



# Crystal structure of 1-hydroxy-2,2,6,6-tetramethylpiperidin-1-ium trifluoromethanesulfonate

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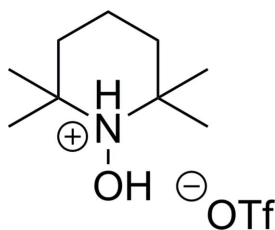
In the cation of the title salt,  $C_9H_{20}NO^+ \cdot CF_3O_3S^-$ , the six-membered heterocyclic ring displays a chair conformation. In the crystal, centrosymmetric pairs of cations and anions are linked by  $N-H \cdots O$  and  $O-H \cdots O$  hydrogen bonds to form rings with a  $R_4^4(14)$  graph-set motif.

**Keywords:** crystal structure; TEMPO; ammonium salt; triflate; hydrogen bonding.

**CCDC reference:** 1435030

## 1. Related literature

For molecular structures and discussions of related compounds, see: Jaitner & Wurst (1997); Spirk *et al.* (2010); Ananchenko *et al.* (2006); Percino *et al.* (2016). For the molecular structure of the neutral TEMPO-H compound, see: Mader *et al.* (2007); Giffin *et al.* (2011).



## 2. Experimental

### 2.1. Crystal data

$C_9H_{20}NO^+ \cdot CF_3O_3S^-$   
 $M_r = 307.33$   
 Triclinic,  $P\bar{1}$

$a = 8.2824$  (2) Å  
 $b = 8.7656$  (2) Å  
 $c = 10.5703$  (3) Å

$\alpha = 79.5417$  (7)°  
 $\beta = 76.5159$  (7)°  
 $\gamma = 75.5022$  (6)°  
 $V = 716.28$  (3) Å<sup>3</sup>  
 $Z = 2$

Mo  $K\alpha$  radiation  
 $\mu = 0.27$  mm<sup>-1</sup>  
 $T = 150$  K  
 $0.55 \times 0.38 \times 0.34$  mm

### 2.2. Data collection

Bruker APEXII CCD diffractometer  
 Absorption correction: multi-scan (SADABS; Bruker, 2014)  
 $T_{\min} = 0.83$ ,  $T_{\max} = 0.86$

22933 measured reflections  
 3452 independent reflections  
 3039 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$

### 2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$   
 $wR(F^2) = 0.094$   
 $S = 1.05$   
 3452 reflections  
 184 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.32$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.28$  e Å<sup>-3</sup>

**Table 1**  
 Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$O1-H1A \cdots O4$	0.84 (2)	1.78 (2)	2.6163 (14)	177 (2)
$N1-H1B \cdots O3^i$	0.875 (16)	1.991 (16)	2.8385 (14)	163.0 (14)

Symmetry code: (i)  $-x + 1, -y + 2, -z + 1$ .

Data collection: APEX2 (Bruker, 2014); cell refinement: SAINT (Bruker, 2013); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL2014 (Sheldrick, 2015); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL2014.

## Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: RZ5176).

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## supporting information

*Acta Cryst.* (2015). E71, o921 [https://doi.org/10.1107/S2056989015020897]

## Crystal structure of 1-hydroxy-2,2,6,6-tetramethylpiperidin-1-ium trifluoromethanesulfonate

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### S1. Synthesis and crystallization

An equimolar mixture of the titanocene(IV) triflate complex  $[(\text{SiMe}_2\text{C}_5\text{Me}_4)_2\text{Ti}(\text{H}_2\text{O})(\text{OH})(\text{OTf})]$  (Godemann & Beweries, unpublished results) and 2,2,6,6-tetramethyl-1-hydroxypiperidine (TEMPO-H) in toluene was cooled to  $-78^\circ\text{C}$ . After two weeks, the formation of colourless crystals of the title compound could be observed on slow evaporation of the solvent. Alternatively, layering a toluene solution of the same titanocene compound and TEMPO-H with *n*-hexane also resulted in the formation of colourless crystals of the title compound.

### S2. Refinement

The H1A and H1B atoms were found from a difference Fourier map and refined freely. All other H atoms were placed in idealized positions with  $d(\text{C}-\text{H}) = 0.99 \text{ \AA}$  ( $\text{CH}_2$ ),  $0.98 \text{ \AA}$  ( $\text{CH}_3$ ) and refined using a riding model with  $U_{\text{iso}}(\text{H})$  fixed at  $1.2 U_{\text{eq}}(\text{C})$  for  $\text{CH}_2$  and  $1.5 U_{\text{eq}}(\text{C})$  for  $\text{CH}_3$ . A rotating model was used for the methyl groups.

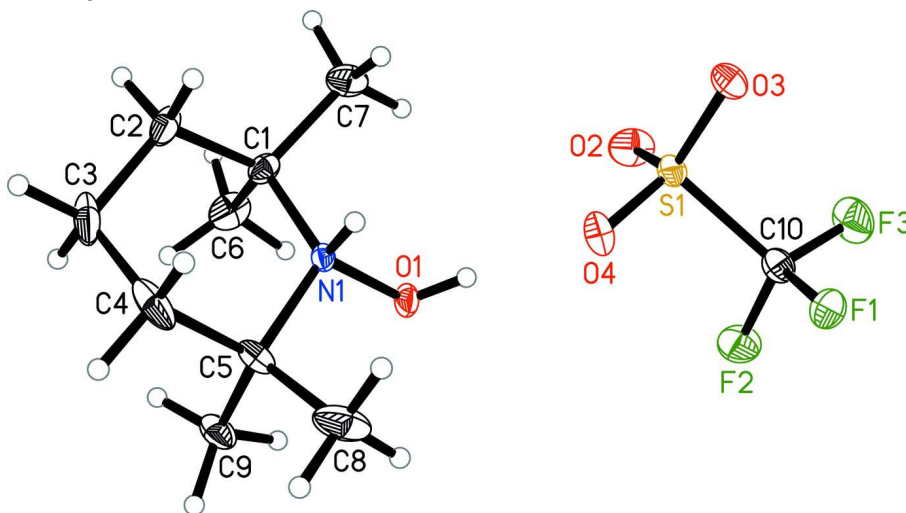


Figure 1

The molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level.

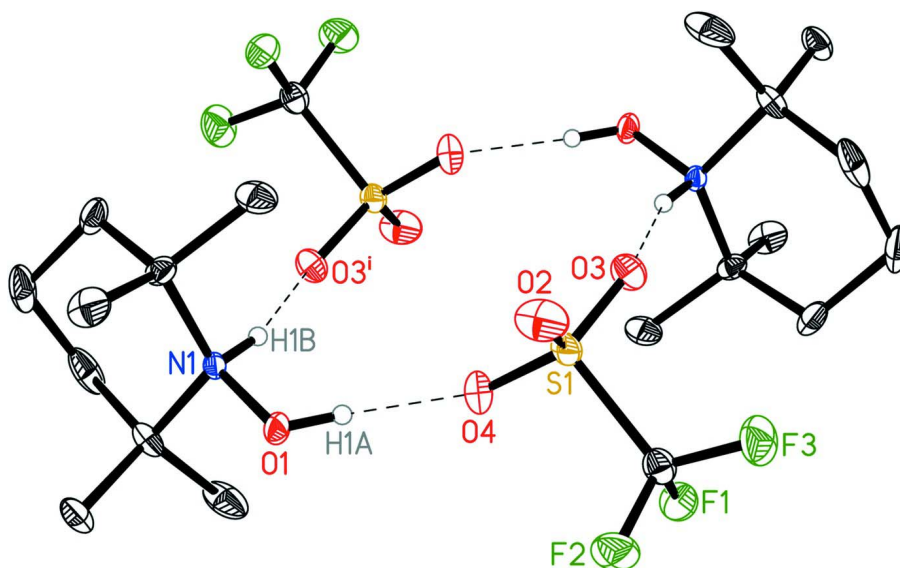


Figure 2

The hydrogen-bonding network (dashed lines) linking centrosymmetric pairs of cations and anions in the title compound. C-bound hydrogen atoms are omitted for clarity. Displacement ellipsoids are drawn at the 30% probability level.

### 1-Hydroxy-2,2,6,6-tetramethylpiperidin-1-ium trifluoromethanesulfonate

#### Crystal data

$C_9H_{20}NO^+ \cdot CF_3O_3S^-$

$M_r = 307.33$

Triclinic,  $P\bar{1}$

$a = 8.2824(2) \text{ \AA}$

$b = 8.7656(2) \text{ \AA}$

$c = 10.5703(3) \text{ \AA}$

$\alpha = 79.5417(7)^\circ$

$\beta = 76.5159(7)^\circ$

$\gamma = 75.5022(6)^\circ$

$V = 716.28(3) \text{ \AA}^3$

$Z = 2$

$F(000) = 324$

$D_x = 1.425 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 9968 reflections

$\theta = 2.6\text{--}28.9^\circ$

$\mu = 0.27 \text{ mm}^{-1}$

$T = 150 \text{ K}$

Prism, colourless

$0.55 \times 0.38 \times 0.34 \text{ mm}$

#### Data collection

Bruker APEXII CCD  
diffractometer

Radiation source: fine-focus sealed tube

Detector resolution:  $8.3333 \text{ pixels mm}^{-1}$

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2014)

$T_{\min} = 0.83$ ,  $T_{\max} = 0.86$

22933 measured reflections

3452 independent reflections

3039 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.020$

$\theta_{\max} = 28.0^\circ$ ,  $\theta_{\min} = 2.0^\circ$

$h = -10 \rightarrow 10$

$k = -11 \rightarrow 11$

$l = -13 \rightarrow 13$

#### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.034$

$wR(F^2) = 0.094$

$S = 1.05$

3452 reflections

184 parameters

0 restraints

Hydrogen site location: mixed

H atoms treated by a mixture of independent  
and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0476P)^2 + 0.2597P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$

$$\Delta\rho_{\max} = 0.32 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.28 \text{ e } \text{\AA}^{-3}$$

### Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.96872 (15)	0.80108 (15)	0.28534 (12)	0.0251 (3)
C2	0.96590 (19)	0.86192 (17)	0.14106 (13)	0.0371 (3)
H2A	0.8640	0.9487	0.1352	0.045*
H2B	1.0673	0.9072	0.1019	0.045*
C3	0.9637 (2)	0.7356 (2)	0.06137 (14)	0.0503 (5)
H3A	1.0691	0.6517	0.0611	0.060*
H3B	0.9595	0.7834	-0.0305	0.060*
C4	0.8101 (2)	0.66311 (19)	0.11958 (15)	0.0456 (4)
H4A	0.8100	0.5820	0.0655	0.055*
H4B	0.7057	0.7472	0.1151	0.055*
C5	0.80421 (17)	0.58576 (15)	0.26184 (13)	0.0287 (3)
C6	1.13968 (18)	0.69450 (19)	0.30494 (17)	0.0396 (3)
H6A	1.2234	0.7600	0.2937	0.059*
H6B	1.1782	0.6194	0.2402	0.059*
H6C	1.1273	0.6356	0.3936	0.059*
C7	0.92379 (19)	0.93950 (17)	0.36579 (16)	0.0360 (3)
H7A	0.9328	0.8983	0.4571	0.054*
H7B	0.8072	0.9985	0.3625	0.054*
H7C	1.0026	1.0105	0.3294	0.054*
C8	0.6305 (2)	0.5497 (2)	0.3250 (2)	0.0511 (5)
H8A	0.6018	0.4827	0.2724	0.077*
H8B	0.5446	0.6494	0.3293	0.077*
H8C	0.6334	0.4938	0.4138	0.077*
C9	0.94213 (19)	0.43502 (16)	0.27559 (15)	0.0343 (3)
H9A	0.9560	0.4085	0.3671	0.051*
H9B	1.0498	0.4523	0.2192	0.051*
H9C	0.9093	0.3473	0.2493	0.051*
C10	0.48030 (17)	0.77783 (16)	0.86989 (13)	0.0301 (3)
F1	0.32733 (11)	0.76163 (11)	0.86237 (9)	0.0403 (2)
F2	0.57731 (13)	0.63247 (11)	0.88758 (10)	0.0488 (3)
F3	0.46049 (15)	0.84651 (13)	0.97606 (9)	0.0522 (3)
N1	0.82282 (12)	0.71302 (11)	0.33670 (9)	0.0180 (2)
O1	0.82554 (13)	0.64554 (11)	0.46877 (8)	0.0285 (2)
O2	0.74038 (13)	0.88719 (14)	0.74267 (14)	0.0495 (3)
O3	0.45998 (13)	1.04716 (11)	0.72367 (10)	0.0352 (2)
O4	0.56749 (14)	0.80767 (14)	0.61969 (10)	0.0425 (3)

S1	0.57433 (4)	0.89340 (4)	0.72262 (3)	0.02752 (10)
H1A	0.745 (3)	0.700 (2)	0.516 (2)	0.050 (5)*
H1B	0.730 (2)	0.7873 (18)	0.3356 (15)	0.025 (4)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0187 (6)	0.0249 (6)	0.0285 (6)	-0.0058 (4)	-0.0015 (5)	0.0019 (5)
C2	0.0327 (7)	0.0331 (7)	0.0270 (6)	0.0054 (6)	0.0074 (5)	0.0086 (5)
C3	0.0644 (11)	0.0464 (9)	0.0167 (6)	0.0215 (8)	0.0011 (6)	-0.0030 (6)
C4	0.0617 (10)	0.0393 (8)	0.0359 (8)	0.0188 (7)	-0.0286 (7)	-0.0221 (6)
C5	0.0290 (6)	0.0228 (6)	0.0368 (7)	0.0021 (5)	-0.0114 (5)	-0.0145 (5)
C6	0.0203 (6)	0.0402 (8)	0.0531 (9)	-0.0038 (6)	-0.0086 (6)	0.0050 (7)
C7	0.0354 (7)	0.0303 (7)	0.0484 (8)	-0.0133 (6)	-0.0121 (6)	-0.0068 (6)
C8	0.0323 (8)	0.0330 (8)	0.0957 (15)	-0.0088 (6)	-0.0148 (8)	-0.0230 (8)
C9	0.0393 (8)	0.0223 (6)	0.0381 (7)	0.0055 (5)	-0.0090 (6)	-0.0112 (5)
C10	0.0286 (7)	0.0293 (6)	0.0315 (7)	-0.0007 (5)	-0.0078 (5)	-0.0068 (5)
F1	0.0280 (4)	0.0431 (5)	0.0489 (5)	-0.0120 (4)	0.0008 (4)	-0.0084 (4)
F2	0.0473 (6)	0.0309 (5)	0.0588 (6)	0.0034 (4)	-0.0127 (4)	0.0043 (4)
F3	0.0693 (7)	0.0580 (6)	0.0319 (5)	-0.0098 (5)	-0.0128 (4)	-0.0148 (4)
N1	0.0177 (5)	0.0180 (4)	0.0160 (4)	-0.0009 (4)	-0.0013 (3)	-0.0031 (3)
O1	0.0368 (5)	0.0252 (4)	0.0151 (4)	0.0002 (4)	0.0018 (4)	-0.0004 (3)
O2	0.0206 (5)	0.0412 (6)	0.0850 (9)	-0.0062 (4)	-0.0104 (5)	-0.0052 (6)
O3	0.0290 (5)	0.0264 (5)	0.0432 (6)	0.0058 (4)	-0.0055 (4)	-0.0057 (4)
O4	0.0435 (6)	0.0459 (6)	0.0332 (5)	-0.0042 (5)	0.0055 (4)	-0.0172 (5)
S1	0.01852 (16)	0.02389 (17)	0.03577 (18)	0.00056 (11)	-0.00008 (12)	-0.00683 (12)

*Geometric parameters (Å, °)*

C1—C6	1.5247 (18)	C7—H7A	0.9800
C1—C2	1.5251 (18)	C7—H7B	0.9800
C1—C7	1.5279 (19)	C7—H7C	0.9800
C1—N1	1.5362 (15)	C8—H8A	0.9800
C2—C3	1.514 (2)	C8—H8B	0.9800
C2—H2A	0.9900	C8—H8C	0.9800
C2—H2B	0.9900	C9—H9A	0.9800
C3—C4	1.516 (3)	C9—H9B	0.9800
C3—H3A	0.9900	C9—H9C	0.9800
C3—H3B	0.9900	C10—F3	1.3262 (16)
C4—C5	1.528 (2)	C10—F1	1.3314 (16)
C4—H4A	0.9900	C10—F2	1.3323 (16)
C4—H4B	0.9900	C10—S1	1.8202 (15)
C5—C8	1.522 (2)	N1—O1	1.4168 (12)
C5—C9	1.5244 (17)	N1—H1B	0.875 (16)
C5—N1	1.5354 (15)	O1—H1A	0.84 (2)
C6—H6A	0.9800	O2—S1	1.4260 (11)
C6—H6B	0.9800	O3—S1	1.4406 (9)
C6—H6C	0.9800	O4—S1	1.4486 (11)

C6—C1—C2	112.20 (11)	C1—C7—H7B	109.5
C6—C1—C7	110.16 (12)	H7A—C7—H7B	109.5
C2—C1—C7	110.70 (11)	C1—C7—H7C	109.5
C6—C1—N1	111.77 (10)	H7A—C7—H7C	109.5
C2—C1—N1	106.75 (11)	H7B—C7—H7C	109.5
C7—C1—N1	104.98 (10)	C5—C8—H8A	109.5
C3—C2—C1	113.88 (12)	C5—C8—H8B	109.5
C3—C2—H2A	108.8	H8A—C8—H8B	109.5
C1—C2—H2A	108.8	C5—C8—H8C	109.5
C3—C2—H2B	108.8	H8A—C8—H8C	109.5
C1—C2—H2B	108.8	H8B—C8—H8C	109.5
H2A—C2—H2B	107.7	C5—C9—H9A	109.5
C2—C3—C4	109.99 (11)	C5—C9—H9B	109.5
C2—C3—H3A	109.7	H9A—C9—H9B	109.5
C4—C3—H3A	109.7	C5—C9—H9C	109.5
C2—C3—H3B	109.7	H9A—C9—H9C	109.5
C4—C3—H3B	109.7	H9B—C9—H9C	109.5
H3A—C3—H3B	108.2	F3—C10—F1	107.06 (11)
C3—C4—C5	113.93 (13)	F3—C10—F2	108.19 (12)
C3—C4—H4A	108.8	F1—C10—F2	107.48 (11)
C5—C4—H4A	108.8	F3—C10—S1	111.63 (10)
C3—C4—H4B	108.8	F1—C10—S1	111.42 (9)
C5—C4—H4B	108.8	F2—C10—S1	110.87 (10)
H4A—C4—H4B	107.7	O1—N1—C5	108.32 (9)
C8—C5—C9	109.71 (12)	O1—N1—C1	109.24 (9)
C8—C5—C4	111.57 (14)	C5—N1—C1	120.28 (9)
C9—C5—C4	112.76 (11)	O1—N1—H1B	108.1 (10)
C8—C5—N1	105.18 (11)	C5—N1—H1B	105.5 (10)
C9—C5—N1	111.54 (10)	C1—N1—H1B	104.8 (10)
C4—C5—N1	105.78 (11)	N1—O1—H1A	107.7 (13)
C1—C6—H6A	109.5	O2—S1—O3	115.62 (7)
C1—C6—H6B	109.5	O2—S1—O4	115.69 (7)
H6A—C6—H6B	109.5	O3—S1—O4	113.73 (7)
C1—C6—H6C	109.5	O2—S1—C10	103.77 (7)
H6A—C6—H6C	109.5	O3—S1—C10	103.23 (6)
H6B—C6—H6C	109.5	O4—S1—C10	102.32 (7)
C1—C7—H7A	109.5		
C6—C1—C2—C3	71.46 (16)	C2—C1—N1—O1	176.56 (9)
C7—C1—C2—C3	-165.02 (13)	C7—C1—N1—O1	-65.88 (12)
N1—C1—C2—C3	-51.30 (15)	C6—C1—N1—C5	-72.58 (14)
C1—C2—C3—C4	58.13 (16)	C2—C1—N1—C5	50.45 (13)
C2—C3—C4—C5	-59.31 (16)	C7—C1—N1—C5	168.01 (10)
C3—C4—C5—C8	166.79 (12)	F3—C10—S1—O2	65.11 (11)
C3—C4—C5—C9	-69.21 (15)	F1—C10—S1—O2	-175.25 (9)
C3—C4—C5—N1	52.95 (14)	F2—C10—S1—O2	-55.59 (12)
C8—C5—N1—O1	64.27 (13)	F3—C10—S1—O3	-55.88 (11)

C9—C5—N1—O1	-54.59 (13)	F1—C10—S1—O3	63.76 (10)
C4—C5—N1—O1	-177.53 (10)	F2—C10—S1—O3	-176.58 (10)
C8—C5—N1—C1	-169.19 (12)	F3—C10—S1—O4	-174.21 (10)
C9—C5—N1—C1	71.95 (14)	F1—C10—S1—O4	-54.57 (11)
C4—C5—N1—C1	-50.99 (14)	F2—C10—S1—O4	65.09 (11)
C6—C1—N1—O1	53.53 (13)		

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O1—H1A $\cdots$ O4	0.84 (2)	1.78 (2)	2.6163 (14)	177 (2)
N1—H1B $\cdots$ O3 <sup>i</sup>	0.875 (16)	1.991 (16)	2.8385 (14)	163.0 (14)

Symmetry code: (i)  $-x+1, -y+2, -z+1$ .