

# Crystal structure of poly[[2,2'-bipyridine)manganese(II)]-di- $\mu$ -thiocyanato]

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Received 1 November 2014; accepted 7 November 2014

Edited by M. Weil, Vienna University of Technology, Austria

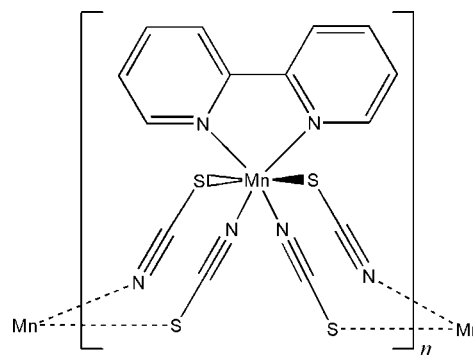
In the crystal structure of the polymeric title compound,  $[\text{Mn}(\text{NCS})_2(\text{C}_{10}\text{H}_8\text{N}_2)]_n$ , the  $\text{Mn}^{\text{II}}$  cations are coordinated by one chelating 2,2'-bipyridine ligand and four thiocyanate anions (two *N*- and two *S*-coordinating), forming a distorted  $[\text{MnN}_4\text{S}_2]$  octahedron. The asymmetric unit consists of one manganese cation located on a twofold rotation axis and half of a 2,2'-bipyridine ligand, the other half being generated by the same twofold rotation axis, as well as one thiocyanate anion in a general position. The  $\text{Mn}^{\text{II}}$  cations are linked by two pairs of  $\mu_{1,3}$ -bridging thiocyanate ligands into chains along the *c* axis; because the N atoms of the 2,2'-bipyridine ligands, as well as the N and the S atoms of the thiocyanate anions, are each *cis*-coordinating, these chains show a zigzag arrangement.

**Keywords:** crystal structure; coordination polymer; Mn in octahedral coordination; bipyridine ligand.

**CCDC reference:** 1033178

## 1. Related literature

For the magnetic properties of the title compound, see: Dockum *et al.* (1983). For general background to this work, see: Näther *et al.* (2013).



## 2. Experimental

### 2.1. Crystal data

$[\text{Mn}(\text{NCS})_2(\text{C}_{10}\text{H}_8\text{N}_2)]$	$V = 1317.51 (17) \text{ \AA}^3$
$M_r = 327.28$	$Z = 4$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
$a = 7.6158 (5) \text{ \AA}$	$\mu = 1.31 \text{ mm}^{-1}$
$b = 16.2007 (14) \text{ \AA}$	$T = 180 \text{ K}$
$c = 10.6784 (7) \text{ \AA}$	$0.24 \times 0.18 \times 0.11 \text{ mm}$
$\beta = 90.129 (8)^\circ$	

### 2.2. Data collection

STOE IPDS-1 diffractometer	5163 measured reflections
Absorption correction: numerical ( <i>X-SHAPE</i> and <i>X-RED32</i> ; Stoe & Cie, 2008)	1424 independent reflections
$T_{\min} = 0.714$ , $T_{\max} = 0.825$	1208 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.029$

### 2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.026$	87 parameters
$wR(F^2) = 0.065$	H-atom parameters constrained
$S = 1.10$	$\Delta\rho_{\max} = 0.22 \text{ e \AA}^{-3}$
1424 reflections	$\Delta\rho_{\min} = -0.35 \text{ e \AA}^{-3}$

**Table 1**

Selected bond lengths ( $\text{\AA}$ ).

Mn1—N1 <sup>i</sup>	2.1318 (13)	Mn1—S1	2.8138 (5)
Mn1—N10	2.2433 (12)		

Symmetry code: (i)  $-x, -y + 1, -z + 1$ .

Data collection: *X-AREA* (Stoe & Cie, 2008); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

## Acknowledgements

We gratefully acknowledge financial support by the DFG (project number NA 720/5-1) and the State of Schleswig-Holstein. We thank Professor Dr Wolfgang Bensch for access to his experimental facilities.

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Supporting information for this paper is available from the IUCr electronic archives (Reference: WM5088).

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### References

Brandenburg, K. (1999). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.

Dockum, B. W., Eisman, G. A., Witten, E. H. & Reiff, W. M. (1983). *Inorg. Chem.* **22**, 150–156.

Näther, C., Wöhlert, S., Boeckmann, J., Wriedt, M. & Jess, I. (2013). *Z. Anorg. Allg. Chem.* **639**, 2696–2714.

Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

Stoe & Cie (2008). *X-AREA, X-RED32 and X-SHAPE*. Stoe & Cie, Darmstadt, Germany.

Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

## supporting information

*Acta Cryst.* (2014). E70, m401–m402 [doi:10.1107/S1600536814024490]

## Crystal structure of poly[[*(2,2'*-bipyridine)manganese(II)]-di- $\mu$ -thiocyanato]

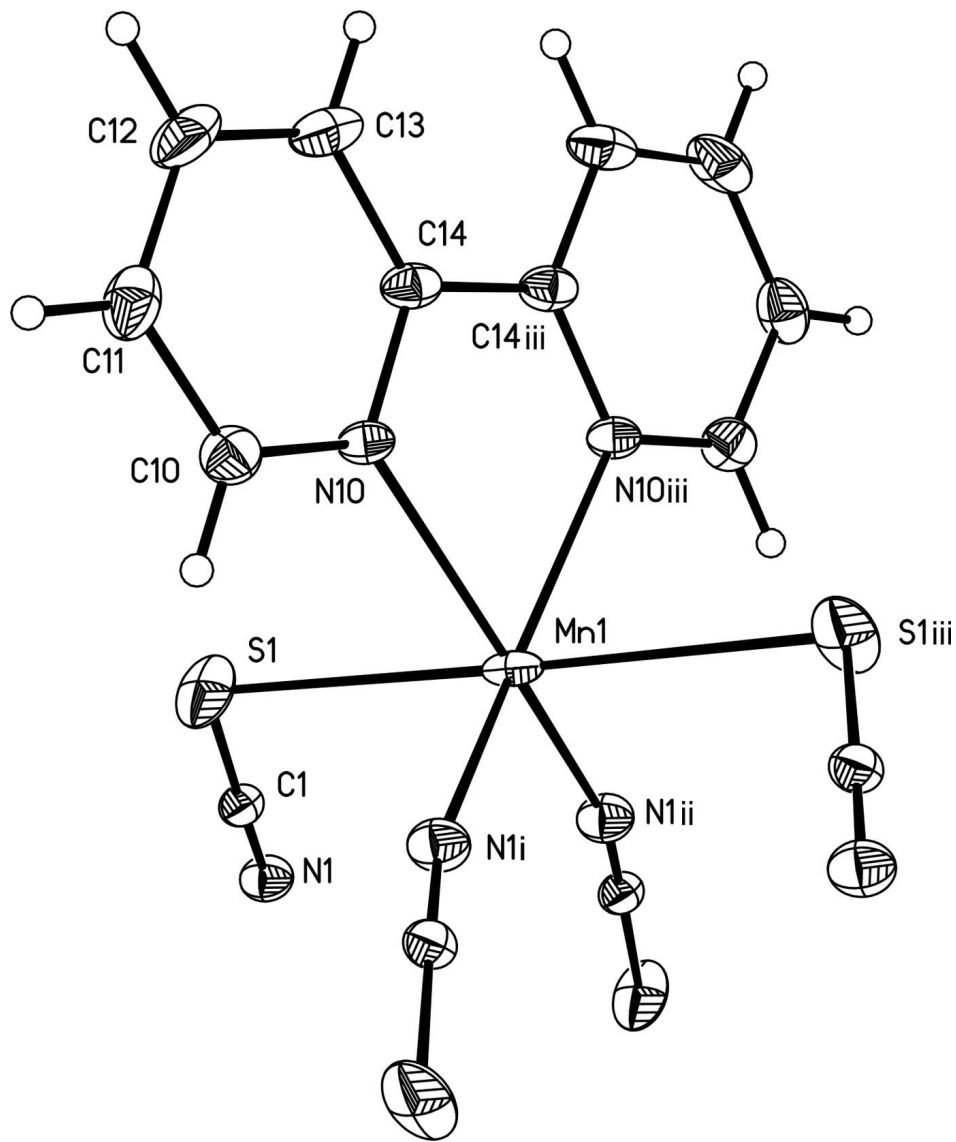
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### S1. Synthesis and crystallization

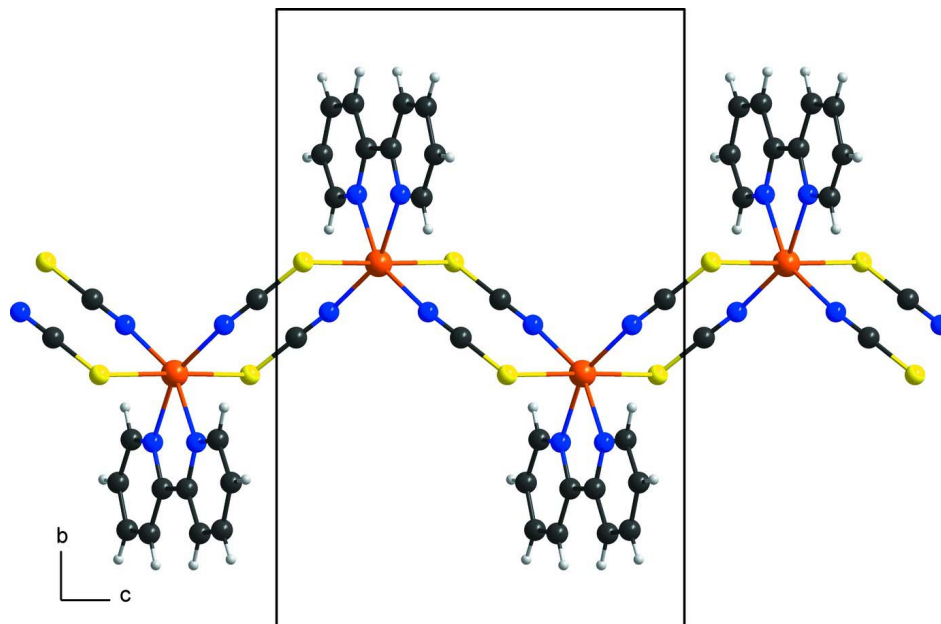
MnSO<sub>4</sub>·H<sub>2</sub>O was purchased from Merck and 2,2'-bipyridine and Ba(NCS)<sub>2</sub>·3 H<sub>2</sub>O were purchased from Alfa Aesar. Mn(NCS)<sub>2</sub> was synthesized by stirring 17.97 g (58.44 mmol) Ba(NCS)<sub>2</sub>·3H<sub>2</sub>O and 9.88 g (58.44 mmol) MnSO<sub>4</sub>·H<sub>2</sub>O in 300 ml water at RT for three hours. The white precipitate of BaSO<sub>4</sub> was filtered off and the solvent removed with a rotary evaporator. The homogeneity of the product was investigated by X-ray powder diffraction and elemental analysis. The title compound was prepared by the reaction of (0.4 mmol) 70.0 mg Mn(NCS)<sub>2</sub> and (0.05 mmol) 7.0 mg 2,2'-bipyridine in 1.0 ml acetonitrile at RT. After few days, yellow block-shaped crystals of the title compound were obtained.

### S2. Refinement

The H atoms were positioned with idealized geometry and were refined with C—H = 0.93 Å and  $U_{\text{eq}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  using a riding model.

**Figure 1**

The coordination of the Mn<sup>II</sup> atom in the title compound with atom labelling and displacement ellipsoids drawn at the 50% probability level. [Symmetry codes: i)  $x,-y+1,z-1/2$ ; ii)  $-x,-y+1,-z+1$ ; iii)  $-x,y,-z+1/2$ .]

**Figure 2**

The polymeric arrangement of the chains in the crystal structure of the title compound in a view along the *a* axis. Colour code: Mn orange; N blue; S yellow; C black; H white.

### Poly[[*(2,2'*-bipyridine)manganese(II)]-di- $\mu$ -thiocyanato]

#### Crystal data

[Mn(NCS)<sub>2</sub>(C<sub>10</sub>H<sub>8</sub>N<sub>2</sub>)]

*M<sub>r</sub>* = 327.28

Monoclinic, *C2/c*

Hall symbol: -C 2yc

*a* = 7.6158 (5) Å

*b* = 16.2007 (14) Å

*c* = 10.6784 (7) Å

$\beta$  = 90.129 (8)°

*V* = 1317.51 (17) Å<sup>3</sup>

*Z* = 4

*F*(000) = 660

*D<sub>x</sub>* = 1.650 Mg m<sup>-3</sup>

Mo *K*α radiation,  $\lambda$  = 0.71073 Å

Cell parameters from 5163 reflections

$\theta$  = 3.2–27.0°

$\mu$  = 1.31 mm<sup>-1</sup>

*T* = 180 K

Block, yellow

0.24 × 0.18 × 0.11 mm

#### Data collection

STOE IPDS-1

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

phi scans

Absorption correction: numerical

(*X-SHAPE* and *X-RED32*; Stoe & Cie, 2008)

*T<sub>min</sub>* = 0.714, *T<sub>max</sub>* = 0.825

5163 measured reflections

1424 independent reflections

1208 reflections with *I* > 2σ(*I*)

*R<sub>int</sub>* = 0.029

$\theta_{\max}$  = 27.0°,  $\theta_{\min}$  = 3.2°

*h* = -9→9

*k* = -20→20

*l* = -13→13

#### Refinement

Refinement on *F*<sup>2</sup>

Least-squares matrix: full

*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.026

*wR*(*F*<sup>2</sup>) = 0.065

*S* = 1.10

1424 reflections

87 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods  
Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.041P)^2 + 0.0505P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.22 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.35 \text{ e } \text{\AA}^{-3}$

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Mn1	0.0000	0.589163 (17)	0.2500	0.01511 (12)
S1	0.26426 (6)	0.59114 (3)	0.43456 (4)	0.02879 (14)
C1	0.1959 (2)	0.53098 (9)	0.54809 (13)	0.0157 (3)
N1	0.14566 (19)	0.48959 (8)	0.62863 (13)	0.0191 (3)
N10	0.15848 (17)	0.70040 (7)	0.19663 (11)	0.0150 (3)
C10	0.3171 (2)	0.69584 (10)	0.14411 (14)	0.0205 (3)
H10	0.3650	0.6440	0.1287	0.025*
C11	0.4133 (2)	0.76537 (11)	0.11149 (15)	0.0246 (4)
H11	0.5240	0.7605	0.0756	0.029*
C12	0.3400 (2)	0.84198 (11)	0.13384 (16)	0.0277 (4)
H12	0.4008	0.8897	0.1122	0.033*
C13	0.1770 (3)	0.84748 (9)	0.18819 (16)	0.0256 (4)
H13	0.1267	0.8988	0.2037	0.031*
C14	0.0878 (2)	0.77516 (9)	0.21990 (13)	0.0175 (3)

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Mn1	0.0203 (2)	0.00785 (16)	0.01723 (18)	0.000	0.00916 (12)	0.000
S1	0.0261 (3)	0.0358 (3)	0.0244 (2)	-0.01473 (18)	0.00026 (18)	0.01435 (16)
C1	0.0150 (8)	0.0134 (6)	0.0186 (7)	-0.0014 (6)	0.0007 (5)	-0.0002 (5)
N1	0.0218 (7)	0.0167 (6)	0.0187 (6)	-0.0016 (5)	0.0031 (5)	0.0031 (5)
N10	0.0193 (7)	0.0116 (5)	0.0141 (6)	-0.0013 (5)	0.0024 (5)	0.0023 (4)
C10	0.0215 (9)	0.0203 (7)	0.0195 (7)	-0.0010 (6)	0.0016 (6)	0.0026 (6)
C11	0.0196 (9)	0.0319 (8)	0.0222 (8)	-0.0067 (7)	-0.0002 (6)	0.0060 (7)
C12	0.0319 (11)	0.0242 (8)	0.0269 (9)	-0.0145 (7)	-0.0025 (7)	0.0075 (6)
C13	0.0352 (11)	0.0140 (7)	0.0275 (8)	-0.0055 (7)	-0.0005 (7)	0.0002 (6)
C14	0.0254 (9)	0.0133 (7)	0.0137 (6)	-0.0011 (6)	-0.0009 (6)	0.0018 (5)

*Geometric parameters (Å, °)*

Mn1—N1 <sup>i</sup>	2.1318 (13)	N10—C14	1.3485 (18)
Mn1—N1 <sup>ii</sup>	2.1318 (13)	C10—C11	1.389 (2)
Mn1—N10 <sup>iii</sup>	2.2433 (12)	C10—H10	0.9300
Mn1—N10	2.2433 (12)	C11—C12	1.382 (3)
Mn1—S1 <sup>iii</sup>	2.8138 (5)	C11—H11	0.9300
Mn1—S1	2.8138 (5)	C12—C13	1.375 (3)
S1—C1	1.6411 (15)	C12—H12	0.9300
C1—N1	1.156 (2)	C13—C14	1.396 (2)
N1—Mn1 <sup>ii</sup>	2.1318 (13)	C13—H13	0.9300
N10—C10	1.335 (2)	C14—C14 <sup>iii</sup>	1.486 (3)
N1 <sup>i</sup> —Mn1—N1 <sup>ii</sup>	106.48 (7)	C10—N10—C14	119.27 (13)
N1 <sup>i</sup> —Mn1—N10 <sup>iii</sup>	156.38 (5)	C10—N10—Mn1	123.37 (11)
N1 <sup>ii</sup> —Mn1—N10 <sup>iii</sup>	92.60 (5)	C14—N10—Mn1	117.36 (10)
N1 <sup>i</sup> —Mn1—N10	92.60 (5)	N10—C10—C11	122.62 (16)
N1 <sup>ii</sup> —Mn1—N10	156.38 (5)	N10—C10—H10	118.7
N10 <sup>iii</sup> —Mn1—N10	73.10 (7)	C11—C10—H10	118.7
N1 <sup>i</sup> —Mn1—S1 <sup>iii</sup>	87.33 (4)	C12—C11—C10	118.14 (16)
N1 <sup>ii</sup> —Mn1—S1 <sup>iii</sup>	93.46 (4)	C12—C11—H11	120.9
N10 <sup>iii</sup> —Mn1—S1 <sup>iii</sup>	77.55 (3)	C10—C11—H11	120.9
N10—Mn1—S1 <sup>iii</sup>	101.38 (3)	C13—C12—C11	119.78 (15)
N1 <sup>i</sup> —Mn1—S1	93.46 (4)	C13—C12—H12	120.1
N1 <sup>ii</sup> —Mn1—S1	87.33 (4)	C11—C12—H12	120.1
N10 <sup>iii</sup> —Mn1—S1	101.38 (3)	C12—C13—C14	119.23 (15)
N10—Mn1—S1	77.55 (3)	C12—C13—H13	120.4
S1 <sup>iii</sup> —Mn1—S1	178.69 (2)	C14—C13—H13	120.4
C1—S1—Mn1	106.45 (6)	N10—C14—C13	120.96 (15)
N1—C1—S1	178.83 (15)	N10—C14—C14 <sup>iii</sup>	116.09 (8)
C1—N1—Mn1 <sup>ii</sup>	166.59 (14)	C13—C14—C14 <sup>iii</sup>	122.95 (10)

Symmetry codes: (i)  $x, -y+1, z-1/2$ ; (ii)  $-x, -y+1, -z+1$ ; (iii)  $-x, y, -z+1/2$ .