

Crystal structure of 2,5-dimethyl-3-(2-methylphenylsulfonyl)-1-benzofuran

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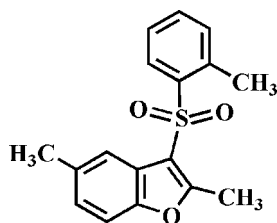
The title compound, C₁₇H₁₆O₃S, crystallized with two independent molecules (*A* and *B*) in the asymmetric unit. The dihedral angle between the benzofuran ring system [r.m.s. deviation of 0.013 (1) for *A* and 0.009 (1) Å for *B*] and the 2-methylphenyl ring is 83.88 (5) for *A* and 86.94 (5)° for *B*. In the crystal, the *B* molecules are linked into a chain along the *b*-axis direction by C—H...O hydrogen bonds. The *A* molecules are connected on either side of this chain by further C—H...O hydrogen bonds. These chains are linked *via* C—H...π interactions, forming sheets parallel to (100). There are also very weak π–π interactions present [centroid–centroid distance = 3.925 (11) Å] involving the 2-methylphenyl rings of neighbouring *A* and *B* molecules.

Keywords: crystal structure; benzofuran; 2-methylphenyl; sulfonyl; C—H...O hydrogen bonds; C—H...π interactions; π–π interactions.

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1. Related literature

For a related structure and background to benzofuran derivatives, see: Choi & Lee (2014). For further synthetic details, see: Choi *et al.* (1999).



2. Experimental

2.1. Crystal data

C ₁₇ H ₁₆ O ₃ S	<i>V</i> = 2932.00 (13) Å ³
<i>M_r</i> = 300.36	<i>Z</i> = 8
Monoclinic, <i>P</i> 2 ₁ / <i>c</i>	Mo <i>K</i> α radiation
<i>a</i> = 16.7338 (4) Å	<i>μ</i> = 0.23 mm ⁻¹
<i>b</i> = 8.0646 (2) Å	<i>T</i> = 173 K
<i>c</i> = 21.8195 (6) Å	0.58 × 0.37 × 0.23 mm
<i>β</i> = 95.296 (1)°	

2.2. Data collection

Bruker SMART APEXII CCD diffractometer	27259 measured reflections
Absorption correction: multi-scan (SADABS; Bruker, 2009)	6773 independent reflections
<i>T</i> _{min} = 0.879, <i>T</i> _{max} = 0.950	5313 reflections with <i>I</i> > 2σ(<i>I</i>)
	<i>R</i> _{int} = 0.042

2.3. Refinement

<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.043	385 parameters
<i>wR</i> (<i>F</i> ²) = 0.118	H-atom parameters constrained
<i>S</i> = 1.04	Δρ _{max} = 0.34 e Å ⁻³
6773 reflections	Δρ _{min} = -0.44 e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

*Cg*1 and *Cg*2 are the centroids of benzene rings C2–C7 and C19–C24, respectively.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C23—H23...O5 ⁱ	0.95	2.52	3.431 (3)	162
C34—H34B...O2 ⁱⁱ	0.98	2.43	3.137 (2)	129
C31—H31... <i>Cg</i> 1 ⁱⁱ	0.95	2.93	3.773 (3)	148
C14—H14... <i>Cg</i> 2 ⁱ	0.95	3.00	3.876 (3)	154

Symmetry codes: (i) $-x, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x, y + 1, z$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012) and *DIAMOND* (Brandenburg, 1998); software used to prepare material for publication: *SHELXL97*.

Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: SU5004).

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supporting information

Acta Cryst. (2014). E70, o1181–o1182 [doi:10.1107/S1600536814022788]

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S1. Comment

As part of our continuing program on benzofuran derivatives (Choi & Lee, 2014), we report herein on the crystal structure of the title compound which crystallizes with two symmetrically independent molecules (A and B) in the asymmetric unit.

In the title molecules (A and B; Fig. 1), the benzofuran units are essentially planar, with mean deviations of 0.013 (1) and 0.009 (1) Å for A and B, respectively, from the mean planes defined by the nine constituent non-H atoms. The dihedral angles formed by the benzofuran ring systems and the 2-methylphenyl rings are 83.88 (5) and 86.94 (5)° for A and B, respectively.

In the crystal, the B molecules are linked into a chain along the *b* axis direction by C—H···O hydrogen bonds (Table 1 and Fig. 2). The A molecules are connected on either side of this chain by further C—H···O hydrogen bonds (Table 1). These chains are linked via C—H··· π interactions forming sheets parallel to (100); see Table 1. There are also very weak π ··· π interactions present (Fig. 3) involving 2-methylphenyl rings of neighbouring A and B molecules [$\text{Cg3}\cdots\text{Cg7}^i = 3.925$ (11) Å, where Cg3 and Cg7 are the centroids of rings C11–C16 and C28–C33; symmetry code: (i) = x, -y+3/2, z+1/2].

S2. Experimental

The starting material 2,5-dimethyl-3-(2-methylphenylsulfonyl)-1-benzofuran was prepared by the literature method (Choi *et al.*, 1999). 3-chloroperoxybenzoic acid (77%, 515 mg, 2.3 mmol) was added in small portions to a stirred solution of 2,5-dimethyl-3-(2-methylphenylsulfonyl)-1-benzofuran (295 mg, 1.1 mmol) in dichloromethane (35 ml) at 273 K. After being stirred at room temperature for 8h, the mixture was washed with saturated sodium bicarbonate solution (2 × 10 ml) and the organic layer was separated, dried over magnesium sulfate, filtered and concentrated at reduced pressure. The residue was purified by column chromatography (hexane-ethyl acetate, 4:1 v/v) to afford the title compound as a colourless solid [yield 71%; 222 mg; m.p. 385–386 K; $R_f = 0.54$ (hexane-ethyl acetate, 4:1 v/v)]. Single crystals suitable for X-ray diffraction were prepared by slow evaporation of a solution of the title compound (23 mg) in ethyl acetate (20 ml) at room temperature.

S3. Refinement

All H atoms were positioned geometrically and refined using a riding model, with C—H = 0.95 Å for aryl, and 0.98 Å for methyl H atoms and with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H atoms and = $1.2U_{\text{eq}}(\text{C})$ for other H atoms.

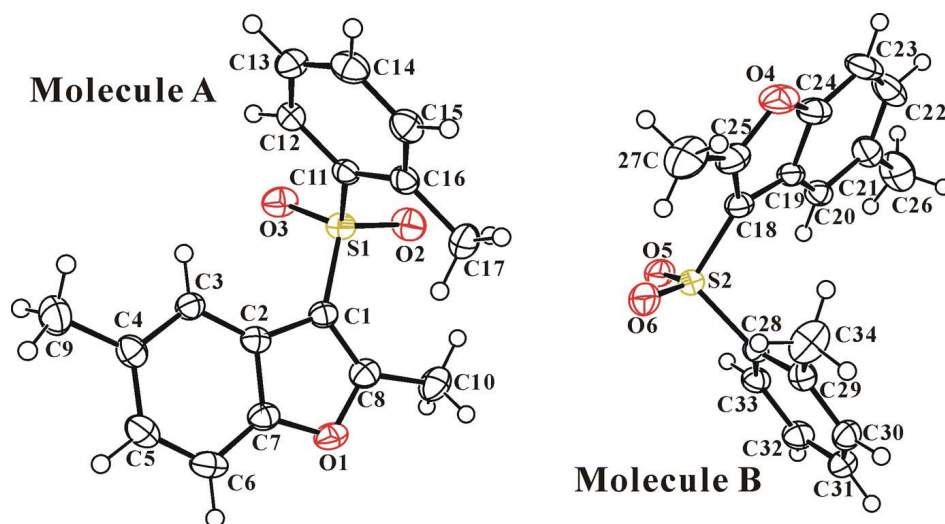


Figure 1

The molecular structure of the two independent molecules (A and B) of the title compound, with atom labelling. Displacement ellipsoids are drawn at the 50% probability level.

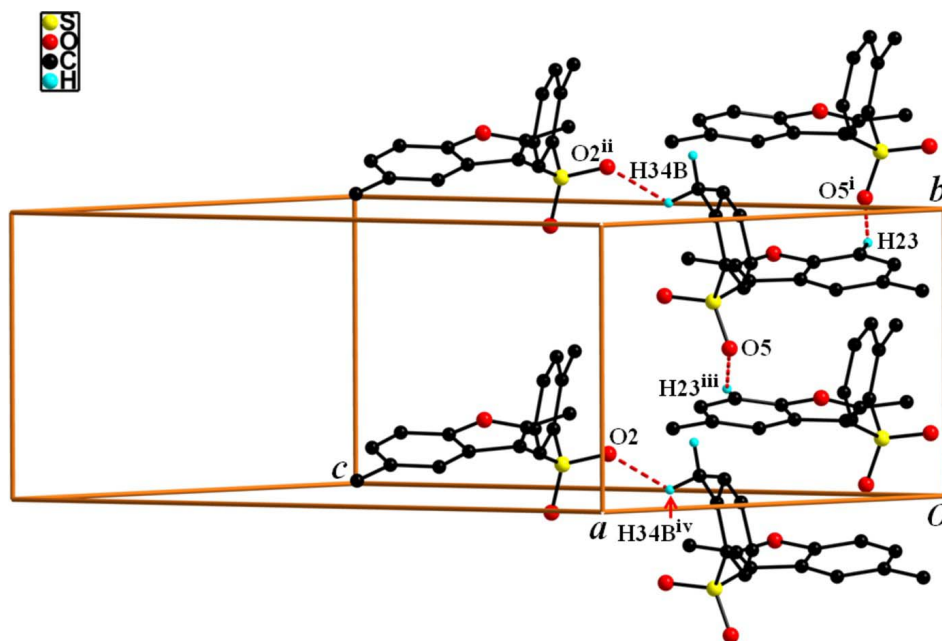
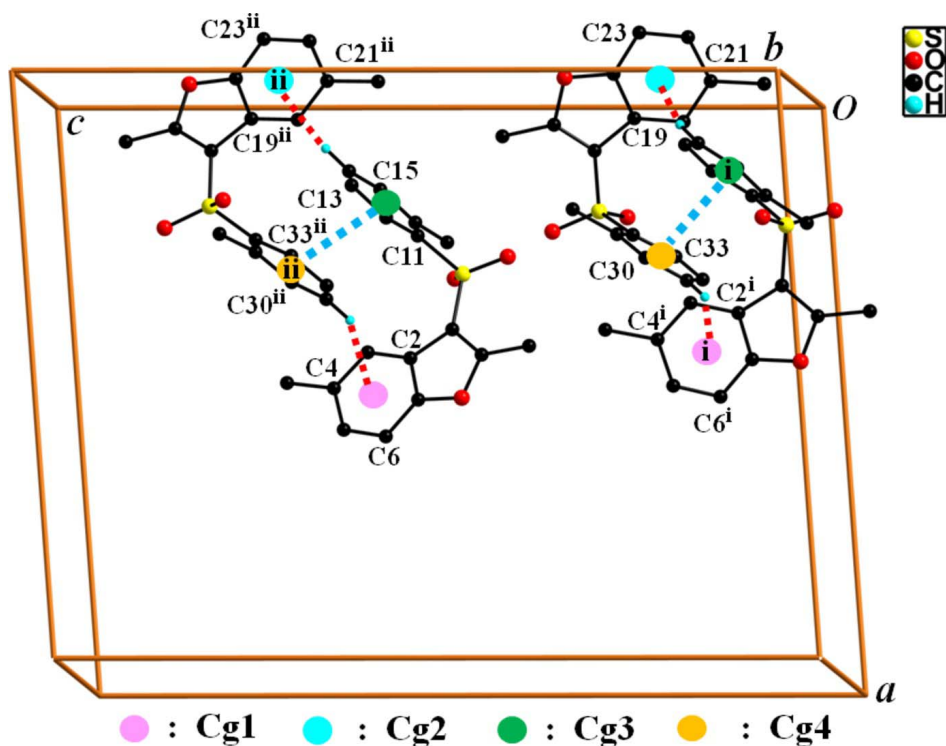


Figure 2

A view of the C—H...O interactions (dashed lines) in the crystal structure of the title compound - see Table 1 for details. H atoms not involved in hydrogen bonding have been omitted for clarity [Symmetry codes: (i) $-x, y + 1/2, -z + 1/2$; (ii) $x, y + 1, z$; (iii) $-x, y - 1/2, -z + 1/2$; (iv) $x, y - 1, z$].

**Figure 3**

A view of the C—H... π and π - π interactions (dashed lines) in the crystal structure of the title compound - see Table 1 for details. H atoms non-participating in hydrogen-bonding have been omitted for clarity [Symmetry codes: (i) $x, -y + 3/2, z - 1/2$; (ii) $x, -y + 3/2, z + 1/2$].

2,5-Dimethyl-3-(2-methylphenylsulfonyl)-1-benzofuran

Crystal data

$C_{17}H_{16}O_3S$
 $M_r = 300.36$
 Monoclinic, $P2_1/c$
 Hall symbol: -P 2ybc
 $a = 16.7338(4) \text{ \AA}$
 $b = 8.0646(2) \text{ \AA}$
 $c = 21.8195(6) \text{ \AA}$
 $\beta = 95.296(1)^\circ$
 $V = 2932.00(13) \text{ \AA}^3$
 $Z = 8$

$F(000) = 1264$
 $D_x = 1.361 \text{ Mg m}^{-3}$
 Melting point = 386–385 K
 Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
 Cell parameters from 5209 reflections
 $\theta = 2.3\text{--}26.6^\circ$
 $\mu = 0.23 \text{ mm}^{-1}$
 $T = 173 \text{ K}$
 Block, colourless
 $0.58 \times 0.37 \times 0.23 \text{ mm}$

Data collection

Bruker SMART APEXII CCD
 diffractometer
 Radiation source: rotating anode
 Graphite multilayer monochromator
 Detector resolution: 10.0 pixels mm^{-1}
 φ and ω scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 2009)
 $T_{\min} = 0.879, T_{\max} = 0.950$

27259 measured reflections
 6773 independent reflections
 5313 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$
 $\theta_{\text{max}} = 27.6^\circ, \theta_{\text{min}} = 1.9^\circ$
 $h = -21 \rightarrow 21$
 $k = -9 \rightarrow 10$
 $l = -23 \rightarrow 28$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.043$ $wR(F^2) = 0.118$ $S = 1.04$

6773 reflections

385 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
map

Hydrogen site location: difference Fourier map

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.056P)^2 + 1.1004P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.34 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.44 \text{ e } \text{\AA}^{-3}$ *Special details***Experimental.** ^1H NMR (δ p.p.m., CDCl_3 , 400 Hz): 8.25 (d, $J=7.52$ Hz, 1H), 7.37-7.48 (m, 3H), 7.30 (d, $J=8.52$ Hz, 1H), 7.26 (d, $J=7.52$ Hz, 1H), 7.08 (d, $J=8.56$ Hz, 1H), 2.74 (s, 3H), 2.52 (s, 3H), 2.37 (s, 3H).**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.**Refinement.** Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)*

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.29003 (3)	0.10514 (6)	0.48076 (2)	0.02658 (12)
O3	0.28718 (8)	-0.06396 (16)	0.50124 (6)	0.0361 (3)
O1	0.51121 (7)	0.27240 (17)	0.48421 (6)	0.0336 (3)
O2	0.26309 (8)	0.14295 (19)	0.41824 (6)	0.0388 (3)
C1	0.38848 (10)	0.1730 (2)	0.49575 (8)	0.0257 (4)
C2	0.43612 (10)	0.1551 (2)	0.55426 (8)	0.0245 (4)
C3	0.42259 (10)	0.0982 (2)	0.61259 (8)	0.0261 (4)
H3	0.3718	0.0542	0.6201	0.031*
C4	0.48413 (11)	0.1065 (2)	0.65957 (9)	0.0306 (4)
C5	0.55882 (11)	0.1708 (3)	0.64695 (9)	0.0356 (4)
H5	0.6008	0.1757	0.6794	0.043*
C6	0.57388 (11)	0.2272 (3)	0.58951 (10)	0.0355 (4)
H6	0.6249	0.2694	0.5815	0.043*
C7	0.51090 (10)	0.2191 (2)	0.54429 (8)	0.0285 (4)
C8	0.43609 (11)	0.2433 (2)	0.45585 (8)	0.0302 (4)
C9	0.47128 (13)	0.0478 (3)	0.72362 (9)	0.0427 (5)
H9A	0.4165	0.0746	0.7327	0.064*
H9B	0.5097	0.1032	0.7535	0.064*
H9C	0.4794	-0.0724	0.7262	0.064*
C10	0.42443 (13)	0.2938 (3)	0.39015 (9)	0.0404 (5)
H10A	0.4554	0.2202	0.3656	0.061*
H10B	0.4428	0.4083	0.3860	0.061*
H10C	0.3674	0.2863	0.3756	0.061*

C11	0.23229 (9)	0.2235 (2)	0.52923 (8)	0.0232 (3)
C12	0.19580 (10)	0.1353 (2)	0.57375 (8)	0.0283 (4)
H12	0.2052	0.0196	0.5784	0.034*
C13	0.14589 (11)	0.2152 (3)	0.61129 (9)	0.0337 (4)
H13	0.1211	0.1553	0.6419	0.040*
C14	0.13262 (11)	0.3823 (3)	0.60383 (9)	0.0351 (4)
H14	0.0978	0.4381	0.6290	0.042*
C15	0.16958 (11)	0.4699 (2)	0.55985 (9)	0.0330 (4)
H15	0.1597	0.5856	0.5557	0.040*
C16	0.22088 (10)	0.3944 (2)	0.52148 (8)	0.0265 (4)
C17	0.26127 (12)	0.4981 (3)	0.47580 (9)	0.0373 (5)
H17A	0.2446	0.6140	0.4790	0.056*
H17B	0.2458	0.4575	0.4341	0.056*
H17C	0.3196	0.4901	0.4846	0.056*
S2	0.22038 (3)	0.67798 (6)	0.26437 (2)	0.02693 (12)
O4	0.00302 (8)	0.83199 (19)	0.28806 (7)	0.0422 (4)
O5	0.22026 (8)	0.51558 (16)	0.23717 (7)	0.0360 (3)
O6	0.25599 (8)	0.69703 (19)	0.32615 (6)	0.0404 (4)
C29	0.27854 (10)	0.9817 (2)	0.22581 (8)	0.0272 (4)
C18	0.12153 (10)	0.7450 (2)	0.26076 (8)	0.0271 (4)
C19	0.06680 (10)	0.7498 (2)	0.20564 (8)	0.0263 (4)
C20	0.07067 (10)	0.7165 (2)	0.14330 (8)	0.0282 (4)
H20	0.1190	0.6767	0.1289	0.034*
C21	0.00326 (11)	0.7420 (2)	0.10269 (10)	0.0365 (5)
C22	-0.06806 (12)	0.7977 (3)	0.12514 (12)	0.0471 (6)
H22	-0.1141	0.8136	0.0969	0.057*
C23	-0.07384 (12)	0.8299 (3)	0.18648 (12)	0.0454 (6)
H23	-0.1225	0.8669	0.2012	0.054*
C24	-0.00521 (11)	0.8056 (2)	0.22530 (10)	0.0352 (4)
C25	0.08057 (12)	0.7949 (3)	0.30844 (9)	0.0365 (5)
C26	0.00614 (15)	0.7143 (3)	0.03466 (10)	0.0534 (6)
H26A	0.0072	0.8216	0.0137	0.080*
H26B	-0.0414	0.6520	0.0184	0.080*
H26C	0.0545	0.6514	0.0276	0.080*
C27	0.10147 (16)	0.8167 (3)	0.37561 (10)	0.0534 (6)
H27A	0.0772	0.7273	0.3980	0.080*
H27B	0.0811	0.9238	0.3886	0.080*
H27C	0.1599	0.8135	0.3846	0.080*
C28	0.26882 (10)	0.8114 (2)	0.21492 (8)	0.0235 (4)
C30	0.32432 (11)	1.0681 (2)	0.18626 (9)	0.0321 (4)
H30	0.3325	1.1837	0.1924	0.038*
C31	0.35822 (11)	0.9915 (3)	0.13847 (9)	0.0339 (4)
H31	0.3903	1.0539	0.1131	0.041*
C32	0.34591 (11)	0.8247 (3)	0.12726 (9)	0.0345 (4)
H32	0.3680	0.7725	0.0936	0.041*
C33	0.30100 (10)	0.7347 (2)	0.16561 (8)	0.0290 (4)
H33	0.2920	0.6197	0.1583	0.035*
C34	0.24201 (15)	1.0720 (3)	0.27628 (10)	0.0453 (5)

H34A	0.2581	1.1889	0.2761	0.068*
H34B	0.2606	1.0223	0.3160	0.068*
H34C	0.1834	1.0641	0.2698	0.068*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0259 (2)	0.0283 (2)	0.0256 (2)	-0.00190 (17)	0.00290 (17)	-0.00472 (18)
O3	0.0361 (7)	0.0248 (7)	0.0485 (8)	-0.0034 (6)	0.0094 (6)	-0.0059 (6)
O1	0.0275 (6)	0.0383 (8)	0.0366 (7)	-0.0025 (6)	0.0121 (6)	0.0011 (6)
O2	0.0379 (8)	0.0539 (9)	0.0241 (7)	-0.0015 (7)	0.0002 (6)	-0.0076 (6)
C1	0.0246 (8)	0.0262 (9)	0.0268 (9)	0.0009 (7)	0.0055 (7)	-0.0019 (7)
C2	0.0228 (8)	0.0204 (8)	0.0306 (9)	0.0022 (7)	0.0049 (7)	-0.0040 (7)
C3	0.0249 (8)	0.0248 (9)	0.0288 (9)	-0.0013 (7)	0.0041 (7)	0.0009 (7)
C4	0.0302 (9)	0.0274 (9)	0.0335 (10)	0.0026 (7)	-0.0015 (8)	-0.0010 (8)
C5	0.0281 (9)	0.0365 (11)	0.0406 (11)	0.0008 (8)	-0.0059 (8)	-0.0020 (9)
C6	0.0216 (9)	0.0370 (11)	0.0485 (12)	-0.0022 (8)	0.0055 (8)	-0.0028 (9)
C7	0.0257 (9)	0.0272 (9)	0.0337 (10)	0.0018 (7)	0.0085 (7)	-0.0020 (8)
C8	0.0308 (9)	0.0299 (10)	0.0309 (10)	0.0026 (8)	0.0086 (8)	-0.0025 (8)
C9	0.0441 (12)	0.0477 (13)	0.0345 (11)	-0.0005 (10)	-0.0063 (9)	0.0054 (10)
C10	0.0471 (12)	0.0446 (12)	0.0314 (11)	-0.0004 (10)	0.0138 (9)	0.0041 (9)
C11	0.0192 (8)	0.0286 (9)	0.0216 (8)	0.0003 (7)	0.0001 (6)	-0.0013 (7)
C12	0.0248 (9)	0.0302 (10)	0.0294 (9)	0.0003 (7)	0.0006 (7)	0.0031 (8)
C13	0.0291 (9)	0.0438 (12)	0.0292 (10)	-0.0006 (8)	0.0076 (8)	0.0038 (8)
C14	0.0286 (10)	0.0431 (12)	0.0341 (10)	0.0044 (8)	0.0048 (8)	-0.0092 (9)
C15	0.0308 (9)	0.0290 (10)	0.0386 (11)	0.0035 (8)	0.0006 (8)	-0.0033 (8)
C16	0.0235 (8)	0.0279 (9)	0.0275 (9)	-0.0015 (7)	-0.0018 (7)	-0.0006 (7)
C17	0.0418 (11)	0.0302 (10)	0.0401 (11)	-0.0007 (8)	0.0053 (9)	0.0076 (9)
S2	0.0236 (2)	0.0282 (2)	0.0292 (2)	-0.00071 (17)	0.00328 (17)	0.00437 (18)
O4	0.0341 (8)	0.0445 (9)	0.0512 (9)	0.0028 (6)	0.0212 (7)	-0.0050 (7)
O5	0.0304 (7)	0.0246 (7)	0.0543 (9)	0.0007 (5)	0.0106 (6)	0.0034 (6)
O6	0.0377 (8)	0.0534 (9)	0.0292 (7)	-0.0056 (7)	-0.0015 (6)	0.0109 (6)
C29	0.0284 (9)	0.0278 (9)	0.0249 (9)	-0.0003 (7)	-0.0010 (7)	-0.0017 (7)
C18	0.0247 (8)	0.0273 (9)	0.0301 (9)	-0.0007 (7)	0.0074 (7)	-0.0001 (7)
C19	0.0202 (8)	0.0223 (9)	0.0370 (10)	-0.0015 (7)	0.0059 (7)	-0.0001 (7)
C20	0.0236 (8)	0.0244 (9)	0.0366 (10)	-0.0011 (7)	0.0020 (7)	-0.0027 (8)
C21	0.0319 (10)	0.0284 (10)	0.0473 (12)	-0.0030 (8)	-0.0069 (9)	-0.0045 (9)
C22	0.0265 (10)	0.0380 (12)	0.0736 (16)	0.0032 (9)	-0.0128 (10)	-0.0057 (11)
C23	0.0207 (9)	0.0381 (12)	0.0781 (17)	0.0032 (8)	0.0084 (10)	-0.0052 (11)
C24	0.0269 (9)	0.0319 (10)	0.0483 (12)	-0.0003 (8)	0.0118 (9)	-0.0051 (9)
C25	0.0379 (11)	0.0358 (11)	0.0382 (11)	-0.0013 (9)	0.0157 (9)	0.0000 (9)
C26	0.0538 (14)	0.0554 (15)	0.0465 (13)	0.0008 (12)	-0.0190 (11)	-0.0091 (11)
C27	0.0671 (16)	0.0598 (16)	0.0365 (12)	0.0019 (13)	0.0226 (11)	-0.0050 (11)
C28	0.0190 (8)	0.0277 (9)	0.0234 (8)	-0.0006 (7)	0.0003 (6)	0.0015 (7)
C30	0.0334 (10)	0.0281 (10)	0.0339 (10)	-0.0045 (8)	-0.0009 (8)	0.0032 (8)
C31	0.0286 (9)	0.0390 (11)	0.0344 (10)	-0.0030 (8)	0.0050 (8)	0.0095 (8)
C32	0.0306 (10)	0.0407 (11)	0.0336 (10)	0.0030 (8)	0.0108 (8)	-0.0004 (9)
C33	0.0247 (9)	0.0280 (9)	0.0348 (10)	0.0019 (7)	0.0061 (7)	-0.0022 (8)

C34	0.0689 (15)	0.0323 (11)	0.0366 (11)	-0.0067 (10)	0.0157 (11)	-0.0113 (9)
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Geometric parameters (Å, °)

S1—O2	1.4295 (14)	S2—O6	1.4309 (14)
S1—O3	1.4373 (14)	S2—O5	1.4378 (14)
S1—C1	1.7379 (18)	S2—C18	1.7351 (18)
S1—C11	1.7754 (17)	S2—C28	1.7719 (17)
O1—C8	1.369 (2)	O4—C25	1.366 (2)
O1—C7	1.380 (2)	O4—C24	1.380 (2)
C1—C8	1.357 (2)	C29—C30	1.392 (3)
C1—C2	1.449 (2)	C29—C28	1.401 (2)
C2—C7	1.389 (2)	C29—C34	1.497 (3)
C2—C3	1.391 (2)	C18—C25	1.358 (3)
C3—C4	1.386 (2)	C18—C19	1.443 (2)
C3—H3	0.9500	C19—C24	1.391 (2)
C4—C5	1.404 (3)	C19—C20	1.394 (2)
C4—C9	1.510 (3)	C20—C21	1.384 (2)
C5—C6	1.378 (3)	C20—H20	0.9500
C5—H5	0.9500	C21—C22	1.405 (3)
C6—C7	1.377 (3)	C21—C26	1.506 (3)
C6—H6	0.9500	C22—C23	1.376 (3)
C8—C10	1.486 (3)	C22—H22	0.9500
C9—H9A	0.9800	C23—C24	1.377 (3)
C9—H9B	0.9800	C23—H23	0.9500
C9—H9C	0.9800	C25—C27	1.485 (3)
C10—H10A	0.9800	C26—H26A	0.9800
C10—H10B	0.9800	C26—H26B	0.9800
C10—H10C	0.9800	C26—H26C	0.9800
C11—C12	1.390 (2)	C27—H27A	0.9800
C11—C16	1.400 (3)	C27—H27B	0.9800
C12—C13	1.382 (3)	C27—H27C	0.9800
C12—H12	0.9500	C28—C33	1.392 (2)
C13—C14	1.373 (3)	C30—C31	1.378 (3)
C13—H13	0.9500	C30—H30	0.9500
C14—C15	1.383 (3)	C31—C32	1.379 (3)
C14—H14	0.9500	C31—H31	0.9500
C15—C16	1.393 (3)	C32—C33	1.382 (3)
C15—H15	0.9500	C32—H32	0.9500
C16—C17	1.508 (3)	C33—H33	0.9500
C17—H17A	0.9800	C34—H34A	0.9800
C17—H17B	0.9800	C34—H34B	0.9800
C17—H17C	0.9800	C34—H34C	0.9800
O2—S1—O3	118.85 (9)	O6—S2—O5	118.11 (9)
O2—S1—C1	108.81 (9)	O6—S2—C18	108.77 (9)
O3—S1—C1	107.34 (8)	O5—S2—C18	107.46 (8)
O2—S1—C11	108.22 (8)	O6—S2—C28	109.53 (8)

O3—S1—C11	106.82 (8)	O5—S2—C28	106.55 (8)
C1—S1—C11	106.09 (8)	C18—S2—C28	105.72 (8)
C8—O1—C7	106.99 (13)	C25—O4—C24	107.11 (14)
C8—C1—C2	107.39 (15)	C30—C29—C28	116.58 (17)
C8—C1—S1	127.81 (14)	C30—C29—C34	119.84 (17)
C2—C1—S1	124.73 (13)	C28—C29—C34	123.57 (17)
C7—C2—C3	119.39 (16)	C25—C18—C19	107.70 (16)
C7—C2—C1	104.81 (16)	C25—C18—S2	127.13 (15)
C3—C2—C1	135.76 (16)	C19—C18—S2	125.11 (13)
C4—C3—C2	119.23 (17)	C24—C19—C20	118.89 (17)
C4—C3—H3	120.4	C24—C19—C18	104.65 (16)
C2—C3—H3	120.4	C20—C19—C18	136.46 (16)
C3—C4—C5	119.14 (18)	C21—C20—C19	119.18 (17)
C3—C4—C9	120.67 (17)	C21—C20—H20	120.4
C5—C4—C9	120.19 (17)	C19—C20—H20	120.4
C6—C5—C4	122.77 (18)	C20—C21—C22	119.5 (2)
C6—C5—H5	118.6	C20—C21—C26	120.71 (19)
C4—C5—H5	118.6	C22—C21—C26	119.77 (19)
C7—C6—C5	116.35 (17)	C23—C22—C21	122.55 (19)
C7—C6—H6	121.8	C23—C22—H22	118.7
C5—C6—H6	121.8	C21—C22—H22	118.7
C6—C7—O1	126.59 (16)	C22—C23—C24	116.24 (19)
C6—C7—C2	123.11 (18)	C22—C23—H23	121.9
O1—C7—C2	110.30 (16)	C24—C23—H23	121.9
C1—C8—O1	110.52 (16)	C23—C24—O4	126.14 (18)
C1—C8—C10	134.49 (18)	C23—C24—C19	123.6 (2)
O1—C8—C10	114.99 (16)	O4—C24—C19	110.24 (17)
C4—C9—H9A	109.5	C18—C25—O4	110.29 (17)
C4—C9—H9B	109.5	C18—C25—C27	134.5 (2)
H9A—C9—H9B	109.5	O4—C25—C27	115.18 (17)
C4—C9—H9C	109.5	C21—C26—H26A	109.5
H9A—C9—H9C	109.5	C21—C26—H26B	109.5
H9B—C9—H9C	109.5	H26A—C26—H26B	109.5
C8—C10—H10A	109.5	C21—C26—H26C	109.5
C8—C10—H10B	109.5	H26A—C26—H26C	109.5
H10A—C10—H10B	109.5	H26B—C26—H26C	109.5
C8—C10—H10C	109.5	C25—C27—H27A	109.5
H10A—C10—H10C	109.5	C25—C27—H27B	109.5
H10B—C10—H10C	109.5	H27A—C27—H27B	109.5
C12—C11—C16	121.63 (16)	C25—C27—H27C	109.5
C12—C11—S1	116.08 (14)	H27A—C27—H27C	109.5
C16—C11—S1	122.22 (13)	H27B—C27—H27C	109.5
C13—C12—C11	120.27 (18)	C33—C28—C29	121.31 (16)
C13—C12—H12	119.9	C33—C28—S2	115.62 (14)
C11—C12—H12	119.9	C29—C28—S2	122.98 (14)
C14—C13—C12	119.17 (18)	C31—C30—C29	122.19 (18)
C14—C13—H13	120.4	C31—C30—H30	118.9
C12—C13—H13	120.4	C29—C30—H30	118.9

C13—C14—C15	120.40 (18)	C30—C31—C32	120.42 (18)
C13—C14—H14	119.8	C30—C31—H31	119.8
C15—C14—H14	119.8	C32—C31—H31	119.8
C14—C15—C16	122.23 (18)	C31—C32—C33	119.08 (18)
C14—C15—H15	118.9	C31—C32—H32	120.5
C16—C15—H15	118.9	C33—C32—H32	120.5
C15—C16—C11	116.27 (17)	C32—C33—C28	120.36 (18)
C15—C16—C17	119.56 (17)	C32—C33—H33	119.8
C11—C16—C17	124.15 (16)	C28—C33—H33	119.8
C16—C17—H17A	109.5	C29—C34—H34A	109.5
C16—C17—H17B	109.5	C29—C34—H34B	109.5
H17A—C17—H17B	109.5	H34A—C34—H34B	109.5
C16—C17—H17C	109.5	C29—C34—H34C	109.5
H17A—C17—H17C	109.5	H34A—C34—H34C	109.5
H17B—C17—H17C	109.5	H34B—C34—H34C	109.5

Hydrogen-bond geometry (Å, °)

Cg1 and Cg2 are the centroids of benzene rings C2—C7 and C19—C24, respectively.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C23—H23...O5 ⁱ	0.95	2.52	3.431 (3)	162
C34—H34B...O2 ⁱⁱ	0.98	2.43	3.137 (2)	129
C31—H31...Cg1 ⁱⁱⁱ	0.95	2.93	3.773 (3)	148
C14—H14...Cg2 ⁱ	0.95	3.00	3.876 (3)	154

Symmetry codes: (i) $-x, y+1/2, -z+1/2$; (ii) $x, y+1, z$.