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# (4Z)-4-[1-(2-Aminoanilino)ethylidene]-3-methyl-1-phenyl-4,5-dihydro-1H-pyrazol-5-one

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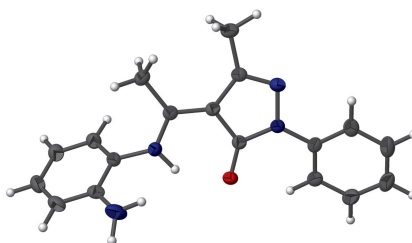
Keywords: crystal structure; hydrogen bond; pyrazolone;  $\pi$ -stacking.

CCDC reference: 1532587

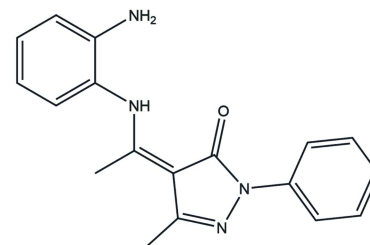
Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

The conformation of the title compound, C<sub>18</sub>H<sub>18</sub>N<sub>4</sub>O, is partly determined by an intramolecular N—H···O hydrogen bond that imposes planarity on the central aminoethylidene-3-methylpyrazol-5-one segment of the molecule. In the crystal, N—H···O hydrogen and N—H···N hydrogen bonds both form centrosymmetric dimers that enclose R<sub>2</sub><sup>2</sup>(18) rings. These, together with C—H···N and  $\pi$ – $\pi$  stacking interactions between centrosymmetrically related pyrazolone rings, stack the molecules along the *b*-axis direction.

## 3D view



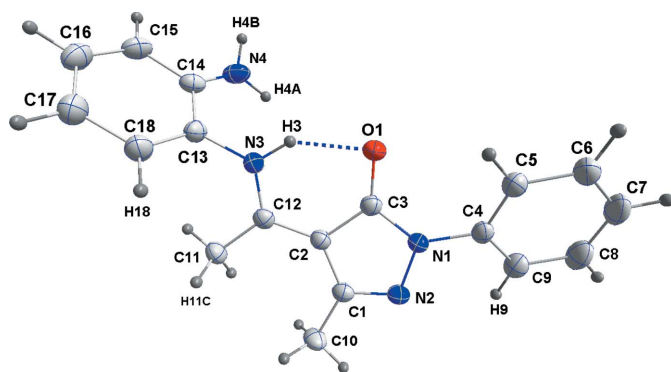
## Chemical scheme



## Structure description

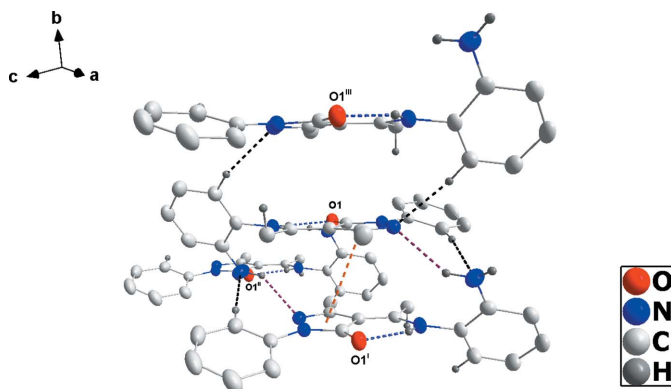
Pyrazole derivatives are known to possess a broad spectrum of biological activities, acting as cannabinoid receptor antagonists (Lan *et al.*, 1999), antibacterial (Tanitame *et al.*, 2004), anti-inflammatory and antimicrobial agents (Bekhit & Abdel-Aziem, 2004). In a continuation of our research using acetoacetyl pyrazole as starting material in the synthesis of several heterocyclic systems with potent pharmacological properties (Djerrari *et al.*, 2001,2003), we report in this work the crystal structure of the title pyrazolone derivative.

The conformation of the title molecule is partially determined by an intramolecular N3—H3···O1 hydrogen bond (Table 1, Figs. 1 and 2). This ensures the planarity of the central C13/N3/C12(C11)[N1,N2,C1(C10)C2,C3(O1)] section of the molecule, r.m.s. deviation 0.0403 Å. The dihedral angle between the mean planes of the N1/N2/C1–C3 and C4–C9 rings is 21.31 (4)° while that between the former ring and the C13–C18 ring is 62.60 (5)°.

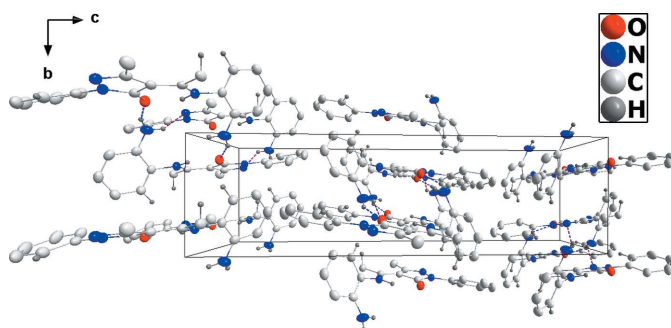


**Figure 1**  
The title molecule with the labelling scheme and 50% probability displacement ellipsoids. The intramolecular hydrogen bond is shown as a dashed line.

In the crystal, the molecules form centrosymmetric dimers through pairwise  $N4-H4B \cdots O1^{ii}$  hydrogen bonds, Table 1.  $N1-H4A \cdots N2^i$  hydrogen bonds also form inversion dimers with both sets of dimers forming  $R_2^2(18)$  ring motifs. The dimers stack along the  $b$ -axis direction aided by  $C9-H9 \cdots N4^i$ ,  $C11-H11C \cdots N1^{iii}$  and  $C18-H18 \cdots N2^{iii}$  hydrogen bonds and offset  $\pi$ - $\pi$ -stacking interactions [ $Cg1 \cdots Cg1^{iv} =$



**Figure 2**  
Detail of the intra- and intermolecular molecular interactions (dashed lines:  $N-H \cdots O$ , blue;  $N-H \cdots N$ , purple;  $C-H \cdots O$ ; black;  $\pi$ - $\pi$  stacking, orange). For symmetry codes, see Table 1.



**Figure 3**  
Crystal packing, viewed along the  $a$  axis (the colour codes for the hydrogen bonds are given in Fig. 2).

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N3-H3 \cdots O1$	0.955 (18)	1.863 (18)	2.6965 (14)	144.3 (15)
$N4-H4A \cdots N2^i$	0.92 (2)	2.47 (2)	3.2542 (17)	144.1 (16)
$N4-H4B \cdots O1^{ii}$	0.93 (2)	2.14 (2)	3.0090 (16)	153.3 (17)
$C9-H9 \cdots N4^i$	1.008 (17)	2.567 (17)	3.571 (2)	173.7 (13)
$C11-H11C \cdots N1^{iii}$	0.98 (2)	2.68 (2)	3.5669 (19)	152.0 (16)
$C18-H18 \cdots N2^{iii}$	0.997 (17)	2.439 (17)	3.3984 (18)	161.2 (13)

Symmetry codes: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x, -y+1, -z+1$ ; (iii)  $-x+1, -y+2, -z+1$ .

**Table 2**  
Experimental details.

Crystal data	
Chemical formula	$C_{18}H_{18}N_4O$
$M_r$	306.36
Crystal system, space group	Monoclinic, $P2_1/c$
Temperature (K)	150
$a, b, c$ ( $\text{\AA}$ )	8.6575 (2), 7.1749 (1), 25.3062 (5)
$\beta$ ( $^\circ$ )	93.782 (1)
$V$ ( $\text{\AA}^3$ )	1568.51 (5)
$Z$	4
Radiation type	$Cu K\alpha$
$\mu$ ( $\text{mm}^{-1}$ )	0.67
Crystal size (mm)	$0.19 \times 0.05 \times 0.05$
Data collection	
Diffractometer	Bruker D8 VENTURE PHOTON 100 CMOS
Absorption correction	Multi-scan ( <i>SADABS</i> ; Bruker, 2016)
$T_{min}, T_{max}$	0.87, 0.97
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	11756, 3032, 2596
$R_{int}$	0.035
$(\sin \theta/\lambda)_{max}$ ( $\text{\AA}^{-1}$ )	0.618
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.038, 0.096, 1.04
No. of reflections	3032
No. of parameters	281
H-atom treatment	All H-atom parameters refined
$\Delta\rho_{max}, \Delta\rho_{min}$ ( $e \text{\AA}^{-3}$ )	0.29, $-0.18$

Computer programs: *APEX3* and *SAINT* (Bruker, 2016), *SHELXT* (Sheldrick, 2015a), *SHELXL2014* (Sheldrick, 2015b), *DIAMOND* (Brandenburg & Putz, 2012) and *SHELXTL* (Sheldrick, 2008).

$3.758$  (1)  $\text{\AA}$ ; symmetry code: (iv)  $1-x, 1-y, 1-z$ ] between pairs of centrosymmetrically related pyrazolone rings (Table 1, Figs. 2 and 3). The individual stacks show only normal van der Waals contacts between them.

### Synthesis and crystallization

To a solution of acetoacetyl pyrazole (0.001 mol) in 40 ml of ethanol was added *o*-phenylenediamine (0.002 mol). The reaction mixture was refluxed for 2 h. After cooling, the solvent was removed under reduced pressure. The residue obtained was recrystallized from ethanol to afford colourless block-like crystals of the title compound (m.p. 485–487 K, yield: 80%).

## Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

## Acknowledgements

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## full crystallographic data

*IUCrData* (2017). 2, x170256 [https://doi.org/10.1107/S2414314617002565]

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**(4Z)-4-[1-(2-Aminoanilino)ethylidene]-3-methyl-1-phenyl-4,5-dihydro-1H-pyrazol-5-one**

*Crystal data*

$C_{18}H_{18}N_4O$

$M_r = 306.36$

Monoclinic,  $P2_1/c$

$a = 8.6575$  (2) Å

$b = 7.1749$  (1) Å

$c = 25.3062$  (5) Å

$\beta = 93.782$  (1)°

$V = 1568.51$  (5) Å<sup>3</sup>

$Z = 4$

$F(000) = 648$

$D_x = 1.297$  Mg m<sup>-3</sup>

Cu  $K\alpha$  radiation,  $\lambda = 1.54178$  Å

Cell parameters from 8108 reflections

$\theta = 5.1\text{--}72.4^\circ$

$\mu = 0.67$  mm<sup>-1</sup>

$T = 150$  K

Column, colourless

$0.19 \times 0.05 \times 0.05$  mm

*Data collection*

Bruker D8 VENTURE PHOTON 100 CMOS diffractometer

Radiation source: INCOATEC  $I\mu$ S micro-focus source

Mirror monochromator

Detector resolution: 10.4167 pixels mm<sup>-1</sup>

$\omega$  scans

Absorption correction: multi-scan (*SADABS*; Bruker, 2016)

$T_{\min} = 0.87$ ,  $T_{\max} = 0.97$

11756 measured reflections

3032 independent reflections

2596 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$

$\theta_{\max} = 72.4^\circ$ ,  $\theta_{\min} = 5.1^\circ$

$h = -10 \rightarrow 10$

$k = -8 \rightarrow 8$

$l = -31 \rightarrow 30$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.038$

$wR(F^2) = 0.096$

$S = 1.04$

3032 reflections

281 parameters

0 restraints

Hydrogen site location: difference Fourier map

All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0455P)^2 + 0.4468P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.29$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.18$  e Å<sup>-3</sup>

Extinction correction: *SHELXL2014* (Sheldrick, 2015b),  $F_c^* = kF_c[1 + 0.001 \times F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0050 (4)

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.23306 (10)	0.69400 (14)	0.45467 (4)	0.0305 (2)
N1	0.48083 (12)	0.71074 (15)	0.42251 (4)	0.0248 (2)
N2	0.63327 (12)	0.73552 (16)	0.44323 (4)	0.0270 (3)
N3	0.25320 (13)	0.75094 (16)	0.56022 (4)	0.0276 (3)
C1	0.62652 (14)	0.75905 (17)	0.49448 (5)	0.0248 (3)
C2	0.47020 (14)	0.74833 (17)	0.50980 (5)	0.0234 (3)
C3	0.37694 (14)	0.71531 (17)	0.46141 (5)	0.0241 (3)
H3	0.203 (2)	0.721 (2)	0.5265 (7)	0.044 (5)*
C4	0.45262 (15)	0.68737 (17)	0.36708 (5)	0.0258 (3)
H4A	0.132 (2)	0.447 (3)	0.5721 (8)	0.051 (5)*
H4B	-0.032 (2)	0.404 (3)	0.5913 (8)	0.057 (6)*
N4	0.05992 (15)	0.47138 (18)	0.59596 (6)	0.0405 (3)
C5	0.30760 (17)	0.7256 (2)	0.34286 (5)	0.0316 (3)
H5	0.221 (2)	0.771 (2)	0.3632 (7)	0.040 (4)*
C6	0.28126 (19)	0.6947 (2)	0.28880 (6)	0.0406 (4)
H6	0.179 (2)	0.722 (3)	0.2718 (7)	0.047 (5)*
C7	0.3984 (2)	0.6291 (3)	0.25907 (6)	0.0451 (4)
H7	0.379 (2)	0.604 (3)	0.2209 (8)	0.057 (5)*
C8	0.5442 (2)	0.5994 (2)	0.28307 (6)	0.0417 (4)
H8	0.631 (2)	0.552 (3)	0.2629 (7)	0.052 (5)*
C9	0.57317 (17)	0.6281 (2)	0.33724 (6)	0.0332 (3)
H9	0.680 (2)	0.607 (2)	0.3542 (6)	0.037 (4)*
C10	0.77427 (16)	0.7937 (2)	0.52661 (6)	0.0338 (3)
H10A	0.769 (2)	0.910 (3)	0.5479 (7)	0.045 (5)*
H10B	0.795 (2)	0.693 (3)	0.5525 (8)	0.054 (5)*
H10C	0.862 (2)	0.799 (3)	0.5015 (8)	0.049 (5)*
C11	0.50207 (16)	0.8055 (2)	0.60909 (5)	0.0294 (3)
H11A	0.450 (2)	0.766 (3)	0.6404 (8)	0.056 (5)*
H11B	0.601 (2)	0.738 (3)	0.6094 (8)	0.054 (5)*
H11C	0.527 (2)	0.938 (3)	0.6128 (8)	0.063 (6)*
C12	0.40601 (15)	0.76834 (17)	0.55887 (5)	0.0242 (3)
C13	0.16041 (15)	0.78645 (19)	0.60397 (5)	0.0266 (3)
C14	0.05963 (14)	0.64562 (19)	0.61874 (5)	0.0284 (3)
C15	-0.04064 (16)	0.6879 (2)	0.65830 (6)	0.0341 (3)
H15	-0.113 (2)	0.588 (3)	0.6692 (7)	0.045 (5)*

C16	−0.04014 (16)	0.8615 (2)	0.68174 (6)	0.0366 (3)
H16	−0.112 (2)	0.889 (3)	0.7101 (7)	0.046 (5)*
C17	0.06066 (17)	0.9995 (2)	0.66678 (6)	0.0367 (3)
H17	0.062 (2)	1.130 (3)	0.6819 (7)	0.050 (5)*
C18	0.15980 (16)	0.9615 (2)	0.62736 (5)	0.0325 (3)
H18	0.2297 (18)	1.060 (2)	0.6150 (6)	0.037 (4)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0219 (4)	0.0418 (6)	0.0276 (5)	−0.0053 (4)	0.0004 (3)	−0.0017 (4)
N1	0.0222 (5)	0.0283 (6)	0.0236 (5)	−0.0018 (4)	0.0002 (4)	0.0002 (4)
N2	0.0223 (5)	0.0285 (6)	0.0301 (6)	−0.0015 (4)	0.0006 (4)	0.0002 (5)
N3	0.0266 (6)	0.0329 (6)	0.0232 (5)	−0.0051 (5)	0.0022 (4)	−0.0021 (5)
C1	0.0243 (6)	0.0210 (6)	0.0289 (6)	−0.0004 (5)	−0.0003 (5)	0.0008 (5)
C2	0.0244 (6)	0.0215 (6)	0.0243 (6)	−0.0010 (5)	−0.0002 (5)	0.0003 (5)
C3	0.0244 (6)	0.0226 (6)	0.0253 (6)	−0.0015 (5)	0.0016 (5)	0.0010 (5)
C4	0.0325 (7)	0.0213 (6)	0.0235 (6)	−0.0048 (5)	0.0025 (5)	0.0009 (5)
N4	0.0266 (6)	0.0327 (7)	0.0626 (9)	−0.0049 (5)	0.0061 (6)	−0.0072 (6)
C5	0.0326 (7)	0.0354 (8)	0.0267 (6)	−0.0071 (6)	0.0006 (5)	0.0046 (6)
C6	0.0432 (8)	0.0506 (9)	0.0274 (7)	−0.0149 (7)	−0.0036 (6)	0.0073 (7)
C7	0.0627 (10)	0.0494 (10)	0.0233 (7)	−0.0182 (8)	0.0034 (7)	0.0001 (7)
C8	0.0562 (10)	0.0385 (9)	0.0321 (7)	−0.0047 (7)	0.0146 (7)	−0.0037 (7)
C9	0.0386 (8)	0.0292 (7)	0.0323 (7)	0.0000 (6)	0.0067 (6)	−0.0004 (6)
C10	0.0249 (7)	0.0394 (8)	0.0363 (7)	−0.0009 (6)	−0.0041 (6)	−0.0039 (7)
C11	0.0323 (7)	0.0300 (7)	0.0254 (6)	−0.0018 (6)	−0.0021 (5)	−0.0013 (6)
C12	0.0275 (6)	0.0193 (6)	0.0257 (6)	−0.0019 (5)	0.0010 (5)	0.0017 (5)
C13	0.0243 (6)	0.0328 (7)	0.0229 (6)	−0.0019 (5)	0.0014 (5)	0.0021 (5)
C14	0.0218 (6)	0.0303 (7)	0.0324 (7)	−0.0008 (5)	−0.0032 (5)	0.0045 (6)
C15	0.0246 (6)	0.0406 (8)	0.0375 (7)	−0.0016 (6)	0.0038 (5)	0.0093 (6)
C16	0.0305 (7)	0.0493 (9)	0.0309 (7)	0.0033 (6)	0.0075 (6)	0.0031 (7)
C17	0.0374 (8)	0.0386 (8)	0.0345 (7)	0.0002 (6)	0.0060 (6)	−0.0040 (7)
C18	0.0337 (7)	0.0328 (7)	0.0316 (7)	−0.0057 (6)	0.0061 (6)	−0.0009 (6)

*Geometric parameters (Å, °)*

O1—C3	1.2554 (15)	C7—H7	0.985 (19)
N1—C3	1.3769 (16)	C8—C9	1.392 (2)
N1—N2	1.3990 (15)	C8—H8	0.999 (19)
N1—C4	1.4178 (16)	C9—H9	1.008 (17)
N2—C1	1.3132 (17)	C10—H10A	0.99 (2)
N3—C12	1.3316 (17)	C10—H10B	0.98 (2)
N3—C13	1.4330 (16)	C10—H10C	1.021 (19)
N3—H3	0.955 (18)	C11—C12	1.4962 (18)
C1—C2	1.4342 (17)	C11—H11A	0.98 (2)
C1—C10	1.4906 (18)	C11—H11B	0.99 (2)
C2—C12	1.4007 (17)	C11—H11C	0.98 (2)
C2—C3	1.4413 (17)	C13—C18	1.388 (2)

C4—C5	1.3878 (19)	C13—C14	1.4017 (18)
C4—C9	1.3945 (19)	C14—C15	1.4007 (19)
N4—C14	1.3768 (19)	C15—C16	1.380 (2)
N4—H4A	0.92 (2)	C15—H15	1.000 (18)
N4—H4B	0.93 (2)	C16—C17	1.389 (2)
C5—C6	1.390 (2)	C16—H16	1.002 (18)
C5—H5	0.991 (17)	C17—C18	1.385 (2)
C6—C7	1.385 (2)	C17—H17	1.01 (2)
C6—H6	0.976 (19)	C18—H18	0.997 (17)
C7—C8	1.380 (2)		
C3—N1—N2	111.94 (10)	C8—C9—H9	119.9 (9)
C3—N1—C4	129.21 (11)	C4—C9—H9	121.1 (9)
N2—N1—C4	118.84 (10)	C1—C10—H10A	111.7 (10)
C1—N2—N1	106.51 (10)	C1—C10—H10B	110.7 (12)
C12—N3—C13	127.72 (11)	H10A—C10—H10B	105.5 (15)
C12—N3—H3	112.9 (11)	C1—C10—H10C	108.2 (11)
C13—N3—H3	119.2 (11)	H10A—C10—H10C	111.5 (14)
N2—C1—C2	111.37 (11)	H10B—C10—H10C	109.2 (15)
N2—C1—C10	117.88 (12)	C12—C11—H11A	112.0 (12)
C2—C1—C10	130.74 (12)	C12—C11—H11B	110.7 (11)
C12—C2—C1	132.12 (12)	H11A—C11—H11B	107.4 (16)
C12—C2—C3	122.50 (11)	C12—C11—H11C	111.3 (12)
C1—C2—C3	105.35 (11)	H11A—C11—H11C	108.4 (16)
O1—C3—N1	126.09 (11)	H11B—C11—H11C	106.8 (16)
O1—C3—C2	129.10 (11)	N3—C12—C2	117.79 (11)
N1—C3—C2	104.81 (10)	N3—C12—C11	119.49 (11)
C5—C4—C9	120.57 (13)	C2—C12—C11	122.72 (12)
C5—C4—N1	120.36 (12)	C18—C13—C14	121.23 (12)
C9—C4—N1	119.06 (12)	C18—C13—N3	120.55 (12)
C14—N4—H4A	118.0 (13)	C14—C13—N3	117.93 (12)
C14—N4—H4B	120.0 (12)	N4—C14—C15	120.93 (13)
H4A—N4—H4B	115.7 (17)	N4—C14—C13	121.64 (13)
C4—C5—C6	119.22 (14)	C15—C14—C13	117.42 (13)
C4—C5—H5	121.8 (10)	C16—C15—C14	121.21 (13)
C6—C5—H5	118.9 (10)	C16—C15—H15	120.8 (10)
C7—C6—C5	120.70 (15)	C14—C15—H15	118.0 (10)
C7—C6—H6	120.2 (11)	C15—C16—C17	120.71 (13)
C5—C6—H6	119.1 (11)	C15—C16—H16	120.0 (11)
C8—C7—C6	119.64 (14)	C17—C16—H16	119.3 (11)
C8—C7—H7	119.9 (11)	C18—C17—C16	119.10 (14)
C6—C7—H7	120.5 (11)	C18—C17—H17	117.6 (10)
C7—C8—C9	120.70 (14)	C16—C17—H17	123.2 (10)
C7—C8—H8	121.8 (11)	C17—C18—C13	120.32 (13)
C9—C8—H8	117.5 (11)	C17—C18—H18	120.3 (9)
C8—C9—C4	119.05 (14)	C13—C18—H18	119.4 (9)
C3—N1—N2—C1	-1.13 (14)	C6—C7—C8—C9	2.4 (3)

C4—N1—N2—C1	178.18 (11)	C7—C8—C9—C4	0.1 (2)
N1—N2—C1—C2	0.78 (14)	C5—C4—C9—C8	-3.0 (2)
N1—N2—C1—C10	-178.48 (12)	N1—C4—C9—C8	177.95 (13)
N2—C1—C2—C12	-178.39 (13)	C13—N3—C12—C2	-172.18 (12)
C10—C1—C2—C12	0.7 (2)	C13—N3—C12—C11	8.6 (2)
N2—C1—C2—C3	-0.19 (14)	C1—C2—C12—N3	179.71 (13)
C10—C1—C2—C3	178.94 (14)	C3—C2—C12—N3	1.77 (18)
N2—N1—C3—O1	-178.75 (12)	C1—C2—C12—C11	-1.1 (2)
C4—N1—C3—O1	2.0 (2)	C3—C2—C12—C11	-179.04 (12)
N2—N1—C3—C2	0.99 (14)	C12—N3—C13—C18	58.78 (19)
C4—N1—C3—C2	-178.23 (12)	C12—N3—C13—C14	-127.44 (14)
C12—C2—C3—O1	-2.3 (2)	C18—C13—C14—N4	-179.68 (13)
C1—C2—C3—O1	179.25 (13)	N3—C13—C14—N4	6.59 (19)
C12—C2—C3—N1	177.93 (11)	C18—C13—C14—C15	-0.54 (19)
C1—C2—C3—N1	-0.48 (13)	N3—C13—C14—C15	-174.28 (12)
C3—N1—C4—C5	21.3 (2)	N4—C14—C15—C16	179.20 (13)
N2—N1—C4—C5	-157.89 (12)	C13—C14—C15—C16	0.1 (2)
C3—N1—C4—C9	-159.63 (13)	C14—C15—C16—C17	-0.2 (2)
N2—N1—C4—C9	21.19 (17)	C15—C16—C17—C18	0.8 (2)
C9—C4—C5—C6	3.4 (2)	C16—C17—C18—C13	-1.3 (2)
N1—C4—C5—C6	-177.52 (13)	C14—C13—C18—C17	1.2 (2)
C4—C5—C6—C7	-1.0 (2)	N3—C13—C18—C17	174.77 (13)
C5—C6—C7—C8	-1.9 (3)		

## Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N3—H3...O1	0.955 (18)	1.863 (18)	2.6965 (14)	144.3 (15)
N4—H4 <i>A</i> ...N2 <sup>i</sup>	0.92 (2)	2.47 (2)	3.2542 (17)	144.1 (16)
N4—H4 <i>B</i> ...O1 <sup>ii</sup>	0.93 (2)	2.14 (2)	3.0090 (16)	153.3 (17)
C9—H9...N4 <sup>i</sup>	1.008 (17)	2.567 (17)	3.571 (2)	173.7 (13)
C11—H11 <i>C</i> ...N1 <sup>iii</sup>	0.98 (2)	2.68 (2)	3.5669 (19)	152.0 (16)
C18—H18...N2 <sup>iii</sup>	0.997 (17)	2.439 (17)	3.3984 (18)	161.2 (13)

Symmetry codes: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x, -y+1, -z+1$ ; (iii)  $-x+1, -y+2, -z+1$ .