

Whole-molecule disorder of the Schiff base compound 4-chloro-*N*-(4-nitrobenzylidene)aniline: crystal structure and Hirshfeld surface analysis

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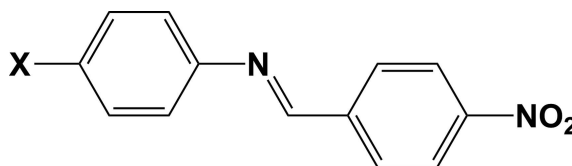
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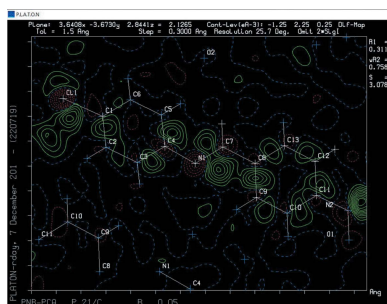
In the crystal of the title Schiff base compound, C₁₃H₉ClN₂O₂, [*E*]-*N*-(4-chlorophenyl)-1-(4-nitrophenyl)methanimine], the CNBA molecule shows whole-molecule disorder (occupancy ratio 0.65:0.35), with the disorder components related by a twofold rotation about the shorter axis of the molecule. The aromatic rings are inclined to each other by 39.3 (5)° in the major component and by 35.7 (9)° in the minor component. In the crystal, C—H···O hydrogen bonds predominate in linking the major components, while weak C—H···Cl interactions predominate in linking the minor components. The result is the formation of corrugated layers lying parallel to the *ac* plane. The crystal packing was analysed using Hirshfeld surface analysis and compared with related structures.

1. Chemical context

A number of benzylideneaniline derivatives crystallize in non-centrosymmetric space groups and are therefore of interest for their non-linear optical properties (Bar & Bernstein, 1977; Batra *et al.*, 2004). In 1970, Bürgi & Dunitz (1970) analysed a number of *N*-benzylideneanilines and found that they were twisted about the N=C bond unlike *trans*-stilbenes (see for example: Behrnd *et al.*, 2010; De Borger *et al.*, 2005) or *trans*-azobenzenes (see for example: Huang *et al.*, 2002; Bushuyev *et al.*, 2016), which are almost planar.



- X = H (HNBA)
- X = F (FNBA)
- X = Cl (CNBA)
- X = Br (BNBA)
- X = CH₃ (MNBA)
- X = CH₃O (MONBA)
- X = OH (HONBA)



Benzylideneaniline derivatives are known to exhibit disorder and Bernstein and collaborators (Bar & Bernstein, 1983; Kluge *et al.*, 2003) have defined the different types of orientational disorder of these compounds, where the molecules may be oriented in different ways but in the two or more dispositions each atom is essentially superimposed on another at any one crystallographic site. Static disorder around the C=N bond is also responsible for the apparent shortening of the C=N bond at room temperature (Bar & Bernstein, 1984). This phenomenon has also been studied by Harada *et al.* (2004a), who, by means of a variable temperature study, concluded that the shortening depends on temperature and is due to a torsional vibration of the C–phenyl and N–phenyl bonds in the crystals.

The crystal structures of a number of disordered benzylideneaniline compounds have been reported on and various forms of the disorder have been analysed (Bar & Bernstein, 1977, 1984; Harada *et al.*, 2004a,b). The disorder appears to fall into three categories (Fig. 1): *D1* – twofold rotation about the longer axis of the molecule, *D2* – the molecule is located about a crystallographic center of symmetry, and *D3* – twofold rotation about the shorter axis of the molecule.

Type *D1* disorder has been observed for one of the two independent molecules in the crystal of *N*-(4-nitrobenzylidene)aniline at 300 and 200 K, but the disorder is not present at 90 K (Harada *et al.*, 2004b). Orientational disorder about a center of symmetry (type *D2*) was found in *N*-(*p*-chlorobenzylidene)-*p*-chloroaniline (Bar & Bernstein, 1982; Bernstein & Schmidt, 1972). Type *D3* disorder has been observed for *N*-benzylideneaniline (Bernstein & Izak, 1976; Harada *et al.*, 2004a) and for 4-methyl-4'-methoxybenzylideneaniline (Harada *et al.*, 2004a).

Three forms of *p*-methyl-*N*-(*p*-methylbenzylidene)aniline (Bernstein, Bar & Christensen, 1976; Bar & Bernstein, 1982; Bar & Bernstein, 1977) have been shown to exist: Form I (Bar & Bernstein, 1982), crystallizes in space group $P2_1/c$ and the C=N bond of the molecule is located about a center of symmetry, hence the molecule has type *D2* disorder; form II (Bar & Bernstein, 1977) crystallizes in space group $P2_1$ and the molecule is not disordered; form III (Bar & Bernstein, 1977; Harada *et al.*, 2004b), has a fourfold disorder with the mol-

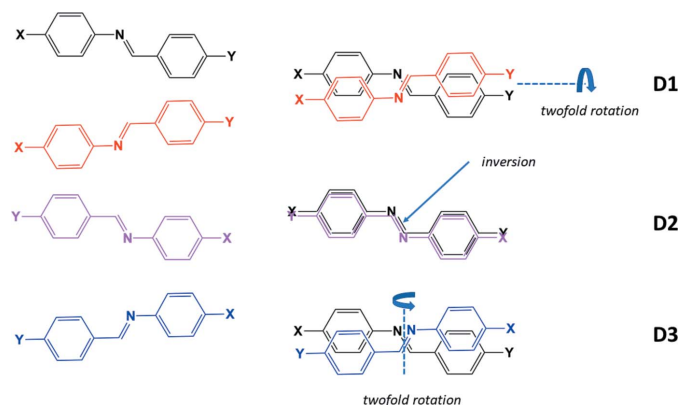


Figure 1
Disorder types in benzylideneanilines.

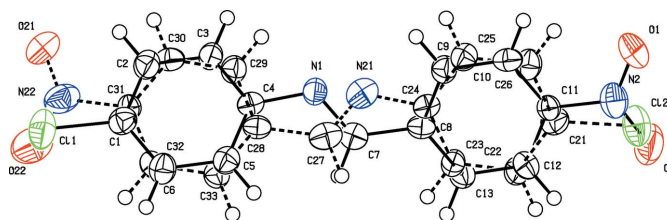


Figure 2
Molecular structure of CNBA, with atom labelling. The displacement ellipsoids are drawn at the 50% probability level. The major component is shown with solid bonds, while the minor component is shown with dashed bonds.

ecule being located about a center of symmetry and has a twofold rotation about the longer axis of the molecule (*D1* + *D2*).

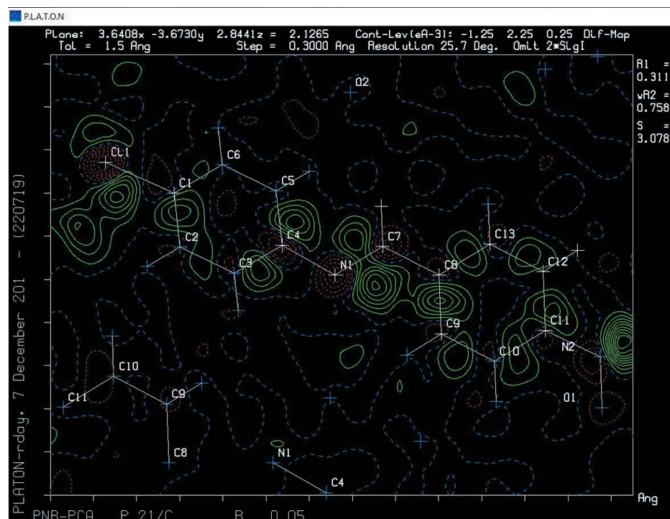
In the past few years some benzylideneaniline compounds have been synthesized using *p*-nitrobenzaldehyde as one of the reactants; for example, 4-nitro-benzylideneaniline (HNBA; Harada *et al.*, 2004b), 4-fluoro-4'-nitro-benzylideneaniline (FNBA; Subashini *et al.*, 2013b), 4-bromo-4'-nitro-benzylideneaniline (BNBA; Subashini *et al.*, 2013a) and 4-hydroxy-4'-nitro-benzylideneaniline [systematic name: 4-[(*E*)-(4-nitrobenzylidene)amino]phenol] (HONBA; Atioğlu *et al.*, 2015).

To continue the series of 4-halogen species, we report herein on the crystal structure of 4-chloro-4'-nitro-benzylideneaniline (CNBA). It was previously synthesized by Batra *et al.* (2004), who found that the crystals they obtained showed good second harmonic generation (SHG) of 1.064 micron wavelength radiation. The crystal structure analysis carried out for CNBA in this work shows that it crystallizes in the centrosymmetric space group $P2_1/c$, and that the molecule has positional disorder (type *D3*), hence no SHG properties are expected for this particular sample. It is interesting to note that the structure of 4-bromo-4'-nitrobenzylideneaniline (BNBA) crystallizes in a non-centrosymmetric space group (*A2*), while the title compound and 4-fluoro-4'-nitrobenzylideneaniline (FNBA; Subashini *et al.*, 2013b) both crystallize in space group $P2_1/c$.

2. Structural commentary

The molecular structure of CNBA is shown in Fig. 2. It crystallizes in the centrosymmetric monoclinic space group $P2_1/c$, and is disordered with a twofold rotation about the shorter axis of the molecule – type *D3*. The twofold axis almost bisects the central C=N bond, so that the two component molecules are superimposed head-to-tail, as shown clearly in a difference-Fourier map (Fig. 3). They have an occupancy ratio that, after initial refinement, was fixed at 0.649:0.351. As mentioned above, this type of disorder (*D3*) has been observed previously for related phases.

The configuration about the C=N bond is *E* in both components. The dihedral angle between the benzene rings of the major component CNBA_1 (C1–C6 and C8–C13) is 38.6 (2)°, and that between rings C21–C26 and C28–C33 of the

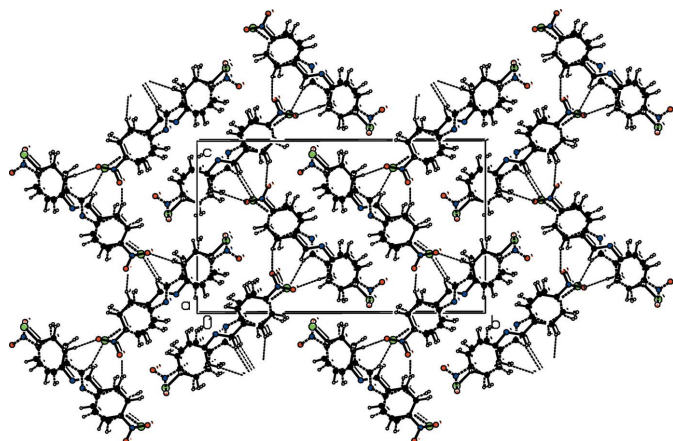

Figure 3

A difference electron-density map showing the density peaks related to the minor disordered component.

minor component (CNBA_2) is $36.5(4)^\circ$. In CNBA_1 the $N1=C7$ bond length is $1.291(6)$ Å, while for CNBA_2 the equivalent $N21=C27$ bond length is $1.234(12)$ Å. The NO_2 group, $N2/O1/O2$, in CNBA_1 is inclined to benzene ring $C8-C13$ by $2.2(7)^\circ$, and atom $C11$ is displaced by $0.016(3)$ Å from benzene ring $C1-C6$. In component CNBA_2, the NO_2 group, $N22/O21/O22$, is inclined to benzene ring $C28-C33$ by $9.0(15)^\circ$, while atom $C12$ lies in the plane of the benzene ring $C21-C26$ [deviation $0.002(5)$ Å].

3. Supramolecular features

A view along the a axis of the crystal packing of CNBA is presented in Fig. 4, and details of the hydrogen bonding are given in Table 1. The crystal packing of the individual components, CBNA_1 and CBNA_2, are given in Fig. 5a and 5b, respectively. In Fig. 5a it can be seen that the molecular


Figure 4

A view along the a axis of the crystal packing of CNBA. The hydrogen bonds (see Table 1) are shown as dashed lines.

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C5-H5\cdots O2^i$	0.95	2.55	3.489 (7)	172
$C13-H13\cdots O1^{ii}$	0.95	2.57	3.426 (5)	151
$C27-H27\cdots Cl2^i$	0.95	2.78	3.661 (11)	155

Symmetry codes: (i) $x, -y + \frac{1}{2}, z - \frac{1}{2}$; (ii) $x + 1, -y + \frac{1}{2}, z - \frac{1}{2}$.

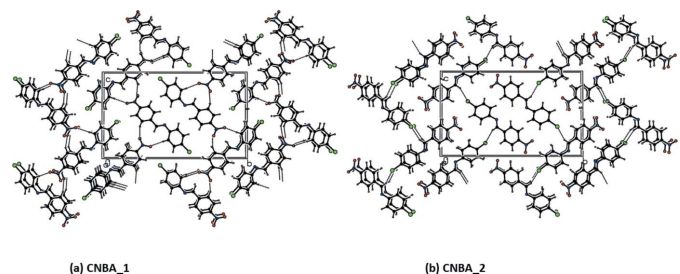
packing for CNBA_1 is influenced by two $C-H\cdots O$ interactions: namely, $C5-H5\cdots O2$ and $C13-H13\cdots O1$. The first of these links the molecules into $C(11)$ chains and the second generates $C(6)$ chains. In Fig. 5b, it can be seen that for CNBA_2 the molecular packing features weak $C-H\cdots Cl$ interactions (Table 1). As a result of these interactions, corrugated layers are formed, lying parallel to the ac plane.

4. Database survey

A search of the Cambridge Structural Database (Version 5.41, last update November 2019; Groom *et al.*, 2016) for $N,1$ -diphenylmethanimines gave 73 hits for 63 compounds, while a search for 1-(4-nitrophenyl)- N -phenylmethanimines gave 25 hits for six compounds. In these searches a number of compounds have multiple reports, or have been studied at different temperatures, or concern polymorphs.

The most relevant compounds that concern us here include those reported above in §1 (*Chemical context*), *viz.* N -(4-nitrobenzylidene)aniline (CSD refcodes QQQAIY01, QQQAIY02, QQQAIY03; Harada *et al.*, 2004b), the 4-fluoro derivative (MIMDUJ; Subashini *et al.*, 2013b), the 4-bromo derivative (FIBXIZ01; Subashini *et al.*, 2013a), the 4-methyl derivative (NMBYAN; Filipenko *et al.*, 1976; NMBYAN22; Filipenko *et al.*, 1977; NMBYAN01; Cole *et al.*, 2001; NMBYAN25, NMBYAN26; Harada *et al.*, 2004a), the 4-methoxy benzylidene derivative (NMBZYA01, NMBZYA02; Harada *et al.*, 2004a) and the 4-hydroxy derivative (WOTQED; Atioğlu *et al.*, 2015).

For N -(4-nitrobenzylidene)aniline measured at 300 K (QQQAIY01), one of the two independent molecules in the asymmetric unit has type $D1$ disorder. At 200 K (QQQAIY02) a difference-Fourier map indicated only a few weak residual density peaks corresponding to the minor


Figure 5

A view along the a axis of the crystal packing of (a) the major disorder component and (b) the minor component. The hydrogen bonds (see Table 1) are shown as dashed lines.

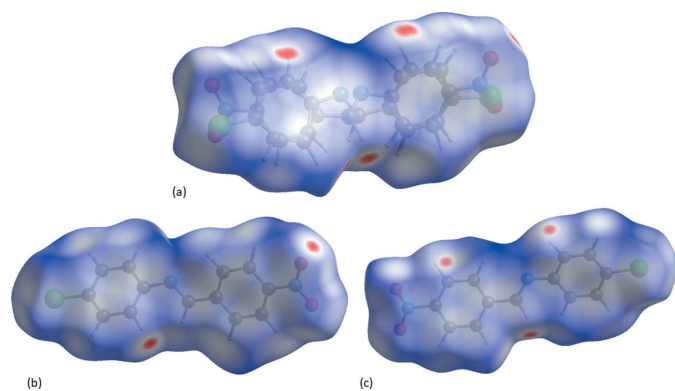


Figure 6
 (a) The Hirshfeld surface of CNBA mapped over d_{norm} , in the colour range -0.15 to 1.13 a.u., (b) the Hirshfeld surface of CNBA_1 mapped over d_{norm} , in the colour range -0.14 to 1.32 a.u., (c) the Hirshfeld surface of CNBA_2 mapped over d_{norm} , in the colour range -0.15 to 1.29 a.u.

component, while at 90 K (QQQAIY03) no disorder was observed. For the 4-fluoro derivative measured at 173 K (MIMDUJ) no disorder was observed. For the 4-bromo derivative (FIBXIZ01), the crystals were incommensurate and twinned and the structure was refined in space group $A2$. A triclinic polymorph of the 4-methyl derivative (NMBYAN) with two independent molecules in the asymmetric unit was reported on by Filipenko *et al.* (1976). A monoclinic polymorph, with one molecule in the asymmetric unit, was reported on first by Filipenko *et al.* (1976) for NMBYAN22, and later a neutron diffraction study at 20 K was carried out by Cole *et al.* (2001) for NMBYAN01. The triclinic polymorph was also studied by Harada *et al.* (2004a), at 300 K (NMBYAN25) and at 90 K (NMBYAN26) and showed only disorder of the methyl hydrogen atoms at both temperatures. The 4-methoxy benzylidene derivative, measured at 300 K

(NMBZYA01) and 90 K (NMBZYA02), showed no disorder at either temperature. Finally, the 4-hydroxy derivative, WOTQED, crystallizes with four independent molecules in the asymmetric unit, and one of the molecules has type $D1$ disorder.

The N=C bond lengths vary from as short as *ca.* 1.187 Å, in one of the four independent molecules of WOTQED, to *ca.* 1.281 Å in NMBZYA02. In the title compound, the N1=C7 bond length in the major component is 1.291 (6) Å, while for the minor component the N21=C27 bond length is 1.234 (12) Å. In the above-mentioned compounds, the benzene rings are inclined to each other by dihedral angles varying from *ca.* 2.24° in one of the independent molecules of WOTQED to *ca.* 55.76° in one of the two independent molecules in NMBYAN26; thus the dihedral angles for the disorder components of the title compound fall roughly in the middle of this range.

5. Hirshfeld surface analysis and two-dimensional fingerprint plots

The Hirshfeld surface analysis (Spackman & Jayatilaka, 2009) and the associated two-dimensional fingerprint plots (McKinnon *et al.*, 2007) were performed with *Crystal-Explorer17* (Turner *et al.*, 2017) following the protocol of Tiekink and collaborators (Tan *et al.*, 2019).

The Hirshfeld surface of CNBA mapped over d_{norm} is given in Fig. 6a, where short interatomic contacts are indicated by the red spots. The Hirshfeld surfaces of the individual components, CNBA_1 and CNBA_2, mapped over d_{norm} are given in Fig. 6b and 6c, respectively.

The full two-dimensional fingerprint plots for CNBA and for the individual components, CNBA_1 and CNBA_2, are given in Fig. 7a, 7b and 7c, respectively. The relative percen-

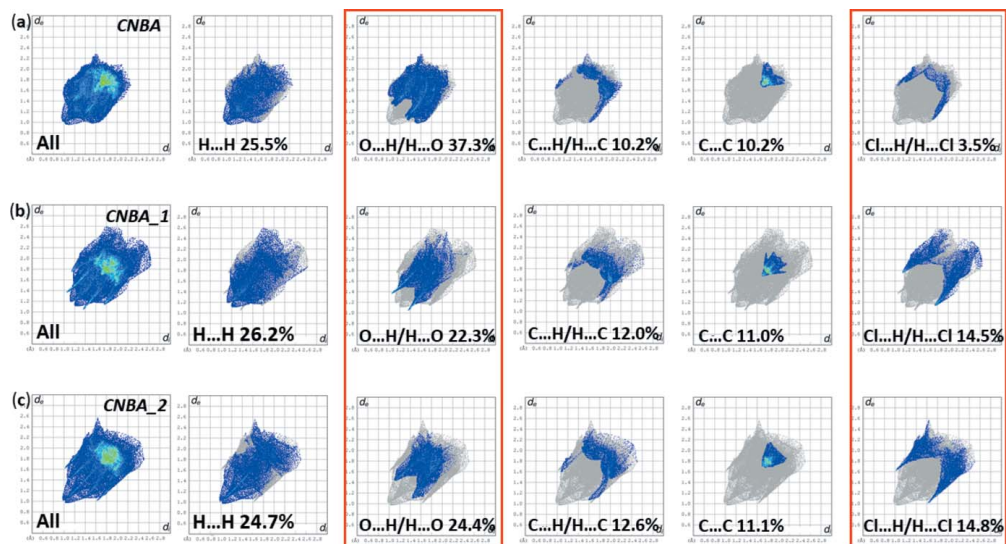


Figure 7
 (a) The full two-dimensional fingerprint plot for CNBA, and fingerprint plots delineated into H...H, O...H/H...O, C...H/H...C, C...C and Cl...H/H...Cl contacts, (b) the full two-dimensional fingerprint plot for CNBA_1, and fingerprint plots delineated into H...H, O...H/H...O, C...H/H...C, C...C and Cl...H/H...Cl contacts, (c) the full two-dimensional fingerprint plot for CNBA_2, and fingerprint plots delineated into H...H, O...H/H...O, C...H/H...C, C...C and Cl...H/H...Cl contacts.

Table 2

Percentage contributions of interatomic contacts to the Hirshfeld surface for CNBA, and for the individual disordered components, CNBA_1 and CNBA_2.

Contact	Percentage contributions		
	CNBA	CNBA_1	CNBA_2
H···H	25.5	26.2	24.7
O···H/H···O	37.3	22.3	24.4
Cl···H/H···Cl	3.5	14.5	14.8
C···H/H···C	10.2	12.0	12.6
C···C	10.2	11.0	11.1
N···H/H···N	3.5	4.2	3.4
C···N	3.2	2.4	2.4
O···O	2.3	0.3	0.6
C···O	1.6	1.9	0.7
O···Cl	0.8	2.0	1.4
C···Cl	0.4	1.0	1.9

Percentage contributions of close contacts to the Hirshfeld surface for CNBA and for the individual components are compared in Table 2. For CNBA the principal intermolecular interactions are delineated into O···H/H···O at 37.3%, H···H at 25.5%, C···H/H···C and C···C both at 10.2%, followed by Cl···H/H···Cl and N···H/H···N contacts both at 3.5%. For CNBA_1 and CNBA_2 the order is somewhat different with H···H contributions being superior or almost equal to the contributions of the O···H/H···O contacts, the latter contributions being 22.3 and 24.4%, respectively, compared to 37.3% for CNBA. In contrast, the Cl···H/H···Cl contacts contribute 14.5% and 14.8% for CNBA_1 and CNBA_2, respectively, compared to only 3.5% for CNBA. This situation reflects the details of the hydrogen bonding in the crystal structure (see Fig. 5a and 5b, and Table 1).

6. Synthesis and crystallization

The commercially available organic compounds *p*-nitrobenzaldehyde and *p*-chloroaniline were used without further purification and the title compound was synthesized following reported procedures (Batra *et al.*, 2004; Subashini *et al.*, 2013a): the two reactants were taken in equimolar ratio and refluxed in ethanol for 6 h. On cooling, the synthesized compound was deposited at room temperature as a deep-yellow microcrystalline powder. The material was purified by repeated recrystallization using ethanol at room temperature and the purity of the sample was confirmed by thin layer chromatography. A saturated solution of CNBA was prepared using mixed solvents of ethanol and ethylacetate (1:1, *v:v*) and single crystals were obtained as yellow rods by slow evaporation of the solvents at room temperature over a period of 18 days. The ¹H NMR spectrum of CNBA recorded in CDCl₃ is shown in the supporting information, Fig. 1S, and the FTIR and FT Raman spectra are shown in Fig. 2S.

7. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 3. The C-bound H atoms were included in calculated positions and refined as riding: C—H =

Table 3

Experimental details.

Crystal data	
Chemical formula	C ₁₃ H ₉ ClN ₂ O ₂
<i>M_r</i>	260.67
Crystal system, space group	Monoclinic, <i>P</i> ₂ ₁ / <i>c</i>
Temperature (K)	173
<i>a</i> , <i>b</i> , <i>c</i> (Å)	3.8195 (5), 22.826 (4), 13.6381 (19)
β (°)	92.829 (11)
<i>V</i> (Å ³)	1187.6 (3)
<i>Z</i>	4
Radiation type	Mo <i>K</i> α
μ (mm ⁻¹)	0.32
Crystal size (mm)	0.34 × 0.17 × 0.09
Data collection	
Diffractometer	STOE IPDS2
Absorption correction	Multi-scan (<i>MULABS</i> ; Spek, 2020)
<i>T</i> _{min} , <i>T</i> _{max}	0.923, 1.000
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	9572, 2256, 1498
<i>R</i> _{int}	0.046
(<i>sin</i> θ / λ) _{max} (Å ⁻¹)	0.610
Refinement	
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.049, 0.121, 0.98
No. of reflections	2256
No. of parameters	206
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}$, $\Delta\rho_{\text{min}}$ (e Å ⁻³)	0.30, -0.19

Computer programs: *X-AREA* and *X-RED32* (Stoe & Cie, 2006), *SHELXS97* (Sheldrick, 2008), *SHELXL2018* (Sheldrick, 2015), *PLATON* (Spek, 2020), *Mercury* (Macrae *et al.*, 2020) and *publCIF* (Westrip, 2010).

0.95 Å with *U*_{iso}(H) = 1.2*U*_{eq}(C). The molecule is disordered with an occupancy ratio that after refinement was fixed at 0.649:0.351. The benzene rings in the two components were refined as rigid bodies and the anisotropic displacement parameters of corresponding C atoms were made equal.

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supporting information

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Whole-molecule disorder of the Schiff base compound 4-chloro-*N*-(4-nitrobenzylidene)aniline: crystal structure and Hirshfeld surface analysis

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Computing details

Data collection: *X-AREA* (Stoe & Cie, 2006); cell refinement: *X-AREA* (Stoe & Cie, 2006); data reduction: *X-RED32* (Stoe & Cie, 2006); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2018* (Sheldrick, 2015); molecular graphics: *PLATON* (Spek, 2020) and *Mercury* (Macrae *et al.*, 2020); software used to prepare material for publication: *SHELXL2018* (Sheldrick, 2015), *PLATON* (Spek, 2020) and *pubCIF* (Westrip, 2010).

4-Chloro-*N*-(4-nitrobenzylidene)aniline

Crystal data

C₁₃H₉ClN₂O₂

M_r = 260.67

Monoclinic, *P2₁/c*

a = 3.8195 (5) Å

b = 22.826 (4) Å

c = 13.6381 (19) Å

β = 92.829 (11)°

V = 1187.6 (3) Å³

Z = 4

F(000) = 536

D_x = 1.458 Mg m⁻³

Mo *K*α radiation, λ = 0.71073 Å

Cell parameters from 3565 reflections

θ = 1.7–26.0°

μ = 0.32 mm⁻¹

T = 173 K

Rod, yellow

0.34 × 0.17 × 0.09 mm

Data collection

STOE IPDS-2

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

phi & ω scans

Absorption correction: multi-scan

(MULABS; Spek, 2020)

T_{min} = 0.923, *T_{max}* = 1.000

9572 measured reflections

2256 independent reflections

1498 reflections with *I* > 2σ(*I*)

R_{int} = 0.046

θ_{\max} = 25.7°, θ_{\min} = 1.7°

h = -4→4

k = -27→27

l = -16→16

Refinement

Refinement on *F*²

Least-squares matrix: full

R [*F*² > 2σ(*F*²)] = 0.049

wR(*F*²) = 0.121

S = 0.98

2256 reflections

206 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier

map

Hydrogen site location: inferred from

neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0668P)^2]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.30 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$$

Extinction correction: (SHELXL2018;
Sheldrick, 2015),

$$F_c^* = kFc[1 + 0.001x\lambda^3/\sin(2\theta)]^{-1/4}$$

Extinction coefficient: 0.036 (4)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C11	1.0727 (7)	0.60002 (12)	0.0780 (3)	0.0575 (7)	0.649
O1	0.0813 (9)	0.23891 (13)	0.7308 (2)	0.0628 (8)	0.649
O2	0.212 (2)	0.1651 (2)	0.6402 (7)	0.069 (2)	0.649
N1	0.7162 (7)	0.42998 (14)	0.3862 (3)	0.0326 (7)	0.649
N2	0.1963 (14)	0.2188 (2)	0.6565 (4)	0.0444 (13)	0.649
C1	0.9720 (11)	0.54801 (15)	0.16579 (17)	0.0369 (11)	0.649
C2	1.0661 (9)	0.55941 (11)	0.26368 (19)	0.0379 (10)	0.649
H2	1.184626	0.594746	0.281349	0.045*	0.649
C3	0.9869 (8)	0.51912 (12)	0.33572 (13)	0.0334 (8)	0.649
H3	1.051256	0.526918	0.402629	0.040*	0.649
C4	0.8136 (8)	0.46743 (11)	0.30987 (16)	0.0297 (8)	0.649
C5	0.7194 (11)	0.45602 (14)	0.21198 (19)	0.0353 (10)	0.649
H5	0.600963	0.420689	0.194313	0.042*	0.649
C6	0.7987 (12)	0.49631 (18)	0.13994 (13)	0.0391 (11)	0.649
H6	0.734329	0.488517	0.073031	0.047*	0.649
C7	0.7008 (13)	0.3742 (2)	0.3713 (5)	0.0335 (11)	0.649
H7	0.774112	0.358824	0.310888	0.040*	0.649
C8	0.5727 (10)	0.33367 (11)	0.44542 (19)	0.0305 (10)	0.649
C9	0.4512 (9)	0.35627 (9)	0.5321 (2)	0.0346 (9)	0.649
H9	0.451447	0.397391	0.542816	0.042*	0.649
C10	0.3292 (9)	0.31870 (11)	0.60298 (18)	0.0337 (9)	0.649
H10	0.246154	0.334147	0.662194	0.040*	0.649
C11	0.3289 (11)	0.25854 (11)	0.5872 (2)	0.0278 (9)	0.649
C12	0.4504 (13)	0.23594 (9)	0.5006 (3)	0.0331 (10)	0.649
H12	0.450148	0.194817	0.489855	0.040*	0.649
C13	0.5724 (12)	0.27351 (12)	0.4297 (2)	0.0333 (10)	0.649
H13	0.655443	0.258061	0.370476	0.040*	0.649
Cl2	0.1792 (12)	0.1871 (2)	0.6481 (4)	0.0553 (12)	0.351
O21	1.1934 (18)	0.6433 (2)	0.1833 (4)	0.0704 (17)	0.351
O22	1.091 (5)	0.5975 (7)	0.0487 (13)	0.099 (8)	0.351
N21	0.6719 (15)	0.3947 (3)	0.4304 (6)	0.0429 (14)	0.351
N22	1.095 (2)	0.6024 (4)	0.1364 (8)	0.053 (2)	0.351
C21	0.331 (3)	0.2457 (2)	0.5774 (7)	0.0369 (11)	0.351
C22	0.476 (3)	0.2384 (3)	0.4867 (7)	0.0379 (10)	0.351

H22	0.496956	0.200313	0.459614	0.045*	0.351
C23	0.590 (3)	0.2869 (4)	0.4356 (5)	0.0334 (8)	0.351
H23	0.689079	0.281974	0.373553	0.040*	0.351
C24	0.559 (2)	0.3427 (3)	0.4752 (5)	0.0297 (8)	0.351
C25	0.414 (2)	0.3500 (2)	0.5659 (5)	0.0353 (10)	0.351
H25	0.393402	0.388076	0.592940	0.042*	0.351
C26	0.300 (2)	0.3015 (3)	0.6170 (4)	0.0391 (11)	0.351
H26	0.201277	0.306416	0.679004	0.047*	0.351
C27	0.652 (2)	0.3978 (5)	0.3399 (7)	0.039 (2)	0.351
H27	0.556380	0.365770	0.303238	0.047*	0.351
C28	0.7743 (17)	0.4503 (2)	0.2864 (5)	0.0305 (10)	0.351
C29	0.9255 (16)	0.4963 (3)	0.3401 (3)	0.0346 (9)	0.351
H29	0.953565	0.493632	0.409583	0.042*	0.351
C30	1.0355 (18)	0.5463 (2)	0.2922 (5)	0.0337 (9)	0.351
H30	1.138834	0.577751	0.328889	0.040*	0.351
C31	0.994 (2)	0.5502 (3)	0.1906 (5)	0.0278 (9)	0.351
C32	0.843 (2)	0.5042 (4)	0.1369 (3)	0.0331 (10)	0.351
H32	0.815122	0.506868	0.067426	0.040*	0.351
C33	0.733 (2)	0.4542 (3)	0.1848 (5)	0.0333 (10)	0.351
H33	0.629851	0.422748	0.148118	0.040*	0.351

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0600 (10)	0.0436 (10)	0.070 (2)	0.0011 (8)	0.0144 (12)	0.0180 (13)
O1	0.080 (2)	0.070 (2)	0.0402 (16)	-0.0056 (16)	0.0215 (16)	0.0067 (15)
O2	0.086 (4)	0.044 (4)	0.078 (3)	0.004 (3)	0.019 (3)	0.017 (3)
N1	0.0346 (17)	0.0294 (16)	0.0340 (17)	-0.0007 (13)	0.0049 (13)	0.0029 (17)
N2	0.039 (2)	0.048 (3)	0.046 (2)	0.005 (3)	-0.0023 (16)	0.014 (3)
C1	0.038 (2)	0.0319 (18)	0.040 (2)	0.0038 (16)	0.000 (2)	-0.0006 (18)
C2	0.039 (2)	0.0315 (18)	0.044 (3)	0.0009 (15)	0.0093 (17)	0.0016 (15)
C3	0.0362 (19)	0.0280 (18)	0.0363 (19)	-0.0026 (14)	0.0043 (15)	0.0024 (14)
C4	0.0270 (17)	0.0295 (18)	0.0327 (19)	0.0002 (14)	0.0011 (15)	-0.0044 (17)
C5	0.0339 (19)	0.0328 (19)	0.039 (3)	-0.0006 (14)	0.0036 (18)	0.0001 (18)
C6	0.036 (2)	0.039 (2)	0.043 (2)	-0.0028 (17)	0.0006 (16)	0.0048 (17)
C7	0.029 (2)	0.037 (3)	0.034 (3)	0.005 (2)	0.0000 (19)	-0.001 (2)
C8	0.0267 (16)	0.035 (2)	0.0289 (19)	-0.0004 (14)	-0.0039 (16)	-0.0050 (16)
C9	0.0364 (19)	0.0323 (17)	0.035 (2)	-0.0008 (14)	0.0005 (17)	0.0035 (15)
C10	0.0367 (19)	0.0319 (19)	0.0320 (18)	0.0049 (16)	-0.0027 (15)	-0.0071 (16)
C11	0.0271 (19)	0.029 (2)	0.0267 (17)	0.0014 (15)	-0.0009 (14)	0.0031 (16)
C12	0.032 (2)	0.0383 (19)	0.030 (2)	0.0029 (15)	0.0022 (16)	0.0033 (15)
C13	0.0344 (19)	0.033 (2)	0.0329 (18)	0.0014 (17)	0.0026 (15)	0.0015 (15)
C12	0.0509 (15)	0.057 (4)	0.058 (2)	0.007 (3)	0.0044 (13)	0.023 (3)
O21	0.101 (5)	0.041 (3)	0.069 (4)	-0.019 (3)	0.007 (3)	-0.008 (3)
O22	0.153 (13)	0.064 (7)	0.079 (12)	-0.017 (6)	-0.015 (8)	-0.009 (6)
N21	0.036 (3)	0.040 (4)	0.053 (4)	0.002 (3)	0.001 (3)	-0.004 (4)
N22	0.045 (4)	0.052 (5)	0.063 (6)	-0.014 (3)	0.010 (5)	-0.018 (5)
C21	0.038 (2)	0.0319 (18)	0.040 (2)	0.0038 (16)	0.000 (2)	-0.0006 (18)

C22	0.039 (2)	0.0315 (18)	0.044 (3)	0.0009 (15)	0.0093 (17)	0.0016 (15)
C23	0.0362 (19)	0.0280 (18)	0.0363 (19)	-0.0026 (14)	0.0043 (15)	0.0024 (14)
C24	0.0270 (17)	0.0295 (18)	0.0327 (19)	0.0002 (14)	0.0011 (15)	-0.0044 (17)
C25	0.0339 (19)	0.0328 (19)	0.039 (3)	-0.0006 (14)	0.0036 (18)	0.0001 (18)
C26	0.036 (2)	0.039 (2)	0.043 (2)	-0.0028 (17)	0.0006 (16)	0.0048 (17)
C27	0.029 (4)	0.041 (6)	0.048 (6)	0.003 (4)	0.001 (4)	-0.010 (5)
C28	0.0267 (16)	0.035 (2)	0.0289 (19)	-0.0004 (14)	-0.0039 (16)	-0.0050 (16)
C29	0.0364 (19)	0.0323 (17)	0.035 (2)	-0.0008 (14)	0.0005 (17)	0.0035 (15)
C30	0.0367 (19)	0.0319 (19)	0.0320 (18)	0.0049 (16)	-0.0027 (15)	-0.0071 (16)
C31	0.0271 (19)	0.029 (2)	0.0267 (17)	0.0014 (15)	-0.0009 (14)	0.0031 (16)
C32	0.032 (2)	0.0383 (19)	0.030 (2)	0.0029 (15)	0.0022 (16)	0.0033 (15)
C33	0.0344 (19)	0.033 (2)	0.0329 (18)	0.0014 (17)	0.0026 (15)	0.0015 (15)

Geometric parameters (Å, °)

C11—C1	1.743 (4)	C12—C21	1.763 (7)
O1—N2	1.214 (6)	O21—N22	1.182 (9)
O2—N2	1.248 (7)	O22—N22	1.20 (2)
N1—C7	1.291 (6)	N21—C27	1.234 (12)
N1—C4	1.411 (4)	N21—C24	1.412 (9)
N2—C11	1.420 (6)	N22—C31	1.462 (11)
C1—C2	1.3900	C21—C22	1.3900
C1—C6	1.3900	C21—C26	1.3900
C2—C3	1.3900	C22—C23	1.3900
C2—H2	0.9500	C22—H22	0.9500
C3—C4	1.3900	C23—C24	1.3900
C3—H3	0.9500	C23—H23	0.9500
C4—C5	1.3900	C24—C25	1.3900
C5—C6	1.3900	C25—C26	1.3900
C5—H5	0.9500	C25—H25	0.9500
C6—H6	0.9500	C26—H26	0.9500
C7—C8	1.471 (6)	C27—C28	1.488 (11)
C7—H7	0.9500	C27—H27	0.9500
C8—C9	1.3900	C28—C29	1.3900
C8—C13	1.3900	C28—C33	1.3900
C9—C10	1.3900	C29—C30	1.3900
C9—H9	0.9500	C29—H29	0.9500
C10—C11	1.3900	C30—C31	1.3900
C10—H10	0.9500	C30—H30	0.9500
C11—C12	1.3900	C31—C32	1.3900
C12—C13	1.3900	C32—C33	1.3900
C12—H12	0.9500	C32—H32	0.9500
C13—H13	0.9500	C33—H33	0.9500
C7—N1—C4	119.5 (4)	C27—N21—C24	118.5 (9)
O1—N2—O2	122.8 (8)	O21—N22—O22	126.8 (14)
O1—N2—C11	117.9 (4)	O21—N22—C31	117.0 (9)
O2—N2—C11	119.2 (8)	O22—N22—C31	115.9 (12)

C2—C1—C6	120.0	C22—C21—C26	120.0
C2—C1—C11	118.5 (2)	C22—C21—C12	123.6 (5)
C6—C1—C11	121.5 (2)	C26—C21—C12	116.4 (5)
C3—C2—C1	120.0	C23—C22—C21	120.0
C3—C2—H2	120.0	C23—C22—H22	120.0
C1—C2—H2	120.0	C21—C22—H22	120.0
C2—C3—C4	120.0	C22—C23—C24	120.0
C2—C3—H3	120.0	C22—C23—H23	120.0
C4—C3—H3	120.0	C24—C23—H23	120.0
C5—C4—C3	120.0	C25—C24—C23	120.0
C5—C4—N1	122.0 (2)	C25—C24—N21	115.4 (6)
C3—C4—N1	117.8 (2)	C23—C24—N21	124.6 (6)
C4—C5—C6	120.0	C24—C25—C26	120.0
C4—C5—H5	120.0	C24—C25—H25	120.0
C6—C5—H5	120.0	C26—C25—H25	120.0
C5—C6—C1	120.0	C25—C26—C21	120.0
C5—C6—H6	120.0	C25—C26—H26	120.0
C1—C6—H6	120.0	C21—C26—H26	120.0
N1—C7—C8	121.7 (5)	N21—C27—C28	122.1 (10)
N1—C7—H7	119.1	N21—C27—H27	118.9
C8—C7—H7	119.1	C28—C27—H27	118.9
C9—C8—C13	120.0	C29—C28—C33	120.0
C9—C8—C7	119.2 (3)	C29—C28—C27	118.8 (6)
C13—C8—C7	120.8 (3)	C33—C28—C27	121.2 (6)
C8—C9—C10	120.0	C28—C29—C30	120.0
C8—C9—H9	120.0	C28—C29—H29	120.0
C10—C9—H9	120.0	C30—C29—H29	120.0
C11—C10—C9	120.0	C31—C30—C29	120.0
C11—C10—H10	120.0	C31—C30—H30	120.0
C9—C10—H10	120.0	C29—C30—H30	120.0
C10—C11—C12	120.0	C30—C31—C32	120.0
C10—C11—N2	121.8 (3)	C30—C31—N22	122.4 (7)
C12—C11—N2	118.2 (3)	C32—C31—N22	117.5 (7)
C13—C12—C11	120.0	C33—C32—C31	120.0
C13—C12—H12	120.0	C33—C32—H32	120.0
C11—C12—H12	120.0	C31—C32—H32	120.0
C12—C13—C8	120.0	C32—C33—C28	120.0
C12—C13—H13	120.0	C32—C33—H33	120.0
C8—C13—H13	120.0	C28—C33—H33	120.0
C6—C1—C2—C3	0.0	C26—C21—C22—C23	0.0
C11—C1—C2—C3	179.4 (3)	C12—C21—C22—C23	-179.9 (8)
C1—C2—C3—C4	0.0	C21—C22—C23—C24	0.0
C2—C3—C4—C5	0.0	C22—C23—C24—C25	0.0
C2—C3—C4—N1	-175.4 (3)	C22—C23—C24—N21	178.9 (8)
C7—N1—C4—C5	34.9 (4)	C27—N21—C24—C25	-147.7 (7)
C7—N1—C4—C3	-149.8 (4)	C27—N21—C24—C23	33.4 (10)
C3—C4—C5—C6	0.0	C23—C24—C25—C26	0.0

N1—C4—C5—C6	175.2 (3)	N21—C24—C25—C26	-179.0 (7)
C4—C5—C6—C1	0.0	C24—C25—C26—C21	0.0
C2—C1—C6—C5	0.0	C22—C21—C26—C25	0.0
C11—C1—C6—C5	-179.4 (3)	C12—C21—C26—C25	179.9 (7)
C4—N1—C7—C8	-174.8 (3)	C24—N21—C27—C28	-178.3 (6)
N1—C7—C8—C9	2.8 (6)	N21—C27—C28—C29	2.2 (11)
N1—C7—C8—C13	-177.2 (4)	N21—C27—C28—C33	-176.6 (7)
C13—C8—C9—C10	0.0	C33—C28—C29—C30	0.0
C7—C8—C9—C10	-180.0 (4)	C27—C28—C29—C30	-178.8 (7)
C8—C9—C10—C11	0.0	C28—C29—C30—C31	0.0
C9—C10—C11—C12	0.0	C29—C30—C31—C32	0.0
C9—C10—C11—N2	-178.0 (4)	C29—C30—C31—N22	178.0 (8)
O1—N2—C11—C10	0.1 (6)	O21—N22—C31—C30	-4.7 (12)
O2—N2—C11—C10	-177.9 (7)	O22—N22—C31—C30	170.1 (13)
O1—N2—C11—C12	-178.0 (4)	O21—N22—C31—C32	173.3 (7)
O2—N2—C11—C12	4.1 (9)	O22—N22—C31—C32	-11.8 (15)
C10—C11—C12—C13	0.0	C30—C31—C32—C33	0.0
N2—C11—C12—C13	178.1 (4)	N22—C31—C32—C33	-178.1 (8)
C11—C12—C13—C8	0.0	C31—C32—C33—C28	0.0
C9—C8—C13—C12	0.0	C29—C28—C33—C32	0.0
C7—C8—C13—C12	180.0 (4)	C27—C28—C33—C32	178.7 (7)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C5—H5 \cdots O2 ⁱ	0.95	2.55	3.489 (7)	172
C13—H13 \cdots O1 ⁱⁱ	0.95	2.57	3.426 (5)	151
C27—H27 \cdots C12 ⁱ	0.95	2.78	3.661 (11)	155

Symmetry codes: (i) *x*, -*y*+1/2, *z*-1/2; (ii) *x*+1, -*y*+1/2, *z*-1/2.