

Crystal structure of (*E*)-2-[(4-chloro-2*H*-chromen-3-yl)methylidene]-*N*-cyclohexylhydrazinecarbothioamide

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In the title compound, C₁₇H₂₀ClN₃OS, the mean plane of the central thiourea core makes dihedral angles of 26.56 (9) and 47.62 (12)° with the mean planes of the chromene moiety and the cyclohexyl ring, respectively. The cyclohexyl ring adopts a chair conformation. The N–H atoms of the thiourea unit adopt an *anti* conformation. The chromene group is positioned *trans*, whereas the cyclohexyl ring lies in the *cis* position to the thione S atom, with respect to the thiourea C–N bond. In the crystal, molecules are linked by N–H···S hydrogen bonds, forming inversion dimers enclosing R₂²(8) ring motifs. The dimers are linked by C–H···Cl hydrogen bonds, enclosing R₆⁶(44) ring motifs, forming sheets lying parallel to (010).

Keywords: crystal structure; chromene; hydrazine; thioamide; cyclohexyl; hydrogen bonds.

CCDC reference: 1016441

1. Related literature

For the biological properties of thiosemicarbazones, see: Prabhakaran *et al.* (2007); Kelly *et al.* (1996); West *et al.* (1993); Pérez *et al.* (1999). For their optical properties and applications, see: Tian *et al.* (1997); Uesugi *et al.* (1994). For a related structure, see: Jayakumar *et al.* (2011).

2. Experimental

2.1. Crystal data

C ₁₇ H ₂₀ ClN ₃ OS	<i>V</i> = 3483.9 (6) Å ³
<i>M_r</i> = 349.87	<i>Z</i> = 8
Orthorhombic, <i>Pbca</i>	Mo <i>K</i> α radiation
<i>a</i> = 12.2857 (12) Å	<i>μ</i> = 0.35 mm ⁻¹
<i>b</i> = 15.3082 (16) Å	<i>T</i> = 296 K
<i>c</i> = 18.5241 (18) Å	0.30 × 0.25 × 0.20 mm

2.2. Data collection

Bruker SMART APEXII area-detector diffractometer	18568 measured reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2008)	4291 independent reflections
<i>T</i> _{min} = 0.901, <i>T</i> _{max} = 0.933	2762 reflections with <i>I</i> > 2σ(<i>I</i>)
	<i>R</i> _{int} = 0.033

2.3. Refinement

<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.050	208 parameters
<i>wR</i> (<i>F</i> ²) = 0.152	H-atom parameters constrained
<i>S</i> = 1.02	Δρ _{max} = 0.32 e Å ⁻³
4291 reflections	Δρ _{min} = -0.31 e Å ⁻³

Table 1
Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N2–H2A···S1 ⁱ	0.86	2.73	3.507 (2)	151
C12–H12···Cl1 ⁱⁱ	0.98	2.83	3.689 (2)	147

Symmetry codes: (i) $-x + 1, -y, -z + 1$; (ii) $-x + \frac{3}{2}, -y, z + \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINT* (Bruker, 2008); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2009).

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Supporting information for this paper is available from the IUCr electronic archives (Reference: SU2769).

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Crystal structure of (*E*)-2-[(4-chloro-2*H*-chromen-3-yl)methylidene]-*N*-cyclohexylhydrazinecarbothioamide

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S1. Experimental

An ethanol solution of *N*-cyclohexylhydrazinecarbothioamide (1.736 g, 0.01 mole) was added to a ethanol solution (50 cm³) of 4-chloro-2*H*-chromene-3-carbaldehyde (1.94 g, 0.01 mole). The mixture was refluxed for 2 h during which time a yellow precipitate separated out. The reaction mixture was then cooled to room temperature and the precipitate was filtered off. It was then washed with ethanol and dried under vacuum (Yield: 85%). Crystals of the title compound were obtained by slow evaporation of a solution in ethanol.

S2. Refinement

The positions of the H atoms were localized from difference electron density maps and they were refined as riding atoms: N-H = 0.86 Å, C-H = 0.93 - 0.98 Å, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C-methyl})$ and $= 1.2U_{\text{eq}}(\text{N,C})$ for other H atoms.

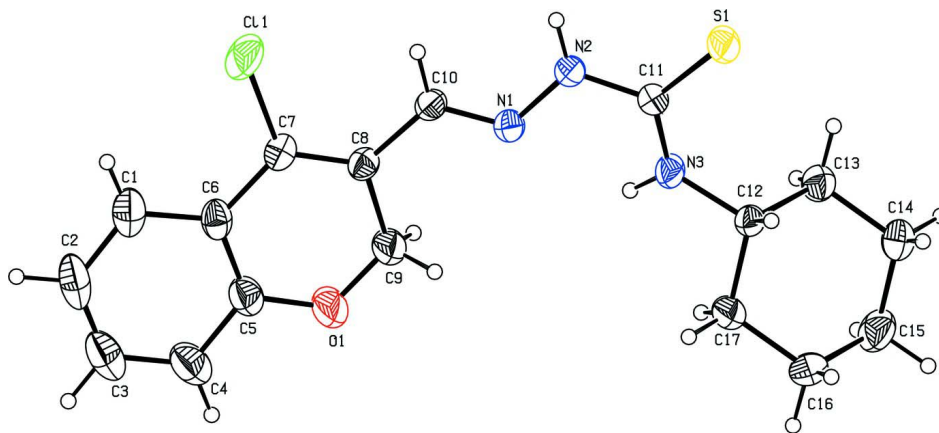
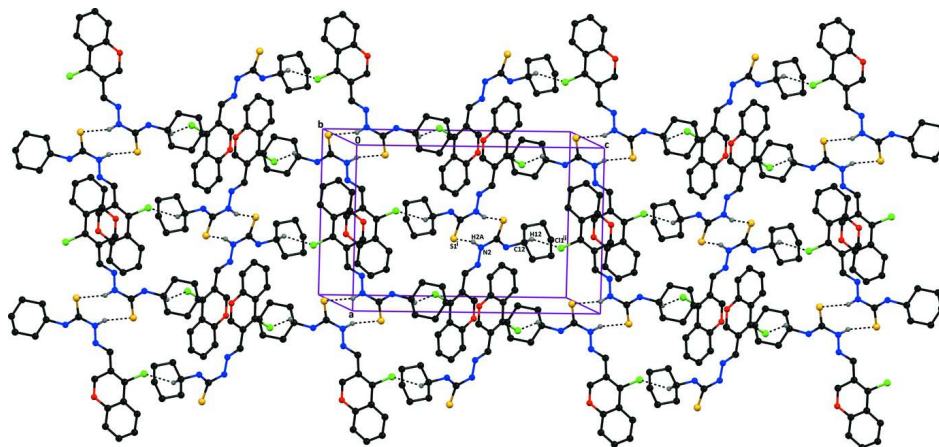


Figure 1

The molecular structure of the title molecule, with atom labelling. The displacement ellipsoids are drawn at the 30% probability level.

**Figure 2**

The crystal packing of the title compound viewed along the *a* axis. The hydrogen bonds are shown as dashed lines (see Table 1 for details; H atoms not involved in hydrogen bonding have been omitted for clarity).

(*E*)-2-[(4-Chloro-2*H*-chromen-3-yl)methylidene]-*N*-cyclohexylhydrazine carbothioamide

Crystal data

$C_{17}H_{20}ClN_3OS$

$M_r = 349.87$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 12.2857$ (12) Å

$b = 15.3082$ (16) Å

$c = 18.5241$ (18) Å

$V = 3483.9$ (6) Å³

$Z = 8$

$F(000) = 1472$

$D_x = 1.334$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2762 reflections

$\theta = 2.2$ – 28.3°

$\mu = 0.35$ mm⁻¹

$T = 296$ K

Block, colourless

$0.30 \times 0.25 \times 0.20$ mm

Data collection

Bruker SMART APEXII area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω and φ scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2008)

$T_{\min} = 0.901$, $T_{\max} = 0.933$

18568 measured reflections

4291 independent reflections

2762 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.033$

$\theta_{\max} = 28.3^\circ$, $\theta_{\min} = 2.2^\circ$

$h = -15 \rightarrow 16$

$k = -19 \rightarrow 19$

$l = -24 \rightarrow 23$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.050$

$wR(F^2) = 0.152$

$S = 1.02$

4291 reflections

208 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0745P)^2 + 0.9377P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.32$ e Å⁻³

$\Delta\rho_{\min} = -0.31$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	1.1410 (2)	0.0421 (2)	0.35313 (15)	0.0735 (8)
H1	1.1134	0.0105	0.3144	0.088*
C2	1.2530 (2)	0.0541 (2)	0.36038 (19)	0.0835 (10)
H2	1.2997	0.0306	0.3259	0.100*
C3	1.2948 (2)	0.0990 (2)	0.41635 (19)	0.0824 (9)
H3	1.3698	0.1057	0.4205	0.099*
C4	1.2276 (2)	0.1348 (2)	0.46699 (17)	0.0790 (8)
H4	1.2568	0.1662	0.5054	0.095*
C5	1.11588 (18)	0.12440 (18)	0.46109 (13)	0.0600 (6)
C6	1.07055 (17)	0.07780 (15)	0.40441 (12)	0.0514 (5)
C7	0.95198 (17)	0.06950 (15)	0.40340 (11)	0.0485 (5)
C8	0.88960 (16)	0.10411 (14)	0.45507 (10)	0.0438 (5)
C9	0.94477 (19)	0.1572 (2)	0.51200 (14)	0.0712 (8)
H9A	0.9170	0.2164	0.5085	0.085*
H9B	0.9222	0.1345	0.5585	0.085*
C10	0.77399 (16)	0.08929 (15)	0.46230 (10)	0.0453 (5)
H10	0.7364	0.0579	0.4273	0.054*
C11	0.56710 (15)	0.10953 (15)	0.58999 (10)	0.0444 (5)
C12	0.59224 (15)	0.17836 (15)	0.71073 (10)	0.0423 (5)
H12	0.5665	0.1240	0.7327	0.051*
C13	0.50398 (19)	0.24676 (16)	0.71814 (12)	0.0563 (6)
H13A	0.4375	0.2255	0.6960	0.068*
H13B	0.5259	0.2995	0.6930	0.068*
C14	0.4829 (2)	0.2677 (2)	0.79728 (14)	0.0729 (8)
H14A	0.4279	0.3129	0.8007	0.087*
H14B	0.4555	0.2160	0.8215	0.087*
C15	0.5855 (2)	0.2981 (2)	0.83439 (14)	0.0733 (8)
H15A	0.6093	0.3527	0.8130	0.088*
H15B	0.5704	0.3087	0.8850	0.088*
C16	0.6747 (2)	0.2314 (2)	0.82767 (13)	0.0713 (8)
H16A	0.6547	0.1794	0.8546	0.086*
H16B	0.7410	0.2546	0.8486	0.086*
C17	0.69562 (17)	0.2068 (2)	0.74917 (12)	0.0645 (7)
H17A	0.7266	0.2566	0.7241	0.077*
H17B	0.7483	0.1597	0.7474	0.077*

N1	0.72378 (13)	0.11972 (13)	0.51739 (8)	0.0456 (4)
N2	0.61712 (13)	0.09549 (13)	0.52482 (8)	0.0481 (4)
H2A	0.5823	0.0721	0.4895	0.058*
N3	0.62011 (13)	0.16098 (12)	0.63520 (9)	0.0486 (4)
H3A	0.6769	0.1870	0.6185	0.058*
Cl1	0.89480 (7)	0.01105 (6)	0.33357 (4)	0.0936 (3)
O1	1.05307 (15)	0.16182 (18)	0.51223 (12)	0.1067 (9)
S1	0.44825 (4)	0.05982 (5)	0.60746 (3)	0.0645 (2)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0760 (17)	0.0745 (19)	0.0699 (15)	0.0181 (15)	0.0253 (13)	0.0081 (14)
C2	0.0698 (17)	0.086 (2)	0.095 (2)	0.0314 (16)	0.0410 (16)	0.0311 (19)
C3	0.0520 (14)	0.089 (2)	0.106 (2)	0.0117 (15)	0.0193 (15)	0.039 (2)
C4	0.0504 (13)	0.093 (2)	0.0933 (19)	-0.0055 (14)	0.0029 (13)	0.0141 (18)
C5	0.0483 (12)	0.0669 (16)	0.0649 (13)	-0.0009 (12)	0.0085 (10)	0.0079 (13)
C6	0.0532 (11)	0.0488 (13)	0.0523 (11)	0.0072 (10)	0.0149 (9)	0.0133 (10)
C7	0.0581 (12)	0.0446 (12)	0.0429 (10)	-0.0022 (10)	0.0071 (9)	-0.0021 (9)
C8	0.0481 (10)	0.0453 (12)	0.0381 (9)	-0.0035 (9)	0.0032 (8)	0.0006 (9)
C9	0.0466 (12)	0.100 (2)	0.0675 (15)	-0.0087 (13)	0.0033 (10)	-0.0310 (15)
C10	0.0467 (10)	0.0522 (13)	0.0370 (9)	-0.0054 (10)	0.0004 (8)	-0.0019 (9)
C11	0.0390 (9)	0.0516 (13)	0.0425 (9)	0.0002 (9)	0.0007 (7)	-0.0016 (10)
C12	0.0405 (9)	0.0481 (12)	0.0385 (9)	-0.0048 (9)	0.0027 (7)	-0.0014 (9)
C13	0.0556 (13)	0.0614 (16)	0.0519 (11)	0.0092 (12)	0.0019 (9)	-0.0019 (11)
C14	0.0633 (15)	0.092 (2)	0.0634 (14)	0.0143 (15)	0.0081 (12)	-0.0221 (15)
C15	0.0829 (18)	0.0764 (19)	0.0608 (14)	-0.0063 (16)	0.0083 (13)	-0.0251 (14)
C16	0.0611 (14)	0.097 (2)	0.0556 (13)	-0.0050 (15)	-0.0097 (11)	-0.0245 (14)
C17	0.0412 (11)	0.0898 (19)	0.0625 (13)	-0.0023 (12)	-0.0016 (10)	-0.0232 (14)
N1	0.0417 (8)	0.0540 (11)	0.0412 (8)	-0.0050 (8)	0.0034 (6)	0.0009 (8)
N2	0.0396 (8)	0.0637 (12)	0.0410 (8)	-0.0071 (8)	0.0016 (6)	-0.0058 (8)
N3	0.0423 (8)	0.0598 (12)	0.0436 (8)	-0.0144 (8)	0.0096 (7)	-0.0081 (8)
Cl1	0.0938 (6)	0.1186 (7)	0.0683 (4)	-0.0167 (5)	0.0133 (3)	-0.0505 (4)
O1	0.0485 (10)	0.173 (3)	0.0992 (15)	-0.0152 (12)	0.0062 (9)	-0.0721 (16)
S1	0.0446 (3)	0.0870 (5)	0.0621 (4)	-0.0208 (3)	0.0102 (2)	-0.0205 (3)

Geometric parameters (Å, °)

C1—C2	1.395 (4)	C11—S1	1.678 (2)
C1—C6	1.397 (3)	C12—N3	1.465 (2)
C1—H1	0.9300	C12—C13	1.514 (3)
C2—C3	1.346 (5)	C12—C17	1.520 (3)
C2—H2	0.9300	C12—H12	0.9800
C3—C4	1.364 (4)	C13—C14	1.523 (3)
C3—H3	0.9300	C13—H13A	0.9700
C4—C5	1.387 (3)	C13—H13B	0.9700
C4—H4	0.9300	C14—C15	1.510 (4)
C5—O1	1.349 (3)	C14—H14A	0.9700

C5—C6	1.386 (3)	C14—H14B	0.9700
C6—C7	1.462 (3)	C15—C16	1.503 (4)
C7—C8	1.336 (3)	C15—H15A	0.9700
C7—C11	1.723 (2)	C15—H15B	0.9700
C8—C10	1.445 (3)	C16—C17	1.524 (3)
C8—C9	1.494 (3)	C16—H16A	0.9700
C9—O1	1.332 (3)	C16—H16B	0.9700
C9—H9A	0.9700	C17—H17A	0.9700
C9—H9B	0.9700	C17—H17B	0.9700
C10—N1	1.280 (3)	N1—N2	1.369 (2)
C10—H10	0.9300	N2—H2A	0.8600
C11—N3	1.321 (3)	N3—H3A	0.8600
C11—N2	1.372 (2)		
C2—C1—C6	119.7 (3)	C13—C12—H12	108.6
C2—C1—H1	120.2	C17—C12—H12	108.6
C6—C1—H1	120.2	C12—C13—C14	110.75 (19)
C3—C2—C1	121.2 (3)	C12—C13—H13A	109.5
C3—C2—H2	119.4	C14—C13—H13A	109.5
C1—C2—H2	119.4	C12—C13—H13B	109.5
C2—C3—C4	120.3 (3)	C14—C13—H13B	109.5
C2—C3—H3	119.9	H13A—C13—H13B	108.1
C4—C3—H3	119.9	C15—C14—C13	111.2 (2)
C3—C4—C5	119.9 (3)	C15—C14—H14A	109.4
C3—C4—H4	120.0	C13—C14—H14A	109.4
C5—C4—H4	120.0	C15—C14—H14B	109.4
O1—C5—C6	121.4 (2)	C13—C14—H14B	109.4
O1—C5—C4	117.5 (3)	H14A—C14—H14B	108.0
C6—C5—C4	121.1 (2)	C16—C15—C14	111.2 (2)
C5—C6—C1	117.9 (2)	C16—C15—H15A	109.4
C5—C6—C7	117.04 (19)	C14—C15—H15A	109.4
C1—C6—C7	125.1 (2)	C16—C15—H15B	109.4
C8—C7—C6	121.9 (2)	C14—C15—H15B	109.4
C8—C7—C11	120.68 (17)	H15A—C15—H15B	108.0
C6—C7—C11	117.45 (16)	C15—C16—C17	111.7 (2)
C7—C8—C10	124.63 (19)	C15—C16—H16A	109.3
C7—C8—C9	117.47 (19)	C17—C16—H16A	109.3
C10—C8—C9	117.78 (18)	C15—C16—H16B	109.3
O1—C9—C8	119.0 (2)	C17—C16—H16B	109.3
O1—C9—H9A	107.6	H16A—C16—H16B	107.9
C8—C9—H9A	107.6	C12—C17—C16	112.15 (18)
O1—C9—H9B	107.6	C12—C17—H17A	109.2
C8—C9—H9B	107.6	C16—C17—H17A	109.2
H9A—C9—H9B	107.0	C12—C17—H17B	109.2
N1—C10—C8	119.37 (18)	C16—C17—H17B	109.2
N1—C10—H10	120.3	H17A—C17—H17B	107.9
C8—C10—H10	120.3	C10—N1—N2	116.28 (17)
N3—C11—N2	115.49 (17)	N1—N2—C11	118.36 (16)

N3—C11—S1	125.24 (15)	N1—N2—H2A	120.8
N2—C11—S1	119.25 (15)	C11—N2—H2A	120.8
N3—C12—C13	112.32 (17)	C11—N3—C12	126.76 (17)
N3—C12—C17	107.71 (16)	C11—N3—H3A	116.6
C13—C12—C17	110.98 (19)	C12—N3—H3A	116.6
N3—C12—H12	108.6	C9—O1—C5	123.1 (2)
C6—C1—C2—C3	0.5 (4)	C7—C8—C10—N1	174.1 (2)
C1—C2—C3—C4	-0.7 (5)	C9—C8—C10—N1	-1.8 (3)
C2—C3—C4—C5	0.4 (5)	N3—C12—C13—C14	-176.0 (2)
C3—C4—C5—O1	-179.4 (3)	C17—C12—C13—C14	-55.4 (3)
C3—C4—C5—C6	0.1 (4)	C12—C13—C14—C15	57.3 (3)
O1—C5—C6—C1	179.1 (3)	C13—C14—C15—C16	-56.7 (3)
C4—C5—C6—C1	-0.4 (4)	C14—C15—C16—C17	54.5 (3)
O1—C5—C6—C7	-1.3 (4)	N3—C12—C17—C16	176.9 (2)
C4—C5—C6—C7	179.3 (2)	C13—C12—C17—C16	53.6 (3)
C2—C1—C6—C5	0.1 (4)	C15—C16—C17—C12	-53.2 (3)
C2—C1—C6—C7	-179.5 (2)	C8—C10—N1—N2	-173.26 (19)
C5—C6—C7—C8	-0.8 (3)	C10—N1—N2—C11	165.58 (19)
C1—C6—C7—C8	178.8 (2)	N3—C11—N2—N1	13.0 (3)
C5—C6—C7—C11	-179.54 (18)	S1—C11—N2—N1	-165.46 (16)
C1—C6—C7—C11	0.1 (3)	N2—C11—N3—C12	-172.13 (19)
C6—C7—C8—C10	-171.9 (2)	S1—C11—N3—C12	6.2 (3)
C11—C7—C8—C10	6.7 (3)	C13—C12—N3—C11	-81.4 (3)
C6—C7—C8—C9	4.0 (3)	C17—C12—N3—C11	156.1 (2)
C11—C7—C8—C9	-177.36 (19)	C8—C9—O1—C5	3.6 (5)
C7—C8—C9—O1	-5.4 (4)	C6—C5—O1—C9	-0.3 (5)
C10—C8—C9—O1	170.8 (3)	C4—C5—O1—C9	179.2 (3)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2A...S1 ⁱ	0.86	2.73	3.507 (2)	151
C12—H12...C11 ⁱⁱ	0.98	2.83	3.689 (2)	147

Symmetry codes: (i) $-x+1, -y, -z+1$; (ii) $-x+3/2, -y, z+1/2$.