}_5\text{S}$
$M_r$	308.34
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	100
$a, b, c$ ( $\text{\AA}$ )	7.8066 (2), 8.9238 (2), 11.9303 (3)
$\alpha, \beta, \gamma$ ( $^{\circ}$ )	106.583 (2), 94.701 (2), 111.358 (2)
$V$ ( $\text{\AA}^3$ )	725.58 (3)
$Z$	2
Radiation type	Mo $K\alpha$
$\mu$ ( $\text{mm}^{-1}$ )	0.24
Crystal size (mm)	$0.26 \times 0.25 \times 0.24$
Data collection	
Diffractometer	Oxford Diffraction Xcalibur
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	4805, 2795, 2316
$R_{\text{int}}$	0.013
$(\sin \theta/\lambda)_{\text{max}}$ ( $\text{\AA}^{-1}$ )	0.617
Refinement	
$R[F^2 > 2\sigma(F^2)]$ , $wR(F^2)$ , $S$	0.029, 0.077, 1.05
No. of reflections	2795
No. of parameters	193
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}$ , $\Delta\rho_{\text{min}}$ ( $\text{e \AA}^{-3}$ )	0.38, $-0.35$

Computer programs: *CrysAlis CCD* and *CrysAlis RED* (Oxford Diffraction, 2008), *SHELXS2014* (Sheldrick, 2008), *SHELXL2014* (Sheldrick, 2015) and *SHELXTL* (Sheldrick, 2008).

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## full crystallographic data

*IUCrData* (2017). 2, x170980 [https://doi.org/10.1107/S2414314617009804]

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*Crystal data*

$C_{15}H_{16}O_5S$	$Z = 2$
$M_r = 308.34$	$F(000) = 324$
Triclinic, $P\bar{1}$	$D_x = 1.411 \text{ Mg m}^{-3}$
$a = 7.8066 (2) \text{ \AA}$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$b = 8.9238 (2) \text{ \AA}$	Cell parameters from 4805 reflections
$c = 11.9303 (3) \text{ \AA}$	$\theta = 3.1\text{--}26.0^\circ$
$\alpha = 106.583 (2)^\circ$	$\mu = 0.24 \text{ mm}^{-1}$
$\beta = 94.701 (2)^\circ$	$T = 100 \text{ K}$
$\gamma = 111.358 (2)^\circ$	Irregular, colourless
$V = 725.58 (3) \text{ \AA}^3$	$0.26 \times 0.25 \times 0.24 \text{ mm}$

*Data collection*

Oxford Diffraction Xcalibur diffractometer	2795 independent reflections
Radiation source: fine-focus sealed tube	2316 reflections with $I > 2\sigma(I)$
Detector resolution: 1024 x 1024 with blocks 2 x 2 pixels $\text{mm}^{-1}$	$R_{\text{int}} = 0.013$
$\omega$ -scan	$\theta_{\text{max}} = 26.0^\circ$ , $\theta_{\text{min}} = 3.1^\circ$
4805 measured reflections	$h = -8 \rightarrow 9$
	$k = -10 \rightarrow 10$
	$l = -14 \rightarrow 14$

*Refinement*

Refinement on $F^2$	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.029$	$w = 1/[\sigma^2(F_o^2) + (0.0437P)^2 + 0.0255P]$
$wR(F^2) = 0.077$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.05$	$(\Delta/\sigma)_{\text{max}} < 0.001$
2795 reflections	$\Delta\rho_{\text{max}} = 0.38 \text{ e \AA}^{-3}$
193 parameters	$\Delta\rho_{\text{min}} = -0.35 \text{ e \AA}^{-3}$
0 restraints	

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** All H atoms were found in a difference map but set to idealized positions and treated as riding with  $C_{\text{Ar}}\text{---H} = 0.93 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  and with  $C_{\text{methyl}}\text{---H} = 0.96 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ .

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.73013 (5)	0.83595 (4)	0.47367 (3)	0.01539 (11)
O1	0.81146 (14)	0.96873 (12)	0.42577 (9)	0.0195 (2)
C1	0.80548 (19)	0.67126 (18)	0.41824 (12)	0.0147 (3)
C2	0.91926 (19)	0.68105 (18)	0.33471 (12)	0.0157 (3)
H2	0.9673	0.7794	0.3145	0.019*
O2	0.73567 (14)	0.87416 (13)	0.59863 (9)	0.0217 (2)
O3	0.50777 (13)	0.74229 (12)	0.41772 (8)	0.0157 (2)
C3	0.95988 (19)	0.54111 (18)	0.28197 (12)	0.0168 (3)
H3	1.0349	0.5460	0.2253	0.020*
C4	0.89064 (19)	0.39332 (18)	0.31217 (12)	0.0166 (3)
O4	0.26779 (15)	0.29839 (13)	0.04463 (9)	0.0234 (3)
C5	0.78274 (19)	0.39000 (18)	0.39973 (12)	0.0173 (3)
H5	0.7394	0.2940	0.4229	0.021*
O5	0.32254 (15)	0.85268 (13)	0.06289 (9)	0.0221 (2)
C6	0.7393 (2)	0.52739 (18)	0.45258 (12)	0.0165 (3)
H6	0.6665	0.5236	0.5105	0.020*
C8	0.44594 (18)	0.69113 (18)	0.29240 (12)	0.0148 (3)
C7	0.9329 (2)	0.24146 (19)	0.25186 (14)	0.0224 (3)
H7A	0.8954	0.2084	0.1667	0.034*
H7B	0.8649	0.1479	0.2774	0.034*
H7C	1.0653	0.2714	0.2730	0.034*
C9	0.39824 (19)	0.52299 (18)	0.22586 (12)	0.0163 (3)
H9	0.4173	0.4479	0.2607	0.020*
C10	0.3204 (2)	0.46833 (18)	0.10448 (13)	0.0173 (3)
C11	0.29768 (19)	0.58182 (18)	0.05261 (12)	0.0172 (3)
H11	0.2472	0.5449	-0.0287	0.021*
C12	0.3511 (2)	0.75243 (19)	0.12328 (13)	0.0163 (3)
C13	0.42417 (19)	0.80989 (18)	0.24526 (12)	0.0157 (3)
H13	0.4570	0.9227	0.2931	0.019*
C14	0.1783 (2)	0.2341 (2)	-0.07898 (13)	0.0261 (4)
H14A	0.2629	0.2905	-0.1225	0.039*
H14B	0.0668	0.2557	-0.0868	0.039*
H14C	0.1451	0.1132	-0.1106	0.039*
C15	0.3825 (2)	1.0314 (2)	0.12916 (14)	0.0275 (4)
H15A	0.3640	1.0896	0.0759	0.041*
H15B	0.5134	1.0791	0.1661	0.041*
H15C	0.3102	1.0449	0.1897	0.041*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.01689 (19)	0.01626 (19)	0.01268 (19)	0.00745 (15)	0.00219 (13)	0.00373 (14)
O1	0.0201 (5)	0.0157 (5)	0.0223 (6)	0.0064 (4)	0.0022 (4)	0.0074 (4)
C1	0.0143 (7)	0.0157 (7)	0.0124 (7)	0.0067 (6)	-0.0007 (5)	0.0026 (6)
C2	0.0148 (7)	0.0159 (7)	0.0139 (7)	0.0036 (6)	0.0006 (6)	0.0058 (6)

O2	0.0265 (6)	0.0261 (6)	0.0124 (5)	0.0139 (5)	0.0019 (4)	0.0024 (4)
O3	0.0157 (5)	0.0206 (5)	0.0119 (5)	0.0081 (4)	0.0040 (4)	0.0056 (4)
C3	0.0144 (7)	0.0214 (8)	0.0142 (7)	0.0069 (6)	0.0037 (6)	0.0057 (6)
C4	0.0142 (7)	0.0172 (7)	0.0160 (7)	0.0065 (6)	-0.0008 (6)	0.0033 (6)
O4	0.0343 (6)	0.0172 (6)	0.0176 (6)	0.0134 (5)	0.0002 (5)	0.0020 (4)
C5	0.0160 (7)	0.0167 (7)	0.0189 (7)	0.0048 (6)	0.0014 (6)	0.0086 (6)
O5	0.0329 (6)	0.0195 (6)	0.0188 (5)	0.0137 (5)	0.0041 (5)	0.0097 (4)
C6	0.0158 (7)	0.0202 (8)	0.0139 (7)	0.0067 (6)	0.0040 (6)	0.0069 (6)
C8	0.0120 (7)	0.0207 (8)	0.0121 (7)	0.0067 (6)	0.0036 (5)	0.0056 (6)
C7	0.0242 (8)	0.0205 (8)	0.0237 (8)	0.0106 (7)	0.0069 (7)	0.0066 (7)
C9	0.0166 (7)	0.0190 (7)	0.0185 (8)	0.0102 (6)	0.0049 (6)	0.0097 (6)
C10	0.0174 (7)	0.0162 (7)	0.0187 (8)	0.0090 (6)	0.0048 (6)	0.0033 (6)
C11	0.0176 (7)	0.0213 (8)	0.0129 (7)	0.0089 (6)	0.0027 (6)	0.0047 (6)
C12	0.0156 (7)	0.0199 (8)	0.0188 (7)	0.0097 (6)	0.0068 (6)	0.0103 (6)
C13	0.0150 (7)	0.0147 (7)	0.0180 (7)	0.0068 (6)	0.0060 (6)	0.0047 (6)
C14	0.0329 (9)	0.0219 (8)	0.0185 (8)	0.0128 (7)	-0.0013 (7)	-0.0006 (7)
C15	0.0440 (10)	0.0207 (8)	0.0242 (9)	0.0174 (8)	0.0080 (7)	0.0112 (7)

*Geometric parameters (Å, °)*

S1—O2	1.4244 (10)	C6—H6	0.9300
S1—O1	1.4254 (10)	C8—C9	1.373 (2)
S1—O3	1.6101 (10)	C8—C13	1.3871 (19)
S1—C1	1.7556 (14)	C7—H7A	0.9600
S1—O1 <sup>i</sup>	3.2929 (10)	C7—H7B	0.9600
C1—C2	1.3885 (19)	C7—H7C	0.9600
C1—C6	1.3910 (19)	C9—C10	1.3931 (19)
C2—C3	1.3887 (19)	C9—H9	0.9300
C2—H2	0.9300	C10—C11	1.382 (2)
O3—C8	1.4171 (15)	C11—C12	1.396 (2)
C3—C4	1.395 (2)	C11—H11	0.9300
C3—H3	0.9300	C12—C13	1.3896 (19)
C4—C5	1.3947 (19)	C13—H13	0.9300
C4—C7	1.5070 (19)	C14—H14A	0.9600
O4—C10	1.3633 (17)	C14—H14B	0.9600
O4—C14	1.4313 (17)	C14—H14C	0.9600
C5—C6	1.383 (2)	C15—H15A	0.9600
C5—H5	0.9300	C15—H15B	0.9600
O5—C12	1.3623 (17)	C15—H15C	0.9600
O5—C15	1.4366 (18)		
O2—S1—O1	120.88 (6)	C4—C7—H7A	109.5
O2—S1—O3	102.37 (6)	C4—C7—H7B	109.5
O1—S1—O3	108.43 (5)	H7A—C7—H7B	109.5
O2—S1—C1	110.89 (6)	C4—C7—H7C	109.5
O1—S1—C1	110.05 (6)	H7A—C7—H7C	109.5
O3—S1—C1	102.28 (6)	H7B—C7—H7C	109.5
O2—S1—O1 <sup>i</sup>	80.61 (5)	C8—C9—C10	117.91 (13)

O1—S1—O1 <sup>i</sup>	68.92 (5)	C8—C9—H9	121.0
O3—S1—O1 <sup>i</sup>	176.85 (4)	C10—C9—H9	121.0
C1—S1—O1 <sup>i</sup>	77.41 (5)	O4—C10—C11	124.24 (13)
C2—C1—C6	121.02 (13)	O4—C10—C9	115.06 (12)
C2—C1—S1	119.61 (11)	C11—C10—C9	120.70 (13)
C6—C1—S1	119.22 (11)	C10—C11—C12	119.46 (13)
C1—C2—C3	118.68 (13)	C10—C11—H11	120.3
C1—C2—H2	120.7	C12—C11—H11	120.3
C3—C2—H2	120.7	O5—C12—C13	124.31 (13)
C8—O3—S1	118.80 (8)	O5—C12—C11	114.44 (13)
C2—C3—C4	121.42 (13)	C13—C12—C11	121.24 (13)
C2—C3—H3	119.3	C8—C13—C12	116.89 (13)
C4—C3—H3	119.3	C8—C13—H13	121.6
C5—C4—C3	118.48 (13)	C12—C13—H13	121.6
C5—C4—C7	120.97 (13)	O4—C14—H14A	109.5
C3—C4—C7	120.55 (13)	O4—C14—H14B	109.5
C10—O4—C14	117.21 (11)	H14A—C14—H14B	109.5
C6—C5—C4	120.99 (13)	O4—C14—H14C	109.5
C6—C5—H5	119.5	H14A—C14—H14C	109.5
C4—C5—H5	119.5	H14B—C14—H14C	109.5
C12—O5—C15	117.31 (12)	O5—C15—H15A	109.5
C5—C6—C1	119.33 (13)	O5—C15—H15B	109.5
C5—C6—H6	120.3	H15A—C15—H15B	109.5
C1—C6—H6	120.3	O5—C15—H15C	109.5
C9—C8—C13	123.77 (13)	H15A—C15—H15C	109.5
C9—C8—O3	117.73 (12)	H15B—C15—H15C	109.5
C13—C8—O3	118.33 (12)		
O2—S1—C1—C2	-139.74 (11)	S1—C1—C6—C5	173.38 (11)
O1—S1—C1—C2	-3.35 (13)	S1—O3—C8—C9	98.87 (13)
O3—S1—C1—C2	111.74 (11)	S1—O3—C8—C13	-85.70 (13)
O1 <sup>i</sup> —S1—C1—C2	-65.06 (11)	C13—C8—C9—C10	-1.2 (2)
O2—S1—C1—C6	44.73 (13)	O3—C8—C9—C10	173.96 (11)
O1—S1—C1—C6	-178.88 (11)	C14—O4—C10—C11	-2.0 (2)
O3—S1—C1—C6	-63.80 (12)	C14—O4—C10—C9	176.90 (13)
O1 <sup>i</sup> —S1—C1—C6	119.40 (11)	C8—C9—C10—O4	-176.98 (12)
C6—C1—C2—C3	2.6 (2)	C8—C9—C10—C11	1.9 (2)
S1—C1—C2—C3	-172.83 (10)	O4—C10—C11—C12	177.91 (12)
O2—S1—O3—C8	177.23 (9)	C9—C10—C11—C12	-0.9 (2)
O1—S1—O3—C8	48.40 (11)	C15—O5—C12—C13	3.8 (2)
C1—S1—O3—C8	-67.86 (10)	C15—O5—C12—C11	-177.20 (12)
C1—C2—C3—C4	-0.7 (2)	C10—C11—C12—O5	179.96 (12)
C2—C3—C4—C5	-1.7 (2)	C10—C11—C12—C13	-1.0 (2)
C2—C3—C4—C7	178.63 (13)	C9—C8—C13—C12	-0.6 (2)
C3—C4—C5—C6	2.3 (2)	O3—C8—C13—C12	-175.70 (11)
C7—C4—C5—C6	-178.07 (13)	O5—C12—C13—C8	-179.37 (13)

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C4—C5—C6—C1	-0.4 (2)	C11—C12—C13—C8	1.7 (2)
C2—C1—C6—C5	-2.1 (2)		

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Symmetry code: (i)  $-x+2, -y+2, -z+1$ .

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

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<i>D—H<math>\cdots</math>A</i>	<i>D—H</i>	<i>H<math>\cdots</math>A</i>	<i>D<math>\cdots</math>A</i>	<i>D—H<math>\cdots</math>A</i>
C15—H15A $\cdots$ O4 <sup>ii</sup>	0.96	2.35	3.2051 (18)	147
C15—H15C $\cdots$ O2 <sup>iii</sup>	0.96	2.51	3.4063 (19)	155

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Symmetry codes: (ii)  $x, y+1, z$ ; (iii)  $-x+1, -y+2, -z+1$ .