



ISSN 2414-3146

Tetrakis(μ_2 -diphenylphosphinato- $\kappa^2 O, O'$)tetra- μ_3 -oxido-tetraoxido-hexamolybdenum(V)

Sara Raquel Mota Merelo de Aguiar,^a Berthold Stöger,^b Matthias Weil^{b*} and Karl Kirchner^a

^aInstitute of Applied Synthetic Chemistry, TU Wien, Getreidemarkt 9/163, A-1060 Vienna, Austria, and ^bInstitute for Chemical Technologies and Analytics, Division of Structural Chemistry, TU Wien, Getreidemarkt 9/164-SC, A-1060 Vienna, Austria. *Correspondence e-mail: Matthias.Weil@tuwien.ac.at

Received 5 January 2016

Accepted 7 January 2016

Edited by J. Simpson, University of Otago, New Zealand

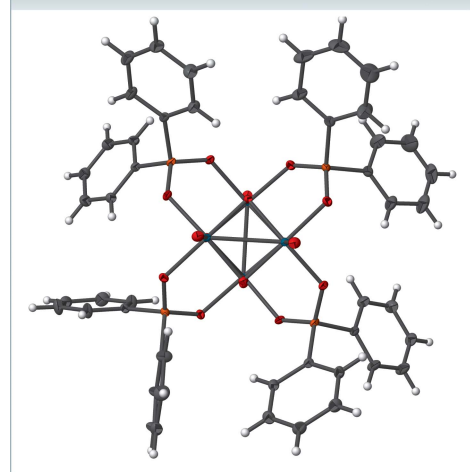
Keywords: crystal structure; heterocubane; molybdenum; Mo₄O₄ core.

CCDC reference: 1446068

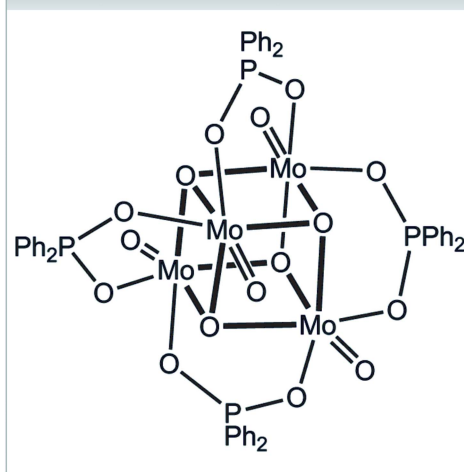
Structural data: full structural data are available from iucrdata.iucr.org

The molecule of the title compound, [Mo₄(μ_2 -C₁₂H₁₀OP₂)₄(μ_3 -O)₄O₄], exhibits point group symmetry 2 with the twofold rotation axis passing through two opposite P atoms. Each Mo^V atom is bridged by three O atoms resulting in an Mo₄O₄ heterocubane core. In the crystal, weak C—H...O interactions may help to consolidate packing of the molecules.

3D view



Chemical scheme



Structure description

The title compound had formed accidentally due to the presence of oxygen in a solution originally intended to crystallize the PNP pincer compound [Mo(PNP^{Me}—Ph)(CO)₂F₂] [(PNP^{Me}—Ph) = *N,N'*-bis(diphenylphosphino)-*N,N'*-methyl-2,6-diaminopyridine; de Aguiar *et al.*, 2014]. A chloroform disolvate of the title compound was reported by Schirmer *et al.* (1989). The molecule of the title compound [MoO(μ_3 -O)(μ_2 -C₁₂H₁₀PO₂)₄], exhibits point group symmetry 2 with the twofold rotation axis passing through two opposite P atoms (Fig. 1). Each molybdenum atom is bridged by three oxygen atoms resulting in a Mo₄O₄ heterocubane core. The distorted octahedral coordination spheres of the two unique molybdenum atoms [bond lengths range from 1.6686 (19) to 2.3982 (19) Å] are completed by a double-bonded terminal oxygen atom and two oxygen atoms of two diphenylphosphinato anions, each bridging two opposite molybdenum atoms in the heterocube. The short Mo...Mo distance of 2.6395 (3) Å indicates Mo^V...Mo^V interactions and causes a distortion of the Mo₄O₄ heterocubane core with O—Mo—O angles ranging from 77.94 (7) to 89.87 (7)° and Mo—O—Mo angles from 84.60 (7) to 102.02 (7)°. For characteristic bond lengths and angles of {Mo₄O₄(μ_3 -

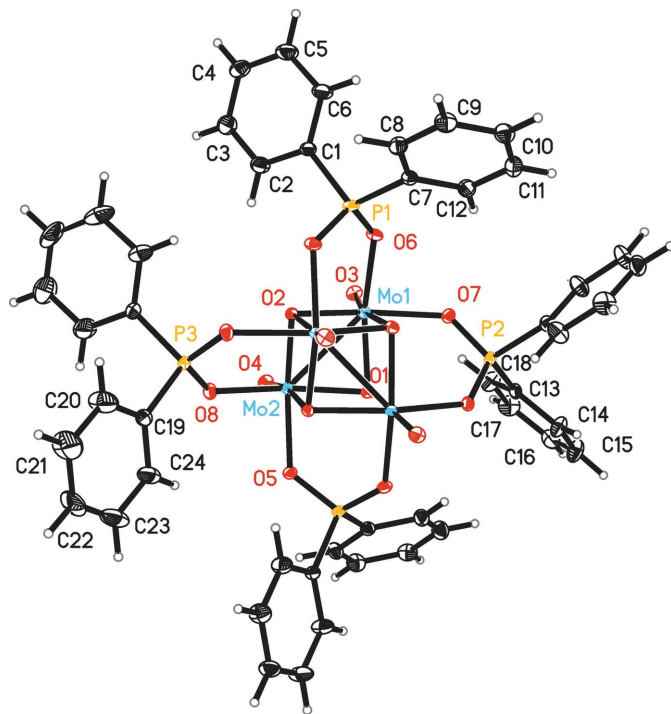


Figure 1
The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level; H atoms are shown as spheres of arbitrary radius. Non-labelled atoms are generated by the symmetry code $-x + 2, y, -z + \frac{1}{2}$.

$O_4\}^{4+}$ heterocubane cores, see: Modéc *et al.* (2003). In the crystal, weak C—H...O interactions, Table 1, may help to consolidate packing of the molecules, Fig. 2.

Synthesis and crystallization

The original intention was to crystallize the compound $[Mo(PNP^{Me}-Ph)(CO)_2F_2]$ (de Aguiar *et al.*, 2014). One

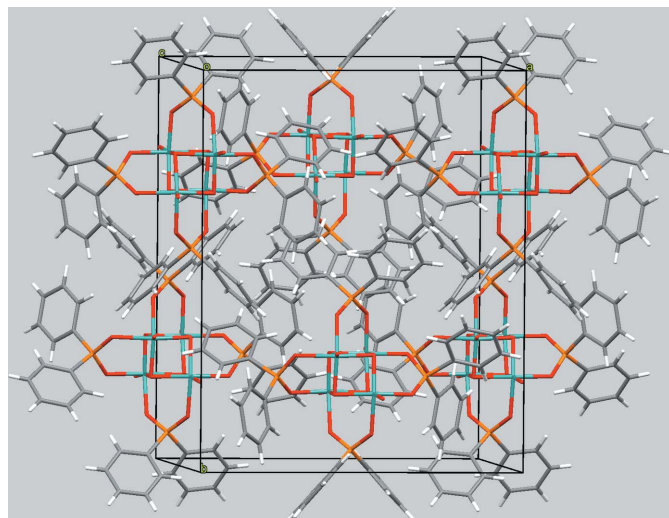


Figure 2
The packing of the molecules in the crystal structure of the title compound in a view approximately along [001].

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C4-H4\cdots O3^i$	0.93	2.52	3.188 (4)	129
$C21-H21\cdots O4^{ii}$	0.93	2.57	3.420 (5)	152

Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{1}{2}, -z$; (ii) $x, -y, z + \frac{1}{2}$.

equivalent of 1-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate was added to a solution of $[Mo(PNP^{Me}-Ph)(CO)_3]$ in 10 ml CH_2Cl_2 and stirred for three days. After filtration, the solution was layered with pentane and left to stand for several days. Instead of the desired compound $[Mo(PNP^{Me}-Ph)(CO)_2F_2]$, the title compound crystallized in form of dark-red blocks, apparently caused by the presence of larger amounts of oxygen in the reaction vessel.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

Financial support by the Austrian Science Fund (FWF) (Project No. P24202-N17) is gratefully acknowledged. The

Table 2
Experimental details.

Crystal data	
Chemical formula	$[Mo_4(C_{12}H_{10}O_2P)_4O_8]$
M_r	1380.44
Crystal system, space group	Orthorhombic, $Pbcn$
Temperature (K)	100
a, b, c (\AA)	16.1061 (19), 19.891 (2), 15.254 (3)
V (\AA^3)	4887.0 (11)
Z	4
Radiation type	Mo $K\alpha$
μ (mm^{-1})	1.21
Crystal size (mm)	$0.61 \times 0.49 \times 0.30$
Data collection	
Diffractometer	Bruker Kappa APEXII CCD diffractometer
Absorption correction	Multi-scan (SADABS; Bruker, 2014)
T_{\min}, T_{\max}	0.45, 0.66
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	255304, 8962, 7558
R_{int}	0.061
$(\sin \theta/\lambda)_{\text{max}}$ (\AA^{-1})	0.760
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.039, 0.090, 1.27
No. of reflections	8962
No. of parameters	326
H-atom treatment	H-atom parameters constrained
	$w = 1/[\sigma^2(F_o^2) + 23.6442P]$ where $P = (F_o^2 + 2F_c^2)/3$
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ ($e \text{\AA}^{-3}$)	0.92, -1.05

Computer programs: APEX2 (Bruker, 2014), SAINT (Bruker, 2014), SUPERFLIP (Palatinus & Chapuis, 2007), SHELXL2014 (Sheldrick, 2015), XP in SHELXTL (Sheldrick, 2008), Mercury (Macrae *et al.*, 2006), publCIF (Westrip, 2010).

X-ray centre of TU Wien is acknowledged for providing access to the single-crystal diffractometer.

References

- Aguiar, S. R. M. M. de, Stöger, B., Pittenauer, E., Puchberger, M., Allmaier, G., Veiros, L. F. & Kirchner, K. (2014). *J. Organomet. Chem.* **760**, 74–83.
- Bruker (2014). *APEX2*, *SAINTE* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Macrae, C. F., Edgington, P. R., McCabe, P., Pidcock, E., Shields, G. P., Taylor, R., Towler, M. & van de Streek, J. (2006). *J. Appl. Cryst.* **39**, 453–457.
- Modec, B., Brenčič, J., Burkholder, E. M. & Zubieta, J. (2003). *Dalton Trans.* pp. 4618–4625.
- Palatinus, L. & Chapuis, G. (2007). *J. Appl. Cryst.* **40**, 786–790.
- Schirmer, W., Flörke, U. & Haupt, H.-J. (1989). *Z. Anorg. Allg. Chem.* **574**, 239–255.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Sheldrick, G. M. (2015). *Acta Cryst.* **C71**, 3–8.
- Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

full crystallographic data

IUCrData (2016). **1**, x160036 [doi:10.1107/S2414314616000365]

Tetrakis(μ_2 -diphenylphosphinato- κ^2O,O')tetra- μ_3 -oxido-tetraoxidohexamolybdenum(V)

Sara Raquel Mota Merelo de Aguiar, Berthold Stöger, Matthias Weil and Karl Kirchner

Tetrakis(μ_2 -diphenylphosphinato- κ^2O,O')tetra- μ_3 -oxido-tetraoxidohexamolybdenum(V)

Crystal data

[Mo₄(C₁₂H₁₀O₂P)₄O₈]

$M_r = 1380.44$

Orthorhombic, *Pbcn*

$a = 16.1061$ (19) Å

$b = 19.891$ (2) Å

$c = 15.254$ (3) Å

$V = 4887.0$ (11) Å³

$Z = 4$

$F(000) = 2736$

$D_x = 1.876$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 9220 reflections

$\theta = 2.4$ – 30.0°

$\mu = 1.21$ mm⁻¹

$T = 100$ K

Block, dark red

$0.61 \times 0.49 \times 0.30$ mm

Data collection

Bruker Kappa APEXII CCD
diffractometer

Graphite monochromator

ω - and ϕ -scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2014)

$T_{\min} = 0.45$, $T_{\max} = 0.66$

255304 measured reflections

8962 independent reflections

7558 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.061$

$\theta_{\max} = 32.7^\circ$, $\theta_{\min} = 1.6^\circ$

$h = -24 \rightarrow 24$

$k = -30 \rightarrow 30$

$l = -23 \rightarrow 23$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.039$

$wR(F^2) = 0.090$

$S = 1.27$

8962 reflections

326 parameters

0 restraints

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + 23.6442P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.92$ e Å⁻³

$\Delta\rho_{\min} = -1.05$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Mo1	0.94264 (2)	0.30980 (2)	0.15695 (2)	0.00840 (5)
Mo2	1.05855 (2)	0.21599 (2)	0.15724 (2)	0.00870 (5)
P1	0.76853 (4)	0.26945 (3)	0.25500 (4)	0.01124 (11)
P2	1.0000	0.45135 (4)	0.2500	0.01029 (15)
P3	1.0000	0.07459 (5)	0.2500	0.01125 (15)
O1	1.06085 (11)	0.31230 (9)	0.18613 (12)	0.0102 (3)
O2	0.93968 (11)	0.21393 (9)	0.18558 (12)	0.0095 (3)
O3	0.93008 (12)	0.31794 (10)	0.04868 (12)	0.0133 (3)
O4	1.07105 (12)	0.20672 (10)	0.04932 (12)	0.0142 (3)
O5	1.18201 (11)	0.21577 (10)	0.19500 (13)	0.0121 (3)
O6	0.81967 (11)	0.31237 (10)	0.19196 (12)	0.0120 (3)
O7	0.94434 (12)	0.41077 (9)	0.18832 (12)	0.0114 (3)
O8	1.05526 (12)	0.11516 (9)	0.18828 (13)	0.0133 (3)
C1	0.69240 (16)	0.22699 (13)	0.19004 (18)	0.0136 (4)
C2	0.72255 (18)	0.17894 (16)	0.1312 (2)	0.0199 (5)
H2	0.7791	0.1696	0.1289	0.024*
C3	0.6680 (2)	0.14529 (17)	0.0762 (2)	0.0226 (6)
H3	0.6880	0.1134	0.0369	0.027*
C4	0.58371 (19)	0.15910 (17)	0.0796 (2)	0.0218 (6)
H4	0.5471	0.1361	0.0433	0.026*
C5	0.5542 (2)	0.20701 (18)	0.1370 (2)	0.0260 (6)
H5	0.4976	0.2162	0.1389	0.031*
C6	0.60803 (18)	0.24191 (17)	0.1924 (2)	0.0212 (5)
H6	0.5878	0.2747	0.2303	0.025*
C7	0.71892 (16)	0.32266 (13)	0.33376 (17)	0.0128 (4)
C8	0.66756 (17)	0.29415 (14)	0.39712 (19)	0.0166 (5)
H8	0.6589	0.2479	0.3977	0.020*
C9	0.62922 (19)	0.33445 (16)	0.4595 (2)	0.0208 (5)
H9	0.5942	0.3153	0.5010	0.025*
C10	0.64329 (19)	0.40329 (16)	0.4597 (2)	0.0207 (5)
H10	0.6179	0.4303	0.5017	0.025*
C11	0.6953 (2)	0.43199 (15)	0.3972 (2)	0.0215 (5)
H11	0.7045	0.4782	0.3975	0.026*
C12	0.73355 (18)	0.39208 (14)	0.3343 (2)	0.0175 (5)
H12	0.7686	0.4113	0.2928	0.021*
C13	1.06447 (16)	0.50353 (13)	0.18254 (17)	0.0130 (4)
C14	1.1143 (2)	0.55275 (15)	0.2199 (2)	0.0220 (6)
H14	1.1133	0.5597	0.2802	0.026*
C15	1.1655 (2)	0.59151 (16)	0.1671 (2)	0.0254 (6)
H15	1.1998	0.6238	0.1922	0.031*
C16	1.1659 (2)	0.58250 (16)	0.0774 (2)	0.0226 (6)
H16	1.1989	0.6098	0.0422	0.027*
C17	1.1175 (2)	0.53317 (18)	0.0397 (2)	0.0245 (6)
H17	1.1187	0.5265	−0.0206	0.029*
C18	1.0667 (2)	0.49336 (16)	0.09217 (19)	0.0210 (5)

H18	1.0342	0.4599	0.0669	0.025*
C19	1.06746 (18)	0.02056 (14)	0.31128 (19)	0.0167 (5)
C20	1.0336 (2)	-0.0317 (2)	0.3592 (3)	0.0397 (10)
H20	0.9762	-0.0371	0.3614	0.048*
C21	1.0845 (3)	-0.0761 (2)	0.4042 (4)	0.0500 (13)
H21	1.0613	-0.1109	0.4366	0.060*
C22	1.1683 (3)	-0.0686 (2)	0.4008 (3)	0.0392 (10)
H22	1.2023	-0.0988	0.4303	0.047*
C23	1.2032 (2)	-0.0169 (2)	0.3543 (3)	0.0347 (9)
H23	1.2605	-0.0117	0.3534	0.042*
C24	1.1529 (2)	0.02794 (18)	0.3083 (2)	0.0240 (6)
H24	1.1766	0.0626	0.2759	0.029*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Mo1	0.00826 (8)	0.00886 (8)	0.00810 (8)	-0.00049 (6)	-0.00012 (6)	0.00068 (6)
Mo2	0.00835 (8)	0.00926 (9)	0.00849 (8)	-0.00033 (6)	0.00031 (6)	-0.00117 (6)
P1	0.0070 (2)	0.0143 (3)	0.0124 (3)	-0.0001 (2)	0.0006 (2)	0.0027 (2)
P2	0.0124 (4)	0.0087 (3)	0.0099 (4)	0.000	0.0000 (3)	0.000
P3	0.0108 (4)	0.0090 (3)	0.0140 (4)	0.000	-0.0004 (3)	0.000
O1	0.0085 (7)	0.0114 (7)	0.0107 (7)	-0.0012 (6)	0.0007 (6)	-0.0018 (6)
O2	0.0082 (7)	0.0090 (7)	0.0113 (7)	-0.0008 (6)	0.0015 (6)	0.0006 (6)
O3	0.0136 (8)	0.0160 (8)	0.0102 (8)	-0.0018 (7)	0.0002 (6)	0.0013 (6)
O4	0.0156 (8)	0.0163 (8)	0.0106 (8)	-0.0012 (7)	0.0007 (6)	-0.0023 (7)
O5	0.0086 (7)	0.0138 (8)	0.0140 (8)	0.0000 (6)	0.0006 (6)	-0.0045 (6)
O6	0.0082 (7)	0.0150 (8)	0.0128 (8)	0.0016 (6)	0.0006 (6)	0.0036 (6)
O7	0.0134 (8)	0.0093 (7)	0.0116 (7)	-0.0009 (6)	-0.0026 (6)	0.0012 (6)
O8	0.0144 (8)	0.0104 (8)	0.0151 (8)	0.0001 (6)	0.0030 (7)	-0.0005 (6)
C1	0.0104 (10)	0.0141 (10)	0.0161 (11)	-0.0001 (8)	-0.0012 (8)	0.0021 (9)
C2	0.0151 (11)	0.0237 (13)	0.0209 (13)	0.0016 (10)	0.0005 (10)	-0.0026 (11)
C3	0.0228 (14)	0.0254 (14)	0.0196 (13)	0.0006 (11)	-0.0033 (11)	-0.0058 (11)
C4	0.0189 (13)	0.0261 (14)	0.0204 (13)	-0.0030 (11)	-0.0072 (10)	-0.0008 (11)
C5	0.0151 (12)	0.0329 (16)	0.0300 (16)	0.0029 (12)	-0.0070 (11)	-0.0062 (13)
C6	0.0126 (11)	0.0276 (14)	0.0233 (13)	0.0030 (10)	-0.0019 (10)	-0.0052 (11)
C7	0.0112 (10)	0.0141 (10)	0.0130 (10)	0.0000 (8)	0.0007 (8)	0.0006 (8)
C8	0.0159 (11)	0.0167 (11)	0.0174 (11)	-0.0026 (9)	0.0039 (9)	0.0014 (9)
C9	0.0192 (12)	0.0248 (14)	0.0182 (12)	-0.0009 (11)	0.0053 (10)	0.0011 (11)
C10	0.0186 (12)	0.0250 (14)	0.0184 (12)	0.0014 (11)	0.0017 (10)	-0.0048 (11)
C11	0.0226 (13)	0.0171 (12)	0.0247 (14)	-0.0005 (10)	0.0015 (11)	-0.0031 (11)
C12	0.0160 (11)	0.0153 (11)	0.0212 (12)	-0.0007 (9)	0.0028 (10)	0.0019 (10)
C13	0.0143 (10)	0.0099 (9)	0.0149 (10)	0.0003 (8)	0.0018 (8)	-0.0014 (8)
C14	0.0251 (14)	0.0185 (12)	0.0223 (13)	-0.0083 (11)	0.0022 (11)	-0.0047 (10)
C15	0.0308 (16)	0.0184 (13)	0.0271 (15)	-0.0106 (12)	0.0056 (12)	-0.0028 (11)
C16	0.0207 (13)	0.0182 (12)	0.0289 (15)	0.0008 (10)	0.0083 (11)	0.0060 (11)
C17	0.0263 (15)	0.0329 (16)	0.0144 (12)	-0.0008 (12)	0.0057 (11)	0.0023 (11)
C18	0.0246 (13)	0.0247 (14)	0.0138 (11)	-0.0064 (11)	0.0036 (10)	-0.0017 (10)
C19	0.0164 (11)	0.0139 (11)	0.0197 (12)	0.0007 (9)	-0.0033 (9)	0.0027 (9)

C20	0.0243 (16)	0.039 (2)	0.055 (3)	-0.0017 (15)	-0.0042 (16)	0.0294 (19)
C21	0.048 (3)	0.040 (2)	0.062 (3)	0.0002 (19)	-0.014 (2)	0.032 (2)
C22	0.044 (2)	0.0339 (19)	0.039 (2)	0.0173 (17)	-0.0218 (18)	0.0018 (16)
C23	0.0236 (15)	0.050 (2)	0.0302 (17)	0.0138 (15)	-0.0116 (13)	-0.0103 (16)
C24	0.0177 (13)	0.0319 (16)	0.0226 (14)	0.0053 (11)	-0.0051 (11)	-0.0037 (12)

Geometric parameters (Å, °)

Mo1—O3	1.6717 (19)	C5—C6	1.396 (4)
Mo1—O1	1.9559 (18)	C5—H5	0.9300
Mo1—O2	1.9569 (18)	C6—H6	0.9300
Mo1—O6	2.0519 (18)	C7—C8	1.393 (4)
Mo1—O7	2.0648 (18)	C7—C12	1.401 (4)
Mo1—O1 ⁱ	2.3950 (19)	C8—C9	1.389 (4)
Mo1—Mo2	2.6395 (3)	C8—H8	0.9300
Mo2—O4	1.6686 (19)	C9—C10	1.388 (4)
Mo2—O2	1.9632 (18)	C9—H9	0.9300
Mo2—O1	1.9661 (18)	C10—C11	1.392 (4)
Mo2—O8	2.0615 (19)	C10—H10	0.9300
Mo2—O5	2.0701 (19)	C11—C12	1.390 (4)
Mo2—O2 ⁱ	2.3982 (19)	C11—H11	0.9300
P1—O6	1.527 (2)	C12—H12	0.9300
P1—O5 ⁱ	1.5351 (19)	C13—C14	1.389 (4)
P1—C1	1.789 (3)	C13—C18	1.394 (4)
P1—C7	1.789 (3)	C14—C15	1.388 (4)
P2—O7	1.5298 (19)	C14—H14	0.9300
P2—O7 ⁱ	1.5298 (19)	C15—C16	1.379 (5)
P2—C13 ⁱ	1.793 (3)	C15—H15	0.9300
P2—C13	1.793 (3)	C16—C17	1.379 (5)
P3—O8 ⁱ	1.526 (2)	C16—H16	0.9300
P3—O8	1.526 (2)	C17—C18	1.391 (4)
P3—C19	1.791 (3)	C17—H17	0.9300
P3—C19 ⁱ	1.791 (3)	C18—H18	0.9300
O1—Mo1 ⁱ	2.3950 (19)	C19—C20	1.383 (5)
O2—Mo2 ⁱ	2.3982 (19)	C19—C24	1.385 (4)
O5—P1 ⁱ	1.5351 (19)	C20—C21	1.386 (5)
C1—C6	1.391 (4)	C20—H20	0.9300
C1—C2	1.398 (4)	C21—C22	1.360 (7)
C2—C3	1.387 (4)	C21—H21	0.9300
C2—H2	0.9300	C22—C23	1.369 (7)
C3—C4	1.386 (4)	C22—H22	0.9300
C3—H3	0.9300	C23—C24	1.393 (5)
C4—C5	1.379 (5)	C23—H23	0.9300
C4—H4	0.9300	C24—H24	0.9300
O3—Mo1—O1	109.88 (9)	C6—C1—P1	123.7 (2)
O3—Mo1—O2	108.17 (9)	C2—C1—P1	116.1 (2)
O1—Mo1—O2	89.87 (7)	C3—C2—C1	119.9 (3)

O3—Mo1—O6	97.93 (9)	C3—C2—H2	120.1
O1—Mo1—O6	151.61 (7)	C1—C2—H2	120.1
O2—Mo1—O6	86.71 (7)	C4—C3—C2	120.1 (3)
O3—Mo1—O7	97.84 (8)	C4—C3—H3	119.9
O1—Mo1—O7	84.81 (7)	C2—C3—H3	119.9
O2—Mo1—O7	153.70 (7)	C5—C4—C3	119.9 (3)
O6—Mo1—O7	85.89 (8)	C5—C4—H4	120.0
O3—Mo1—O1 ⁱ	169.29 (8)	C3—C4—H4	120.0
O1—Mo1—O1 ⁱ	78.16 (7)	C4—C5—C6	120.9 (3)
O2—Mo1—O1 ⁱ	78.27 (7)	C4—C5—H5	119.6
O6—Mo1—O1 ⁱ	73.54 (7)	C6—C5—H5	119.6
O7—Mo1—O1 ⁱ	75.43 (7)	C1—C6—C5	119.0 (3)
O3—Mo1—Mo2	98.95 (7)	C1—C6—H6	120.5
O1—Mo1—Mo2	47.86 (5)	C5—C6—H6	120.5
O2—Mo1—Mo2	47.78 (5)	C8—C7—C12	119.8 (3)
O6—Mo1—Mo2	134.41 (5)	C8—C7—P1	119.4 (2)
O7—Mo1—Mo2	132.67 (5)	C12—C7—P1	120.8 (2)
O1 ⁱ —Mo1—Mo2	91.70 (4)	C9—C8—C7	120.3 (3)
O4—Mo2—O2	109.42 (9)	C9—C8—H8	119.9
O4—Mo2—O1	109.06 (9)	C7—C8—H8	119.9
O2—Mo2—O1	89.39 (7)	C10—C9—C8	119.9 (3)
O4—Mo2—O8	97.02 (9)	C10—C9—H9	120.1
O2—Mo2—O8	84.49 (7)	C8—C9—H9	120.1
O1—Mo2—O8	153.77 (8)	C9—C10—C11	120.1 (3)
O4—Mo2—O5	99.12 (9)	C9—C10—H10	120.0
O2—Mo2—O5	151.09 (7)	C11—C10—H10	120.0
O1—Mo2—O5	85.51 (7)	C12—C11—C10	120.4 (3)
O8—Mo2—O5	87.63 (8)	C12—C11—H11	119.8
O4—Mo2—O2 ⁱ	169.42 (8)	C10—C11—H11	119.8
O2—Mo2—O2 ⁱ	77.94 (7)	C11—C12—C7	119.5 (3)
O1—Mo2—O2 ⁱ	78.02 (7)	C11—C12—H12	120.2
O8—Mo2—O2 ⁱ	75.76 (7)	C7—C12—H12	120.2
O5—Mo2—O2 ⁱ	73.16 (7)	C14—C13—C18	119.6 (3)
O4—Mo2—Mo1	99.31 (7)	C14—C13—P2	120.5 (2)
O2—Mo2—Mo1	47.57 (5)	C18—C13—P2	119.9 (2)
O1—Mo2—Mo1	47.54 (5)	C15—C14—C13	119.8 (3)
O8—Mo2—Mo1	132.06 (5)	C15—C14—H14	120.1
O5—Mo2—Mo1	132.95 (5)	C13—C14—H14	120.1
O2 ⁱ —Mo2—Mo1	91.27 (4)	C16—C15—C14	120.4 (3)
O6—P1—O5 ⁱ	114.95 (10)	C16—C15—H15	119.8
O6—P1—C1	106.55 (12)	C14—C15—H15	119.8
O5 ⁱ —P1—C1	107.61 (12)	C17—C16—C15	120.2 (3)
O6—P1—C7	109.45 (12)	C17—C16—H16	119.9
O5 ⁱ —P1—C7	108.04 (12)	C15—C16—H16	119.9
C1—P1—C7	110.19 (12)	C16—C17—C18	119.8 (3)
O7—P2—O7 ⁱ	116.31 (15)	C16—C17—H17	120.1
O7—P2—C13 ⁱ	108.61 (11)	C18—C17—H17	120.1
O7 ⁱ —P2—C13 ⁱ	106.97 (11)	C17—C18—C13	120.1 (3)

O7—P2—C13	106.97 (11)	C17—C18—H18	119.9
O7 ⁱ —P2—C13	108.61 (11)	C13—C18—H18	119.9
C13 ⁱ —P2—C13	109.25 (17)	C20—C19—C24	119.3 (3)
O8 ⁱ —P3—O8	116.17 (15)	C20—C19—P3	119.2 (2)
O8 ⁱ —P3—C19	110.42 (12)	C24—C19—P3	121.5 (2)
O8—P3—C19	106.58 (12)	C19—C20—C21	120.4 (4)
O8 ⁱ —P3—C19 ⁱ	106.58 (12)	C19—C20—H20	119.8
O8—P3—C19 ⁱ	110.42 (12)	C21—C20—H20	119.8
C19—P3—C19 ⁱ	106.27 (19)	C22—C21—C20	119.9 (4)
Mo1—O1—Mo2	84.60 (7)	C22—C21—H21	120.0
Mo1—O1—Mo1 ⁱ	101.77 (7)	C20—C21—H21	120.0
Mo2—O1—Mo1 ⁱ	101.73 (8)	C21—C22—C23	120.6 (3)
Mo1—O2—Mo2	84.65 (7)	C21—C22—H22	119.7
Mo1—O2—Mo2 ⁱ	101.90 (7)	C23—C22—H22	119.7
Mo2—O2—Mo2 ⁱ	102.02 (7)	C22—C23—C24	120.2 (4)
P1 ⁱ —O5—Mo2	129.43 (11)	C22—C23—H23	119.9
P1—O6—Mo1	132.12 (11)	C24—C23—H23	119.9
P2—O7—Mo1	131.57 (11)	C19—C24—C23	119.5 (3)
P3—O8—Mo2	132.15 (12)	C19—C24—H24	120.2
C6—C1—C2	120.1 (3)	C23—C24—H24	120.2

Symmetry code: (i) $-x+2, y, -z+1/2$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C4—H4 \cdots O3 ⁱⁱ	0.93	2.52	3.188 (4)	129
C21—H21 \cdots O4 ⁱⁱⁱ	0.93	2.57	3.420 (5)	152

Symmetry codes: (ii) $x-1/2, -y+1/2, -z$; (iii) $x, -y, z+1/2$.