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# Crystal structure of dimethyl 4,4'-dimethoxy-biphenyl-3,3'-dicarboxylate 

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In the title compound, $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{6}$, the benzene rings are coplanar due to the centrosymmetric nature of the molecule, with an inversion centre located at the midpoint of the $\mathrm{C}-\mathrm{C}$ bond between the two rings. Consequently, the methyl carboxylate substituents are oriented in a trans fashion with regards to the bond between the benzene rings. The methyl carboxylate and methoxy substituents are rotated slightly out of plane relative to their parent benzene rings, with dihedral and torsion angles of $18.52(8)$ and $-5.22(15)^{\circ}$, respectively. The shortest $\mathrm{O} \cdots \mathrm{H}$ contact between neighbouring molecules is about $2.5 \AA$. Although some structure-directing contributions from $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogenbonding interactions are possible, the crystal packing seems primarily directed by weak van der Waals forces.

## 1. Chemical context

The title compound is an intermediate in the synthesis of $4,4^{\prime}$ -dimethoxybiphenyl-3, $3^{\prime}$-biphenyldicarboxylic acid, an organic linker for use in the synthesis of coordination polymers (Lundvall et al., 2016). The title compound, $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{6}$, has previously been reported (Wang et al., 2009; Kar et al., 2009), however, its crystal structure was undetermined up until now.


## 2. Structural commentary

The asymmetric unit of the title compound comprises one half of the molecule, with an inversion centre located at the midpoint of the benzene-benzene bond (Fig. 1). The benzene rings are coplanar due to symmetry. This is somewhat unexpected since a slight torsion between the two rings is a common feature in biphenyl compounds. The methyl carboxylate substituents are oriented trans relative to the benzene-benzene bond, and the plane of the substituent makes a dihedral angle of $18.52(8)^{\circ}$ relative to the parent benzene ring. The methoxy substituent is nearly coplanar with the parent benzene ring, and a torsion angle

Table 1
Hydrogen-bond geometry ( $\mathrm{A},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| C9-H9C $\cdots \mathrm{O}^{\text {i }}$ | 0.98 | 2.55 | $3.4759(15)$ | 158 |
| ${\text { C } 8-\mathrm{H} 8 B \cdots 3^{\mathrm{ii}}}^{\mathrm{i}}$ | 0.98 | 2.50 | $3.3407(15)$ | 144 |

Symmetry codes: (i) $x, y-1, z$; (ii) $-x, y,-z-\frac{1}{2}$.


Figure 1
The molecular structure of the title compound with atom labels and $50 \%$ probability displacement ellipsoids. Non-labelled atoms are generated by the symmetry code $\left(-x+\frac{1}{2},-y+\frac{1}{2},-z\right)$. H atoms have been omitted for clarity.


Figure 2
Packing diagram of the title compound viewed along the $b$ axis.


Figure 3
Packing diagram of the title compound viewed along the $a$ axis. H atoms have been omitted for clarity.


Figure 4
Graphical representation of the shortest intermolecular $\mathrm{O} \cdots \mathrm{H}$ contacts, illustrated as dashed blue lines.
$\mathrm{C} 5-\mathrm{C} 4-\mathrm{O} 1-\mathrm{C} 9$ of only $-5.22(15)^{\circ}$ is observed. The methyl groups of the methyl carboxylate and methoxy substituents are oriented away from each other to accommodate the steric demands of these groups.

## 3. Supramolecular features

The molecules are packed in the unit cell with the axis of the biphenyl scaffolds parallel to each other. The axis of the biphenyl moiety is oriented approximately $20^{\circ}$ off the $a$ axis of the unit cell (Fig. 2), and the molecules form corrugated layers extending parallel to the ac plane (Fig. 3). The packing is not directed by strong intermolecular bonding since the shortest $\mathrm{O} \cdots \mathrm{H}$ contact is about $2.5 \AA$ (Table 1 ). However, weak C$\mathrm{H} \cdots \mathrm{O}$ interactions between neighbouring molecules seem to have an influence on the crystal packing (Fig. 4).

Table 2
Experimental details.

| Crystal data |  |
| :--- | :--- |
| Chemical formula | $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{6}$ |
| $M_{\mathrm{r}}$ | 330.32 |
| Crystal system, space group | Monoclinic, $C 2 / c$ |
| Temperature $(\mathrm{K})$ | 105 |
| $a, b, c(\AA)$ | $28.5800(14), 4.0632(2)$, |
| $\beta\left(^{\circ}\right)$ | $14.4806(7)$ |
| $V\left(\AA^{3}\right)$ | $115.100(1)$ |
| $Z$ | $1522.78(13)$ |
| Radiation type | 4 |
| $\mu\left(\mathrm{~mm}^{-1}\right)$ | Mo $K \alpha$ |
| Crystal size (mm) | 0.11 |
|  | $0.56 \times 0.29 \times 0.22$ |
| Data collection |  |
| Diffractometer | Bruker PHOTON CCD |
| Absorption correction | Multi-scan $(S A D A B S ;$ Bruker, |
|  | $2007)$ |
| $T_{\text {min }}, T_{\text {max }}$ | $0.602,0.746$ |
| No. of measured, independent and | $17902,2053,1819$ |
| $\quad$ observed $[I>2 \sigma(I)]$ reflections |  |
| $R_{\text {int }}$ | 0.037 |
| $(\text { sin } \theta / \lambda)_{\text {max }}\left(\AA \AA^{-1}\right)$ | 0.685 |
| Refinement |  |
| $R\left[F^{2}>2 \sigma\left(F^{2}\right)\right], w R\left(F^{2}\right), S$ | $0.038,0.109,1.10$ |
| No. of reflections | 2053 |
| No. of parameters | 111 |
| H-atom treatment | H -atom parameters constrained |
| $\Delta \rho_{\text {max }}, \Delta \rho_{\text {min }}\left(\mathrm{e} \AA \AA^{-3}\right)$ | $0.41,-0.24$ |

Computer programs: APEX2 and SAINT (Bruker, 2007), SHELXS97 (Sheldrick, 2008), SHELXL2014 (Sheldrick, 2015), WinGX (Farrugia, 2012), DIAMOND (Brandenburg, 2004), ChemBioDraw (Cambridge Soft, 2009) and publCIF (Westrip, 2010).

## 4. Synthesis and crystallization

The title compound was synthesized by a slightly modified procedure of the method described by Wang et al. (2009). Synthetic details are given in the Supporting Information of our recent contribution (Lundvall et al., 2016). Single crystals suitable for structure determination were obtained by recrystallizing the title compound from chloroform solution.

## 5. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2. H atoms were positioned geometrically at distances of $0.95(\mathrm{CH})$ and $0.98 \AA\left(\mathrm{CH}_{3}\right)$ and were refined using a riding model with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{CH})$ and $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}\left(\mathrm{CH}_{3}\right)$.

## Acknowledgements

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structure for X-ray diffraction and scattering (RECX) and the Department of Chemistry, UiO.

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## supporting information

## Crystal structure of dimethyl 4,4'-dimethoxybiphenyl-3,3'-dicarboxylate

## Fredrik Lundvall, Pascal D. C. Dietzel and Helmer Fjellvåg

## Computing details

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT (Bruker, 2007); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL2014 (Sheldrick, 2015) and WinGX (Farrugia, 2012); molecular graphics: DIAMOND (Brandenburg, 2004) and ChemBioDraw (Cambridge Soft, 2009); software used to prepare material for publication: publCIF (Westrip, 2010).

Dimethyl 4,4'-dimethoxybiphenyl-3,3'-dicarboxylate

## Crystal data

$\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{6}$
$M_{r}=330.32$
Monoclinic, $C 2 / c$
$a=28.5800(14) \AA$
$b=4.0632(2) \AA$
$c=14.4806(7) \AA$
$\beta=115.100(1)^{\circ}$
$V=1522.78(13) \AA^{3}$
$Z=4$

## Data collection

## Bruker PHOTON CCD

diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
$\omega$ scans
Absorption correction: multi-scan
(SADABS; Bruker, 2007)
$T_{\min }=0.602, T_{\text {max }}=0.746$

## Refinement

Refinement on $F^{2}$
Least-squares matrix: full
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.038$
$w R\left(F^{2}\right)=0.109$
$S=1.10$
2053 reflections
111 parameters
0 restraints
Primary atom site location: structure-invariant direct methods

$$
F(000)=696
$$

$D_{\mathrm{x}}=1.441 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation, $\lambda=0.71073 \AA$
Cell parameters from 9942 reflections
$\theta=2.8-29.1^{\circ}$
$\mu=0.11 \mathrm{~mm}^{-1}$
$T=105 \mathrm{~K}$
Needle, colourless
$0.56 \times 0.29 \times 0.22 \mathrm{~mm}$

17902 measured reflections
2053 independent reflections
1819 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.037$
$\theta_{\text {max }}=29.2^{\circ}, \theta_{\text {min }}=2.8^{\circ}$
$h=-38 \rightarrow 38$
$k=-5 \rightarrow 5$
$l=-19 \rightarrow 19$

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{0}^{2}\right)+(0.0534 P)^{2}+1.1903 P\right]$ where $P=\left(F_{0}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.41 \mathrm{e}^{-3}$
$\Delta \rho_{\text {min }}=-0.24 \mathrm{e}^{-3}$

## Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\AA^{2}$ )

|  | $x$ | $y$ | $z$ | $U_{\text {iso }} * / U_{\text {eq }}$ |
| :--- | :--- | :--- | :--- | :--- |
| C1 | $0.22923(4)$ | $0.2306(3)$ | $0.01755(7)$ | $0.0163(2)$ |
| C2 | $0.17986(4)$ | $0.3583(3)$ | $-0.03782(7)$ | $0.0175(2)$ |
| H2 | 0.1730 | 0.4703 | -0.0998 | $0.021^{*}$ |
| C3 | $0.14011(4)$ | $0.3300(3)$ | $-0.00694(7)$ | $0.0170(2)$ |
| C4 | $0.14924(4)$ | $0.1625(3)$ | $0.08424(8)$ | $0.0170(2)$ |
| C5 | $0.19832(4)$ | $0.0299(3)$ | $0.14041(8)$ | $0.0193(2)$ |
| H5 | 0.2052 | -0.0851 | 0.2020 | $0.023^{*}$ |
| C6 | $0.23708(4)$ | $0.0637(3)$ | $0.10753(8)$ | $0.0193(2)$ |
| H6 | 0.2701 | -0.0293 | 0.1473 | $0.023^{*}$ |
| C7 | $0.09039(4)$ | $0.4858(3)$ | $-0.07720(8)$ | $0.0192(2)$ |
| C8 | $0.01027(4)$ | $0.7077(3)$ | $-0.10223(9)$ | $0.0256(3)$ |
| H8A | -0.0111 | 0.7454 | -0.0653 | $0.038^{*}$ |
| H8B | -0.0086 | 0.5685 | -0.1618 | $0.038^{*}$ |
| H8C | 0.0184 | 0.9191 | -0.1246 | $0.038^{*}$ |
| C9 | $0.11816(5)$ | $-0.0599(3)$ | $0.20034(8)$ | $0.0229(2)$ |
| H9A | 0.0868 | -0.0586 | 0.2121 | $0.034^{*}$ |
| H9B | 0.1470 | 0.0309 | 0.2601 | $0.034^{*}$ |
| H9C | 0.1262 | -0.2863 | 0.1888 | $0.034^{*}$ |
| O1 | $0.10999(3)$ | $0.1361(2)$ | $0.11283(6)$ | $0.02064(19)$ |
| O2 | $0.05778(3)$ | $0.5450(2)$ | $-0.03563(6)$ | $0.0232(2)$ |
| O3 | $0.08127(3)$ | $0.5572(3)$ | $-0.16406(7)$ | $0.0354(3)$ |

Atomic displacement parameters $\left(\AA^{2}\right)$

|  | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| C1 | $0.0173(5)$ | $0.0161(5)$ | $0.0141(4)$ | $-0.0009(4)$ | $0.0053(4)$ | $-0.0023(4)$ |
| C2 | $0.0182(5)$ | $0.0194(5)$ | $0.0133(4)$ | $-0.0012(4)$ | $0.0051(4)$ | $-0.0004(4)$ |
| C3 | $0.0162(4)$ | $0.0181(5)$ | $0.0142(4)$ | $-0.0010(4)$ | $0.0040(4)$ | $-0.0018(4)$ |
| C4 | $0.0187(5)$ | $0.0168(5)$ | $0.0156(4)$ | $-0.0020(4)$ | $0.0072(4)$ | $-0.0027(4)$ |
| C5 | $0.0216(5)$ | $0.0201(5)$ | $0.0153(4)$ | $0.0013(4)$ | $0.0071(4)$ | $0.0023(4)$ |
| C6 | $0.0186(5)$ | $0.0208(5)$ | $0.0164(5)$ | $0.0025(4)$ | $0.0054(4)$ | $0.0008(4)$ |
| C7 | $0.0165(5)$ | $0.0223(5)$ | $0.0170(5)$ | $-0.0015(4)$ | $0.0055(4)$ | $0.0001(4)$ |
| C8 | $0.0185(5)$ | $0.0324(6)$ | $0.0240(5)$ | $0.0064(5)$ | $0.0073(4)$ | $0.0036(5)$ |
| C9 | $0.0288(6)$ | $0.0234(6)$ | $0.0203(5)$ | $0.0022(4)$ | $0.0141(4)$ | $0.0036(4)$ |
| O1 | $0.0207(4)$ | $0.0247(4)$ | $0.0184(4)$ | $0.0013(3)$ | $0.0101(3)$ | $0.0038(3)$ |
| O2 | $0.0193(4)$ | $0.0323(5)$ | $0.0172(4)$ | $0.0060(3)$ | $0.0068(3)$ | $0.0015(3)$ |
| O3 | $0.0223(4)$ | $0.0622(7)$ | $0.0219(4)$ | $0.0115(4)$ | $0.0096(3)$ | $0.0158(4)$ |

Geometric parameters (A, ${ }^{\circ}$ )

| C1-C2 | 1.3940 (14) | C6-H6 | 0.9500 |
| :---: | :---: | :---: | :---: |
| C1-C6 | 1.3998 (14) | C7-03 | 1.2067 (14) |
| C1- $\mathrm{Cl}^{\text {i }}$ | 1.4846 (19) | C7-02 | 1.3285 (13) |
| C2-C3 | 1.3909 (14) | C8-O2 | 1.4490 (13) |
| C2-H2 | 0.9500 | C8-H8A | 0.9800 |
| C3-C4 | 1.4080 (14) | C8-H8B | 0.9800 |
| C3-C7 | 1.4928 (14) | C8-H8C | 0.9800 |
| C4-O1 | 1.3549 (12) | C9-O1 | 1.4293 (13) |
| C4-C5 | 1.3971 (15) | C9-H9A | 0.9800 |
| C5-C6 | 1.3859 (15) | C9-H9B | 0.9800 |
| C5-H5 | 0.9500 | C9-H9C | 0.9800 |
| C2- $\mathrm{C} 1-\mathrm{C} 6$ | 116.04 (9) | O3-C7-O2 | 123.09 (10) |
| C2-C1-C1 ${ }^{\text {i }}$ | 121.65 (11) | $\mathrm{O} 3-\mathrm{C} 7-\mathrm{C} 3$ | 122.32 (10) |
| C6-C1-C1 ${ }^{\text {i }}$ | 122.31 (11) | $\mathrm{O} 2-\mathrm{C} 7-\mathrm{C} 3$ | 114.58 (9) |
| C3-C2-C1 | 123.36 (9) | $\mathrm{O} 2-\mathrm{C} 8-\mathrm{H} 8 \mathrm{~A}$ | 109.5 |
| C3-C2-H2 | 118.3 | O2-C8-H8B | 109.5 |
| C1-C2-H2 | 118.3 | H8A-C8-H8B | 109.5 |
| C2-C3-C4 | 119.32 (9) | $\mathrm{O} 2-\mathrm{C} 8-\mathrm{H} 8 \mathrm{C}$ | 109.5 |
| C2-C3-C7 | 114.66 (9) | H8A-C8-H8C | 109.5 |
| C4-C3-C7 | 126.03 (9) | H8B-C8-H8C | 109.5 |
| $\mathrm{O} 1-\mathrm{C} 4-\mathrm{C} 5$ | 123.27 (9) | O1-C9-H9A | 109.5 |
| $\mathrm{O} 1-\mathrm{C} 4-\mathrm{C} 3$ | 118.48 (9) | O1-C9- H 9 B | 109.5 |
| C5-C4-C3 | 118.25 (10) | H9A-C9-H9B | 109.5 |
| C6-C5-C4 | 120.87 (10) | O1-C9- H 9 C | 109.5 |
| C6-C5-H5 | 119.6 | H9A-C9-H9C | 109.5 |
| C4-C5-H5 | 119.6 | H9B- $\mathrm{C} 9-\mathrm{H} 9 \mathrm{C}$ | 109.5 |
| C5-C6-C1 | 122.15 (10) | C4-O1-C9 | 118.03 (8) |
| C5-C6-H6 | 118.9 | C7-O2-C8 | 114.87 (9) |
| C1-C6-H6 | 118.9 |  |  |
| C6-C1-C2-C3 | 0.89 (16) | C2-C1-C6-C5 | -0.64 (16) |
| $\mathrm{C} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | -179.29 (11) | C1- ${ }^{\text {i }} 10-\mathrm{C} 6-\mathrm{C} 5$ | 179.55 (12) |
| C1-C2-C3-C4 | -0.55 (16) | C2-C3-C7-O3 | 17.33 (16) |
| C1-C2-C3-C7 | 179.71 (10) | $\mathrm{C} 4-\mathrm{C} 3-\mathrm{C} 7-\mathrm{O} 3$ | -162.39 (12) |
| $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{O} 1$ | -179.68 (9) | C2-C3-C7-O2 | -161.30 (10) |
| C7-C3-C4-O1 | 0.03 (16) | $\mathrm{C} 4-\mathrm{C} 3-\mathrm{C} 7-\mathrm{O} 2$ | 18.97 (16) |
| C2-C3-C4-C5 | -0.08 (15) | C5-C4-O1-C9 | -5.22 (15) |
| C7-C3-C4-C5 | 179.63 (10) | C3-C4-O1-C9 | 174.36 (10) |
| O1-C4-C5-C6 | 179.90 (10) | $\mathrm{O} 3-\mathrm{C} 7-\mathrm{O} 2-\mathrm{C} 8$ | -0.89 (17) |
| C3-C4-C5-C6 | 0.32 (16) | C3-C7-O2-C8 | 177.74 (10) |
| C4-C5-C6-C1 | 0.05 (17) | C2-C1-C1 ${ }^{\text {i }}$ - $6^{\text {i }}$ | 0.2 (2) |

[^0]
## supporting information

Hydrogen-bond geometry (A, ${ }^{\circ}$ )

| $D — \mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| C9—H9C $\cdots \mathrm{O} 1^{\mathrm{ii}}$ | 0.98 | 2.55 | $3.4759(15)$ | 158 |
| C8—H8B $\cdots 3^{\text {iii }}$ | 0.98 | 2.50 | $3.3407(15)$ | 144 |

Symmetry codes: (ii) $x, y-1, z$; (iii) $-x, y,-z-1 / 2$.


[^0]:    Symmetry code: (i) $-x+1 / 2,-y+1 / 2,-z$.

