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(*E*)-4-Methoxy-*N'*-(2,4,5-trifluorobenzylidene)-benzohydrazide monohydrate

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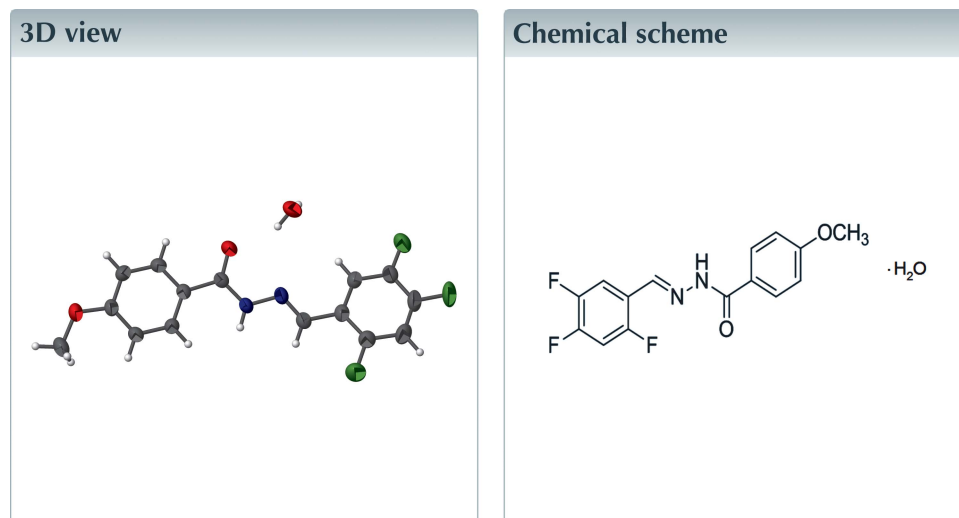
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Keywords: crystal structure; benzohydrazide; Schiff base; bifurcated hydrogen bonds; hydrogen bonding.

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Structural data: full structural data are available from iucrdata.iucr.org

The title Schiff base compound, C₁₅H₁₁F₃N₂O₂·H₂O, crystallized as a monohydrate. The conformation about the C=N bond is *E*. The molecule is almost planar, with the dihedral angle between the planes of the methoxybenzene and trifluorobenzylidene rings being 7.46 (6)°. In the crystal, molecules are linked by bifurcated O_{water}—H···(O,N) hydrogen bonds and N—H···O_{water} and O_{water}—H···O and C—H···O hydrogen bonds, forming chains along [100]. The chains are linked by C—H···O_{water} hydrogen bonds, forming slabs parallel to the *bc* plane. Within the slabs there are offset π - π interactions present [intercentroid distance = 3.7883 (7) Å].



Structure description

Benzohydrazide Schiff base compounds exhibit a wide range of biological activities and act as a promising lead compounds for the design of more efficient drugs (Negi *et al.*, 2012). Recently, benzohydrazones have attracted attention due to their versatile applications in various fields such as biology (Ibrahim *et al.*, 2016), medicine (Velezheva *et al.*, 2016) and catalysis (Selvamurugan *et al.*, 2016). These derivatives also exhibit antimicrobial (Pieczonka *et al.*, 2013), anti-proliferative (Yadagiri *et al.*, 2014) and antiplatelet (Mashayekhi *et al.*, 2013) activities.

The geometric parameters of the title molecule (Fig. 1) agree well with those reported for similar structures (Maheswari *et al.*, 2016; Nair *et al.*, 2012). The conformation about the C7=N1 bond is *E*. The dihedral angle between methoxyphenyl ring and trifluorobenzylidene ring is 7.46 (6)°.

In the crystal, molecules are linked by bifurcated O_{water}—H···(O,N) hydrogen bonds and N—H···O_{water} and O_{water}—H···O and C—H···O hydrogen bonds, forming chains

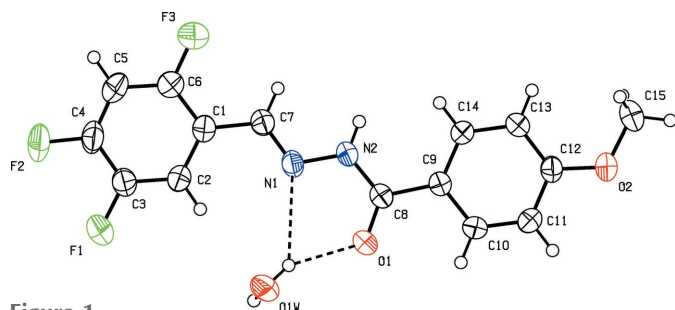


Figure 1
The molecular structure of the title compound, showing the atom labelling. Displacement ellipsoids are drawn at the 30% probability level. The bifurcated $O_{\text{water}}-H \cdots (O,N)$ hydrogen bonds are shown as dashed lines (see Table 1).

along [100]; see Table 1 and Fig. 2. The chains are linked by $C-H \cdots O_{\text{water}}$ hydrogen bonds, forming slabs parallel to the bc plane (Table 1 and Fig. 3). Within the slabs, there are offset $\pi-\pi$ interactions present [$Cg1 \cdots Cg2^i = 3.7883(3) \text{ \AA}$, interplanar distance = $3.5156(6) \text{ \AA}$, slippage = 1.746 \AA ; $Cg1$ and $Cg2$ are the centroids of rings $C1-C6$ and $C9-C14$, respectively; symmetry code: (i) $-x + 1, -y - 1, -z + 1$].

Synthesis and crystallization

To mixture of 4-methoxybenzohydrazide (1.6 g, 0.01 mol) and 2,4,5-trifluorobenzaldehyde (1.6 ml, 0.01 mol) in ethanol

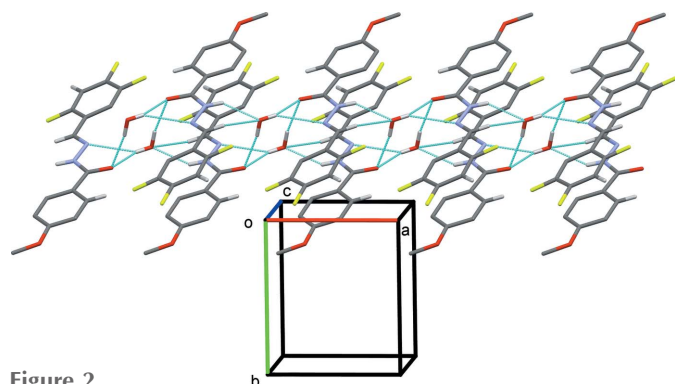


Figure 2
A partial view along the c axis of the crystal packing of the title compound. The hydrogen bonds (see Table 1) are shown as dashed lines and, for clarity, only the H atoms involved in these interactions are included.

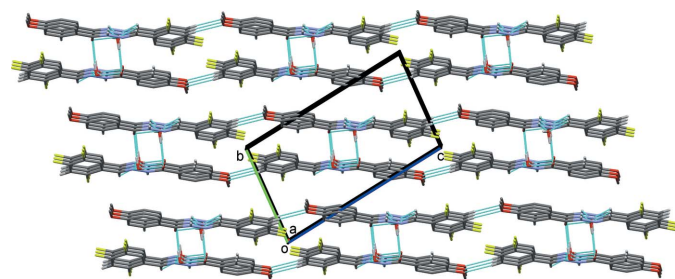


Figure 3
A view along the a axis of the crystal packing of the title compound. The hydrogen bonds (see Table 1) are shown as dashed lines and, for clarity, only the H atoms involved in these interactions are included.

Table 1
Hydrogen-bond geometry ($\text{\AA}, ^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$O1W-H1WA \cdots O1$	0.85 (2)	2.05 (2)	2.8376 (13)	154 (2)
$O1W-H1WA \cdots N1$	0.85 (2)	2.61 (2)	3.2777 (14)	137 (2)
$N2-H2 \cdots O1W^i$	0.86	2.12	2.8681 (14)	145
$O1W-H1WB \cdots O1^{ii}$	0.86 (2)	2.01 (2)	2.8635 (16)	170 (2)
$C5-H5 \cdots O2^{iii}$	0.93	2.49	3.4128 (15)	172
$C7-H7 \cdots O1W^i$	0.93	2.54	3.2191 (15)	130

Symmetry codes: (i) $x - 1, y, z$; (ii) $-x + 2, -y - 1, -z + 1$; (iii) $x, y - 1, z + 1$.

Table 2
Experimental details.

Crystal data	
Chemical formula	$C_{15}H_{11}F_3N_2O_2 \cdot H_2O$
M_r	326.27
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	296
a, b, c (\AA)	6.6872 (2), 7.9309 (2), 13.9723 (4)
α, β, γ ($^\circ$)	82.348 (2), 84.783 (1), 88.710 (1)
V (\AA^3)	731.35 (4)
Z	2
Radiation type	Mo $K\alpha$
μ (mm^{-1})	0.13
Crystal size (mm)	$0.19 \times 0.16 \times 0.11$
Data collection	
Diffractometer	Bruker APEXII CCD
Absorption correction	Multi-scan (<i>SADABS</i> ; Bruker, 2008)
$T_{\text{min}}, T_{\text{max}}$	0.976, 0.986
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	10841, 3036, 2559
R_{int}	0.023
$(\sin \theta/\lambda)_{\text{max}}$ (\AA^{-1})	0.629
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.036, 0.111, 1.04
No. of reflections	3036
No. of parameters	218
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ ($e \text{ \AA}^{-3}$)	0.16, -0.20

Computer programs: *APEX2* and *SAINT* (Bruker, 2008), *SHELXS97* and *SHELXL97* (Sheldrick, 2008), *PLATON* (Spek, 2009) and *Mercury* (Macrae *et al.*, 2008).

(10 ml) were added a few drops of concentrated HCl. The reaction mixture was stirred for 30 min at room temperature. The insoluble solid that gradually formed was filtered and washed with petroleum ether and dried in a vacuum desiccator. The crude solid was recrystallized from DMSO solution, giving colourless block-like crystals (yield 96%).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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full crystallographic data

IUCrData (2016). **1**, x160846 [doi:10.1107/S2414314616008464]

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(*E*)-4-Methoxy-*N'*-(2,4,5-trifluorobenzylidene)benzohydrazide monohydrate*Crystal data*

$C_{15}H_{11}F_3N_2O_2 \cdot H_2O$

$M_r = 326.27$

Triclinic, $P\bar{1}$

$a = 6.6872$ (2) Å

$b = 7.9309$ (2) Å

$c = 13.9723$ (4) Å

$\alpha = 82.348$ (2)°

$\beta = 84.783$ (1)°

$\gamma = 88.710$ (1)°

$V = 731.35$ (4) Å³

$Z = 2$

$F(000) = 336$

$D_x = 1.482$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 3036 reflections

$\theta = 1.5$ – 26.6 °

$\mu = 0.13$ mm⁻¹

$T = 296$ K

Block, colourless

$0.19 \times 0.16 \times 0.11$ mm

Data collection

Bruker APEXII CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 0 pixels mm⁻¹

ω and ϕ scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2008)

$T_{\min} = 0.976$, $T_{\max} = 0.986$

10841 measured reflections

3036 independent reflections

2559 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.023$

$\theta_{\max} = 26.6$ °, $\theta_{\min} = 1.5$ °

$h = -8 \rightarrow 8$

$k = -9 \rightarrow 9$

$l = -17 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.036$

$wR(F^2) = 0.111$

$S = 1.04$

3036 reflections

218 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier

map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0587P)^2 + 0.1124P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.16$ e Å⁻³

$\Delta\rho_{\min} = -0.20$ e Å⁻³

Extinction correction: *SHELXL97* (Sheldrick,
2008), $F_c^* = kFc[1 + 0.001xFe^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.105 (7)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O2	0.27163 (14)	0.17533 (13)	0.17677 (6)	0.0534 (3)

C15	0.0702 (2)	0.23667 (19)	0.18294 (11)	0.0595 (4)
H15A	-0.0200	0.1425	0.1898	0.089*
H15B	0.0478	0.3119	0.1251	0.089*
H15C	0.0467	0.2971	0.2381	0.089*
O1W	1.07816 (14)	-0.40072 (17)	0.59527 (7)	0.0567 (3)
C9	0.50245 (16)	-0.14171 (14)	0.40181 (8)	0.0360 (3)
N2	0.49491 (14)	-0.31489 (12)	0.55853 (7)	0.0406 (2)
H2	0.3684	-0.2927	0.5657	0.049*
C8	0.60665 (17)	-0.25169 (14)	0.47707 (8)	0.0383 (3)
N1	0.58713 (15)	-0.41549 (12)	0.62992 (7)	0.0412 (3)
C2	0.75138 (18)	-0.64277 (15)	0.77805 (8)	0.0428 (3)
H2A	0.8315	-0.6199	0.7199	0.051*
C12	0.33642 (18)	0.07140 (15)	0.25356 (8)	0.0394 (3)
C1	0.56035 (18)	-0.56972 (15)	0.78779 (8)	0.0401 (3)
C14	0.30309 (17)	-0.08984 (16)	0.41155 (8)	0.0415 (3)
H14	0.2240	-0.1272	0.4683	0.050*
C13	0.21914 (17)	0.01643 (16)	0.33850 (8)	0.0426 (3)
H13	0.0854	0.0506	0.3464	0.051*
C11	0.53540 (19)	0.01912 (17)	0.24265 (9)	0.0471 (3)
H11	0.6139	0.0549	0.1854	0.057*
O1	0.78818 (13)	-0.28143 (12)	0.46533 (7)	0.0538 (3)
F2	0.77810 (17)	-0.89156 (14)	1.01158 (6)	0.0863 (4)
C10	0.61712 (17)	-0.08505 (16)	0.31573 (9)	0.0437 (3)
H10	0.7512	-0.1184	0.3077	0.052*
C7	0.47776 (18)	-0.45989 (15)	0.70817 (8)	0.0420 (3)
H7	0.3451	-0.4224	0.7147	0.050*
F1	1.00444 (13)	-0.82074 (12)	0.84455 (6)	0.0698 (3)
C3	0.82137 (19)	-0.74793 (17)	0.85365 (9)	0.0470 (3)
F3	0.26498 (14)	-0.53644 (14)	0.88906 (7)	0.0779 (3)
C4	0.7048 (2)	-0.78435 (19)	0.93994 (9)	0.0552 (4)
C6	0.4494 (2)	-0.60790 (18)	0.87636 (9)	0.0503 (3)
C5	0.5175 (2)	-0.7150 (2)	0.95292 (9)	0.0613 (4)
H5	0.4386	-0.7390	1.0113	0.074*
H1WA	0.967 (3)	-0.372 (2)	0.5732 (14)	0.079 (5)*
H1WB	1.104 (3)	-0.500 (3)	0.5787 (16)	0.098 (8)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O2	0.0504 (5)	0.0632 (6)	0.0421 (5)	0.0077 (4)	-0.0084 (4)	0.0111 (4)
C15	0.0544 (8)	0.0629 (9)	0.0585 (8)	0.0126 (6)	-0.0161 (6)	0.0071 (7)
O1W	0.0366 (5)	0.0764 (8)	0.0573 (6)	0.0056 (5)	-0.0114 (4)	-0.0054 (5)
C9	0.0356 (6)	0.0366 (6)	0.0357 (6)	0.0019 (4)	-0.0055 (4)	-0.0028 (5)
N2	0.0345 (5)	0.0445 (5)	0.0405 (5)	0.0045 (4)	-0.0074 (4)	0.0048 (4)
C8	0.0359 (6)	0.0383 (6)	0.0404 (6)	0.0040 (4)	-0.0060 (4)	-0.0029 (5)
N1	0.0417 (5)	0.0398 (5)	0.0412 (5)	0.0021 (4)	-0.0105 (4)	0.0030 (4)
C2	0.0453 (6)	0.0443 (6)	0.0368 (6)	-0.0012 (5)	-0.0033 (5)	0.0016 (5)
C12	0.0425 (6)	0.0401 (6)	0.0351 (6)	0.0006 (5)	-0.0088 (5)	0.0001 (5)

C1	0.0448 (6)	0.0382 (6)	0.0372 (6)	-0.0027 (5)	-0.0077 (5)	-0.0014 (5)
C14	0.0379 (6)	0.0484 (7)	0.0351 (6)	0.0045 (5)	0.0008 (4)	0.0019 (5)
C13	0.0348 (6)	0.0497 (7)	0.0417 (6)	0.0070 (5)	-0.0039 (5)	-0.0006 (5)
C11	0.0410 (6)	0.0597 (8)	0.0368 (6)	-0.0013 (5)	0.0009 (5)	0.0047 (5)
O1	0.0367 (5)	0.0669 (6)	0.0524 (5)	0.0125 (4)	-0.0029 (4)	0.0091 (4)
F2	0.1011 (8)	0.1021 (8)	0.0465 (5)	0.0264 (6)	-0.0128 (5)	0.0238 (5)
C10	0.0324 (6)	0.0540 (7)	0.0423 (6)	0.0037 (5)	-0.0005 (5)	-0.0003 (5)
C7	0.0403 (6)	0.0439 (6)	0.0412 (6)	0.0013 (5)	-0.0076 (5)	-0.0011 (5)
F1	0.0587 (5)	0.0800 (6)	0.0655 (5)	0.0242 (4)	-0.0089 (4)	0.0075 (4)
C3	0.0487 (7)	0.0468 (7)	0.0449 (7)	0.0052 (5)	-0.0089 (5)	-0.0012 (5)
F3	0.0581 (5)	0.1054 (8)	0.0624 (5)	0.0235 (5)	0.0085 (4)	0.0039 (5)
C4	0.0698 (9)	0.0573 (8)	0.0361 (6)	0.0064 (7)	-0.0121 (6)	0.0061 (6)
C6	0.0475 (7)	0.0582 (8)	0.0434 (7)	0.0048 (6)	-0.0013 (5)	-0.0027 (6)
C5	0.0686 (9)	0.0751 (10)	0.0351 (6)	0.0047 (7)	0.0039 (6)	0.0052 (6)

Geometric parameters (Å, °)

O2—C12	1.3606 (13)	C12—C11	1.3860 (17)
O2—C15	1.4210 (16)	C12—C13	1.3863 (16)
C15—H15A	0.9600	C1—C6	1.3846 (17)
C15—H15B	0.9600	C1—C7	1.4605 (15)
C15—H15C	0.9600	C14—C13	1.3861 (15)
O1W—H1WA	0.84 (2)	C14—H14	0.9300
O1W—H1WB	0.86 (2)	C13—H13	0.9300
C9—C14	1.3865 (16)	C11—C10	1.3713 (17)
C9—C10	1.3930 (16)	C11—H11	0.9300
C9—C8	1.4879 (15)	F2—C4	1.3451 (15)
N2—C8	1.3473 (15)	C10—H10	0.9300
N2—N1	1.3747 (13)	C7—H7	0.9300
N2—H2	0.8600	F1—C3	1.3445 (15)
C8—O1	1.2318 (14)	C3—C4	1.3754 (19)
N1—C7	1.2720 (15)	F3—C6	1.3522 (16)
C2—C3	1.3668 (16)	C4—C5	1.363 (2)
C2—C1	1.3936 (17)	C6—C5	1.3794 (18)
C2—H2A	0.9300	C5—H5	0.9300
C12—O2—C15	118.76 (10)	C13—C14—C9	121.42 (10)
O2—C15—H15A	109.5	C13—C14—H14	119.3
O2—C15—H15B	109.5	C9—C14—H14	119.3
H15A—C15—H15B	109.5	C14—C13—C12	119.40 (11)
O2—C15—H15C	109.5	C14—C13—H13	120.3
H15A—C15—H15C	109.5	C12—C13—H13	120.3
H15B—C15—H15C	109.5	C10—C11—C12	120.36 (11)
H1WA—O1W—H1WB	104.7 (19)	C10—C11—H11	119.8
C14—C9—C10	118.11 (10)	C12—C11—H11	119.8
C14—C9—C8	125.20 (10)	C11—C10—C9	121.02 (11)
C10—C9—C8	116.68 (10)	C11—C10—H10	119.5
C8—N2—N1	118.70 (9)	C9—C10—H10	119.5

C8—N2—H2	120.6	N1—C7—C1	120.07 (11)
N1—N2—H2	120.6	N1—C7—H7	120.0
O1—C8—N2	121.75 (10)	C1—C7—H7	120.0
O1—C8—C9	120.98 (10)	F1—C3—C2	120.63 (11)
N2—C8—C9	117.27 (10)	F1—C3—C4	118.54 (11)
C7—N1—N2	115.79 (10)	C2—C3—C4	120.81 (12)
C3—C2—C1	120.19 (11)	F2—C4—C5	120.24 (12)
C3—C2—H2A	119.9	F2—C4—C3	118.74 (13)
C1—C2—H2A	119.9	C5—C4—C3	121.01 (12)
O2—C12—C11	115.34 (10)	F3—C6—C5	118.01 (12)
O2—C12—C13	124.96 (11)	F3—C6—C1	118.64 (12)
C11—C12—C13	119.69 (10)	C5—C6—C1	123.34 (13)
C6—C1—C2	117.04 (11)	C4—C5—C6	117.59 (12)
C6—C1—C7	120.95 (11)	C4—C5—H5	121.2
C2—C1—C7	122.01 (11)	C6—C5—H5	121.2
N1—N2—C8—O1	-0.19 (18)	C14—C9—C10—C11	0.03 (18)
N1—N2—C8—C9	-179.31 (9)	C8—C9—C10—C11	-178.94 (11)
C14—C9—C8—O1	-174.51 (12)	N2—N1—C7—C1	178.74 (9)
C10—C9—C8—O1	4.38 (17)	C6—C1—C7—N1	174.55 (12)
C14—C9—C8—N2	4.62 (18)	C2—C1—C7—N1	-6.50 (18)
C10—C9—C8—N2	-176.49 (11)	C1—C2—C3—F1	179.14 (11)
C8—N2—N1—C7	174.47 (11)	C1—C2—C3—C4	0.5 (2)
C15—O2—C12—C11	-179.01 (12)	F1—C3—C4—F2	-0.2 (2)
C15—O2—C12—C13	1.29 (19)	C2—C3—C4—F2	178.51 (12)
C3—C2—C1—C6	0.47 (18)	F1—C3—C4—C5	-179.57 (13)
C3—C2—C1—C7	-178.52 (11)	C2—C3—C4—C5	-0.9 (2)
C10—C9—C14—C13	-0.51 (18)	C2—C1—C6—F3	178.65 (12)
C8—C9—C14—C13	178.37 (11)	C7—C1—C6—F3	-2.35 (19)
C9—C14—C13—C12	0.39 (19)	C2—C1—C6—C5	-1.1 (2)
O2—C12—C13—C14	179.90 (11)	C7—C1—C6—C5	177.93 (13)
C11—C12—C13—C14	0.21 (19)	F2—C4—C5—C6	-179.08 (14)
O2—C12—C11—C10	179.60 (11)	C3—C4—C5—C6	0.3 (2)
C13—C12—C11—C10	-0.68 (19)	F3—C6—C5—C4	-179.02 (13)
C12—C11—C10—C9	0.6 (2)	C1—C6—C5—C4	0.7 (2)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1 <i>W</i> —H1 <i>WA</i> ⋯O1	0.85 (2)	2.05 (2)	2.8376 (13)	154 (2)
O1 <i>W</i> —H1 <i>WA</i> ⋯N1	0.85 (2)	2.61 (2)	3.2777 (14)	137 (2)
N2—H2⋯O1 <i>W</i> ⁱ	0.86	2.12	2.8681 (14)	145
O1 <i>W</i> —H1 <i>WB</i> ⋯O1 ⁱⁱ	0.86 (2)	2.01 (2)	2.8635 (16)	170 (2)
C5—H5⋯O2 ⁱⁱⁱ	0.93	2.49	3.4128 (15)	172
C7—H7⋯O1 <i>W</i> ⁱ	0.93	2.54	3.2191 (15)	130

Symmetry codes: (i) $x-1, y, z$; (ii) $-x+2, -y-1, -z+1$; (iii) $x, y-1, z+1$.