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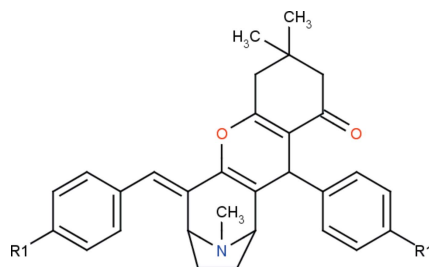
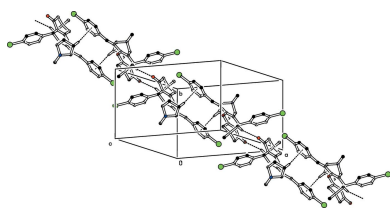
Crystal structures of (12*E*)-12-(4-benzylidene)- 7,7,16-trimethyl-3-(4-methylphenyl)-1-oxa-16-aza- tetracyclo[11.2.1.0^{2,11}.0^{4,9}]hexadeca-2(11),4(9)- dien-5-one and (12*E*)-12-(4-bromobenzylidene)-7,7- 3-(4-bromophenyl)-,7,16-trimethyl-10-oxa-16-aza- tetracyclo[11.2.1.0^{2,11}.0^{4,9}]hexadeca-2(11),4(9)- dien-5-one

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The title compounds, C₃₂H₃₅NO₂, (I), and C₃₀H₂₉Br₂NO₂, (II), differ by the presence of a bromine atom instead of a methyl atom in the *para* position of two benzene rings of compound (II). The two compounds have a structural overlap r.m.s. deviation of 0.27 Å. The pyran and seven-membered cycloheptene rings in both structures adopt boat and boat-sofa conformations, respectively. Intra- and intermolecular C—H...O hydrogen bonds are responsible for the consolidation of the crystal packing of both molecules. In addition to this, weak C—H...π interactions are also observed. The intermolecular interactions were quantified and analysed using Hirshfeld surface analysis.

1. Chemical context

The tropane skeleton is found widely in both natural and manufactured medications. It is the fundamental component of many beneficial alkaloids, including atropine, scopolamine, and cocaine, whose derivatives are important in the treatment of neurological and psychiatric conditions such depression and panic disorder (Cheenpracha *et al.*, 2013; Afewerki *et al.*, 2019; Dongbang *et al.*, 2021). It is also a key component in the synthesis of newer types of drugs. Tropane derivatives are used to treat irritable bowel syndrome, peptic ulcers, colic, cystitis, and pancreatitis thanks to their anti-spasmodic properties. In view of the above importance, we have undertaken a single-crystal X-ray diffraction study for the title compounds, and the results are presented herein.



R1 - Methyl in (I) and Bromine in (II)



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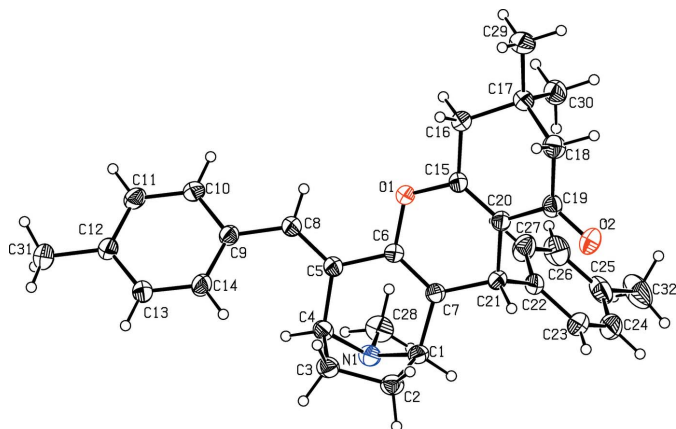


Figure 1
A view of the molecular structure of compound (I), showing the atom labelling. Displacement ellipsoids are drawn at the 30% probability level.

2. Structural commentary

The molecular structure of the title compounds (I) and (II) are illustrated in Figs. 1 and 2, respectively. Fig. 3 shows the superposition of the two compounds except for atom C21 using *Qmol* (Gans & Shalloway, 2001); the r.m.s. deviation is 0.27 Å. The methylphenyl rings in (I) are oriented at a dihedral angle of 57.7 (1)°. The methyl atoms C31 and C32 in (I) deviate by -0.036 (1) and 0.053 (1) Å, respectively, from the rings to which they are attached. The bromophenyl rings in (II) are oriented at a dihedral angle of 54.3 (1)°. Bromine atoms Br1 and Br2 deviate by 0.050 (1) and 0.037 (1) Å, respectively, from the rings to which they are attached.

The seven-membered ring (C1–C7) in both compounds has a boat-sofa conformation, with puckering parameters (Boesenkool & Boeyens, 1980) $q_2 = 1.021$ (2) and $q_3 = 0.391$ (2) Å in (I) and $q_2 = 1.053$ (2) and $q_3 = 0.374$ (2) Å in (II). The piperidine ring (N1/C4–C7/C1) has a half-boat conformation in both compounds, with atoms C4–C7/C1 in the plane and atom N1 deviating by -0.793 (1) [in (I)] or 0.785 (1) Å [in (II)] from this plane. The pyran ring (C6/O1/C15/C20/C21/C7) adopts boat conformation in both structures with puckering

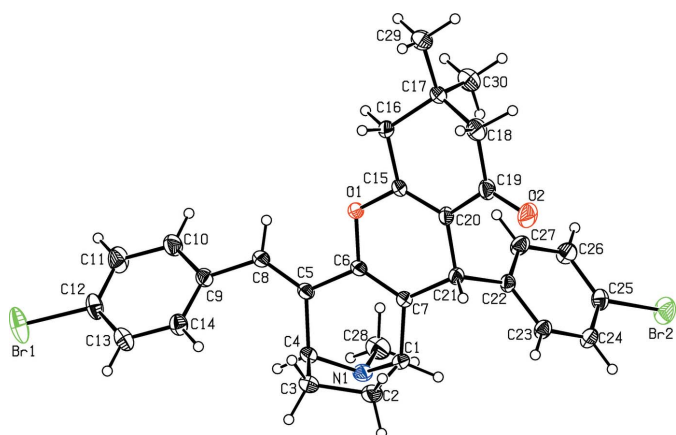


Figure 2
A view of the molecular structure of compound (II), showing the atom labelling. Displacement ellipsoids are drawn at the 30% probability level.

Table 1
Hydrogen-bond geometry (Å, °) for (I).

C_g is the centroid of the C22–C27 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C8–H8 \cdots O1	0.93	2.37	2.766 (2)	105
C14–H14 \cdots O2 ⁱ	0.93	2.59	3.285 (2)	131
C3–H3B \cdots C _g ⁱ	0.97	2.88	3.795 (2)	152

Symmetry code: (i) $x - \frac{1}{2}, -y + \frac{3}{2}, -z$.

Table 2
Hydrogen-bond geometry (Å, °) for (II).

C_g is the centroid of the C9–C14 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C8–H8 \cdots O1	0.93	2.36	2.742 (3)	105
C30–H30A \cdots O2 ⁱ	0.96	2.58	3.306 (5)	132
C3–H3B \cdots C _g ⁱⁱ	0.97	2.84	3.798 (3)	169

Symmetry codes: (i) $-x + 1, -y + 1, -z$; (ii) $-x + 2, -y + 2, -z$.

parameters (Cremer & Pople, 1975) $q_2 = Q_T = 0.185$ (2) Å and $\varphi = 43.9$ (1)° [in (I)] and $q_2 = Q_T = 0.087$ (1) Å and $\varphi = 47.1$ (1)° [in (II)]. The cyclohexene ring (C15–C20) has a distorted sofa conformation in both (I) and (II), with $\Delta C_s(C17)$ asymmetry parameters (Nardelli, 1983) of 0.058 (1) and 0.004 (1), respectively.

Intramolecular C–H \cdots O hydrogen bonds are observed (Tables 1 and 2)

3. Supramolecular features

In the crystal of (I), molecules associate *via* C–H \cdots O intermolecular interactions (C14–H14 \cdots O2ⁱ, Table 1), forming chains propagating along the [100] direction, see Fig. 4. In addition to this, inversion-related molecules are linked into chains by C–H \cdots π interactions [C3–H3B \cdots C_g, where C_g is the centroid of the C22–C27 benzene ring of the symmetry-

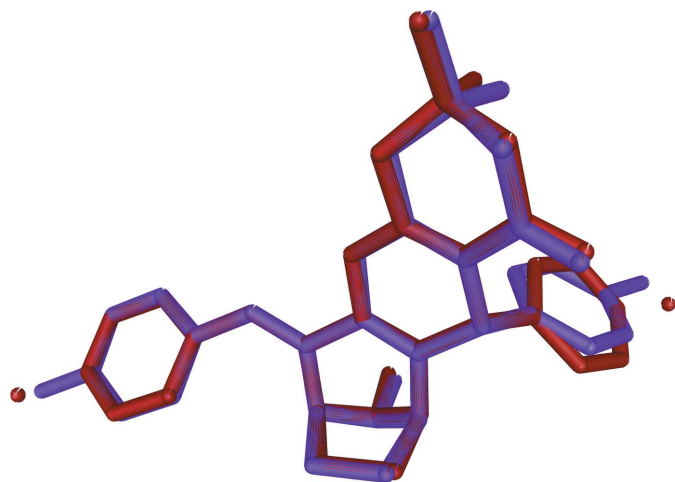


Figure 3
Superposition of compound (I) (violet) and compound (II) (brown) except for atoms Br1, Br2 and C21.

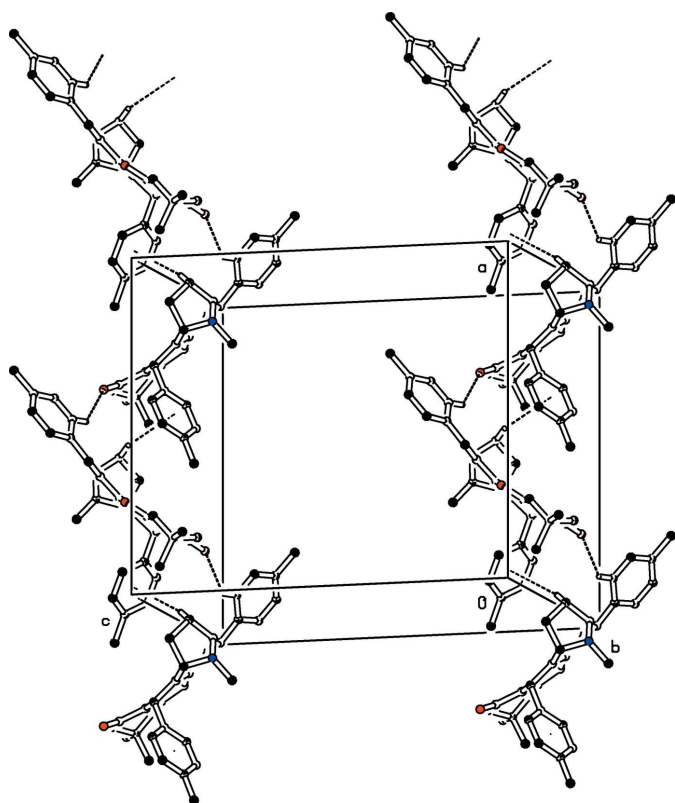


Figure 4
The crystal packing of the title compound (I) viewed along *b* axis. The C–H···O and C–H··· π intermolecular interactions are shown as dashed lines. For clarity, H atoms not involved in these interactions have been omitted.

related molecule at $(-\frac{1}{2} + x, \frac{3}{2} - y, -z)$, see Fig. 4]. These two interactions form chains running in a helical manner along [101].

In the crystal of (II), molecules are also linked *via* C–H···O intermolecular interaction (C30–H30A···O2ⁱ, Table 2), here forming centrosymmetrical dimers with an $R_2^2(12)$ ring motif (Fig. 5). The dimers are further linked into chains by C–H··· π interactions (C3–H3B···Cg, where Cg is the centroid of the C9–C14 benzene ring of the symmetry-

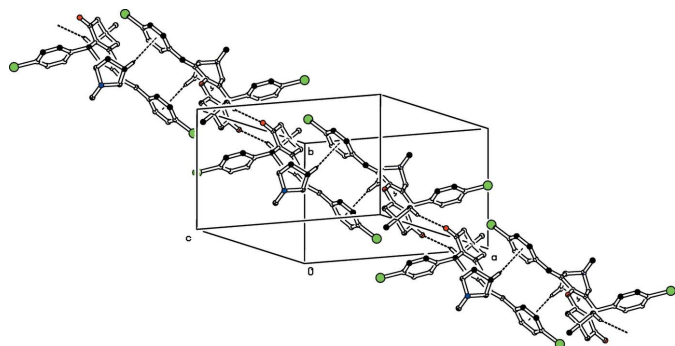


Figure 5
The centrosymmetrical dimer formed in compound (II) *via* C–H···O hydrogen bonds (dashed lines). The dimers are linked by C–H··· π interactions (dashed lines). For clarity H atoms, not involved in these interactions have been omitted.

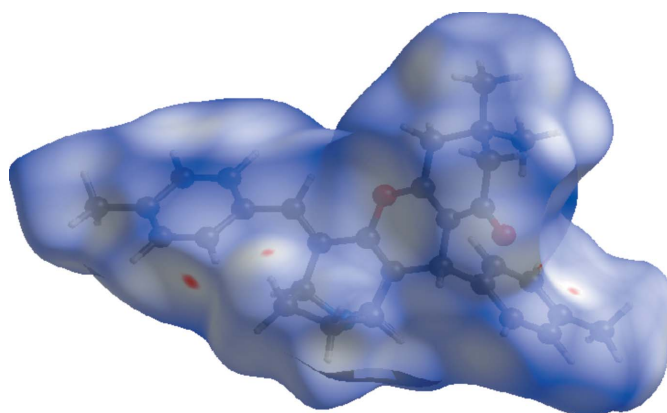


Figure 6
A view of the Hirshfeld surface mapped over d_{norm} in the range -0.0701 to $+1.6693$ arbitrary units for compound (I).

related molecule at $(2 - x, 2 - y, -z)$, see Fig. 4]. These two interactions form chains running diagonally along [110].

4. Hirshfeld surface analysis

To further characterize the intermolecular interactions in the title compound, we carried out a Hirshfeld surface (HS) analysis (Spackman & Jayatilaka, 2009) using *Crystal Explorer 21* (Spackman *et al.*, 2021) and generated the associated two dimensional fingerprint plots (McKinnon *et al.*, 2007). The HS mapped over d_{norm} in the range -0.0701 to $+1.6693$ a.u. for compound (I) and -0.1162 to $+1.5964$ a.u. for compound (II) are illustrated in Figs. 6 and 7, using colours to indicate contacts that are shorter (red areas), equal to (white areas), or longer than (blue areas) the sum of the van der Waals radii (Ashfaq *et al.*, 2021).

The two-dimensional fingerprint plots provide quantitative information about the non-covalent interactions and the crystal packing in terms of the percentage contribution of the interatomic contacts (Spackman & McKinnon, 2002; Ashfaq *et al.*, 2021). The HS analysis reveals that H···H (74.2%) and

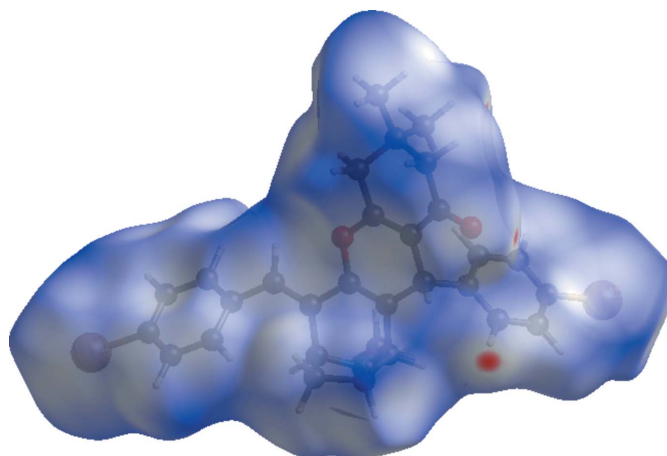
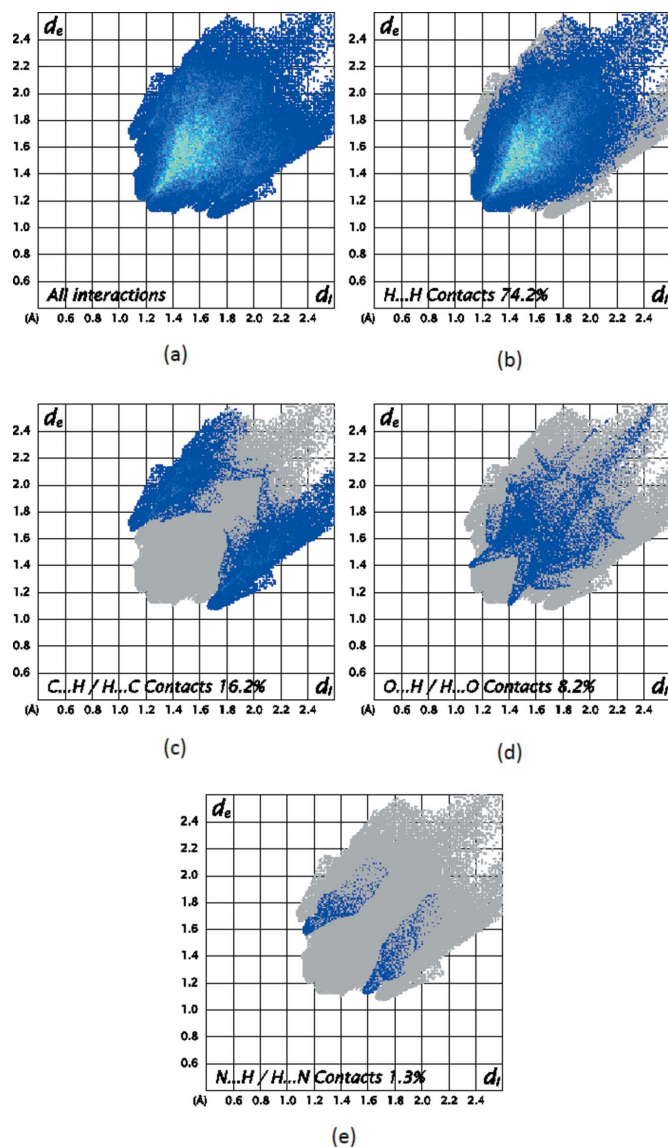


Figure 7
A view of the Hirshfeld surface mapped over d_{norm} in the range -0.1162 to $+1.5964$ arbitrary units for compound (II).

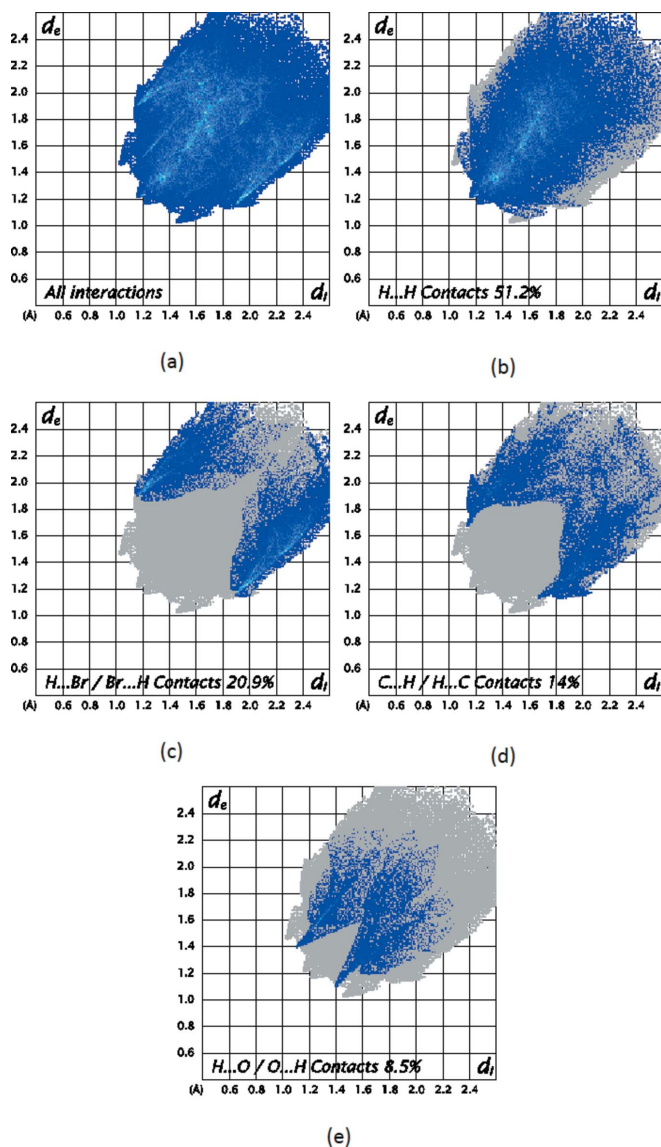

Figure 8

Two-dimensional fingerprint plots for compound (I), showing (a) all interactions, and delineated into (b) H...H, (c) C...H/H...C, (d) O...H/H...O and (e) N...H/H...N interactions. The d_i and d_e values are the closest internal and external distances (in Å) from given points on the Hirshfeld surface.

H...C/C...H (16.2%) contacts are the main contributors to the crystal packing, followed by O...H/H...O (8.2%) and N...H/H...N (1.3%) contacts for compound (I) (Fig. 8). In compound (II), H...H (51.2%) and H...Br/Br...H (20.9%) contacts are the main contributors to the crystal packing, followed by C...H/H...C (14%) and O...H/H...O (8.5%) contacts (Fig. 9). The fragment patches on the HS provide an easy way to investigate the nearest neighbour coordination environment of a molecule (coordination number), which is 14 for compound (I) and 15 for compound (II).

5. Synthesis and crystallization

Compound (I) was synthesized from a mixture of 8-methyl-8-azabicyclo[3.2.1]octan-3-one and two equivalents of


Figure 9

Two-dimensional fingerprint plots for compound (II), showing (a) all interactions, and delineated into (b) H...H, (c) Br...H/H...Br, (d) C...H/H...C and (e) O...H/H...O interactions. The d_i and d_e values are the closest internal and external distances (in Å) from given points on the Hirshfeld surface.

4-methylbenzaldehyde and 5,5-dimethylcyclohexane-1,3-dione dissolved in ethanol/acetic acid and refluxed for 12 h. After completion of the reaction, as indicated by thin layer chromatography (TLC), the mixture was cooled to room temperature, poured into ice-cold water and neutralized with a saturated solution of sodium bicarbonate. The compound was further recrystallized from ethanol to obtain crystals suitable for single crystal X-ray analysis.

Compound (II) was synthesized from a mixture of 8-methyl-8-azabicyclo[3.2.1]octan-3-one, two equivalents of 4-bromobenzaldehyde and 5,5-dimethylcyclohexane-1,3-dione dissolved in ethanol/acetic acid and refluxed for 12 h. After completion of the reaction, as indicated by thin layer chromatography (TLC), the mixture was cooled to room

Table 3
Experimental details.

	(I)	(II)
Crystal data		
Chemical formula	C ₃₂ H ₃₅ NO ₂	C ₃₀ H ₂₉ Br ₂ NO ₂
<i>M_r</i>	465.61	595.36
Crystal system, space group	Orthorhombic, <i>Pbca</i>	Monoclinic, <i>P2₁/n</i>
Temperature (K)	293	293
<i>a</i> , <i>b</i> , <i>c</i> (Å)	16.7576 (6), 16.3223 (7), 19.2416 (8)	16.051 (3), 9.7793 (17), 17.005 (3)
α , β , γ (°)	90, 90, 90	90, 96.312 (5), 90
<i>V</i> (Å ³)	5263.0 (4)	2653.1 (8)
<i>Z</i>	8	4
Radiation type	Mo <i>K</i> α	Mo <i>K</i> α
μ (mm ⁻¹)	0.07	3.08
Crystal size (mm)	0.21 × 0.18 × 0.16	0.20 × 0.17 × 0.15
Data collection		
Diffractometer	Bruker SMART APEX CCD area-detector	Bruker SMART APEX CCD area-detector
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	186331, 8013, 5206	133143, 6556, 4933
<i>R</i> _{int}	0.097	0.091
(<i>sin</i> θ / λ) _{max} (Å ⁻¹)	0.714	0.667
Refinement		
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> [<i>F</i> ²], <i>S</i>	0.059, 0.213, 1.04	0.050, 0.141, 1.09
No. of reflections	8013	6556
No. of parameters	319	316
H-atom treatment	H-atom parameters constrained	H-atom parameters constrained
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	0.31, -0.23	0.54, -0.71

Computer programs: *APEX2* and *SAINT* (Bruker, 2009), *SHELXT2018/2* (Sheldrick, 2015a), *SHELXL2018/3* (Sheldrick, 2015b), *ORTEP-3 for Windows* (Farrugia, 2012) and *PLATON* (Spek, 2020).

temperature, poured into ice-cold water and neutralized with a saturated solution of sodium bicarbonate. The compound was further recrystallized from ethanol to obtain crystals suitable for single crystal X-ray analysis.

6. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 3. In both (I) and (II), H atoms were placed in idealized positions and allowed to ride on their parent atoms: C–H = 0.93–0.98 Å, with *U*_{iso}(H) = 1.5*U*_{eq}(C–methyl) and 1.2*U*_{eq}(C) for other H atoms.

Acknowledgements

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supporting information

Acta Cryst. (2023). E79, 392-396 [https://doi.org/10.1107/S205698902300275X]

Crystal structures of (12*E*)-12-(4-benzylidene)-7,7,16-trimethyl-3-(4-methylphenyl)-1-oxa-16-azatetracyclo[11.2.1.0^{2,11}.0^{4,9}]hexadeca-2(11),4(9)-dien-5-one and (12*E*)-12-(4-bromobenzylidene)-7,7,16-trimethyl-10-oxa-16-azatetracyclo[11.2.1.0^{2,11}.0^{4,9}]hexadeca-2(11),4(9)-dien-5-one

C. Selva Meenatchi, R. Vishnupriya, J. Suresh, S. Raja Rubina, S. Selvanayagam and S. R. Bhandari

Computing details

For both structures, data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT* (Bruker, 2009); program(s) used to solve structure: *SHELXT2018/2* (Sheldrick, 2015a); program(s) used to refine structure: *SHELXL2018/3* (Sheldrick, 2015b); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012) and *PLATON* (Spek, 2020); software used to prepare material for publication: *SHELXL2018/3* (Sheldrick, 2015b) and *PLATON* (Spek, 2020).

(12*E*)-12-(4-Benzylidene)-7,7,16-trimethyl-3-(4-methylphenyl)-1-oxa-16-azatetracyclo[11.2.1.0^{2,11}.0^{4,9}]hexadeca-2(11),4(9)-dien-5-one (I)

Crystal data

$C_{32}H_{35}NO_2$	$D_x = 1.175 \text{ Mg m}^{-3}$
$M_r = 465.61$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Orthorhombic, <i>Pbca</i>	Cell parameters from 118400 reflections
$a = 16.7576 (6) \text{ \AA}$	$\theta = 3.1\text{--}27.8^\circ$
$b = 16.3223 (7) \text{ \AA}$	$\mu = 0.07 \text{ mm}^{-1}$
$c = 19.2416 (8) \text{ \AA}$	$T = 293 \text{ K}$
$V = 5263.0 (4) \text{ \AA}^3$	Block, colourless
$Z = 8$	$0.21 \times 0.18 \times 0.16 \text{ mm}$
$F(000) = 2000$	

Data collection

Bruker SMART APEX CCD area-detector diffractometer	5206 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.097$
ω and φ scans	$\theta_{\text{max}} = 30.5^\circ$, $\theta_{\text{min}} = 2.9^\circ$
186331 measured reflections	$h = -23 \rightarrow 23$
8013 independent reflections	$k = -23 \rightarrow 23$
	$l = -27 \rightarrow 27$

Refinement

Refinement on F^2	$S = 1.04$
Least-squares matrix: full	8013 reflections
$R[F^2 > 2\sigma(F^2)] = 0.059$	319 parameters
$wR(F^2) = 0.213$	0 restraints

Hydrogen site location: inferred from neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.1059P)^2 + 1.8765P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.31 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL2018/3*
 (Sheldrick 2015b),
 $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
 Extinction coefficient: 0.0063 (9)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} [*] / <i>U</i> _{eq}
O1	0.21074 (7)	0.58367 (7)	0.03767 (7)	0.0427 (3)
O2	0.34366 (9)	0.72866 (9)	-0.13951 (8)	0.0577 (4)
N1	0.16512 (8)	0.78470 (9)	0.16257 (8)	0.0424 (3)
C1	0.18856 (10)	0.80464 (10)	0.09126 (10)	0.0405 (4)
H1	0.228435	0.848551	0.090424	0.049*
C2	0.10849 (11)	0.83394 (11)	0.06103 (11)	0.0463 (4)
H2A	0.105807	0.823989	0.011381	0.056*
H2B	0.100310	0.891866	0.069747	0.056*
C3	0.04652 (10)	0.78136 (11)	0.10060 (11)	0.0459 (4)
H3A	0.009062	0.815815	0.125555	0.055*
H3B	0.017141	0.746445	0.068828	0.055*
C4	0.09680 (10)	0.72978 (10)	0.15142 (9)	0.0400 (4)
H4	0.067771	0.720868	0.194982	0.048*
C5	0.12141 (9)	0.64808 (10)	0.11912 (9)	0.0370 (3)
C6	0.18632 (9)	0.65699 (9)	0.06908 (9)	0.0371 (3)
C7	0.21783 (9)	0.72967 (10)	0.05213 (9)	0.0376 (3)
C8	0.08674 (10)	0.57522 (10)	0.13228 (9)	0.0396 (4)
H8	0.109249	0.529878	0.110539	0.048*
C9	0.01749 (10)	0.55924 (10)	0.17700 (9)	0.0382 (3)
C10	0.01149 (11)	0.48435 (10)	0.21179 (10)	0.0435 (4)
H10	0.050998	0.445040	0.205445	0.052*
C11	-0.05206 (12)	0.46743 (11)	0.25556 (10)	0.0475 (4)
H11	-0.053642	0.417504	0.278801	0.057*
C12	-0.11325 (11)	0.52321 (11)	0.26546 (10)	0.0452 (4)
C13	-0.10909 (12)	0.59637 (12)	0.22870 (11)	0.0511 (5)
H13	-0.149884	0.634551	0.233678	0.061*
C14	-0.04585 (11)	0.61396 (11)	0.18485 (11)	0.0486 (4)
H14	-0.045578	0.663009	0.160261	0.058*
C15	0.25631 (10)	0.59020 (10)	-0.02083 (9)	0.0377 (3)
C16	0.26745 (11)	0.50897 (11)	-0.05506 (10)	0.0447 (4)
H16A	0.218260	0.493661	-0.078242	0.054*
H16B	0.278380	0.468083	-0.019734	0.054*
C17	0.33577 (11)	0.50880 (11)	-0.10836 (10)	0.0459 (4)

C18	0.32508 (13)	0.58496 (12)	-0.15407 (10)	0.0511 (4)
H18A	0.369117	0.587793	-0.186698	0.061*
H18B	0.276320	0.579014	-0.180759	0.061*
C19	0.32134 (10)	0.66405 (11)	-0.11423 (9)	0.0420 (4)
C20	0.28567 (9)	0.66154 (10)	-0.04453 (9)	0.0368 (3)
C21	0.28068 (9)	0.73973 (10)	-0.00295 (9)	0.0386 (3)
H21	0.264501	0.784237	-0.034095	0.046*
C22	0.36107 (10)	0.76184 (11)	0.02941 (10)	0.0418 (4)
C23	0.39866 (12)	0.83551 (12)	0.01558 (13)	0.0555 (5)
H23	0.375068	0.873040	-0.014421	0.067*
C24	0.47229 (14)	0.85379 (15)	0.04666 (15)	0.0689 (7)
H24	0.496623	0.903724	0.037033	0.083*
C25	0.50927 (12)	0.79995 (16)	0.09086 (13)	0.0624 (6)
C26	0.47107 (13)	0.72732 (17)	0.10501 (14)	0.0688 (6)
H26	0.494539	0.690036	0.135313	0.083*
C27	0.39826 (12)	0.70872 (14)	0.07493 (13)	0.0596 (5)
H27	0.373750	0.659182	0.085642	0.071*
C28	0.22827 (13)	0.75329 (15)	0.20727 (12)	0.0588 (5)
H28A	0.206981	0.742153	0.252617	0.088*
H28B	0.249391	0.703701	0.187734	0.088*
H28C	0.270044	0.793302	0.210916	0.088*
C29	0.33113 (16)	0.43051 (14)	-0.15184 (14)	0.0669 (6)
H29A	0.337967	0.383735	-0.122184	0.100*
H29B	0.280025	0.427542	-0.174248	0.100*
H29C	0.372469	0.431112	-0.186370	0.100*
C30	0.41730 (12)	0.51242 (14)	-0.07213 (13)	0.0590 (5)
H30A	0.420629	0.561454	-0.044726	0.088*
H30B	0.423397	0.465485	-0.042538	0.088*
H30C	0.458865	0.512600	-0.106463	0.088*
C31	-0.18224 (15)	0.50520 (14)	0.31352 (13)	0.0645 (6)
H31A	-0.198385	0.449151	0.307978	0.097*
H31B	-0.226148	0.540633	0.302414	0.097*
H31C	-0.166058	0.514370	0.360747	0.097*
C32	0.59000 (14)	0.8188 (2)	0.12213 (18)	0.0941 (10)
H32A	0.605958	0.873155	0.109106	0.141*
H32B	0.628525	0.780055	0.105291	0.141*
H32C	0.586766	0.815138	0.171847	0.141*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0463 (6)	0.0325 (6)	0.0492 (7)	-0.0009 (5)	0.0145 (5)	0.0013 (5)
O2	0.0641 (9)	0.0546 (8)	0.0545 (8)	-0.0103 (6)	0.0075 (7)	0.0150 (6)
N1	0.0379 (7)	0.0422 (8)	0.0471 (8)	0.0005 (6)	-0.0033 (6)	-0.0055 (6)
C1	0.0366 (7)	0.0319 (7)	0.0530 (10)	-0.0008 (6)	0.0002 (7)	-0.0028 (7)
C2	0.0473 (9)	0.0391 (9)	0.0525 (10)	0.0060 (7)	-0.0018 (8)	0.0024 (7)
C3	0.0367 (8)	0.0425 (9)	0.0585 (11)	0.0037 (7)	-0.0035 (7)	-0.0012 (8)
C4	0.0364 (7)	0.0396 (8)	0.0442 (9)	0.0021 (6)	0.0024 (6)	-0.0032 (7)

C5	0.0334 (7)	0.0377 (8)	0.0400 (8)	0.0012 (6)	0.0020 (6)	0.0010 (6)
C6	0.0361 (7)	0.0327 (7)	0.0424 (8)	0.0013 (6)	0.0035 (6)	0.0000 (6)
C7	0.0323 (7)	0.0346 (8)	0.0460 (9)	-0.0001 (6)	0.0003 (6)	-0.0004 (6)
C8	0.0377 (8)	0.0374 (8)	0.0437 (9)	0.0020 (6)	0.0041 (6)	0.0028 (7)
C9	0.0391 (8)	0.0360 (8)	0.0395 (8)	-0.0014 (6)	0.0010 (6)	0.0042 (6)
C10	0.0477 (9)	0.0359 (8)	0.0468 (9)	0.0039 (7)	0.0027 (7)	0.0051 (7)
C11	0.0587 (11)	0.0376 (8)	0.0460 (10)	-0.0014 (8)	0.0041 (8)	0.0098 (7)
C12	0.0485 (9)	0.0413 (9)	0.0459 (10)	-0.0049 (7)	0.0073 (7)	0.0027 (7)
C13	0.0441 (9)	0.0442 (10)	0.0649 (12)	0.0032 (7)	0.0139 (8)	0.0083 (8)
C14	0.0420 (9)	0.0422 (9)	0.0617 (11)	0.0027 (7)	0.0073 (8)	0.0168 (8)
C15	0.0343 (7)	0.0379 (8)	0.0410 (8)	0.0003 (6)	0.0052 (6)	0.0025 (6)
C16	0.0452 (9)	0.0362 (8)	0.0526 (10)	-0.0033 (7)	0.0095 (8)	-0.0028 (7)
C17	0.0479 (9)	0.0410 (9)	0.0489 (10)	0.0037 (7)	0.0111 (8)	0.0010 (7)
C18	0.0616 (11)	0.0527 (11)	0.0390 (9)	0.0050 (9)	0.0069 (8)	0.0044 (8)
C19	0.0353 (7)	0.0464 (9)	0.0442 (9)	0.0007 (7)	0.0010 (6)	0.0084 (7)
C20	0.0337 (7)	0.0359 (8)	0.0409 (8)	0.0003 (6)	0.0019 (6)	0.0038 (6)
C21	0.0362 (7)	0.0322 (7)	0.0475 (9)	-0.0008 (6)	0.0039 (6)	0.0051 (6)
C22	0.0360 (8)	0.0395 (8)	0.0499 (9)	-0.0043 (6)	0.0069 (7)	-0.0033 (7)
C23	0.0500 (10)	0.0446 (10)	0.0720 (14)	-0.0095 (8)	0.0080 (9)	-0.0017 (9)
C24	0.0543 (12)	0.0584 (13)	0.0940 (18)	-0.0226 (10)	0.0181 (12)	-0.0237 (13)
C25	0.0369 (9)	0.0826 (15)	0.0677 (13)	-0.0041 (10)	0.0078 (9)	-0.0285 (12)
C26	0.0439 (10)	0.0877 (17)	0.0748 (15)	0.0026 (10)	-0.0091 (10)	0.0013 (13)
C27	0.0437 (10)	0.0609 (12)	0.0741 (14)	-0.0087 (9)	-0.0059 (9)	0.0121 (11)
C28	0.0528 (11)	0.0680 (13)	0.0557 (12)	0.0042 (10)	-0.0169 (9)	-0.0056 (10)
C29	0.0819 (16)	0.0496 (11)	0.0690 (14)	0.0036 (11)	0.0217 (12)	-0.0115 (10)
C30	0.0459 (10)	0.0580 (12)	0.0730 (14)	0.0081 (9)	0.0069 (9)	0.0132 (10)
C31	0.0683 (13)	0.0578 (12)	0.0674 (14)	-0.0054 (10)	0.0250 (11)	0.0097 (10)
C32	0.0405 (11)	0.129 (3)	0.112 (2)	-0.0109 (14)	0.0027 (13)	-0.055 (2)

Geometric parameters (Å, °)

O1—C15	1.364 (2)	C16—H16B	0.9700
O1—C6	1.4018 (19)	C17—C29	1.529 (3)
O2—C19	1.220 (2)	C17—C18	1.533 (3)
N1—C28	1.457 (2)	C17—C30	1.535 (3)
N1—C1	1.464 (2)	C18—C19	1.503 (3)
N1—C4	1.470 (2)	C18—H18A	0.9700
C1—C7	1.518 (2)	C18—H18B	0.9700
C1—C2	1.539 (2)	C19—C20	1.469 (2)
C1—H1	0.9800	C20—C21	1.509 (2)
C2—C3	1.547 (3)	C21—C22	1.527 (2)
C2—H2A	0.9700	C21—H21	0.9800
C2—H2B	0.9700	C22—C27	1.381 (3)
C3—C4	1.541 (2)	C22—C23	1.383 (2)
C3—H3A	0.9700	C23—C24	1.403 (3)
C3—H3B	0.9700	C23—H23	0.9300
C4—C5	1.528 (2)	C24—C25	1.371 (4)
C4—H4	0.9800	C24—H24	0.9300

C5—C8	1.348 (2)	C25—C26	1.375 (4)
C5—C6	1.460 (2)	C25—C32	1.512 (3)
C6—C7	1.339 (2)	C26—C27	1.384 (3)
C7—C21	1.503 (2)	C26—H26	0.9300
C8—C9	1.468 (2)	C27—H27	0.9300
C8—H8	0.9300	C28—H28A	0.9600
C9—C14	1.395 (2)	C28—H28B	0.9600
C9—C10	1.397 (2)	C28—H28C	0.9600
C10—C11	1.385 (3)	C29—H29A	0.9600
C10—H10	0.9300	C29—H29B	0.9600
C11—C12	1.384 (3)	C29—H29C	0.9600
C11—H11	0.9300	C30—H30A	0.9600
C12—C13	1.390 (3)	C30—H30B	0.9600
C12—C31	1.509 (3)	C30—H30C	0.9600
C13—C14	1.385 (3)	C31—H31A	0.9600
C13—H13	0.9300	C31—H31B	0.9600
C14—H14	0.9300	C31—H31C	0.9600
C15—C20	1.344 (2)	C32—H32A	0.9600
C15—C16	1.492 (2)	C32—H32B	0.9600
C16—C17	1.537 (2)	C32—H32C	0.9600
C16—H16A	0.9700		
C15—O1—C6	116.89 (12)	C29—C17—C16	109.18 (16)
C28—N1—C1	115.90 (15)	C18—C17—C16	107.11 (15)
C28—N1—C4	115.93 (15)	C30—C17—C16	111.09 (17)
C1—N1—C4	102.00 (13)	C19—C18—C17	114.13 (16)
N1—C1—C7	111.86 (14)	C19—C18—H18A	108.7
N1—C1—C2	100.92 (14)	C17—C18—H18A	108.7
C7—C1—C2	110.16 (15)	C19—C18—H18B	108.7
N1—C1—H1	111.2	C17—C18—H18B	108.7
C7—C1—H1	111.2	H18A—C18—H18B	107.6
C2—C1—H1	111.2	O2—C19—C20	120.85 (17)
C1—C2—C3	103.11 (14)	O2—C19—C18	121.77 (17)
C1—C2—H2A	111.1	C20—C19—C18	117.30 (15)
C3—C2—H2A	111.1	C15—C20—C19	118.89 (15)
C1—C2—H2B	111.1	C15—C20—C21	122.19 (15)
C3—C2—H2B	111.1	C19—C20—C21	118.89 (14)
H2A—C2—H2B	109.1	C7—C21—C20	108.68 (13)
C4—C3—C2	104.38 (14)	C7—C21—C22	110.88 (15)
C4—C3—H3A	110.9	C20—C21—C22	111.55 (13)
C2—C3—H3A	110.9	C7—C21—H21	108.6
C4—C3—H3B	110.9	C20—C21—H21	108.6
C2—C3—H3B	110.9	C22—C21—H21	108.6
H3A—C3—H3B	108.9	C27—C22—C23	117.53 (18)
N1—C4—C5	112.42 (13)	C27—C22—C21	120.54 (16)
N1—C4—C3	100.68 (13)	C23—C22—C21	121.92 (17)
C5—C4—C3	111.48 (15)	C22—C23—C24	120.2 (2)
N1—C4—H4	110.6	C22—C23—H23	119.9

C5—C4—H4	110.6	C24—C23—H23	119.9
C3—C4—H4	110.6	C25—C24—C23	121.7 (2)
C8—C5—C6	122.22 (15)	C25—C24—H24	119.1
C8—C5—C4	125.27 (15)	C23—C24—H24	119.1
C6—C5—C4	112.48 (13)	C24—C25—C26	117.7 (2)
C7—C6—O1	122.44 (15)	C24—C25—C32	121.4 (2)
C7—C6—C5	122.87 (15)	C26—C25—C32	120.9 (3)
O1—C6—C5	114.63 (13)	C25—C26—C27	121.1 (2)
C6—C7—C21	123.02 (15)	C25—C26—H26	119.4
C6—C7—C1	117.77 (15)	C27—C26—H26	119.4
C21—C7—C1	119.21 (14)	C22—C27—C26	121.7 (2)
C5—C8—C9	127.43 (16)	C22—C27—H27	119.2
C5—C8—H8	116.3	C26—C27—H27	119.2
C9—C8—H8	116.3	N1—C28—H28A	109.5
C14—C9—C10	116.96 (15)	N1—C28—H28B	109.5
C14—C9—C8	123.43 (15)	H28A—C28—H28B	109.5
C10—C9—C8	119.55 (15)	N1—C28—H28C	109.5
C11—C10—C9	121.39 (16)	H28A—C28—H28C	109.5
C11—C10—H10	119.3	H28B—C28—H28C	109.5
C9—C10—H10	119.3	C17—C29—H29A	109.5
C10—C11—C12	121.46 (16)	C17—C29—H29B	109.5
C10—C11—H11	119.3	H29A—C29—H29B	109.5
C12—C11—H11	119.3	C17—C29—H29C	109.5
C11—C12—C13	117.25 (17)	H29A—C29—H29C	109.5
C11—C12—C31	121.57 (17)	H29B—C29—H29C	109.5
C13—C12—C31	121.17 (18)	C17—C30—H30A	109.5
C14—C13—C12	121.78 (17)	C17—C30—H30B	109.5
C14—C13—H13	119.1	H30A—C30—H30B	109.5
C12—C13—H13	119.1	C17—C30—H30C	109.5
C13—C14—C9	121.03 (16)	H30A—C30—H30C	109.5
C13—C14—H14	119.5	H30B—C30—H30C	109.5
C9—C14—H14	119.5	C12—C31—H31A	109.5
C20—C15—O1	123.55 (15)	C12—C31—H31B	109.5
C20—C15—C16	125.05 (15)	H31A—C31—H31B	109.5
O1—C15—C16	111.39 (13)	C12—C31—H31C	109.5
C15—C16—C17	112.91 (14)	H31A—C31—H31C	109.5
C15—C16—H16A	109.0	H31B—C31—H31C	109.5
C17—C16—H16A	109.0	C25—C32—H32A	109.5
C15—C16—H16B	109.0	C25—C32—H32B	109.5
C17—C16—H16B	109.0	H32A—C32—H32B	109.5
H16A—C16—H16B	107.8	C25—C32—H32C	109.5
C29—C17—C18	110.96 (18)	H32A—C32—H32C	109.5
C29—C17—C30	109.03 (18)	H32B—C32—H32C	109.5
C18—C17—C30	109.47 (16)		
C28—N1—C1—C7	62.30 (19)	C8—C9—C14—C13	-178.88 (19)
C4—N1—C1—C7	-64.57 (16)	C6—O1—C15—C20	7.7 (2)
C28—N1—C1—C2	179.41 (15)	C6—O1—C15—C16	-171.12 (14)

C4—N1—C1—C2	52.53 (15)	C20—C15—C16—C17	17.0 (3)
N1—C1—C2—C3	-32.38 (17)	O1—C15—C16—C17	-164.20 (15)
C7—C1—C2—C3	85.97 (17)	C15—C16—C17—C29	-167.58 (18)
C1—C2—C3—C4	2.13 (18)	C15—C16—C17—C18	-47.4 (2)
C28—N1—C4—C5	-58.9 (2)	C15—C16—C17—C30	72.1 (2)
C1—N1—C4—C5	67.99 (16)	C29—C17—C18—C19	174.65 (17)
C28—N1—C4—C3	-177.62 (16)	C30—C17—C18—C19	-65.0 (2)
C1—N1—C4—C3	-50.76 (15)	C16—C17—C18—C19	55.6 (2)
C2—C3—C4—N1	28.68 (17)	C17—C18—C19—O2	150.97 (18)
C2—C3—C4—C5	-90.75 (17)	C17—C18—C19—C20	-32.2 (2)
N1—C4—C5—C8	146.10 (17)	O1—C15—C20—C19	-168.96 (15)
C3—C4—C5—C8	-101.7 (2)	C16—C15—C20—C19	9.7 (3)
N1—C4—C5—C6	-36.0 (2)	O1—C15—C20—C21	8.9 (3)
C3—C4—C5—C6	76.16 (18)	C16—C15—C20—C21	-172.43 (16)
C15—O1—C6—C7	-12.9 (2)	O2—C19—C20—C15	174.93 (17)
C15—O1—C6—C5	164.29 (14)	C18—C19—C20—C15	-1.9 (2)
C8—C5—C6—C7	175.52 (17)	O2—C19—C20—C21	-3.0 (2)
C4—C5—C6—C7	-2.4 (2)	C18—C19—C20—C21	-179.88 (15)
C8—C5—C6—O1	-1.7 (2)	C6—C7—C21—C20	13.3 (2)
C4—C5—C6—O1	-179.63 (14)	C1—C7—C21—C20	-167.26 (14)
O1—C6—C7—C21	1.4 (3)	C6—C7—C21—C22	-109.67 (18)
C5—C6—C7—C21	-175.61 (15)	C1—C7—C21—C22	69.79 (19)
O1—C6—C7—C1	-178.09 (15)	C15—C20—C21—C7	-18.3 (2)
C5—C6—C7—C1	4.9 (3)	C19—C20—C21—C7	159.56 (14)
N1—C1—C7—C6	30.9 (2)	C15—C20—C21—C22	104.24 (18)
C2—C1—C7—C6	-80.5 (2)	C19—C20—C21—C22	-77.90 (19)
N1—C1—C7—C21	-148.60 (15)	C7—C21—C22—C27	61.8 (2)
C2—C1—C7—C21	100.01 (17)	C20—C21—C22—C27	-59.5 (2)
C6—C5—C8—C9	-174.93 (16)	C7—C21—C22—C23	-117.10 (19)
C4—C5—C8—C9	2.7 (3)	C20—C21—C22—C23	121.62 (19)
C5—C8—C9—C14	33.7 (3)	C27—C22—C23—C24	0.7 (3)
C5—C8—C9—C10	-149.18 (19)	C21—C22—C23—C24	179.61 (18)
C14—C9—C10—C11	-4.0 (3)	C22—C23—C24—C25	0.4 (3)
C8—C9—C10—C11	178.71 (17)	C23—C24—C25—C26	-1.2 (3)
C9—C10—C11—C12	1.5 (3)	C23—C24—C25—C32	177.8 (2)
C10—C11—C12—C13	1.1 (3)	C24—C25—C26—C27	0.8 (4)
C10—C11—C12—C31	-179.5 (2)	C32—C25—C26—C27	-178.1 (2)
C11—C12—C13—C14	-1.2 (3)	C23—C22—C27—C26	-1.0 (3)
C31—C12—C13—C14	179.5 (2)	C21—C22—C27—C26	-179.9 (2)
C12—C13—C14—C9	-1.4 (3)	C25—C26—C27—C22	0.2 (4)
C10—C9—C14—C13	3.9 (3)		

Hydrogen-bond geometry (\AA , $^\circ$)

Cg is the centroid of the C22–C27 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C8—H8 \cdots O1	0.93	2.37	2.766 (2)	105

C14—H14 \cdots O2 ⁱ	0.93	2.59	3.285 (2)	131
C3—H3B \cdots Cg ⁱ	0.97	2.88	3.795 (2)	152

Symmetry code: (i) $x-1/2, -y+3/2, -z$.

(12*E*)-12-(4-Bromobenzylidene)-73-(4-bromophenyl)-,7,16-trimethyl-10-oxa-16-azatetracyclo[11.2.1.0^{2,11}.0^{4,9}]hexadeca-2(11),4(9)-dien-5-one (II)

Crystal data

C ₃₀ H ₂₉ Br ₂ NO ₂	$F(000) = 1208$
$M_r = 595.36$	$D_x = 1.491 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 16.051 (3) \text{ \AA}$	Cell parameters from 102112 reflections
$b = 9.7793 (17) \text{ \AA}$	$\theta = 3.5\text{--}27.5^\circ$
$c = 17.005 (3) \text{ \AA}$	$\mu = 3.08 \text{ mm}^{-1}$
$\beta = 96.312 (5)^\circ$	$T = 293 \text{ K}$
$V = 2653.1 (8) \text{ \AA}^3$	Block, colourless
$Z = 4$	$0.20 \times 0.17 \times 0.15 \text{ mm}$

Data collection

Bruker SMART APEX CCD area-detector diffractometer	4933 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.091$
ω and φ scans	$\theta_{\text{max}} = 28.3^\circ, \theta_{\text{min}} = 3.3^\circ$
133143 measured reflections	$h = -21 \rightarrow 21$
6556 independent reflections	$k = -13 \rightarrow 13$
	$l = -22 \rightarrow 22$

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.050$	$w = 1/[\sigma^2(F_o^2) + (0.0695P)^2 + 2.235P]$
$wR(F^2) = 0.141$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.09$	$(\Delta/\sigma)_{\text{max}} = 0.004$
6556 reflections	$\Delta\rho_{\text{max}} = 0.54 \text{ e \AA}^{-3}$
316 parameters	$\Delta\rho_{\text{min}} = -0.71 \text{ e \AA}^{-3}$
0 restraints	

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Br1	1.13657 (3)	1.48892 (4)	-0.08618 (3)	0.07424 (18)
Br2	0.33512 (2)	0.92555 (5)	0.21664 (2)	0.05914 (15)
O1	0.74355 (12)	0.8697 (2)	-0.04204 (11)	0.0324 (4)
O2	0.60743 (18)	0.5052 (3)	0.07177 (15)	0.0552 (7)
N1	0.77700 (15)	1.1142 (3)	0.15803 (14)	0.0350 (5)
C1	0.75767 (19)	0.9713 (3)	0.17268 (16)	0.0349 (6)
H1	0.715499	0.963908	0.209986	0.042*

C2	0.8429 (2)	0.9194 (4)	0.21006 (17)	0.0429 (7)
H2A	0.848064	0.821331	0.203353	0.052*
H2B	0.851861	0.941265	0.265987	0.052*
C3	0.90510 (19)	0.9981 (3)	0.16300 (17)	0.0367 (6)
H3A	0.947246	1.044848	0.198476	0.044*
H3B	0.932793	0.936504	0.129481	0.044*
C4	0.84906 (17)	1.1019 (3)	0.11262 (16)	0.0299 (6)
H4	0.877685	1.190003	0.109710	0.036*
C5	0.82292 (16)	1.0451 (3)	0.03006 (16)	0.0278 (5)
C6	0.76206 (16)	0.9339 (3)	0.03146 (15)	0.0277 (5)
C7	0.73047 (16)	0.8929 (3)	0.09691 (15)	0.0280 (5)
C8	0.85228 (18)	1.0867 (3)	-0.03686 (16)	0.0319 (6)
H8	0.827454	1.047826	-0.083593	0.038*
C9	0.91934 (18)	1.1868 (3)	-0.04467 (17)	0.0327 (6)
C10	0.9139 (2)	1.2727 (4)	-0.11047 (19)	0.0451 (8)
H10	0.866585	1.268859	-0.147241	0.054*
C11	0.9776 (2)	1.3637 (4)	-0.1222 (2)	0.0529 (9)
H11	0.972788	1.420918	-0.166140	0.064*
C12	1.0477 (2)	1.3684 (3)	-0.0684 (2)	0.0443 (8)
C13	1.05595 (19)	1.2852 (3)	-0.0033 (2)	0.0445 (7)
H13	1.103820	1.289674	0.032729	0.053*
C14	0.99204 (18)	1.1939 (3)	0.00834 (18)	0.0385 (7)
H14	0.997839	1.136580	0.052242	0.046*
C15	0.69997 (16)	0.7497 (3)	-0.04300 (15)	0.0280 (5)
C16	0.69721 (19)	0.6819 (3)	-0.12194 (16)	0.0370 (7)
H16A	0.749908	0.634979	-0.125406	0.044*
H16B	0.691302	0.751430	-0.162876	0.044*
C17	0.6257 (2)	0.5797 (3)	-0.13722 (18)	0.0415 (7)
C18	0.6246 (3)	0.4881 (4)	-0.0646 (2)	0.0509 (9)
H18A	0.573307	0.434798	-0.070523	0.061*
H18B	0.671080	0.424447	-0.063439	0.061*
C19	0.63029 (19)	0.5615 (3)	0.01397 (18)	0.0367 (6)
C20	0.66752 (16)	0.6980 (3)	0.01977 (16)	0.0283 (5)
C21	0.67489 (16)	0.7691 (3)	0.09920 (15)	0.0274 (5)
H21	0.702592	0.705638	0.138323	0.033*
C22	0.59016 (16)	0.8069 (3)	0.12632 (15)	0.0276 (5)
C23	0.57771 (17)	0.7910 (3)	0.20543 (16)	0.0332 (6)
H23	0.620545	0.755204	0.240721	0.040*
C24	0.50238 (18)	0.8278 (3)	0.23264 (17)	0.0363 (6)
H24	0.494637	0.817617	0.285727	0.044*
C25	0.43931 (17)	0.8796 (3)	0.17940 (18)	0.0354 (6)
C26	0.44916 (19)	0.8963 (4)	0.10059 (18)	0.0412 (7)
H26	0.405744	0.930913	0.065491	0.049*
C27	0.52543 (19)	0.8603 (3)	0.07442 (17)	0.0382 (7)
H27	0.533120	0.872186	0.021438	0.046*
C28	0.7072 (2)	1.1975 (4)	0.1221 (2)	0.0519 (9)
H28A	0.726219	1.289261	0.114943	0.078*
H28B	0.686260	1.159519	0.071727	0.078*

H28C	0.663230	1.198414	0.156123	0.078*
C29	0.6372 (3)	0.4933 (4)	-0.2101 (2)	0.0601 (11)
H29A	0.591792	0.429284	-0.219197	0.090*
H29B	0.637705	0.551816	-0.255393	0.090*
H29C	0.689284	0.444488	-0.201590	0.090*
C30	0.5420 (2)	0.6584 (5)	-0.1527 (2)	0.0626 (11)
H30A	0.496677	0.594438	-0.162336	0.094*
H30B	0.533914	0.713077	-0.107253	0.094*
H30C	0.543537	0.716471	-0.197980	0.094*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Br1	0.0454 (2)	0.0471 (2)	0.1313 (4)	-0.01677 (16)	0.0145 (2)	0.0214 (2)
Br2	0.03657 (19)	0.0783 (3)	0.0654 (3)	0.01380 (17)	0.01814 (16)	0.00151 (19)
O1	0.0402 (11)	0.0323 (10)	0.0253 (9)	-0.0166 (8)	0.0059 (8)	-0.0024 (8)
O2	0.0722 (17)	0.0448 (14)	0.0517 (14)	-0.0212 (12)	0.0207 (13)	0.0075 (11)
N1	0.0362 (13)	0.0354 (13)	0.0337 (12)	-0.0015 (10)	0.0056 (10)	-0.0114 (10)
C1	0.0394 (15)	0.0426 (16)	0.0234 (13)	-0.0106 (13)	0.0066 (11)	-0.0059 (12)
C2	0.0471 (18)	0.0498 (19)	0.0301 (14)	-0.0089 (14)	-0.0039 (13)	0.0040 (13)
C3	0.0332 (14)	0.0429 (17)	0.0324 (14)	-0.0025 (12)	-0.0036 (11)	-0.0013 (12)
C4	0.0300 (13)	0.0301 (13)	0.0294 (13)	-0.0059 (11)	0.0018 (10)	-0.0052 (11)
C5	0.0275 (12)	0.0268 (13)	0.0286 (13)	-0.0040 (10)	0.0007 (10)	-0.0023 (10)
C6	0.0287 (12)	0.0276 (13)	0.0260 (12)	-0.0067 (10)	0.0002 (10)	-0.0020 (10)
C7	0.0260 (12)	0.0318 (13)	0.0262 (12)	-0.0039 (10)	0.0036 (10)	-0.0035 (11)
C8	0.0342 (14)	0.0321 (14)	0.0289 (13)	-0.0085 (11)	0.0014 (11)	-0.0017 (11)
C9	0.0354 (14)	0.0273 (13)	0.0360 (14)	-0.0064 (11)	0.0064 (11)	-0.0034 (11)
C10	0.0436 (17)	0.0460 (18)	0.0444 (17)	-0.0138 (14)	-0.0003 (14)	0.0095 (14)
C11	0.055 (2)	0.0427 (19)	0.061 (2)	-0.0121 (16)	0.0045 (17)	0.0216 (16)
C12	0.0358 (15)	0.0281 (15)	0.070 (2)	-0.0075 (12)	0.0128 (15)	0.0017 (15)
C13	0.0317 (15)	0.0413 (17)	0.060 (2)	-0.0050 (13)	0.0021 (14)	-0.0040 (15)
C14	0.0361 (15)	0.0357 (16)	0.0439 (16)	-0.0044 (12)	0.0054 (13)	0.0050 (13)
C15	0.0258 (12)	0.0285 (13)	0.0300 (13)	-0.0077 (10)	0.0035 (10)	-0.0024 (11)
C16	0.0402 (15)	0.0423 (16)	0.0303 (14)	-0.0176 (13)	0.0123 (12)	-0.0092 (12)
C17	0.0429 (16)	0.0484 (18)	0.0340 (15)	-0.0232 (14)	0.0079 (13)	-0.0104 (13)
C18	0.065 (2)	0.0374 (18)	0.0516 (19)	-0.0236 (16)	0.0127 (17)	-0.0066 (15)
C19	0.0369 (15)	0.0321 (15)	0.0422 (16)	-0.0084 (12)	0.0095 (12)	0.0013 (13)
C20	0.0267 (12)	0.0274 (13)	0.0317 (13)	-0.0044 (10)	0.0075 (10)	0.0009 (11)
C21	0.0264 (12)	0.0304 (13)	0.0256 (12)	-0.0019 (10)	0.0036 (10)	0.0043 (10)
C22	0.0267 (12)	0.0273 (13)	0.0294 (13)	-0.0032 (10)	0.0052 (10)	0.0044 (10)
C23	0.0315 (13)	0.0398 (15)	0.0288 (13)	0.0026 (12)	0.0048 (11)	0.0076 (12)
C24	0.0373 (15)	0.0426 (16)	0.0305 (14)	0.0011 (13)	0.0107 (12)	0.0044 (12)
C25	0.0285 (13)	0.0360 (15)	0.0430 (16)	0.0017 (11)	0.0093 (12)	0.0006 (12)
C26	0.0333 (15)	0.0524 (19)	0.0374 (15)	0.0102 (13)	0.0014 (12)	0.0091 (14)
C27	0.0381 (15)	0.0481 (18)	0.0287 (14)	0.0040 (13)	0.0052 (12)	0.0097 (13)
C28	0.0456 (18)	0.049 (2)	0.061 (2)	0.0079 (16)	0.0042 (16)	-0.0160 (17)
C29	0.074 (3)	0.059 (2)	0.049 (2)	-0.037 (2)	0.0153 (18)	-0.0224 (17)
C30	0.0434 (19)	0.088 (3)	0.054 (2)	-0.014 (2)	-0.0024 (16)	-0.001 (2)

Geometric parameters (Å, °)

Br1—C12	1.899 (3)	C15—C20	1.338 (4)
Br2—C25	1.906 (3)	C15—C16	1.493 (4)
O1—C15	1.365 (3)	C16—C17	1.524 (4)
O1—C6	1.401 (3)	C16—H16A	0.9700
O2—C19	1.218 (4)	C16—H16B	0.9700
N1—C1	1.458 (4)	C17—C29	1.528 (5)
N1—C4	1.464 (4)	C17—C18	1.528 (5)
N1—C28	1.464 (4)	C17—C30	1.544 (5)
C1—C7	1.522 (4)	C18—C19	1.510 (5)
C1—C2	1.531 (4)	C18—H18A	0.9700
C1—H1	0.9800	C18—H18B	0.9700
C2—C3	1.550 (4)	C19—C20	1.462 (4)
C2—H2A	0.9700	C20—C21	1.512 (4)
C2—H2B	0.9700	C21—C22	1.529 (4)
C3—C4	1.551 (4)	C21—H21	0.9800
C3—H3A	0.9700	C22—C27	1.389 (4)
C3—H3B	0.9700	C22—C23	1.390 (4)
C4—C5	1.525 (4)	C23—C24	1.389 (4)
C4—H4	0.9800	C23—H23	0.9300
C5—C8	1.342 (4)	C24—C25	1.378 (4)
C5—C6	1.464 (4)	C24—H24	0.9300
C6—C7	1.335 (4)	C25—C26	1.376 (4)
C7—C21	1.507 (4)	C26—C27	1.393 (4)
C8—C9	1.472 (4)	C26—H26	0.9300
C8—H8	0.9300	C27—H27	0.9300
C9—C10	1.394 (4)	C28—H28A	0.9600
C9—C14	1.396 (4)	C28—H28B	0.9600
C10—C11	1.386 (4)	C28—H28C	0.9600
C10—H10	0.9300	C29—H29A	0.9600
C11—C12	1.371 (5)	C29—H29B	0.9600
C11—H11	0.9300	C29—H29C	0.9600
C12—C13	1.368 (5)	C30—H30A	0.9600
C13—C14	1.391 (4)	C30—H30B	0.9600
C13—H13	0.9300	C30—H30C	0.9600
C14—H14	0.9300		
C15—O1—C6	117.0 (2)	C15—C16—H16B	109.0
C1—N1—C4	101.9 (2)	C17—C16—H16B	109.0
C1—N1—C28	115.9 (3)	H16A—C16—H16B	107.8
C4—N1—C28	116.0 (3)	C16—C17—C29	110.2 (3)
N1—C1—C7	112.6 (2)	C16—C17—C18	108.8 (3)
N1—C1—C2	101.0 (2)	C29—C17—C18	110.2 (3)
C7—C1—C2	110.1 (3)	C16—C17—C30	109.1 (3)
N1—C1—H1	110.9	C29—C17—C30	108.5 (3)
C7—C1—H1	110.9	C18—C17—C30	110.1 (3)
C2—C1—H1	110.9	C19—C18—C17	115.5 (3)

C1—C2—C3	102.6 (2)	C19—C18—H18A	108.4
C1—C2—H2A	111.3	C17—C18—H18A	108.4
C3—C2—H2A	111.3	C19—C18—H18B	108.4
C1—C2—H2B	111.3	C17—C18—H18B	108.4
C3—C2—H2B	111.3	H18A—C18—H18B	107.5
H2A—C2—H2B	109.2	O2—C19—C20	121.0 (3)
C4—C3—C2	104.1 (2)	O2—C19—C18	120.3 (3)
C4—C3—H3A	110.9	C20—C19—C18	118.6 (3)
C2—C3—H3A	110.9	C15—C20—C19	118.9 (2)
C4—C3—H3B	110.9	C15—C20—C21	122.7 (2)
C2—C3—H3B	110.9	C19—C20—C21	118.1 (2)
H3A—C3—H3B	109.0	C7—C21—C20	109.6 (2)
N1—C4—C5	111.2 (2)	C7—C21—C22	111.3 (2)
N1—C4—C3	101.9 (2)	C20—C21—C22	113.3 (2)
C5—C4—C3	110.7 (2)	C7—C21—H21	107.5
N1—C4—H4	110.9	C20—C21—H21	107.5
C5—C4—H4	110.9	C22—C21—H21	107.5
C3—C4—H4	110.9	C27—C22—C23	118.7 (3)
C8—C5—C6	122.4 (2)	C27—C22—C21	121.8 (2)
C8—C5—C4	125.7 (2)	C23—C22—C21	119.5 (2)
C6—C5—C4	111.9 (2)	C24—C23—C22	121.1 (3)
C7—C6—O1	123.1 (2)	C24—C23—H23	119.4
C7—C6—C5	123.5 (2)	C22—C23—H23	119.4
O1—C6—C5	113.2 (2)	C25—C24—C23	118.7 (3)
C6—C7—C21	122.9 (2)	C25—C24—H24	120.7
C6—C7—C1	117.0 (2)	C23—C24—H24	120.7
C21—C7—C1	119.9 (2)	C26—C25—C24	121.9 (3)
C5—C8—C9	127.3 (3)	C26—C25—Br2	119.6 (2)
C5—C8—H8	116.3	C24—C25—Br2	118.4 (2)
C9—C8—H8	116.3	C25—C26—C27	118.7 (3)
C10—C9—C14	117.5 (3)	C25—C26—H26	120.7
C10—C9—C8	119.4 (3)	C27—C26—H26	120.7
C14—C9—C8	122.9 (3)	C22—C27—C26	121.0 (3)
C11—C10—C9	121.4 (3)	C22—C27—H27	119.5
C11—C10—H10	119.3	C26—C27—H27	119.5
C9—C10—H10	119.3	N1—C28—H28A	109.5
C12—C11—C10	119.4 (3)	N1—C28—H28B	109.5
C12—C11—H11	120.3	H28A—C28—H28B	109.5
C10—C11—H11	120.3	N1—C28—H28C	109.5
C13—C12—C11	121.2 (3)	H28A—C28—H28C	109.5
C13—C12—Br1	119.4 (2)	H28B—C28—H28C	109.5
C11—C12—Br1	119.4 (3)	C17—C29—H29A	109.5
C12—C13—C14	119.4 (3)	C17—C29—H29B	109.5
C12—C13—H13	120.3	H29A—C29—H29B	109.5
C14—C13—H13	120.3	C17—C29—H29C	109.5
C13—C14—C9	121.1 (3)	H29A—C29—H29C	109.5
C13—C14—H14	119.4	H29B—C29—H29C	109.5
C9—C14—H14	119.4	C17—C30—H30A	109.5

C20—C15—O1	124.0 (2)	C17—C30—H30B	109.5
C20—C15—C16	125.0 (2)	H30A—C30—H30B	109.5
O1—C15—C16	110.9 (2)	C17—C30—H30C	109.5
C15—C16—C17	113.1 (2)	H30A—C30—H30C	109.5
C15—C16—H16A	109.0	H30B—C30—H30C	109.5
C17—C16—H16A	109.0		
C4—N1—C1—C7	-63.9 (3)	C8—C9—C14—C13	176.6 (3)
C28—N1—C1—C7	63.1 (3)	C6—O1—C15—C20	6.7 (4)
C4—N1—C1—C2	53.6 (2)	C6—O1—C15—C16	-170.6 (2)
C28—N1—C1—C2	-179.4 (3)	C20—C15—C16—C17	23.8 (4)
N1—C1—C2—C3	-36.9 (3)	O1—C15—C16—C17	-158.9 (3)
C7—C1—C2—C3	82.4 (3)	C15—C16—C17—C29	-168.1 (3)
C1—C2—C3—C4	8.1 (3)	C15—C16—C17—C18	-47.3 (4)
C1—N1—C4—C5	70.2 (3)	C15—C16—C17—C30	72.8 (3)
C28—N1—C4—C5	-56.7 (3)	C16—C17—C18—C19	47.9 (4)
C1—N1—C4—C3	-47.8 (2)	C29—C17—C18—C19	168.7 (3)
C28—N1—C4—C3	-174.7 (2)	C30—C17—C18—C19	-71.6 (4)
C2—C3—C4—N1	23.3 (3)	C17—C18—C19—O2	159.9 (3)
C2—C3—C4—C5	-95.1 (3)	C17—C18—C19—C20	-23.2 (5)
N1—C4—C5—C8	140.3 (3)	O1—C15—C20—C19	-173.2 (3)
C3—C4—C5—C8	-107.1 (3)	C16—C15—C20—C19	3.7 (4)
N1—C4—C5—C6	-41.3 (3)	O1—C15—C20—C21	0.8 (4)
C3—C4—C5—C6	71.2 (3)	C16—C15—C20—C21	177.8 (3)
C15—O1—C6—C7	-8.7 (4)	O2—C19—C20—C15	172.8 (3)
C15—O1—C6—C5	168.1 (2)	C18—C19—C20—C15	-4.1 (4)
C8—C5—C6—C7	-178.4 (3)	O2—C19—C20—C21	-1.5 (4)
C4—C5—C6—C7	3.2 (4)	C18—C19—C20—C21	-178.5 (3)
C8—C5—C6—O1	4.8 (4)	C6—C7—C21—C20	4.1 (4)
C4—C5—C6—O1	-173.6 (2)	C1—C7—C21—C20	-171.7 (2)
O1—C6—C7—C21	3.0 (4)	C6—C7—C21—C22	-122.0 (3)
C5—C6—C7—C21	-173.5 (2)	C1—C7—C21—C22	62.2 (3)
O1—C6—C7—C1	178.9 (2)	C15—C20—C21—C7	-6.0 (4)
C5—C6—C7—C1	2.4 (4)	C19—C20—C21—C7	168.1 (2)
N1—C1—C7—C6	29.9 (4)	C15—C20—C21—C22	118.9 (3)
C2—C1—C7—C6	-82.1 (3)	C19—C20—C21—C22	-67.0 (3)
N1—C1—C7—C21	-154.1 (2)	C7—C21—C22—C27	82.3 (3)
C2—C1—C7—C21	93.9 (3)	C20—C21—C22—C27	-41.7 (4)
C6—C5—C8—C9	-173.3 (3)	C7—C21—C22—C23	-96.0 (3)
C4—C5—C8—C9	4.9 (5)	C20—C21—C22—C23	140.0 (3)
C5—C8—C9—C10	-145.2 (3)	C27—C22—C23—C24	-0.1 (4)
C5—C8—C9—C14	39.5 (5)	C21—C22—C23—C24	178.2 (3)
C14—C9—C10—C11	-1.1 (5)	C22—C23—C24—C25	0.5 (5)
C8—C9—C10—C11	-176.7 (3)	C23—C24—C25—C26	-0.3 (5)
C9—C10—C11—C12	0.6 (6)	C23—C24—C25—Br2	178.4 (2)
C10—C11—C12—C13	-0.1 (6)	C24—C25—C26—C27	-0.4 (5)
C10—C11—C12—Br1	178.0 (3)	Br2—C25—C26—C27	-179.1 (3)
C11—C12—C13—C14	0.1 (5)	C23—C22—C27—C26	-0.6 (5)

Br1—C12—C13—C14	-178.0 (2)	C21—C22—C27—C26	-178.9 (3)
C12—C13—C14—C9	-0.7 (5)	C25—C26—C27—C22	0.8 (5)
C10—C9—C14—C13	1.2 (5)		

*Hydrogen-bond geometry (Å, °)*C_g is the centroid of the C9–C14 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C8—H8 \cdots O1	0.93	2.36	2.742 (3)	105
C30—H30 <i>A</i> \cdots O2 ⁱ	0.96	2.58	3.306 (5)	132
C3—H3 <i>B</i> \cdots C _g ⁱⁱ	0.97	2.84	3.798 (3)	169

Symmetry codes: (i) $-x+1, -y+1, -z$; (ii) $-x+2, -y+2, -z$.