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3-(3-Bromophenylsulfinyl)-5-cyclohexyl-2-methyl-1-benzofuran

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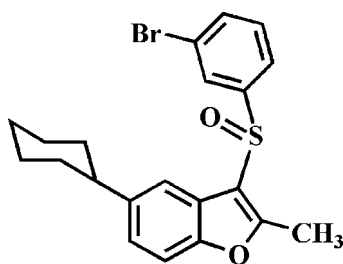
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.031; wR factor = 0.081; data-to-parameter ratio = 14.5.

In the title compound, $\text{C}_{21}\text{H}_{21}\text{BrO}_2\text{S}$, the cyclohexyl ring adopts a chair conformation. The dihedral angle between the mean plane [r.m.s. deviation = 0.178 (2) Å] of the benzofuran ring system and the mean plane of the 3-bromophenyl ring is 86.52 (6)°. In the crystal, molecules are linked by weak C—H...O and C—H... π hydrogen bonds, and by a slipped π – π interaction between the furan rings of neighbouring molecules [centroid–centroid distance = 3.518 (3) Å, interplanar distance = 3.471 (3) Å and slippage = 0.573 (3) Å], resulting in a three-dimensional network.

Related literature

For background information and the crystal structures of related compounds, see: Choi *et al.* (2011, 2012*a,b*).



Experimental

Crystal data

$\text{C}_{21}\text{H}_{21}\text{BrO}_2\text{S}$
 $M_r = 417.35$
 Monoclinic, $P2_1/c$
 $a = 17.6432$ (9) Å
 $b = 8.9425$ (5) Å
 $c = 12.4776$ (7) Å
 $\beta = 108.277$ (3)°

$V = 1869.33$ (18) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 2.32$ mm⁻¹
 $T = 173$ K
 $0.33 \times 0.25 \times 0.18$ mm

Data collection

Bruker SMART APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2009)
 $T_{\min} = 0.482$, $T_{\max} = 0.746$

13952 measured reflections
 3290 independent reflections
 2635 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.031$
 $wR(F^2) = 0.081$
 $S = 1.02$
 3290 reflections

227 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.33$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.35$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C2–C7 benzene ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C15—H15A...O1 ⁱ	0.98	2.58	3.358 (3)	136
C21—H21...O2 ⁱⁱ	0.95	2.45	3.306 (3)	151
C15—H15C...Cg1 ⁱⁱⁱ	0.98	2.82	3.502 (3)	127

Symmetry codes: (i) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x, -y + \frac{1}{2}, z - \frac{1}{2}$; (iii) $-x + 1, -y + 1, -z + 1$.

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012) and DIAMOND (Brandenburg, 1998); software used to prepare material for publication: SHELXL97.

Supporting information for this paper is available from the IUCr electronic archives (Reference: GK2602).

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supporting information

Acta Cryst. (2014). E70, o309 [doi:10.1107/S160053681400316X]

3-(3-Bromophenylsulfinyl)-5-cyclohexyl-2-methyl-1-benzofuran

Hong Dae Choi, Pil Ja Seo and Uk Lee

S1. Comment

As a part of our continuing study of 5-cyclohexyl-2-methyl-1-benzofuran derivatives containing 4-fluorophenylsulfinyl (Choi *et al.*, 2011), 4-bromophenylsulfinyl (Choi *et al.*, 2012*a*) and 4-methylphenylsulfinyl (Choi *et al.*, 2012*b*) substituents in 3-position, we report herein the crystal structure of the title compound.

In the title molecule (Fig. 1), the cyclohexyl ring has a chair conformation. The benzofuran ring system is essentially planar, with a mean deviation of 0.178 (2) Å from the least-squares plane defined by the nine constituent atoms. The 3-bromophenyl ring is essentially planar, with a mean deviation of 0.005 (2) Å from the least-squares plane defined by the six constituent atoms. The dihedral angle formed by the benzofuran ring system and the 3-bromophenyl ring is 86.52 (6)°. In the crystal structure (Fig. 2), molecules are connected by C—H...O and C—H... π hydrogen bonds (Table 1, Cg1 is the centroid of the C2–C7 benzene ring). The crystal packing (Fig. 2) also exhibits a slipped π ... π interaction between the furan rings of neighbouring molecules, with a Cg2...Cg2ⁱⁱⁱ distance of 3.518 (3) Å and an interplanar distance of 3.471 (3) Å resulting in a slippage of 0.573 (3) Å (Cg2 is the centroid of the C1/C2/C7/O1/C8 furan ring).

S2. Experimental

3-Chloroperoxybenzoic acid (77%, 224 mg, 1.0 mmol) was added in small portions to a stirred solution of 3-(3-bromophenylsulfonyl)-5-cyclohexyl-2-methyl-1-benzofuran (361 mg, 0.9 mmol) in dichloromethane (40 mL) at 273 K. After being stirred at room temperature for 4h, the mixture was washed with saturated sodium bicarbonate solution and the organic layer was separated, dried over magnesium sulfate, filtered and concentrated at reduced pressure. The residue was purified by column chromatography (hexane–ethyl acetate, 2:1 v/v) to afford the title compound as a colorless solid [yield 71%, m.p. 415–416 K; R_f = 0.47 (hexane–ethyl acetate, 2:1 v/v)]. Single crystals suitable for X-ray diffraction were prepared by slow evaporation of a solution of the title compound in ethyl acetate at room temperature.

S3. Refinement

All H atoms were positioned geometrically and refined using a riding model, with C—H = 0.95 Å for aryl, 1.00 Å for methine, 0.99 Å for methylene and 0.98 Å for methyl H atoms, respectively. $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aryl, methine and methylene, and $1.5U_{\text{eq}}(\text{C})$ for methyl H atoms. The positions of methyl hydrogens were optimized rotationally.

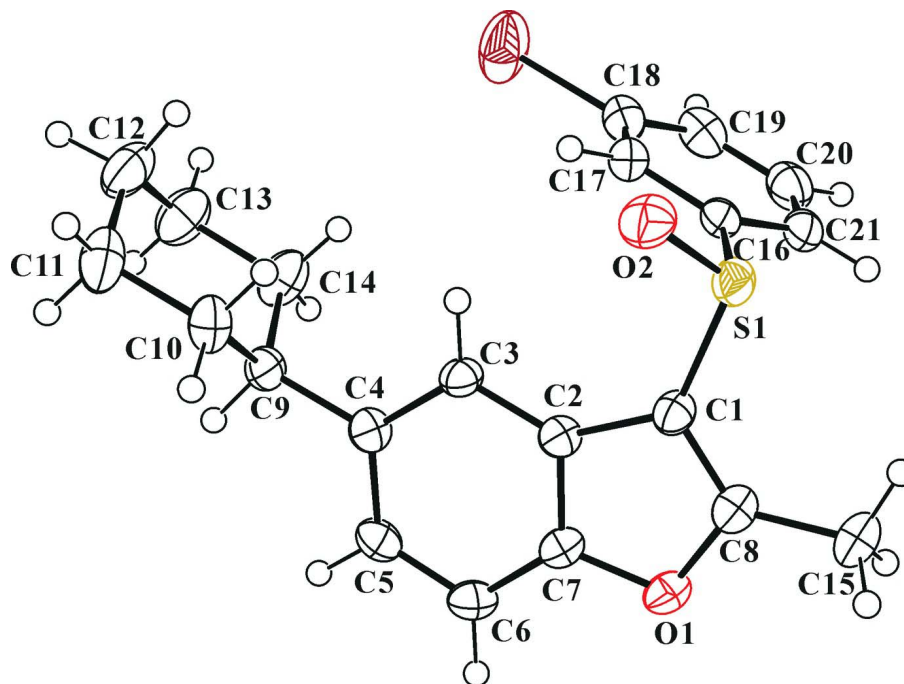


Figure 1

The molecular structure of the title compound with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are presented as small spheres of arbitrary radius.

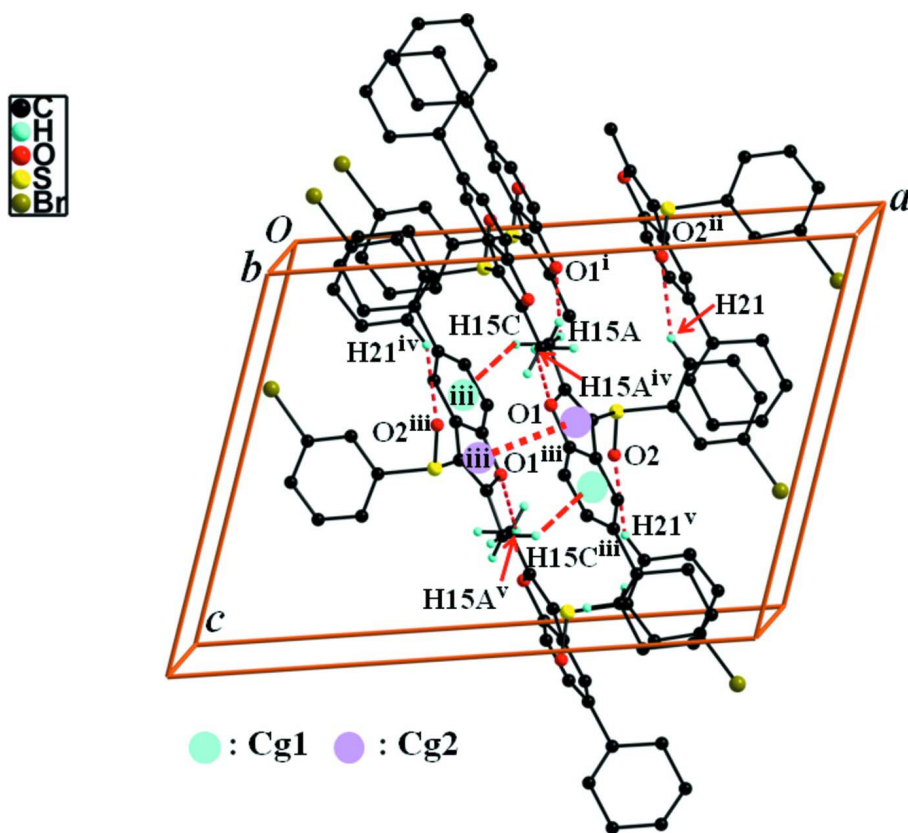


Figure 2

A view of the C—H \cdots O, C—H \cdots π and $\pi\cdots\pi$ interactions (dotted lines) in the crystal structure of the title compound. H atoms non-participating in hydrogen bonding were omitted for clarity. [Symmetry codes : (i) $-x + 1, y - 1/2, -z + 1/2$; (ii) $x, -y + 1/2, z - 1/2$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $-x + 1, y + 1/2, -z + 1/2$; (v) $x, -y + 1/2, z + 1/2$.]

3-(3-Bromophenylsulfinyl)-5-cyclohexyl-2-methyl-1-benzofuran

Crystal data

$C_{21}H_{21}BrO_2S$
 $M_r = 417.35$
 Monoclinic, $P2_1/c$
 Hall symbol: $-P 2_1/c$
 $a = 17.6432$ (9) Å
 $b = 8.9425$ (5) Å
 $c = 12.4776$ (7) Å
 $\beta = 108.277$ (3)°
 $V = 1869.33$ (18) Å³
 $Z = 4$

$F(000) = 856$
 $D_x = 1.483$ Mg m⁻³
 Melting point = 415–416 K
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
 Cell parameters from 5628 reflections
 $\theta = 2.4$ – 27.9 °
 $\mu = 2.32$ mm⁻¹
 $T = 173$ K
 Block, colourless
 $0.33 \times 0.25 \times 0.18$ mm

Data collection

Bruker SMART APEXII CCD
 diffractometer
 Radiation source: rotating anode
 Graphite multilayer monochromator
 Detector resolution: 10.0 pixels mm⁻¹
 φ and ω scans

Absorption correction: multi-scan
 (SADABS; Bruker, 2009)
 $T_{\min} = 0.482, T_{\max} = 0.746$
 13952 measured reflections
 3290 independent reflections
 2635 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$
 $\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 2.4^\circ$
 $h = -20 \rightarrow 20$

$k = -9 \rightarrow 10$
 $l = -14 \rightarrow 14$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.031$
 $wR(F^2) = 0.081$
 $S = 1.02$
 3290 reflections
 227 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: difference Fourier map
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0334P)^2 + 1.1209P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.33 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.35 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Br1	0.957798 (17)	0.38271 (5)	0.67946 (3)	0.07366 (16)
S1	0.63694 (3)	0.26562 (6)	0.46130 (5)	0.03447 (16)
O1	0.53393 (8)	0.66112 (17)	0.40386 (12)	0.0319 (4)
O2	0.64909 (10)	0.19977 (18)	0.57522 (14)	0.0427 (4)
C1	0.60397 (12)	0.4510 (2)	0.46411 (18)	0.0282 (5)
C2	0.63088 (12)	0.5613 (2)	0.55301 (17)	0.0256 (5)
C3	0.68650 (12)	0.5662 (2)	0.66079 (18)	0.0273 (5)
H3	0.7173	0.4802	0.6915	0.033*
C4	0.69659 (12)	0.6982 (2)	0.72299 (18)	0.0280 (5)
C5	0.64951 (13)	0.8227 (3)	0.67603 (19)	0.0319 (5)
H5	0.6567	0.9126	0.7186	0.038*
C6	0.59308 (13)	0.8197 (3)	0.57033 (19)	0.0336 (5)
H6	0.5612	0.9046	0.5399	0.040*
C7	0.58539 (12)	0.6878 (2)	0.51138 (18)	0.0274 (5)
C8	0.54640 (12)	0.5152 (3)	0.37802 (18)	0.0296 (5)
C9	0.75683 (13)	0.7056 (3)	0.84023 (18)	0.0316 (5)
H9	0.7543	0.8085	0.8704	0.038*
C10	0.73622 (14)	0.5948 (3)	0.92025 (19)	0.0380 (6)
H10A	0.6824	0.6180	0.9245	0.046*
H10B	0.7348	0.4924	0.8895	0.046*
C11	0.79601 (17)	0.5999 (3)	1.0382 (2)	0.0499 (7)
H11A	0.7937	0.6990	1.0723	0.060*

H11B	0.7820	0.5231	1.0860	0.060*
C12	0.87977 (16)	0.5719 (3)	1.0348 (2)	0.0515 (7)
H12A	0.8834	0.4687	1.0081	0.062*
H12B	0.9179	0.5810	1.1120	0.062*
C13	0.90249 (16)	0.6819 (4)	0.9573 (2)	0.0512 (7)
H13A	0.9560	0.6560	0.9528	0.061*
H13B	0.9054	0.7839	0.9892	0.061*
C14	0.84220 (14)	0.6799 (3)	0.8393 (2)	0.0433 (6)
H14A	0.8451	0.5822	0.8035	0.052*
H14B	0.8564	0.7586	0.7931	0.052*
C15	0.49706 (14)	0.4623 (3)	0.26513 (19)	0.0373 (6)
H15A	0.5103	0.3578	0.2551	0.056*
H15B	0.5078	0.5241	0.2067	0.056*
H15C	0.4404	0.4699	0.2589	0.056*
C16	0.73449 (14)	0.3121 (2)	0.45342 (18)	0.0319 (5)
C17	0.79756 (14)	0.3239 (3)	0.55204 (19)	0.0353 (5)
H17	0.7903	0.3067	0.6233	0.042*
C18	0.87176 (15)	0.3615 (3)	0.5440 (2)	0.0418 (6)
C19	0.88362 (16)	0.3840 (3)	0.4411 (2)	0.0451 (6)
H19	0.9350	0.4092	0.4373	0.054*
C20	0.81998 (17)	0.3694 (3)	0.3439 (2)	0.0446 (7)
H20	0.8278	0.3846	0.2727	0.054*
C21	0.74495 (16)	0.3330 (3)	0.34816 (19)	0.0378 (6)
H21	0.7013	0.3224	0.2808	0.045*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Br1	0.03672 (18)	0.1296 (4)	0.0485 (2)	0.00172 (17)	0.00453 (14)	-0.02246 (18)
S1	0.0418 (3)	0.0248 (3)	0.0302 (3)	-0.0027 (2)	0.0018 (3)	-0.0026 (2)
O1	0.0302 (8)	0.0323 (10)	0.0295 (8)	0.0027 (7)	0.0040 (7)	0.0072 (7)
O2	0.0528 (10)	0.0314 (10)	0.0403 (10)	0.0006 (8)	0.0097 (8)	0.0081 (8)
C1	0.0291 (11)	0.0263 (13)	0.0268 (11)	-0.0029 (9)	0.0054 (9)	0.0020 (10)
C2	0.0262 (11)	0.0219 (12)	0.0287 (12)	-0.0036 (9)	0.0085 (9)	0.0008 (9)
C3	0.0277 (11)	0.0225 (12)	0.0288 (12)	0.0012 (9)	0.0046 (9)	0.0028 (9)
C4	0.0283 (11)	0.0252 (13)	0.0309 (12)	-0.0043 (9)	0.0099 (10)	-0.0016 (10)
C5	0.0370 (12)	0.0217 (12)	0.0380 (13)	0.0002 (10)	0.0132 (11)	-0.0040 (10)
C6	0.0335 (12)	0.0271 (13)	0.0392 (13)	0.0071 (10)	0.0100 (11)	0.0060 (11)
C7	0.0249 (11)	0.0282 (13)	0.0279 (11)	0.0000 (9)	0.0065 (9)	0.0047 (10)
C8	0.0288 (11)	0.0300 (13)	0.0289 (12)	-0.0053 (10)	0.0074 (10)	0.0030 (10)
C9	0.0367 (12)	0.0239 (13)	0.0297 (12)	-0.0019 (10)	0.0039 (10)	-0.0051 (10)
C10	0.0412 (14)	0.0461 (16)	0.0284 (12)	-0.0063 (11)	0.0133 (11)	-0.0051 (11)
C11	0.0648 (18)	0.0553 (19)	0.0289 (13)	-0.0076 (14)	0.0135 (13)	-0.0022 (12)
C12	0.0511 (16)	0.0556 (18)	0.0351 (14)	-0.0001 (13)	-0.0049 (13)	0.0036 (13)
C13	0.0372 (14)	0.0610 (19)	0.0454 (16)	-0.0093 (13)	-0.0016 (12)	0.0001 (14)
C14	0.0335 (13)	0.0559 (17)	0.0375 (14)	-0.0107 (12)	0.0067 (11)	0.0049 (12)
C15	0.0336 (12)	0.0450 (16)	0.0275 (12)	-0.0061 (11)	0.0012 (10)	0.0021 (11)
C16	0.0426 (13)	0.0197 (12)	0.0293 (12)	0.0048 (10)	0.0054 (10)	-0.0030 (9)

C17	0.0420 (14)	0.0350 (14)	0.0266 (12)	0.0079 (11)	0.0073 (11)	-0.0029 (10)
C18	0.0408 (14)	0.0444 (16)	0.0372 (14)	0.0054 (11)	0.0078 (11)	-0.0099 (12)
C19	0.0492 (15)	0.0421 (16)	0.0476 (16)	0.0020 (12)	0.0205 (13)	-0.0061 (12)
C20	0.0670 (18)	0.0348 (15)	0.0366 (14)	0.0071 (13)	0.0226 (14)	0.0012 (11)
C21	0.0556 (16)	0.0256 (13)	0.0267 (12)	0.0084 (11)	0.0050 (11)	-0.0009 (10)

Geometric parameters (Å, °)

Br1—C18	1.894 (2)	C11—C12	1.512 (4)
S1—O2	1.4907 (17)	C11—H11A	0.9900
S1—C1	1.761 (2)	C11—H11B	0.9900
S1—C16	1.803 (2)	C12—C13	1.519 (4)
O1—C8	1.378 (3)	C12—H12A	0.9900
O1—C7	1.385 (3)	C12—H12B	0.9900
C1—C8	1.353 (3)	C13—C14	1.522 (3)
C1—C2	1.449 (3)	C13—H13A	0.9900
C2—C7	1.390 (3)	C13—H13B	0.9900
C2—C3	1.395 (3)	C14—H14A	0.9900
C3—C4	1.393 (3)	C14—H14B	0.9900
C3—H3	0.9500	C15—H15A	0.9800
C4—C5	1.402 (3)	C15—H15B	0.9800
C4—C9	1.516 (3)	C15—H15C	0.9800
C5—C6	1.381 (3)	C16—C17	1.380 (3)
C5—H5	0.9500	C16—C21	1.395 (3)
C6—C7	1.374 (3)	C17—C18	1.386 (4)
C6—H6	0.9500	C17—H17	0.9500
C8—C15	1.482 (3)	C18—C19	1.379 (4)
C9—C14	1.527 (3)	C19—C20	1.376 (4)
C9—C10	1.529 (3)	C19—H19	0.9500
C9—H9	1.0000	C20—C21	1.380 (4)
C10—C11	1.519 (3)	C20—H20	0.9500
C10—H10A	0.9900	C21—H21	0.9500
C10—H10B	0.9900		
O2—S1—C1	107.61 (10)	C10—C11—H11B	109.5
O2—S1—C16	106.78 (10)	H11A—C11—H11B	108.1
C1—S1—C16	96.38 (10)	C11—C12—C13	111.5 (2)
C8—O1—C7	106.61 (16)	C11—C12—H12A	109.3
C8—C1—C2	107.62 (19)	C13—C12—H12A	109.3
C8—C1—S1	123.66 (17)	C11—C12—H12B	109.3
C2—C1—S1	128.71 (16)	C13—C12—H12B	109.3
C7—C2—C3	118.9 (2)	H12A—C12—H12B	108.0
C7—C2—C1	104.83 (18)	C12—C13—C14	111.4 (2)
C3—C2—C1	136.2 (2)	C12—C13—H13A	109.3
C4—C3—C2	119.5 (2)	C14—C13—H13A	109.3
C4—C3—H3	120.3	C12—C13—H13B	109.3
C2—C3—H3	120.3	C14—C13—H13B	109.3
C3—C4—C5	119.0 (2)	H13A—C13—H13B	108.0

C3—C4—C9	120.2 (2)	C13—C14—C9	112.5 (2)
C5—C4—C9	120.8 (2)	C13—C14—H14A	109.1
C6—C5—C4	122.6 (2)	C9—C14—H14A	109.1
C6—C5—H5	118.7	C13—C14—H14B	109.1
C4—C5—H5	118.7	C9—C14—H14B	109.1
C7—C6—C5	116.6 (2)	H14A—C14—H14B	107.8
C7—C6—H6	121.7	C8—C15—H15A	109.5
C5—C6—H6	121.7	C8—C15—H15B	109.5
C6—C7—O1	126.27 (19)	H15A—C15—H15B	109.5
C6—C7—C2	123.4 (2)	C8—C15—H15C	109.5
O1—C7—C2	110.36 (19)	H15A—C15—H15C	109.5
C1—C8—O1	110.57 (19)	H15B—C15—H15C	109.5
C1—C8—C15	133.6 (2)	C17—C16—C21	121.4 (2)
O1—C8—C15	115.83 (19)	C17—C16—S1	119.04 (18)
C4—C9—C14	112.37 (19)	C21—C16—S1	119.52 (18)
C4—C9—C10	111.31 (18)	C16—C17—C18	118.0 (2)
C14—C9—C10	110.16 (19)	C16—C17—H17	121.0
C4—C9—H9	107.6	C18—C17—H17	121.0
C14—C9—H9	107.6	C19—C18—C17	121.7 (2)
C10—C9—H9	107.6	C19—C18—Br1	120.2 (2)
C11—C10—C9	112.2 (2)	C17—C18—Br1	118.06 (19)
C11—C10—H10A	109.2	C20—C19—C18	119.1 (2)
C9—C10—H10A	109.2	C20—C19—H19	120.4
C11—C10—H10B	109.2	C18—C19—H19	120.4
C9—C10—H10B	109.2	C19—C20—C21	121.0 (2)
H10A—C10—H10B	107.9	C19—C20—H20	119.5
C12—C11—C10	110.8 (2)	C21—C20—H20	119.5
C12—C11—H11A	109.5	C20—C21—C16	118.7 (2)
C10—C11—H11A	109.5	C20—C21—H21	120.7
C12—C11—H11B	109.5	C16—C21—H21	120.7
O2—S1—C1—C8	-139.65 (19)	C7—O1—C8—C15	-179.84 (18)
C16—S1—C1—C8	110.4 (2)	C3—C4—C9—C14	62.5 (3)
O2—S1—C1—C2	40.6 (2)	C5—C4—C9—C14	-118.4 (2)
C16—S1—C1—C2	-69.3 (2)	C3—C4—C9—C10	-61.6 (3)
C8—C1—C2—C7	0.1 (2)	C5—C4—C9—C10	117.5 (2)
S1—C1—C2—C7	179.85 (17)	C4—C9—C10—C11	179.8 (2)
C8—C1—C2—C3	179.3 (2)	C14—C9—C10—C11	54.5 (3)
S1—C1—C2—C3	-0.9 (4)	C9—C10—C11—C12	-56.3 (3)
C7—C2—C3—C4	-1.6 (3)	C10—C11—C12—C13	56.0 (3)
C1—C2—C3—C4	179.3 (2)	C11—C12—C13—C14	-55.0 (3)
C2—C3—C4—C5	0.9 (3)	C12—C13—C14—C9	54.1 (3)
C2—C3—C4—C9	-179.93 (19)	C4—C9—C14—C13	-178.0 (2)
C3—C4—C5—C6	0.2 (3)	C10—C9—C14—C13	-53.3 (3)
C9—C4—C5—C6	-178.9 (2)	O2—S1—C16—C17	-18.3 (2)
C4—C5—C6—C7	-0.7 (3)	C1—S1—C16—C17	92.32 (19)
C5—C6—C7—O1	-178.7 (2)	O2—S1—C16—C21	160.74 (18)
C5—C6—C7—C2	0.0 (3)	C1—S1—C16—C21	-88.7 (2)

C8—O1—C7—C6	179.7 (2)	C21—C16—C17—C18	1.8 (3)
C8—O1—C7—C2	0.8 (2)	S1—C16—C17—C18	-179.20 (18)
C3—C2—C7—C6	1.1 (3)	C16—C17—C18—C19	-1.3 (4)
C1—C2—C7—C6	-179.5 (2)	C16—C17—C18—Br1	178.45 (18)
C3—C2—C7—O1	-179.96 (18)	C17—C18—C19—C20	0.3 (4)
C1—C2—C7—O1	-0.5 (2)	Br1—C18—C19—C20	-179.37 (19)
C2—C1—C8—O1	0.5 (2)	C18—C19—C20—C21	0.1 (4)
S1—C1—C8—O1	-179.36 (14)	C19—C20—C21—C16	0.4 (4)
C2—C1—C8—C15	179.3 (2)	C17—C16—C21—C20	-1.4 (3)
S1—C1—C8—C15	-0.5 (4)	S1—C16—C21—C20	179.61 (18)
C7—O1—C8—C1	-0.8 (2)		

Hydrogen-bond geometry (Å, °)

Cg1 is the centroid of the C2–C7 benzene ring.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C15—H15 <i>A</i> ...O1 ⁱ	0.98	2.58	3.358 (3)	136
C21—H21...O2 ⁱⁱ	0.95	2.45	3.306 (3)	151
C15—H15 <i>C</i> ...Cg1 ⁱⁱⁱ	0.98	2.82	3.502 (3)	127

Symmetry codes: (i) $-x+1, y-1/2, -z+1/2$; (ii) $x, -y+1/2, z-1/2$; (iii) $-x+1, -y+1, -z+1$.