



Crystal structure of *N*-(4-chlorophenyl)-benzothioamide

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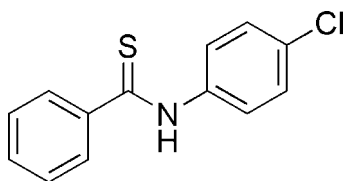
The title compound, C₁₃H₁₀CINS, exhibits a *trans* conformation with regard to the axis of the C–N bond. The benzene and phenyl rings are inclined to one another by 85.06 (8)°. In the crystal, molecules are linked by N–H···S=C hydrogen bonds, forming chains along [001].

Keywords: crystal structure; benzothioamide; N–H···S hydrogen bonding.

CCDC reference: 1061304

1. Related literature

For hydrogen bonding of amides, see: Taylor *et al.* (1984); Leiserowitz & Schmidt (1969). For the preparation and for the use of thioamides as intermediates in chemical transformations, see: Li *et al.* (2012, 2015). For related structures, see: Omondi *et al.* (2012); Nagasawa *et al.* (2014).



2. Experimental

2.1. Crystal data

C₁₃H₁₀CINS

$M_r = 247.73$

Monoclinic, $P2_1/c$
 $a = 11.943$ (2) Å
 $b = 12.689$ (3) Å
 $c = 7.9764$ (16) Å
 $\beta = 109.30$ (3)°
 $V = 1140.9$ (4) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.49$ mm⁻¹
 $T = 113$ K
 $0.22 \times 0.20 \times 0.12$ mm

2.2. Data collection

Rigaku Saturn CCD area-detector diffractometer
Absorption correction: multi-scan (*CrystalClear*; Rigaku/MS, 2005)
 $T_{\min} = 0.901$, $T_{\max} = 0.944$

7517 measured reflections
2010 independent reflections
1701 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.030$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.028$
 $wR(F^2) = 0.077$
 $S = 1.05$
2010 reflections
150 parameters
1 restraint

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.22$ e Å⁻³
 $\Delta\rho_{\min} = -0.18$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N1-H1\cdots S1^i$	0.89 (1)	2.49 (1)	3.346 (15)	163 (1)

Symmetry code: (i) $x, -y + \frac{1}{2}, z - \frac{1}{2}$.

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

Supporting information for this paper is available from the IUCr electronic archives (Reference: SU5123).

References

- Leiserowitz, L. & Schmidt, G. M. (1969). *J. Chem. Soc. A*, pp. 2372–2382.
Li, J. S., Cheng, C., Zhang, X. R., Li, Z. W., Cai, F. F., Xue, Y. & Liu, W. D. (2012). *Chin. J. Chem.* **30**, 1687–1689.
Li, J. S., Xue, Y., Li, P. Y., Li, Z. W., Lu, C. H., Liu, W. D., Pang, H. L., Liu, D. H., Lin, M. S., Luo, B. B. & Jiang, W. (2015). *Res. Chem. Intermed.* **41**, 2235–2247.
Nagasawa, M., Sasanuma, Y. & Masu, H. (2014). *Acta Cryst.* **E70**, o639.
Omondi, B. & Levendis, D. C. (2012). *Acta Cryst.* **E68**, o2604.
Rigaku/MS (2005). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Taylor, R., Kennard, O. & Versichel, W. (1984). *Acta Cryst.* **B40**, 280–288.

supporting information

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Crystal structure of *N*-(4-chlorophenyl)benzothioamide

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S1. Comment

Thioamides are extensively used as extractant for heavy metals in environmental chemistry, as intermediate for important chemical transformations (Li *et al.*, 2015), and also used to replace the amide bonds as isosteres. It's well known that amide units can be connected by a N—H···O=C hydrogen bonds (Taylor *et al.*, 1984; Leiserowitz & Schmidt, 1969), and also the structures of some thioamides have been documented (Omondi *et al.*, 2012; Nagasawa *et al.*, 2014). Herein, we report the crystal structure of the title compound (I).

Figure 1 has shows the molecular structure of the title compound, whose thioamide unit adopts a *trans* conformation around the central C—N bond. The C=S double bond is deviated from its connected phenyl ring [torsion angles: S1/C7/C8/C9 -145.34 (13)°, S1/C7/C8/C13 33.62 (19)°]. The benzene and phenyl rings are inclined to one another by 85.06 (8) °.

In the crystal, molecules are linked *via* N—H···S=C hydrogen bonds, forming chains along the *c* axis direction (Table 1 and Fig. 2).

S2. Experimental

The title compound was prepared from the Beckmann rearrangement from its corresponding ketoximes following a published procedure (Li *et al.*, 2012). It was isolated by flash chromatography and yellow block-like crystal of the title compound were obtained *via* natural evaporation from the diluent.

S3. Refinement

The thioamide N—H atom was located in a difference Fourier map and freely refined. The C-bound H atoms were included in calculated positions and refined as riding atoms: C—H = 0.93 Å with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$.

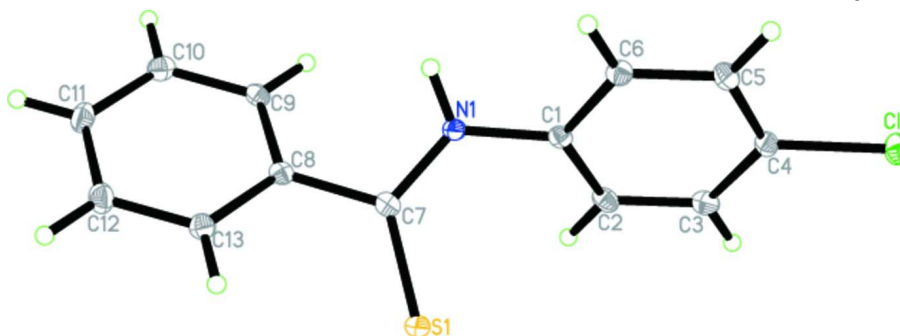


Figure 1

The molecular structure of the title compound, with atom labelling. Displacement ellipsoids are drawn at the 30% probability level.

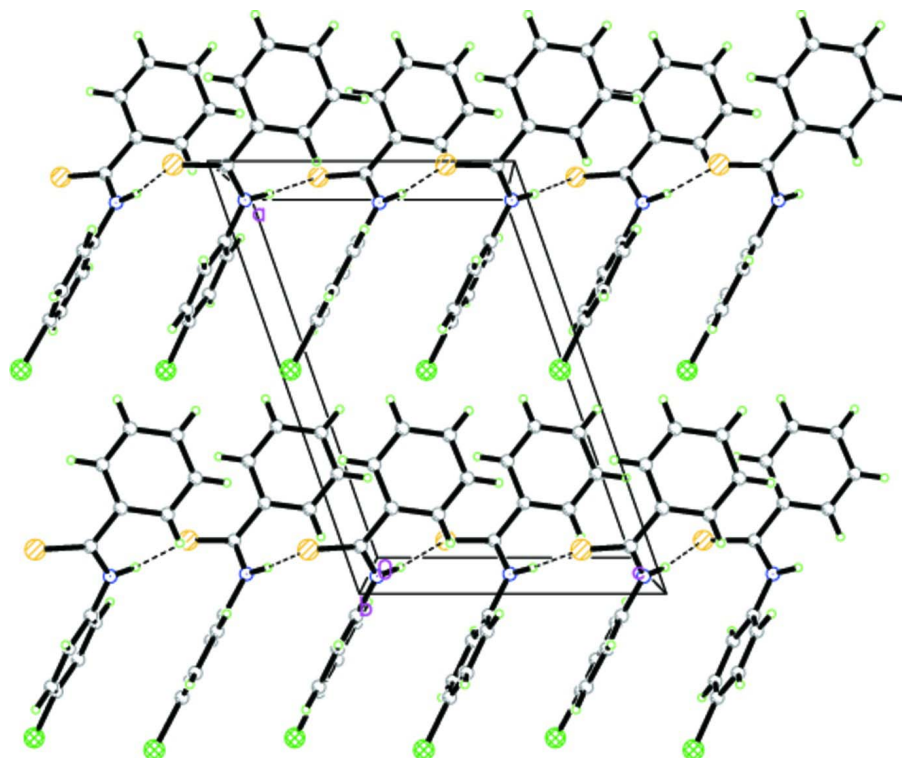


Figure 2

A view along the *b* axis of the crystal packing of the title compound. The N—H...S hydrogen bonds are shown as dashed lines (see Table 1 for details).

N-(4-Chlorophenyl)benzothioamide

Crystal data

$C_{13}H_{10}ClNS$

$M_r = 247.73$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 11.943\ (2)\ \text{\AA}$

$b = 12.689\ (3)\ \text{\AA}$

$c = 7.9764\ (16)\ \text{\AA}$

$\beta = 109.30\ (3)^\circ$

$V = 1140.9\ (4)\ \text{\AA}^3$

$Z = 4$

$F(000) = 512$

$D_x = 1.442\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 3499 reflections

$\theta = 1.8\text{--}27.9^\circ$

$\mu = 0.49\ \text{mm}^{-1}$

$T = 113\ \text{K}$

Block, yellow

$0.22 \times 0.20 \times 0.12\ \text{mm}$

Data collection

Rigaku Saturn CCD area-detector
diffractometer

Radiation source: rotating anode

Confocal monochromator

Detector resolution: $7.31\ \text{pixels mm}^{-1}$

ω and ϕ scans

Absorption correction: multi-scan

(*CrystalClear*; Rigaku/MS, 2005)

$T_{\min} = 0.901$, $T_{\max} = 0.944$

7517 measured reflections

2010 independent reflections

1701 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.030$

$\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 2.4^\circ$

$h = -11 \rightarrow 14$

$k = -15 \rightarrow 15$

$l = -9 \rightarrow 9$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.028$

$wR(F^2) = 0.077$

$S = 1.05$

2010 reflections

150 parameters

1 restraint

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0477P)^2 + 0.1743P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.002$

$\Delta\rho_{\max} = 0.22 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$

Extinction correction: *SHELXL97* (Sheldrick,
2008), $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.050 (4)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.94504 (4)	0.58973 (3)	1.20150 (5)	0.01919 (16)
Cl1	1.47747 (4)	0.84240 (4)	1.54329 (6)	0.02760 (17)
N1	1.02619 (11)	0.71403 (10)	0.99722 (17)	0.0160 (3)
C1	1.13474 (13)	0.74209 (12)	1.1311 (2)	0.0159 (3)
C2	1.21244 (14)	0.66609 (13)	1.2321 (2)	0.0201 (4)
H2	1.1936	0.5949	1.2147	0.024*
C3	1.31777 (14)	0.69722 (13)	1.3583 (2)	0.0218 (4)
H3	1.3698	0.6470	1.4264	0.026*
C4	1.34551 (14)	0.80302 (13)	1.3828 (2)	0.0186 (4)
C5	1.26984 (15)	0.87901 (13)	1.2816 (2)	0.0219 (4)
H5	1.2895	0.9501	1.2982	0.026*
C6	1.16448 (15)	0.84782 (13)	1.1550 (2)	0.0194 (4)
H6	1.1134	0.8982	1.0857	0.023*
C7	0.94159 (14)	0.64895 (12)	1.0128 (2)	0.0159 (4)
C8	0.83987 (14)	0.63239 (12)	0.8467 (2)	0.0161 (4)
C9	0.85648 (15)	0.62942 (12)	0.6821 (2)	0.0182 (4)
H9	0.9320	0.6389	0.6751	0.022*
C10	0.76076 (16)	0.61237 (13)	0.5280 (2)	0.0226 (4)
H10	0.7725	0.6106	0.4184	0.027*
C11	0.64826 (16)	0.59810 (13)	0.5376 (2)	0.0249 (4)
H11	0.5842	0.5868	0.4347	0.030*
C12	0.63130 (15)	0.60079 (13)	0.7015 (2)	0.0251 (4)
H12	0.5557	0.5910	0.7082	0.030*

C13	0.72593 (14)	0.61784 (12)	0.8545 (2)	0.0201 (4)
H13	0.7137	0.6197	0.9637	0.024*
H1	1.0058 (16)	0.7552 (13)	0.9013 (17)	0.030 (5)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0205 (2)	0.0192 (2)	0.0179 (3)	-0.00251 (17)	0.00650 (18)	0.00248 (16)
C11	0.0174 (2)	0.0353 (3)	0.0268 (3)	-0.00442 (18)	0.00281 (18)	-0.00048 (18)
N1	0.0177 (7)	0.0151 (7)	0.0155 (7)	-0.0010 (6)	0.0057 (6)	0.0010 (6)
C1	0.0158 (8)	0.0181 (8)	0.0164 (8)	-0.0003 (7)	0.0090 (7)	-0.0010 (6)
C2	0.0187 (9)	0.0154 (8)	0.0280 (9)	0.0004 (7)	0.0102 (7)	0.0010 (7)
C3	0.0170 (8)	0.0215 (9)	0.0279 (10)	0.0042 (7)	0.0089 (7)	0.0066 (7)
C4	0.0138 (8)	0.0245 (9)	0.0186 (8)	-0.0014 (7)	0.0070 (7)	-0.0008 (7)
C5	0.0231 (9)	0.0177 (8)	0.0249 (9)	-0.0026 (7)	0.0081 (8)	-0.0024 (7)
C6	0.0192 (9)	0.0175 (8)	0.0211 (9)	0.0029 (7)	0.0063 (7)	0.0016 (6)
C7	0.0173 (8)	0.0118 (7)	0.0210 (9)	0.0028 (6)	0.0096 (7)	-0.0021 (6)
C8	0.0186 (8)	0.0098 (7)	0.0201 (8)	0.0007 (6)	0.0065 (7)	-0.0005 (6)
C9	0.0202 (9)	0.0133 (8)	0.0223 (9)	-0.0018 (7)	0.0085 (7)	0.0000 (7)
C10	0.0300 (10)	0.0194 (8)	0.0186 (9)	-0.0017 (8)	0.0081 (8)	-0.0018 (7)
C11	0.0234 (9)	0.0257 (9)	0.0202 (9)	-0.0010 (8)	0.0001 (7)	-0.0017 (7)
C12	0.0171 (9)	0.0280 (9)	0.0285 (10)	0.0010 (8)	0.0052 (7)	0.0009 (8)
C13	0.0192 (9)	0.0210 (9)	0.0212 (9)	0.0019 (7)	0.0080 (7)	0.0011 (7)

Geometric parameters (Å, °)

S1—C7	1.6705 (16)	C6—H6	0.9300
C11—C4	1.7430 (17)	C7—C8	1.487 (2)
N1—C7	1.342 (2)	C8—C9	1.391 (2)
N1—C1	1.426 (2)	C8—C13	1.395 (2)
N1—H1	0.891 (9)	C9—C10	1.392 (2)
C1—C6	1.385 (2)	C9—H9	0.9300
C1—C2	1.395 (2)	C10—C11	1.383 (2)
C2—C3	1.384 (2)	C10—H10	0.9300
C2—H2	0.9300	C11—C12	1.388 (3)
C3—C4	1.381 (2)	C11—H11	0.9300
C3—H3	0.9300	C12—C13	1.379 (2)
C4—C5	1.384 (2)	C12—H12	0.9300
C5—C6	1.385 (2)	C13—H13	0.9300
C5—H5	0.9300		
C7—N1—C1	127.60 (13)	N1—C7—C8	115.00 (13)
C7—N1—H1	116.1 (12)	N1—C7—S1	124.40 (13)
C1—N1—H1	114.8 (12)	C8—C7—S1	120.59 (12)
C6—C1—C2	119.92 (15)	C9—C8—C13	118.95 (15)
C6—C1—N1	118.25 (14)	C9—C8—C7	121.01 (15)
C2—C1—N1	121.77 (14)	C13—C8—C7	120.03 (15)
C3—C2—C1	119.60 (15)	C8—C9—C10	120.42 (16)

C3—C2—H2	120.2	C8—C9—H9	119.8
C1—C2—H2	120.2	C10—C9—H9	119.8
C4—C3—C2	119.90 (15)	C11—C10—C9	120.03 (16)
C4—C3—H3	120.0	C11—C10—H10	120.0
C2—C3—H3	120.0	C9—C10—H10	120.0
C3—C4—C5	120.96 (15)	C10—C11—C12	119.76 (16)
C3—C4—C11	119.94 (13)	C10—C11—H11	120.1
C5—C4—C11	119.09 (13)	C12—C11—H11	120.1
C4—C5—C6	119.13 (15)	C13—C12—C11	120.34 (16)
C4—C5—H5	120.4	C13—C12—H12	119.8
C6—C5—H5	120.4	C11—C12—H12	119.8
C1—C6—C5	120.46 (15)	C12—C13—C8	120.50 (16)
C1—C6—H6	119.8	C12—C13—H13	119.7
C5—C6—H6	119.8	C8—C13—H13	119.7
C7—N1—C1—C6	-131.37 (16)	C1—N1—C7—S1	2.6 (2)
C7—N1—C1—C2	51.1 (2)	N1—C7—C8—C9	35.2 (2)
C6—C1—C2—C3	1.3 (2)	S1—C7—C8—C9	-145.34 (13)
N1—C1—C2—C3	178.80 (14)	N1—C7—C8—C13	-145.86 (15)
C1—C2—C3—C4	-0.3 (2)	S1—C7—C8—C13	33.62 (19)
C2—C3—C4—C5	-0.7 (2)	C13—C8—C9—C10	0.1 (2)
C2—C3—C4—C11	179.35 (12)	C7—C8—C9—C10	179.08 (14)
C3—C4—C5—C6	0.5 (2)	C8—C9—C10—C11	-0.1 (2)
C11—C4—C5—C6	-179.46 (12)	C9—C10—C11—C12	0.0 (2)
C2—C1—C6—C5	-1.5 (2)	C10—C11—C12—C13	0.2 (3)
N1—C1—C6—C5	-179.01 (14)	C11—C12—C13—C8	-0.1 (2)
C4—C5—C6—C1	0.5 (2)	C9—C8—C13—C12	0.0 (2)
C1—N1—C7—C8	-177.96 (13)	C7—C8—C13—C12	-178.98 (14)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...S1 ⁱ	0.89 (1)	2.49 (1)	3.346 (15)	163 (1)

Symmetry code: (i) *x*, -*y*+3/2, *z*-1/2.