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Unexpected reactions of NHC*—Cu^I and —Ag^I bromides with potassium thio- or selenocyanate

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The reactions of N-heterocyclic carbene Cu^I and Ag^I halides with potassium thio- or selenocyanate gave unexpected products. The attempted substitution of bromido(1,3-dibenzyl-4,5-diphenylimidazol-2-ylidene)silver reaction (NHC*-Ag-Br) with KSCN yielded bis[bis(1,3-dibenzyl-4,5-diphenylimidazol-2-ylidene)silver(I)] tris(thiocyanato)argentate(I) diethyl ether disolvate, $[Ag(C_{29}H_{24}N_2)_2][Ag(NCS)_3] \cdot 2C_4H_{10}O$ or $[NHC_2Ag]_2[Ag(SCN)_3] \cdot 2Et_2O$, (1), while reaction with KSeCN led to $bis(\mu-1,3-dibenzyl-4,5-diphenyl-2$ selenoimidazole- κ^2 Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole- κSe)silver(I)] dichloromethane hexasolvate, $[Ag_2Br_2(C_{29}H_{24}N_2Se)_4]$. 6CH₂Cl₂ or (NHC*Se)₄Ag₂Br₂·6CH₂Cl₂, (2), via oxidation of the NHC* fragment to 2-selenoimidazole. This oxidation was observed again in the reaction of NHC*-Cu-Br with KSeCN, yielding *catena*-poly[[[(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole- κSe)copper(I)]- μ -cyanido- $\kappa^2 C:N$] acetonitrile monosolvate], {[Cu(CN)($C_{29}H_{24}N_2Se$)]· C_2H_3N }, or NHC*Se-CuCN·CH₃CN, (3). Compound (1) represents an organic/inorganic salt with Ag^{I} in a linear coordination in each of the two cations and in a trigonal coordination in the anion, accompanied by diethyl ether solvent molecules. The tri-blade boomerang-shaped complex anion $[Ag(SCN)_3]^{2-}$ present in (1) is characterized by X-ray diffraction for the first time. Compound (2) comprises an isolated centrosymmetric molecule with Ag^I in a distorted tetrahedral BrSe₃ coordination, together with dichloromethane solvent molecules. Compound (3) exhibits a linear polymeric ${}^{1}_{\infty}$ [Cu-C=N-Cu-] chain structure with a selenoimidazole moiety additionally coordinating to each Cu^I atom, and completed by acetonitrile solvent molecules. Electron densities associated with an additional ether solvent molecule in (1) and two additional dichloromethane solvent molecules in (2) were removed with the SQUEEZE procedure [Spek (2015). Acta Cryst. C71, 9–18] in PLATON.

1. Chemical context

Copper and silver and their compounds exhibit fungicidal properties. For example, a copper dihydroxide suspension in water - The Bordeaux Mixture - is a well-known early fungicide used in vineyards, while metallic silver has been used as an antimicrobial agent to purify drinking water for a long time. Alexander the Great stored drinking water in silver vessels during his military campaigns (White, 2002). Later, silver nitrate was used to treat wounds and infectious diseases even before the discovery of bacteria (Klasen, 2000). Silver sulfadiazine, discovered in the 1960s, was found to be more effective and safer than silver nitrate in treating burn wounds and is currently the most widely used remedy in burn centres (Fox, 1968). Silver is considered nontoxic to mammalian cells within the determined exposure limits of $0.01-0.1 \text{ mg m}^{-3}$ (Drake & Hazelwood, 2005). Nevertheless, silver compounds may cause skin discoloration, known as Argyria (Kim et al., 2009). The active species, the Ag^I cation, inhibits the respiratory path of

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Table 1	
Selected internuclear distances and bond angles $(\text{\AA}, ^{\circ})$ for (1), (2)	and (3)

Atoms	Distance	Atoms	Angle
(1)			
Ag1-C8	2.091 (2)	\$1-Ag3-\$2	127.68 (2)
Ag1-C37	2.085 (2)	\$2-Ag3-\$3	103.96 (2)
Ag2-C66	2.097 (2)	\$3-Ag3-\$1	126.94 (2)
Ag2-C95	2.102(2)	Sum	358.58
Ag3-S1	2.4657 (5)	C8-Ag1-C37	173.06 (8)
Ag3-S2	2.5377 (6)	C66-Ag2-C95	172.01 (7)
Ag3-S3	2.4940 (6)	Ag3-S-C117	100.90(7)
		Ag3-S2-C118	93.45 (8)
Ag3···N3	4.161 (2)	Ag3-S3-C119	105.56 (9)
Ag3···C66	3.069 (2)	S1-C117-N9	177.8 (2)
		S2-C118-N10	179.4 (3)
		S3-C119-N11	177.1 (2)
(2)			
Ag-Se1	2.6899 (4)	Br-Ag-Se1	102.274 (13)
Ag-Se2	2.7677 (4)	Br-Ag-Se2	109.628 (12)
Ag-Se2#1	2.7187 (4)	Br-Ag-Se2#1	126.883 (14)
Ag-Br	2.6631 (4)	Se1-Ag-Se2	110.623 (12)
e		Se1-Ag-Se2#1	100.026 (11)
		Se2-Ag-Se2#1	106.352 (11)
		Ag-Se1-C8	94.72 (8)
		Ag-Se2-C37	100.72 (8)
		Ag#1-Se2-C37	108.43 (8)
		Ag-Se2-Ag#1	73.649 (11)
(3)			
Cu1-Se1	2.3900 (6)	Se1-Cu1-C59	125.06 (12)
Cu1-C59	1.898 (4)	Se1-Cu1-N6#2	110.25 (9)
Cu1-N6#2	1.939 (3)	C59-Cu1-N6#2	124.68 (14)
		Sum	359.99
Cu2-Se2	2.3861 (6)	Se2-Cu2-C60	126.88 (11)
Cu2-C60	1.895 (4)	Se2-Cu2-N5	110.28 (10)
Cu2-N5	1.937 (3)	C60-Cu2-N5	122.84 (15)
		Sum	360.00
		Cu1-Se1-C8	95.96 (10)
		Cu2-Se2-C37	93.84 (11)

Symmetry codes: (#1) -x + 1, -y + 1, -z + 2; (#2) -x + 2, $y + \frac{1}{2}$, $-z + \frac{3}{2}$.

sensitive strain organisms, destroys the cell wall, impairs essential enzymes, obstructs metabolic activity and/or causes RNA and DNA alteration (Silver, 2003; Starodub & Trevors, 1989). This topic saw renewed interest when Youngs and coworkers published the clean synthesis of N-heterocyclic (NHC) silver acetate derivatives from easily accessible imidazolium halides (Liang *et al.*, 2018); NHC–silver acetates exhibit good chemical stability through covalently bonded silver, which results in significant antibiotic activity. Particularly well suited are complexes with benzyl-substituted ligands like 1,3-dibenzyl-4,5-diphenylimidazol-2-ylidene (NHC*;



Lewis structures of the ligands NHC* and NHC*Se.

Fig. 1), e.g. NHC*-Ag-OAc (SBC3) (Patil et al., 2011; Streciwilk et al., 2014: Hackenberg & Tacke, 2014), which combine synthesis from a commercially available precursor (4,5-diphenylimidazole) with high antimicrobial activity in vitro (Sharkey et al., 2012) and in vivo (Browne et al. 2014). In the homologue gold series, the introduction of pseudohalide anions led to stable and bioactive compounds (Dada et al., 2017). In this context, the attempted syntheses of the analogous NHC*-Ag^I and -Cu^I thio- and selenocyanates as potential antimicrobial species led to unexpected results with the formation of bis[bis(1,3-dibenzyl-4,5-diphenylimidazol-2ylidene)silver(I)] tris(thiocyanato)argentate(I) diethyl ether disolvate, (1), bis(μ -1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole- $\kappa^2 Se:Se$)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole- κSe)silver(I)] dichloromethane hexasolvate, (2), and *catena*-poly[[[(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-*kSe*)copper(I)]- μ -cyanido- $\kappa^2 C:N$] acetonitrile monosolvate], $\{[Cu(CN)(C_{29}H_{24}N_2Se)]\cdot C_2H_3N\}_n$ or NHC*Se-CuCN·CH₃CN, (3), the crystal structures of which are reported in this communication.



2. Structural commentary

2.1. $[NHC_{2}^{*}Ag]_{2}[Ag(SCN)_{3}] \cdot 2Et_{2}O$, (1)

Both of the crystallographically distinct $[NHC^*_2Ag]^+$ cations (Ag1 and Ag2) in the organic/inorganic salt show a nearly linear C-Ag-C angle and two almost identical Ag-C bond lengths (Table 1). In between these two bulky cations is located one $[Ag(SCN)_3]^{2-}$ anion with a rare coordination number of three for monovalent silver(I) (Ag3). The corresponding Ag-S bond lengths cover the range 2.4657 (5)-2.5377 (6) Å. The flexibility of the $[Ag(SCN)_3]^{2-}$ anion also shows itself in the bond angles, with Ag-S-C angles ranging from 93.45 (8) to 105.56 (9)°, and S-Ag-S

angles ranging from 104.96 (2) to 127.68 (2) $^{\circ}$ (Table 1). As expected, all three SCN⁻ ligands are virtually linear. The sum of the S-Ag-S bond angles (Table 1) indicates that the anion is almost planar [the deviation of the Ag3 from the leastsquares plane of the three S atoms is 0.1270 (5) Å]. The $[Ag(SCN)_3]^{2-}$ anion is situated between the two crystallographically independent cations, but not in the middle (Fig. 2): cation 1 (Ag1) has a shortest distance of 4.161 (2) Å from N3 to Ag3 (line 2 in Fig. 2), whereas cation 2 (Ag2) has a shortest distance of 3.069 (2) Å from C66 to Ag3 (line 1 in Fig. 2). As a consequence of this close association, the benzyl groups in cation 2 are all aligned away from the anion. Due to its greater distance from the anion, the benzyl groups of cation 1 have greater flexibility, allowing it to take a shape suitable to fill gaps in the packing caused by the constraint on cation 2 (Fig. 3). The remaining gaps are filled by two noncoordinating diethyl ether molecules, one of which is highly disordered and could not be refined in terms of atomic sites. The SQUEEZE option (Spek, 2015) in PLATON was used to compensate for the ill-defined electron density.



2.2. (NHC*Se)₄Ag₂Br₂·6CH₂Cl₂, (2)

Compound (2) is characterized by a molecular structure complemented by dichloromethane solvent molecules. Two Ag^{I} cations and two bridging NHC*Se ligands build up a



Figure 2

The $[Ag(SCN)_3]^{2-}$ anion in (1) with the two closest $[Ag(NHC^*)_2]^+$ cations. Substituents on the imidazole moiety have been omitted for clarity and displacement ellipsoids are drawn at the 50% probability level.



The $[Ag(SCN)_3]^{2-}$ anion in (1) with the two closest $[Ag(NHC^*)_2]^+$ cations. Displacement ellipsoids are drawn at the 50% probability level.

centrosymmetric four-membered Ag_2Se_2 ring. Each silver cation carries a further terminal NHC*Se ligand and a terminal bromide ligand, in each case leading to a coordination number of 4 in the shape of a distorted tetrahedron (Table 1 and Fig. 4). One of the bridging Ag–Se distances is 2.7677 (4) Å, significantly longer than the other [2.7187 (4) Å]



Figure 4

The molecular structure of (2), with phenyl groups represented by their *ipso* carbon atoms only. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry code: (I) -x + 1, -y + 1, -z + 1.]

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Table 2	
Hydrogen-bond geometry (Å, $^{\circ}$) for (1).	

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$\begin{array}{c} C31 - H31 \cdots O1 \\ C32 - H32 \cdots N10^{i} \\ C40 - H40 \cdots S1^{ii} \end{array}$	0.95	2.57	3.361 (4)	141
	0.95	2.43	3.327 (4)	157
	0.95	2.82	3.684 (2)	152

Symmetry codes: (i) -x + 2, -y + 2, -z + 1; (ii) x, y, z - 1.

or the terminal one [2.6899 (4) Å], suggesting that two AgBr(NHC*Se)₂ moieties are weakly attached to each other. The Ag-Se-C angles are all strongly bent (Table 1), as one would expect. The bridging and terminal NHC*Se ligand pairs, as well as the two bromide ligands, end up in pseudo-*trans* positions with respect to each other, allowing an overall compact shape of the uncharged [AgBr(NHC*Se)₂]₂ complex. Gaps in the packing are filled by dichloromethane solvent molecules, two of which were treated with the SQUEEZE option in *PLATON*.

2.3. NHC*Se—CuCN·CH₃CN, (3)

The structure of (3) is polymeric in nature and contains two distinct Cu^I atoms. The backbone of the structure is a linear copper-cyanide polymer $^{1}_{\infty}$ [Cu1-C=N-Cu2-], where every Cu^I atom is also coordinated by selenium from a terminal NHC*Se ligand (Fig. 5). The Cu-Se-C angles are in the same region as those in (2) (Table 1). The carbon and nitrogen atoms of the two cyanide anions can be distinguished, not only by their electron densities, but also by their different bond lengths to Cu^I atoms, with Cu-N shorter by $\simeq 0.04$ Å (Table 1). The relatively bulky NHC*Se ligands, which lead to the rare coordination number of 3 of the two Cu^I cations, move to opposite positions with respect to the copper cyanide polymer, allowing better packing for the overall structure



Figure 6

The $[Ag(SCN)_3]^{2-}$ anion in (1) involved in nonclassical hydrogenbonding interactions, shown as pink dashed lines. Phenyl groups are represented by their *ipso* carbons only and displacement ellipsoids are drawn at the 50% probability level. [Symmetry codes: (I) -x + 2, -y + 2, -z + 1; (II) x, y, z + 1.]

(Fig. 5). The sum of the three angles at Cu1 and Cu2 (Table 1) indicate that the coordination is practically planar at the central copper(I) atoms (the displacement of Cu1 from the least-squares Se1/N6/C5 plane is 0.099 Å and of Cu2 from the least-squares Se2/N5/C60 plane is 0.054 Å). Acetonitrile solvent molecules fill voids in the crystal packing.

All reactions reported here include cleavage of an Ag– or Cu–carbene bond, suggesting that even at room temperature (Cu) or in refluxing dichloromethane (Ag) the targeted



Figure 5

Section of the polymeric structure of (3), with displacement ellipsoids drawn at the 50% probability level. Phenyl groups are represented by their *ipso* carbons only and acetonitrile solvent molecules have been omitted for clarity. [Symmetry codes: (I) -x + 2, $y + \frac{1}{2}$, $-z + \frac{3}{2}$; (II) -x + 2, $y - \frac{1}{2}$, $-z + \frac{3}{2}$.]



Figure 7

View of the crystal structure of (1) along [100]. Displacement ellipsoids are drawn at the 50% probability level and hydrogen-bonding interactions are shown as pink dashed lines.

Table 3Experimental details.

	(1)	(2)	(3)
Crystal data			
Chemical formula	$\frac{[Ag(C_{29}H_{24})}{N_2} = \frac{[Ag(NCS)_3] \cdot 2C_4 H_{10}O}{N_2}$	$[Ag_2-Br_2(C_{29}H_{24}N_2Se)_4]\cdot 6CH_2Cl_2$	$[Cu(CN)(C_{29}H_{24}N_2Se)]\cdot C_2H_3N$
M _r	2248.09	2802.96	610.07
Crystal system, space group	Triclinic, P1	Triclinic, $P\overline{1}$	Monoclinic, $P2_1/c$
Temperature (K)	100	150	100
<i>a</i> , <i>b</i> , <i>c</i> (Å)	14.86462 (8), 19.3714 (1), 19.85451 (9)	13.6265 (1), 14.7422 (1), 16.9397 (2)	13.7704 (3), 14.3398 (3), 28.4102 (7)
$lpha,eta,\gamma(^\circ)$	102.7710 (4), 100.8268 (4), 99.5778 (4)	106.4172 (7), 112.2820 (8), 96.2211 (6)	90, 93.024 (2), 90
$V(\text{\AA}^3)$	5345.84 (5)	2930.04 (5)	5602.2 (2)
Z	2	1	8
Radiation type	Cu Ka	Μο Κα	Μο Κα
$\mu \ (\mathrm{mm}^{-1})$	5.37	2.58	2.11
Crystal size (mm)	$0.32 \times 0.06 \times 0.04$	$0.37 \times 0.26 \times 0.20$	$0.35 \times 0.12 \times 0.11$
Data collection			
Diffractometer	Rigaku SuperNova, Dual, Cu at zero, Atlas	Rigaku SuperNova, Dual, Cu at zero, Atlas	Rigaku SuperNova, Dual, Cu at zero, Atlas
Absorption correction	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2015)	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2015)	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2015)
T_{\min}, T_{\max}	0.437, 0.848	0.517, 0.664	0.643, 0.851
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	140384, 22376, 19576	119415, 12002, 10726	71725, 11392, 10122
R _{int}	0.045	0.037	0.038
$(\sin \theta / \lambda)_{\rm max} ({\rm \AA}^{-1})$	0.632	0.626	0.626
Refinement			
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.030, 0.080, 1.03	0.035, 0.096, 1.03	0.048, 0.119, 1.06
No. of reflections	22376	12002	11392
No. of parameters	1309	654	687
H-atom treatment	H-atom parameters constrained	H-atom parameters constrained	H-atom parameters constrained
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} \ ({ m e} \ { m \AA}^{-3})$	0.98, -1.13	1.05, -1.49	2.04, -0.56

Computer programs: CrysAlis PRO (Rigaku OD, 2015), SHELXT (Sheldrick, 2015a), SHELXL2014 (Sheldrick, 2015b), DIAMOND (Brandenburg, 1999) and publCIF (Westrip, 2010).

NHC*-M-SeCN complexes are not very stable in solution, but are liable to Schlenk-type equilibrium exchange. It is worth mentioning that the syntheses of (2) and (3) require SeCN⁻ acting as a selenating agent similar to selenium powder (Verlinden *et al.*, 2015) or Woollins Reagent (Bockfeld *et al.*, 2017), even under the relatively mild conditions reported here. Thus, neither thiocyanate nor selenocyanate take up their roles as unreactive substituents in these planned substitution reactions. These findings raise questions about the suitability of NHC-M-X (M = Cu and Ag; X = pseudohalide) as drugs because drug molecules need to be stable in solution under ambient conditions.

3. Supramolecular features

In (1), there are some weak nonclassical hydrogen bonds between the cations and the solvent molecules, as detailed in Table 2. One anion has connections to the two cations closest to it, as described above, and one to the cation of an adjacent ion triplet (Fig. 6), linking the ion triplets into a one-dimensional chain. Fig. 7 shows a view of (1) along [100] with these contacts shown as dashed lines. All nonclassical hydrogen bonds in (2) and (3) are intramolecular, and we are not aware of any other noteworthy intermolecular features in these structures.

4. Database survey

To the best of our knowledge, the crystal structure of (1) is the first report of a salt with the tri-blade boomerang-shaped $[Ag(SCN)_3]^{2-}$ ion. The alkaline metals salts $Rb_2Ag(SCN)_3$ and Rb₂Ag(SCN)₃ have one-dimensional polymeric chains as anions rather than isolated $[Ag(SCN)_3]^{2-}$ (Thiele & Kehr, 1984). Hathaway et al. (1970) reported the spectroscopic properties of [Cu(NH₃)₂Ag(SCN)₃] and indicated that they had determined its crystal structure as well. However, in this article, only the space group type $(P\overline{6}2c)$ and the number of formula units (Z = 2) were given, not the crystal structure itself. From what is reported it can be gleaned that the anion must be situated on a $\overline{6}$ position, *i.e.* it is planar and adhering to threefold rotation symmetry. If this is true then this is in stark contrast to the $[Ag(SCN)_3]^{2-}$ anion reported here, where the Ag-S bond lengths and S-Ag-S and Ag-S-C bond angles cover a wide range. However, since together with the information above only a schematic drawing of the surrounding of the Cu^{II} atom was given, we cannot establish

structural details of the anion in $[Cu(NH_3)_2Ag(SCN)_3]$ with any degree of certainty.

The Ag—Se distances in (2) fall well within the region reported for similar compounds (Perras *et al.*, 2018; Nahra *et al.*, 2018). Remarkably, at least for the neutral compounds, the distances do not depend on whether the coordination number around the silver is 3 or 4: In *N*,*N*-dimesitylselenoimidazole–silver nitrate, the Ag—Se bond lengths range from 2.65 to 2.68 Å for the four-coordinate atom and from 2.63 to 2.71 for the three-coordinate atom (Perras *et al.*, 2018).

A comparison with the compounds reported by Kimani *et al.* (2011) shows a remarkable impact of the cyanide anion on the Cu–Se bond length compared with the corresponding halides. For threefold-coordinated Cu, the distances between Cu and nonbridging Se range from 2.33 to 2.35 Å, whereas both of them in (**3**) are about 2.39 Å (Table 1). This is closer to the Cu– μ -Se distances (2.41–2.42 Å) reported by Kimani *et al.* (2011). In other words, cyanide is the better ligand for Cu^I when compared with halides, and as a result relatively long Cu–Se distances are observed for cyanide derivative (**3**).

5. Synthesis and crystallization

5.1. $[NHC_{2}^{*}Ag]_{2}[Ag(SCN)_{3}] \cdot 2Et_{2}O$, (1)

1,3-Dibenzyl-4,5-diphenylimidazolium bromide (481 mg, 1.00 mmol), silver oxide (116 mg, 0.500 mmol) and potassium thiocyanate (107 mg, 1.10 mmol) were suspended in dichloromethane (35 ml). After stirring for 20 h under reflux, the solution was filtered and the volume of the solvent was reduced to approximately 5 ml. Pentane (40 ml) was then added and a colourless solid precipitated. The product was filtered off, washed with pentane and dried *in vacuo* (yield: 366 mg, 0.647 mmol, 65%) as a colourless powder. ¹H NMR (300 MHz, CDCl₃): δ 1.60 (*s*, 2H, H₂O), 5.49 (*s*, 4H, C_{aliph}-H), 6.90-7.43 (*m*, 20H, C_{ar}-H). IR (ATR, cm⁻¹): 3028 [ν (CH_{ar})], 2924 [ν (CH_{aliph})], 2053 [ν (S-C \equiv N)], 1445, 1348, 1021, 764, 731, 696. Elemental analysis calculated (%): C 63.61, H 4.27, N 7.42; found: C 64.52, H 4.69, N 6.64.

Diethyl ether was diffused into a saturated solution of the crude product in THF; from this solution, that was kept for 10 d at 277 K, the title compound $[NHC*_2Ag]_2[Ag(SCN)_3]$ crystallized in the form of needles of the diethyl ether disolvate.

5.2. $(NHC^*Se)_4Ag_2Br_2 \cdot 6CH_2Cl_2$, (2)

1,3-Dibenzyl-4,5-diphenylimidazolium bromide (481 mg, 1.00 mmol), silver oxide (116 mg, 0.500 mmol) and potassium selenocyanate (159 mg, 1.10 mmol) were suspended in 35 ml of dichloromethane. After stirring for 20 h under reflux, the solution was filtered and the volume of the solvent was reduced to approximately 5 ml. Then 40 ml of pentane were added and a yellow solid precipitated. The product was filtered off, washed with pentane and dried *in vacuo* (yield: 365 mg, 0.159 mmol, 64%) as a yellow powder. ¹H NMR (300 MHz, CDCl₃): δ 5.23–5.57 (*m*, 16H, C_{aliph}–H), 6.77–7.43 (*m*, 80H, C_{ar}–H). IR (ATR, cm⁻¹): 3029 [ν (CH_{ar})], 2926 [ν (CH_{aliph})],

1447, 1402, 764, 732, 696, 520. Elemental analysis calculated (%): C 59.08, H 4.15, N 4.71; found: C 58.39, H 3.86, N 5.08.

A saturated solution of the compound in dichloromethane was prepared at 313 K; from this solution, that was kept for 7 d at 253 K, the title compound $(NHC*Se)_4Ag_2Br_2$ crystallized as clear pale-yellow block-like prisms of the dichloromethane hexasolvate.

5.3. NHC*Se—CuCN·CH₃CN, (3)

(1,3-Dibenzyl-4,5-diphenylimidazol-2-ylidene)copper(I) bromide (270 mg, 0.50 mmol) and potassium selenocyanate (72 mg, 0.50 mmol) were suspended in 15 ml of dichloromethane followed by 5 ml of water. After stirring for 16 h at room temperature (ca 295K) under nitrogen, the solutions were filtered and the two phases separated. The aqueous phase was washed with dichloromethane $(2 \times 10 \text{ ml})$ and the organic phase was washed with deionized water (2 \times 10 ml). The organic phases were combined and dried over magnesium sulfate. The volume of the solvent was reduced to approximately 5 ml before 20 ml of pentane were added and a colourless solid precipitated. The product was filtered off, washed with pentane and dried in vacuo (yield: 121 mg, 0.212 mmol, 42%) as a colourless powder. ¹H NMR (300MHz, CDCl₃): δ 7.29–6.97 (*m*, 20H, CH_{arom}), 5.50 (*s*, 4H, CH_{2benzyl}). IR (ATR, cm⁻¹): 3048 (w), 2105 (s) (ν_{CN}), 1445 (m, v), 695 (s). M.p. 368 K. Elemental analysis calculated (%) for C₃₀H₂₅N₃CuSe: C 63.21, H 4.97, N 7.37; found: C 61.68, H 4.06, N 7.11.

Pentane was diffused into a saturated solution of the crude product in dichloromethane/acetonitrile 1:1 (ν : ν). From this solution, kept for 10 d at 277 K, the title compound [(NHC*Se)CuCN]_{∞} crystallized as needles of the acetonitrile solvate.

6. Refinement details

Crystal data, data collection and structure refinement details are summarized in Table 3. H atoms were placed at calculated positions and treated as riders, with $U_{\rm iso}$ values set at $1.2U_{\rm eq}$ or $1.5U_{\rm eq}$ of their respective bonding partners.

In the crystal structure of (1), one of the phenyl groups was refined as partially disordered over two positions rotated against each other around the *ipso-para* axis [occupancy ratio 0.55 (2):0.45 (2)]. In the crystal structure of (2), one of the dichoromethane solvent molecules was refined over two positions [ratio 0.898 (4):0.102 (4)] due to positional disorder around one C-Cl bond.

The SQUEEZE procedure (Spek, 2015) in *PLATON* was used to treat regions of disordered solvent molecules in (1) and (2) which could not be modelled in terms of atomic sites. In (1), the number of electrons found in these regions in the unit cell, 82, was assigned to two diethyl ether solvent molecules (ideal 84 electrons). In (2), 84 electrons were found and assigned to two solvent molecules of dichloromethane in the unit cell (ideal 84 electrons). Since Z = 2 for (1) and Z = 1 for (2), one solvent molecule of diethyl ether in (1) and two

solvent molecules of dichloromethane in (2) are missing in the final models and the given chemical formulae and other crystal data given in Table 3 take into account these solvent molecules.

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Unexpected reactions of NHC*—Cu¹ and —Ag¹ bromides with potassium thioor selenocyanate

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Computing details

For all structures, data collection: *CrysAlis PRO* (Rigaku OD, 2015); cell refinement: *CrysAlis PRO* (Rigaku OD, 2015); data reduction: *CrysAlis PRO* (Rigaku OD, 2015); program(s) used to solve structure: SHELXT (Sheldrick, 2015a); program(s) used to refine structure: *SHELXL2014* (Sheldrick, 2015b); molecular graphics: *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *publCIF* (Westrip, 2010).

Bis[bis(1,3-dibenzyl-4,5-diphenylimidazol-2-ylidene)silver(I)] tris(thiocyanato)argentate(I) diethyl ether disolvate (1)

Crystal data

 $[Ag(C_{29}H_{24}N_{2})_2][Ag(NCS)_3] \cdot 2C_4H_{10}O$ $M_r = 2248.09$ Triclinic, $P\overline{1}$ a = 14.86462 (8) Å b = 19.3714 (1) Å c = 19.85451 (9) Å a = 102.7710 (4)° $\beta = 100.8268$ (4)° $\gamma = 99.5778$ (4)° V = 5345.84 (5) Å³

Data collection

Rigaku SuperNova, Dual, Cu at zero, Atlas diffractometer Radiation source: micro-focus sealed X-ray tube Detector resolution: 10.3196 pixels mm⁻¹ ω scans Absorption correction: gaussian (CrysAlis PRO; Rigaku OD, 2015) $T_{\min} = 0.437, T_{\max} = 0.848$

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.030$ $wR(F^2) = 0.080$ S = 1.0322376 reflections 1309 parameters Z = 2 F(000) = 2320 $D_x = 1.397 \text{ Mg m}^{-3}$ Cu K\alpha radiation, $\lambda = 1.54184 \text{ Å}$ Cell parameters from 59132 reflections $\theta = 3.7-76.7^{\circ}$ $\mu = 5.37 \text{ mm}^{-1}$ T = 100 KNeedle, colourless $0.32 \times 0.06 \times 0.04 \text{ mm}$

140384 measured reflections 22376 independent reflections 19576 reflections with $I > 2\sigma(I)$ $R_{int} = 0.045$ $\theta_{max} = 76.9^{\circ}, \theta_{min} = 3.4^{\circ}$ $h = -18 \rightarrow 17$ $k = -24 \rightarrow 24$ $l = -24 \rightarrow 25$

0 restraints Primary atom site location: heavy-atom method Secondary atom site location: other Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0438P)^{2} + 3.189P] \qquad \Delta \rho_{max} = 0.98 \text{ e } \text{\AA}^{-3}$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3 \qquad \Delta \rho_{min} = -1.13 \text{ e } \text{\AA}^{-3}$ $(\Delta / \sigma)_{max} = 0.003$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. The structure was first solved by the Patterson method as implemented in SHELXS. The three silver atoms and one sulfur atom were located. The result was subjected to the Tangens Expansion formula as implemented in SHELXS. Almost all non-hydrogen atoms were located. The remainder was found in the difference fourier map from SHELXL refinements. The PLATON SQUEEZE procedure (A. L. Spek. Acta Cryst. C71, 2015, 9-18) was used to treat regions of disordered solvent which could not be modelled in terms of atomic sites. The number of electrons found in these regions, 82, was assigned to 2 molecules of diethylether. 2 diethylethers would give 84 electrons.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
Ag1	0.98105 (2)	0.78382 (2)	-0.03379 (2)	0.02968 (4)	
C1	1.21022 (18)	0.70910 (14)	-0.00838 (13)	0.0331 (5)	
H1	1.2339	0.7517	0.0297	0.040*	
C2	1.2083 (2)	0.64179 (15)	0.00485 (16)	0.0424 (6)	
H2	1.2296	0.6383	0.0519	0.051*	
C3	1.1753 (2)	0.57962 (15)	-0.05035 (18)	0.0507 (7)	
H3	1.1755	0.5334	-0.0415	0.061*	
C4	1.1422 (2)	0.58496 (15)	-0.11839 (18)	0.0519 (7)	
H4	1.1180	0.5423	-0.1562	0.062*	
C5	1.14406 (19)	0.65222 (14)	-0.13158 (13)	0.0364 (5)	
Н5	1.1221	0.6555	-0.1787	0.044*	
C6	1.17778 (15)	0.71508 (12)	-0.07678 (11)	0.0252 (4)	
C7	1.17739 (17)	0.78740 (12)	-0.09409 (11)	0.0266 (4)	
H7A	1.2299	0.7994	-0.1166	0.032*	
H7B	1.1181	0.7839	-0.1284	0.032*	
N1	1.18670 (13)	0.84520 (10)	-0.03008 (9)	0.0227 (3)	
C8	1.11579 (15)	0.85001 (12)	0.00299 (11)	0.0249 (4)	
C9	1.27183 (15)	0.88709 (11)	0.01320 (11)	0.0222 (4)	
C10	1.36079 (17)	0.89170 (12)	-0.01006 (13)	0.0295 (5)	
C11	1.3764 (2)	0.92388 (17)	-0.06298 (14)	0.0426 (6)	
H11	1.3296	0.9450	-0.0850	0.051*	
C12	1.4622 (2)	0.9258 (2)	-0.08513 (16)	0.0554 (9)	
H12	1.4715	0.9461	-0.1234	0.066*	
C13	1.5306 (3)	0.8988 (2)	-0.0515 (2)	0.0630 (9)	
H13	1.5880	0.9003	-0.0662	0.076*	
C14	1.5177 (3)	0.8688 (2)	0.0043 (3)	0.0773 (13)	
H14	1.5669	0.8513	0.0285	0.093*	
C15	1.4337 (2)	0.8643 (2)	0.0246 (2)	0.0573 (9)	
H15	1.4246	0.8427	0.0620	0.069*	
C16	1.25234 (14)	0.92016 (11)	0.07505 (11)	0.0226 (4)	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

C17	1.31529 (15)	0.96909 (12)	0.14136 (11)	0.0256 (4)
C18	1.31124 (18)	1.04121 (14)	0.16332 (13)	0.0344 (5)
H18	1.2679	1.0604	0.1351	0.041*
C19	1.3699 (2)	1.08602 (16)	0.22635 (14)	0.0423 (6)
H19	1.3666	1.1355	0.2406	0.051*
C20	1.4316 (2)	1.05965 (17)	0.26750 (14)	0.0467 (7)
H20	1.4726	1.0908	0.3098	0.056*
C21	1.4348 (3)	0.9874 (2)	0.24781 (18)	0.0632 (10)
H21	1.4764	0.9684	0.2776	0.076*
C22	1.3773 (2)	0.94223 (17)	0.18435 (17)	0.0533 (8)
H22	1.3808	0.8927	0.1706	0.064*
N2	1.15577 (13)	0.89728 (10)	0.06686 (9)	0.0247 (4)
C23	1.10662 (16)	0.91377 (14)	0.12398 (12)	0.0326 (5)
H23A	1.0382	0.8958	0.1046	0.039*
H23B	1.1189	0.9669	0.1437	0.039*
C24	1.13906 (18)	0.87875 (14)	0.18219 (13)	0.0343 (5)
C25	1.1708 (2)	0.92001 (16)	0.25172 (14)	0.0421 (6)
H25	1.1703	0.9702	0.2630	0.051*
C26	1.2033 (3)	0.88852 (19)	0.30501 (15)	0.0544 (8)
H26	1.2244	0.9171	0.3528	0.065*
C27	1.2051 (3)	0.8163 (2)	0.28929 (18)	0.0653 (10)
H27	1.2287	0.7951	0.3259	0.078*
C28	1.1725 (3)	0.77426 (18)	0.21979 (18)	0.0627 (10)
H28	1.1735	0.7241	0.2087	0.075*
C29	1.1384 (2)	0.80514 (15)	0.16650 (15)	0.0467 (7)
H29	1.1146	0.7760	0.1191	0.056*
C30	0.7594 (2)	0.86196 (16)	0.02635 (14)	0.0430 (6)
H30	0.7151	0.8203	0.0260	0.052*
C31	0.7872 (3)	0.9201 (2)	0.08573 (18)	0.0684 (11)
H31	0.7614	0.9181	0.1257	0.082*
C32	0.8500 (4)	0.9794 (2)	0.0880 (2)	0.0846 (17)
H32	0.8678	1.0193	0.1290	0.102*
C33	0.8881 (3)	0.98228 (17)	0.0312 (3)	0.0818 (16)
H33	0.9329	1.0242	0.0331	0.098*
C34	0.8622 (2)	0.92402 (15)	-0.03031 (18)	0.0521 (8)
H34	0.8898	0.9259	-0.0695	0.063*
C35	0.79556 (17)	0.86395 (12)	-0.03230 (12)	0.0312 (5)
C36	0.76054 (18)	0.80201 (12)	-0.09808 (11)	0.0297 (5)
H36A	0.6947	0.8014	-0.1199	0.036*
H36B	0.7986	0.8093	-0.1328	0.036*
N3	0.76572 (13)	0.73222 (10)	-0.08206 (9)	0.0238 (4)
C37	0.84843 (15)	0.71588 (12)	-0.05818 (11)	0.0252 (4)
C38	0.68993 (15)	0.67697 (11)	-0.08559 (10)	0.0220 (4)
C39	0.59068 (15)	0.67846 (11)	-0.11029 (11)	0.0233 (4)
C40	0.55652 (17)	0.68480 (13)	-0.17863 (11)	0.0288 (5)
H40	0.5977	0.6885	-0.2095	0.035*
C41	0.46292 (18)	0.68568 (15)	-0.20165 (13)	0.0358 (5)
H41	0.4404	0.6901	-0.2482	0.043*

C42	0.40217 (18)	0.68017 (16)	-0.15754 (15)	0.0404 (6)	
H42	0.3382	0.6813	-0.1734	0.049*	
C43	0.4349 (2)	0.67302 (18)	-0.08997 (15)	0.0450 (7)	
H43	0.3931	0.6687	-0.0596	0.054*	
C44	0.52835 (18)	0.67213 (15)	-0.06656 (13)	0.0344 (5)	
H44	0.5502	0.6672	-0.0201	0.041*	
C45	0.72822 (15)	0.62438 (12)	-0.06349 (11)	0.0231 (4)	
C46	0.68002 (15)	0.55211 (12)	-0.06161 (11)	0.0250 (4)	
C47A	0.6756 (10)	0.4947 (3)	-0.1191 (3)	0.0382 (19)	0.55(2)
H47A	0.7054	0.5019	-0.1561	0.046*	0.55 (2)
C48A	0.6263 (11)	0.4262 (4)	-0.1213 (4)	0.047 (2)	0.55 (2)
H48A	0.6223	0.3858	-0.1598	0.057*	0.55 (2)
C47B	0.6300 (10)	0.4998 (4)	-0.1238(4)	0.030(2)	0.45 (2)
H47B	0.6306	0.5101	-0.1683	0.036*	0.45(2)
C48B	0.5799 (10)	0.4335 (4)	-0.1225(4)	0.035(2)	0.45(2)
H48B	0.5449	0.3999	-0.1657	0.042*	0.45(2)
C49	0.5803 (2)	0.41677 (14)	-0.06271 (16)	0.0413 (6)	
H49A	0 5496	0 3695	-0.0633	0.050*	0.45(2)
H49	0.5445	0.3710	-0.0637	0.050*	0.15(2) 0.55(2)
C50A	0.5907 (9)	0.4743(5)	-0.0091(7)	0.026	0.55(2)
H50A	0.5638	0.4690	0.0297	0.056*	0.55(2)
C51A	0.6397 (9)	0.5417(5)	-0.0082(6)	0.0385(19)	0.55(2)
H51A	0.6451	0.5817	0.0310	0.046*	0.55(2) 0.55(2)
C50B	0.6251 (10)	0.3017 0.4670 (7)	0.0033 (6)	0.039(2)	0.35(2) 0.45(2)
H50B	0.6214	0.4548	0.0467	0.046*	0.15(2) 0.45(2)
C51B	0.6748 (9)	0.5350 (6)	0.0034 (5)	0.040	0.45(2) 0.45(2)
H51B	0.7052	0.5696	0.0472	0.032 (2)	0.15(2) 0.45(2)
N4	0.82539(13)	0.64955 (10)	-0.04759(9)	0.0253(4)	0.45 (2)
C52	0.82538(16)	0.60961 (13)	-0.02429(12)	0.0209(1) 0.0309(5)	
H52A	0.05558 (10)	0.6241	-0.02429(12)	0.037*	
H52R	0.8699	0.5572	-0.0464	0.037*	
C53	0.0077 (16)	0.5572 0.62147(12)	0.0501(12)	0.037 0.0275(4)	
C54	0.92407(10)	0.02147(12) 0.58529(14)	0.03321(12) 0.08136(15)	0.0273(4)	
H54	1 0170	0.5561	0.0403	0.0407 (0)	
C55	1.0175(2)	0.5501 0 59117 (17)	0.0493 0.15341 (17)	0.0513 (8)	
U55	1.0175 (2)	0.59117 (17)	0.13341 (17)	0.0513 (8)	
C56	1.0025 0.9789 (2)	0.5058 0.6341(2)	0.1700 0.20018 (17)	0.002	
U56	0.9789 (2)	0.6376	0.20018 (17)	0.0500 (9)	
C57	0.9905 0.0145 (2)	0.0370 0.6710 (2)	0.2490	0.000	
H57	0.9145 (2)	0.0719 (2)	0.17551 (10)	0.0555 (8)	
C58	0.0092 0.99662 (19)	0.7023	0.2077 0.10261 (12)	0.007°	
U58	0.00003 (10)	0.00319(17)	0.10201 (13)	0.0387 (0)	
1158	0.0410 0.75271(2)	0.0907	0.0850	0.040°	
Ag2	0.75571(2) 0.76475(15)	0.09309(2) 0.50700(11)	0.34002(2) 0.46307(11)	0.01911(4)	
U50	0.70475(15)	0.50705(11)	0.40307 (11)	0.0223 (4)	
C60	0.0247	0.3304	0.4390	0.027°	
	0.02000 (10)	0.40741 (12)	0.40137 (11)	0.0200 (4)	
C61	0.7100	0.4070 0.42546 (12)	0.3304	0.031°	
001	0.00914 (10)	0.43340 (12)	0.40384 (12)	0.0203 (4)	

H61	0.5633	0.4095	0.3638	0.032*
C62	0.58851 (16)	0.43949 (12)	0.47170 (12)	0.0274 (4)
H62	0.5282	0.4166	0.4747	0.033*
C63	0.65595 (15)	0.47693 (12)	0.53319 (11)	0.0249 (4)
H63	0.6416	0.4796	0.5782	0.030*
C64	0.74483 (14)	0.51063 (10)	0.52924 (11)	0.0198 (4)
C65	0.81645 (14)	0.54679 (11)	0.59853 (10)	0.0199 (4)
H65A	0.7839	0.5696	0.6337	0.024*
H65B	0.8430	0.5090	0.6165	0.024*
N5	0.89381 (11)	0.60198 (9)	0.59393 (8)	0.0176(3)
C66	0 88138 (14)	0.66826 (11)	0.58793(10)	0.0183(4)
C67	0.98878(14)	0.60157 (11)	0.61353 (10)	0.0180(4)
C68	1.02285(14)	0.53624(11)	0.61333(10)	0.0100(1)
C69	0.98347(14)	0.35024(11) 0.46886(11)	0.02292(10) 0.57308(11)	0.0190(4) 0.0218(4)
H60	0.03/3	0.4652	0.5334	0.0218 (4)
C70	1 01507 (16)	0.4052 0.40742 (11)	0.5554	0.020
C70 H70	0.0884	0.40/42 (11)	0.58200 (12)	0.0233(4)
H70	0.9884	0.3019	0.3488	0.031°
C/1 1171	1.08858 (17)	0.41245 (12)	0.03940 (13)	0.0286 (5)
H/I	1.1109	0.3705	0.0455	0.034*
C72	1.12857(17)	0.4/899 (13)	0.68824 (12)	0.0288 (5)
H/2	1.1/82	0.4825	0.7276	0.035*
C/3	1.09603 (15)	0.54048 (12)	0.6/962 (11)	0.0239 (4)
H73	1.1239	0.5859	0.7131	0.029*
C74	1.03660 (14)	0.66983 (10)	0.62029 (10)	0.0176 (4)
C75	1.13793 (14)	0.70362 (10)	0.64412 (10)	0.0190 (4)
C76	1.20124 (15)	0.68387 (11)	0.60387 (11)	0.0229 (4)
H76	1.1799	0.6467	0.5607	0.027*
C77	1.29550 (15)	0.71837 (12)	0.62661 (12)	0.0262 (4)
H77	1.3383	0.7049	0.5988	0.031*
C78	1.32722 (15)	0.77236 (12)	0.68963 (13)	0.0278 (4)
H78	1.3915	0.7963	0.7047	0.033*
C79	1.26479 (16)	0.79140 (12)	0.73066 (11)	0.0265 (4)
H79	1.2868	0.8279	0.7743	0.032*
C80	1.17026 (15)	0.75747 (11)	0.70843 (11)	0.0231 (4)
H80	1.1278	0.7707	0.7367	0.028*
N6	0.96896 (11)	0.70968 (9)	0.60418 (8)	0.0174 (3)
C81	0.98813 (14)	0.78589 (10)	0.60104 (10)	0.0203 (4)
H81A	1.0433	0.8135	0.6395	0.024*
H81B	0.9337	0.8070	0.6093	0.024*
C82	1.00677 (15)	0.79387 (10)	0.53060 (11)	0.0208 (4)
C83	1.09775 (16)	0.80758 (12)	0.52096 (12)	0.0266 (4)
H83	1.1495	0.8135	0.5595	0.032*
C84	1,11334 (18)	0.81269 (13)	0.45497 (14)	0.0337(5)
H84	1.1756	0.8219	0.4487	0.040*
C85	1.0382 (2)	0.80441(14)	0.39856 (13)	0.0379 (6)
H85	1.0488	0.8070	0.3534	0.045*
C86	0.9477(2)	0 79239 (15)	0 40830 (13)	0.0377 (6)
H86	0.8961	0.7876	0 3699	0.045*
1100	0.0701	0.7070	0.0077	0.040

C87	0.93200 (17)	0.78729 (13)	0.47376 (12)	0.0293 (5)
H87	0.8696	0.7792	0.4800	0.035*
C88	0.66481 (17)	0.64638 (13)	0.33252 (12)	0.0301 (5)
H88	0.6923	0.6888	0.3701	0.036*
C89	0.6978 (2)	0.63552 (17)	0.27051 (13)	0.0405 (6)
H89	0.7481	0.6703	0.2662	0.049*
C90	0.6575 (2)	0.57434 (18)	0.21551 (14)	0.0491 (7)
H90	0.6797	0.5672	0.1732	0.059*
C91	0.5850 (3)	0.52357 (17)	0.22198 (14)	0.0540 (8)
H91	0.5574	0.4813	0.1842	0.065*
C92	0.5522 (2)	0.53423 (14)	0.28367(13)	0.0412 (6)
H92	0.5020	0.4993	0.2878	0.049*
C93	0 59255 (16)	0.59583(12)	0.33951(11)	0.0246 (4)
C94	0.55979 (15)	0.60271(11)	0.40821(11)	0.0214(4)
H94A	0 4955	0 5727	0 3977	0.026*
H94B	0.6012	0.5830	0.4408	0.026*
N7	0.55944(12)	0.67719 (9)	0 44431 (8)	0.020
C95	0.63428(14)	0.72101(11)	0.49374(10)	0.0102(3)
C96	0.03120(11) 0.48627(14)	0.72101(11) 0.71328(11)	0.43243(10)	0.0100(1)
C97	0.38961(14)	0.67823(11)	0.39126(11)	0.0202(1) 0.0230(4)
C98	0.36889(17)	0.63429 (14)	0.32221(12)	0.0220 (1) 0.0321 (5)
H98	0.4179	0.6280	0.2984	0.039*
C99	0 27620 (19)	0 59946 (15)	0.29787(14)	0.0408 (6)
H99	0.2625	0 5686	0.2412	0.049*
C100	0.2025 0.20452(17)	0.60977 (14)	0.2112 0.32153(15)	0.0395 (6)
H100	0.1417	0.5856	0.2981	0.0373 (0)
C101	0.22370 (16)	0.65493 (13)	0.38890 (15)	0.0357(5)
H101	0.1740	0.6632	0.4114	0.0337 (3)
C102	0.31620 (16)	0.68858(12)	0.42410(13)	0.0292(5)
H102	0 3293	0.7189	0.4710	0.035*
C103	0 51846 (14)	0.78202(11)	0.47479(10)	0.0198(4)
C104	0 46956 (14)	0.84285(11)	0 48426 (11)	0.0230(4)
C105	0 43043 (16)	0.85935(12)	0.54279(12)	0.0236(1) 0.0276(4)
H105	0.4387	0.8338	0 5784	0.033*
C106	0 37929 (18)	0.91328 (14)	0 54867 (14)	0.0355 (5)
H106	0 3520	0 9244	0 5883	0.044*
C107	0.3679 (2)	0.95088 (15)	0.49722 (17)	0.0437 (6)
H107	0 3328	0.9877	0 5015	0.052*
C108	0.4073(2)	0.93514(17)	0.43954(18)	0.032
H108	0.3997	0.9614	0.4044	0.057*
C109	0.45788(19)	0.88118 (14)	0.43285(14)	0.037
H109	0.4847	0.8703	0.3930	0.043*
N8	0.60883 (12)	0.78514 (9)	0.51169 (8)	0.0188(3)
C110	0.67195 (14)	0.84941 (11)	0.56130 (10)	0.0212 (4)
H11A	0.7090	0.8347	0.6005	0.025*
H11B	0.6342	0.8823	0.5821	0.025*
C111	0.73840 (14)	0.89022 (11)	0.52734 (11)	0.0214 (4)
C112	0.81556 (17)	0.94234 (12)	0.57147 (12)	0.0294 (5)
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H112	0.8272	0.9490	0.6214	0.035*
C113	0.87523 (19)	0.98436 (14)	0.54315 (15)	0.0394 (6)
H113	0.9270	1.0201	0.5738	0.047*
C114	0.85991 (19)	0.97460 (14)	0.47042 (16)	0.0394 (6)
H114	0.9006	1.0039	0.4511	0.047*
C115	0.78475 (18)	0.92178 (14)	0.42575 (13)	0.0333 (5)
H115	0.7748	0.9142	0.3757	0.040*
C116	0.72393 (16)	0.87995 (12)	0.45408 (12)	0.0258 (4)
H116	0.6723	0.8442	0.4233	0.031*
Ag3	0.79941 (2)	0.72828 (2)	0.71384 (2)	0.02795 (4)
S1	0.64189 (4)	0.65134 (3)	0.65701 (3)	0.02631 (10)
C117	0.57874 (15)	0.71425 (12)	0.64988 (11)	0.0240 (4)
N9	0.53253 (14)	0.75603 (11)	0.64467 (11)	0.0297 (4)
S2	0.84839 (4)	0.86528 (3)	0.74115 (3)	0.03126 (12)
C118	0.96150 (18)	0.86638 (12)	0.77086 (12)	0.0295 (5)
N10	1.04011 (16)	0.86674 (12)	0.79091 (13)	0.0412 (5)
S3	0.93489 (4)	0.69294 (3)	0.78243 (3)	0.02904 (11)
C119	0.89197 (18)	0.61118 (13)	0.79230 (12)	0.0322 (5)
N11	0.8653 (2)	0.55532 (13)	0.80169 (13)	0.0496 (6)
C120	0.9168 (4)	0.8774 (4)	0.2474 (3)	0.136 (3)
H12A	0.8818	0.8463	0.2708	0.204*
H12B	0.9833	0.8758	0.2591	0.204*
H12C	0.8922	0.8603	0.1958	0.204*
C121	0.9060 (3)	0.9548 (4)	0.2729 (2)	0.099 (2)
H12D	0.9371	0.9747	0.3239	0.119*
H12E	0.9350	0.9858	0.2458	0.119*
01	0.81017 (17)	0.95296 (14)	0.26223 (11)	0.0551 (5)
C122	0.7881 (4)	1.0237 (2)	0.2809 (2)	0.0799 (14)
H12F	0.8157	1.0552	0.2536	0.096*
H12G	0.8141	1.0469	0.3322	0.096*
C123	0.6848 (4)	1.0135 (2)	0.2638 (3)	0.0803 (14)
H12H	0.6600	0.9920	0.2127	0.120*
H12I	0.6678	1.0606	0.2773	0.120*
H12J	0.6581	0.9812	0.2901	0.120*

Atomic displacement parameters (\mathring{A}^2)

	U^{11}	U^{22}	U ³³	U^{12}	U^{13}	U^{23}
Ag1	0.02005 (8)	0.03206 (9)	0.02983 (8)	-0.00063 (6)	0.00112 (6)	0.00219 (6)
C1	0.0323 (12)	0.0319 (12)	0.0317 (11)	0.0030 (10)	0.0026 (9)	0.0088 (9)
C2	0.0401 (15)	0.0414 (14)	0.0482 (15)	0.0082 (12)	0.0044 (12)	0.0220 (12)
C3	0.0563 (19)	0.0295 (13)	0.069 (2)	0.0084 (12)	0.0133 (15)	0.0201 (13)
C4	0.062 (2)	0.0268 (13)	0.0566 (17)	0.0042 (13)	0.0075 (15)	0.0003 (12)
C5	0.0387 (14)	0.0317 (12)	0.0320 (12)	0.0053 (10)	0.0048 (10)	-0.0007 (10)
C6	0.0200 (10)	0.0262 (10)	0.0272 (10)	0.0038 (8)	0.0049 (8)	0.0039 (8)
C7	0.0324 (12)	0.0255 (10)	0.0180 (9)	0.0041 (9)	0.0037 (8)	0.0015 (8)
N1	0.0224 (9)	0.0235 (8)	0.0191 (8)	0.0029 (7)	0.0026 (7)	0.0029 (6)
C8	0.0208 (10)	0.0271 (10)	0.0235 (10)	0.0031 (8)	0.0027 (8)	0.0033 (8)

C9	0.0197 (10)	0.0222 (9)	0.0224 (9)	0.0023 (8)	0.0028 (8)	0.0047 (7)
C10	0.0263 (11)	0.0249 (10)	0.0349 (11)	0.0015 (9)	0.0135 (9)	0.0006 (9)
C11	0.0337 (14)	0.0584 (17)	0.0352 (13)	0.0010 (12)	0.0091 (10)	0.0169 (12)
C12	0.0501 (18)	0.068 (2)	0.0377 (14)	-0.0167 (16)	0.0157 (13)	0.0104 (14)
C13	0.0454 (18)	0.060 (2)	0.093 (3)	0.0130 (16)	0.0425 (19)	0.0166 (19)
C14	0.046 (2)	0.085 (3)	0.135 (4)	0.0338 (19)	0.046 (2)	0.063 (3)
C15	0.0392 (16)	0.065 (2)	0.089 (2)	0.0229 (15)	0.0283 (16)	0.0452 (19)
C16	0.0175 (10)	0.0250 (10)	0.0227 (9)	0.0016 (8)	0.0046 (8)	0.0036 (8)
C17	0.0201 (10)	0.0302 (11)	0.0216 (9)	-0.0008(8)	0.0038 (8)	0.0022 (8)
C18	0.0320 (12)	0.0346 (12)	0.0287 (11)	0.0054 (10)	0.0024 (9)	-0.0027(9)
C19	0.0379 (14)	0.0408 (14)	0.0337(13)	-0.0011(11)	0.0046(11)	-0.0093(11)
C20	0.0368(14)	0.0554 (17)	0.0295(12)	-0.0097(13)	-0.0041(10)	-0.0020(11)
C21	0.055(2)	0.061(2)	0.0525(12)	0.0052(16)	-0.0274(15)	0.0109(15)
C22	0.033(2) 0.0471(17)	0.001(2)	0.0523(10) 0.0533(17)	0.0002(10) 0.0046(13)	-0.0198(14)	$0.010^{\circ}(12)$
N2	0.0189(9)	0.0286 (9)	0.0231(8)	0.0021(7)	0.0052 (7)	0.0019(7)
C23	0.0105(5)	0.0200(5)	0.0291(0) 0.0298(11)	0.0021(9)	0.0002(7)	-0.0021(9)
C24	0.0227(11) 0.0317(12)	0.0361(13)	0.0290(11) 0.0306(11)	-0.0012(3)	0.0163(10)	0.0021(9)
C25	0.0317(12) 0.0488(16)	0.0301(13) 0.0429(14)	0.0308(11)	0.0032(10) 0.0014(12)	0.0169(11)	0.0015(11)
C26	0.073(2)	0.0429(14) 0.0573(19)	0.0308(12) 0.0298(13)	0.0014(12)	0.0109(11) 0.0200(14)	0.0013(11) 0.0077(12)
C27	0.073(2) 0.102(3)	0.0575(17)	0.0290(15) 0.0434(16)	0.0010(10)	0.0200(11) 0.0323(18)	0.0077(12)
C28	0.102(3)	0.000(2) 0.0379(16)	0.0434(10) 0.0513(17)	-0.007(2)	0.0325(10) 0.0335(19)	0.0207(13)
C20	0.102(3)	0.0379(10) 0.0358(14)	0.0313(17) 0.0361(13)	-0.0103(13)	0.0333(13)	0.0105(15)
C30	0.0029(19) 0.0438(15)	0.0350(14) 0.0458(15)	0.0301(13) 0.0346(13)	0.0105(12)	0.0210(13) 0.0040(11)	0.0040(11) 0.0024(11)
C31	0.0430(15) 0.082(3)	0.0450(15) 0.070(2)	0.0340(15) 0.0405(16)	0.0130(12)	-0.0040(11)	-0.0142(16)
C32	0.002(3)	0.070(2)	0.070(3)	0.041(2) 0.036(2)	-0.050(3)	-0.0195(18)
C32	0.103(3)	0.0420(1)) 0.0232(14)	0.070(3)	-0.0062(15)	-0.058(3)	0.0122(17)
C34	0.077(3)	0.0232(14) 0.0340(14)	0.105(3)	-0.0002(13)	-0.0200(14)	0.0122(17) 0.0238(13)
C35	0.0423(10) 0.0315(12)	0.0340(14) 0.0228(10)	0.0034(1))	0.0093(12)	-0.0200(14)	0.0236(13)
C36	0.0313(12) 0.0367(13)	0.0220(10) 0.0270(11)	0.0330(11) 0.0233(10)	0.0024(9)	0.0048(9)	0.0070(9)
N3	0.0307(13)	0.0270(11) 0.0221(8)	0.0233(10)	-0.0010(9)	0.0017(3)	0.0110 (8)
C37	0.0248(0)	0.0221(0)	0.0200(8)	0.0005 (7)	0.0020(7)	0.0070(0)
C38	0.0213(10)	0.0233(11) 0.0224(10)	0.0212(9)	0.0000(8)	0.0020(3)	0.0024(3)
C30	0.0223(10)	0.0224(10)	0.0193(9)	0.0021(8)	0.0032(7)	0.0047(7)
C39	0.0230(10)	0.0213(10)	0.0241(10)	0.0031(8)	0.0038(8)	0.0055(8)
C40	0.0284(11) 0.0316(13)	0.0302(12)	0.0222(10)	0.0102(9)	0.0030(8)	0.0007(9)
C41 C42	0.0310(13)	0.0481(13)	0.0283(11) 0.0450(14)	0.0137(11) 0.0175(11)	0.0014(9)	0.0100(10)
C42	0.0241(12) 0.0220(14)	0.0303(10)	0.0430(14)	0.0173(11) 0.0200(12)	0.0034(10)	0.0175(12)
C45	0.0320(14)	0.071(2)	0.0443(14)	0.0209(13)	0.0101(11)	0.0233(14)
C44 C45	0.0316(13)	0.0463(14)	0.0300(11)	0.0140(11) 0.0052(8)	0.0103(9)	0.0172(10)
C45	0.0210(10)	0.0202(10)	0.0203(9)	0.0033(8)	0.0037(8)	0.0030(8)
C40	0.0237(10)	0.0232(10)	0.0204(10)	0.0070(8)	0.0025(8)	0.0087(8)
C4/A	0.051(6)	0.030(3)	0.032(3)	0.004(3)	0.008(3)	0.0094(19)
C48A	0.069(7)	0.023(3)	0.039(3)	0.003(3)	-0.004(3)	0.007(2)
C4/B	0.037(5)	0.026(3)	0.025(3)	0.002(3)	0.003(3)	0.008(2)
C48B	0.043(3)	0.025(3)	0.052(3)	0.003(3)	0.001(3)	0.002(2)
C49	0.0378(14)	0.0285(12)	0.0535(16)	0.0001(10)	0.0000(12)	0.0105(11)
CSUA	0.049 (6)	0.037(3)	0.005(5)	0.010(4)	0.028 (5)	0.027(3)
COLA	0.043 (5)	0.029 (3)	0.043 (4)	0.006 (3)	0.012 (4)	0.009 (3)
COOR	0.044 (6)	0.037(4)	0.035 (4)	0.000 (4)	0.012 (4)	0.013 (3)

C51B	0.035 (5)	0.037 (4)	0.018 (3)	-0.007 (4)	0.003 (3)	0.008 (2)
N4	0.0211 (9)	0.0272 (9)	0.0237 (8)	0.0052 (7)	0.0012 (7)	0.0023 (7)
C52	0.0224 (11)	0.0345 (12)	0.0315 (11)	0.0105 (9)	-0.0001 (9)	0.0020 (9)
C53	0.0207 (10)	0.0269 (11)	0.0326 (11)	0.0009 (8)	0.0019 (8)	0.0099 (9)
C54	0.0360 (14)	0.0294 (12)	0.0478 (15)	0.0095 (10)	-0.0071 (11)	0.0045 (11)
C55	0.0502 (17)	0.0448 (16)	0.0539 (17)	0.0109 (13)	-0.0110 (14)	0.0216 (13)
C56	0.0475 (17)	0.090 (3)	0.0409 (15)	0.0136 (17)	0.0064 (13)	0.0389 (16)
C57	0.0427 (16)	0.099 (3)	0.0359 (14)	0.0271 (17)	0.0180 (12)	0.0252 (16)
C58	0.0291 (12)	0.0615 (17)	0.0331 (12)	0.0190 (12)	0.0101 (10)	0.0188 (12)
Ag2	0.01516(7)	0.02111 (7)	0.02099 (7)	0.00652 (5)	0.00277 (5)	0.00464 (5)
C59	0.0174 (9)	0.0244 (10)	0.0258 (10)	0.0048 (8)	0.0069 (8)	0.0048 (8)
C60	0.0256 (11)	0.0266 (10)	0.0239 (10)	0.0059 (9)	0.0068 (8)	0.0015 (8)
C61	0.0241 (11)	0.0218 (10)	0.0271 (10)	0.0019 (8)	0.0019 (8)	0.0000 (8)
C62	0.0207 (10)	0.0269 (11)	0.0313 (11)	-0.0019 (8)	0.0057 (8)	0.0060 (9)
C63	0.0214 (10)	0.0267 (10)	0.0257 (10)	0.0010 (8)	0.0074 (8)	0.0066 (8)
C64	0.0171 (9)	0.0171 (9)	0.0247 (9)	0.0046 (7)	0.0045 (7)	0.0045 (7)
C65	0.0163 (9)	0.0213 (9)	0.0226 (9)	0.0034 (7)	0.0059 (7)	0.0063 (7)
N5	0.0148 (8)	0.0178 (8)	0.0195 (7)	0.0044 (6)	0.0031 (6)	0.0037 (6)
C66	0.0170 (9)	0.0207 (9)	0.0175 (8)	0.0058 (7)	0.0044 (7)	0.0040 (7)
C67	0.0160 (9)	0.0209 (9)	0.0172 (8)	0.0056 (7)	0.0047 (7)	0.0029 (7)
C68	0.0165 (9)	0.0198 (9)	0.0235 (9)	0.0063 (7)	0.0077 (7)	0.0072 (7)
C69	0.0182 (9)	0.0215 (9)	0.0261 (9)	0.0042 (8)	0.0064 (8)	0.0058 (8)
C70	0.0245 (11)	0.0193 (9)	0.0341 (11)	0.0055 (8)	0.0120 (9)	0.0048 (8)
C71	0.0297 (12)	0.0250 (10)	0.0395 (12)	0.0124 (9)	0.0147 (9)	0.0150 (9)
C72	0.0275 (11)	0.0328 (11)	0.0299 (11)	0.0122 (9)	0.0047 (9)	0.0138 (9)
C73	0.0236 (10)	0.0254 (10)	0.0230 (9)	0.0073 (8)	0.0044 (8)	0.0064 (8)
C74	0.0158 (9)	0.0202 (9)	0.0173 (8)	0.0067 (7)	0.0044 (7)	0.0035 (7)
C75	0.0163 (9)	0.0181 (9)	0.0222 (9)	0.0041 (7)	0.0033 (7)	0.0056 (7)
C76	0.0206 (10)	0.0190 (9)	0.0282 (10)	0.0059 (8)	0.0056 (8)	0.0035 (8)
C77	0.0198 (10)	0.0243 (10)	0.0373 (11)	0.0078 (8)	0.0109 (9)	0.0077 (9)
C78	0.0171 (10)	0.0258 (10)	0.0388 (12)	0.0025 (8)	0.0022 (8)	0.0103 (9)
C79	0.0243 (11)	0.0234 (10)	0.0258 (10)	0.0016 (8)	-0.0006 (8)	0.0021 (8)
C80	0.0204 (10)	0.0254 (10)	0.0223 (9)	0.0046 (8)	0.0055 (8)	0.0038 (8)
N6	0.0156 (8)	0.0181 (8)	0.0192 (7)	0.0057 (6)	0.0043 (6)	0.0045 (6)
C81	0.0209 (10)	0.0171 (9)	0.0227 (9)	0.0055 (7)	0.0052 (7)	0.0034 (7)
C82	0.0230 (10)	0.0157 (9)	0.0247 (9)	0.0051 (7)	0.0066 (8)	0.0054 (7)
C83	0.0250 (11)	0.0246 (10)	0.0315 (11)	0.0055 (8)	0.0071 (9)	0.0097 (8)
C84	0.0338 (13)	0.0337 (12)	0.0413 (13)	0.0085 (10)	0.0193 (10)	0.0160 (10)
C85	0.0512 (16)	0.0351 (13)	0.0322 (12)	0.0061 (11)	0.0168 (11)	0.0153 (10)
C86	0.0406 (14)	0.0409 (14)	0.0306 (12)	0.0025 (11)	0.0019 (10)	0.0170 (10)
C87	0.0255 (11)	0.0316 (11)	0.0327 (11)	0.0052 (9)	0.0051 (9)	0.0146 (9)
C88	0.0298 (12)	0.0352 (12)	0.0263 (10)	0.0080 (10)	0.0099 (9)	0.0065 (9)
C89	0.0397 (14)	0.0568 (17)	0.0314 (12)	0.0135 (12)	0.0169 (11)	0.0152 (11)
C90	0.067 (2)	0.0628 (19)	0.0240 (12)	0.0252 (16)	0.0191 (12)	0.0097 (12)
C91	0.086 (2)	0.0454 (16)	0.0249 (12)	0.0105 (16)	0.0151 (14)	-0.0025 (11)
C92	0.0588 (18)	0.0314 (13)	0.0261 (11)	0.0009 (12)	0.0077 (11)	0.0007 (9)
C93	0.0260 (11)	0.0257 (10)	0.0216 (9)	0.0085 (8)	0.0046 (8)	0.0041 (8)
C94	0.0208 (10)	0.0191 (9)	0.0238 (9)	0.0052 (7)	0.0052 (8)	0.0042 (7)

N7	0.0163 (8)	0.0189 (8)	0.0197(7)	0 0052 (6)	0 0034 (6)	0.0050 (6)
C95	0.0173(9)	0.0207(9)	0.0193 (9)	0.0052(0)	0.0058(7)	0.0058(7)
C96	0.0175 (9)	0.0218 (9)	0.0230 (9)	0.0069 (8)	0.0045 (7)	0.0079 (7)
C97	0.0175 (10)	0.0204 (9)	0.0300 (10)	0.0041 (8)	-0.0005(8)	0.0096 (8)
C98	0.0248 (11)	0.0384 (13)	0.0283 (11)	0.0035 (10)	-0.0021(9)	0.0082 (9)
C99	0.0340 (14)	0.0413 (14)	0.0355 (13)	-0.0008(11)	-0.0102(10)	0.0085 (11)
C100	0.0207 (11)	0.0328 (12)	0.0566 (16)	-0.0027(9)	-0.0117 (10)	0.0182 (11)
C101	0.0169 (11)	0.0299 (12)	0.0615 (16)	0.0046 (9)	0.0062 (10)	0.0171 (11)
C102	0.0217 (11)	0.0245 (10)	0.0410 (12)	0.0052 (8)	0.0044 (9)	0.0096 (9)
C103	0.0168 (9)	0.0221 (9)	0.0206 (9)	0.0063 (7)	0.0023 (7)	0.0061 (7)
C104	0.0170 (9)	0.0224 (10)	0.0283 (10)	0.0062 (8)	0.0018 (8)	0.0051 (8)
C105	0.0250 (11)	0.0268 (11)	0.0278 (10)	0.0094 (9)	0.0019 (8)	0.0017 (8)
C106	0.0303 (12)	0.0353 (13)	0.0402 (13)	0.0149 (10)	0.0071 (10)	-0.0024 (10)
C107	0.0358 (14)	0.0348 (13)	0.0654 (18)	0.0231 (11)	0.0098 (13)	0.0130 (12)
C108	0.0467 (16)	0.0485 (16)	0.0673 (19)	0.0299 (14)	0.0192 (14)	0.0349 (15)
C109	0.0350 (13)	0.0412 (14)	0.0444 (13)	0.0202 (11)	0.0155 (11)	0.0228 (11)
N8	0.0167 (8)	0.0202 (8)	0.0192 (7)	0.0066 (6)	0.0024 (6)	0.0042 (6)
C110	0.0193 (10)	0.0212 (9)	0.0206 (9)	0.0044 (8)	0.0027 (7)	0.0021 (7)
C111	0.0193 (10)	0.0192 (9)	0.0271 (10)	0.0083 (8)	0.0054 (8)	0.0056 (7)
C112	0.0268 (11)	0.0262 (11)	0.0315 (11)	0.0031 (9)	0.0046 (9)	0.0039 (9)
C113	0.0318 (13)	0.0320 (12)	0.0484 (14)	-0.0032 (10)	0.0084 (11)	0.0066 (11)
C114	0.0359 (14)	0.0339 (13)	0.0558 (16)	0.0050 (11)	0.0217 (12)	0.0192 (11)
C115	0.0374 (13)	0.0367 (12)	0.0362 (12)	0.0163 (10)	0.0175 (10)	0.0170 (10)
C116	0.0251 (11)	0.0260 (10)	0.0286 (10)	0.0098 (8)	0.0073 (8)	0.0077 (8)
Ag3	0.02101 (8)	0.03363 (9)	0.02613 (8)	0.00446 (6)	0.00344 (6)	0.00431 (6)
S1	0.0207 (2)	0.0273 (2)	0.0305 (2)	0.00565 (19)	0.00707 (19)	0.0052 (2)
C117	0.0205 (10)	0.0260 (10)	0.0217 (9)	-0.0006 (8)	0.0063 (8)	0.0015 (8)
N9	0.0246 (9)	0.0307 (10)	0.0347 (10)	0.0072 (8)	0.0097 (8)	0.0069 (8)
S2	0.0326 (3)	0.0309 (3)	0.0323 (3)	0.0114 (2)	0.0085 (2)	0.0083 (2)
C118	0.0350 (13)	0.0242 (10)	0.0266 (10)	0.0047 (9)	0.0089 (9)	0.0006 (8)
N10	0.0338 (12)	0.0299 (11)	0.0511 (13)	0.0025 (9)	0.0049 (10)	-0.0003 (9)
S3	0.0251 (3)	0.0338 (3)	0.0254 (2)	0.0032 (2)	0.0011 (2)	0.0084 (2)
C119	0.0343 (13)	0.0339 (12)	0.0232 (10)	0.0057 (10)	0.0007 (9)	0.0032 (9)
N11	0.0614 (16)	0.0344 (12)	0.0413 (12)	-0.0008 (11)	-0.0036 (11)	0.0076 (10)
C120	0.090 (4)	0.268 (9)	0.169 (6)	0.122 (5)	0.092 (4)	0.182 (6)
C121	0.047 (2)	0.195 (6)	0.073 (3)	0.014 (3)	0.0118 (19)	0.083 (4)
01	0.0558 (13)	0.0739 (15)	0.0417 (11)	0.0120 (11)	0.0185 (10)	0.0230 (10)
C122	0.132 (4)	0.049 (2)	0.057 (2)	-0.007 (2)	0.046 (2)	0.0105 (16)
C123	0.120 (4)	0.056 (2)	0.102 (3)	0.041 (2)	0.077 (3)	0.036 (2)

Geometric parameters (Å, °)

Ag1—C37	2.085 (2)	С60—Н60	0.9500	
Ag1—C8	2.091 (2)	C61—C62	1.387 (3)	
C1—C2	1.383 (4)	C61—H61	0.9500	
C1—C6	1.390 (3)	C62—C63	1.388 (3)	
C1—H1	0.9500	С62—Н62	0.9500	
C2—C3	1.382 (4)	C63—C64	1.397 (3)	

С2—Н2	0.9500	С63—Н63	0.9500
C3—C4	1.381 (5)	C64—C65	1.514 (3)
С3—Н3	0.9500	C65—N5	1.465 (2)
C4—C5	1.382 (4)	С65—Н65А	0.9900
C4—H4	0.9500	С65—Н65В	0.9900
C5—C6	1.387 (3)	N5—C66	1.355 (3)
С5—Н5	0.9500	N5—C67	1.394 (3)
C6—C7	1.515 (3)	C66—N6	1.350 (3)
C7—N1	1.463 (3)	C67—C74	1.358 (3)
C7—H7A	0.9900	С67—С68	1.472 (3)
С7—Н7В	0.9900	C68—C73	1.396 (3)
N1—C8	1.348 (3)	C68—C69	1.403 (3)
N1—C9	1.393 (3)	С69—С70	1.391 (3)
C8—N2	1.350 (3)	С69—Н69	0.9500
C9—C16	1.360 (3)	C70—C71	1.387 (3)
C9—C10	1.476 (3)	С70—Н70	0.9500
C10—C11	1.371 (4)	C71—C72	1.390 (3)
C10—C15	1.414 (4)	C71—H71	0.9500
C11—C12	1.423 (4)	C72—C73	1.390 (3)
C11—H11	0.9500	С72—Н72	0.9500
C12—C13	1.353 (5)	С73—Н73	0.9500
С12—Н12	0.9500	C74—N6	1.398 (2)
C13—C14	1.389 (6)	C74—C75	1.479 (3)
С13—Н13	0.9500	C75—C76	1.392 (3)
C14—C15	1.378 (5)	C75—C80	1.401 (3)
C14—H14	0.9500	C76—C77	1.390 (3)
С15—Н15	0.9500	С76—Н76	0.9500
C16—N2	1.397 (3)	С77—С78	1.385 (3)
C16—C17	1.482 (3)	С77—Н77	0.9500
C17—C22	1.382 (4)	C78—C79	1.388 (3)
C17—C18	1.383 (3)	С78—Н78	0.9500
C18—C19	1.392 (3)	С79—С80	1.390 (3)
C18—H18	0.9500	С79—Н79	0.9500
C19—C20	1.352 (4)	С80—Н80	0.9500
С19—Н19	0.9500	N6—C81	1.474 (2)
C20—C21	1.378 (5)	C81—C82	1.511 (3)
C20—H20	0.9500	C81—H81A	0.9900
C21—C22	1.394 (4)	C81—H81B	0.9900
C21—H21	0.9500	C82—C83	1.391 (3)
С22—Н22	0.9500	C82—C87	1.394 (3)
N2—C23	1.467 (3)	C83—C84	1.394 (3)
C23—C24	1.508 (4)	С83—Н83	0.9500
С23—Н23А	0.9900	C84—C85	1.385 (4)
C23—H23B	0.9900	C84—H84	0.9500
C24—C25	1.382 (3)	C85—C86	1.385 (4)
C24—C29	1.389 (4)	С85—Н85	0.9500
C25—C26	1.384 (5)	C86—C87	1.384 (3)
C25—H25	0.9500	C86—H86	0.9500

C26—C27	1.371 (5)	С87—Н87	0.9500
С26—Н26	0.9500	C88—C93	1.378 (3)
C27—C28	1.387 (5)	C88—C89	1.396 (3)
С27—Н27	0.9500	C88—H88	0.9500
C28—C29	1.383 (5)	C89—C90	1.379 (4)
C28—H28	0.9500	С89—Н89	0.9500
С29—Н29	0.9500	C90—C91	1.379 (5)
C30—C35	1.376 (4)	С90—Н90	0.9500
C30—C31	1.380 (4)	C91—C92	1.387 (4)
C30—H30	0.9500	C91—H91	0.9500
$C_{31} - C_{32}$	1 341 (7)	C92-C93	1 393 (3)
C31—H31	0.9500	C92_H92	0.9500
C_{32}	1 360 (8)	C92 - C94	1.519(3)
$C_{32} = C_{33}$	0.9500	C94 N7	1.317(3) 1.463(3)
$C_{32} = C_{32}$	1 400 (5)	C_{94} H04 A	0.0000
C_{22} H_{22}	0.0500	C_{94} H04P	0.9900
C34 C25	0.9300	C94—194D	0.9900
C34—C35	1.385 (4)	N/	1.352 (3)
C34—H34	0.9500	N/	1.399 (3)
C35—C36	1.503 (3)	C95—N8	1.351 (3)
C36—N3	1.466 (3)	C96—C103	1.361 (3)
С36—Н36А	0.9900	С96—С97	1.477 (3)
С36—Н36В	0.9900	C97—C98	1.392 (3)
N3—C37	1.347 (3)	C97—C102	1.392 (3)
N3—C38	1.398 (3)	C98—C99	1.397 (3)
C37—N4	1.346 (3)	С98—Н98	0.9500
C38—C45	1.361 (3)	C99—C100	1.379 (4)
C38—C39	1.472 (3)	С99—Н99	0.9500
C39—C44	1.392 (3)	C100-C101	1.375 (4)
C39—C40	1.396 (3)	C100—H100	0.9500
C40—C41	1.385 (3)	C101—C102	1.394 (3)
C40—H40	0.9500	C101—H101	0.9500
C41—C42	1.378 (4)	C102—H102	0.9500
C41—H41	0.9500	C103—N8	1.389 (3)
C42—C43	1.384 (4)	C103—C104	1.480 (3)
C42—H42	0.9500	C104—C109	1.391 (3)
C43—C44	1.384 (4)	C104—C105	1.394 (3)
C43—H43	0.9500	C105—C106	1.388 (3)
C44—H44	0.9500	C105—H105	0.9500
C45—N4	1 396 (3)	C106-C107	1 381 (4)
C_{45}	1.376(3)	C106—H106	0.9500
$C_{45} = C_{40}$	1.475(3) 1.348(9)	$C_{100} = 1100$	1.380(4)
$C_{46} = C_{47A}$	1.348(9) 1 300(7)	$C_{107} = C_{108}$	0.9500
$C_{40} = C_{47} R$	1.390(7) 1.200(8)	C_{10}^{-110}	1.382(4)
$C_{40} - C_{4/B}$	1.377 (0)	C108 H108	0.0500
C_{40} C_{40} C_{40} C_{40}	1.413 (7)	C100_H100	0.9300
C47A = U40A	1.392 (9)	$ \begin{array}{c} 109 \\ 109 \\ 10 \\ 10 \\ 10 \\ 10 \\ 10 \\ 10$	0.9300
$C4/A - \Pi4/A$	0.9300		1.457 (5)
C48A—C49	1.484 (12)		1.510(3)
C48A—H48A	0.9500	C110—H11A	0.9900

C47B—C48B	1.382 (10)	C110—H11B	0.9900
C47B—H47B	0.9500	C111—C116	1.394 (3)
C48B—C49	1 297 (8)	C111—C112	1 396 (3)
C48B - H48B	0.9500	C_{112} C_{113}	1.390(3) 1 384(4)
$C49 - C50 \Delta$	1 325 (11)	C112_H112	0.9500
C_{40} C_{50R}	1.325(11) 1.415(11)	$C_{112} = C_{114}$	1.384(4)
$C_{49} = C_{50B}$	0.0500	C113 H113	0.0500
C40 H40	0.9500		1.399(4)
$C_{49} = 1149$	0.9500	$C_{114} = C_{115}$	1.388 (4)
C50A - C51A	1.579 (15)	C114—H114	0.9300
CSUA—HSUA	0.9500		1.393 (3)
CSIA—HSIA	0.9500		0.9500
CSOB-CSIB	1.399 (15)		0.9500
C50B—H50B	0.9500	Ag3—S1	2.4657 (5)
C51B—H51B	0.9500	Ag3—S3	2.4940 (6)
N4—C52	1.463 (3)	Ag3—S2	2.5377 (6)
C52—C53	1.510 (3)	S1—C117	1.671 (2)
C52—H52A	0.9900	C117—N9	1.154 (3)
C52—H52B	0.9900	S2—C118	1.669 (3)
C53—C58	1.384 (4)	C118—N10	1.160 (3)
C53—C54	1.392 (3)	S3—C119	1.674 (3)
C54—C55	1.385 (4)	C119—N11	1.156 (4)
С54—Н54	0.9500	C120—C121	1.517 (9)
C55—C56	1.379 (5)	C120—H12A	0.9800
С55—Н55	0.9500	C120—H12B	0.9800
C56—C57	1.384 (5)	C120—H12C	0.9800
С56—Н56	0.9500	C121—O1	1.394 (5)
С57—С58	1.395 (4)	C121—H12D	0.9900
С57—Н57	0.9500	C121—H12E	0.9900
С58—Н58	0.9500	O1—C122	1.447 (5)
Ag2—C66	2.097 (2)	C122—C123	1.479 (7)
Ag2—C95	2.102 (2)	C122—H12F	0.9900
C59—C64	1390(3)	C122—H12G	0.9900
C59 - C60	1 393 (3)	C123—H12H	0.9900
C59_H59	0.9500	C123 H12H	0.9800
C60-C61	1 386 (3)	C123—H12I	0.9800
200-201	1.560 (5)	0125—11125	0.9800
C37 Ag1 $C8$	173 06 (8)	C60 C61 H61	120.1
C_{2} C_{1} C_{6}	175.00(0)	$C_{60} = C_{61} = H_{61}$	120.1
$C_2 = C_1 = C_0$	120.0 (2)	C_{02} C_{01} C_{01} C_{01} C_{02} C	120.1
	119.7	$C_{01} = C_{02} = C_{03}$	120.0 (2)
	119.7	C61—C62—H62	120.0
$C_3 = C_2 = C_1$	120.0 (3)	C63—C62—H62	120.0
$C_3 = C_2 = H_2$	120.0	C62—C63—C64	120.4 (2)
C1—C2—H2	120.0	С62—С63—Н63	119.8
C4—C3—C2	119.7 (3)	С64—С63—Н63	119.8
С4—С3—Н3	120.1	C59—C64—C63	119.32 (19)
C2—C3—H3	120.1	C59—C64—C65	123.21 (18)
C3—C4—C5	120.2 (3)	C63—C64—C65	117.42 (18)
C3—C4—H4	119.9	N5—C65—C64	114.90 (16)

C5—C4—H4	119.9	N5—C65—H65A	108.5
C4—C5—C6	120.7 (3)	С64—С65—Н65А	108.5
C4—C5—H5	119.7	N5—C65—H65B	108.5
С6—С5—Н5	119.7	С64—С65—Н65В	108.5
C5—C6—C1	118.7 (2)	H65A—C65—H65B	107.5
C5—C6—C7	118.5 (2)	C66—N5—C67	111.26 (16)
C1—C6—C7	122.8 (2)	C66—N5—C65	121.43 (16)
N1—C7—C6	111.19 (17)	C67—N5—C65	125.21 (16)
N1—C7—H7A	109.4	N6—C66—N5	104.81 (16)
C6—C7—H7A	109.4	N6—C66—Ag2	128.78 (14)
N1—C7—H7B	109.4	N5-C66-Ag2	125 76 (14)
C6-C7-H7B	109.4	C74-C67-N5	106.39(17)
H7A - C7 - H7B	108.0	C74-C67-C68	130.61(18)
C8 - N1 - C9	111 66 (17)	N_{5} C_{67} C_{68}	122.99(17)
C8-N1-C7	121 59 (18)	C73 - C68 - C69	122.99(17) 118.72(19)
C9-N1-C7	121.57 (18)	C73 - C68 - C67	110.72(19) 120.38(18)
C_{2} N1 C8 N2	124.37(18) 104.60(18)	$C_{13} = C_{03} = C_{07}$	120.38(18) 120.00(18)
$N1 = C_0 = N_2$	104.09(18) 125.14(15)	C70 C60 C68	120.90(18)
$N1 - C_0 - Ag_1$	123.14(13) 120.44(16)	C70 - C69 - U68	120.41 (19)
$N_2 = C_0 = Ag_1$	129.44 (10)	С/0—С09—Н09	119.8
C16 - C9 - N1	100.11(18) 121.0((10)	C08—C09—H09	119.8
C10 - C9 - C10	131.06 (19)	C/1 = C/0 = C69	120.2 (2)
NI = C9 = C10	122.82 (19)	C/1 - C/0 - H/0	119.9
	118.6 (2)	C69—C70—H70	119.9
	122.0 (2)	C/0_C/1_C/2	119.9 (2)
C15—C10—C9	119.4 (2)	С70—С71—Н71	120.0
C10—C11—C12	120.4 (3)	С72—С71—Н71	120.0
C10—C11—H11	119.8	C71—C72—C73	120.1 (2)
C12—C11—H11	119.8	С71—С72—Н72	120.0
C13—C12—C11	119.8 (3)	С73—С72—Н72	120.0
C13—C12—H12	120.1	C72—C73—C68	120.7 (2)
C11—C12—H12	120.1	С72—С73—Н73	119.7
C12—C13—C14	120.6 (3)	С68—С73—Н73	119.7
C12—C13—H13	119.7	C67—C74—N6	106.25 (17)
C14—C13—H13	119.7	C67—C74—C75	131.52 (18)
C15—C14—C13	120.1 (4)	N6—C74—C75	122.10 (17)
C15—C14—H14	119.9	C76—C75—C80	119.53 (19)
C13—C14—H14	119.9	C76—C75—C74	121.65 (18)
C14—C15—C10	120.3 (3)	C80—C75—C74	118.82 (18)
C14—C15—H15	119.8	C77—C76—C75	120.21 (19)
C10—C15—H15	119.8	С77—С76—Н76	119.9
C9—C16—N2	106.17 (18)	С75—С76—Н76	119.9
C9—C16—C17	130.6 (2)	C78—C77—C76	120.3 (2)
N2—C16—C17	123.25 (18)	С78—С77—Н77	119.9
C22—C17—C18	118.4 (2)	С76—С77—Н77	119.9
C22—C17—C16	120.0 (2)	C77—C78—C79	119.8 (2)
C18—C17—C16	121.6 (2)	С77—С78—Н78	120.1
C17—C18—C19	120.7 (3)	С79—С78—Н78	120.1
C17—C18—H18	119.6	C78—C79—C80	120.5 (2)

C19—C18—H18	119.6	С78—С79—Н79	119.8
C20-C19-C18	120.5 (3)	С80—С79—Н79	119.8
С20—С19—Н19	119.7	C79—C80—C75	119.7 (2)
С18—С19—Н19	119.7	С79—С80—Н80	120.2
C19-C20-C21	119.8 (2)	C75—C80—H80	120.2
C_{19} C_{20} H_{20}	120.1	$C_{66} = N_{6} = C_{74}$	111 27 (16)
C_{21} C_{20} H_{20}	120.1	C66 N6 $C81$	122.02(17)
$C_{21} = C_{20} = H_{20}$	120.1 120.2(2)	C74 N6 $C81$	122.92(17) 125.72(17)
$C_{20} = C_{21} = C_{22}$	120.2 (3)	C/4— $N0$ — $C01$	123.75(17)
C20—C21—H21	119.9	$N_0 = C_{81} = C_{82}$	112.39 (16)
C22—C21—H21	119.9	N6—C81—H81A	109.1
C17—C22—C21	120.3 (3)	C82—C81—H81A	109.1
C17—C22—H22	119.8	N6—C81—H81B	109.1
C21—C22—H22	119.8	C82—C81—H81B	109.1
C8—N2—C16	111.32 (18)	H81A—C81—H81B	107.8
C8—N2—C23	123.91 (19)	C83—C82—C87	118.8 (2)
C16—N2—C23	124.17 (18)	C83—C82—C81	121.23 (19)
N2-C23-C24	110.6 (2)	C87—C82—C81	119.93 (19)
N2—C23—H23A	109.5	C82—C83—C84	120.3 (2)
C24—C23—H23A	109.5	С82—С83—Н83	119.8
N2—C23—H23B	109.5	C84—C83—H83	119.8
C24—C23—H23B	109.5	C85—C84—C83	120.1 (2)
H23A—C23—H23B	108.1	C85—C84—H84	119.9
C_{25} C_{24} C_{29}	1194(3)	C83—C84—H84	119.9
$C_{25} = C_{24} = C_{23}$	1201(2)	$C_{86} C_{85} C_{84}$	119.5 119.7(2)
$C_{23} C_{24} C_{23} C_{24} C_{23}$	120.1(2) 120.5(2)	C_{86}^{86} C_{85}^{85} H85	120.1
$C_{2}^{2} = C_{2}^{2} + C_{2}^{2}$	120.3(2) 120.3(3)	$C_{80} = C_{85} = 1185$	120.1
$C_{24} = C_{25} = C_{20}$	120.3 (3)	$C_{04} = C_{05} = 1185$	120.1
$C_{24} = C_{25} = H_{25}$	119.9	$C_{87} - C_{80} - C_{83}$	120.2(2)
C26-C25-H25	119.9	C8/-C80-H80	119.9
$C_2/-C_{26}-C_{25}$	120.4 (3)	C85—C86—H86	119.9
C2/C26H26	119.8		120.7 (2)
C25—C26—H26	119.8	С86—С87—Н87	119.7
C26—C27—C28	119.7 (3)	С82—С87—Н87	119.7
С26—С27—Н27	120.1	C93—C88—C89	120.3 (2)
C28—C27—H27	120.1	C93—C88—H88	119.8
C29—C28—C27	120.1 (3)	C89—C88—H88	119.8
C29—C28—H28	119.9	C90—C89—C88	120.1 (3)
С27—С28—Н28	119.9	С90—С89—Н89	120.0
C28—C29—C24	120.0 (3)	С88—С89—Н89	120.0
С28—С29—Н29	120.0	C91—C90—C89	120.0 (2)
С24—С29—Н29	120.0	С91—С90—Н90	120.0
C35—C30—C31	120.3 (3)	С89—С90—Н90	120.0
C35—C30—H30	119.8	C90—C91—C92	120.0 (3)
C_{31} $-C_{30}$ $-H_{30}$	119.8	C90—C91—H91	120.0
C_{32} C_{31} C_{30}	121.2 (4)	C92—C91—H91	120.0
C_{32} C_{31} H_{31}	119.4	C91-C92-C93	120.0 120.4(3)
C_{30} C_{31} H_{31}	119.4	C91_C92_H92	119.8
C_{31} C_{32} C_{32}	110.7 (3)	$C_{02} = C_{02} = C$	110.8
$C_{21} = C_{22} = C_{23}$	120.1	$C_{22} - C_{22} - C_{1122}$	117.0
UJ1-UJ2	120.1	000-073-092	117.2 (2)

С33—С32—Н32	120.1	C88—C93—C94	121.89 (19)
C32—C33—C34	121.0 (4)	С92—С93—С94	118.8 (2)
С32—С33—Н33	119.5	N7—C94—C93	113.90 (17)
С34—С33—Н33	119.5	N7—C94—H94A	108.8
C35—C34—C33	118.4 (4)	С93—С94—Н94А	108.8
C35—C34—H34	120.8	N7—C94—H94B	108.8
C33—C34—H34	120.8	C93—C94—H94B	108.8
C30—C35—C34	119.3 (3)	H94A—C94—H94B	107.7
C_{30} C_{35} C_{36}	119.7 (2)	C95 - N7 - C96	111.11 (16)
C_{34} — C_{35} — C_{36}	121.0 (3)	C95—N7—C94	122.37 (17)
N3-C36-C35	11153(17)	C96—N7—C94	126.49(17)
N3—C36—H36A	109.3	N8—C95—N7	104.77(17)
C35—C36—H36A	109.3	$N8-C95-A\sigma^2$	125 79 (14)
N3-C36-H36B	109.3	N7 - C95 - Ag2	128.50(14)
C35—C36—H36B	109.3	C103 - C96 - N7	126.33(11) 106.17(17)
H36A—C36—H36B	108.0	C103 - C96 - C97	128.07(19)
C_{37} N3 C_{38}	111 70 (18)	N7	125.00(18)
$C_{37} N_{3} C_{36}$	121 70 (19)	C98 - C97 - C102	123.00(10) 118.8(2)
C_{38} N3 $-C_{36}$	126 54 (19)	C98 - C97 - C96	123.2(2)
N4—C37—N3	10474(18)	C102 - C97 - C96	123.2(2) 118.02(19)
N4-C37-Ag1	128 26 (16)	C97 - C98 - C99	1201(2)
$N_3 - C_37 - A_g I$	126.73 (16)	C97—C98—H98	119.9
C45-C38-N3	105.77 (18)	С99—С98—Н98	119.9
C45-C38-C39	129 59 (19)	C100-C99-C98	120.2(2)
N3-C38-C39	124 63 (19)	С100—С99—Н99	119.9
C44-C39-C40	118.5 (2)	С98—С99—Н99	119.9
C44—C39—C38	120.78 (19)	C101—C100—C99	120.2(2)
C40—C39—C38	120.7 (2)	C101—C100—H100	119.9
C41—C40—C39	120.4 (2)	C99—C100—H100	119.9
C41—C40—H40	119.8	C100—C101—C102	119.9 (2)
С39—С40—Н40	119.8	C100—C101—H101	120.1
C42-C41-C40	120.5 (2)	C102—C101—H101	120.1
C42—C41—H41	119.7	C97—C102—C101	120.7 (2)
C40—C41—H41	119.7	C97—C102—H102	119.6
C41—C42—C43	119.7 (2)	C101—C102—H102	119.6
C41—C42—H42	120.2	C96—C103—N8	106.35 (17)
C43—C42—H42	120.2	C96—C103—C104	129.11 (19)
C44—C43—C42	120.2 (2)	N8—C103—C104	124.51 (18)
C44—C43—H43	119.9	C109—C104—C105	119.6 (2)
C42—C43—H43	119.9	C109—C104—C103	120.2(2)
C43—C44—C39	120.8 (2)	C105-C104-C103	120.07(19)
C43—C44—H44	119.6	C106—C105—C104	119.6 (2)
C39—C44—H44	119.6	C106—C105—H105	120.2
C38—C45—N4	106.32 (18)	C104—C105—H105	120.2
C38—C45—C46	128.4 (2)	C107—C106—C105	120.3 (2)
N4—C45—C46	125.13 (19)	C107—C106—H106	119.9
C51A—C46—C47A	121.0 (5)	C105—C106—H106	119.9
C47B—C46—C51B	116.9 (5)	C108—C107—C106	120.2 (2)
	(-)		(-)

C51A—C46—C45	122.4 (4)	C108—C107—H107	119.9
C47A—C46—C45	116.6 (3)	C106—C107—H107	119.9
C47B—C46—C45	121.5 (3)	C107—C108—C109	120.0 (3)
C51B—C46—C45	121.4 (4)	C107—C108—H108	120.0
C46—C47A—C48A	118.4 (6)	C109—C108—H108	120.0
С46—С47А—Н47А	120.8	C108—C109—C104	120.2 (2)
C48A—C47A—H47A	120.8	C108—C109—H109	119.9
C47A—C48A—C49	119.4 (6)	C104—C109—H109	119.9
C47A—C48A—H48A	120.3	C95—N8—C103	111.58 (16)
C49—C48A—H48A	120.3	C95—N8—C110	122.79 (17)
C48B-C47B-C46	122.0 (6)	C103 - N8 - C110	125.57(17)
C48B-C47B-H47B	119.0	N8-C110-C111	113 24 (16)
C46-C47B-H47B	119.0	N8-C110-H11A	108.9
C49-C48B-C47B	120.5 (6)	C111—C110—H11A	108.9
C49 - C48B - H48B	119.8	N8-C110-H11B	108.9
C47B - C48B - H48B	119.8	C111—C110—H11B	108.9
C48B-C49-C50B	121.9 (6)	H114_C110_H11B	107.7
$C_{40} = C_{40} = C_{40} = C_{40}$	121.9(0) 117.9(5)	C116_C111_C112	107.7 118.8(2)
C48B-C49-H49A	119.0	C116-C111-C110	122.77(19)
C50B C49 H49A	119.0	$C_{112} = C_{111} = C_{110}$	122.77(19) 118.43(19)
$C_{50} = C_{49} = H_{49}$	119.0	C112—C112—C111	120.7(2)
C_{48A} C_{49} H_{49}	121.1	$C_{113} = C_{112} = C_{111}$	110.7 (2)
C_{49} C_{50A} C_{51A}	121.1 121.7(8)	$C_{113} = C_{112} = H_{112}$	119.7
$C_{49} = C_{50A} = C_{51A}$	121.7 (8)	$C_{114} = C_{112} = C_{112}$	119.7 120.4(2)
$C_{49} = C_{50A} = H_{50A}$	119.2	C114 - C113 - C112 C114 - C113 - H113	120.4 (2)
$C_{14} = C_{504} = 1150 \text{ A}$	119.2	$C_{114} - C_{113} - H_{113}$	119.8
$C_{40} = C_{51A} = C_{50A}$	121.0 (6)	$C_{112} = C_{113} = III_{13}$	119.6
C_{40} C_{51A} H_{51A}	119.2	$C_{113} = C_{114} = C_{115}$	119.0 (2)
C_{51} C_{50} C_{50} C_{40}	119.2	C115_C114_H114	120.2
$C_{51D} = C_{50D} = U_{50D}$	110.5 (0)	C113 - C114 - H114	120.2
C10 C50D U50D	120.7	C114 - C115 - C116	120.2 (2)
C49—C50B—H50B	120.7	C114—C115—H115	119.9
C50B—C51B—C46	120.0 (8)	C116—C115—H115	119.9
CSUB-CSIB-HSIB	120.0		120.4 (2)
C46—C51B—H51B	120.0	C115—C116—H116	119.8
$C_3/-N_4-C_{45}$	111.46 (18)		119.8
$C_3/-N_4-C_52$	122.88 (19)	SI - Ag3 - S3	126.936 (19)
C45—N4—C52	125.64 (19)	S1—Ag3—S2	127.68 (2)
N4—C52—C53	114.39 (19)	S3—Ag3—S2	103.957 (19)
N4—C52—H52A	108.7	C117—S1—Ag3	100.90 (7)
C53—C52—H52A	108.7	N9—C117—S1	177.8 (2)
N4—C52—H52B	108.7	C118—S2—Ag3	93.45 (8)
C53—C52—H52B	108.7	N10—C118—S2	179.4 (3)
H52A—C52—H52B	107.6	C119—S3—Ag3	105.56 (9)
C58—C53—C54	118.9 (2)	N11—C119—S3	177.1 (2)
C58—C53—C52	123.4 (2)	C121—C120—H12A	109.5
C54—C53—C52	117.7 (2)	C121—C120—H12B	109.5
C55—C54—C53	120.9 (3)	H12A—C120—H12B	109.5
С55—С54—Н54	119.5	C121—C120—H12C	109.5

С53—С54—Н54	119.5	H12A—C120—H12C	109.5
C56—C55—C54	119.7 (3)	H12B-C120-H12C	109.5
С56—С55—Н55	120.1	O1—C121—C120	107.2 (4)
С54—С55—Н55	120.1	O1—C121—H12D	110.3
C55—C56—C57	120.1 (3)	C120—C121—H12D	110.3
С55—С56—Н56	119.9	O1—C121—H12E	110.3
С57—С56—Н56	119.9	C120—C121—H12E	110.3
C56—C57—C58	120.0 (3)	H12D-C121-H12E	108.5
С56—С57—Н57	120.0	C121—O1—C122	113.9 (4)
С58—С57—Н57	120.0	O1—C122—C123	107.7 (3)
C53—C58—C57	120.3 (3)	O1—C122—H12F	110.2
С53—С58—Н58	119.9	C123—C122—H12F	110.2
С57—С58—Н58	119.9	O1—C122—H12G	110.2
C66—Ag2—C95	172.01 (7)	C123—C122—H12G	110.2
C64—C59—C60	120.02 (19)	H12F—C122—H12G	108.5
С64—С59—Н59	120.0	C122—C123—H12H	109.5
С60—С59—Н59	120.0	C122—C123—H12I	109.5
C61—C60—C59	120.4 (2)	H12H—C123—H12I	109.5
С61—С60—Н60	119.8	C122—C123—H12J	109.5
С59—С60—Н60	119.8	H12H—C123—H12J	109.5
C60—C61—C62	119.8 (2)	H12I—C123—H12J	109.5
C6—C1—C2—C3	1.1 (4)	C46—C45—N4—C52	-2.1(3)
C1—C2—C3—C4	-1.6 (5)	C37—N4—C52—C53	92.3 (3)
C2—C3—C4—C5	1.6 (5)	C45—N4—C52—C53	-89.6 (3)
C3—C4—C5—C6	-1.0(5)	N4—C52—C53—C58	1.2 (3)
C4—C5—C6—C1	0.4 (4)	N4—C52—C53—C54	-179.9 (2)
C4—C5—C6—C7	-179.4 (3)	C58—C53—C54—C55	1.6 (4)
C2—C1—C6—C5	-0.5 (4)	C52—C53—C54—C55	-177.4 (3)
C2—C1—C6—C7	179.4 (2)	C53—C54—C55—C56	-0.7 (5)
C5—C6—C7—N1	162.3 (2)	C54—C55—C56—C57	-1.0(5)
C1—C6—C7—N1	-17.5 (3)	C55—C56—C57—C58	1.7 (5)
C6—C7—N1—C8	-72.3 (3)	C54—C53—C58—C57	-0.8 (4)
C6—C7—N1—C9	89.5 (2)	C52—C53—C58—C57	178.1 (3)
C9—N1—C8—N2	2.0 (2)	C56—C57—C58—C53	-0.8(5)
C7—N1—C8—N2	165.95 (18)	C64—C59—C60—C61	0.2 (3)
C9—N1—C8—Ag1	-169.00(15)	C59—C60—C61—C62	0.5 (3)
C7—N1—C8—Ag1	-5.1 (3)	C60—C61—C62—C63	-0.6(3)
C8—N1—C9—C16	-1.1(2)	C61—C62—C63—C64	-0.1(3)
C7—N1—C9—C16	-164.48(19)	C60—C59—C64—C63	-0.8(3)
C8—N1—C9—C10	-179.9(2)	C60—C59—C64—C65	176.41 (19)
C7—N1—C9—C10	16.8 (3)	C62—C63—C64—C59	0.8 (3)
C16-C9-C10-C11	-111.4(3)	C62—C63—C64—C65	-176.6(2)
N1-C9-C10-C11	67.0 (3)	C59 - C64 - C65 - N5	23.7(3)
C16—C9—C10—C15	66.5 (4)	C63-C64-C65-N5	-158.98(18)
N1-C9-C10-C15	-115.1(3)	C64—C65—N5—C66	74.7 (2)
C15-C10-C11-C12	3.5 (4)	C64—C65—N5—C67	-123.2(2)
C9-C10-C11-C12	-178.5 (2)	C67 - N5 - C66 - N6	-0.3(2)
			··· (-)

C10-C11-C12-C13	-3.1 (5)	C65—N5—C66—N6	164.03 (16)
C11—C12—C13—C14	0.3 (6)	C67—N5—C66—Ag2	171.15 (13)
C12—C13—C14—C15	2.0 (7)	C65—N5—C66—Ag2	-24.5 (2)
C13—C14—C15—C10	-1.5 (7)	C66—N5—C67—C74	0.4 (2)
C11—C10—C15—C14	-1.3 (5)	C65—N5—C67—C74	-163.23(17)
C9—C10—C15—C14	-179.3(4)	C66—N5—C67—C68	-178.55 (17)
N1—C9—C16—N2	-0.3(2)	C65—N5—C67—C68	17.8 (3)
C10-C9-C16-N2	178.3 (2)	C74—C67—C68—C73	43.9 (3)
N1—C9—C16—C17	177.8 (2)	N5-C67-C68-C73	-137.5(2)
C10-C9-C16-C17	-3.7(4)	C74—C67—C68—C69	-135.4(2)
C9-C16-C17-C22	-695(4)	N5-C67-C68-C69	43 3 (3)
N_{2} C16 C17 C22	108.2(3)	C73 - C68 - C69 - C70	10(3)
C9-C16-C17-C18	100.2(3) 113.1(3)	C67 - C68 - C69 - C70	-17974(19)
N_{2} C16 C17 C18	-69.2(3)	C68 - C69 - C70 - C71	-0.8(3)
$C_{22} = C_{12} = C_{13} = C_{10} = C$	1.6(4)	C60 C70 C71 C72	0.0(3)
$C_{22} = C_{17} = C_{18} = C_{19}$	1.0(4) 170 0 (2)	$C_{0}^{-0} = C_{1}^{-0} = C_{1}^{-0} = C_{1}^{-0}$	-0.1(3)
$C_{10} - C_{17} - C_{10} - C_{19}$	179.0(2)	C71 C72 C72 C73	0.1(3)
C17 - C18 - C19 - C20	-0.4(4)	C/1 = C/2 = C/3 = C08	0.3(3)
C18 - C19 - C20 - C21	-1.6(5)	C69 - C68 - C73 - C72	-0.8(3)
C19 - C20 - C21 - C22	2.5 (6)	C6/-C68-C/3-C/2	180.00 (19)
C18—C17—C22—C21	-0.7 (5)	N5—C67—C74—N6	-0.3(2)
C16—C17—C22—C21	-178.1(3)	C68—C67—C74—N6	178.52 (19)
C20—C21—C22—C17	-1.3 (6)	N5—C67—C74—C75	175.57 (19)
N1—C8—N2—C16	-2.2 (2)	C68—C67—C74—C75	-5.6 (4)
Ag1—C8—N2—C16	168.31 (16)	C67—C74—C75—C76	68.8 (3)
N1—C8—N2—C23	-173.7 (2)	N6—C74—C75—C76	-115.9 (2)
Ag1—C8—N2—C23	-3.2 (3)	C67—C74—C75—C80	-112.0 (2)
C9—C16—N2—C8	1.6 (3)	N6—C74—C75—C80	63.3 (3)
C17—C16—N2—C8	-176.6 (2)	C80—C75—C76—C77	-1.3 (3)
C9-C16-N2-C23	173.0 (2)	C74—C75—C76—C77	177.84 (19)
C17—C16—N2—C23	-5.2 (3)	C75—C76—C77—C78	0.3 (3)
C8—N2—C23—C24	106.1 (2)	C76—C77—C78—C79	0.8 (3)
C16—N2—C23—C24	-64.3 (3)	C77—C78—C79—C80	-1.0(3)
N2-C23-C24-C25	126.8 (2)	C78—C79—C80—C75	0.0 (3)
N2—C23—C24—C29	-52.5 (3)	C76—C75—C80—C79	1.1 (3)
C29—C24—C25—C26	1.2 (4)	C74—C75—C80—C79	-178.06 (19)
C23—C24—C25—C26	-178.0(3)	N5-C66-N6-C74	0.1 (2)
C24—C25—C26—C27	0.6 (5)	Ag2-C66-N6-C74	-170.99(13)
C_{25} C_{26} C_{27} C_{28}	-1.3(6)	N5—C66—N6—C81	177.12 (16)
C_{26} C_{27} C_{28} C_{29}	0.2 (6)	Ag2-C66-N6-C81	6.0 (3)
C_{27} C_{28} C_{29} C_{24}	1.6(5)	C67 - C74 - N6 - C66	0.1(2)
C_{25} C_{26} C_{29} C_{24}	-23(4)	C75 - C74 - N6 - C66	-176.22(17)
$C_{23} = C_{24} = C_{29} = C_{28}$	2.5 (T) 176 9 (3)	$C_{7} = C_{7} = 10 = C_{00}$	-17679(17)
$C_{25} - C_{27} - C_{25} - C_{26}$	-0.4(5)	C75 - C74 - N6 - C81	60(3)
$C_{30} = C_{31} = C_{31} = C_{32}$	-0.8 (6)	$C_{13} - C_{14} - NO - Col$	-036(3)
$C_{30} - C_{31} - C_{32} - C_{33}$	0.0(0)	C00 - 10 - C01 - C02	93.0 (2) 82.0 (2)
$C_{31} - C_{32} - C_{33} - C_{34}$	0.5(5)	C/4 - NO - C01 - C02	03.0(2) -04.5(2)
C_{22} C_{23} C_{24} C_{25} C_{24}	1.1(3)	100-001-02-003	-94.3 (2) 85 2 (2)
$C_{21} = C_{20} = C_{25} = C_{24}$	2.0 (4)	$1 \times 0 - \times 1 - \times 2 - \times 2$	03.3 (2)
U31-U30-U35-U36	-1/0.5(3)	U8/	-1.7(3)

C33—C34—C35—C30	-2.3(4)	C81—C82—C83—C84	178.2 (2)
C33—C34—C35—C36	176.2 (3)	C82—C83—C84—C85	0.2 (4)
C_{30} C_{35} C_{36} N_{3}	-52.4(3)	C83 - C84 - C85 - C86	12(4)
C_{34} C_{35} C_{36} N_{3}	1291(2)	C84 - C85 - C86 - C87	-1.2(1)
C_{35} C_{36} N_{3} C_{37}	-653(3)	C85 - C86 - C87 - C82	-0.2(4)
C_{35} C_{36} C_{36} C_{38} C_{38}	111 4 (2)	$C_{00}^{83} = C_{00}^{83} = C_{00}^{87} = C_{00}^{86}$	1.7(3)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	111.4(2)	$C_{81} C_{82} C_{87} C_{86}$	-1781(2)
$C_{36} = N_{3} = C_{37} = N_{4}$	1.0(2) 178 10(18)	$C_{01} = C_{02} = C_{01} = C_{00}$	178.1(2)
$C_{30} = 103 = C_{37} = 104$	-172.49(10)	$C_{95} = C_{80} = C_{90} = C_{90}$	0.7(4)
$C_{30} = N_3 = C_{37} = A_{21}$	-1/5.40(14)	$C_{88} = C_{89} = C_{90} = C_{91}$	-0.3(3)
C30—N3— $C37$ —Agi	5.7(5)	$C_{89} = C_{90} = C_{91} = C_{92}$	0.4 (5)
$C_3/-N_3-C_{38}-C_{45}$	-0.6(2)	C90 - C91 - C92 - C93	-0.4 (5)
C36—N3—C38—C45	-177.58 (19)	C89—C88—C93—C92	-0.7 (4)
C37—N3—C38—C39	-179.36 (19)	C89—C88—C93—C94	175.1 (2)
C36—N3—C38—C39	3.6 (3)	C91—C92—C93—C88	0.5 (4)
C45—C38—C39—C44	59.2 (3)	C91—C92—C93—C94	-175.4 (3)
N3—C38—C39—C44	-122.3 (2)	C88—C93—C94—N7	38.4 (3)
C45—C38—C39—C40	-119.5 (3)	C92—C93—C94—N7	-145.8 (2)
N3—C38—C39—C40	59.0 (3)	C93—C94—N7—C95	-91.0 (2)
C44—C39—C40—C41	0.8 (4)	C93—C94—N7—C96	87.2 (2)
C38—C39—C40—C41	179.6 (2)	C96—N7—C95—N8	-1.4(2)
C39—C40—C41—C42	-0.1 (4)	C94—N7—C95—N8	177.10 (16)
C40—C41—C42—C43	-0.6 (4)	C96—N7—C95—Ag2	167.96 (14)
C41—C42—C43—C44	0.6 (5)	C94—N7—C95—Ag2	-13.6 (3)
C42—C43—C44—C39	0.1 (5)	C95—N7—C96—C103	1.5 (2)
C40—C39—C44—C43	-0.8(4)	C94—N7—C96—C103	-176.87 (18)
C38—C39—C44—C43	-179.5(2)	C95—N7—C96—C97	-169.15(19)
N3-C38-C45-N4	-0.1(2)	C94—N7—C96—C97	12.5 (3)
C_{39} C_{38} C_{45} N_{4}	178.6(2)	$C_{103} - C_{96} - C_{97} - C_{98}$	12.0(3) 137.9(2)
N_{3} C_{38} C_{45} C_{46}	-1754(2)	N7-C96-C97-C98	-53.6(3)
C_{39} C_{38} C_{45} C_{46}	33(4)	$C_{103} - C_{96} - C_{97} - C_{102}$	-43.7(3)
C_{38} C_{45} C_{46} C_{514}	-820(7)	N7 - C96 - C97 - C102	124.9(2)
$C_{36} - C_{43} - C_{40} - C_{51A}$	102.5(7)	$(102 \ C102 \ $	-22(2)
104 - 043 - 040 - 051A	103.3(7)	$C_{102} - C_{97} - C_{98} - C_{99}$	-2.2(3)
$C_{30} - C_{43} - C_{40} - C_{47A}$	90.2(7)	C90 - C97 - C98 - C99	170.3(2)
N4-C45-C46-C4/A	-78.3(7)	C97 - C98 - C99 - C100	1.5 (4)
$C_{38} - C_{45} - C_{46} - C_{47B}$	03.0 (8)		0.6 (4)
N4—C45—C46—C47B	-110.8 (7)	C99—C100—C101—C102	-2.0(4)
C38—C45—C46—C51B	-110.3 (8)	C98—C97—C102—C101	0.7 (3)
N4—C45—C46—C51B	75.2 (8)	C96—C97—C102—C101	-177.8 (2)
C51A—C46—C47A—C48A	1.6 (9)	C100—C101—C102—C97	1.3 (4)
C47B—C46—C47A—C48A	-69.1 (10)	N7—C96—C103—N8	-1.0(2)
C51B—C46—C47A—C48A	28.6 (9)	C97—C96—C103—N8	169.28 (19)
C45—C46—C47A—C48A	-176.6 (5)	N7—C96—C103—C104	-178.8 (2)
C46—C47A—C48A—C49	0.2 (10)	C97—C96—C103—C104	-8.5 (4)
C51A—C46—C47B—C48B	-26.2 (9)	C96—C103—C104—C109	-76.7 (3)
C47A—C46—C47B—C48B	95.3 (12)	N8-C103-C104-C109	105.8 (3)
C51B—C46—C47B—C48B	-1.8 (10)	C96—C103—C104—C105	99.7 (3)
C45—C46—C47B—C48B	-176.0 (6)	N8-C103-C104-C105	-77.8 (3)
C46—C47B—C48B—C49	-2.2 (12)	C109—C104—C105—C106	0.8 (3)

$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C47B—C48B—C49—C50A	30.1 (10)	C103—C104—C105—C106	-175.7 (2)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C47B—C48B—C49—C50B	5.1 (11)	C104—C105—C106—C107	-0.6 (4)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C47B—C48B—C49—C48A	-77.0 (11)	C105—C106—C107—C108	0.0 (4)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C47A—C48A—C49—C48B	84.6 (11)	C106—C107—C108—C109	0.5 (5)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C47A—C48A—C49—C50A	-2.2 (9)	C107—C108—C109—C104	-0.3 (5)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C47A—C48A—C49—C50B	-28.6 (9)	C105—C104—C109—C108	-0.3 (4)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C48B—C49—C50A—C51A	-29.5 (10)	C103—C104—C109—C108	176.1 (2)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C50B—C49—C50A—C51A	89 (2)	N7-C95-N8-C103	0.7 (2)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C48A—C49—C50A—C51A	2.3 (10)	Ag2-C95-N8-C103	-168.99 (14)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C47A—C46—C51A—C50A	-1.5 (9)	N7-C95-N8-C110	-176.58 (17)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C47B—C46—C51A—C50A	27.0 (9)	Ag2-C95-N8-C110	13.7 (3)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C51B—C46—C51A—C50A	-88.1 (18)	C96—C103—N8—C95	0.2 (2)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C45—C46—C51A—C50A	176.6 (6)	C104—C103—N8—C95	178.16 (19)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C49—C50A—C51A—C46	-0.6 (11)	C96—C103—N8—C110	177.40 (18)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C48B—C49—C50B—C51B	-3.8 (12)	C104—C103—N8—C110	-4.6 (3)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C50A—C49—C50B—C51B	-77.2 (19)	C95—N8—C110—C111	82.2 (2)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C48A—C49—C50B—C51B	28.1 (10)	C103—N8—C110—C111	-94.7 (2)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C49—C50B—C51B—C46	-0.4 (13)	N8-C110-C111-C116	16.9 (3)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C51A—C46—C51B—C50B	77.3 (16)	N8-C110-C111-C112	-165.55 (19)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C47A—C46—C51B—C50B	-29.3 (10)	C116—C111—C112—C113	1.7 (3)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C47B—C46—C51B—C50B	3.0 (11)	C110—C111—C112—C113	-175.9 (2)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C45—C46—C51B—C50B	177.2 (7)	C111—C112—C113—C114	-0.9 (4)
Ag1—C37—N4—C45173.29 (15)C113—C114—C115—C1161.4 (4)N3—C37—N4—C52177.28 (18)C114—C115—C116—C111 -0.6 (4)Ag1—C37—N4—C52 -8.4 (3)C112—C111—C116—C115 -0.9 (3)C38—C45—N4—C370.8 (2)C110—C111—C116—C115176.6 (2)C46—C45—N4—C37176.21 (19)C120—C121—O1—C122 -177.1 (3)C38—C45—N4—C52 -177.54 (19)C121—O1—C122—C123177.5 (3)	N3—C37—N4—C45	-1.1 (2)	C112—C113—C114—C115	-0.7 (4)
N3-C37-N4-C52 177.28 (18) C114-C115-C116-C111 -0.6 (4) Ag1-C37-N4-C52 -8.4 (3) C112-C111-C116-C115 -0.9 (3) C38-C45-N4-C37 0.8 (2) C110-C111-C116-C115 176.6 (2) C46-C45-N4-C37 176.21 (19) C120-C121-O1-C122 -177.1 (3) C38-C45-N4-C52 -177.54 (19) C121-O1-C122-C123 177.5 (3)	Ag1-C37-N4-C45	173.29 (15)	C113—C114—C115—C116	1.4 (4)
Ag1—C37—N4—C52-8.4 (3)C112—C111—C116—C115-0.9 (3)C38—C45—N4—C370.8 (2)C110—C111—C116—C115176.6 (2)C46—C45—N4—C37176.21 (19)C120—C121—O1—C122-177.1 (3)C38—C45—N4—C52-177.54 (19)C121—O1—C122—C123177.5 (3)	N3—C37—N4—C52	177.28 (18)	C114—C115—C116—C111	-0.6 (4)
C38—C45—N4—C37 0.8 (2) C110—C111—C116—C115 176.6 (2) C46—C45—N4—C37 176.21 (19) C120—C121—O1—C122 -177.1 (3) C38—C45—N4—C52 -177.54 (19) C121—O1—C122—C123 177.5 (3)	Ag1—C37—N4—C52	-8.4 (3)	C112—C111—C116—C115	-0.9 (3)
C46—C45—N4—C37176.21 (19)C120—C121—O1—C122-177.1 (3)C38—C45—N4—C52-177.54 (19)C121—O1—C122—C123177.5 (3)	C38—C45—N4—C37	0.8 (2)	C110—C111—C116—C115	176.6 (2)
C38—C45—N4—C52 -177.54 (19) C121—O1—C122—C123 177.5 (3)	C46—C45—N4—C37	176.21 (19)	C120-C121-O1-C122	-177.1 (3)
	C38—C45—N4—C52	-177.54 (19)	C121—O1—C122—C123	177.5 (3)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	Н…А	$D \cdots A$	D—H··· A	
С31—Н31…О1	0.95	2.57	3.361 (4)	141	
C32—H32…N10 ⁱ	0.95	2.43	3.327 (4)	157	
C40—H40…S1 ⁱⁱ	0.95	2.82	3.684 (2)	152	

Symmetry codes: (i) –*x*+2, –*y*+2, –*z*+1; (ii) *x*, *y*, *z*–1.

 $Bis(\mu-1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-\kappa^2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-2-selenoimidazole-k-2Se:Se)bis[bromido(1,3-dibenzyl-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidazole-k-2-selenoimidaz$ selenoimidazole-κSe)silver(I)] dichloromethane hexasolvate (2)

Crystal data	
$[Ag_2Br_2(C_{29}H_{24}N_2Se)_4]$ ·6CH ₂ Cl ₂	$\alpha = 106.4172 \ (7)^{\circ}$
$M_r = 2802.96$	$\beta = 112.2820 \ (8)^{\circ}$
Triclinic, $P\overline{1}$	$\gamma = 96.2211 \ (6)^{\circ}$
a = 13.6265 (1) Å	$V = 2930.04 (5) Å^3$
b = 14.7422 (1) Å	Z = 1
c = 16.9397 (2) Å	F(000) = 1400

 $D_x = 1.589 \text{ Mg m}^{-3}$ Mo K α radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 42526 reflections $\theta = 3.0-26.4^{\circ}$

Data collection

Rigaku SuperNova, Dual, Cu at zero, Atlas	
diffractometer	
Radiation source: micro-focus sealed X-ray tub	e
Detector resolution: 10.3196 pixels mm ⁻¹	
ω scans	
Absorption correction: gaussian	
(CrysAlis PRO; Rigaku OD, 2015)	
$T_{\min} = 0.517, \ T_{\max} = 0.664$	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.035$	Hydrogen site location: inferred from
$wR(F^2) = 0.096$	neighbouring sites
S = 1.03	H-atom parameters constrained
12002 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0468P)^2 + 5.4618P]$
654 parameters	where $P = (F_o^2 + 2F_o^2)/3$
0 restraints	$(\Delta/\sigma)_{max} = 0.002$
Primary atom site location: structure-invariant	$\Delta\rho_{max} = 1.05 \text{ e } \text{\AA}^{-3}$
direct methods	$\Delta\rho_{min} = -1.48 \text{ e } \text{\AA}^{-3}$

 $\mu = 2.58 \text{ mm}^{-1}$

Prism, clear pale yellow

119415 measured reflections 12002 independent reflections 10726 reflections with $I > 2\sigma(I)$

 $0.37 \times 0.26 \times 0.20$ mm

 $\theta_{\rm max} = 26.4^{\circ}, \ \theta_{\rm min} = 2.8^{\circ}$

T = 150 K

 $R_{\rm int} = 0.037$

 $h = -17 \rightarrow 17$ $k = -18 \rightarrow 18$ $l = -21 \rightarrow 21$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. The PLATON SQUEEZE procedure (A. L. Spek. Acta Cryst. C71, 2015, 9-18) was used to treat regions of disordered solvent which could not be modelled in terms of atomic sites. The number of electrons found in these regions, 84, was assigned to 2 molecules of dichloromethane. 2 dichloromethanes would give 84 electrons.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

	X	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
C1	0.2991 (3)	0.2585 (2)	0.5958 (2)	0.0303 (6)	
H1	0.3686	0.2453	0.6224	0.036*	
C2	0.2264 (3)	0.2008 (3)	0.5078 (2)	0.0390 (8)	
H2	0.2460	0.1483	0.4743	0.047*	
C3	0.1255 (3)	0.2194 (3)	0.4687 (3)	0.0476 (9)	
Н3	0.0752	0.1796	0.4083	0.057*	
C4	0.0978 (3)	0.2950 (3)	0.5167 (3)	0.0530 (10)	
H4	0.0283	0.3078	0.4894	0.064*	
C5	0.1709 (3)	0.3540 (3)	0.6060 (3)	0.0422 (8)	
Н5	0.1509	0.4065	0.6390	0.051*	
C6	0.2721 (2)	0.3356 (2)	0.6459 (2)	0.0281 (6)	
C7	0.3529 (2)	0.3973 (2)	0.7424 (2)	0.0277 (6)	
H7A	0.3201	0.4477	0.7680	0.033*	

H7B	0.3677	0.3555	0.7801	0.033*
N1	0.45650 (19)	0.44492 (16)	0.74753 (15)	0.0228 (5)
C8	0.5529 (2)	0.42207 (19)	0.78510 (18)	0.0227 (5)
Se1	0.57786 (3)	0.32825 (2)	0.83916 (2)	0.02863 (8)
C9	0.4711 (2)	0.52088 (19)	0.71564 (18)	0.0232 (5)
C10	0.3796 (2)	0.5572 (2)	0.66477 (19)	0.0246 (6)
C11	0.3552 (3)	0.5471 (2)	0.5752 (2)	0.0351 (7)
H11	0.3998	0.5197	0.5481	0.042*
C12	0.2666 (3)	0.5764 (3)	0.5246 (2)	0.0444 (9)
H12	0.2507	0.5691	0.4632	0.053*
C13	0.2019(3)	0.6158 (3)	0.5630(3)	0.0462 (9)
H13	0.1414	0.6362	0.5283	0.055*
C14	0.2241(4)	0.6259(3)	0.6511(3)	0.0519(10)
H14	0.1781	0.6521	0.6771	0.062*
C15	0.1701 0.3137 (3)	0.0521 0.5981 (3)	0.0771 0.7029(2)	0.002
H15	0.3137 (3)	0.5981 (5)	0.7629(2)	0.0412 (8)
C16	0.5300	0.0070	0.7047 0.72592 (19)	0.049°
C10 C17	0.3788(2)	0.34319(19)	0.73585(18) 0.7255((10))	0.0227(3)
C17	0.0400(2)	0.6262(2)	0.72556(19)	0.0252(6)
	0.6984 (3)	0.6095 (2)	0.6728 (2)	0.0329(7)
HI8	0.69/8	0.5446	0.6418	0.039*
C19	0.7568 (3)	0.6868 (3)	0.6651 (3)	0.0418 (8)
H19	0.7959	0.6749	0.6288	0.050*
C20	0.7583 (3)	0.7811 (3)	0.7102 (3)	0.0441 (8)
H20	0.7991	0.8341	0.7055	0.053*
C21	0.7003 (3)	0.7986 (2)	0.7623 (3)	0.0436 (8)
H21	0.7011	0.8638	0.7929	0.052*
C22	0.6411 (3)	0.7217 (2)	0.7702 (2)	0.0331 (7)
H22	0.6011	0.7340	0.8058	0.040*
N2	0.62857 (19)	0.48262 (16)	0.77814 (15)	0.0220 (5)
C23	0.7469 (2)	0.4912 (2)	0.82079 (19)	0.0267 (6)
H23A	0.7644	0.4699	0.8739	0.032*
H23B	0.7852	0.5608	0.8439	0.032*
C24	0.7915 (2)	0.4329 (2)	0.75860 (19)	0.0273 (6)
C25	0.7432 (3)	0.3347 (2)	0.7062 (2)	0.0311 (6)
H25	0.6773	0.3042	0.7050	0.037*
C26	0.7911 (3)	0.2812 (3)	0.6557 (2)	0.0387 (7)
H26	0.7580	0.2142	0.6206	0.046*
C27	0.8863 (3)	0.3247 (3)	0.6561 (2)	0.0393 (8)
H27	0.9187	0.2875	0.6216	0.047*
C28	0.9342(3)	0.4220 (3)	0.7065(2)	0.0413 (8)
H28	0.9996	0.4524	0 7069	0.050*
C29	0.8862 (3)	0.1521 0.4759(3)	0.7571(2)	0.0351(7)
H29	0.9190	0.5432	0.7912	0.042*
C30	0.5190	0.3452 0.1771 (3)	1.0909(2)	0.042
U30	0.5505 (5)	0.2240	1 1018	0.0491 (9)
C31	0.0202	0.22+0 0.1424 (4)	1.1010	0.057
	0.5507 (4)	0.1424 (4)	1.1500 (5)	0.0040(13)
пэт С22	0.0199	0.1000	1.210/	0.078°
C32	0.4709(4)	0.0730(3)	1.1427 (3)	0.0363 (12)

H32	0.4721	0.0474	1.1885	0.068*
C33	0.3819 (4)	0.0412 (3)	1.0607 (3)	0.0651 (13)
H33	0.3199	-0.0052	1.0498	0.078*
C34	0.3818 (4)	0.0765 (3)	0.9930 (3)	0.0558 (11)
H34	0.3197	0.0531	0.9356	0.067*
C35	0.4684 (3)	0.1437 (2)	1.00705 (19)	0.0283 (6)
C36	0.4722 (2)	0.1824 (2)	0.93410 (18)	0.0236 (5)
H36A	0.5324	0.2419	0.9626	0.028*
H36B	0.4878	0.1331	0.8900	0.028*
N3	0.36874 (18)	0.20544 (16)	0.88532 (14)	0.0195 (4)
C37	0.3351 (2)	0.28258 (19)	0.92241 (17)	0.0208 (5)
Se2	0.41926(2)	0.38665 (2)	1.03116 (2)	0.02432(7)
C38	0.2820(2)	0.14190(19)	0.80476(17)	0.02132(7)
C39	0.2828(2)	0.0524(2)	0.00170(17) 0.74254(18)	0.0200(5) 0.0222(5)
C40	0.2000(2) 0.3653(2)	0.0521(2) 0.0541(2)	0.71231(10)	0.0222(5)
H40	0.4178	0.1129	0.7258	0.0200 (0)
C41	0.3640 (3)	-0.0308(2)	0.7230	0.034
U41	0.3049(3)	-0.0207	0.6100	0.0344(7)
C42	0.4173 0.2887 (3)	-0.1161(2)	0.0199	0.041
C42	0.2887 (3)	-0.1726	0.0103 (2)	0.0348 (7)
П42 С42	0.2007	-0.1/30	0.5755	0.042°
C43	0.2124 (5)	-0.1183 (2)	0.0313(2)	0.0313(7)
П43	0.1001	-0.1775	0.0525	0.038°
C44	0.2119 (2)	-0.0345(2)	0.71439 (19)	0.0201 (0)
H44	0.1592	-0.0362	0.7383	0.031*
C45	0.1938 (2)	0.18073 (19)	0.79410 (18)	0.0217 (5)
C46	0.0815 (2)	0.14143 (19)	0.72006 (18)	0.0224 (5)
C47	0.0640 (2)	0.1324 (2)	0.63145 (19)	0.0273 (6)
H47	0.1229	0.1550	0.6194	0.033*
C48	-0.0397 (3)	0.0901 (2)	0.5609 (2)	0.0336 (7)
H48	-0.0521	0.0847	0.5004	0.040*
C49	-0.1246 (3)	0.0560 (2)	0.5781 (2)	0.0359 (7)
H49	-0.1954	0.0268	0.5294	0.043*
C50	-0.1075 (3)	0.0638 (2)	0.6657 (2)	0.0358 (7)
H50	-0.1664	0.0395	0.6769	0.043*
C51	-0.0047 (2)	0.1073 (2)	0.7372 (2)	0.0301 (6)
H51	0.0068	0.1136	0.7976	0.036*
N4	0.22814 (18)	0.26809 (16)	0.86718 (15)	0.0210 (4)
C52	0.1591 (2)	0.3352 (2)	0.87987 (19)	0.0236 (5)
H52A	0.1031	0.3288	0.8193	0.028*
H52B	0.2048	0.4028	0.9088	0.028*
C53	0.1023 (2)	0.3184 (2)	0.93731 (19)	0.0262 (6)
C54	0.1310 (3)	0.2618 (2)	0.9917 (2)	0.0325 (7)
H54	0.1874	0.2290	0.9922	0.039*
C55	0.0770 (3)	0.2531 (3)	1.0453 (2)	0.0434 (8)
H55	0.0962	0.2139	1.0820	0.052*
C56	-0.0046 (3)	0.3012 (3)	1.0454 (3)	0.0464 (9)
Н56	-0.0407	0.2956	1.0827	0.056*
C57	-0.0336 (3)	0.3573 (3)	0.9915 (3)	0.0463 (9)

H57	-0.0898	0.3903	0.9914	0.056*	
C58	0.0191 (3)	0.3656 (3)	0.9375 (2)	0.0375 (7)	
H58	-0.0017	0.4040	0.9001	0.045*	
Ag	0.59867 (2)	0.44437 (2)	1.00177 (2)	0.03306 (7)	
Br	0.77487 (3)	0.41077 (3)	1.11754 (2)	0.04318 (9)	
C59	0.4858 (4)	0.7506 (4)	0.5306 (3)	0.0671 (13)	
H59A	0.5389	0.7634	0.5941	0.081*	
H59B	0.4115	0.7398	0.5281	0.081*	
C11	0.49731 (8)	0.64544 (7)	0.45851 (6)	0.0459 (2)	
Cl2	0.50991 (12)	0.85328 (9)	0.50091 (11)	0.0817 (4)	
C60	0.7655 (4)	0.1514 (3)	0.8195 (3)	0.0628 (12)	
H60A	0.7439	0.1961	0.8623	0.075*	0.898 (4)
H60B	0.8157	0.1914	0.8058	0.075*	0.898 (4)
H60C	0.7912	0.2167	0.8192	0.075*	0.102 (4)
H60D	0.8253	0.1173	0.8271	0.075*	0.102 (4)
C13	0.64844 (9)	0.08367 (10)	0.71812 (10)	0.0764 (4)	
Cl4A	0.83340 (11)	0.07387 (9)	0.87093 (9)	0.0670 (5)	0.898 (4)
Cl4B	0.7203 (18)	0.1617 (16)	0.9129 (15)	0.120 (7)*	0.102 (4)

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0294 (15)	0.0298 (15)	0.0306 (15)	0.0035 (12)	0.0132 (13)	0.0099 (12)
C2	0.0415 (19)	0.0386 (18)	0.0322 (17)	-0.0023 (14)	0.0181 (15)	0.0073 (14)
C3	0.041 (2)	0.055 (2)	0.0329 (18)	-0.0089 (17)	0.0050 (16)	0.0178 (17)
C4	0.0272 (18)	0.057 (2)	0.064 (3)	0.0042 (16)	0.0014 (17)	0.032 (2)
C5	0.0316 (17)	0.0375 (18)	0.056 (2)	0.0100 (14)	0.0151 (16)	0.0194 (16)
C6	0.0261 (15)	0.0273 (14)	0.0339 (16)	0.0040 (11)	0.0143 (13)	0.0138 (12)
C7	0.0294 (15)	0.0283 (14)	0.0292 (15)	0.0058 (12)	0.0173 (13)	0.0094 (12)
N1	0.0287 (12)	0.0201 (11)	0.0191 (11)	0.0052 (9)	0.0111 (10)	0.0054 (9)
C8	0.0306 (15)	0.0183 (12)	0.0171 (12)	0.0058 (11)	0.0103 (11)	0.0030 (10)
Se1	0.04242 (18)	0.02315 (14)	0.02676 (15)	0.01194 (12)	0.01801 (13)	0.01214 (12)
C9	0.0297 (15)	0.0211 (13)	0.0175 (12)	0.0055 (11)	0.0101 (11)	0.0053 (10)
C10	0.0269 (14)	0.0212 (13)	0.0252 (14)	0.0067 (11)	0.0115 (12)	0.0068 (11)
C11	0.0392 (18)	0.0455 (18)	0.0283 (16)	0.0242 (15)	0.0173 (14)	0.0147 (14)
C12	0.048 (2)	0.054 (2)	0.0320 (17)	0.0304 (18)	0.0127 (16)	0.0164 (16)
C13	0.041 (2)	0.047 (2)	0.057 (2)	0.0287 (17)	0.0187 (18)	0.0240 (18)
C14	0.061 (2)	0.057 (2)	0.068 (3)	0.042 (2)	0.045 (2)	0.032 (2)
C15	0.059 (2)	0.0440 (19)	0.0388 (18)	0.0285 (17)	0.0317 (17)	0.0190 (15)
C16	0.0283 (14)	0.0220 (13)	0.0169 (12)	0.0072 (11)	0.0090 (11)	0.0060 (10)
C17	0.0267 (14)	0.0259 (14)	0.0211 (13)	0.0065 (11)	0.0067 (11)	0.0107 (11)
C18	0.0386 (17)	0.0327 (16)	0.0321 (16)	0.0095 (13)	0.0182 (14)	0.0134 (13)
C19	0.044 (2)	0.047 (2)	0.048 (2)	0.0112 (16)	0.0277 (17)	0.0258 (17)
C20	0.0414 (19)	0.0384 (18)	0.056 (2)	0.0012 (15)	0.0201 (17)	0.0258 (17)
C21	0.051 (2)	0.0247 (16)	0.050 (2)	0.0044 (14)	0.0191 (18)	0.0113 (15)
C22	0.0392 (17)	0.0256 (15)	0.0346 (17)	0.0069 (13)	0.0166 (14)	0.0102 (13)
N2	0.0263 (12)	0.0204 (11)	0.0182 (11)	0.0069 (9)	0.0085 (9)	0.0063 (9)
C23	0.0255 (14)	0.0277 (14)	0.0185 (13)	0.0042 (11)	0.0031 (11)	0.0059 (11)

C24	0.0247 (14)	0.0347 (15)	0.0221 (14)	0.0121 (12)	0.0058 (11)	0.0133 (12)
C25	0.0315 (16)	0.0305 (15)	0.0330 (16)	0.0108 (12)	0.0140 (13)	0.0120 (13)
C26	0.0450 (19)	0.0378 (17)	0.0352 (17)	0.0201 (15)	0.0166 (15)	0.0127 (14)
C27	0.0399 (18)	0.054 (2)	0.0351 (17)	0.0282 (16)	0.0193 (15)	0.0210 (16)
C28	0.0298 (17)	0.062 (2)	0.0378 (18)	0.0155 (16)	0.0146 (15)	0.0229 (17)
C29	0.0288 (16)	0.0428 (18)	0.0293 (16)	0.0059 (13)	0.0086 (13)	0.0128 (14)
C30	0.038 (2)	0.080 (3)	0.0338 (18)	0.0264 (19)	0.0095 (16)	0.0301 (19)
C31	0.064 (3)	0.106 (4)	0.040 (2)	0.045 (3)	0.017 (2)	0.046 (2)
C32	0.104 (4)	0.054 (2)	0.041 (2)	0.054 (3)	0.041 (2)	0.0347 (19)
C33	0.096 (4)	0.051 (2)	0.046 (2)	-0.004 (2)	0.030 (2)	0.024 (2)
C34	0.070 (3)	0.050(2)	0.0279 (18)	-0.0117 (19)	0.0063 (18)	0.0170 (16)
C35	0.0396 (17)	0.0255 (14)	0.0216 (14)	0.0170 (12)	0.0115 (13)	0.0096 (11)
C36	0.0210 (13)	0.0240 (13)	0.0201 (13)	0.0064 (10)	0.0026 (11)	0.0080 (11)
N3	0.0195 (11)	0.0202 (11)	0.0158 (10)	0.0044 (8)	0.0049 (9)	0.0060 (9)
C37	0.0234 (13)	0.0202 (12)	0.0154 (12)	0.0034 (10)	0.0061 (10)	0.0054 (10)
Se2	0.02608 (15)	0.02286 (14)	0.01577 (13)	0.00262 (11)	0.00472 (11)	0.00219 (10)
C38	0.0199 (13)	0.0222 (13)	0.0154 (12)	0.0030 (10)	0.0035 (10)	0.0062 (10)
C39	0.0215 (13)	0.0240 (13)	0.0156 (12)	0.0093 (10)	0.0020 (10)	0.0062 (10)
C40	0.0247 (14)	0.0323 (15)	0.0244 (14)	0.0096 (12)	0.0077 (12)	0.0093 (12)
C41	0.0331 (16)	0.0427 (18)	0.0276 (15)	0.0167 (14)	0.0142 (13)	0.0084 (13)
C42	0.0388 (17)	0.0315 (16)	0.0228 (15)	0.0169 (13)	0.0053 (13)	0.0012 (12)
C43	0.0307 (16)	0.0229 (14)	0.0270 (15)	0.0073 (12)	0.0012 (12)	0.0041 (12)
C44	0.0258 (14)	0.0253 (14)	0.0221 (13)	0.0068 (11)	0.0053 (11)	0.0080 (11)
C45	0.0233 (13)	0.0209 (13)	0.0164 (12)	0.0044 (10)	0.0055 (11)	0.0049 (10)
C46	0.0202 (13)	0.0210 (13)	0.0200 (13)	0.0074 (10)	0.0036 (11)	0.0050 (10)
C47	0.0294 (15)	0.0276 (14)	0.0213 (14)	0.0084 (12)	0.0071 (12)	0.0082 (11)
C48	0.0373 (17)	0.0317 (16)	0.0198 (14)	0.0104 (13)	0.0013 (13)	0.0064 (12)
C49	0.0260 (15)	0.0278 (15)	0.0321 (16)	0.0048 (12)	-0.0040 (13)	0.0031 (13)
C50	0.0234 (15)	0.0359 (17)	0.0383 (18)	0.0030 (12)	0.0079 (13)	0.0083 (14)
C51	0.0265 (15)	0.0343 (16)	0.0253 (15)	0.0063 (12)	0.0093 (12)	0.0076 (12)
N4	0.0216 (11)	0.0202 (11)	0.0175 (11)	0.0054 (9)	0.0059 (9)	0.0046 (9)
C52	0.0249 (14)	0.0214 (13)	0.0221 (13)	0.0094 (11)	0.0080 (11)	0.0057 (11)
C53	0.0246 (14)	0.0258 (14)	0.0197 (13)	0.0034 (11)	0.0065 (11)	0.0014 (11)
C54	0.0337 (16)	0.0324 (16)	0.0297 (16)	0.0061 (13)	0.0127 (13)	0.0107 (13)
C55	0.047 (2)	0.045 (2)	0.0382 (19)	0.0003 (16)	0.0190 (16)	0.0175 (16)
C56	0.042 (2)	0.051 (2)	0.0402 (19)	-0.0052 (16)	0.0238 (17)	0.0052 (16)
C57	0.0378 (19)	0.050(2)	0.048 (2)	0.0116 (16)	0.0242 (17)	0.0057 (17)
C58	0.0346 (17)	0.0434 (19)	0.0366 (18)	0.0158 (14)	0.0170 (15)	0.0125 (15)
Ag	0.03885 (14)	0.03004 (12)	0.02210 (12)	0.00221 (10)	0.00806 (10)	0.00678 (9)
Br	0.03466 (18)	0.0598 (2)	0.03813 (19)	0.01376 (16)	0.01402 (15)	0.02296 (16)
C59	0.078 (3)	0.079 (3)	0.044 (2)	0.035 (3)	0.031 (2)	0.008 (2)
Cl1	0.0467 (5)	0.0523 (5)	0.0417 (5)	0.0182 (4)	0.0221 (4)	0.0143 (4)
Cl2	0.0795 (8)	0.0507 (6)	0.1100 (11)	0.0188 (6)	0.0553 (8)	0.0001 (6)
C60	0.056 (3)	0.046 (2)	0.070 (3)	0.0032 (19)	0.023 (2)	0.006 (2)
C13	0.0393 (6)	0.0724 (8)	0.0863 (9)	0.0018 (5)	0.0126 (6)	0.0080 (6)
Cl4A	0.0709 (9)	0.0580 (8)	0.0583 (8)	-0.0081 (6)	0.0150 (6)	0.0272 (6)

Geometric parameters (Å, °)

C1—C2	1.381 (5)	C33—C34	1.387 (5)
C1—C6	1.390 (4)	С33—Н33	0.9500
C1—H1	0.9500	C34—C35	1.352 (5)
C2—C3	1.379 (5)	C34—H34	0.9500
C2—H2	0.9500	C35—C36	1.515 (4)
C3—C4	1.362 (6)	C36—N3	1.471 (3)
С3—Н3	0.9500	C36—H36A	0.9900
C4—C5	1.400 (6)	С36—Н36В	0.9900
C4—H4	0.9500	N3—C37	1.351 (3)
C5—C6	1.383 (5)	N3—C38	1.397 (3)
С5—Н5	0.9500	C37—N4	1.353 (3)
C6—C7	1.512 (4)	C37—Se2	1.866 (3)
C7—N1	1.466 (4)	Se2—Ag ⁱ	2.7187 (4)
С7—Н7А	0.9900	Se2—Ag	2.7677 (4)
С7—Н7В	0.9900	C38—C45	1.359 (4)
N1—C8	1.352 (4)	C38—C39	1.475 (4)
N1—C9	1.400 (3)	C39—C40	1.386 (4)
C8—N2	1.351 (4)	C39—C44	1.397 (4)
C8—Se1	1.857 (3)	C40—C41	1.395 (4)
Sel—Ag	2.6899 (4)	C40—H40	0.9500
C9—C16	1.354 (4)	C41—C42	1.377 (5)
C9—C10	1.475 (4)	C41—H41	0.9500
C10—C11	1.384 (4)	C42—C43	1.380 (5)
C10—C15	1.386 (4)	C42—H42	0.9500
C11—C12	1.384 (4)	C43—C44	1.387 (4)
C11—H11	0.9500	C43—H43	0.9500
C12—C13	1.368 (5)	C44—H44	0.9500
C12—H12	0.9500	C45—N4	1.392 (3)
C13—C14	1.365 (6)	C45—C46	1.481 (4)
C13—H13	0.9500	C46—C51	1.389 (4)
C14—C15	1.389 (5)	C46—C47	1.392 (4)
C14—H14	0.9500	C47—C48	1.387 (4)
C15—H15	0.9500	C47—H47	0.9500
C16—N2	1.402 (3)	C48—C49	1.374 (5)
C16—C17	1.476 (4)	C48—H48	0.9500
C17—C18	1.389 (4)	C49—C50	1.380 (5)
C17—C22	1.396 (4)	C49—H49	0.9500
C18—C19	1.384 (5)	C50—C51	1.386 (4)
C18—H18	0.9500	C50—H50	0.9500
C19—C20	1.377 (5)	C51—H51	0.9500
C19—H19	0.9500	N4—C52	1.465 (3)
C20—C21	1.384 (5)	C52—C53	1.507 (4)
С20—Н20	0.9500	C52—H52A	0.9900
C21—C22	1.387 (5)	C52—H52B	0.9900
C21—H21	0.9500	C53—C54	1.389 (4)
С22—Н22	0.9500	C53—C58	1.395 (4)

N2—C23	1.469 (4)	C54—C55	1.392 (5)
C23—C24	1.518 (4)	C54—H54	0.9500
C23—H23A	0.9900	C55—C56	1.383 (6)
С23—Н23В	0.9900	С55—Н55	0.9500
C24—C29	1.387 (4)	C56—C57	1.378 (6)
C24—C25	1.393 (4)	C56—H56	0.9500
C25—C26	1.391 (5)	C57—C58	1.381 (5)
С25—Н25	0.9500	С57—Н57	0.9500
C26—C27	1.381 (5)	C58—H58	0.9500
C26—H26	0.9500	Ag—Br	2.6631 (4)
C27—C28	1.377 (5)	Ag—Se2 ⁱ	2.7186 (4)
C27—H27	0.9500	C59—C11	1.746 (4)
C28-C29	1 395 (5)	C59—Cl2	1 761 (6)
C28—H28	0.9500	C59—H59A	0.9900
C29—H29	0.9500	C59—H59B	0,9900
C_{30} $-C_{31}$	1 384 (6)	C60-C13	1.760(5)
C30-C35	1.385(5)	C60 - C14A	1.762(5)
C30—H30	0.9500	C60 - C14B	1.702(0)
C_{31} C_{32}	1.372(7)	C60—H60A	0.9900
C31—H31	0.9500	C60 - H60R	0.9900
C_{32} C_{33}	1.360(7)	C60—H60C	0.9900
С32—Н32	0.9500	C60—H60D	0.9900
052 1152	0.9500		0.9900
C2-C1-C6	120.9 (3)	C35—C34—H34	119.4
C2—C1—H1	119.5	C33—C34—H34	119.4
C6-C1-H1	119.5	C34 - C35 - C30	118.8 (3)
C3-C2-C1	119.9 (3)	C_{34} C_{35} C_{36}	122.7(3)
C3—C2—H2	120.0	C30—C35—C36	118.4 (3)
C1—C2—H2	120.0	N3-C36-C35	112.0(2)
C4-C3-C2	1199(3)	N3-C36-H36A	109.2
C4-C3-H3	120.0	C35—C36—H36A	109.2
C2-C3-H3	120.0	N3-C36-H36B	109.2
$C_{3} - C_{4} - C_{5}$	120.0 120.7(3)	C35—C36—H36B	109.2
C3—C4—H4	119.7	H36A—C36—H36B	107.9
C5-C4-H4	119.7	C_{37} N3 C_{38}	107.9 109.6 (2)
C6C5C4	119.8 (4)	C_{37} N3 C_{36}	109.0(2) 123.6(2)
Сб-С5-Н5	120.1	C_{38} N3 $-C_{36}$	125.0(2) 125.3(2)
C4	120.1	N3-C37-N4	125.5(2) 106.8(2)
$C_{-}^{-}C$	120.1 118.7(3)	$N_3 = C_37 = Se^2$	100.0(2) 1264(2)
$C_{5} = C_{6} = C_{7}$	110.7(3) 121 $4(3)$	$N_{1} = C_{3}^{3} = S_{2}^{3}$	126.4(2)
C_{1} C_{6} C_{7}	121.4(3) 110.8(3)	C37 Se2 Ag ⁱ	120.0(2) 108.43(8)
CI = CO = C/	119.8(3) 112.7(2)	C37 = Sc2 = Ag	108.43(8) 100.72(8)
NI = C7 = H7A	112.7 (2)	Δg^{i} So2 Δg	73640(11)
$\Gamma = C = \Pi / \Lambda$	109.1	$\frac{Ag}{C45} \frac{-3}{C38} \frac{N3}{N3}$	1060(2)
N1 C7 H7P	102.1	$C_{45} = C_{50} = N_5$	100.7(2) 127.7(2)
мі—с/—п/в С6_С7_ H7R	107.1	N3 - C38 - C39	127.7(2) 125A(2)
$U_{\rm II}$ $U_{\rm II}$ $U_{\rm II}$ $U_{\rm II}$ $U_{\rm II}$ $U_{\rm II}$	107.1	C40 C20 C44	123.4(2) 110.5(2)
$\Pi/A = U/= \Pi/D$	107.0	C40 - C39 - C44	119.3(3)
CO-INI-C9	109.8 (2)	C40-C39-C38	121.4 (3)

C8—N1—C7	124.5 (2)	C44—C39—C38	118.9 (3)
C9—N1—C7	125.7 (2)	C39—C40—C41	119.9 (3)
N2—C8—N1	106.8 (2)	C39—C40—H40	120.1
N2—C8—Se1	126.4 (2)	C41—C40—H40	120.1
N1—C8—Se1	126.8 (2)	C42—C41—C40	120.3 (3)
C8—Se1—Ag	94 72 (8)	C42—C41—H41	119.9
C16 - C9 - N1	1067(2)	C40-C41-H41	119.9
$C_{16} - C_{9} - C_{10}$	130.3(3)	C_{41} C_{42} C_{43}	120.2(3)
N1 - C9 - C10	122 8 (3)	$C_{41} - C_{42} - H_{42}$	119.9
$C_{11} C_{10} C_{15}$	122.0(3) 118.5(3)	$C_{41} C_{42} H_{42}$	110.0
$C_{11} = C_{10} = C_{13}$	110.5(3)	C_{42} C_{42} C_{42} C_{44}	119.9 120.2(3)
$C_{11} = C_{10} = C_{9}$	119.0(3)	$C_{42} = C_{43} = C_{44}$	120.2 (3)
$C_{10} = C_{10} = C_{10}$	121.3(3) 120.7(2)	$C_{42} = C_{43} = 1143$	119.9
C10 - C11 - C12	120.7 (5)	$C_{44} = C_{43} = H_{43}$	119.9
	119.0	C43 - C44 - C39	120.0 (3)
	119.6	C43—C44—H44	120.0
C13—C12—C11	120.1 (3)	C39—C44—H44	120.0
C13—C12—H12	120.0	C38—C45—N4	106.9 (2)
C11—C12—H12	120.0	C38—C45—C46	128.4 (2)
C14—C13—C12	120.2 (3)	N4—C45—C46	124.7 (2)
C14—C13—H13	119.9	C51—C46—C47	119.9 (3)
C12—C13—H13	119.9	C51—C46—C45	120.6 (3)
C13—C14—C15	120.2 (3)	C47—C46—C45	119.3 (3)
C13—C14—H14	119.9	C48—C47—C46	119.7 (3)
C15—C14—H14	119.9	C48—C47—H47	120.2
C10-C15-C14	120.3 (3)	C46—C47—H47	120.2
C10—C15—H15	119.8	C49—C48—C47	120.2 (3)
C14—C15—H15	119.8	C49—C48—H48	119.9
C9—C16—N2	107.2 (2)	C47—C48—H48	119.9
C9—C16—C17	129.5 (3)	C48—C49—C50	120.4 (3)
N2-C16-C17	123.1 (2)	C48—C49—H49	119.8
C18—C17—C22	119.4 (3)	С50—С49—Н49	119.8
C18—C17—C16	121.6 (3)	C49—C50—C51	120.2(3)
C^{22} C^{17} C^{16}	1190(3)	C49—C50—H50	119.9
C19 - C18 - C17	120 5 (3)	C51—C50—H50	119.9
C19 - C18 - H18	119.8	C_{50} C_{51} C_{46}	119.5 119.6(3)
C17 - C18 - H18	119.8	C_{50} C_{51} H_{51}	120.2
C_{20} C_{19} C_{18}	120.0 (3)	$C_{46} C_{51} H_{51}$	120.2
$C_{20} = C_{10} = C_{10}$	120.0 (3)	$C_{40} = C_{51} = H_{51}$	120.2 100.7 (2)
$C_{20} = C_{19} = 1119$	120.0	$C_{37} = N_{4} = C_{43}$	109.7(2) 125.6(2)
$C_{10} = C_{20} = C_{21}$	120.0 120.1(2)	$C_{37} = 104 = C_{32}$	123.0(2)
C19 - C20 - C21	120.1 (5)	C43 - 104 - C32	124.7(2)
C19 - C20 - H20	120.0	N4-C52-C53	114.1 (2)
C21—C20—H20	120.0	N4—C52—H52A	108.7
C20—C21—C22	120.4 (3)	C53—C52—H52A	108.7
C20—C21—H21	119.8	N4—C52—H52B	108.7
C22—C21—H21	119.8	C53—C52—H52B	108.7
C21—C22—C17	119.6 (3)	H52A—C52—H52B	107.6
C21—C22—H22	120.2	C54—C53—C58	118.9 (3)
C17—C22—H22	120.2	C54—C53—C52	123.7 (3)

C9 NO $C1($	100.5(2)	C59 C52 C52	1172(2)
$C_0 = N_2 = C_{10}$	109.5 (2)	$C_{38} = C_{33} = C_{32}$	117.5 (5)
C8 - N2 - C23	124.8 (2)	053-054-055	119.9 (3)
C16—N2—C23	125.1 (2)	С53—С54—Н54	120.0
N2—C23—C24	115.0 (2)	С55—С54—Н54	120.0
N2—C23—H23A	108.5	C56—C55—C54	120.4 (3)
C24—C23—H23A	108.5	С56—С55—Н55	119.8
N2—C23—H23B	108.5	С54—С55—Н55	119.8
С24—С23—Н23В	108.5	C57—C56—C55	120.0 (3)
H23A—C23—H23B	107.5	С57—С56—Н56	120.0
C29—C24—C25	118.5 (3)	С55—С56—Н56	120.0
C29—C24—C23	119.8 (3)	C56—C57—C58	120.0 (3)
$C_{25} - C_{24} - C_{23}$	121.6(3)	C56—C57—H57	120.0
$C_{25} = C_{25} = C_{25}$	121.0(3) 120.2(3)	C58—C57—H57	120.0
$C_{26} C_{25} C_{24}$	110.0	C_{57} C_{58} C_{53}	120.8 (3)
$C_{20} = C_{23} = H_{23}$	119.9	$C_{57} = C_{58} = C_{55}$	120.8 (5)
$C_{24} = C_{23} = H_{23}$	119.9	$C_{52} = C_{58} = H_{58}$	119.0
$C_2/-C_{20}-C_{23}$	120.7 (3)	C33—C38—H38	119.0
C27—C26—H26	119.7	Br—Ag—Sel	102.274 (13)
С25—С26—Н26	119.7	Br—Ag—Se ²¹	126.883 (14)
C28—C27—C26	119.8 (3)	$Se1$ —Ag— $Se2^{1}$	100.026 (11)
С28—С27—Н27	120.1	Br—Ag—Se2	109.628 (12)
С26—С27—Н27	120.1	Se1—Ag—Se2	110.623 (12)
C27—C28—C29	119.7 (3)	Se2 ⁱ —Ag—Se2	106.352 (11)
C27—C28—H28	120.2	Cl1—C59—Cl2	112.3 (3)
C29—C28—H28	120.2	Cl1—C59—H59A	109.1
C24—C29—C28	121.2 (3)	Cl2—C59—H59A	109.1
С24—С29—Н29	119.4	Cl1—C59—H59B	109.1
С28—С29—Н29	119.4	C12—C59—H59B	109.1
C31—C30—C35	120.1 (4)	H59A—C59—H59B	107.9
C31—C30—H30	119.9	C13—C60—C14A	110.9 (2)
$C_{35} - C_{30} - H_{30}$	119.9	C_{13} C_{60} C_{14B}	1049(7)
C_{32} C_{31} C_{30}	120.2 (4)	C13 - C60 - H60A	109.5
$C_{32} = C_{31} = C_{30}$	110.0		109.5
$C_{32} = C_{31} = H_{31}$	119.9	$C_{14}^{12} = C_{00}^{10} = H_{00}^{10} R_{00}^{10}$	109.5
C_{22} C_{22} C_{21}	119.9	$C_{13} = C_{00} = H_{00} B$	109.5
$C_{33} = C_{32} = C_{31}$	119.5 (4)		109.5
C33—C32—H32	120.2	H60A - C60 - H60B	108.0
С31—С32—Н32	120.2	Cl3—C60—H60C	110.8
C32—C33—C34	120.0 (4)	Cl4B—C60—H60C	110.8
С32—С33—Н33	120.0	Cl3—C60—H60D	110.8
С34—С33—Н33	120.0	Cl4B—C60—H60D	110.8
C35—C34—C33	121.2 (4)	H60C—C60—H60D	108.8
C6_C1_C2_C3	0.0(5)	C_{30} C_{31} C_{32} C_{33}	-23(7)
$C_1 = C_2 = C_3$	0.0(3)	$C_{30} = C_{31} = C_{32} = C_{33}$	2.3(7)
$C_1 - C_2 - C_3 - C_4$	-0.3(5)	$C_{31} - C_{32} - C_{33} - C_{34}$	2.0(7)
$C_2 = C_4 = C_5 = C_4$	0.5(0)	$C_{22} = C_{23} = C_{24} = C_{25} = C_{20}$	1.0 (8)
$C_{4} = C_{5} = C_{6} = C_{1}$	0.1(0)	$C_{22} = C_{24} = C_{25} = C_{26}$	0.1(0)
	0.2 (3)	$C_{33} - C_{34} - C_{35} - C_{36}$	1/8./(4)
C4—C5—C6—C7	-179.6 (3)	C31—C30—C35—C34	-0.4 (6)
C2-C1-C6-C5	-0.2(5)	C31—C30—C35—C36	-179.0(4)

C2—C1—C6—C7	179.6 (3)	C34—C35—C36—N3	44.2 (4)
C5—C6—C7—N1	-120.5 (3)	C30-C35-C36-N3	-137.3 (3)
C1—C6—C7—N1	59.7 (4)	C35—C36—N3—C37	72.1 (3)
C6—C7—N1—C8	-111.0 (3)	C35—C36—N3—C38	-92.7 (3)
C6—C7—N1—C9	69.7 (3)	C38—N3—C37—N4	0.8 (3)
C9—N1—C8—N2	0.4 (3)	C36—N3—C37—N4	-166.1 (2)
C7—N1—C8—N2	-179.0 (2)	C38—N3—C37—Se2	178.82 (19)
C9—N1—C8—Se1	178.68 (19)	C36—N3—C37—Se2	11.9 (4)
C7—N1—C8—Se1	-0.7(4)	$N3$ — $C37$ — $Se2$ — Ag^i	128.8 (2)
N2—C8—Se1—Ag	94.9 (2)	$N4-C37-Se2-Ag^{i}$	-53.6(2)
N1—C8—Se1—Ag	-831(2)	$N3-C37-Se^2-Ag$	52.6(2)
C8-N1-C9-C16	-10(3)	$N4-C37-Se^2-Ag$	-1297(2)
C7 - N1 - C9 - C16	1784(2)	C_{37} N3 C_{38} C45	-1.2(3)
C_{8} N1 C_{9} C10	175.7(2)	$C_{36} = N_{3} = C_{38} = C_{45}$	1655(2)
C7 - N1 - C9 - C10	-5.5(4)	C_{37} N3 C_{38} C39	1765(2)
C_16 C_9 C_{10} C_{11}	59.5 (4)	C_{36}^{-1} N3 C_{38}^{-2} C39	-169(4)
N1 = C9 = C10 = C11	-1158(3)	$C_{30} = \frac{1}{13} = \frac{1}{230} = \frac{1}{230$	10.9(4)
11 - 0 - 010 - 011	-122.2(4)	$N_{2} = C_{3}^{20} = C_{3}^{20} = C_{40}^{40}$	-54.1(4)
C10 - C9 - C10 - C15	-123.2(4)	$N_{3} = C_{30} = C_{40} = C_{40}$	-52.8(4)
NI = C9 = C10 = C13	01.0(4)	C43 - C36 - C39 - C44	-32.0(4)
C15 - C10 - C11 - C12	-0.5(5)	N_{3} C_{38} C_{39} C_{44} $C_$	130.1(3)
C_{9} C_{10} C_{11} C_{12} C_{12}	1/0.9(3)	C44 - C39 - C40 - C41	-0.1(4)
C10-C11-C12-C13	0.0 (6)	$C_{38} - C_{39} - C_{40} - C_{41}$	-1/6.0(3)
	-0.3(6)	C39 - C40 - C41 - C42	0.2 (5)
C12—C13—C14—C15	1.2 (7)	C40—C41—C42—C43	-0.3 (5)
C11—C10—C15—C14	1.4 (5)	C41—C42—C43—C44	0.2 (5)
C9—C10—C15—C14	-176.0 (3)	C42—C43—C44—C39	-0.1 (4)
C13—C14—C15—C10	-1.8 (6)	C40—C39—C44—C43	0.1 (4)
N1—C9—C16—N2	1.2 (3)	C38—C39—C44—C43	176.0 (2)
C10—C9—C16—N2	-174.6 (3)	N3—C38—C45—N4	1.0 (3)
N1—C9—C16—C17	-173.6 (3)	C39—C38—C45—N4	-176.5 (3)
C10-C9-C16-C17	10.6 (5)	N3—C38—C45—C46	-179.5 (3)
C9—C16—C17—C18	-121.8 (3)	C39—C38—C45—C46	2.9 (5)
N2-C16-C17-C18	64.1 (4)	C38—C45—C46—C51	111.2 (3)
C9—C16—C17—C22	58.3 (4)	N4—C45—C46—C51	-69.5 (4)
N2-C16-C17-C22	-115.7 (3)	C38—C45—C46—C47	-64.2 (4)
C22—C17—C18—C19	0.6 (5)	N4—C45—C46—C47	115.2 (3)
C16—C17—C18—C19	-179.3 (3)	C51—C46—C47—C48	0.7 (4)
C17—C18—C19—C20	0.2 (5)	C45—C46—C47—C48	176.1 (3)
C18—C19—C20—C21	-0.8 (6)	C46—C47—C48—C49	-1.0 (4)
C19—C20—C21—C22	0.5 (6)	C47—C48—C49—C50	0.3 (5)
C20—C21—C22—C17	0.3 (5)	C48—C49—C50—C51	0.7 (5)
C18—C17—C22—C21	-0.9(5)	C49—C50—C51—C46	-1.0(5)
C16—C17—C22—C21	179.0 (3)	C47—C46—C51—C50	0.3 (4)
N1—C8—N2—C16	0.4 (3)	C45—C46—C51—C50	-175.0 (3)
Se1-C8-N2-C16	-177.93 (19)	N3—C37—N4—C45	-0.1 (3)
N1-C8-N2-C23	171.8 (2)	Se2—C37—N4—C45	-178.15 (19)
Se1—C8—N2—C23	-6.5 (4)	N3—C37—N4—C52	-178.2 (2)
C9—C16—N2—C8	-1.0(3)	Se2—C37—N4—C52	3.8 (4)
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C17—C16—N2—C8	174.2 (2)	C38—C45—N4—C37	-0.6(3)	
C9—C16—N2—C23	-172.4 (2)	C46—C45—N4—C37	180.0 (2)	
C17—C16—N2—C23	2.8 (4)	C38—C45—N4—C52	177.5 (2)	
C8—N2—C23—C24	97.6 (3)	C46—C45—N4—C52	-2.0 (4)	
C16—N2—C23—C24	-92.3 (3)	C37—N4—C52—C53	-89.4 (3)	
N2—C23—C24—C29	134.4 (3)	C45—N4—C52—C53	92.9 (3)	
N2—C23—C24—C25	-49.4 (4)	N4—C52—C53—C54	14.8 (4)	
C29—C24—C25—C26	1.5 (4)	N4—C52—C53—C58	-167.5 (3)	
C23—C24—C25—C26	-174.7 (3)	C58—C53—C54—C55	-0.1 (5)	
C24—C25—C26—C27	-0.6 (5)	C52—C53—C54—C55	177.5 (3)	
C25—C26—C27—C28	-0.3 (5)	C53—C54—C55—C56	-0.5 (5)	
C26—C27—C28—C29	0.2 (5)	C54—C55—C56—C57	0.7 (6)	
C25—C24—C29—C28	-1.7 (5)	C55—C56—C57—C58	-0.1 (6)	
C23—C24—C29—C28	174.7 (3)	C56—C57—C58—C53	-0.5 (5)	
C27—C28—C29—C24	0.8 (5)	C54—C53—C58—C57	0.7 (5)	
C35—C30—C31—C32	1.5 (7)	C52—C53—C58—C57	-177.2 (3)	

Symmetry code: (i) -x+1, -y+1, -z+2.

 $catena - Poly[[[(1,3-dibenzyl-4,5-diphenyl-2-selenoimidazole-\kappa Se)copper(I)] - \mu - cyanido-\kappa^2 C:N] \ acetonitrile$

monosolvate] (3)

Crystal data

 $[Cu(CN)(C_{29}H_{24}N_2Se)] \cdot C_2H_3N$ $M_r = 610.07$ Monoclinic, $P2_1/c$ a = 13.7704 (3) Å b = 14.3398 (3) Å c = 28.4102 (7) Å $\beta = 93.024$ (2)° V = 5602.2 (2) Å³ Z = 8

Data collection

Rigaku SuperNova, Dual, Cu at zero, Atlas diffractometer Radiation source: micro-focus sealed X-ray tube Detector resolution: 10.3196 pixels mm⁻¹ ω scans Absorption correction: gaussian (CrysAlis PRO; Rigaku OD, 2015) $T_{\min} = 0.643, T_{\max} = 0.851$

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.048$ $wR(F^2) = 0.119$ S = 1.0611392 reflections 687 parameters 0 restraints F(000) = 2480 $D_x = 1.447 \text{ Mg m}^{-3}$ Mo K\alpha radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 23662 reflections $\theta = 3.0-26.3^{\circ}$ $\mu = 2.11 \text{ mm}^{-1}$ T = 100 KRod, colourless $0.35 \times 0.12 \times 0.11 \text{ mm}$

71725 measured reflections 11392 independent reflections 10122 reflections with $I > 2\sigma(I)$ $R_{int} = 0.038$ $\theta_{max} = 26.4^\circ, \ \theta_{min} = 2.8^\circ$ $h = -17 \rightarrow 17$ $k = -17 \rightarrow 17$ $l = -34 \rightarrow 35$

Primary atom site location: structure-invariant direct methods Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.045P)^{2} + 17.9881P] \qquad \Delta \rho_{max} = 2.04 \text{ e } \text{\AA}^{-3}$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3 \qquad \Delta \rho_{min} = -0.56 \text{ e } \text{\AA}^{-3}$ $(\Delta/\sigma)_{max} = 0.002$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
C1	0.7661 (3)	0.8105 (3)	0.56492 (13)	0.0266 (8)
H1	0.8151	0.7909	0.5448	0.032*
C2	0.6869 (3)	0.8612 (3)	0.54673 (14)	0.0324 (9)
H2	0.6819	0.8758	0.5141	0.039*
C3	0.6153 (3)	0.8906 (3)	0.57562 (15)	0.0332 (9)
Н3	0.5620	0.9265	0.5631	0.040*
C4	0.6218 (3)	0.8673 (3)	0.62283 (15)	0.0343 (9)
H4	0.5720	0.8859	0.6427	0.041*
C5	0.7008 (3)	0.8171 (3)	0.64125 (13)	0.0253 (7)
Н5	0.7051	0.8019	0.6738	0.030*
C6	0.7738 (2)	0.7886 (2)	0.61249 (12)	0.0189 (7)
C7	0.8551 (2)	0.7292 (2)	0.63438 (12)	0.0204 (7)
H7A	0.8727	0.7532	0.6664	0.024*
H7B	0.8310	0.6646	0.6377	0.024*
N1	0.9424 (2)	0.72729 (19)	0.60732 (9)	0.0183 (6)
C8	0.9750 (2)	0.6511 (2)	0.58509 (11)	0.0200 (7)
C9	1.0034 (2)	0.8028 (2)	0.60041 (11)	0.0174 (6)
C10	0.9878 (2)	0.8956 (2)	0.62140 (11)	0.0177 (6)
C11	0.9911 (2)	0.9067 (2)	0.67034 (12)	0.0218 (7)
H11	1.0039	0.8545	0.6903	0.026*
C12	0.9757 (3)	0.9937 (3)	0.68992 (13)	0.0262 (8)
H12	0.9773	1.0007	0.7232	0.031*
C13	0.9580 (3)	1.0703 (3)	0.66121 (14)	0.0281 (8)
H13	0.9476	1.1298	0.6747	0.034*
C14	0.9554 (3)	1.0602 (3)	0.61274 (14)	0.0284 (8)
H14	0.9438	1.1130	0.5930	0.034*
C15	0.9697 (3)	0.9730 (2)	0.59275 (12)	0.0237 (7)
H15	0.9671	0.9663	0.5594	0.028*
C16	1.0749 (2)	0.7722 (2)	0.57321 (11)	0.0188 (7)
C17	1.1600 (2)	0.8218 (2)	0.55600 (11)	0.0180 (6)
C18	1.1500 (3)	0.8798 (3)	0.51648 (13)	0.0301 (8)
H18	1.0883	0.8867	0.5003	0.036*
C19	1.2298 (3)	0.9274 (3)	0.50083 (15)	0.0357 (9)
H19	1.2227	0.9667	0.4739	0.043*
C20	1.3199 (3)	0.9176 (3)	0.52444 (14)	0.0313 (8)
H20	1.3743	0.9508	0.5139	0.038*

C21	1.3312 (3)	0.8598 (3)	0.56325 (13)	0.0281 (8)
H21	1.3933	0.8525	0.5790	0.034*
C22	1.2508 (3)	0.8122 (3)	0.57926 (12)	0.0235 (7)
H22	1.2582	0.7730	0.6062	0.028*
N2	1.0562 (2)	0.67822 (19)	0.56411 (9)	0.0195 (6)
C23	1.1121 (3)	0.6179 (2)	0.53376 (12)	0.0224 (7)
H23A	1.1495	0.6576	0.5127	0.027*
H23B	1.0663	0.5800	0.5137	0.027*
C24	1.1821 (3)	0.5528 (3)	0.56083 (13)	0.0253 (7)
C25	1.2045 (3)	0.5611 (3)	0.60863 (13)	0.0269 (8)
H25	1.1780	0.6111	0.6257	0.032*
C26	1.2657 (3)	0.4965 (3)	0.63204 (15)	0.0384 (10)
H26	1.2812	0.5033	0.6648	0.046*
C27	1.3040 (3)	0.4226 (4)	0.60772 (16)	0.0453 (11)
H27	1.3435	0.3774	0.6239	0.054*
C28	1.2840 (3)	0.4152 (3)	0.55930 (16)	0.0464 (11)
H28	1.3121	0.3661	0.5421	0.056*
C29	1.2230 (3)	0.4793 (3)	0.53623 (15)	0.0390 (10)
H29	1.2089	0.4733	0.5033	0.047*
Se1	0.92117 (3)	0.53117 (2)	0.58409 (2)	0.02398 (10)
Cu1	0.95522 (3)	0.49927 (3)	0.66587 (2)	0.02577 (11)
C59	0.9042 (3)	0.3973 (3)	0.69925 (13)	0.0282 (8)
N5	0.8749 (2)	0.3390 (2)	0.72193 (12)	0.0322 (7)
Cu2	0.82825 (3)	0.24571 (3)	0.76454 (2)	0.02719 (12)
Se2	0.66152 (3)	0.27327 (3)	0.77892 (2)	0.02914 (10)
C30	0.5614 (4)	0.3668 (4)	0.59119 (16)	0.0474 (11)
H30	0.5037	0.4033	0.5905	0.057*
C31	0.6036 (5)	0.3434 (5)	0.54929 (18)	0.0644 (16)
H31	0.5758	0.3649	0.5200	0.077*
C32	0.6879 (4)	0.2878 (4)	0.55079 (18)	0.0641 (17)
H32	0.7136	0.2659	0.5224	0.077*
C33	0.7316 (5)	0.2661 (5)	0.5926 (2)	0.084 (2)
H33	0.7924	0.2347	0.5937	0.101*
C34	0.6894 (4)	0.2887 (4)	0.63443 (18)	0.0614 (16)
H34	0.7205	0.2706	0.6636	0.074*
C35	0.6040 (3)	0.3366 (3)	0.63423 (13)	0.0280 (8)
C36	0.5573 (3)	0.3625 (3)	0.67936 (13)	0.0262 (8)
H36A	0.4960	0.3965	0.6714	0.031*
H36B	0.5404	0.3047	0.6961	0.031*
N3	0.6190 (2)	0.4204 (2)	0.71106 (10)	0.0216 (6)
C37	0.6612 (3)	0.3937 (2)	0.75286 (12)	0.0232 (7)
C38	0.6364 (2)	0.5153 (2)	0.70361 (12)	0.0199 (7)
C39	0.6004 (2)	0.5646 (2)	0.66026 (12)	0.0208 (7)
C40	0.6554 (3)	0.5642 (3)	0.62088 (13)	0.0283 (8)
H40	0.7164	0.5330	0.6219	0.034*
C41	0.6218 (3)	0.6090 (3)	0.57994 (14)	0.0364 (10)
H41	0.6600	0.6091	0.5531	0.044*
C42	0.5336 (3)	0.6530(3)	0.57830 (15)	0.0427 (11)

H42	0.5104	0.6835	0.5502	0.051*
C43	0.4786 (3)	0.6529 (4)	0.61719 (19)	0.0559 (14)
H43	0.4170	0.6831	0.6157	0.067*
C44	0.5114 (3)	0.6097 (3)	0.65840 (16)	0.0430 (11)
H44	0.4732	0.6109	0.6853	0.052*
C45	0.6898 (2)	0.5460 (2)	0.74189 (12)	0.0194 (7)
C46	0.7299 (2)	0.6395 (2)	0.75337 (11)	0.0194 (7)
C47	0.8295 (3)	0.6547 (2)	0.75481 (12)	0.0233 (7)
H47	0.8723	0.6051	0.7480	0.028*
C48	0.8669 (3)	0.7425 (3)	0.76618 (12)	0.0276 (8)
H48	0.9352	0.7525	0.7673	0.033*
C49	0.8050 (3)	0.8150 (3)	0.77590 (13)	0.0313 (9)
H49	0.8305	0.8750	0.7836	0.038*
C50	0.7054 (3)	0.7999 (3)	0.77436 (13)	0.0314 (8)
H50	0.6629	0.8497	0.7811	0.038*
C51	0.6676 (3)	0.7132 (3)	0.76306 (12)	0.0251 (7)
H51	0.5993	0.7035	0.7619	0.030*
N4	0.7045 (2)	0.4698(2)	0.77188 (10)	0.0214 (6)
C52	0.7539 (3)	0.4747 (2)	0.81887 (12)	0.0235(7)
H52A	0.8209	0.4984	0.8159	0.028*
H52B	0.7586	0.4112	0.8325	0.028*
C53	0.7010 (3)	0.5374 (2)	0.85174 (12)	0.0223(7)
C54	0.5996 (3)	0.5417 (3)	0.84997 (13)	0.0292(8)
H54	0.5627	0.5049	0.8277	0.035*
C55	0.5525(3)	0 5998 (3)	0.8277 0.88072 (15)	0.0381 (10)
H55	0.4835	0.6028	0.8793	0.046*
C56	0.6058 (3)	0.6533(3)	0.91339(16)	0.0426(10)
H56	0.5736	0.6932	0.9342	0.051*
C57	0,7057 (3)	0.6483(3)	0.9312 0.91558 (15)	0.0411 (10)
H57	0.7424	0.6846	0.9381	0.049*
C58	0.7534(3)	0 5903 (3)	0 88486 (13)	0.0302(8)
H58	0.8224	0.5871	0.8867	0.036*
C60	0.0221 0.9083 (3)	0.1490(2)	0.79035(12)	0.0238(7)
N6	0.9596(2)	0.0943(2)	0.80625(11)	0.0280(7)
N7	0.5766(3)	0.0913(2) 0.4987(3)	0.24282(14)	0.0200(7) 0.0445(9)
C61	0.5700(3)	0.4766 (3)	0.21202(11) 0.22229(16)	0.0365(9)
C62	0.0400(5) 0.7238(4)	0.4700(5) 0.4507(4)	0.22229(10) 0.1964(2)	0.0505(9)
Н62А	0.7024	0.4155	0.1682	0.094*
H62R	0.7677	0.4121	0.2164	0.094*
H62C	0.7581	0.5072	0.1871	0.094*
N8	0.9405 (3)	0.7629 (3)	0.1071 0.47523(14)	0.094
C63	0.9333 (3)	0.7029(3)	0.47925(14) 0.43914(15)	0.0404(10)
C64	0.9355(3)	0.6019 (3)	0 30148 (16)	0.0370(9) 0.0476(11)
Ц64 A	0.9203 (+)	0.7245	0.3727	0.071*
H6/B	0.0755	0.7273	0.3020	0.071*
H64C	0.9109	0.0233	0.3929	0.071*
11040	0.7000	0.7001	0.3709	0.071

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0223 (18)	0.035 (2)	0.0229 (18)	0.0024 (15)	0.0034 (14)	0.0067 (15)
C2	0.030 (2)	0.039 (2)	0.028 (2)	-0.0039 (17)	-0.0033 (16)	0.0129 (17)
C3	0.0228 (19)	0.029 (2)	0.047 (2)	0.0033 (15)	-0.0051 (17)	0.0048 (17)
C4	0.0237 (19)	0.038 (2)	0.041 (2)	0.0055 (16)	0.0071 (17)	-0.0092 (18)
C5	0.0246 (18)	0.0293 (19)	0.0222 (18)	0.0007 (14)	0.0043 (14)	-0.0037 (14)
C6	0.0183 (16)	0.0174 (15)	0.0212 (16)	-0.0028 (12)	0.0018 (13)	-0.0007 (13)
C7	0.0200 (16)	0.0244 (17)	0.0173 (16)	0.0015 (13)	0.0062 (13)	0.0041 (13)
N1	0.0178 (14)	0.0198 (14)	0.0172 (13)	0.0013 (11)	0.0015 (11)	0.0004 (11)
C8	0.0218 (17)	0.0209 (16)	0.0170 (16)	0.0032 (13)	-0.0014 (13)	0.0002 (13)
C9	0.0169 (16)	0.0216 (16)	0.0135 (15)	0.0027 (12)	-0.0005 (12)	0.0018 (12)
C10	0.0155 (15)	0.0195 (16)	0.0184 (16)	0.0019 (12)	0.0037 (12)	-0.0004 (12)
C11	0.0187 (16)	0.0263 (17)	0.0204 (17)	0.0001 (13)	0.0004 (13)	-0.0008 (14)
C12	0.0219 (18)	0.0318 (19)	0.0251 (18)	-0.0009 (14)	0.0047 (14)	-0.0084 (15)
C13	0.0225 (18)	0.0231 (18)	0.039 (2)	0.0000 (14)	0.0069 (16)	-0.0098 (16)
C14	0.0283 (19)	0.0226 (18)	0.035 (2)	0.0040 (15)	0.0062 (16)	0.0042 (15)
C15	0.0266 (18)	0.0240 (17)	0.0207 (17)	0.0031 (14)	0.0040 (14)	0.0003 (14)
C16	0.0223 (17)	0.0209 (16)	0.0131 (15)	0.0050 (13)	-0.0019 (12)	0.0016 (12)
C17	0.0193 (16)	0.0207 (16)	0.0144 (15)	0.0036 (13)	0.0035 (12)	-0.0014 (12)
C18	0.0244 (18)	0.040 (2)	0.0260 (19)	0.0019 (16)	0.0010 (15)	0.0080 (16)
C19	0.032 (2)	0.043 (2)	0.033 (2)	-0.0007 (18)	0.0044 (17)	0.0143 (18)
C20	0.0261 (19)	0.032 (2)	0.037 (2)	-0.0038 (16)	0.0085 (16)	0.0020 (17)
C21	0.0207 (18)	0.033 (2)	0.031 (2)	0.0021 (15)	0.0011 (15)	-0.0050 (16)
C22	0.0243 (18)	0.0279 (18)	0.0185 (16)	0.0024 (14)	0.0021 (14)	0.0005 (14)
N2	0.0196 (14)	0.0224 (14)	0.0165 (14)	0.0035 (11)	0.0022 (11)	-0.0019 (11)
C23	0.0254 (18)	0.0247 (17)	0.0172 (16)	0.0039 (14)	0.0035 (14)	-0.0032 (13)
C24	0.0227 (18)	0.0296 (19)	0.0235 (18)	0.0052 (14)	0.0008 (14)	-0.0017 (15)
C25	0.0266 (19)	0.0305 (19)	0.0237 (18)	0.0027 (15)	0.0015 (14)	-0.0045 (15)
C26	0.035 (2)	0.053 (3)	0.026 (2)	0.0072 (19)	-0.0058 (17)	-0.0002 (18)
C27	0.038 (2)	0.057 (3)	0.040 (2)	0.016 (2)	-0.0028 (19)	0.009 (2)
C28	0.048 (3)	0.050 (3)	0.041 (2)	0.025 (2)	0.000 (2)	-0.004(2)
C29	0.046 (2)	0.044 (2)	0.027 (2)	0.017 (2)	-0.0013 (18)	-0.0035 (18)
Se1	0.02574 (19)	0.02198 (18)	0.02412 (19)	-0.00181 (13)	0.00024 (14)	-0.00282 (13)
Cu1	0.0309 (2)	0.0204 (2)	0.0263 (2)	-0.00198 (17)	0.00483 (18)	0.00115 (17)
C59	0.036 (2)	0.0238 (18)	0.0250 (19)	-0.0014 (15)	0.0041 (16)	-0.0006 (15)
N5	0.0349 (18)	0.0259 (17)	0.0360 (18)	0.0016 (14)	0.0032 (14)	0.0023 (14)
Cu2	0.0365 (3)	0.0186 (2)	0.0263 (2)	0.00050 (18)	0.00004 (19)	0.00181 (17)
Se2	0.0374 (2)	0.02157 (18)	0.0290 (2)	-0.00150 (15)	0.00708 (16)	0.00452 (14)
C30	0.043 (3)	0.065 (3)	0.035 (2)	-0.007(2)	-0.001(2)	-0.007(2)
C31	0.072 (4)	0.090 (4)	0.031 (3)	-0.021 (3)	0.002 (2)	-0.013 (3)
C32	0.066 (4)	0.089 (4)	0.039 (3)	-0.022 (3)	0.024 (3)	-0.037 (3)
C33	0.076 (4)	0.116 (6)	0.060 (4)	0.047 (4)	0.015 (3)	-0.030 (4)
C34	0.065 (3)	0.083 (4)	0.036 (3)	0.034 (3)	0.002 (2)	-0.013 (3)
C35	0.033 (2)	0.0274 (19)	0.0243 (18)	-0.0089 (15)	0.0049 (15)	-0.0091 (15)
C36	0.0266 (19)	0.0271 (18)	0.0253 (18)	-0.0062 (15)	0.0028 (15)	-0.0051 (15)
N3	0.0247 (15)	0.0220 (14)	0.0185 (14)	-0.0006 (12)	0.0041 (12)	-0.0025 (11)

C37	0.0268 (18)	0.0210 (17)	0.0224 (17)	0.0010 (14)	0.0070 (14)	-0.0007 (14)
C38	0.0188 (16)	0.0213 (16)	0.0202 (17)	0.0006 (13)	0.0056 (13)	-0.0002 (13)
C39	0.0215 (17)	0.0225 (16)	0.0181 (16)	-0.0036 (13)	-0.0009 (13)	-0.0023 (13)
C40	0.033 (2)	0.0285 (19)	0.0243 (18)	-0.0025 (15)	0.0062 (15)	-0.0016 (15)
C41	0.054 (3)	0.036 (2)	0.0202 (19)	-0.0099 (19)	0.0053 (18)	-0.0018 (16)
C42	0.052 (3)	0.044 (2)	0.030 (2)	-0.014 (2)	-0.018 (2)	0.0143 (19)
C43	0.033 (2)	0.069 (3)	0.065 (3)	0.012 (2)	-0.003 (2)	0.035 (3)
C44	0.033 (2)	0.057 (3)	0.040 (2)	0.015 (2)	0.0140 (19)	0.020 (2)
C45	0.0185 (16)	0.0216 (16)	0.0188 (16)	0.0041 (13)	0.0068 (13)	-0.0007 (13)
C46	0.0248 (17)	0.0213 (16)	0.0123 (15)	0.0006 (13)	0.0010 (13)	0.0010 (12)
C47	0.0267 (18)	0.0257 (18)	0.0175 (16)	0.0010 (14)	0.0017 (14)	0.0015 (13)
C48	0.030 (2)	0.032 (2)	0.0205 (17)	-0.0070 (16)	0.0002 (15)	0.0021 (15)
C49	0.049 (2)	0.0245 (19)	0.0204 (18)	-0.0069 (17)	-0.0023 (16)	0.0005 (14)
C50	0.044 (2)	0.0229 (18)	0.0271 (19)	0.0058 (16)	0.0042 (17)	-0.0024 (15)
C51	0.0247 (18)	0.0271 (18)	0.0235 (18)	0.0025 (14)	0.0032 (14)	-0.0005 (14)
N4	0.0261 (15)	0.0209 (14)	0.0174 (14)	0.0020 (12)	0.0027 (11)	0.0013 (11)
C52	0.0244 (18)	0.0251 (17)	0.0209 (17)	0.0003 (14)	-0.0015 (14)	0.0051 (14)
C53	0.0267 (18)	0.0249 (17)	0.0155 (16)	-0.0027 (14)	0.0035 (13)	0.0037 (13)
C54	0.0261 (19)	0.036 (2)	0.0251 (19)	-0.0069 (16)	0.0014 (15)	-0.0022 (16)
C55	0.028 (2)	0.054 (3)	0.032 (2)	0.0029 (19)	0.0047 (17)	-0.0054 (19)
C56	0.048 (3)	0.044 (2)	0.038 (2)	-0.002 (2)	0.014 (2)	-0.0121 (19)
C57	0.046 (3)	0.046 (2)	0.032 (2)	-0.018 (2)	0.0035 (19)	-0.0149 (19)
C58	0.029 (2)	0.036 (2)	0.0264 (19)	-0.0104 (16)	0.0034 (15)	-0.0013 (16)
C60	0.0297 (19)	0.0225 (17)	0.0195 (17)	-0.0037 (15)	0.0045 (14)	0.0009 (14)
N6	0.0349 (18)	0.0267 (16)	0.0229 (16)	-0.0004 (14)	0.0078 (13)	-0.0035 (13)
N7	0.037 (2)	0.047 (2)	0.050 (2)	-0.0062 (17)	0.0076 (18)	0.0022 (18)
C61	0.031 (2)	0.037 (2)	0.042 (2)	-0.0038 (18)	-0.0033 (19)	0.0119 (18)
C62	0.044 (3)	0.064 (3)	0.081 (4)	0.023 (3)	0.020 (3)	0.030 (3)
N8	0.042 (2)	0.072 (3)	0.032 (2)	-0.003 (2)	-0.0019 (17)	-0.0028 (19)
C63	0.032 (2)	0.036 (2)	0.034 (2)	0.0025 (17)	-0.0027 (17)	0.0068 (18)
C64	0.064 (3)	0.040 (2)	0.038 (2)	0.012 (2)	0.003 (2)	-0.005 (2)

Geometric parameters (Å, °)

C1—C6	1.386 (5)	С30—Н30	0.9500
C1—C2	1.388 (5)	C31—C32	1.407 (9)
С1—Н1	0.9500	C31—H31	0.9500
C2—C3	1.382 (6)	C32—C33	1.340 (9)
С2—Н2	0.9500	C32—H32	0.9500
C3—C4	1.381 (6)	C33—C34	1.389 (7)
С3—Н3	0.9500	С33—Н33	0.9500
C4—C5	1.384 (5)	C34—C35	1.360 (6)
C4—H4	0.9500	C34—H34	0.9500
C5—C6	1.391 (5)	C35—C36	1.512 (5)
С5—Н5	0.9500	C36—N3	1.464 (4)
C6—C7	1.513 (5)	C36—H36A	0.9900
C7—N1	1.461 (4)	C36—H36B	0.9900
С7—Н7А	0.9900	N3—C37	1.350 (5)

С7—Н7В	0.9900	N3—C38	1,399 (4)
N1—C8	1.351 (4)	C37—N4	1.343 (5)
N1—C9	1 391 (4)	C38—C45	1354(5)
C8—N2	1 351 (4)	C_{38} C_{39}	1483(5)
C8—Se1	1.872 (3)	C39—C40	1.103(5)
C9—C16	1.372(5) 1.356(5)	C39—C44	1.385(5)
C9-C10	1.556(5) 1 478(4)	C40-C41	1.305(5) 1.387(6)
C10-C15	1 391 (5)	C40 H40	0.9500
C10-C11	1.391(5) 1 398(5)	C41 - C42	1.367(7)
C_{11} C_{12}	1.396 (5)	C41 H41	0.0500
C11_U11	0.0500	C42 - C42	1,272(7)
C_{11}	0.9300	$C_{42} = C_{43}$	1.372(7)
C_{12} C_{13} C_{12} C_{13} C_{12} C_{13} C	0.0500	C_{42} C_{42} C_{44}	1 270 (6)
C12 $-C14$	0.9300	$C_{43} = C_{44}$	1.379(0)
$C_{12} = U_{12}$	1.365 (3)	C43—n43	0.9300
C13—H13	0.9500	C44—H44	0.9500
	1.391 (3)	C45 - N4	1.394 (4)
	0.9500	C45-C46	1.480 (5)
	0.9500		1.387 (5)
C16—N2	1.393 (4)	C46—C51	1.397 (5)
C16—C17	1.478 (5)	C47—C48	1.391 (5)
C17—C22	1.389 (5)	C47—H47	0.9500
C17—C18	1.398 (5)	C48—C49	1.382 (6)
C18—C19	1.386 (6)	C48—H48	0.9500
C18—H18	0.9500	C49—C50	1.386 (6)
C19—C20	1.386 (6)	C49—H49	0.9500
С19—Н19	0.9500	C50—C51	1.380 (5)
C20—C21	1.382 (6)	С50—Н50	0.9500
C20—H20	0.9500	C51—H51	0.9500
C21—C22	1.398 (5)	N4—C52	1.468 (4)
C21—H21	0.9500	C52—C53	1.512 (5)
С22—Н22	0.9500	С52—Н52А	0.9900
N2—C23	1.468 (4)	С52—Н52В	0.9900
C23—C24	1.521 (5)	C53—C58	1.382 (5)
С23—Н23А	0.9900	C53—C54	1.395 (5)
С23—Н23В	0.9900	C54—C55	1.392 (6)
C24—C25	1.382 (5)	С54—Н54	0.9500
C24—C29	1.399 (5)	C55—C56	1.384 (6)
C25—C26	1.397 (6)	С55—Н55	0.9500
С25—Н25	0.9500	C56—C57	1.375 (6)
C26—C27	1.384 (6)	С56—Н56	0.9500
C26—H26	0.9500	С57—С58	1.395 (6)
C27—C28	1.393 (6)	С57—Н57	0.9500
С27—Н27	0.9500	С58—Н58	0.9500
C28—C29	1.386 (6)	C60—N6	1.134 (5)
C28—H28	0.9500	N6—Cu1 ⁱⁱ	1.939 (3)
С29—Н29	0.9500	N7—C61	1.121 (6)
Se1—Cu1	2.3900 (6)	C61—C62	1.450 (7)
Cu1—C59	1.898 (4)	С62—Н62А	0.9800
	× /		

C-1 NG	1 020 (2)	C(2) 11(2D	0.0000
C_{1} C_{1} C_{2} C_{2	1.939 (3)	C_{02} — $H_{02}B$ C_{02} — $H_{02}C$	0.9800
N5 Cu2	1.141(3) 1.037(3)	N8 C63	1 127 (6)
n_{3} — $c_{u_{2}}$	1.937(3)	C_{63}	1.127(0) 1.460(6)
$Cu^2 = So^2$	1.095(4)	C64 H64A	0.0800
So2 C37	2.3801(0) 1.870(2)	C64 H64P	0.9800
$Se_2 - C_3 / C_3 $	1.079(3) 1.202(7)		0.9800
C30—C31	1.395 (7)	С04—п04С	0.9800
0.30-0.35	1.397 (0)		
C6—C1—C2	120.0 (3)	C30—C31—C32	119.4 (5)
C6-C1-H1	120.0	C_{30} C_{31} H_{31}	120.3
C2-C1-H1	120.0	C_{32} — C_{31} — H_{31}	120.3
C_{3} C_{2} C_{1}	120.0 120.7(4)	C_{33} C_{32} C_{31} C_{31}	119 3 (5)
C3-C2-H2	1197	C_{33} C_{32} H_{32}	120.3
C1 - C2 - H2	119.7	C_{31} C_{32} H_{32}	120.3
C4-C3-C2	119.7	C_{32} C_{33} C_{34}	121.0 (6)
C4-C3-H3	120.3	C_{32} C_{33} H_{33}	119.5
C_{2} C_{3} H_{3}	120.3	C_{34} C_{33} H_{33}	119.5
$C_2 = C_3 = H_3$	120.3 120.2(4)	C_{35} C_{36} C	117.5 121.0 (5)
$C_3 - C_4 - H_4$	110.0	C_{35} C_{34} H_{34}	119.5
C_{5} C_{4} H_{4}	119.9	C_{33} C_{34} H_{34}	119.5
C4-C5-C6	120.6 (3)	C_{34} C_{35} C_{30}	119.9 118.9(4)
C4 - C5 - H5	110.7	C_{34} C_{35} C_{36}	110.9(4) 121.8(4)
C6 C5 H5	119.7	C_{30} C_{35} C_{36}	121.0(4) 110.2(4)
$C_1 C_6 C_5$	119.7	$N_3 = C_3^{-36} = C_3^{-35}$	113.2(4)
C1 - C6 - C7	119.1(3) 123.1(3)	N3-C36-H364	108.8
$C_1 = C_0 = C_1$	123.1(3) 117.8(3)	C_{35} C_{36} H_{36A}	108.8
$C_3 = C_0 = C_7$	117.8(3) 113.9(3)	N3 C36 H36B	108.8
N1 C7 H7A	108.8	C35 C36 H36B	108.8
C6-C7-H7A	108.8	H_{364} C_{36} H_{36B}	107.7
N1 - C7 - H7B	108.8	C37_N3_C38	107.7 109.9(3)
C6-C7-H7B	108.8	C_{37} N3 C_{36}	107.7(3) 126.1(3)
H7A - C7 - H7B	107.7	C_{38} N3 C_{36}	120.1(3) 123.8(3)
C8-N1-C9	110 2 (3)	N4-C37-N3	125.0(3) 106 5 (3)
C8-N1-C7	1244(3)	$N4-C37-Se^2$	126.6(3)
C9-N1-C7	1254(3)	N3-C37-Se2	126.9(3)
N2-C8-N1	106.0(3)	C45-C38-N3	106.6(3)
N2-C8-Se1	1265(2)	C45-C38-C39	130.9(3)
N1 - C8 - Se1	127.5 (3)	N3-C38-C39	122.5(3)
C16-C9-N1	106.8 (3)	C40—C39—C44	119.5 (3)
C16-C9-C10	1303(3)	C40-C39-C38	119.8 (3)
N1-C9-C10	122.9 (3)	C44-C39-C38	120.7(3)
C15-C10-C11	119.1 (3)	C39—C40—C41	120.2(4)
C15—C10—C9	120.5 (3)	C39—C40—H40	119.9
C11—C10—C9	120.4 (3)	C41—C40—H40	119.9
C12—C11—C10	120.3 (3)	C42—C41—C40	120.0 (4)
C12—C11—H11	119.9	C42—C41—H41	120.0
C10-C11-H11	119.9	C40—C41—H41	120.0

C13—C12—C11	120.3 (3)	C41—C42—C43	120.0 (4)
C13—C12—H12	119.9	C41—C42—H42	120.0
C11—C12—H12	119.9	C43—C42—H42	120.0
C12—C13—C14	119.9 (3)	C42—C43—C44	120.9 (4)
С12—С13—Н13	120.0	C42—C43—H43	119.6
C14—C13—H13	120.0	C44-C43-H43	119.6
C_{13} C_{14} C_{15}	120.3(3)	C_{43} — C_{44} — C_{39}	119.0 119.5(4)
C_{13} C_{14} H_{14}	119.9	C43 - C44 - H44	120.3
C_{15} C_{14} H_{14}	119.9	C_{39} C_{44} H_{44}	120.3
C_{10} C_{15} C_{14}	120.2(3)	C_{38} C_{45} N_4	120.5 106.9(3)
$C_{10} = C_{15} = C_{14}$	120.2 (3)	$C_{38} = C_{45} = C_{46}$	100.9(3)
$C_{10} - C_{15} - H_{15}$	119.9	$V_{40} = C_{40} = C_{40}$	130.8(3)
C_{14} C_{15} H_{15}	119.9	$\mathbf{N4} - \mathbf{C43} - \mathbf{C40}$	122.4(3)
$C_{9} = C_{16} = N_{2}$	100.7(3)	C47 - C40 - C51	119.4 (3)
C9—C16—C17	130.1 (3)	C4/-C46-C45	120.4 (3)
N2—C16—C17	123.1 (3)	C51—C46—C45	120.1 (3)
C22—C17—C18	119.3 (3)	C46—C47—C48	120.2 (3)
C22—C17—C16	120.3 (3)	C46—C47—H47	119.9
C18—C17—C16	120.4 (3)	C48—C47—H47	119.9
C19—C18—C17	120.3 (3)	C49—C48—C47	120.2 (4)
C19—C18—H18	119.9	C49—C48—H48	119.9
C17—C18—H18	119.9	C47—C48—H48	119.9
C18—C19—C20	119.9 (4)	C48—C49—C50	119.7 (4)
С18—С19—Н19	120.0	C48—C49—H49	120.2
С20—С19—Н19	120.0	С50—С49—Н49	120.2
C21—C20—C19	120.5 (4)	C51—C50—C49	120.6 (4)
C21—C20—H20	119.8	С51—С50—Н50	119.7
С19—С20—Н20	119.8	С49—С50—Н50	119.7
C20—C21—C22	119.7 (3)	C50—C51—C46	119.9 (3)
C20—C21—H21	120.2	C50-C51-H51	120.0
$C_{22} = C_{21} = H_{21}$	120.2	C46-C51-H51	120.0
C17 - C22 - C21	120.2 120.3(3)	C_{37} N4 C_{45}	1101(3)
C_{17} C_{22} C_{21}	110.8	C_{37} N/ C_{52}	125 4 (3)
$C_{11} = C_{22} = H_{22}$	119.8	$C_{37} = N_{4} = C_{32}$	123.4(3)
$C_{21} = C_{22} = 1122$	119.0 110.2(2)	$C_{43} - 104 - C_{52}$	124.3(3)
C_{0} N2 C_{2}	110.2(3)	N4 = C52 = C53	112.0(3)
$C_0 = N_2 = C_{23}$	124.1(3)	N4 - C52 - H52A	109.2
C10 - N2 - C23	125.5(3)	C55—C52—H52A	109.2
N2-C23-C24	113.8 (3)	N4—C52—H52B	109.2
N2—C23—H23A	108.8	С53—С52—Н52В	109.2
С24—С23—Н23А	108.8	H52A—C52—H52B	107.9
N2—C23—H23B	108.8	C58—C53—C54	119.1 (3)
C24—C23—H23B	108.8	C58—C53—C52	119.7 (3)
H23A—C23—H23B	107.7	C54—C53—C52	121.2 (3)
C25—C24—C29	118.7 (3)	C55—C54—C53	120.2 (4)
C25—C24—C23	123.1 (3)	С55—С54—Н54	119.9
C29—C24—C23	118.1 (3)	С53—С54—Н54	119.9
C24—C25—C26	120.6 (4)	C56—C55—C54	120.3 (4)
C24—C25—H25	119.7	С56—С55—Н55	119.9
С26—С25—Н25	119.7	С54—С55—Н55	119.9

C27—C26—C25	120.4 (4)	C57—C56—C55	119.7 (4)
C27—C26—H26	119.8	С57—С56—Н56	120.2
С25—С26—Н26	119.8	С55—С56—Н56	120.2
C26—C27—C28	119.4 (4)	C56—C57—C58	120.4 (4)
С26—С27—Н27	120.3	С56—С57—Н57	119.8
C28—C27—H27	120.3	C58—C57—H57	119.8
C_{29} C_{28} C_{27}	119.9 (4)	C53-C58-C57	120.4 (4)
C29—C28—H28	120.0	C53-C58-H58	119.8
C27—C28—H28	120.0	C57—C58—H58	119.8
C_{28} C_{29} C_{24}	120.9 (4)	N6	176.7 (3)
C_{28} C_{29} H_{29}	119.6	$C60 - N6 - Cu1^{ii}$	178.6(3)
C_{24} C_{29} H_{29}	119.6	N7-C61-C62	178.0(5)
C8—Se1—Cu1	95.96 (10)	C_{61} C_{62} H_{62A}	109.5
$C59$ — $Cu1$ — $N6^{i}$	124 68 (14)	C61 - C62 - H62R	109.5
C59 Cul $R61$	124.06(14) 125.06(12)	H62A - C62 - H62B	109.5
$N6^{i}$ $Cu1$ Sel	110 25 (9)	C_{61} C_{62} H_{62} H_{62}	109.5
N5 $C59$ $Cu1$	175.6(3)	H62A $C62$ $H62C$	109.5
$C_{50} = N_5 = C_{11}^2$	175.0(3) 175.7(3)	H62R - C62 - H62C	109.5
C_{3} C_{1} C_{2} C_{2} C_{2} C_{2} C_{3} C_{2} C_{3} C_{3	173.7(3) 122.84(15)	$N_{2}C_{6}C_{6}C_{6}C_{6}C_{6}C_{6}C_{6}C_{6$	109.3 177.2(5)
$C60 Cu^2 Sc^2$	122.04(13) 126.88(11)	C63 C64 H64A	177.2 (5)
$Coo-Cu_2 - Se_2$	120.88(11) 110.28(10)	C63 - C64 - H64R	109.5
C_{27} C_{27} C_{27} C_{27}	110.28(10)	H64A C64 H64P	109.5
C_{3}^{-} Se2—Cu2	95.04 (11) 110 8 (5)	C63 $C64$ $H64C$	109.5
$C_{21} = C_{20} = U_{20}$	119.6 (5)		109.5
C35 C30 H30	120.1	H04A - C04 - H04C	109.5
C35-C30-H30	120.1	H04B—C04—H04C	109.5
C6_C1_C2_C3	03(6)	C_{35} C_{30} C_{31} C_{32}	13(8)
$C_1 - C_2 - C_3 - C_4$	-1.5(6)	C_{30} C_{31} C_{32} C_{33}	-66(9)
$C_2 = C_3 = C_4 = C_5$	17(6)	C_{31} C_{32} C_{33} C_{34}	72(11)
$C_{2}^{-} = C_{3}^{-} = C_{4}^{-} = C_{5}^{-} = C_{6}^{-}$	-0.7(6)	C_{32} C_{32} C_{33} C_{34} C_{35}	-23(11)
C_{2}^{-} C_{1}^{-} C_{2}^{-} C_{2	0.7(0)	C_{33} C_{34} C_{35} C_{35} C_{30}	-3.1(9)
$C_2 = C_1 = C_6 = C_7$	176.6(3)	C_{33} C_{34} C_{35} C_{36}	-1795(6)
C_{2}^{-} C_{1}^{-} C_{0}^{-} C_{1}^{-}	-0.5(5)	C_{31} C_{30} C_{35} C_{30}	35(7)
C4 - C5 - C6 - C7	-1766(3)	C_{31} C_{30} C_{35} C_{34}	-180.0(4)
$C_1 - C_6 - C_7 - N_1$	22.7(5)	C_{34} C_{35} C_{36} C	58 9 (6)
$C_{5} - C_{6} - C_{7} - N_{1}$	-1614(3)	C_{30} C_{35} C_{36} N_{3}	-1175(4)
C6-C7-N1-C8	-1127(3)	C_{35} C_{35} C_{36} N_{3} C_{37}	-1102(4)
C6-C7-N1-C9	66.9(4)	C_{35} C_{36} N_{3} C_{38}	75.0(4)
C9-N1-C8-N2	-0.1(4)	C_{38} N_{3} C_{37} N_{4}	(1, 2, 3, 3, 5, 5, 5, 5, 5, 5, 5, 5, 5, 5, 5, 5, 5,
C7-N1-C8-N2	1795(3)	C_{36} N3 C_{37} N4	-1752(3)
$C_{1}^{0} = N_{1}^{1} = C_{2}^{0} = S_{2}^{0}$	179.5(3) 178.8(2)	C_{38} N3 C_{37} Se ²	-178.8(2)
C7 - N1 - C8 - Se1	-1.5(5)	C_{36} N3 C_{37} Se2	5 8 (5)
C_{8} N1_C_0_C_16	(3)	C_{112} S_{e2} C_{27} N_{a}	-720(3)
$C7_N1_C9_C16$	-170 5 (3)	$C_{112} = S_{12} = C_{12} = $	1067(3)
C_{8} N1 C_{9} C10	-1785(3)	$C_{12} = S_{12} = C_{13} = C_{13}$	-0.3(4)
$C7_N1_C9_C10$	170.3(3) 18(5)	$C_{3} = 0.00 -$	175 2 (3)
$C_1 = C_1 = C_2 = C_1 $	63.8(5)	C_{37} N3 C_{38} C30	170.2(3)
$N1_0$	-117 8 (A)	C_{36} N3 C_{38} C_{20}	-55(5)
111 - 02 - 010 - 013	11/.0(4)	UJU-11J-UJ0-UJ7	5.5 (5)

C16—C9—C10—C11	-116.2 (4)	C45—C38—C39—C40	92.5 (5)
N1-C9-C10-C11	62.2 (4)	N3-C38-C39-C40	-86.6 (4)
C15—C10—C11—C12	0.5 (5)	C45—C38—C39—C44	-88.8(5)
C9-C10-C11-C12	-179.5 (3)	N3-C38-C39-C44	92.1 (5)
C10-C11-C12-C13	-0.7 (5)	C44—C39—C40—C41	0.4 (6)
C11—C12—C13—C14	0.1 (5)	C38—C39—C40—C41	179.1 (3)
C12—C13—C14—C15	0.6 (6)	C39—C40—C41—C42	-0.7 (6)
C11—C10—C15—C14	0.2 (5)	C40—C41—C42—C43	0.2 (7)
C9-C10-C15-C14	-179.8 (3)	C41—C42—C43—C44	0.6 (8)
C13—C14—C15—C10	-0.7 (6)	C42—C43—C44—C39	-0.9 (8)
N1-C9-C16-N2	-0.1 (3)	C40—C39—C44—C43	0.4 (7)
C10—C9—C16—N2	178.4 (3)	C38—C39—C44—C43	-178.3 (4)
N1—C9—C16—C17	-178.4(3)	N3—C38—C45—N4	0.3 (4)
C10—C9—C16—C17	0.1 (6)	C39—C38—C45—N4	-178.9 (3)
C9—C16—C17—C22	98.5 (4)	N3—C38—C45—C46	179.5 (3)
N2-C16-C17-C22	-79.5 (4)	C39—C38—C45—C46	0.3 (6)
C9—C16—C17—C18	-80.8(5)	C38—C45—C46—C47	-111.4 (4)
N2-C16-C17-C18	101.2 (4)	N4—C45—C46—C47	67.7 (4)
C22—C17—C18—C19	-0.2(6)	C38—C45—C46—C51	68.9 (5)
C16—C17—C18—C19	179.1 (4)	N4—C45—C46—C51	-112.1 (4)
C17—C18—C19—C20	0.0 (6)	C51—C46—C47—C48	0.5 (5)
C18—C19—C20—C21	0.7 (6)	C45—C46—C47—C48	-179.2(3)
C19—C20—C21—C22	-1.1 (6)	C46—C47—C48—C49	-0.3(5)
C18—C17—C22—C21	-0.1(5)	C47—C48—C49—C50	0.1 (5)
C16—C17—C22—C21	-179.4 (3)	C48—C49—C50—C51	-0.2 (6)
C20—C21—C22—C17	0.8 (5)	C49—C50—C51—C46	0.4 (6)
N1 - C8 - N2 - C16	0.0 (4)	C47—C46—C51—C50	-0.6(5)
Se1—C8—N2—C16	-178.9(2)	C45—C46—C51—C50	179.1 (3)
N1 - C8 - N2 - C23	-176.4(3)	N3-C37-N4-C45	0.0 (4)
Se1-C8-N2-C23	4.6 (5)	Se2-C37-N4-C45	178.9(2)
C9—C16—N2—C8	0.0 (4)	N3—C37—N4—C52	175.7 (3)
C17-C16-N2-C8	178.5 (3)	Se2-C37-N4-C52	-5.3(5)
C9-C16-N2-C23	176.4 (3)	C38—C45—N4—C37	-0.1(4)
C17-C16-N2-C23	-5.1(5)	C46—C45—N4—C37	-179.4(3)
C8 - N2 - C23 - C24	-81.7(4)	C_{38} C_{45} N_{4} C_{52}	-175.9(3)
$C_{16} N_{2} C_{23} C_{24}$	102.4 (4)	C46-C45-N4-C52	4.8 (5)
N_{2} C23 C24 C25	-11.3(5)	C_{37} N4 C_{52} C_{53}	-113.1(4)
N_{2} C_{23} C_{24} C_{29}	166 4 (4)	C45 - N4 - C52 - C53	62.0 (4)
$C_{29} - C_{24} - C_{25} - C_{26}$	-0.9(6)	N4-C52-C53-C58	-1451(3)
C_{23} C_{24} C_{25} C_{20}	1769(4)	N4 - C52 - C53 - C54	36.2 (5)
C_{24} C_{25} C_{26} C_{26} C_{27}	-0.8(7)	C_{58} C_{53} C_{54} C_{55}	11(6)
C_{25} C	2.5(7)	$C_{52} - C_{53} - C_{54} - C_{55}$	179 8 (4)
$C_{26} = C_{27} = C_{28} = C_{29}$	-2.6(8)	C_{53} C_{54} C_{55} C_{56}	-0.3(6)
C_{27} C_{28} C_{29} C_{24}	10(8)	C_{54} C_{55} C_{56} C_{57}	-0.4(7)
C_{25} C_{24} C_{29} C_{28}	0.8(7)	C_{5} C_{5	0.5(7)
C_{23} C_{24} C_{29} C_{28}	-1771(4)	C_{54} C_{53} C_{58} C_{57}	-10(6)
	· / / · · · · · / · /		1.0 (0)

N2-C8-Se1-Cu1	111.9 (3)	C52—C53—C58—C57	-179.8 (4)
N1—C8—Se1—Cu1	-66.9 (3)	C56—C57—C58—C53	0.3 (7)

Symmetry codes: (i) -x+2, y+1/2, -z+3/2; (ii) -x+2, y-1/2, -z+3/2.