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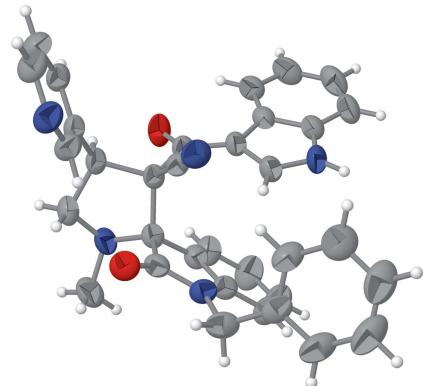
1-Benzyl-3'-(1*H*-indol-3-yl)carbonyl]-1'-methyl-2-oxo-4'-(pyridin-3-yl)spiro[indoline-3,2'-pyrrolidine]-3'-carbonitrile

P. Seethalakshmi^a and C. Palanivel^{b,a*}

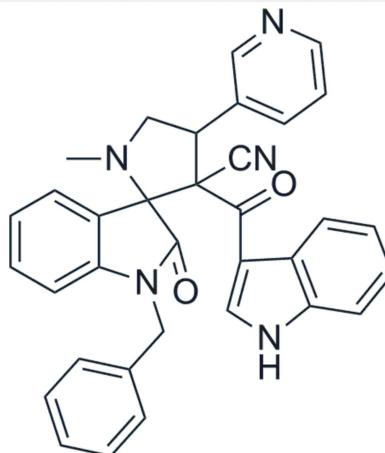
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In the title compound, $C_{34}H_{27}N_5O_2$, the central pyrrolidine ring adopts an envelope conformation, with the N atom as the flap. The mean planes of the two indoline ring systems are inclined to the mean plane of the central pyrrolidine ring by 86.26 (9) and 69.30 (9) $^\circ$, respectively. The dihedral angle between the benzene and pyridine rings is 75.09 (11) $^\circ$. In the crystal, molecules are linked by N—H···N and C—H···N hydrogen bonds, forming sheets parallel to the *ab* plane.

3D view



Chemical scheme



Structure description

Spiro-pyrrolidine derivatives are unique tetracyclic 5-HT(2A) receptor antagonists (Obniska *et al.*, 2003; Peddi *et al.*, 2004). These derivatives possess anticancer (Zapf *et al.*, 2011) and anti-influenza virus (Stylianakis *et al.*, 2003) activities. Highly functionalized pyrrolidines have attracted much interest in recent years as they constitute the main structural element of many natural and synthetic pharmacologically active compounds (Waldmann, 1995). Optically active pyrrolidines have been used as intermediates, chiral ligands or auxiliaries in controlled asymmetric synthesis (Suzuki *et al.*, 1994; Huryn *et al.*, 1991). In view of their importance and in continuation of our work on the crystal structure analysis of spiro-pyrrolidine derivatives, we report herein on the synthesis and crystal structure of a new spiro-pyrrolidine derivative.

The molecular structure of the title compound is illustrated in Fig. 1. The five-membered ring (N2/C6–C9) in the pyrrolidine moiety adopts an envelope conformation, with atom N2 as the flap atom [puckering parameters: $q_2 = 0.410$ (2) Å and $\varphi_2 = 181.7$ (3) $^\circ$], and the pyridine ring (N1/C1–C5) exhibits a slightly twisted conformation [puckering parameters: $q_2 = 0.093$ (2) Å and $\Phi_2 = 49.9$ (2) $^\circ$]. The sum of angles at atom

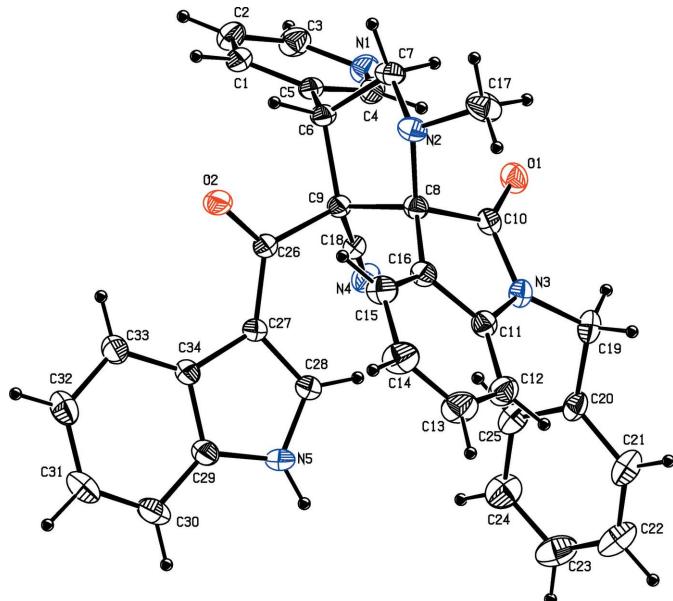


Figure 1

The molecular structure of the title compound, showing the atom labelling and 20% probability displacement ellipsoids.

N2 [328.4 (15) $^\circ$] is in accordance with sp^2 -hybridization, and the sum of angles at atom N3 [313.6 (14) $^\circ$] is in accordance with sp^3 -hybridization. The dihedral angle between the benzene (C20–C25) and pyridine (N1/C1–C5) rings is 75.09 (11) $^\circ$.

In the crystal, molecules are linked by N–H \cdots N and C–H \cdots N hydrogen bonds, forming sheets parallel to the *ab* plane (Table 1 and Fig. 2).

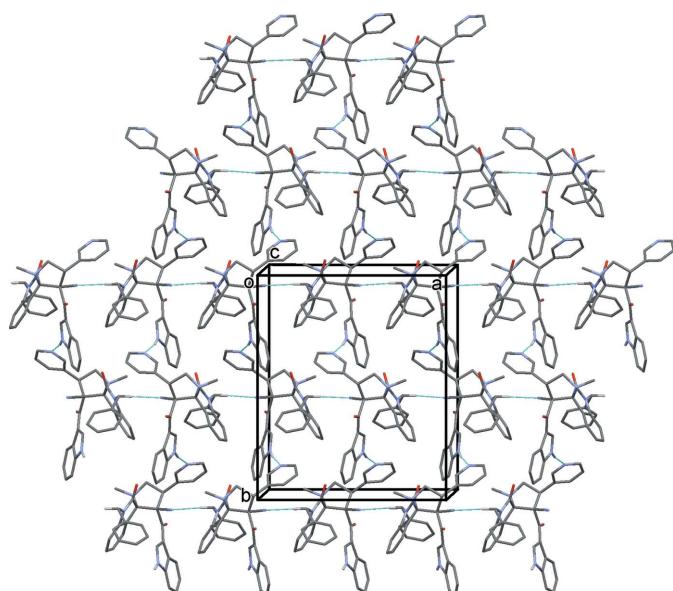


Figure 2

The crystal packing of the title compound, viewed along the *c* axis. Hydrogen bonds are shown as dashed lines (see Table 1) and, for clarity, only H atoms H5 and H19A have been included.

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{--H}\cdots A$	$D\text{--H}$	$H\cdots A$	$D\cdots A$	$D\text{--H}\cdots A$
N5–H5 \cdots N1 ⁱ	0.86	1.96	2.812 (2)	170
C19–H19A \cdots N4 ⁱⁱ	0.97	2.46	3.370 (3)	157

Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x - \frac{1}{2}, y, -z + \frac{1}{2}$.

Table 2
Experimental details.

Crystal data	
Chemical formula	$\text{C}_{34}\text{H}_{27}\text{N}_5\text{O}_2$
M_r	537.60
Crystal system, space group	Orthorhombic, <i>Pbca</i>
Temperature (K)	296
a, b, c (\AA)	14.7421 (6), 17.5573 (6), 21.4675 (9)
V (\AA^3)	5556.5 (4)
Z	8
Radiation type	Mo $K\alpha$
μ (mm^{-1})	0.08
Crystal size (mm)	0.25 \times 0.16 \times 0.12
Data collection	
Diffractometer	Bruker APEXII CCD
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	22838, 4884, 3251
R_{int}	0.031
($\sin \theta/\lambda$) _{max} (\AA^{-1})	0.595
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.040, 0.095, 0.97
No. of reflections	4884
No. of parameters	372
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ ($e \text{\AA}^{-3}$)	0.13, -0.17

Computer programs: APEX2 and SAINT (Bruker, 2014), SHELXS97 and SHELXTL (Sheldrick, 2008), SHELXL2014 (Sheldrick, 2015), Mercury (Macrae *et al.*, 2008) and PLATON (Spek, 2009).

Synthesis and crystallization

N-Benzylisatin, (1) (0.3 mmol), was mixed with sarcosine, (2) (0.3 mmol), and (*E*)-2-[*(1H*-indol-3-yl)carbonyl]-3-(pyridin-3-yl)acrylonitrile in ethanol (10 ml) in a round-bottomed flask. The reaction mixture was heated at 358 K for 3 h. After cooling to ambient temperature, the reaction mixture was filtered to afford the pure title product as a white solid (yield 92%). The filtrate was left to evaporate slowly and after 48 h yellow crystals of the title compound were obtained.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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full crystallographic data

IUCrData (2016). **1**, x161686 [https://doi.org/10.1107/S2414314616016862]

1-Benzyl-3'-(1*H*-indol-3-yl)carbonyl]-1'-methyl-2-oxo-4'-(pyridin-3-yl)spiro-[indoline-3,2'-pyrrolidine]-3'-carbonitrile

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Crystal data

$C_{34}H_{27}N_5O_2$
 $M_r = 537.60$
Orthorhombic, $Pbca$
 $a = 14.7421$ (6) Å
 $b = 17.5573$ (6) Å
 $c = 21.4675$ (9) Å
 $V = 5556.5$ (4) Å³
 $Z = 8$
 $F(000) = 2256$

$D_x = 1.285 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 5384 reflections
 $\theta = 2.3\text{--}24.1^\circ$
 $\mu = 0.08 \text{ mm}^{-1}$
 $T = 296 \text{ K}$
Rectangular, yellow
0.25 × 0.16 × 0.12 mm

Data collection

Bruker APEXII CCD
diffractometer
 φ and ω scans
22838 measured reflections
4884 independent reflections
3251 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.031$
 $\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 1.9^\circ$
 $h = -14 \rightarrow 17$
 $k = -20 \rightarrow 20$
 $l = -18 \rightarrow 25$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.095$
 $S = 0.97$
4884 reflections
372 parameters
0 restraints
Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0237P)^2 + 2.8668P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.13 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.17 \text{ e } \text{\AA}^{-3}$
Extinction correction: SHELXL2014 (Sheldrick
2015), $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Extinction coefficient: 0.00063 (9)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.63234 (14)	-0.06447 (11)	0.42023 (9)	0.0517 (5)
H1	0.6404	-0.0377	0.4572	0.062*
C2	0.70167 (15)	-0.10825 (13)	0.39656 (11)	0.0627 (6)
H2	0.7568	-0.1116	0.4174	0.075*
C3	0.68837 (16)	-0.14676 (12)	0.34183 (11)	0.0644 (6)
H3	0.7358	-0.1758	0.3259	0.077*
C4	0.54388 (15)	-0.10303 (11)	0.33427 (10)	0.0524 (5)
H4	0.4889	-0.1024	0.3131	0.063*
C5	0.55064 (13)	-0.06054 (10)	0.38871 (9)	0.0438 (5)
C6	0.47250 (13)	-0.01328 (10)	0.41220 (9)	0.0445 (5)
H6	0.4907	0.0063	0.4531	0.053*
C7	0.38438 (14)	-0.05742 (11)	0.42225 (10)	0.0537 (5)
H7A	0.3763	-0.0960	0.3904	0.064*
H7B	0.3837	-0.0817	0.4628	0.064*
C8	0.33670 (12)	0.04828 (10)	0.36441 (9)	0.0419 (5)
C9	0.44388 (13)	0.05810 (10)	0.37192 (8)	0.0405 (4)
C10	0.31830 (13)	0.01138 (11)	0.29978 (10)	0.0458 (5)
C11	0.24722 (13)	0.12705 (11)	0.29981 (9)	0.0470 (5)
C12	0.19527 (15)	0.18821 (12)	0.28185 (11)	0.0637 (6)
H12	0.1732	0.1926	0.2414	0.076*
C13	0.17724 (17)	0.24293 (13)	0.32648 (13)	0.0788 (8)
H13	0.1433	0.2855	0.3155	0.095*
C14	0.20803 (18)	0.23604 (13)	0.38641 (12)	0.0757 (7)
H14	0.1939	0.2734	0.4155	0.091*
C15	0.26003 (15)	0.17389 (11)	0.40409 (10)	0.0593 (6)
H15	0.2804	0.1689	0.4449	0.071*
C16	0.28089 (13)	0.11976 (10)	0.35996 (9)	0.0455 (5)
C17	0.22168 (15)	-0.02969 (13)	0.41757 (12)	0.0748 (7)
H17A	0.1792	0.0112	0.4119	0.112*
H17B	0.2097	-0.0547	0.4565	0.112*
H17C	0.2155	-0.0656	0.3841	0.112*
C18	0.48887 (13)	0.05771 (10)	0.31095 (10)	0.0426 (5)
C19	0.24710 (15)	0.04824 (13)	0.19947 (10)	0.0622 (6)
H19A	0.1814	0.0489	0.1969	0.075*
H19B	0.2673	-0.0025	0.1883	0.075*
C20	0.28439 (15)	0.10381 (12)	0.15286 (9)	0.0522 (5)
C21	0.22805 (17)	0.13774 (15)	0.10973 (11)	0.0762 (7)
H21	0.1660	0.1283	0.1113	0.091*
C22	0.2618 (2)	0.18511 (17)	0.06460 (13)	0.0908 (9)
H22	0.2226	0.2070	0.0357	0.109*
C23	0.3517 (2)	0.20027 (16)	0.06171 (13)	0.0888 (9)
H23	0.3743	0.2321	0.0308	0.107*
C24	0.40917 (18)	0.16848 (16)	0.10455 (12)	0.0805 (8)
H24	0.4708	0.1797	0.1033	0.097*
C25	0.37569 (15)	0.11979 (13)	0.14957 (10)	0.0620 (6)

H25	0.4153	0.0975	0.1780	0.074*
C26	0.47020 (13)	0.13396 (10)	0.40690 (9)	0.0453 (5)
C27	0.47548 (13)	0.20507 (10)	0.37413 (9)	0.0455 (5)
C28	0.44590 (14)	0.22198 (11)	0.31459 (10)	0.0528 (5)
H28	0.4239	0.1863	0.2863	0.063*
C29	0.48880 (14)	0.33263 (11)	0.35534 (10)	0.0531 (5)
C30	0.50841 (16)	0.40925 (12)	0.36446 (13)	0.0700 (7)
H30	0.4986	0.4452	0.3334	0.084*
C31	0.54281 (18)	0.42924 (13)	0.42120 (14)	0.0776 (8)
H31	0.5563	0.4801	0.4291	0.093*
C32	0.55800 (16)	0.37571 (13)	0.46725 (12)	0.0710 (7)
H32	0.5813	0.3914	0.5054	0.085*
C33	0.53938 (14)	0.29933 (12)	0.45804 (10)	0.0565 (6)
H33	0.5506	0.2637	0.4892	0.068*
C34	0.50338 (13)	0.27721 (10)	0.40092 (9)	0.0465 (5)
N1	0.61087 (13)	-0.14454 (9)	0.31048 (8)	0.0588 (5)
N2	0.31418 (11)	0.00086 (9)	0.41796 (8)	0.0509 (4)
N3	0.27357 (10)	0.06328 (9)	0.26387 (7)	0.0477 (4)
N4	0.52502 (12)	0.05610 (10)	0.26381 (8)	0.0572 (5)
N5	0.45319 (12)	0.29669 (9)	0.30315 (8)	0.0587 (5)
H5	0.4382	0.3189	0.2690	0.070*
O1	0.34090 (10)	-0.05210 (7)	0.28292 (7)	0.0599 (4)
O2	0.48569 (11)	0.12999 (7)	0.46262 (7)	0.0646 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0643 (14)	0.0493 (12)	0.0416 (12)	-0.0037 (11)	-0.0093 (11)	0.0065 (10)
C2	0.0557 (14)	0.0682 (14)	0.0643 (16)	0.0042 (12)	-0.0136 (12)	0.0109 (13)
C3	0.0679 (16)	0.0581 (13)	0.0671 (16)	0.0158 (12)	-0.0047 (13)	0.0013 (13)
C4	0.0580 (13)	0.0468 (11)	0.0525 (14)	0.0043 (10)	-0.0129 (11)	-0.0027 (11)
C5	0.0566 (13)	0.0371 (10)	0.0377 (11)	-0.0020 (9)	-0.0059 (10)	0.0053 (9)
C6	0.0592 (13)	0.0388 (10)	0.0354 (11)	-0.0025 (9)	-0.0034 (10)	0.0058 (9)
C7	0.0684 (14)	0.0415 (11)	0.0512 (13)	-0.0009 (10)	0.0045 (11)	0.0087 (10)
C8	0.0483 (11)	0.0384 (10)	0.0388 (11)	-0.0013 (8)	0.0022 (9)	-0.0019 (9)
C9	0.0496 (11)	0.0392 (10)	0.0326 (11)	-0.0036 (9)	-0.0018 (9)	0.0026 (9)
C10	0.0423 (11)	0.0433 (11)	0.0516 (13)	-0.0045 (9)	-0.0008 (10)	-0.0034 (10)
C11	0.0439 (11)	0.0456 (11)	0.0515 (13)	0.0002 (9)	0.0001 (10)	-0.0020 (10)
C12	0.0667 (15)	0.0608 (13)	0.0638 (15)	0.0106 (12)	-0.0107 (12)	-0.0001 (12)
C13	0.0919 (19)	0.0590 (14)	0.086 (2)	0.0292 (14)	-0.0058 (16)	-0.0034 (15)
C14	0.100 (2)	0.0584 (14)	0.0689 (18)	0.0263 (14)	0.0033 (15)	-0.0124 (13)
C15	0.0755 (15)	0.0527 (12)	0.0495 (13)	0.0070 (11)	0.0058 (12)	-0.0051 (11)
C16	0.0474 (12)	0.0430 (10)	0.0460 (12)	-0.0008 (9)	0.0037 (10)	-0.0013 (10)
C17	0.0643 (16)	0.0686 (15)	0.0914 (19)	-0.0096 (12)	0.0192 (14)	0.0141 (14)
C18	0.0445 (11)	0.0431 (11)	0.0401 (12)	-0.0027 (9)	-0.0077 (10)	0.0038 (10)
C19	0.0595 (14)	0.0734 (15)	0.0535 (14)	-0.0075 (12)	-0.0152 (12)	-0.0114 (12)
C20	0.0577 (14)	0.0597 (12)	0.0393 (12)	0.0074 (11)	-0.0110 (11)	-0.0101 (11)
C21	0.0650 (16)	0.105 (2)	0.0589 (16)	0.0134 (15)	-0.0171 (14)	-0.0009 (16)

C22	0.099 (2)	0.113 (2)	0.0605 (18)	0.0353 (19)	-0.0144 (17)	0.0123 (17)
C23	0.112 (3)	0.095 (2)	0.0593 (18)	0.0244 (19)	0.0108 (17)	0.0166 (15)
C24	0.0728 (17)	0.103 (2)	0.0654 (17)	0.0064 (15)	0.0134 (15)	0.0091 (16)
C25	0.0571 (15)	0.0769 (15)	0.0520 (14)	0.0146 (12)	-0.0048 (12)	0.0003 (13)
C26	0.0529 (12)	0.0434 (11)	0.0397 (12)	-0.0050 (9)	-0.0050 (10)	0.0033 (10)
C27	0.0533 (13)	0.0407 (10)	0.0426 (12)	-0.0071 (9)	-0.0053 (10)	0.0067 (10)
C28	0.0608 (13)	0.0461 (11)	0.0516 (14)	-0.0110 (10)	-0.0063 (11)	0.0084 (10)
C29	0.0521 (13)	0.0463 (11)	0.0611 (14)	-0.0071 (10)	0.0001 (11)	0.0064 (11)
C30	0.0755 (17)	0.0464 (13)	0.0882 (19)	-0.0106 (11)	0.0065 (15)	0.0124 (13)
C31	0.0872 (19)	0.0482 (13)	0.097 (2)	-0.0185 (13)	0.0103 (17)	-0.0056 (15)
C32	0.0763 (17)	0.0647 (15)	0.0719 (17)	-0.0191 (13)	0.0038 (14)	-0.0151 (14)
C33	0.0570 (13)	0.0539 (12)	0.0585 (14)	-0.0078 (10)	-0.0008 (11)	-0.0026 (11)
C34	0.0448 (11)	0.0435 (11)	0.0512 (13)	-0.0059 (9)	-0.0001 (10)	0.0024 (10)
N1	0.0668 (12)	0.0492 (10)	0.0603 (12)	0.0118 (9)	-0.0086 (10)	-0.0058 (9)
N2	0.0547 (11)	0.0461 (9)	0.0518 (11)	-0.0049 (8)	0.0098 (9)	0.0080 (8)
N3	0.0474 (10)	0.0515 (9)	0.0440 (10)	-0.0014 (8)	-0.0062 (8)	-0.0062 (8)
N4	0.0557 (11)	0.0711 (12)	0.0447 (11)	0.0006 (9)	0.0005 (9)	0.0045 (10)
N5	0.0698 (12)	0.0487 (10)	0.0575 (12)	-0.0070 (9)	-0.0063 (10)	0.0211 (9)
O1	0.0665 (10)	0.0466 (8)	0.0666 (10)	0.0015 (7)	-0.0079 (8)	-0.0144 (7)
O2	0.1075 (13)	0.0464 (8)	0.0399 (9)	-0.0101 (8)	-0.0152 (9)	0.0022 (7)

Geometric parameters (\AA , ^\circ)

C1—C2	1.376 (3)	C17—H17A	0.9600
C1—C5	1.383 (3)	C17—H17B	0.9600
C1—H1	0.9300	C17—H17C	0.9600
C2—C3	1.370 (3)	C18—N4	1.144 (2)
C2—H2	0.9300	C19—N3	1.460 (2)
C3—N1	1.327 (3)	C19—C20	1.502 (3)
C3—H3	0.9300	C19—H19A	0.9700
C4—N1	1.329 (2)	C19—H19B	0.9700
C4—C5	1.390 (3)	C20—C25	1.377 (3)
C4—H4	0.9300	C20—C21	1.379 (3)
C5—C6	1.507 (3)	C21—C22	1.370 (4)
C6—C7	1.528 (3)	C21—H21	0.9300
C6—C9	1.580 (2)	C22—C23	1.353 (4)
C6—H6	0.9800	C22—H22	0.9300
C7—N2	1.458 (2)	C23—C24	1.369 (3)
C7—H7A	0.9700	C23—H23	0.9300
C7—H7B	0.9700	C24—C25	1.381 (3)
C8—N2	1.458 (2)	C24—H24	0.9300
C8—C16	1.504 (2)	C25—H25	0.9300
C8—C10	1.555 (3)	C26—O2	1.220 (2)
C8—C9	1.598 (3)	C26—C27	1.435 (2)
C9—C18	1.467 (3)	C27—C28	1.383 (3)
C9—C26	1.578 (2)	C27—C34	1.451 (2)
C10—O1	1.218 (2)	C28—N5	1.339 (2)
C10—N3	1.364 (2)	C28—H28	0.9300

C11—C12	1.374 (3)	C29—C30	1.390 (3)
C11—C16	1.389 (3)	C29—N5	1.389 (3)
C11—N3	1.414 (2)	C29—C34	1.397 (3)
C12—C13	1.383 (3)	C30—C31	1.365 (3)
C12—H12	0.9300	C30—H30	0.9300
C13—C14	1.370 (3)	C31—C32	1.382 (3)
C13—H13	0.9300	C31—H31	0.9300
C14—C15	1.387 (3)	C32—C33	1.383 (3)
C14—H14	0.9300	C32—H32	0.9300
C15—C16	1.377 (3)	C33—C34	1.391 (3)
C15—H15	0.9300	C33—H33	0.9300
C17—N2	1.465 (3)	N5—H5	0.8600
C2—C1—C5	119.6 (2)	H17A—C17—H17C	109.5
C2—C1—H1	120.2	H17B—C17—H17C	109.5
C5—C1—H1	120.2	N4—C18—C9	178.5 (2)
C3—C2—C1	119.1 (2)	N3—C19—C20	114.55 (17)
C3—C2—H2	120.5	N3—C19—H19A	108.6
C1—C2—H2	120.5	C20—C19—H19A	108.6
N1—C3—C2	123.0 (2)	N3—C19—H19B	108.6
N1—C3—H3	118.5	C20—C19—H19B	108.6
C2—C3—H3	118.5	H19A—C19—H19B	107.6
N1—C4—C5	124.34 (19)	C25—C20—C21	117.8 (2)
N1—C4—H4	117.8	C25—C20—C19	121.6 (2)
C5—C4—H4	117.8	C21—C20—C19	120.5 (2)
C1—C5—C4	116.54 (19)	C22—C21—C20	121.2 (3)
C1—C5—C6	121.97 (18)	C22—C21—H21	119.4
C4—C5—C6	121.49 (18)	C20—C21—H21	119.4
C5—C6—C7	114.71 (15)	C23—C22—C21	120.5 (3)
C5—C6—C9	117.24 (15)	C23—C22—H22	119.8
C7—C6—C9	104.63 (15)	C21—C22—H22	119.8
C5—C6—H6	106.5	C22—C23—C24	119.7 (3)
C7—C6—H6	106.5	C22—C23—H23	120.2
C9—C6—H6	106.5	C24—C23—H23	120.2
N2—C7—C6	103.80 (15)	C23—C24—C25	120.1 (3)
N2—C7—H7A	111.0	C23—C24—H24	120.0
C6—C7—H7A	111.0	C25—C24—H24	120.0
N2—C7—H7B	111.0	C20—C25—C24	120.8 (2)
C6—C7—H7B	111.0	C20—C25—H25	119.6
H7A—C7—H7B	109.0	C24—C25—H25	119.6
N2—C8—C16	113.72 (15)	O2—C26—C27	121.35 (17)
N2—C8—C10	115.22 (15)	O2—C26—C9	117.69 (16)
C16—C8—C10	101.27 (15)	C27—C26—C9	120.97 (17)
N2—C8—C9	101.96 (14)	C28—C27—C26	128.52 (18)
C16—C8—C9	117.23 (15)	C28—C27—C34	105.57 (16)
C10—C8—C9	107.90 (15)	C26—C27—C34	125.49 (18)
C18—C9—C26	108.50 (15)	N5—C28—C27	110.76 (18)
C18—C9—C6	111.34 (15)	N5—C28—H28	124.6

C26—C9—C6	110.09 (14)	C27—C28—H28	124.6
C18—C9—C8	110.89 (15)	C30—C29—N5	129.2 (2)
C26—C9—C8	112.48 (15)	C30—C29—C34	122.9 (2)
C6—C9—C8	103.51 (14)	N5—C29—C34	107.86 (17)
O1—C10—N3	125.13 (19)	C31—C30—C29	116.8 (2)
O1—C10—C8	126.77 (18)	C31—C30—H30	121.6
N3—C10—C8	108.09 (16)	C29—C30—H30	121.6
C12—C11—C16	122.13 (19)	C30—C31—C32	121.6 (2)
C12—C11—N3	128.23 (19)	C30—C31—H31	119.2
C16—C11—N3	109.63 (16)	C32—C31—H31	119.2
C11—C12—C13	117.1 (2)	C33—C32—C31	121.7 (2)
C11—C12—H12	121.4	C33—C32—H32	119.2
C13—C12—H12	121.4	C31—C32—H32	119.2
C14—C13—C12	121.7 (2)	C32—C33—C34	118.2 (2)
C14—C13—H13	119.1	C32—C33—H33	120.9
C12—C13—H13	119.1	C34—C33—H33	120.9
C13—C14—C15	120.7 (2)	C33—C34—C29	118.79 (18)
C13—C14—H14	119.7	C33—C34—C27	134.53 (18)
C15—C14—H14	119.7	C29—C34—C27	106.67 (18)
C16—C15—C14	118.6 (2)	C3—N1—C4	117.45 (19)
C16—C15—H15	120.7	C8—N2—C7	106.80 (15)
C14—C15—H15	120.7	C8—N2—C17	114.61 (17)
C15—C16—C11	119.75 (18)	C7—N2—C17	113.82 (16)
C15—C16—C8	130.88 (19)	C10—N3—C11	110.69 (16)
C11—C16—C8	109.36 (16)	C10—N3—C19	122.93 (17)
N2—C17—H17A	109.5	C11—N3—C19	125.88 (17)
N2—C17—H17B	109.5	C28—N5—C29	109.12 (17)
H17A—C17—H17B	109.5	C28—N5—H5	125.4
N2—C17—H17C	109.5	C29—N5—H5	125.4
C5—C1—C2—C3	0.4 (3)	C21—C22—C23—C24	-0.5 (4)
C1—C2—C3—N1	-0.7 (3)	C22—C23—C24—C25	1.5 (4)
C2—C1—C5—C4	0.8 (3)	C21—C20—C25—C24	0.2 (3)
C2—C1—C5—C6	-179.71 (17)	C19—C20—C25—C24	177.1 (2)
N1—C4—C5—C1	-1.9 (3)	C23—C24—C25—C20	-1.4 (4)
N1—C4—C5—C6	178.56 (18)	C18—C9—C26—O2	-139.91 (19)
C1—C5—C6—C7	-123.4 (2)	C6—C9—C26—O2	-17.8 (2)
C4—C5—C6—C7	56.0 (2)	C8—C9—C26—O2	97.0 (2)
C1—C5—C6—C9	113.2 (2)	C18—C9—C26—C27	40.2 (2)
C4—C5—C6—C9	-67.3 (2)	C6—C9—C26—C27	162.28 (17)
C5—C6—C7—N2	-156.36 (16)	C8—C9—C26—C27	-82.8 (2)
C9—C6—C7—N2	-26.51 (19)	O2—C26—C27—C28	-168.1 (2)
C5—C6—C9—C18	10.9 (2)	C9—C26—C27—C28	11.8 (3)
C7—C6—C9—C18	-117.49 (17)	O2—C26—C27—C34	3.4 (3)
C5—C6—C9—C26	-109.52 (18)	C9—C26—C27—C34	-176.69 (18)
C7—C6—C9—C26	122.14 (17)	C26—C27—C28—N5	172.8 (2)
C5—C6—C9—C8	130.04 (16)	C34—C27—C28—N5	-0.1 (2)
C7—C6—C9—C8	1.70 (18)	N5—C29—C30—C31	179.6 (2)

N2—C8—C9—C18	142.99 (15)	C34—C29—C30—C31	−0.4 (3)
C16—C8—C9—C18	−92.16 (19)	C29—C30—C31—C32	0.4 (4)
C10—C8—C9—C18	21.2 (2)	C30—C31—C32—C33	0.2 (4)
N2—C8—C9—C26	−95.31 (16)	C31—C32—C33—C34	−0.8 (4)
C16—C8—C9—C26	29.5 (2)	C32—C33—C34—C29	0.8 (3)
C10—C8—C9—C26	142.95 (15)	C32—C33—C34—C27	−179.9 (2)
N2—C8—C9—C6	23.50 (17)	C30—C29—C34—C33	−0.2 (3)
C16—C8—C9—C6	148.35 (16)	N5—C29—C34—C33	179.80 (18)
C10—C8—C9—C6	−98.24 (16)	C30—C29—C34—C27	−179.7 (2)
N2—C8—C10—O1	−48.4 (3)	N5—C29—C34—C27	0.3 (2)
C16—C8—C10—O1	−171.61 (19)	C28—C27—C34—C33	−179.5 (2)
C9—C8—C10—O1	64.7 (2)	C26—C27—C34—C33	7.4 (4)
N2—C8—C10—N3	132.87 (16)	C28—C27—C34—C29	−0.2 (2)
C16—C8—C10—N3	9.69 (19)	C26—C27—C34—C29	−173.29 (19)
C9—C8—C10—N3	−113.99 (16)	C2—C3—N1—C4	−0.4 (3)
C16—C11—C12—C13	−0.2 (3)	C5—C4—N1—C3	1.7 (3)
N3—C11—C12—C13	178.6 (2)	C16—C8—N2—C7	−169.48 (16)
C11—C12—C13—C14	−1.3 (4)	C10—C8—N2—C7	74.2 (2)
C12—C13—C14—C15	1.1 (4)	C9—C8—N2—C7	−42.32 (18)
C13—C14—C15—C16	0.7 (4)	C16—C8—N2—C17	63.5 (2)
C14—C15—C16—C11	−2.2 (3)	C10—C8—N2—C17	−52.8 (2)
C14—C15—C16—C8	178.4 (2)	C9—C8—N2—C17	−169.36 (16)
C12—C11—C16—C15	1.9 (3)	C6—C7—N2—C8	44.46 (19)
N3—C11—C16—C15	−177.05 (17)	C6—C7—N2—C17	171.97 (17)
C12—C11—C16—C8	−178.56 (18)	O1—C10—N3—C11	172.32 (19)
N3—C11—C16—C8	2.4 (2)	C8—C10—N3—C11	−9.0 (2)
N2—C8—C16—C15	48.1 (3)	O1—C10—N3—C19	−0.1 (3)
C10—C8—C16—C15	172.3 (2)	C8—C10—N3—C19	178.67 (16)
C9—C8—C16—C15	−70.7 (3)	C12—C11—N3—C10	−174.6 (2)
N2—C8—C16—C11	−131.37 (17)	C16—C11—N3—C10	4.3 (2)
C10—C8—C16—C11	−7.2 (2)	C12—C11—N3—C19	−2.5 (3)
C9—C8—C16—C11	109.90 (19)	C16—C11—N3—C19	176.42 (18)
N3—C19—C20—C25	52.9 (3)	C20—C19—N3—C10	−122.3 (2)
N3—C19—C20—C21	−130.3 (2)	C20—C19—N3—C11	66.5 (3)
C25—C20—C21—C22	0.7 (3)	C27—C28—N5—C29	0.3 (3)
C19—C20—C21—C22	−176.1 (2)	C30—C29—N5—C28	179.6 (2)
C20—C21—C22—C23	−0.6 (4)	C34—C29—N5—C28	−0.4 (2)

Hydrogen-bond geometry (\AA , $^\circ$)

$D\cdots H$	$D—H$	$H\cdots A$	$D\cdots A$	$D—H\cdots A$
N5—H5 ⁱⁱ ···N1 ⁱ	0.86	1.96	2.812 (2)	170
C19—H19A ⁱⁱ ···N4 ⁱⁱ	0.97	2.46	3.370 (3)	157

Symmetry codes: (i) $-x+1, y+1/2, -z+1/2$; (ii) $x-1/2, y, -z+1/2$.