

2-[(5-Amino-1,3,4-thiadiazol-2-yl)sulfanyl]-*N*-(2,4,5-trichlorophenyl)acetamide

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Received 1 July 2016

Accepted 11 July 2016

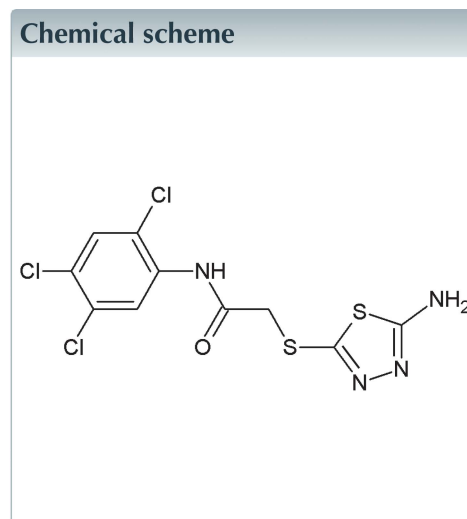
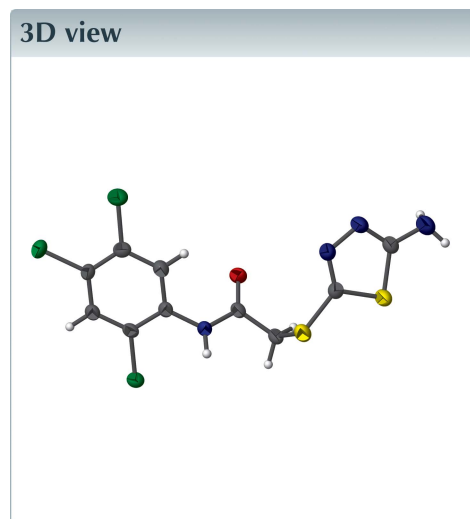
Edited by H. Stoeckli-Evans, University of Neuchâtel, Switzerland

Keywords: crystal structure; imidazothiadiazole; hydrogen bonding.

CCDC reference: 1491987

Structural data: full structural data are available from iucrdata.iucr.org

In the title compound, C₁₀H₇Cl₃N₄OS₂, the dihedral angle between the trichlorobenzene and thiadiazole rings is 29.26 (17)°. In the crystal, molecules are connected by N—H···O and C—H···O hydrogen bonds, forming chains propagating along [001]. The chains are linked *via* N—H···N hydrogen bonds to form slabs parallel to (100).



Structure description

As part of our research on the syntheses and crystal structure analyses of thiadiazole derivatives, we report herein on the synthesis and crystal structure of the title compound.

The molecular structure of the title compound is shown in Fig. 1. The dihedral angle between the trichlorobenzene ring (C5–C10) and the thiadiazole moiety (C1/C2/N1/N2/S1) is 29.26 (17)°.

In the crystal, molecules are connected by N—H···O and C—H···O hydrogen bonds, forming chains propagating along the *c*-axis direction (Table 1 and Fig. 2). The chains are linked *via* N—H···N hydrogen bonds to form slabs parallel to the *bc* plane (Table 1 and Fig. 2).

Synthesis and crystallization

An equimolar ratio of compound 2-((5-amino-1,3,4-thiadiazol-2-yl)thio)-*N*-(trichlorophenyl)acetamide (0.005 mol) and ethyl chloroacetate (0.005 mol) in glacial acetic acid

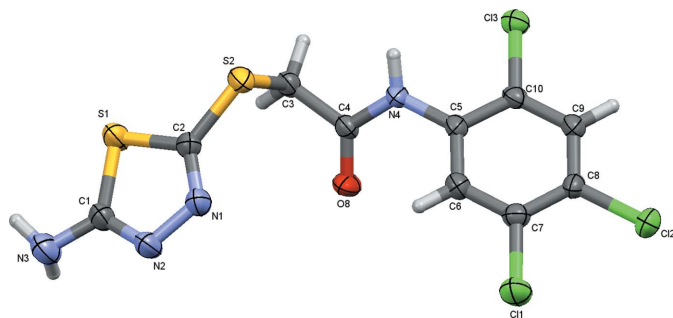


Figure 1
A view of the molecular structure of the title compound, showing the atom labelling. Displacement ellipsoids are drawn at the 50% probability level.

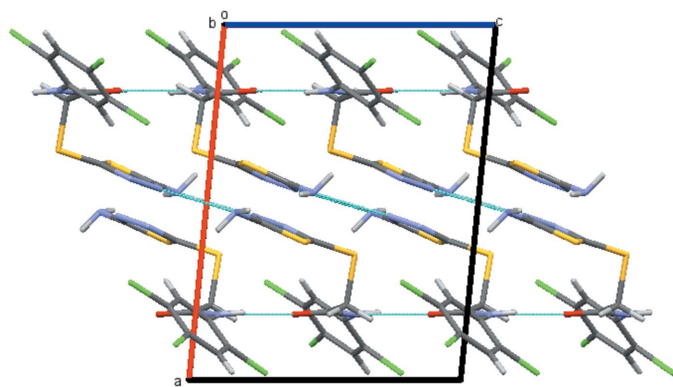


Figure 2
A view along the *b* axis of the crystal packing of the title compound. Hydrogen bonds are drawn as dashed lines (see Table 1).

(20 ml) was heated under reflux for 17 h. The reaction mixture was poured into ice-cold water. The precipitated solid was filtered, dried and recrystallized from ethanol.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

The authors thank DST-PURSE, Mangalore University, Mangaluru, for providing the single-crystal X-ray diffraction facility.

Table 1
Hydrogen-bond geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
N4—H4···O8 ⁱ	0.86	2.09	2.927 (3)	163
C3—H3D···O8 ⁱ	0.97	2.43	3.166 (3)	133
N3—H3A···N1 ⁱⁱ	0.86	2.51	3.304 (4)	154

Symmetry codes: (i) $x, -y + \frac{3}{2}, z - \frac{1}{2}$; (ii) $-x + 1, y + \frac{1}{2}, -z + \frac{3}{2}$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₀ H ₇ Cl ₃ N ₄ OS ₂
<i>M_r</i>	369.67
Crystal system, space group	Monoclinic, <i>P</i> ₂ ₁ / <i>c</i>
Temperature (K)	273
<i>a</i> , <i>b</i> , <i>c</i> (Å)	12.4679 (12), 11.9467 (11), 9.5278 (8)
β (°)	95.701 (7)
<i>V</i> (Å ³)	1412.1 (2)
<i>Z</i>	4
Radiation type	Mo <i>K</i> α
μ (mm ⁻¹)	0.94
Crystal size (mm)	0.32 × 0.23 × 0.1
Data collection	
Diffractometer	Rigaku Saturn724+
Absorption correction	Multi-scan (NUMABS; Rigaku, 1999)
<i>T_{min}</i> , <i>T_{max}</i>	0.771, 0.910
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	9999, 3200, 2292
<i>R_{int}</i>	0.052
(sin θ/λ) _{max} (Å ⁻¹)	0.649
Refinement	
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)], <i>wR</i> [<i>F</i> ²], <i>S</i>	0.051, 0.128, 1.09
No. of reflections	3200
No. of parameters	182
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	0.46, -0.41

Computer programs: *CrystalClear* (Rigaku, 2011), *SHELXS97* (Sheldrick, 2008), *SHELXL2014* (Sheldrick, 2015), *Mercury* (Macrae *et al.*, 2008) and *OLEX2* (Dolomanov *et al.*, 2009).

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full crystallographic data

IUCrData (2016). **1**, x161123 [https://doi.org/10.1107/S2414314616011238]

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$C_{10}H_7Cl_3N_4OS_2$

$M_r = 369.67$

Monoclinic, $P2_1/c$

$a = 12.4679$ (12) Å

$b = 11.9467$ (11) Å

$c = 9.5278$ (8) Å

$\beta = 95.701$ (7)°

$V = 1412.1$ (2) Å³

$Z = 4$

$F(000) = 744$

$D_x = 1.739$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71075$ Å

Cell parameters from 3200 reflections

$\theta = 3.1$ – 27.5 °

$\mu = 0.94$ mm⁻¹

$T = 273$ K

Block, colourless

$0.32 \times 0.23 \times 0.1$ mm

Data collection

Rigaku Saturn724+

diffractometer

profile data from ω -scans

Absorption correction: multi-scan

(NUMABS; Rigaku, 1999)

$T_{\min} = 0.771$, $T_{\max} = 0.910$

9999 measured reflections

3200 independent reflections

2292 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.052$

$\theta_{\max} = 27.5$ °, $\theta_{\min} = 3.1$ °

$h = -16 \rightarrow 16$

$k = -15 \rightarrow 15$

$l = -9 \rightarrow 12$

3200 standard reflections

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.051$

$wR(F^2) = 0.128$

$S = 1.09$

3200 reflections

182 parameters

0 restraints

Primary atom site location: structure-invariant direct methods

Hydrogen site location: mixed

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0554P)^2 + 0.4363P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.46$ e Å⁻³

$\Delta\rho_{\min} = -0.41$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.27431 (8)	0.33324 (7)	0.75005 (10)	0.0522 (3)
Cl2	0.10023 (8)	0.19165 (6)	0.56097 (9)	0.0479 (3)
Cl3	0.01952 (7)	0.54537 (6)	0.22229 (8)	0.0393 (2)
S1	0.38818 (8)	1.08239 (7)	0.62462 (9)	0.0427 (3)
S2	0.35323 (7)	0.87823 (7)	0.43499 (8)	0.0381 (2)
O8	0.1844 (2)	0.73427 (18)	0.6362 (2)	0.0415 (6)
N1	0.4321 (3)	0.8833 (2)	0.7082 (3)	0.0447 (7)
N2	0.4623 (3)	0.9512 (2)	0.8237 (3)	0.0505 (8)
N3	0.4728 (3)	1.1403 (3)	0.8894 (3)	0.0576 (9)
H3A	0.4773	1.2027	0.8453	0.069*
H3B	0.4237	1.1454	0.9466	0.069*
N4	0.1841 (2)	0.6632 (2)	0.4152 (3)	0.0332 (6)
H4	0.1894	0.6797	0.3283	0.040*
C1	0.4452 (3)	1.0570 (3)	0.7939 (3)	0.0390 (8)
C2	0.3929 (3)	0.9389 (3)	0.5991 (3)	0.0333 (7)
C3	0.2102 (3)	0.8607 (2)	0.4479 (3)	0.0331 (7)
H3C	0.1845	0.9190	0.5071	0.040*
H3D	0.1710	0.8665	0.3550	0.040*
C4	0.1910 (2)	0.7478 (2)	0.5102 (3)	0.0291 (7)
C5	0.1687 (2)	0.5499 (2)	0.4494 (3)	0.0285 (7)
C6	0.2261 (3)	0.4997 (2)	0.5656 (3)	0.0336 (7)
H6	0.2784	0.5405	0.6202	0.040*
C7	0.2061 (3)	0.3899 (3)	0.6005 (3)	0.0336 (7)
C8	0.1297 (3)	0.3284 (2)	0.5181 (3)	0.0316 (7)
C9	0.0733 (3)	0.3762 (2)	0.4015 (3)	0.0315 (7)
H9	0.0224	0.3346	0.3457	0.038*
C10	0.0930 (2)	0.4854 (2)	0.3683 (3)	0.0295 (7)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0694 (7)	0.0412 (5)	0.0413 (5)	0.0011 (4)	-0.0173 (4)	0.0066 (4)
Cl2	0.0716 (7)	0.0291 (4)	0.0417 (5)	-0.0121 (4)	-0.0004 (4)	0.0057 (3)
Cl3	0.0487 (5)	0.0355 (4)	0.0316 (4)	-0.0010 (4)	-0.0065 (4)	0.0017 (3)
S1	0.0567 (6)	0.0308 (4)	0.0373 (5)	0.0017 (4)	-0.0116 (4)	-0.0020 (4)
S2	0.0461 (5)	0.0388 (4)	0.0295 (5)	-0.0096 (4)	0.0043 (4)	-0.0046 (4)
O8	0.0680 (17)	0.0346 (12)	0.0224 (12)	-0.0034 (11)	0.0067 (11)	-0.0042 (9)
N1	0.065 (2)	0.0304 (14)	0.0357 (16)	-0.0071 (14)	-0.0080 (14)	0.0003 (12)
N2	0.078 (2)	0.0364 (16)	0.0329 (16)	-0.0065 (15)	-0.0137 (15)	-0.0010 (13)
N3	0.086 (3)	0.0422 (17)	0.0415 (19)	-0.0042 (17)	-0.0088 (17)	-0.0072 (15)
N4	0.0512 (18)	0.0266 (12)	0.0218 (13)	-0.0073 (12)	0.0028 (12)	-0.0026 (10)
C1	0.046 (2)	0.0341 (16)	0.0348 (18)	-0.0052 (15)	-0.0067 (15)	-0.0024 (14)
C2	0.0362 (18)	0.0303 (15)	0.0324 (17)	-0.0055 (13)	-0.0007 (14)	0.0014 (13)
C3	0.0391 (19)	0.0272 (15)	0.0310 (17)	0.0001 (14)	-0.0073 (14)	-0.0026 (13)
C4	0.0306 (17)	0.0295 (15)	0.0266 (17)	-0.0016 (13)	-0.0012 (13)	-0.0043 (13)

C5	0.0368 (18)	0.0255 (14)	0.0236 (15)	-0.0027 (13)	0.0054 (12)	-0.0039 (12)
C6	0.0387 (19)	0.0317 (16)	0.0298 (17)	-0.0047 (14)	0.0003 (14)	-0.0033 (13)
C7	0.0398 (19)	0.0328 (16)	0.0279 (16)	0.0030 (14)	0.0011 (14)	0.0016 (13)
C8	0.045 (2)	0.0233 (14)	0.0271 (16)	-0.0053 (13)	0.0061 (14)	-0.0008 (12)
C9	0.0396 (18)	0.0297 (15)	0.0254 (16)	-0.0053 (14)	0.0032 (13)	-0.0040 (13)
C10	0.0347 (18)	0.0319 (15)	0.0216 (15)	0.0006 (13)	0.0011 (13)	-0.0039 (12)

Geometric parameters (Å, °)

C11—C7	1.724 (3)	N4—H4	0.8600
C12—C8	1.733 (3)	N4—C4	1.354 (4)
C13—C10	1.742 (3)	N4—C5	1.410 (4)
S1—C1	1.723 (3)	C3—H3C	0.9700
S1—C2	1.733 (3)	C3—H3D	0.9700
S2—C2	1.750 (3)	C3—C4	1.503 (4)
S2—C3	1.812 (3)	C5—C6	1.393 (4)
O8—C4	1.222 (4)	C5—C10	1.392 (4)
N1—N2	1.388 (4)	C6—H6	0.9300
N1—C2	1.287 (4)	C6—C7	1.383 (4)
N2—C1	1.308 (4)	C7—C8	1.383 (4)
N3—H3A	0.8607	C8—C9	1.378 (4)
N3—H3B	0.8611	C9—H9	0.9300
N3—C1	1.369 (4)	C9—C10	1.371 (4)
C1—S1—C2	86.57 (15)	O8—C4—N4	123.5 (3)
C2—S2—C3	100.39 (15)	O8—C4—C3	122.4 (3)
C2—N1—N2	113.0 (3)	N4—C4—C3	114.1 (3)
C1—N2—N1	111.5 (3)	C6—C5—N4	121.7 (3)
H3A—N3—H3B	109.4	C10—C5—N4	120.2 (3)
C1—N3—H3A	109.3	C10—C5—C6	118.0 (3)
C1—N3—H3B	109.0	C5—C6—H6	119.7
C4—N4—H4	117.9	C7—C6—C5	120.6 (3)
C4—N4—C5	124.3 (3)	C7—C6—H6	119.7
C5—N4—H4	117.9	C6—C7—C11	119.0 (2)
N2—C1—S1	114.6 (2)	C6—C7—C8	119.9 (3)
N2—C1—N3	122.2 (3)	C8—C7—C11	121.1 (2)
N3—C1—S1	123.2 (3)	C7—C8—C12	121.1 (2)
S1—C2—S2	121.61 (18)	C9—C8—C12	118.5 (2)
N1—C2—S1	114.4 (2)	C9—C8—C7	120.4 (3)
N1—C2—S2	124.0 (3)	C8—C9—H9	120.3
S2—C3—H3C	109.9	C10—C9—C8	119.4 (3)
S2—C3—H3D	109.9	C10—C9—H9	120.3
H3C—C3—H3D	108.3	C5—C10—C13	119.3 (2)
C4—C3—S2	109.1 (2)	C9—C10—C13	118.9 (2)
C4—C3—H3C	109.9	C9—C10—C5	121.8 (3)
C4—C3—H3D	109.9		
C11—C7—C8—C12	-0.5 (4)	C2—N1—N2—C1	-1.0 (5)

C11—C7—C8—C9	177.8 (2)	C3—S2—C2—S1	-87.5 (2)
C12—C8—C9—C10	177.9 (2)	C3—S2—C2—N1	95.8 (3)
S2—C3—C4—O8	95.0 (3)	C4—N4—C5—C6	-44.6 (5)
S2—C3—C4—N4	-83.7 (3)	C4—N4—C5—C10	134.0 (3)
N1—N2—C1—S1	1.5 (4)	C5—N4—C4—O8	-0.2 (5)
N1—N2—C1—N3	-177.3 (3)	C5—N4—C4—C3	178.6 (3)
N2—N1—C2—S1	0.0 (4)	C5—C6—C7—C11	-176.9 (2)
N2—N1—C2—S2	177.0 (3)	C5—C6—C7—C8	1.1 (5)
N4—C5—C6—C7	177.1 (3)	C6—C5—C10—C13	179.9 (2)
N4—C5—C10—C13	1.3 (4)	C6—C5—C10—C9	0.9 (5)
N4—C5—C10—C9	-177.7 (3)	C6—C7—C8—C12	-178.4 (2)
C1—S1—C2—S2	-176.4 (2)	C6—C7—C8—C9	-0.2 (5)
C1—S1—C2—N1	0.6 (3)	C7—C8—C9—C10	-0.4 (5)
C2—S1—C1—N2	-1.2 (3)	C8—C9—C10—C13	-179.0 (2)
C2—S1—C1—N3	177.5 (3)	C8—C9—C10—C5	0.0 (5)
C2—S2—C3—C4	-90.1 (2)	C10—C5—C6—C7	-1.5 (5)

Hydrogen-bond geometry (Å, °)

<i>D—H...A</i>	<i>D—H</i>	<i>H...A</i>	<i>D...A</i>	<i>D—H...A</i>
N4—H4...O8 ⁱ	0.86	2.09	2.927 (3)	163
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