



ISSN 2414-3146

4-Benzyloxy-1,1'-biphenyl

Farook Adam,^{a*} Nadiah Ameram,^a Shevgoor Dhiraj Kamath,^b Pallavi^c and Seranthimata Samshuddin^c

^aSchool of Chemical Sciences, Universiti Sains Malaysia, 11800 Pulau Pinang, Malaysia, ^bDepartment of P.G. Studies in Analytical Chemistry, Alva's College, Moodbidri, Karnataka 574 227, India, and ^cDepartment of P.G. Studies in Chemistry, Alva's College, Moodbidri, Karnataka 574 227, India. *Correspondence e-mail: farook@usm.my

Received 24 February 2016

Accepted 9 March 2016

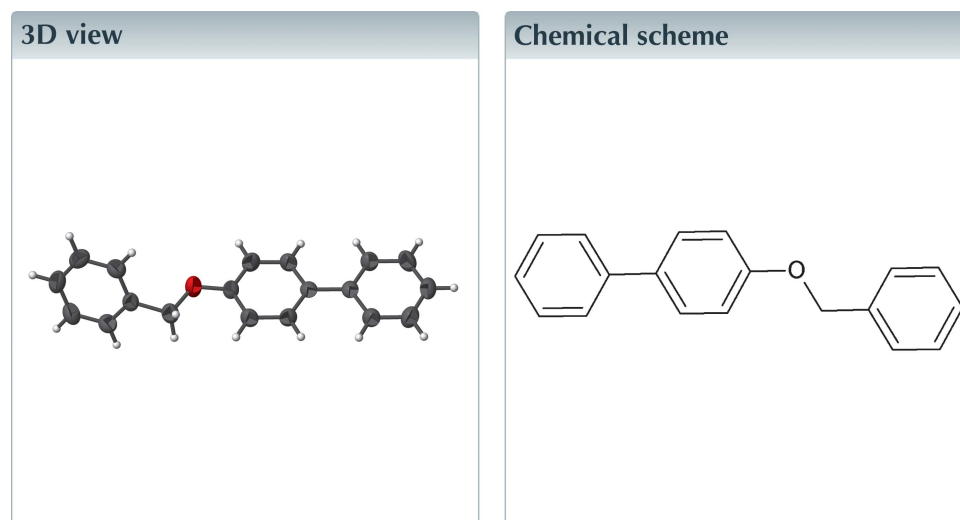
Edited by W. T. A. Harrison, University of Aberdeen, Scotland

Keywords: crystal structure; aryl ether; synthesis.

CCDC reference: 1460794

Structural data: full structural data are available from iucrdata.iucr.org

In the title compound, C₁₉H₁₆O, the dihedral angle between the benzene rings of the biphenyl unit is 1.54 (13)° and the C—O—C—C torsion angle is 174.4 (2)°. In the crystal, very weak C—H···π interactions link the molecules into a three-dimensional network.



Structure description

Many diaryl ethers exhibit various pharmacological properties including anti-bacterial, anti-inflammatory, antifungal and herbicidal activities (Ley & Thomas, 2003; Frlan & Kikelj, 2006). The crystal structures of some aryl ethers *viz.*, 2,4-dichloro-1-[1-(2,4-dichlorobenzyloxy)ethyl]benzene (Jasinski *et al.*, 2010) and 2,6-bis[2-(4-benzyloxyphenyl)ethyl]biphenyl (Moratti *et al.*, 2007) have been reported. As part of our studies in this area, the synthesis and structure of the title compound are reported.

The title molecule (Fig. 1) consists of three benzene rings, C1–C6 (*A*), C7–C12 (*B*) and C14–C19 (*C*). The dihedral angles *A/B*, *A/C* and *B/C* are 1.54 (13), 61.50 (14) and 62.80 (14)°, respectively. Five weak C—H···π interactions are observed in the crystal structure (Table 1). The crystal packing is illustrated in Fig. 2.

Synthesis and crystallization

A mixture of (1,1'-biphenyl)-4-ol (1.70 g, 0.01 mol) and benzyl chloride (5 ml) was refluxed for 30 min. The reaction mixture was cooled and poured into 25 ml hexane. The precipitate was collected by filtration and purified by recrystallization from ethanol. Single crystals were grown from 1,4-dioxane solution by the slow evaporation method; m.p. 421–425 K, yield 85%.

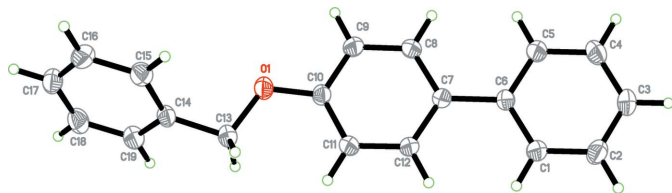


Figure 1
The molecular structure of the title compound, showing 50% probability displacement ellipsoids.

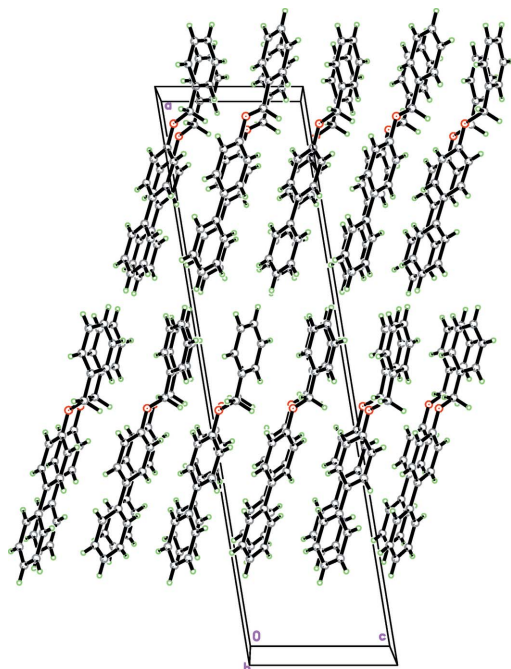


Figure 2
The crystal packing, viewed along the *b* axis.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

SS and SDK thank Alva's Education Foundation, Moodbidri, for providing research facilities. FA is grateful for USM research grants 1001/PKIMIA/846017 and 1001/PKIMIA/811269, which partially supported this research.

References

Bruker (2010). *APEX2*, *SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.

Table 1
Hydrogen-bond geometry (Å, °).

*Cg*1, *Cg*2 and *Cg*3 are the centroids of the C1–C6, C7–C12 and C14–C19 benzene rings, respectively.

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C1–H1A··· <i>Cg</i> 1 ⁱ	0.93	2.99	3.700 (3)	134
C4–H4A··· <i>Cg</i> 1 ⁱⁱ	0.93	2.97	3.700 (3)	134
C8–H8A··· <i>Cg</i> 2 ⁱⁱ	0.93	2.96	3.687 (3)	137
C11–H11A··· <i>Cg</i> 2 ⁱ	0.93	2.93	3.650 (3)	135
C19–H19A··· <i>Cg</i> 3 ⁱⁱⁱ	0.93	2.90	3.591 (3)	133

Symmetry codes: (i) $x, -y + 1, z + \frac{1}{2}$; (ii) $x, -y, z - \frac{1}{2}$; (iii) $x, -y + 1, z - \frac{1}{2}$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₉ H ₁₆ O
<i>M_r</i>	260.32
Crystal system, space group	Monoclinic, <i>Cc</i>
Temperature (K)	296
<i>a</i> , <i>b</i> , <i>c</i> (Å)	31.270 (2), 5.6720 (4), 7.8812 (5)
β (°)	99.271 (3)
<i>V</i> (Å ³)	1379.58 (16)
<i>Z</i>	4
Radiation type	Mo <i>K</i> α
μ (mm ⁻¹)	0.08
Crystal size (mm)	0.39 × 0.31 × 0.09
Data collection	
Diffractometer	Bruker APEXII CCD
Absorption correction	Multi-scan (<i>SADABS</i> ; Bruker, 2010)
<i>T_{min}</i> , <i>T_{max}</i>	0.891, 0.969
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	18200, 2435, 2218
<i>R_{int}</i>	0.027
(<i>sin</i> θ / λ) _{max} (Å ⁻¹)	0.595
Refinement	
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.034, 0.104, 0.87
No. of reflections	2435
No. of parameters	181
No. of restraints	2
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{max}$, $\Delta\rho_{min}$ (e Å ⁻³)	0.12, -0.12

Computer programs: *APEX2* (Bruker, 2010), *SAINT* (Bruker, 2010), *SHELXS97* (Sheldrick 2008), *SHELXL2014* (Sheldrick, 2015), *SHELXTL* (Sheldrick 2008).

Frlan, R. & Kikelj, D. (2006). *Synthesis*, pp. 2271–2285.
 Jasinski, J. P., Butcher, R. J., Chidan Kumar, C. S., Yathirajan, H. S. & Narayana, B. (2010). *Acta Cryst.* **E66**, o165–o166.
 Ley, S. V. & Thomas, A. W. (2003). *Angew. Chem. Int. Ed.* **42**, 5400–5449.
 Moratti, S. C., Simpson, J. & Tierney, S. M. (2007). *Acta Cryst.* **E63**, o3954.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Sheldrick, G. M. (2015). *Acta Cryst.* **C71**, 3–8.

full crystallographic data

IUCrData (2016). **1**, x160398 [doi:10.1107/S2414314616003989]

4-Benzyloxy-1,1'-biphenyl

Farook Adam, Nadiyah Ameram, Shevgoor Dhiraj Kamath, Pallavi and Seranthimata Samshuddin

4-Benzyloxy-1,1'-biphenyl

Crystal data

$C_{19}H_{16}O$
 $M_r = 260.32$
 Monoclinic, *Cc*
 $a = 31.270$ (2) Å
 $b = 5.6720$ (4) Å
 $c = 7.8812$ (5) Å
 $\beta = 99.271$ (3)°
 $V = 1379.58$ (16) Å³
 $Z = 4$

$F(000) = 552$
 $D_x = 1.253$ Mg m⁻³
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
 Cell parameters from 7054 reflections
 $\theta = 2.6$ – 31.7 °
 $\mu = 0.08$ mm⁻¹
 $T = 296$ K
 Plate, colourless
 $0.39 \times 0.31 \times 0.09$ mm

Data collection

Bruker APEXII CCD
 diffractometer
 φ and ω scans
 Absorption correction: multi-scan
 (*SADABS*; Bruker, 2010)
 $T_{\min} = 0.891$, $T_{\max} = 0.969$
 18200 measured reflections

2435 independent reflections
 2218 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.027$
 $\theta_{\max} = 25.0$ °, $\theta_{\min} = 2.6$ °
 $h = -36 \rightarrow 36$
 $k = -6 \rightarrow 6$
 $l = -9 \rightarrow 9$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.104$
 $S = 0.87$
 2435 reflections
 181 parameters
 2 restraints

Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0802P)^2 + 0.3664P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.12$ e Å⁻³
 $\Delta\rho_{\min} = -0.12$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.44220 (6)	0.2333 (3)	1.0798 (2)	0.0538 (5)

C1	0.23730 (9)	0.4280 (5)	0.7983 (4)	0.0517 (7)
H1A	0.2475	0.5521	0.8709	0.062*
C2	0.19447 (9)	0.4248 (5)	0.7236 (4)	0.0579 (8)
H2A	0.1761	0.5458	0.7463	0.069*
C3	0.17851 (9)	0.2446 (5)	0.6158 (4)	0.0538 (7)
H3A	0.1495	0.2416	0.5659	0.065*
C4	0.20609 (9)	0.0691 (5)	0.5831 (4)	0.0588 (8)
H4A	0.1958	-0.0529	0.5088	0.071*
C5	0.24892 (9)	0.0712 (5)	0.6590 (4)	0.0515 (7)
H5A	0.2670	-0.0512	0.6362	0.062*
C6	0.26585 (8)	0.2510 (4)	0.7683 (3)	0.0377 (6)
C7	0.31216 (8)	0.2517 (4)	0.8516 (3)	0.0366 (6)
C8	0.34084 (9)	0.0771 (5)	0.8203 (4)	0.0523 (8)
H8A	0.3309	-0.0435	0.7442	0.063*
C9	0.38322 (9)	0.0756 (5)	0.8973 (4)	0.0553 (8)
H9A	0.4013	-0.0461	0.8738	0.066*
C10	0.39943 (8)	0.2524 (4)	1.0094 (3)	0.0408 (6)
C11	0.37210 (8)	0.4295 (5)	1.0437 (3)	0.0491 (7)
H11A	0.3823	0.5502	1.1194	0.059*
C12	0.32920 (8)	0.4267 (5)	0.9646 (3)	0.0476 (7)
H12A	0.3111	0.5479	0.9886	0.057*
C13	0.45998 (9)	0.4022 (5)	1.2055 (4)	0.0512 (7)
H13A	0.4429	0.4080	1.2977	0.061*
H13B	0.4598	0.5575	1.1539	0.061*
C14	0.50538 (8)	0.3302 (5)	1.2746 (3)	0.0434 (6)
C15	0.51377 (9)	0.1223 (5)	1.3677 (4)	0.0527 (7)
H15A	0.4909	0.0250	1.3852	0.063*
C16	0.55562 (10)	0.0597 (5)	1.4339 (4)	0.0595 (8)
H16A	0.5611	-0.0792	1.4963	0.071*
C17	0.58958 (10)	0.2035 (6)	1.4075 (4)	0.0612 (8)
H17A	0.6179	0.1610	1.4516	0.073*
C18	0.58158 (9)	0.4071 (6)	1.3169 (4)	0.0587 (7)
H18A	0.6045	0.5045	1.3004	0.070*
C19	0.53974 (9)	0.4697 (5)	1.2493 (4)	0.0503 (7)
H19A	0.5347	0.6079	1.1858	0.060*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0395 (10)	0.0632 (12)	0.0561 (11)	0.0062 (8)	-0.0001 (8)	-0.0172 (9)
C1	0.0494 (17)	0.0478 (16)	0.0544 (17)	0.0041 (11)	-0.0016 (13)	-0.0105 (12)
C2	0.0472 (16)	0.0626 (18)	0.0615 (19)	0.0129 (13)	0.0016 (14)	-0.0039 (14)
C3	0.0393 (14)	0.0658 (19)	0.0537 (16)	-0.0012 (12)	-0.0006 (12)	0.0044 (13)
C4	0.0514 (18)	0.0628 (18)	0.0591 (19)	-0.0092 (14)	-0.0009 (14)	-0.0135 (14)
C5	0.0452 (16)	0.0525 (16)	0.0556 (17)	0.0035 (12)	0.0049 (13)	-0.0130 (12)
C6	0.0388 (14)	0.0411 (13)	0.0338 (13)	-0.0009 (9)	0.0077 (10)	0.0044 (9)
C7	0.0416 (14)	0.0372 (13)	0.0317 (12)	-0.0009 (9)	0.0081 (10)	0.0024 (9)
C8	0.0480 (17)	0.0464 (15)	0.0590 (18)	0.0032 (11)	-0.0024 (14)	-0.0199 (12)

C9	0.0491 (17)	0.0514 (16)	0.0630 (19)	0.0128 (12)	0.0019 (14)	-0.0180 (13)
C10	0.0373 (14)	0.0476 (14)	0.0369 (14)	0.0018 (10)	0.0043 (11)	-0.0021 (10)
C11	0.0476 (18)	0.0465 (16)	0.0510 (17)	0.0005 (11)	0.0015 (13)	-0.0140 (12)
C12	0.0429 (16)	0.0441 (14)	0.0547 (17)	0.0073 (11)	0.0041 (13)	-0.0124 (12)
C13	0.0446 (15)	0.0550 (16)	0.0513 (16)	-0.0010 (12)	0.0000 (12)	-0.0087 (12)
C14	0.0415 (14)	0.0481 (13)	0.0384 (13)	0.0008 (11)	-0.0002 (11)	-0.0059 (11)
C15	0.0520 (16)	0.0490 (14)	0.0557 (17)	-0.0058 (12)	0.0045 (13)	0.0000 (13)
C16	0.067 (2)	0.0510 (16)	0.0568 (17)	0.0114 (14)	-0.0022 (15)	0.0046 (13)
C17	0.0471 (16)	0.071 (2)	0.0611 (18)	0.0112 (14)	-0.0058 (13)	-0.0100 (16)
C18	0.0443 (16)	0.0687 (19)	0.0619 (17)	-0.0106 (14)	0.0052 (13)	-0.0074 (15)
C19	0.0522 (17)	0.0509 (14)	0.0464 (14)	-0.0053 (12)	0.0041 (12)	0.0033 (12)

Geometric parameters (Å, °)

O1—C10	1.367 (3)	C9—H9A	0.9300
O1—C13	1.425 (3)	C10—C11	1.374 (4)
C1—C2	1.374 (4)	C11—C12	1.386 (4)
C1—C6	1.389 (4)	C11—H11A	0.9300
C1—H1A	0.9300	C12—H12A	0.9300
C2—C3	1.370 (4)	C13—C14	1.493 (4)
C2—H2A	0.9300	C13—H13A	0.9700
C3—C4	1.369 (4)	C13—H13B	0.9700
C3—H3A	0.9300	C14—C19	1.374 (4)
C4—C5	1.376 (4)	C14—C15	1.391 (4)
C4—H4A	0.9300	C15—C16	1.375 (4)
C5—C6	1.384 (4)	C15—H15A	0.9300
C5—H5A	0.9300	C16—C17	1.381 (5)
C6—C7	1.491 (3)	C16—H16A	0.9300
C7—C12	1.382 (4)	C17—C18	1.359 (4)
C7—C8	1.384 (4)	C17—H17A	0.9300
C8—C9	1.366 (4)	C18—C19	1.378 (4)
C8—H8A	0.9300	C18—H18A	0.9300
C9—C10	1.378 (4)	C19—H19A	0.9300
C10—O1—C13	118.39 (19)	C10—C11—C12	119.4 (2)
C2—C1—C6	121.7 (3)	C10—C11—H11A	120.3
C2—C1—H1A	119.1	C12—C11—H11A	120.3
C6—C1—H1A	119.1	C7—C12—C11	122.9 (2)
C3—C2—C1	120.6 (3)	C7—C12—H12A	118.6
C3—C2—H2A	119.7	C11—C12—H12A	118.6
C1—C2—H2A	119.7	O1—C13—C14	108.1 (2)
C4—C3—C2	118.8 (3)	O1—C13—H13A	110.1
C4—C3—H3A	120.6	C14—C13—H13A	110.1
C2—C3—H3A	120.6	O1—C13—H13B	110.1
C3—C4—C5	120.7 (3)	C14—C13—H13B	110.1
C3—C4—H4A	119.7	H13A—C13—H13B	108.4
C5—C4—H4A	119.7	C19—C14—C15	118.6 (2)
C4—C5—C6	121.7 (2)	C19—C14—C13	120.6 (2)

C4—C5—H5A	119.2	C15—C14—C13	120.8 (2)
C6—C5—H5A	119.2	C16—C15—C14	120.4 (3)
C5—C6—C1	116.5 (2)	C16—C15—H15A	119.8
C5—C6—C7	121.5 (2)	C14—C15—H15A	119.8
C1—C6—C7	121.9 (2)	C15—C16—C17	119.9 (3)
C12—C7—C8	115.8 (2)	C15—C16—H16A	120.1
C12—C7—C6	122.1 (2)	C17—C16—H16A	120.1
C8—C7—C6	122.1 (2)	C18—C17—C16	120.0 (3)
C9—C8—C7	122.4 (2)	C18—C17—H17A	120.0
C9—C8—H8A	118.8	C16—C17—H17A	120.0
C7—C8—H8A	118.8	C17—C18—C19	120.3 (3)
C8—C9—C10	120.7 (2)	C17—C18—H18A	119.8
C8—C9—H9A	119.7	C19—C18—H18A	119.8
C10—C9—H9A	119.7	C14—C19—C18	120.8 (3)
O1—C10—C11	125.2 (2)	C14—C19—H19A	119.6
O1—C10—C9	115.9 (2)	C18—C19—H19A	119.6
C11—C10—C9	118.8 (2)		
C6—C1—C2—C3	-0.1 (5)	C8—C9—C10—C11	0.6 (4)
C1—C2—C3—C4	-0.5 (5)	O1—C10—C11—C12	-179.4 (2)
C2—C3—C4—C5	1.0 (5)	C9—C10—C11—C12	-0.3 (4)
C3—C4—C5—C6	-1.0 (5)	C8—C7—C12—C11	-0.4 (4)
C4—C5—C6—C1	0.4 (4)	C6—C7—C12—C11	179.7 (2)
C4—C5—C6—C7	179.4 (2)	C10—C11—C12—C7	0.2 (4)
C2—C1—C6—C5	0.1 (4)	C10—O1—C13—C14	174.4 (2)
C2—C1—C6—C7	-178.9 (2)	O1—C13—C14—C19	115.6 (3)
C5—C6—C7—C12	-178.2 (3)	O1—C13—C14—C15	-65.1 (3)
C1—C6—C7—C12	0.7 (3)	C19—C14—C15—C16	0.6 (4)
C5—C6—C7—C8	1.9 (3)	C13—C14—C15—C16	-178.7 (2)
C1—C6—C7—C8	-179.2 (3)	C14—C15—C16—C17	-0.2 (4)
C12—C7—C8—C9	0.7 (4)	C15—C16—C17—C18	0.3 (4)
C6—C7—C8—C9	-179.4 (3)	C16—C17—C18—C19	-0.8 (5)
C7—C8—C9—C10	-0.8 (5)	C15—C14—C19—C18	-1.0 (4)
C13—O1—C10—C11	3.3 (4)	C13—C14—C19—C18	178.2 (3)
C13—O1—C10—C9	-175.8 (3)	C17—C18—C19—C14	1.1 (4)
C8—C9—C10—O1	179.7 (3)		

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1, Cg2 and Cg3 are the centroids of the C1—C6, C7—C12 and C14—C19 benzene rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C1—H1A \cdots Cg1 ⁱ	0.93	2.99	3.700 (3)	134
C4—H4A \cdots Cg1 ⁱⁱ	0.93	2.97	3.700 (3)	134
C8—H8A \cdots Cg2 ⁱⁱ	0.93	2.96	3.687 (3)	137
C11—H11A \cdots Cg2 ⁱ	0.93	2.93	3.650 (3)	135
C19—H19A \cdots Cg3 ⁱⁱⁱ	0.93	2.90	3.591 (3)	133

Symmetry codes: (i) $x, -y+1, z+1/2$; (ii) $x, -y, z-1/2$; (iii) $x, -y+1, z-1/2$.