

(*E*)-4-([4-(Benzo[*d*]thiazol-2-yl)phenyl]imino}methyl)-*N,N*-diethylaniline

Shun Yao and Xin Zhang*

Department of Chemistry, Anhui University, Hefei 230601, Peoples Republic of China and, Key Laboratory of Functional Inorganic Materials, Chemistry, Hefei 230601, People's Republic of China. *Correspondence e-mail: 329408172@qq.com

Received 24 September 2016

Accepted 12 October 2016

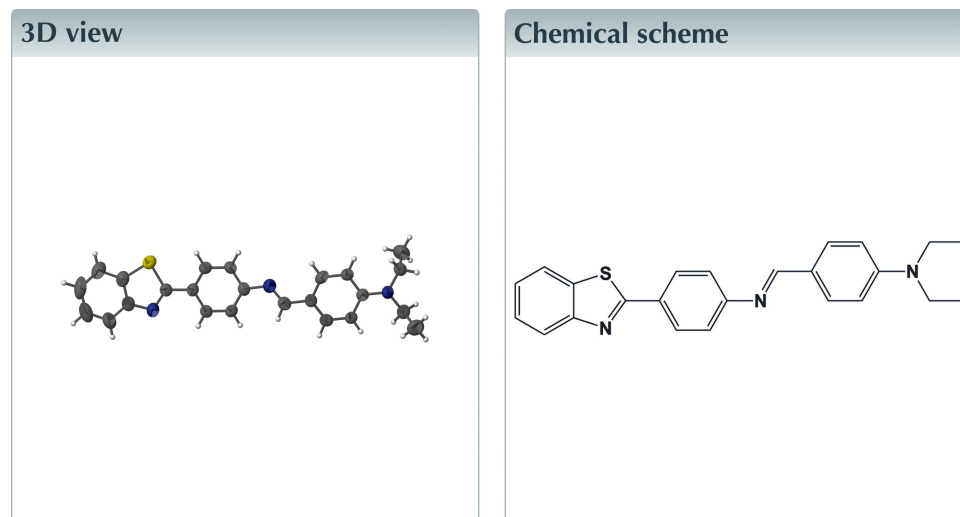
Edited by H. Stoeckli-Evans, University of Neuchâtel, Switzerland

Keywords: crystal structure; benzothiazole; Schiff base; C—H··· π interactions.

CCDC reference: 1509540

Structural data: full structural data are available from iucrdata.iucr.org

The title benzothiazole derivative molecule, C₂₄H₂₃N₃S, is almost planar with the central benzene ring being inclined to the benzothiazole ring and the diethylaniline ring by 4.05 (13) and 5.05 (13)°, respectively. The conformation about the N=C bond is *E*. In the crystal, molecules are linked by C—H··· π interactions, forming layers parallel to the *ab* plane.



Structure description

Benzothiazole derivatives are considered to be important because of their wide range of biological activities (Ali & Siddiqui, 2013). At the same time, their high electron affinity and good planarity make them appropriate building blocks for the construction of optical materials (Wang *et al.*, 2010). Herein, we report on the synthesis and crystal structure of the title compound, a new benzothiazole derivative.

The molecular structure is illustrated in Fig. 1. The molecule is almost planar with the central benzene ring (C8–C13) being inclined to the benzothiazole ring (C15–C20; r.m.s. deviation = 0.021 Å) and the diethylaniline ring (C15–C20) by 4.05 (13) and 5.05 (13)°, respectively. The conformation about the N2=C14 bond is *E*.

In the crystal, molecules are linked by C—H··· π interactions forming layers parallel to the *ab* plane (Table 1 and Fig. 2).

Synthesis and crystallization

4-(Diethylamino)salicylaldehyde (0.52 g, 2.7 mmol) was dissolved in 20 ml ethanol and added drop wise to a 20 ml ethanol solution of 4-(benzothiazol-2-yl)aniline (0.61 g, 2.7 mmol) with stirring. The mixture was stirred at room temperature, and then filtered.

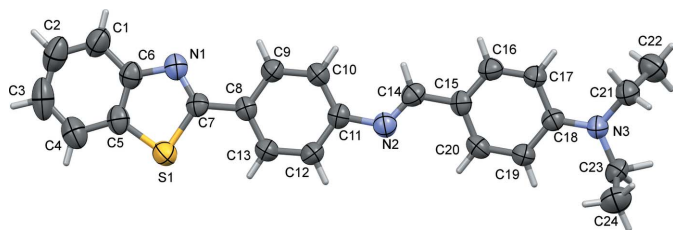


Figure 1
The molecular structure of the title compound, showing the atom labelling. Displacement ellipsoids drawn at the 50% probability level.

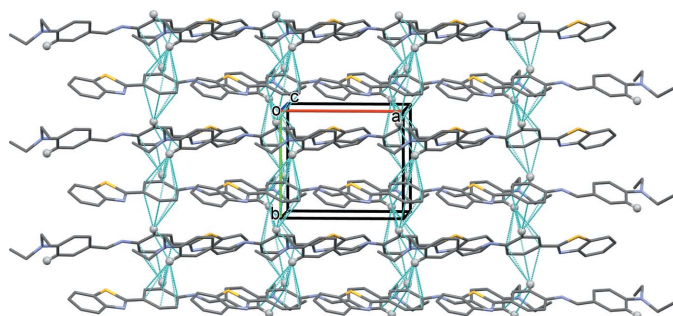


Figure 2
A view along the *c* axis of the crystal packing of the title compound. The C–H... π interactions are represented by dashed lines (see Table 1), and only the H atoms (H12 and H17; grey balls) involved in these interactions have been included.

The filtrate was allowed to evaporate slowly at room temperature and after one week, yellow block-like crystals were obtained (yield 62.0%).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

This work was supported by the Graduate Students Innovative Program of Anhui University (grant Nos. J18515024, J18515019 and 201310357155).

References

- Ali, R. & Siddiqui, N. (2013). *J. Chem.* Article ID 345198, 12 pp. <http://dx.doi.org/10.1155/2013/345198>
 Bruker (2004). *SMART*, *SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

Cg is the centroid of the C8–C13 ring.

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
C12–H12... <i>Cg</i> ⁱ	0.93	2.88	3.577 (5)	133
C17–H17... <i>Cg</i> ⁱⁱ	0.93	2.93	3.646 (4)	135

Symmetry codes: (i) $-x + 2, y - \frac{1}{2}, -z + 2$; (ii) $-x + 1, y + \frac{1}{2}, -z + 2$.

Table 2
Experimental details.

Crystal data	
Chemical formula	$\text{C}_{24}\text{H}_{23}\text{N}_3\text{S}$
<i>M_r</i>	385.51
Crystal system, space group	Monoclinic, <i>P2₁</i>
Temperature (K)	298
<i>a</i> , <i>b</i> , <i>c</i> (\AA)	8.895 (5), 7.648 (5), 15.091 (5)
β ($^\circ$)	98.566 (5)
<i>V</i> (\AA^3)	1015.2 (9)
<i>Z</i>	2
Radiation type	Mo <i>K</i> α
μ (mm^{-1})	0.17
Crystal size (mm)	0.30 \times 0.20 \times 0.20
Data collection	
Diffractometer	Bruker SMART CCD area detector
Absorption correction	Multi-scan (<i>SADABS</i> ; Bruker, 2004)
<i>T_{min}</i> , <i>T_{max}</i>	0.950, 0.966
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	7218, 3278, 2690
<i>R_{int}</i>	0.024
($\sin \theta/\lambda$) _{max} (\AA^{-1})	0.595
Refinement	
$R[F^2 > 2\sigma(F^2)]$, $wR(F^2)$, <i>S</i>	0.044, 0.122, 1.04
No. of reflections	3278
No. of parameters	256
No. of restraints	1
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}$, $\Delta\rho_{\text{min}}$ (e \AA^{-3})	0.23, -0.18
Absolute structure	Flack (1983), 1363 Friedel pairs
Absolute structure parameter	0.27 (10)

Computer programs: *SMART* and *SAINT* (Bruker, 2004), *SHELXS97*, *SHELXL97* and *SHELXTL* (Sheldrick, 2008), *Mercury* (Macrae *et al.*, 2008) and *PLATON* (Spek, 2009).

- Macrae, C. F., Bruno, I. J., Chisholm, J. A., Edgington, P. R., McCabe, P., Pidcock, E., Rodriguez-Monge, L., Taylor, R., van de Streek, J. & Wood, P. A. (2008). *J. Appl. Cryst.* **41**, 466–470.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
 Wang, H., Chen, G., Xu, X., Chen, H. & Ji, S. (2010). *Dyes Pigments*, **86**, 238–248.

full crystallographic data

IUCrData (2016). **1**, x161621 [<https://doi.org/10.1107/S2414314616016217>]

(*E*)-4-([4-(Benzo[*d*]thiazol-2-yl)phenyl]imino)methyl)-*N,N*-diethylaniline

Shun Yao and Xin Zhang

(*E*)-4-([4-(Benzo[*d*]thiazol-2-yl)phenyl]imino)methyl)-*N,N*-diethylaniline*Crystal data*

$C_{24}H_{23}N_3S$

$M_r = 385.51$

Monoclinic, $P2_1$

Hall symbol: $P\ 2y_b$

$a = 8.895\ (5)\ \text{\AA}$

$b = 7.648\ (5)\ \text{\AA}$

$c = 15.091\ (5)\ \text{\AA}$

$\beta = 98.566\ (5)^\circ$

$V = 1015.2\ (9)\ \text{\AA}^3$

$Z = 2$

$F(000) = 408$

$D_x = 1.261\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71069\ \text{\AA}$

Cell parameters from 2407 reflections

$\theta = 2.5\text{--}23.5^\circ$

$\mu = 0.17\ \text{mm}^{-1}$

$T = 298\ \text{K}$

Block, yellow

$0.30 \times 0.20 \times 0.20\ \text{mm}$

Data collection

Bruker SMART CCD area detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

phi scans

Absorption correction: multi-scan

(SADABS; Bruker, 2004)

$T_{\min} = 0.950$, $T_{\max} = 0.966$

7218 measured reflections

3278 independent reflections

2690 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.024$

$\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 1.4^\circ$

$h = -10 \rightarrow 10$

$k = -8 \rightarrow 9$

$l = 0 \rightarrow 17$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.044$

$wR(F^2) = 0.122$

$S = 1.04$

3278 reflections

256 parameters

1 restraint

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier

map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0719P)^2 + 0.0425P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.23\ \text{e \AA}^{-3}$

$\Delta\rho_{\min} = -0.18\ \text{e \AA}^{-3}$

Absolute structure: Flack (1983), 1363 Friedel

pairs

Absolute structure parameter: 0.27 (10)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	1.41511 (8)	0.16492 (14)	1.18673 (5)	0.0654 (3)
N1	1.2597 (3)	0.3273 (4)	1.29530 (16)	0.0583 (6)
N2	0.7097 (3)	0.2212 (3)	0.96102 (15)	0.0578 (7)
N3	0.0747 (3)	0.2513 (4)	0.69361 (15)	0.0600 (6)
C1	1.4589 (4)	0.3630 (6)	1.4264 (2)	0.0781 (11)
H1	1.3944	0.4240	1.4586	0.094*
C2	1.6035 (5)	0.3290 (7)	1.4630 (2)	0.0928 (13)
H2	1.6383	0.3681	1.5207	0.111*
C3	1.7014 (4)	0.2380 (6)	1.4172 (3)	0.0945 (14)
H3	1.7999	0.2151	1.4451	0.113*
C4	1.6555 (3)	0.1801 (6)	1.3303 (2)	0.0773 (10)
H4	1.7214	0.1199	1.2988	0.093*
C5	1.5068 (3)	0.2160 (4)	1.2925 (2)	0.0602 (8)
C6	1.4073 (3)	0.3051 (4)	1.33936 (19)	0.0588 (8)
C7	1.2487 (3)	0.2597 (4)	1.21464 (17)	0.0483 (6)
C8	1.1075 (3)	0.2557 (4)	1.15066 (17)	0.0462 (6)
C9	0.9747 (3)	0.3337 (4)	1.17029 (18)	0.0536 (7)
H9	0.9753	0.3904	1.2249	0.064*
C10	0.8426 (3)	0.3278 (4)	1.10960 (18)	0.0560 (7)
H10	0.7556	0.3824	1.1233	0.067*
C11	0.8375 (3)	0.2414 (4)	1.02829 (17)	0.0490 (6)
C12	0.9689 (3)	0.1634 (5)	1.00924 (17)	0.0541 (7)
H12	0.9677	0.1055	0.9550	0.065*
C13	1.1012 (3)	0.1698 (5)	1.06886 (17)	0.0562 (7)
H13	1.1881	0.1159	1.0544	0.067*
C14	0.5823 (3)	0.2863 (4)	0.96753 (18)	0.0551 (7)
H14	0.5718	0.3493	1.0190	0.066*
C15	0.4502 (3)	0.2683 (4)	0.89868 (17)	0.0470 (6)
C16	0.3150 (3)	0.3497 (4)	0.90676 (19)	0.0567 (7)
H16	0.3071	0.4113	0.9590	0.068*
C17	0.1925 (3)	0.3434 (4)	0.84128 (18)	0.0530 (7)
H17	0.1032	0.3998	0.8500	0.064*
C18	0.1979 (3)	0.2539 (4)	0.76100 (17)	0.0482 (6)
C19	0.3333 (3)	0.1671 (5)	0.75167 (17)	0.0532 (6)
H19	0.3405	0.1053	0.6994	0.064*

C20	0.4569 (3)	0.1725 (5)	0.81967 (18)	0.0548 (7)
H20	0.5452	0.1120	0.8130	0.066*
C21	-0.0560 (3)	0.3699 (5)	0.6928 (2)	0.0655 (9)
H21A	-0.0899	0.4077	0.6318	0.079*
H21B	-0.0234	0.4729	0.7281	0.079*
C22	-0.1861 (4)	0.2892 (6)	0.7288 (2)	0.0884 (12)
H22A	-0.2191	0.1870	0.6944	0.133*
H22B	-0.2682	0.3716	0.7251	0.133*
H22C	-0.1550	0.2568	0.7903	0.133*
C23	0.0718 (3)	0.1430 (5)	0.61427 (18)	0.0647 (9)
H23A	-0.0327	0.1119	0.5921	0.078*
H23B	0.1269	0.0356	0.6309	0.078*
C24	0.1398 (4)	0.2295 (6)	0.5403 (2)	0.0914 (13)
H24A	0.0835	0.3337	0.5217	0.137*
H24B	0.1355	0.1508	0.4905	0.137*
H24C	0.2437	0.2595	0.5614	0.137*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0536 (4)	0.0666 (5)	0.0750 (5)	0.0054 (4)	0.0065 (3)	-0.0112 (5)
N1	0.0498 (14)	0.0609 (16)	0.0624 (14)	-0.0015 (12)	0.0022 (11)	0.0012 (12)
N2	0.0528 (14)	0.0632 (19)	0.0567 (14)	0.0001 (12)	0.0060 (10)	0.0011 (12)
N3	0.0488 (13)	0.0723 (17)	0.0558 (13)	0.0090 (13)	-0.0018 (10)	-0.0137 (13)
C1	0.074 (2)	0.099 (3)	0.0584 (19)	-0.017 (2)	-0.0003 (16)	-0.0030 (18)
C2	0.097 (3)	0.109 (3)	0.064 (2)	-0.024 (3)	-0.016 (2)	0.011 (2)
C3	0.062 (2)	0.095 (3)	0.114 (3)	-0.016 (2)	-0.027 (2)	0.038 (3)
C4	0.0561 (18)	0.071 (2)	0.103 (3)	0.000 (2)	0.0046 (17)	0.008 (2)
C5	0.0527 (16)	0.052 (2)	0.0737 (19)	-0.0072 (14)	0.0014 (14)	0.0095 (14)
C6	0.0565 (17)	0.060 (2)	0.0586 (16)	-0.0093 (15)	0.0049 (13)	0.0057 (15)
C7	0.0511 (15)	0.0399 (15)	0.0534 (15)	-0.0045 (13)	0.0058 (12)	-0.0006 (13)
C8	0.0434 (14)	0.0407 (15)	0.0537 (14)	-0.0021 (13)	0.0047 (11)	0.0032 (13)
C9	0.0520 (16)	0.0570 (18)	0.0510 (15)	0.0000 (14)	0.0046 (12)	-0.0054 (14)
C10	0.0467 (15)	0.062 (2)	0.0584 (16)	0.0064 (14)	0.0067 (13)	-0.0047 (15)
C11	0.0504 (15)	0.0471 (16)	0.0485 (14)	-0.0050 (13)	0.0035 (11)	0.0045 (13)
C12	0.0539 (15)	0.0604 (17)	0.0473 (13)	-0.0001 (17)	0.0048 (11)	-0.0094 (15)
C13	0.0511 (14)	0.0612 (17)	0.0563 (15)	0.0068 (17)	0.0082 (11)	-0.0067 (17)
C14	0.0623 (19)	0.0495 (19)	0.0526 (15)	-0.0010 (15)	0.0057 (13)	-0.0013 (14)
C15	0.0484 (15)	0.0424 (16)	0.0499 (14)	-0.0065 (13)	0.0060 (11)	0.0007 (13)
C16	0.0598 (18)	0.0551 (19)	0.0555 (16)	0.0039 (15)	0.0098 (13)	-0.0068 (14)
C17	0.0475 (15)	0.0539 (18)	0.0572 (16)	0.0046 (14)	0.0061 (13)	-0.0097 (15)
C18	0.0488 (15)	0.0451 (15)	0.0508 (15)	-0.0004 (13)	0.0074 (11)	0.0000 (14)
C19	0.0498 (14)	0.0565 (16)	0.0532 (14)	0.0034 (17)	0.0076 (11)	-0.0069 (16)
C20	0.0428 (13)	0.0542 (16)	0.0693 (17)	0.0049 (15)	0.0146 (12)	0.0076 (17)
C21	0.0592 (19)	0.069 (2)	0.0660 (19)	0.0069 (17)	0.0019 (15)	-0.0037 (16)
C22	0.072 (2)	0.100 (3)	0.097 (3)	-0.003 (2)	0.0224 (19)	-0.009 (2)
C23	0.0601 (17)	0.074 (2)	0.0579 (16)	0.0041 (18)	0.0008 (13)	-0.0168 (18)
C24	0.100 (3)	0.113 (4)	0.064 (2)	0.003 (3)	0.0193 (19)	-0.004 (2)

Geometric parameters (Å, °)

S1—C5	1.725 (3)	C12—C13	1.372 (3)
S1—C7	1.755 (3)	C12—H12	0.9300
N1—C7	1.313 (4)	C13—H13	0.9300
N1—C6	1.390 (4)	C14—C15	1.454 (4)
N2—C14	1.255 (4)	C14—H14	0.9300
N2—C11	1.415 (3)	C15—C16	1.376 (4)
N3—C18	1.379 (3)	C15—C20	1.408 (4)
N3—C23	1.453 (4)	C16—C17	1.358 (4)
N3—C21	1.474 (4)	C16—H16	0.9300
C1—C2	1.348 (5)	C17—C18	1.398 (4)
C1—C6	1.397 (4)	C17—H17	0.9300
C1—H1	0.9300	C18—C19	1.401 (4)
C2—C3	1.377 (6)	C19—C20	1.388 (3)
C2—H2	0.9300	C19—H19	0.9300
C3—C4	1.387 (5)	C20—H20	0.9300
C3—H3	0.9300	C21—C22	1.483 (5)
C4—C5	1.388 (4)	C21—H21A	0.9700
C4—H4	0.9300	C21—H21B	0.9700
C5—C6	1.391 (4)	C22—H22A	0.9600
C7—C8	1.466 (4)	C22—H22B	0.9600
C8—C13	1.392 (4)	C22—H22C	0.9600
C8—C9	1.394 (4)	C23—C24	1.500 (5)
C9—C10	1.379 (4)	C23—H23A	0.9700
C9—H9	0.9300	C23—H23B	0.9700
C10—C11	1.388 (4)	C24—H24A	0.9600
C10—H10	0.9300	C24—H24B	0.9600
C11—C12	1.380 (4)	C24—H24C	0.9600
C5—S1—C7	89.09 (14)	N2—C14—C15	123.3 (3)
C7—N1—C6	109.3 (2)	N2—C14—H14	118.3
C14—N2—C11	122.6 (2)	C15—C14—H14	118.3
C18—N3—C23	122.0 (2)	C16—C15—C20	117.3 (2)
C18—N3—C21	122.4 (2)	C16—C15—C14	121.2 (3)
C23—N3—C21	115.3 (2)	C20—C15—C14	121.5 (2)
C2—C1—C6	118.9 (4)	C17—C16—C15	122.5 (3)
C2—C1—H1	120.5	C17—C16—H16	118.8
C6—C1—H1	120.5	C15—C16—H16	118.8
C1—C2—C3	121.9 (4)	C16—C17—C18	121.3 (3)
C1—C2—H2	119.1	C16—C17—H17	119.3
C3—C2—H2	119.1	C18—C17—H17	119.3
C2—C3—C4	121.1 (3)	N3—C18—C17	121.3 (2)
C2—C3—H3	119.5	N3—C18—C19	121.3 (2)
C4—C3—H3	119.5	C17—C18—C19	117.4 (2)
C3—C4—C5	117.0 (4)	C20—C19—C18	120.7 (3)
C3—C4—H4	121.5	C20—C19—H19	119.7
C5—C4—H4	121.5	C18—C19—H19	119.7

C4—C5—C6	121.8 (3)	C19—C20—C15	120.8 (3)
C4—C5—S1	128.8 (3)	C19—C20—H20	119.6
C6—C5—S1	109.4 (2)	C15—C20—H20	119.6
N1—C6—C5	116.3 (3)	N3—C21—C22	113.5 (3)
N1—C6—C1	124.4 (3)	N3—C21—H21A	108.9
C5—C6—C1	119.2 (3)	C22—C21—H21A	108.9
N1—C7—C8	123.9 (2)	N3—C21—H21B	108.9
N1—C7—S1	115.9 (2)	C22—C21—H21B	108.9
C8—C7—S1	120.3 (2)	H21A—C21—H21B	107.7
C13—C8—C9	117.9 (2)	C21—C22—H22A	109.5
C13—C8—C7	120.7 (2)	C21—C22—H22B	109.5
C9—C8—C7	121.4 (2)	H22A—C22—H22B	109.5
C10—C9—C8	120.7 (3)	C21—C22—H22C	109.5
C10—C9—H9	119.7	H22A—C22—H22C	109.5
C8—C9—H9	119.7	H22B—C22—H22C	109.5
C9—C10—C11	121.0 (3)	N3—C23—C24	113.7 (3)
C9—C10—H10	119.5	N3—C23—H23A	108.8
C11—C10—H10	119.5	C24—C23—H23A	108.8
C12—C11—C10	118.2 (2)	N3—C23—H23B	108.8
C12—C11—N2	114.7 (2)	C24—C23—H23B	108.8
C10—C11—N2	127.1 (3)	H23A—C23—H23B	107.7
C13—C12—C11	121.2 (3)	C23—C24—H24A	109.5
C13—C12—H12	119.4	C23—C24—H24B	109.5
C11—C12—H12	119.4	H24A—C24—H24B	109.5
C12—C13—C8	121.0 (3)	C23—C24—H24C	109.5
C12—C13—H13	119.5	H24A—C24—H24C	109.5
C8—C13—H13	119.5	H24B—C24—H24C	109.5
C6—C1—C2—C3	-0.4 (6)	C14—N2—C11—C12	-179.0 (3)
C1—C2—C3—C4	1.2 (7)	C14—N2—C11—C10	1.9 (5)
C2—C3—C4—C5	-0.8 (6)	C10—C11—C12—C13	0.3 (5)
C3—C4—C5—C6	-0.4 (5)	N2—C11—C12—C13	-178.9 (3)
C3—C4—C5—S1	179.3 (3)	C11—C12—C13—C8	-0.1 (6)
C7—S1—C5—C4	178.3 (3)	C9—C8—C13—C12	0.5 (5)
C7—S1—C5—C6	-2.0 (2)	C7—C8—C13—C12	179.1 (3)
C7—N1—C6—C5	-1.7 (4)	C11—N2—C14—C15	179.4 (3)
C7—N1—C6—C1	179.5 (3)	N2—C14—C15—C16	-176.6 (3)
C4—C5—C6—N1	-177.7 (3)	N2—C14—C15—C20	1.0 (5)
S1—C5—C6—N1	2.6 (3)	C20—C15—C16—C17	-1.6 (5)
C4—C5—C6—C1	1.2 (5)	C14—C15—C16—C17	176.1 (3)
S1—C5—C6—C1	-178.6 (3)	C15—C16—C17—C18	-0.5 (5)
C2—C1—C6—N1	178.0 (4)	C23—N3—C18—C17	-172.8 (3)
C2—C1—C6—C5	-0.8 (5)	C21—N3—C18—C17	13.3 (5)
C6—N1—C7—C8	178.2 (3)	C23—N3—C18—C19	7.3 (5)
C6—N1—C7—S1	0.1 (3)	C21—N3—C18—C19	-166.7 (3)
C5—S1—C7—N1	1.2 (3)	C16—C17—C18—N3	-178.3 (3)
C5—S1—C7—C8	-177.0 (2)	C16—C17—C18—C19	1.6 (5)
N1—C7—C8—C13	-175.0 (3)	N3—C18—C19—C20	179.3 (3)

S1—C7—C8—C13	3.1 (4)	C17—C18—C19—C20	-0.7 (5)
N1—C7—C8—C9	3.5 (4)	C18—C19—C20—C15	-1.4 (5)
S1—C7—C8—C9	-178.4 (2)	C16—C15—C20—C19	2.5 (5)
C13—C8—C9—C10	-1.0 (4)	C14—C15—C20—C19	-175.2 (3)
C7—C8—C9—C10	-179.6 (3)	C18—N3—C21—C22	-97.4 (4)
C8—C9—C10—C11	1.2 (5)	C23—N3—C21—C22	88.3 (3)
C9—C10—C11—C12	-0.8 (5)	C18—N3—C23—C24	-87.5 (4)
C9—C10—C11—N2	178.3 (3)	C21—N3—C23—C24	86.9 (3)

Hydrogen-bond geometry (Å, °)

Cg is the centroid of the C8—C13 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C12—H12 \cdots Cg ⁱ	0.93	2.88	3.577 (5)	133
C17—H17 \cdots Cg ⁱⁱ	0.93	2.93	3.646 (4)	135

Symmetry codes: (i) $-x+2, y-1/2, -z+2$; (ii) $-x+1, y+1/2, -z+2$.