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# Crystal structure of trans-1,4-bis[(trimethylsilyl)-oxy]cyclohexa-2,5-diene-1,4-dicarbonitrile 

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The asymmetric unit of the title compound, $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Si}_{2}$, contains one half of the molecule, which is completed by inversion symmetry. The cyclohexa-2,5diene ring is exactly planar and reflects the bond-length distribution of a pair of located double bonds [1.3224 (14) Å] and two pairs of single bonds [1.5121 (13) and 1.5073 (14) $\AA]$. The tetrahedral angle between the $s p^{3}$-C atom and the two neighbouring $s p^{2}$-C atoms in the cyclohexa-2,5-diene ring is enlarged by about $3^{\circ}$.

## 1. Chemical context

Cyanohydrins (Friedrich, 1983) are an important class of organic compounds. Silylated cyanohydrins are versatile precursor compounds in organic chemistry because the nitrile functional group can be modified by a variety of reactions such as hydrolysis, reduction or addition of organometallic reagents. The molecular and crystal structure of the title compound, a new silylated cyclohexa-2,5-diene with trans nitrile groups in the 1,4 positions, is reported herein.


## 2. Structural commentary

The molecular structure of the title compound is centrosymmetric, leading to a trans-1,4-configuration of the oxy(trimethylsilyl) and carbonitrile groups (Fig. 1). The cyclohexa-2,5-diene ring is exactly planar, but its angles differ from that of an ideal hexagon. Whereas the angle between the $s p^{3}$ - C atom ( C 1 ) and the neighbouring $s p^{2}-\mathrm{C}$ atoms $(\mathrm{C} 2, \mathrm{C} 3)$ is reduced to $112.58(8)^{\circ}$, the other intra-ring angles are enlarged to $123.94(9)^{\circ}(\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3)$ and $123.48(9)^{\circ}\left(\mathrm{C}^{\mathrm{i}}-\mathrm{C} 3-\mathrm{C} 2\right)$ [symmetry code: (i) $-x+1,-y+1,-z$ ]. The tetrahedral angles around C 1 are likewise distorted due to the ring strain. The angles involving the O atom of the oxy(trimethylsilyl) group and the ring C atoms are enlarged to 110.79 (8) ${ }^{\circ}$ and 113.26 ( 8$)^{\circ}$ while the angle involving the O atom and the C atom of the carbonitrile group is reduced to $104.95(8)^{\circ}$. The backbone of the 1,1 -substituents is nearly perpendicular to the cyclohexa-2,5-diene ring, with a dihedral angle of 86.05 (7).


Figure 1
The molecular structure of the title compound, showing the atomlabelling scheme and displacement ellipsoids drawn at the $80 \%$ probability level. Non-labelled atoms are generated by the symmetry code $-x+1,-y+1,-z$.

## 3. Supramolecular features

Notable features in terms of non-classical hydrogen bonding interactions are not observed in the crystal structure of the title compound. As a result of the bulky trimethylsilyl groups, $\pi-\pi$ stacking interactions between the rings are not possible. The packing of the molecules (Fig. 2) seems to be dominated mainly by van der Waals forces.

## 4. Database survey

In the current Cambridge Structural Database (Version 5.35, last update February 2014; Allen, 2002) only one example of a cyclohexa-2,5-diene with trans nitrile groups in the 1,4 posi-


Figure 2
A view of the crystal packing of the title compound along [001]. Colour code: O red, C grey, N light-blue, Si off-white, H white.


Figure 3
Reaction scheme to obtain the title compound.
tions is listed, namely 3,5 -bis(4-(dimethylamino)phenyl)-cyclohexa-2,5-diene-1,1,2,4,4-pentacarbonitrile (Jayamurugan et al., 2011). The $\mathrm{C}-\mathrm{C}$ bond lengths within the cyclohexa-2,5diene are very similar to those of the title compound.

## 5. Synthesis and crystallization

1,4-Bis[(trimethylsilyl)oxy]cyclohexa-2,5-diene-1,4-dicarbonitrile was synthesized by a modified protocol reported by Onaka et al. (1989). The required heterogeneous catalyst Femontmorillonite (K10-FeAA) was prepared according to Pai et al. (2000) and activated at 393 K and 5 mbar for 2 h prior to use.

1,4-Benzoquinone ( $1.62 \mathrm{~g}, 15 \mathrm{mmol}$ ) was dissolved in 75 ml dichloromethane ( 0.2 M ), purged with argon and cooled to 273 K . Trimethylsilyl cyanide $(2.98 \mathrm{~g}, 30 \mathrm{mmol})$ and $\mathrm{Fe}-$ montmorillonite $(0.75 \mathrm{~g})$ were added sequentially and the mixture stirred for 1 h at 273 K under an argon atmosphere. The Fe-montmorillonite was filtered off (Por 4 glass filter) and the solvent was evaporated in vacuo to yield 4.23 g

Table 1
Experimental details.

| Crystal data |  |
| :--- | :--- |
| Chemical formula | $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Si}_{2}$ |
| $M_{\mathrm{r}}$ | 306.5 |
| Crystal system, space group | Monoclinic, $P 2_{1} / n$ |
| Temperature (K) | 100 |
| $a, b, c(\AA)$ | $8.0770(5), 11.2234(6), 9.4377(6)$ |
| $\beta\left({ }^{\circ}\right)$ | $97.7087(19)$ |
| $V\left(\AA^{3}\right)$ | $847.81(9)$ |
| $Z$ | 2 |
| Radiation type | Mo $K \alpha$ |
| $\mu\left(\mathrm{~mm}^{-1}\right)$ | 0.21 |
| Crystal size (mm) | $0.65 \times 0.26 \times 0.12$ |
|  |  |
| Data collection | Bruker Kappa APEXII CCD |
| Diffractometer | Multi-scan $(S A D A B S ;$ Bruker, |
| Absorption correction | $2013)$ |
|  | $0.94,0.98$ |
| $T_{\text {min }}, T_{\text {max }}$ | $15160,2487,2123$ |
| No. of measured, independent and |  |
| observed $[I>3 \sigma(I)]$ reflections | 0.024 |
| $R_{\text {int }}$ | 0.705 |
| (sin $\theta / \lambda)_{\text {max }}\left(\AA \AA^{-1}\right)$ |  |
|  |  |
| Refinement | $0.030,0.042,2.38$ |
| $R\left[F^{2}>3 \sigma\left(F^{2}\right)\right], w R\left(F^{2}\right), S$ | 2487 |
| No. of reflections | 91 |
| No. of parameters | H -atom parameters constrained |
| H-atom treatment | $0.38,-0.20$ |
| $\Delta \rho_{\text {max }}, \Delta \rho_{\text {min }}\left(\mathrm{e} \AA \AA^{-3}\right)$ |  |

Computer programs: APEX2 and SAINT-Plus (Bruker, 2013), SUPERFLIP (Palatinus \& Chapuis, 2007), JANA2006 (Petríček, et al., 2014), Mercury (Macrae et al., 2008) and publCIF (Westrip, 2010).
( $13.8 \mathrm{mmol}, 92 \%$ ) of a cis/trans (3/1) isomeric mixture of $1,4-$ bis[(trimethylsilyl)oxy]cyclohexa-2,5-diene-1,4-dicarbonitrile (Fig. 3). Crystallization from $n$-hexane selectively yielded white crystals of the trans-isomer, which were suitable for single-crystal X-ray diffraction analysis. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right.$, $200 \mathrm{MHz}): \delta=6.19(s, 4 \mathrm{H}), 0.23(\mathrm{~s}, 18 \mathrm{H})$ p.p.m.; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right): \delta=238.3(s), 129.4(d), 1.5(q)$ p.p.m.

## 6. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 1. The H atoms were included in calculated positions $(\mathrm{C}-\mathrm{H}=0.96 \AA)$ and treated as riding atoms with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.

## Acknowledgements

The X-ray centre of the Vienna University of Technology is acknowledged for providing access to the single-crystal diffractometer.

## References

Allen, F. H. (2002). Acta Cryst. B58, 380-388.
Bruker (2013). APEX2, SAINT-Plus and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
Friedrich, K. (1983). The Chemistry of Functional Groups, Supplement C, Part 2, edited by S. Patai \& Z. Rappoport, pp. 1345-1390, New York: Wiley.
Jayamurugan, G., Gisselbrecht, J.-P., Boudon, C., Schoenebeck, F., Schweizer, W. B., Bernet, B. \& Diederich, F. (2011). Chem. Commun. 47, 4510-4522.
Macrae, C. F., Bruno, I. J., Chisholm, J. A., Edgington, P. R., McCabe, P., Pidcock, E., Rodriguez-Monge, L., Taylor, R., van de Streek, J. \& Wood, P. A. (2008). J. Appl. Cryst. 41, 466-470.
Onaka, M., Higuchi, K., Sugita, K. \& Izumi, Y. (1989). Chem. Lett. 18, 1393-1396.
Pai, S. G., Bajpai, A. R., Deshpande, A. B. \& Samant, S. D. (2000). J. Mol. Catal. A Chem. 156, 233-243.
Palatinus, L. \& Chapuis, G. (2007). J. Appl. Cryst. 40, 786-790.
Petříček, V., Dušek, M. \& Palatinus, L. (2014). Z. Kristallogr. 229, 345-352.
Westrip, S. P. (2010). J. Appl. Cryst. 43, 920-925.

## supporting information

## Crystal structure of trans-1,4-bis[(trimethylsilyl)oxy]cyclohexa-2,5-diene-1,4dicarbonitrile

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## Computing details

Data collection: APEX2 (Bruker, 2013); cell refinement: SAINT-Plus (Bruker, 2013); data reduction: SAINT-Plus (Bruker, 2013); program(s) used to solve structure: SUPERFLIP (Palatinus \& Chapuis, 2007); program(s) used to refine structure: JANA2006 (Petříček, et al., 2014); molecular graphics: Mercury (Macrae et al., 2008); software used to prepare material for publication: publCIF (Westrip, 2010).
trans-1,4-Bis[(trimethylsilyl)oxy]cyclohexa-2,5-diene-1,4-dicarbonitrile

## Crystal data

$\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Si}_{2}$
$M_{r}=306.5$
Monoclinic, $P 2_{1} / n$
Hall symbol: -P 2 yn
$a=8.0770$ (5) $\AA$
$b=11.2234$ (6) $\AA$
$c=9.4377$ (6) $\AA$
$\beta=97.7087(19)^{\circ}$
$V=847.81(9) \AA^{3}$
$Z=2$

## Data collection

Bruker Kappa APEXII CCD diffractometer
Radiation source: X-ray tube
Graphite monochromator
$\omega$ and $\varphi$-scans
Absorption correction: multi-scan
(SADABS; Bruker, 2013)
$T_{\text {min }}=0.94, T_{\text {max }}=0.98$

## Refinement

Refinement on $F$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.030$
$w R\left(F^{2}\right)=0.042$
$S=2.38$
2487 reflections
91 parameters
0 restraints
$F(000)=328$
$D_{\mathrm{x}}=1.200 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation, $\lambda=0.71073 \AA$
Cell parameters from 7267 reflections
$\theta=2.8-29.9^{\circ}$
$\mu=0.21 \mathrm{~mm}^{-1}$
$T=100 \mathrm{~K}$
Block, clear colourless
$0.65 \times 0.26 \times 0.12 \mathrm{~mm}$

15160 measured reflections
2487 independent reflections
2123 reflections with $I>3 \sigma(I)$
$R_{\text {int }}=0.024$
$\theta_{\text {max }}=30.1^{\circ}, \theta_{\text {min }}=2.8^{\circ}$
$h=-11 \rightarrow 11$
$k=-15 \rightarrow 15$
$l=-13 \rightarrow 13$

44 constraints
H -atom parameters constrained
Weighting scheme based on measured s.u.'s $w=$

$$
1 /\left(\sigma^{2}(F)+0.0001 F^{2}\right)
$$

$(\Delta / \sigma)_{\text {max }}=0.023$
$\Delta \rho_{\text {max }}=0.38$ e $\AA^{-3}$
$\Delta \rho_{\min }=-0.20 \mathrm{e}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $A^{2}$ )

|  | $x$ | $y$ | $z$ | $U_{\text {iso }}{ }^{\prime} U_{\mathrm{eq}}$ |
| :--- | :--- | :--- | :--- | :--- |
| Si1 | $0.21973(4)$ | $0.48658(3)$ | $0.25729(3)$ | $0.01515(9)$ |
| O1 | $0.34897(9)$ | $0.59176(6)$ | $0.20989(8)$ | $0.0151(2)$ |
| N1 | $0.66734(12)$ | $0.77227(8)$ | $0.18067(10)$ | $0.0203(3)$ |
| C1 | $0.47262(12)$ | $0.58514(9)$ | $0.11702(10)$ | $0.0118(3)$ |
| C2 | $0.39530(12)$ | $0.59953(9)$ | $-0.03704(10)$ | $0.0130(3)$ |
| C3 | $0.41937(12)$ | $0.52449(9)$ | $-0.14034(11)$ | $0.0125(3)$ |
| C4 | $0.58275(13)$ | $0.69117(9)$ | $0.15441(10)$ | $0.0136(3)$ |
| C5 | $0.04800(15)$ | $0.57505(11)$ | $0.31580(13)$ | $0.0244(4)$ |
| C6 | $0.32433(16)$ | $0.39773(11)$ | $0.40911(13)$ | $0.0303(4)$ |
| C7 | $0.14743(14)$ | $0.38628(10)$ | $0.10522(12)$ | $0.0208(3)$ |
| H1c2 | 0.324479 | 0.667221 | -0.061301 | $0.0155^{*}$ |
| H1c3 | 0.365173 | 0.540694 | -0.235238 | $0.015^{*}$ |
| H1c5 | -0.036613 | 0.522349 | 0.341718 | $0.0293^{*}$ |
| H2c5 | 0.09082 | 0.622785 | 0.396915 | $0.0293^{*}$ |
| H3c5 | 0.000656 | 0.625942 | 0.239152 | $0.0293^{*}$ |
| H1c6 | 0.246714 | 0.341141 | 0.438738 | $0.0363^{*}$ |
| H2c6 | 0.417964 | 0.356223 | 0.379805 | $0.0363^{*}$ |
| H3c6 | 0.362472 | 0.449816 | 0.487404 | $0.0363^{*}$ |
| H1c7 | 0.049234 | 0.344384 | 0.124438 | $0.0249^{*}$ |
| H2c7 | 0.121741 | 0.432548 | 0.019545 | $0.0249^{*}$ |
| H3c7 | 0.233911 | 0.330007 | 0.093019 | $0.0249^{*}$ |

Atomic displacement parameters $\left(\hat{A}^{2}\right)$

|  | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Si1 | $0.01621(16)$ | $0.01632(17)$ | $0.01325(15)$ | $-0.00379(11)$ | $0.00316(11)$ | $-0.00069(11)$ |
| O1 | $0.0166(4)$ | $0.0138(4)$ | $0.0161(4)$ | $-0.0014(3)$ | $0.0071(3)$ | $-0.0023(3)$ |
| N 1 | $0.0220(5)$ | $0.0178(5)$ | $0.0210(5)$ | $-0.0034(4)$ | $0.0026(4)$ | $-0.0036(4)$ |
| C 1 | $0.0132(4)$ | $0.0106(5)$ | $0.0118(4)$ | $-0.0006(3)$ | $0.0026(3)$ | $-0.0006(3)$ |
| C 2 | $0.0126(4)$ | $0.0112(5)$ | $0.0147(5)$ | $0.0009(4)$ | $0.0001(4)$ | $0.0016(4)$ |
| C 3 | $0.0126(5)$ | $0.0121(5)$ | $0.0122(4)$ | $0.0000(4)$ | $-0.0005(4)$ | $0.0017(4)$ |
| C4 | $0.0151(5)$ | $0.0143(5)$ | $0.0115(4)$ | $0.0018(4)$ | $0.0021(3)$ | $-0.0004(4)$ |
| C5 | $0.0226(6)$ | $0.0291(7)$ | $0.0236(6)$ | $-0.0046(5)$ | $0.0102(5)$ | $-0.0070(5)$ |
| C6 | $0.0327(7)$ | $0.0314(7)$ | $0.0249(6)$ | $-0.0108(5)$ | $-0.0029(5)$ | $0.0105(5)$ |
| C7 | $0.0215(6)$ | $0.0219(6)$ | $0.0198(5)$ | $-0.0064(4)$ | $0.0058(4)$ | $-0.0038(4)$ |

Geometric parameters ( $\AA,{ }^{\circ}$ )

| Si1-C5 | $1.8495(13)$ | $\mathrm{C} 3-\mathrm{H} 1 \mathrm{c} 3$ | 0.96 |
| :--- | :--- | :--- | :--- |
| $\mathrm{Si} 1-\mathrm{C} 6$ | $1.8537(13)$ | $\mathrm{C} 5-\mathrm{H} 1 \mathrm{c} 5$ | 0.96 |
| $\mathrm{Si}-\mathrm{C} 7$ | $1.8555(11)$ | $\mathrm{C} 5-\mathrm{H} 2 \mathrm{c} 5$ | 0.96 |
| $\mathrm{O} 1-\mathrm{C} 1$ | $1.4163(13)$ | $\mathrm{C} 5-\mathrm{H} 3 \mathrm{c} 5$ | 0.96 |
| $\mathrm{~N} 1-\mathrm{C} 4$ | $1.1451(14)$ | $\mathrm{C} 6-\mathrm{H} 1 \mathrm{c} 6$ | 0.96 |
| $\mathrm{C} 1-\mathrm{C} 2$ | $1.5121(13)$ | $\mathrm{C} 6-\mathrm{H} 2 \mathrm{c} 6$ | 0.96 |
| $\mathrm{C} 1-\mathrm{C}^{\mathrm{i}}$ | $1.5073(14)$ | $\mathrm{C} 6-\mathrm{H} 3 \mathrm{c} 6$ | 0.96 |


| C1-C4 | 1.4993 (14) | C7-H1c7 | 0.96 |
| :---: | :---: | :---: | :---: |
| C2-C3 | 1.3224 (14) | C7-H2c7 | 0.96 |
| C2-H1c2 | 0.96 | C7-H3c7 | 0.96 |
| C5-Si1-C6 | 109.89 (6) | Si1-C5-H2c5 | 109.47 |
| C5-Si1-C7 | 112.70 (5) | Si1-C5-H3c5 | 109.47 |
| C6-Si1-C7 | 109.57 (5) | H1c5-C5-H2c5 | 109.47 |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2$ | 110.79 (8) | H1c5-C5-H3c5 | 109.47 |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C}^{\text {i }}$ | 113.26 (8) | H2c5-C5-H3c5 | 109.47 |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 4$ | 104.95 (8) | Si1-C6-H1c6 | 109.47 |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{C} 3{ }^{\text {i }}$ | 112.58 (8) | Si1-C6-H2c6 | 109.47 |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{C} 4$ | 107.28 (8) | Si1-C6-H3c6 | 109.47 |
| C3- ${ }^{\text {i }} 1-\mathrm{C} 4$ | 107.46 (8) | H1c6-C6-H2c6 | 109.47 |
| C1-C2-C3 | 123.94 (9) | H1c6-C6-H3c6 | 109.47 |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{H} 1 \mathrm{c} 2$ | 118.03 | H2c6-C6-H3c6 | 109.47 |
| $\mathrm{C} 3-\mathrm{C} 2-\mathrm{H} 1 \mathrm{c} 2$ | 118.03 | Si1-C7-H1c7 | 109.47 |
| C1- ${ }^{\text {i }} 3-\mathrm{C} 2$ | 123.48 (9) | Si1-C7-H2c7 | 109.47 |
| $\mathrm{C} 12-\mathrm{C} 3-\mathrm{H} 1 \mathrm{c} 3$ | 118.26 | Si1-C7-H3c7 | 109.47 |
| C2-C3-H1c3 | 118.26 | H1c7-C7-H2c7 | 109.47 |
| N1-C4-C1 | 178.87 (11) | H1c7-C7-H3c7 | 109.47 |
| Si1-C5-H1c5 | 109.47 | H2c7-C7-H3c7 | 109.47 |

Symmetry code: (i) $-x+1,-y+1,-z$.

