



# Crystal structure of (1*E*,1'*E*)-*N,N'*-(ethane-1,2-diyl)bis[(pyridin-2-yl)-methanimine]

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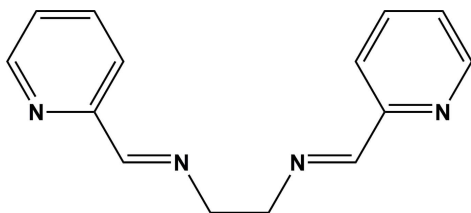
The whole molecule of the title compound, C<sub>14</sub>H<sub>14</sub>N<sub>4</sub>, is generated by twofold rotation symmetry. The twofold axis bisects the central –CH<sub>2</sub>–CH<sub>2</sub>– bond and the planes of the pyridine rings are inclined to one another by 65.60 (7)°. In the crystal, there are no significant intermolecular interactions present.

**Keywords:** crystal structure; pyridinecarbaldehydes; 1,2-diaminopyridine; Schiff base; chelating ligands.

**CCDC reference:** 1402701

## 1. Related literature

For the use of Schiff bases, derived from pyridine-carbaldehydes, in synthetic chemistry, see: Marjani *et al.* (2009). For 1,2-diaminopyridine-derived Schiff bases as bidentate or polydentate chelating ligands and their possible medical applications, see: Warad *et al.* (2014).



## 2. Experimental

### 2.1. Crystal data

|  |                                      |
|--|--------------------------------------|
| C <sub>14</sub> H <sub>14</sub> N <sub>4</sub> | <i>V</i> = 1278.0 (5) Å <sup>3</sup> |
| <i>M<sub>r</sub></i> = 238.29                  | <i>Z</i> = 4                         |
| Monoclinic, <i>C</i> 2/ <i>c</i>               | Cu <i>K</i> α radiation              |
| <i>a</i> = 19.347 (5) Å                        | <i>μ</i> = 0.61 mm <sup>-1</sup>     |
| <i>b</i> = 5.9339 (12) Å                       | <i>T</i> = 296 K                     |
| <i>c</i> = 13.165 (2) Å                        | 0.30 × 0.27 × 0.25 mm                |
| <i>β</i> = 122.266 (8)°                        |                                      |

### 2.2. Data collection

|  |  |
|--|--|
| Bruker X8 Proteum diffractometer                                     | 1539 measured reflections                      |
| Absorption correction: multi-scan<br>( <i>SADABS</i> ; Bruker, 2013) | 933 independent reflections                    |
| <i>T<sub>min</sub></i> = 0.837, <i>T<sub>max</sub></i> = 0.862       | 881 reflections with <i>I</i> > 2σ( <i>I</i> ) |
|  | <i>R<sub>int</sub></i> = 0.015                 |

### 2.3. Refinement

|   |   |
|---|---|
| <i>R</i> [ <i>F</i> <sup>2</sup> > 2σ( <i>F</i> <sup>2</sup> )] = 0.043 | 82 parameters                                       |
| <i>wR</i> ( <i>F</i> <sup>2</sup> ) = 0.120                             | H-atom parameters constrained                       |
| <i>S</i> = 1.05   | Δ <i>ρ</i> <sub>max</sub> = 0.10 e Å <sup>-3</sup>  |
| 933 reflections   | Δ <i>ρ</i> <sub>min</sub> = -0.10 e Å <sup>-3</sup> |

Data collection: *APEX2* (Bruker, 2013); cell refinement: *SAINT* (Bruker, 2013); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *Mercury* (Macrae *et al.*, 2008); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2009).

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Supporting information for this paper is available from the IUCr electronic archives (Reference: SU5142).

## References

- Bruker (2013). *APEX2*, *SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Macrae, C. F., Bruno, I. J., Chisholm, J. A., Edgington, P. R., McCabe, P., Pidcock, E., Rodriguez-Monge, L., Taylor, R., van de Streek, J. & Wood, P. A. (2008). *J. Appl. Cryst.* **41**, 466–470.
- Marjani, K., Asgarian, J., Mousavi, M. & Amani, V. (2009). *Z. Anorg. Allg. Chem.* **635**, 1633–1637.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Warad, I., Khan, A., Azam, M., Al-Resayes, S. I. & Haddad, S. (2014). *J. Mol. Struct.* **1062**, 167–173.

## supporting information

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### S1. Structural commentary

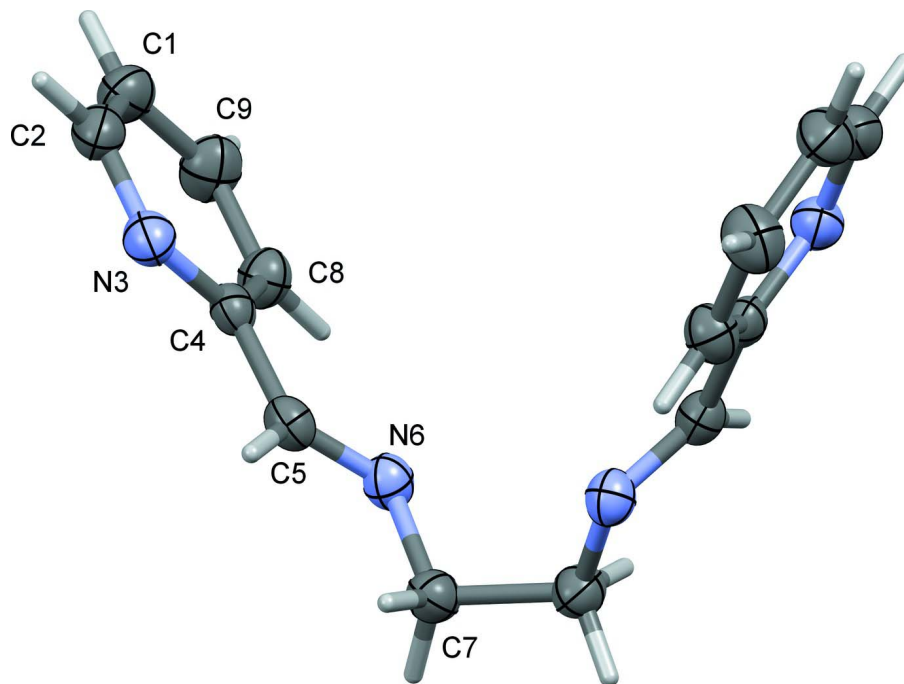
Schiff bases derived from pyridinecarbaldehydes have received considerable interest in synthetic chemistry (Marjani *et al.*, 2009). 1,2-diamine-pyridine derived Schiff base bidentate or polydentate chelating ligand towards metal centers draw major attraction towards synthesis and medical application (Warad *et al.*, 2014). It is still challenging to design and rationally synthesize ligands with unique structures and functions.

### S2. Synthesis and crystallization

To a solution of pyridine-2-carbaldehyde (1 mmol) dissolved in 10 ml of absolute ethanol was added drop wise ethane-1,2-diamine (1 mmol) in 5 ml of absolute ethanol under constant stirring for 10 min. The mixture was refluxed for 4 h and then concentrated under reduced pressure. The title compound was precipitated by the addition of 50 ml of n-hexane. It was filtered off, washed three times with 80 ml of distilled water then with diethyl ether to give the title compound (yield: 86%). Single crystals suitable for X-ray analysis were obtained within two days by slow evaporation of a solution in dichloromethane.

### S3. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 1. The H atoms were fixed geometrically (C—H = 0.93 – 0.97 Å) and allowed to ride on their parent atoms with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

**Figure 1**

View of the molecular structure of the title compound, with atom labelling. Displacement ellipsoids are drawn at the 50% probability level. Unlabelled atoms are related to the labelled atoms by twofold rotation symmetry (symmetry code:  $-x + 1, y, -z - 1/2$ ).

**(1*E*,1'*E*)-*N,N'*-(Ethane-1,2-diyl)bis[(pyridin-2-yl)methanimine]**

*Crystal data*

$C_{14}H_{14}N_4$

$M_r = 238.29$

Monoclinic,  $C2/c$

Hall symbol:  $-C 2yc$

$a = 19.347 (5) \text{ \AA}$

$b = 5.9339 (12) \text{ \AA}$

$c = 13.165 (2) \text{ \AA}$

$\beta = 122.266 (8)^\circ$

$V = 1278.0 (5) \text{ \AA}^3$

$Z = 4$

$F(000) = 504$

$D_x = 1.238 \text{ Mg m}^{-3}$

Cu  $K\alpha$  radiation,  $\lambda = 1.54178 \text{ \AA}$

Cell parameters from 881 reflections

$\theta = 5.4\text{--}63.8^\circ$

$\mu = 0.61 \text{ mm}^{-1}$

$T = 296 \text{ K}$

Block, colourless

$0.30 \times 0.27 \times 0.25 \text{ mm}$

*Data collection*

Bruker X8 Proteum  
diffractometer

Radiation source: Rotating Anode

Graphite monochromator

Detector resolution:  $18.4 \text{ pixels mm}^{-1}$

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2013)

$T_{\min} = 0.837, T_{\max} = 0.862$

1539 measured reflections

933 independent reflections

881 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.015$

$\theta_{\max} = 63.8^\circ, \theta_{\min} = 5.4^\circ$

$h = -8 \rightarrow 21$

$k = -6 \rightarrow 5$

$l = -15 \rightarrow 12$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.043$   
 $wR(F^2) = 0.120$   
 $S = 1.05$   
 933 reflections  
 82 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0763P)^2 + 0.2763P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.10 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.10 \text{ e } \text{\AA}^{-3}$

Special details

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement on  $F^2$  for ALL reflections except those flagged by the user for potential systematic errors. Weighted  $R$ -factors  $wR$  and all goodnesses of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The observed criterion of  $F^2 > \sigma(F^2)$  is used only for calculating  $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|    | $x$          | $y$          | $z$           | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|--------------|--------------|---------------|----------------------------------|
| N3 | 0.69126 (7)  | 0.50665 (18) | 0.02615 (10)  | 0.0576 (4)                       |
| N6 | 0.55349 (7)  | 0.05345 (19) | -0.11311 (9)  | 0.0542 (4)                       |
| C1 | 0.67191 (10) | 0.7085 (3)   | 0.16610 (14)  | 0.0703 (6)                       |
| C2 | 0.70819 (9)  | 0.6792 (2)   | 0.10143 (14)  | 0.0634 (5)                       |
| C4 | 0.63482 (8)  | 0.3583 (2)   | 0.01373 (11)  | 0.0478 (4)                       |
| C5 | 0.61592 (8)  | 0.1747 (2)   | -0.07264 (11) | 0.0495 (4)                       |
| C7 | 0.54261 (9)  | -0.1287 (2)  | -0.19397 (13) | 0.0585 (5)                       |
| C8 | 0.59615 (9)  | 0.3750 (2)   | 0.07686 (12)  | 0.0591 (5)                       |
| C9 | 0.61501 (11) | 0.5541 (3)   | 0.15343 (14)  | 0.0728 (6)                       |
| H1 | 0.68570      | 0.83080      | 0.21760       | 0.0840*                          |
| H2 | 0.55830      | 0.26720      | 0.06750       | 0.0710*                          |
| H4 | 0.74660      | 0.78490      | 0.11040       | 0.0760*                          |
| H5 | 0.65170      | 0.14760      | -0.09810      | 0.0590*                          |
| H7 | 0.58250      | -0.11360     | -0.21680      | 0.0700*                          |
| H8 | 0.55230      | -0.27160     | -0.15260      | 0.0700*                          |
| H9 | 0.58950      | 0.57050      | 0.19620       | 0.0870*                          |

Atomic displacement parameters ( $\text{\AA}^2$ )

|    | $U^{11}$    | $U^{22}$   | $U^{33}$   | $U^{12}$    | $U^{13}$   | $U^{23}$    |
|----|-------------|------------|------------|-------------|------------|-------------|
| N3 | 0.0509 (8)  | 0.0571 (7) | 0.0602 (7) | 0.0007 (5)  | 0.0265 (6) | 0.0064 (5)  |
| N6 | 0.0501 (8)  | 0.0609 (7) | 0.0466 (6) | 0.0031 (5)  | 0.0225 (5) | 0.0005 (5)  |
| C1 | 0.0714 (11) | 0.0636 (9) | 0.0601 (9) | -0.0011 (7) | 0.0246 (8) | -0.0102 (7) |
| C2 | 0.0554 (10) | 0.0564 (9) | 0.0628 (9) | -0.0038 (6) | 0.0212 (7) | 0.0035 (6)  |
| C4 | 0.0410 (8)  | 0.0514 (7) | 0.0430 (7) | 0.0068 (5)  | 0.0171 (6) | 0.0090 (5)  |

|    |             |             |            |             |            |             |
|----|-------------|-------------|------------|-------------|------------|-------------|
| C5 | 0.0463 (8)  | 0.0551 (8)  | 0.0473 (7) | 0.0085 (6)  | 0.0251 (6) | 0.0079 (5)  |
| C7 | 0.0600 (9)  | 0.0532 (8)  | 0.0540 (8) | 0.0048 (6)  | 0.0249 (7) | -0.0001 (6) |
| C8 | 0.0561 (9)  | 0.0674 (9)  | 0.0542 (8) | -0.0052 (7) | 0.0298 (7) | -0.0034 (6) |
| C9 | 0.0743 (12) | 0.0866 (11) | 0.0628 (9) | -0.0047 (8) | 0.0401 (9) | -0.0136 (8) |

*Geometric parameters (Å, °)*

|                          |              |  |              |
|--------------------------|--------------|--|--------------|
| N3—C2                    | 1.3384 (18)  | C8—C9                                  | 1.373 (2)    |
| N3—C4                    | 1.343 (2)    | C1—H1                                  | 0.9300       |
| N6—C5                    | 1.254 (2)    | C2—H4                                  | 0.9300       |
| N6—C7                    | 1.4514 (18)  | C5—H5                                  | 0.9300       |
| C1—C2                    | 1.373 (3)    | C7—H7                                  | 0.9700       |
| C1—C9                    | 1.372 (3)    | C7—H8                                  | 0.9700       |
| C4—C5                    | 1.4730 (18)  | C8—H2                                  | 0.9300       |
| C4—C8                    | 1.387 (2)    | C9—H9                                  | 0.9300       |
| C7—C7 <sup>i</sup>       | 1.516 (2)    |  |              |
|                          |              |  |              |
| C2—N3—C4                 | 116.96 (15)  | N3—C2—H4                               | 118.00       |
| C5—N6—C7                 | 117.93 (15)  | C1—C2—H4                               | 118.00       |
| C2—C1—C9                 | 118.85 (16)  | N6—C5—H5                               | 119.00       |
| N3—C2—C1                 | 123.50 (16)  | C4—C5—H5                               | 119.00       |
| N3—C4—C5                 | 115.43 (14)  | N6—C7—H7                               | 109.00       |
| N3—C4—C8                 | 122.94 (12)  | N6—C7—H8                               | 109.00       |
| C5—C4—C8                 | 121.62 (13)  | H7—C7—H8                               | 108.00       |
| N6—C5—C4                 | 122.55 (15)  | C7 <sup>i</sup> —C7—H7                 | 109.00       |
| N6—C7—C7 <sup>i</sup>    | 111.74 (13)  | C7 <sup>i</sup> —C7—H8                 | 109.00       |
| C4—C8—C9                 | 118.56 (16)  | C4—C8—H2                               | 121.00       |
| C1—C9—C8                 | 119.17 (19)  | C9—C8—H2                               | 121.00       |
| C2—C1—H1                 | 121.00       | C1—C9—H9                               | 120.00       |
| C9—C1—H1                 | 121.00       | C8—C9—H9                               | 120.00       |
|                          |              |  |              |
| C2—N3—C4—C5              | 178.17 (12)  | N3—C4—C5—N6                            | -164.26 (12) |
| C2—N3—C4—C8              | -1.6 (2)     | C5—C4—C8—C9                            | -178.16 (14) |
| C4—N3—C2—C1              | 0.9 (2)      | C8—C4—C5—N6                            | 15.5 (2)     |
| C7—N6—C5—C4              | -177.50 (11) | N3—C4—C8—C9                            | 1.6 (2)      |
| C5—N6—C7—C7 <sup>i</sup> | -131.18 (14) | N6—C7—C7 <sup>i</sup> —N6 <sup>i</sup> | 73.41 (17)   |
| C9—C1—C2—N3              | -0.2 (3)     | C4—C8—C9—C1                            | -0.8 (2)     |
| C2—C1—C9—C8              | 0.1 (3)      |  |              |

Symmetry code: (i)  $-x+1, y, -z-1/2$ .