



Crystal structure of 1-{2-[(2-methoxyphenyl)selenyl]phenyl}-4-phenyl-1*H*-1,2,3-triazole

Leandro R. S. Camargo,^a Julio Zukerman-Schpector,^{a*} Anna M. Deobald,^{b‡} Antonio L. Braga^c and Edward R. T. Tiekink^d

^aDepartamento de Química, Universidade Federal de São Carlos, 13565-905 São Carlos, SP, Brazil, ^bDepartamento de Química, Universidade Federal de Santa Maria, 97105-900 Santa Maria, RS, Brazil, ^cDepartamento de Química, Universidade Federal de Santa Catarina, 88040-900 Florianópolis, SC, Brazil, and ^dDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia. *Correspondence e-mail: julio@power.ufscar.br

Received 12 February 2015; accepted 16 February 2015

Edited by P. C. Healy, Griffith University, Australia

In the title compound, C₂₁H₁₇N₃OSe, the dihedral angles between the central five-membered ring and the C- and N-bound rings are 17.89 (10) and 42.35 (10)°, respectively, indicating the molecule is twisted. The dihedral angle between the Se-bound rings is 85.36 (10)°. A close intramolecular Se···O contact of 2.8507 (13) Å is noted. In the crystal, C—H···O, C—H···N and C—H···π interactions lead to the formation of supramolecular layers parallel to (011); these stack with no specific intermolecular interactions between them.

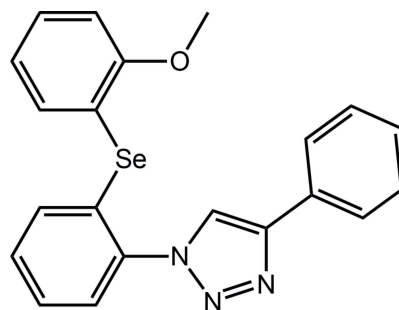
Keywords: crystal structure; organoselenium; Se···O halogen bonding; hydrogen bonding; C—H···π interactions.

CCDC reference: 1049507

1. Related literature

For background to arylseleno-1,2,3-triazoles and to the synthesis of the title compound, see: Deobald *et al.* (2011). For an analysis of intra- and intermolecular Se···O interactions, see: Linden *et al.* (2014). For a related organoselenium compound with a 1,2,3-triazole residue, see: Camargo *et al.* (2015).

[‡] Present address: Instituto Federal de Educação, Ciência e Tecnologia Farroupilha Rua Erechim, 860 - Bairro Planalto, 98280-000 Panambi, RS, Brazil.



2. Experimental

2.1. Crystal data

C ₂₁ H ₁₇ N ₃ OSe	γ = 85.340 (4)°
M _r = 406.33	V = 874.83 (8) Å ³
Triclinic, P $\bar{1}$	Z = 2
a = 5.6565 (3) Å	Mo Kα radiation
b = 10.3682 (5) Å	μ = 2.16 mm ⁻¹
c = 15.3358 (7) Å	T = 100 K
α = 81.604 (4)°	0.30 × 0.20 × 0.10 mm
β = 80.006 (4)°	

2.2. Data collection

Agilent SuperNova CCD diffractometer	6845 measured reflections
Absorption correction: multi-scan (CrysAlis PRO; Agilent, 2011)	3869 independent reflections
T _{min} = 0.759, T _{max} = 1.000	3548 reflections with I > 2σ(I)
	R _{int} = 0.040

2.3. Refinement

R[F ² > 2σ(F ²)] = 0.028	236 parameters
wR(F ²) = 0.064	H-atom parameters constrained
S = 1.01	Δρ _{max} = 0.41 e Å ⁻³
3869 reflections	Δρ _{min} = -0.52 e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C1–C6 ring.

D—H···A	D—H	H···A	D···A	D—H···A
C18—H18···O1 ⁱ	0.95	2.54	3.472 (2)	165
C14—H14···N3 ⁱⁱ	0.95	2.58	3.520 (2)	170
C10—H10···Cg1 ⁱⁱⁱ	0.95	2.82	3.630 (2)	144

Symmetry codes: (i) $-x+2, -y+2, -z+1$; (ii) $x-1, y, z$; (iii) $-x+1, -y+1, -z+2$.

Data collection: CrysAlis PRO (Agilent, 2011); cell refinement: CrysAlis PRO; data reduction: CrysAlis PRO; program(s) used to solve structure: SIR2014 (Burla *et al.*, 2015); program(s) used to refine structure: SHELXL2014 (Sheldrick, 2015); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012) and DIAMOND (Brandenburg, 2006); software used to prepare material for publication: MarvinSketch (ChemAxon, 2010) and publCIF (Westrip, 2010).

Acknowledgements

The Brazilian agencies CNPq (305626/2013-2 to JZ-S), CAPES, FAPESC and FAPESP (2010/10855-5 to LRSC) are acknowledged for financial support.

Supporting information for this paper is available from the IUCr electronic archives (Reference: HG5432).

References

- Agilent (2011). *CrysAlis PRO*. Agilent Technologies, Yarnton, England.
- Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
- Burla, M. C., Caliandro, R., Carrozzini, B., Cascarano, G. L., Cuocci, C., Giovacazzo, C., Mallamo, M., Mazzone, A. & Polidori, G. (2015). *J. Appl. Cryst.* **48**, 306–309.
- Camargo, L. R. S., Zukerman-Schpector, J., Deobald, A. M., Braga, A. L. & Tiekink, E. R. T. (2015). *Acta Cryst.* **E71**, o200–o201.
- ChemAxon (2010). *Marvinsketch*. <http://www.chemaxon.com>.
- Deobald, A. M., Camargo, L. R. S., Hörner, M., Rodrigues, O. E. D., Alves, D. & Braga, A. L. (2011). *Synthesis*, pp. 2397–2406.
- Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.
- Linden, A., Zhou, Y. & Heimgartner, H. (2014). *Acta Cryst.* **C70**, 482–487.
- Sheldrick, G. M. (2015). *Acta Cryst.* **C71**, 3–8.
- Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

supporting information

Acta Cryst. (2015). E71, o202–o203 [doi:10.1107/S2056989015003230]

Crystal structure of 1-{2-[(2-methoxyphenyl)selenanyl]phenyl}-4-phenyl-1*H*-1,2,3-triazole

Leandro R. S. Camargo, Julio Zukerman-Schpector, Anna M. Deobald, Antonio L. Braga and Edward R. T. Tiekink

S1. Experimental

The compound was prepared in accord with the literature (Deobald *et al.*, 2011). Crystals were obtained by taking 200 mg of sample into a sample vial containing methanol (10 ml) and letting it stand at room temperature.

S2. Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H = 0.95 to 0.98 Å) and were included in the refinement in the riding model approximation, with $U_{iso}(H) = 1.2-1.5U_{eq}(C)$.

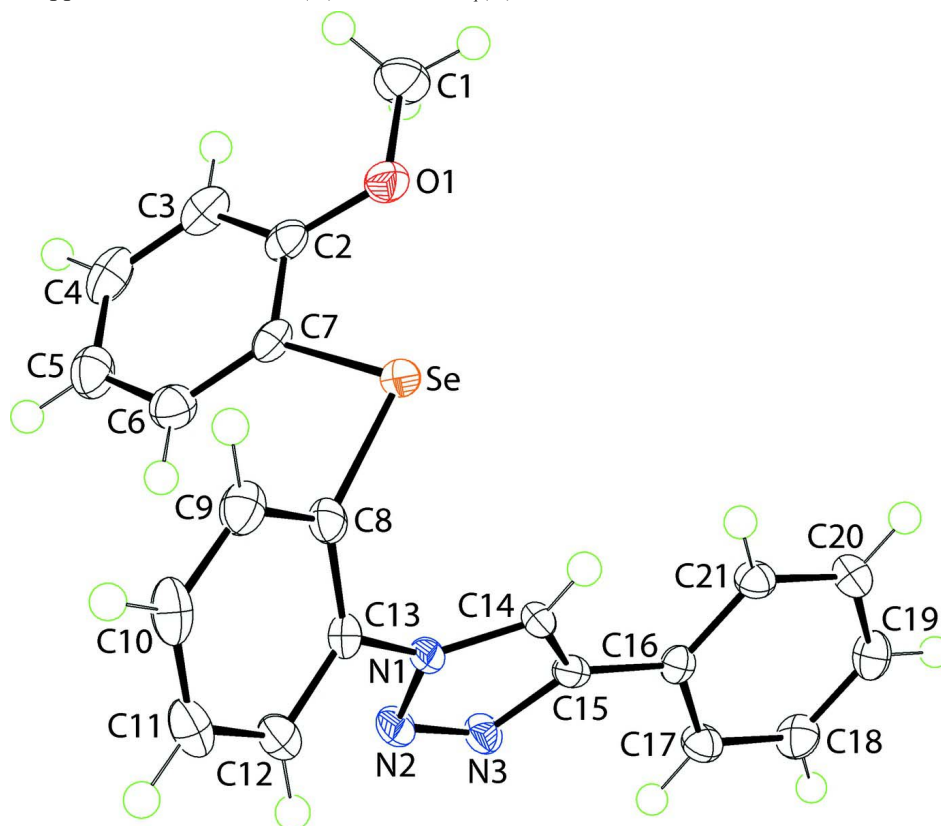
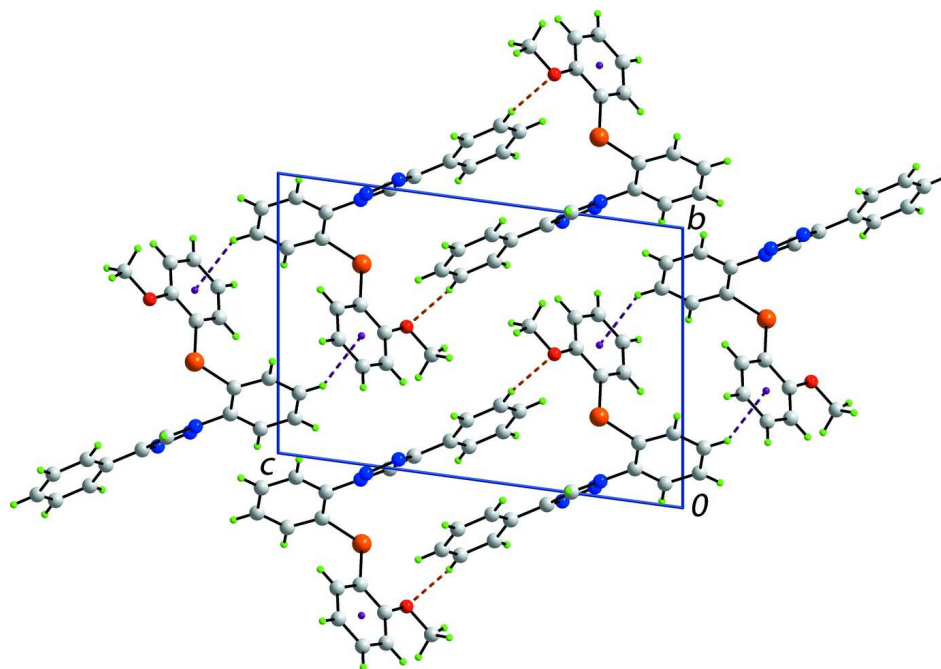


Figure 1

The molecular structure of the title compound showing the atom-labelling scheme and displacement ellipsoids at the 70% probability level.

**Figure 2**

A view in projection down the a axis of the unit-cell contents. The C—H \cdots O, C—H \cdots N and C—H \cdots π interactions are shown as orange, blue and purple dashed lines, respectively.

1-{2-[(2-Methoxyphenyl)selanyl]phenyl}-4-phenyl-1*H*-1,2,3-triazole

Crystal data

$C_{21}H_{17}N_3OSe$

$M_r = 406.33$

Triclinic, $P\bar{1}$

$a = 5.6565$ (3) Å

$b = 10.3682$ (5) Å

$c = 15.3358$ (7) Å

$\alpha = 81.604$ (4)°

$\beta = 80.006$ (4)°

$\gamma = 85.340$ (4)°

$V = 874.83$ (8) Å³

$Z = 2$

$F(000) = 412$

$D_x = 1.543$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 4524 reflections

$\theta = 2.6$ – 29.2 °

$\mu = 2.16$ mm⁻¹

$T = 100$ K

Prism, colourless

$0.30 \times 0.20 \times 0.10$ mm

Data collection

Agilent SuperNova CCD
diffractometer

Radiation source: SuperNova (Cu) X-ray

Source

ω scans

Absorption correction: multi-scan
(*CrysAlis PRO*; Agilent, 2011)

$T_{\min} = 0.759$, $T_{\max} = 1.000$

6845 measured reflections

3869 independent reflections

3548 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.040$

$\theta_{\max} = 27.5$ °, $\theta_{\min} = 2.6$ °

$h = -6 \rightarrow 7$

$k = -13 \rightarrow 12$

$l = -19 \rightarrow 19$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.028$
 $wR(F^2) = 0.064$
 $S = 1.01$
 3869 reflections
 236 parameters
 0 restraints

Hydrogen site location: inferred from neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.026P)^2 + 0.4456P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.002$
 $\Delta\rho_{\max} = 0.41 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.52 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Se	0.46182 (3)	0.71551 (2)	0.79326 (2)	0.01503 (7)
O1	0.5130 (2)	0.51326 (12)	0.68268 (9)	0.0185 (3)
N1	0.8216 (3)	0.94528 (14)	0.79605 (10)	0.0128 (3)
N2	1.0599 (3)	0.96928 (15)	0.78243 (11)	0.0154 (3)
N3	1.1085 (3)	1.03509 (15)	0.70214 (10)	0.0150 (3)
C1	0.4930 (4)	0.4251 (2)	0.62080 (14)	0.0255 (5)
H1A	0.3570	0.4549	0.5897	0.038*
H1B	0.4669	0.3374	0.6533	0.038*
H1C	0.6414	0.4227	0.5771	0.038*
C2	0.6840 (3)	0.48102 (18)	0.73647 (12)	0.0164 (4)
C3	0.8446 (4)	0.37256 (18)	0.73332 (13)	0.0209 (4)
H3	0.8404	0.3139	0.6916	0.025*
C4	1.0113 (4)	0.34994 (19)	0.79132 (14)	0.0238 (4)
H4	1.1196	0.2750	0.7897	0.029*
C5	1.0204 (4)	0.4358 (2)	0.85130 (14)	0.0227 (4)
H5	1.1355	0.4202	0.8905	0.027*
C6	0.8611 (3)	0.54507 (19)	0.85434 (13)	0.0194 (4)
H6	0.8684	0.6041	0.8956	0.023*
C7	0.6917 (3)	0.56852 (17)	0.79755 (12)	0.0150 (4)
C8	0.5446 (3)	0.78849 (18)	0.89132 (12)	0.0153 (4)
C9	0.4431 (4)	0.73766 (19)	0.97760 (13)	0.0194 (4)
H9	0.3236	0.6758	0.9856	0.023*
C10	0.5126 (4)	0.7755 (2)	1.05209 (13)	0.0223 (4)
H10	0.4427	0.7386	1.1104	0.027*
C11	0.6845 (4)	0.8671 (2)	1.04128 (13)	0.0224 (4)
H11	0.7347	0.8922	1.0920	0.027*
C12	0.7825 (4)	0.92197 (19)	0.95603 (13)	0.0182 (4)
H12	0.8974	0.9863	0.9485	0.022*
C13	0.7135 (3)	0.88317 (17)	0.88174 (12)	0.0133 (4)
C14	0.7201 (3)	0.99764 (17)	0.72442 (12)	0.0127 (3)

H14	0.5569	0.9958	0.7172	0.015*
C15	0.9046 (3)	1.05438 (17)	0.66405 (12)	0.0124 (3)
C16	0.8998 (3)	1.12626 (17)	0.57456 (12)	0.0126 (4)
C17	1.0842 (3)	1.20866 (18)	0.53447 (13)	0.0166 (4)
H17	1.2131	1.2173	0.5651	0.020*
C18	1.0791 (3)	1.27779 (19)	0.44998 (13)	0.0192 (4)
H18	1.2047	1.3334	0.4233	0.023*
C19	0.8925 (3)	1.26635 (19)	0.40436 (13)	0.0184 (4)
H19	0.8903	1.3136	0.3465	0.022*
C20	0.7089 (3)	1.18527 (18)	0.44379 (13)	0.0180 (4)
H20	0.5802	1.1772	0.4129	0.022*
C21	0.7127 (3)	1.11581 (18)	0.52828 (12)	0.0152 (4)
H21	0.5861	1.0606	0.5547	0.018*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Se	0.01551 (10)	0.01235 (10)	0.01832 (11)	-0.00120 (7)	-0.00576 (7)	-0.00192 (7)
O1	0.0237 (7)	0.0152 (7)	0.0177 (7)	-0.0024 (5)	-0.0049 (6)	-0.0032 (5)
N1	0.0117 (7)	0.0131 (7)	0.0145 (8)	-0.0021 (6)	-0.0051 (6)	-0.0002 (6)
N2	0.0108 (7)	0.0180 (8)	0.0185 (8)	-0.0012 (6)	-0.0056 (6)	-0.0015 (6)
N3	0.0125 (7)	0.0174 (8)	0.0155 (8)	-0.0012 (6)	-0.0040 (6)	-0.0011 (6)
C1	0.0338 (12)	0.0244 (11)	0.0210 (10)	-0.0045 (9)	-0.0057 (9)	-0.0090 (8)
C2	0.0193 (9)	0.0138 (9)	0.0147 (9)	-0.0045 (7)	0.0008 (8)	0.0010 (7)
C3	0.0252 (10)	0.0148 (9)	0.0200 (10)	-0.0012 (8)	0.0030 (8)	-0.0017 (8)
C4	0.0218 (10)	0.0165 (10)	0.0282 (11)	0.0032 (8)	0.0033 (9)	0.0028 (8)
C5	0.0182 (10)	0.0219 (10)	0.0265 (11)	0.0008 (8)	-0.0053 (9)	0.0021 (8)
C6	0.0182 (9)	0.0181 (9)	0.0218 (10)	-0.0020 (7)	-0.0031 (8)	-0.0024 (8)
C7	0.0148 (9)	0.0112 (8)	0.0168 (9)	-0.0014 (7)	0.0006 (7)	0.0014 (7)
C8	0.0142 (9)	0.0152 (9)	0.0168 (9)	0.0021 (7)	-0.0053 (7)	-0.0016 (7)
C9	0.0177 (9)	0.0185 (10)	0.0200 (10)	-0.0012 (7)	-0.0018 (8)	0.0020 (8)
C10	0.0228 (10)	0.0273 (11)	0.0139 (9)	0.0043 (8)	-0.0026 (8)	0.0031 (8)
C11	0.0243 (10)	0.0285 (11)	0.0154 (10)	0.0017 (8)	-0.0081 (8)	-0.0020 (8)
C12	0.0178 (9)	0.0205 (10)	0.0179 (10)	-0.0017 (7)	-0.0074 (8)	-0.0017 (8)
C13	0.0129 (8)	0.0135 (8)	0.0126 (9)	0.0018 (7)	-0.0029 (7)	0.0006 (7)
C14	0.0129 (8)	0.0135 (8)	0.0129 (9)	-0.0002 (7)	-0.0056 (7)	-0.0021 (7)
C15	0.0110 (8)	0.0123 (8)	0.0148 (9)	-0.0003 (6)	-0.0031 (7)	-0.0043 (7)
C16	0.0122 (8)	0.0121 (8)	0.0127 (9)	0.0015 (7)	-0.0007 (7)	-0.0022 (7)
C17	0.0139 (9)	0.0183 (9)	0.0183 (10)	-0.0031 (7)	-0.0036 (8)	-0.0022 (7)
C18	0.0178 (9)	0.0187 (9)	0.0191 (10)	-0.0047 (7)	0.0002 (8)	0.0018 (8)
C19	0.0199 (10)	0.0193 (9)	0.0136 (9)	0.0020 (8)	0.0000 (8)	0.0012 (7)
C20	0.0168 (9)	0.0208 (10)	0.0174 (10)	-0.0006 (8)	-0.0060 (8)	-0.0021 (8)
C21	0.0134 (9)	0.0160 (9)	0.0163 (9)	-0.0032 (7)	-0.0023 (7)	-0.0018 (7)

Geometric parameters (Å, °)

Se—C8	1.9202 (19)	C8—C13	1.400 (2)
Se—C7	1.9224 (19)	C9—C10	1.388 (3)

O1—C2	1.368 (2)	C9—H9	0.9500
O1—C1	1.434 (2)	C10—C11	1.387 (3)
N1—C14	1.351 (2)	C10—H10	0.9500
N1—N2	1.365 (2)	C11—C12	1.387 (3)
N1—C13	1.433 (2)	C11—H11	0.9500
N2—N3	1.313 (2)	C12—C13	1.388 (3)
N3—C15	1.369 (2)	C12—H12	0.9500
C1—H1A	0.9800	C14—C15	1.378 (2)
C1—H1B	0.9800	C14—H14	0.9500
C1—H1C	0.9800	C15—C16	1.467 (2)
C2—C3	1.388 (3)	C16—C21	1.392 (3)
C2—C7	1.404 (3)	C16—C17	1.402 (2)
C3—C4	1.389 (3)	C17—C18	1.390 (3)
C3—H3	0.9500	C17—H17	0.9500
C4—C5	1.380 (3)	C18—C19	1.385 (3)
C4—H4	0.9500	C18—H18	0.9500
C5—C6	1.390 (3)	C19—C20	1.388 (3)
C5—H5	0.9500	C19—H19	0.9500
C6—C7	1.388 (3)	C20—C21	1.390 (3)
C6—H6	0.9500	C20—H20	0.9500
C8—C9	1.393 (3)	C21—H21	0.9500
C8—Se—C7	96.32 (8)	C11—C10—C9	119.82 (18)
C2—O1—C1	117.14 (16)	C11—C10—H10	120.1
C14—N1—N2	110.78 (14)	C9—C10—H10	120.1
C14—N1—C13	130.06 (15)	C12—C11—C10	119.69 (19)
N2—N1—C13	118.87 (15)	C12—C11—H11	120.2
N3—N2—N1	106.67 (14)	C10—C11—H11	120.2
N2—N3—C15	109.61 (14)	C11—C12—C13	120.24 (18)
O1—C1—H1A	109.5	C11—C12—H12	119.9
O1—C1—H1B	109.5	C13—C12—H12	119.9
H1A—C1—H1B	109.5	C12—C13—C8	120.86 (16)
O1—C1—H1C	109.5	C12—C13—N1	116.82 (16)
H1A—C1—H1C	109.5	C8—C13—N1	122.32 (16)
H1B—C1—H1C	109.5	N1—C14—C15	104.97 (16)
O1—C2—C3	125.05 (17)	N1—C14—H14	127.5
O1—C2—C7	114.90 (17)	C15—C14—H14	127.5
C3—C2—C7	120.05 (19)	N3—C15—C14	107.96 (16)
C2—C3—C4	119.93 (19)	N3—C15—C16	122.74 (15)
C2—C3—H3	120.0	C14—C15—C16	129.29 (17)
C4—C3—H3	120.0	C21—C16—C17	118.71 (17)
C5—C4—C3	120.32 (19)	C21—C16—C15	121.23 (16)
C5—C4—H4	119.8	C17—C16—C15	120.06 (17)
C3—C4—H4	119.8	C18—C17—C16	120.19 (18)
C4—C5—C6	120.0 (2)	C18—C17—H17	119.9
C4—C5—H5	120.0	C16—C17—H17	119.9
C6—C5—H5	120.0	C19—C18—C17	120.64 (17)
C7—C6—C5	120.45 (19)	C19—C18—H18	119.7

C7—C6—H6	119.8	C17—C18—H18	119.7
C5—C6—H6	119.8	C18—C19—C20	119.46 (18)
C6—C7—C2	119.22 (18)	C18—C19—H19	120.3
C6—C7—Se	125.27 (14)	C20—C19—H19	120.3
C2—C7—Se	115.50 (15)	C19—C20—C21	120.26 (18)
C9—C8—C13	117.89 (17)	C19—C20—H20	119.9
C9—C8—Se	117.98 (14)	C21—C20—H20	119.9
C13—C8—Se	123.97 (14)	C20—C21—C16	120.73 (17)
C10—C9—C8	121.44 (18)	C20—C21—H21	119.6
C10—C9—H9	119.3	C16—C21—H21	119.6
C8—C9—H9	119.3		
C14—N1—N2—N3	-0.75 (19)	C9—C8—C13—N1	-177.53 (17)
C13—N1—N2—N3	-175.16 (14)	Se—C8—C13—N1	7.1 (3)
N1—N2—N3—C15	0.31 (19)	C14—N1—C13—C12	-133.19 (19)
C1—O1—C2—C3	2.6 (3)	N2—N1—C13—C12	40.0 (2)
C1—O1—C2—C7	-178.26 (16)	C14—N1—C13—C8	46.3 (3)
O1—C2—C3—C4	179.81 (17)	N2—N1—C13—C8	-140.54 (18)
C7—C2—C3—C4	0.7 (3)	N2—N1—C14—C15	0.86 (19)
C2—C3—C4—C5	-0.9 (3)	C13—N1—C14—C15	174.46 (17)
C3—C4—C5—C6	0.4 (3)	N2—N3—C15—C14	0.2 (2)
C4—C5—C6—C7	0.2 (3)	N2—N3—C15—C16	179.29 (16)
C5—C6—C7—C2	-0.5 (3)	N1—C14—C15—N3	-0.65 (19)
C5—C6—C7—Se	-179.52 (14)	N1—C14—C15—C16	-179.64 (17)
O1—C2—C7—C6	-179.22 (16)	N3—C15—C16—C21	162.95 (17)
C3—C2—C7—C6	0.0 (3)	C14—C15—C16—C21	-18.2 (3)
O1—C2—C7—Se	-0.1 (2)	N3—C15—C16—C17	-17.9 (3)
C3—C2—C7—Se	179.15 (13)	C14—C15—C16—C17	161.00 (18)
C13—C8—C9—C10	-2.4 (3)	C21—C16—C17—C18	-0.2 (3)
Se—C8—C9—C10	173.23 (15)	C15—C16—C17—C18	-179.43 (17)
C8—C9—C10—C11	0.9 (3)	C16—C17—C18—C19	0.0 (3)
C9—C10—C11—C12	1.1 (3)	C17—C18—C19—C20	0.2 (3)
C10—C11—C12—C13	-1.5 (3)	C18—C19—C20—C21	-0.2 (3)
C11—C12—C13—C8	0.0 (3)	C19—C20—C21—C16	0.0 (3)
C11—C12—C13—N1	179.48 (17)	C17—C16—C21—C20	0.2 (3)
C9—C8—C13—C12	1.9 (3)	C15—C16—C21—C20	179.44 (17)
Se—C8—C13—C12	-173.39 (15)		

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C1—C6 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C18—H18 \cdots O1 ⁱ	0.95	2.54	3.472 (2)	165
C14—H14 \cdots N3 ⁱⁱ	0.95	2.58	3.520 (2)	170
C10—H10 \cdots Cg1 ⁱⁱⁱ	0.95	2.82	3.630 (2)	144

Symmetry codes: (i) $-x+2, -y+2, -z+1$; (ii) $x-1, y, z$; (iii) $-x+1, -y+1, -z+2$.