

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## 2-(Phenylselenonyl)pyridine

 Shivani Gulati,<sup>a</sup> K. K. Bhasin,<sup>b</sup> V. A. Potapov,<sup>c</sup> Ekta Arora<sup>b</sup> and Ray J. Butcher<sup>d\*</sup>
<sup>a</sup>Department of Chemistry, D. A. V. College, Sector-10, Chandigarh, India,

<sup>b</sup>Department of Chemistry & Centre of Advanced Studies in Chemistry, Panjab University, Chandigarh 160 014, India, <sup>c</sup>A.E. Favorsky Irkutsk Institute of Chemistry, 1 Favorsky Street, Irkutsk, RUS-664033, Russian Federation, and <sup>d</sup>Department of Chemistry, Howard University, 525 College Street NW, Washington, DC 20059, USA

Correspondence e-mail: rbutcher99@yahoo.com

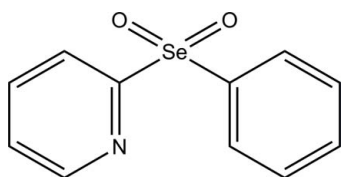
Received 12 June 2013; accepted 1 November 2013

 Key indicators: single-crystal X-ray study;  $T = 123$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.047;  $wR$  factor = 0.090; data-to-parameter ratio = 38.2.

In the title compound,  $\text{C}_{11}\text{H}_9\text{NO}_2\text{Se}$ , the pyridine and phenyl rings are almost perpendicular, with the dihedral angle between their mean planes being  $79.16(7)^\circ$ . In the crystal, the molecules pack so as to form ruffled sheets in the (110) plane connected by weak  $\text{C}-\text{H}\cdots\text{O}$  interactions. In addition, there are weak  $\pi-\pi$  interactions between the mean planes of both the phenyl [centroid-centroid perpendicular distance of  $3.591(2)$  Å and slippage of  $1.854(2)$  Å] and pyridine rings [centroid-centroid perpendicular distance of  $3.348(2)$  Å and slippage of  $1.854(2)$  Å].

## Related literature

For the pharmacological activity of selenone derivatives, see: Abdel-Hafez & Hussein (2008); Zhao *et al.* (2012); Hassan *et al.* (2011); Bhabak *et al.* (2011). For the chemistry of selenium compounds bonded directly to pyridine, see: Bhasin *et al.* (2013). For the synthesis of pharmaceuticals, see: Nogueira & Rocha (2011). For the synthesis of perfumes, fine chemicals and polymers, see: Zeng *et al.* (2013).



## Experimental

## Crystal data

|   |                    |
|---|--------------------|
| $\text{C}_{11}\text{H}_9\text{NO}_2\text{Se}$ | $a = 6.1598(5)$ Å  |
| $M_r = 266.15$                                | $b = 7.7223(6)$ Å  |
| Triclinic, $P\bar{1}$                         | $c = 11.4952(7)$ Å |

|                                |
|--------------------------------|
| $\alpha = 80.683(6)^\circ$     |
| $\beta = 83.494(6)^\circ$      |
| $\gamma = 74.614(7)^\circ$     |
| $V = 518.83(7)$ Å <sup>3</sup> |
| $Z = 2$                        |

|                                   |
|-----------------------------------|
| Mo $K\alpha$ radiation            |
| $\mu = 3.60$ mm <sup>-1</sup>     |
| $T = 123$ K                       |
| $0.50 \times 0.26 \times 0.16$ mm |

## Data collection

|  |
|--|
| Agilent Xcalibur (Ruby, Gemini) diffractometer   |
| Absorption correction: analytical ( <i>CrysAlis PRO</i> and <i>CrysAlis RED</i> ; Agilent, 2012) |
| $T_{\min} = 0.383$ , $T_{\max} = 0.613$  |

|  |
|--|
| 8688 measured reflections              |
| 5196 independent reflections           |
| 3965 reflections with $I > 2\sigma(I)$ |
| $R_{\text{int}} = 0.041$               |

## Refinement

|                                 |
|---------------------------------|
| $R[F^2 > 2\sigma(F^2)] = 0.047$ |
| $wR(F^2) = 0.090$               |
| $S = 1.01$                      |
| 5196 reflections                |

|   |
|---|
| 136 parameters                                |
| H-atom parameters constrained                 |
| $\Delta\rho_{\max} = 0.64$ e Å <sup>-3</sup>  |
| $\Delta\rho_{\min} = -0.76$ e Å <sup>-3</sup> |

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$  | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|-------|-------------|-------------|---------------|
| $\text{C2A}-\text{H2AA}\cdots\text{O1}^{\text{i}}$   | 0.95  | 2.50        | 3.331 (3)   | 146           |
| $\text{C4A}-\text{H4AA}\cdots\text{O1}^{\text{ii}}$  | 0.95  | 2.53        | 3.341 (3)   | 143           |
| $\text{C5A}-\text{H5AA}\cdots\text{O2}^{\text{iii}}$ | 0.95  | 2.35        | 3.188 (3)   | 146           |

 Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $x - 1, y + 1, z$ ; (iii)  $x, y + 1, z$ .

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

This work was supported by the Department of Science and Technology, DST, New Delhi (Research Grant SR/S1/IC-37/2009). RJB wishes to acknowledge the NSF-MRI program (grant CHE-0619278) for funds to purchase the diffractometer as well as the Howard University Nanoscience Facility for access to liquid nitrogen.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: JJ2170).

## References

- Abdel-Hafez, S. H. & Hussein, M. A. (2008). *Arch. Pharm.* **341**, 240–246.
- Agilent (2012). *CrysAlis PRO* and *CrysAlis RED*. Agilent Technologies, Yarnton, England.
- Bhabak, K. P., Vernekar, A. A., Jakka, S. R., Roy, G. & Mughesh, G. (2011). *Org. Biomol. Chem.* **9**, 5193–5200.
- Bhasin K. K., Arora E., Grover A. S., Jyoti, Singh. H., Mehta S. K, Bhasin, A. K. K. & Jacob C. (2013). *J. Organomet. Chem.* **732**, 137–141.
- Hassan, W., Narayanaperumal, S., Santos, M. M., Gul, K., Mohammadzai, I. U., Braga, A. L., Rodrigues, O. D. & Rocha, J. B. T. (2011). *Pharm. Anal. Acta*. doi:10.4172/2153-2435.S3-002.
- Nogueira, C. W. & Rocha, J. B. (2011). *Arch. Toxicol.* **85**, 1313–1359.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Zeng, J., Zhu, J., Pan, X., Zhang, Z., Zhou, N., Cheng, Z., Zhang, W. & Zhu, X. (2013). *Polym. Chem. Advance Article*.
- Zhao, L., Li, J., Li, Y., Liu, J., Wirth, T. & Li, Z. (2012). *Bioorg. Med. Chem.* **20**, 2558–2563.

## supporting information

*Acta Cryst.* (2013). E69, o1791 [doi:10.1107/S1600536813029978]

## 2-(Phenylselenonyl)pyridine

Shivani Gulati, K. K. Bhasin, V. A. Potapov, Ekta Arora and Ray J. Butcher

### S1. Comment

Organochalcogen compounds, especially containing selenium have continued to attract attention of researchers in academia as anti-cancer (Zhao *et al.*, 2012), anti-oxidant (Hassan *et al.*, 2011; Bhabak *et al.*, 2011), anti-inflammatory and anti-allergic agents (Abdel-Hafez, & Hussein, 2008), and in industry because of their wide involvement as key intermediates for the synthesis of pharmaceuticals (Nogueira, & Rocha, 2011), perfumes, fine chemicals and polymers (Zeng *et al.*, 2013). Curiously, compared to alkyl, aryl and mixed alkyl aryl selenium compounds, the chemistry of selenium compounds bonded directly to pyridine has not yet been exploited extensively (Bhasin *et al.* 2013). In continuation of our ongoing program directed at the synthesis of novel organoselenium derivatives, we report here the synthesis and crystal structure of 2-(phenylselenonyl)pyridine.

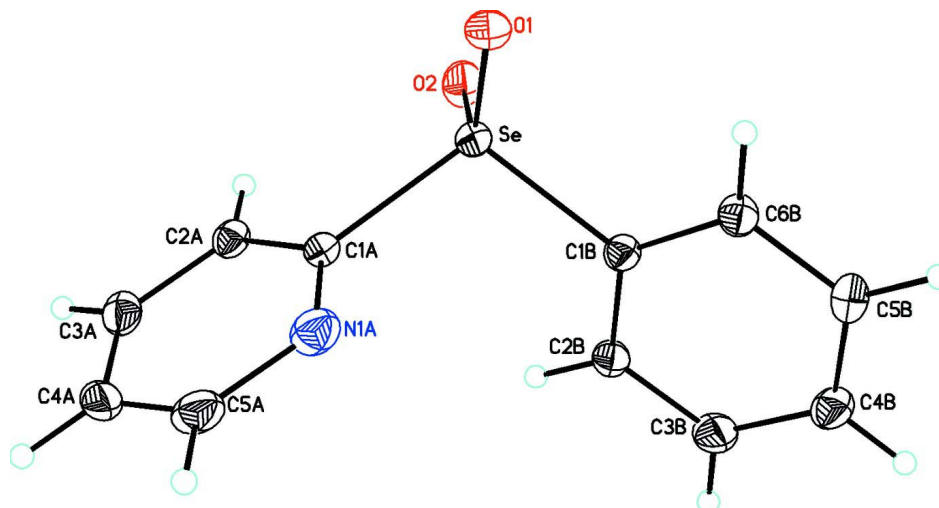
In the title compound, C<sub>11</sub>H<sub>9</sub>NO<sub>2</sub>Se, (I), the pyridine and phenyl rings are almost perpendicular with the dihedral angle between the mean planes being 79.16 (7)° (Fig. 1). The molecules pack so as to form ruffled sheets in the (1 1 0) plane connected by weak C—H···O intermolecular interactions (Fig. 2). In addition there are weak  $\pi$ – $\pi$  interactions between both the phenyl groups (Cg···Cg perpendicular distance of 3.591 (2) Å with slippage of 1.854 (2) Å [2 -x, -y, 1 - z]) and pyridine rings (Cg···Cg perpendicular distance of 3.348 (2) Å with slippage of 1.854 (2) Å [1 -x, 1 - y, -z]) (Fig. 3).

### S2. Experimental

A stirred solution of 2-(phenylseleninyl)pyridine (0.235 g, 1 mmol) in glacial acetic acid (10 ml) was treated with (0.550 g, 3.5 mmol) potassium permanganate in small amounts. The reaction mixture was allowed to stir for 3 h at room temperature. The progress of the reaction mixture was monitored by thin layer chromatography. After completion of the reaction, the reaction mixture was neutralized with excess of saturated solution of sodium bicarbonate and extracted with dichloromethane (4 x 25 ml). The combined organic extracts were washed with water and dried over anhydrous MgSO<sub>4</sub>. Dichloromethane was removed on a rota-evaporator that yielded a white powder. Single crystals of the compound suitable for XRD were prepared by dissolving the obtained white powder in a (1:1) mixture of CHCl<sub>3</sub> and CCl<sub>4</sub> followed by slow evaporation. Yield = 85%. *M.p.* = 453–455°K.

### S3. Refinement

All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms with a C—H distance of 0.95 and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

**Figure 1**

Molecular diagram of (I) illustrating the atom numbering scheme used. Thermal ellipsoids are at the 30% probability level.

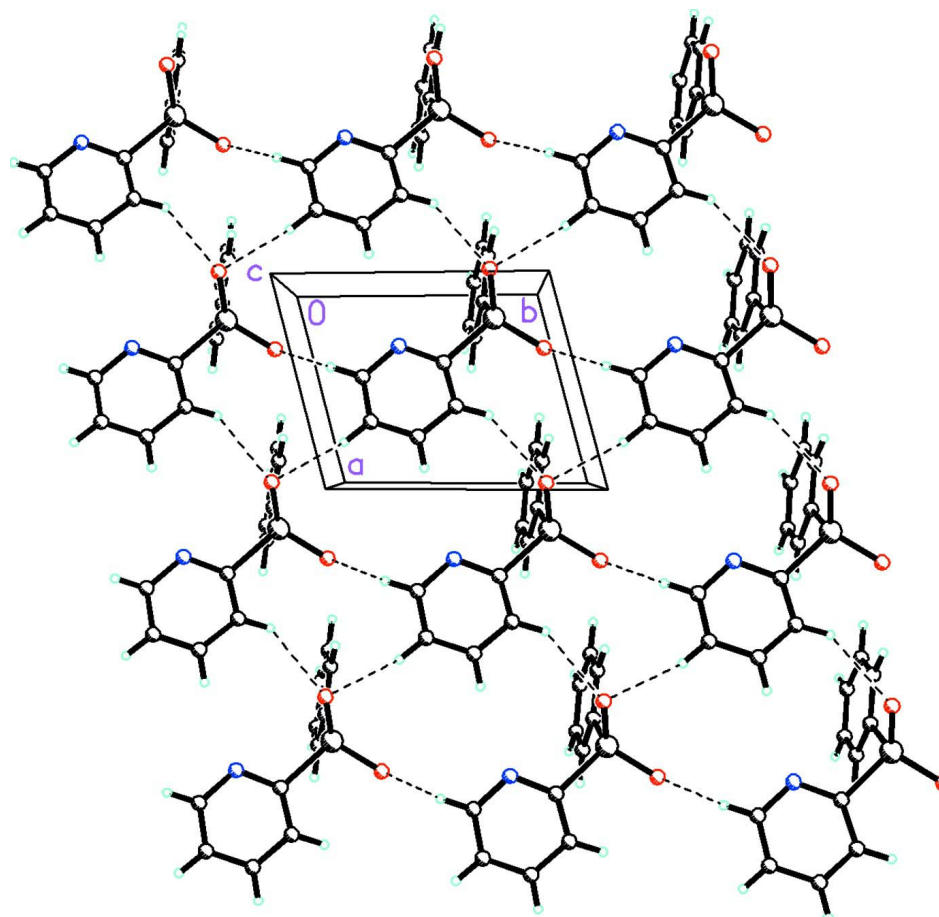
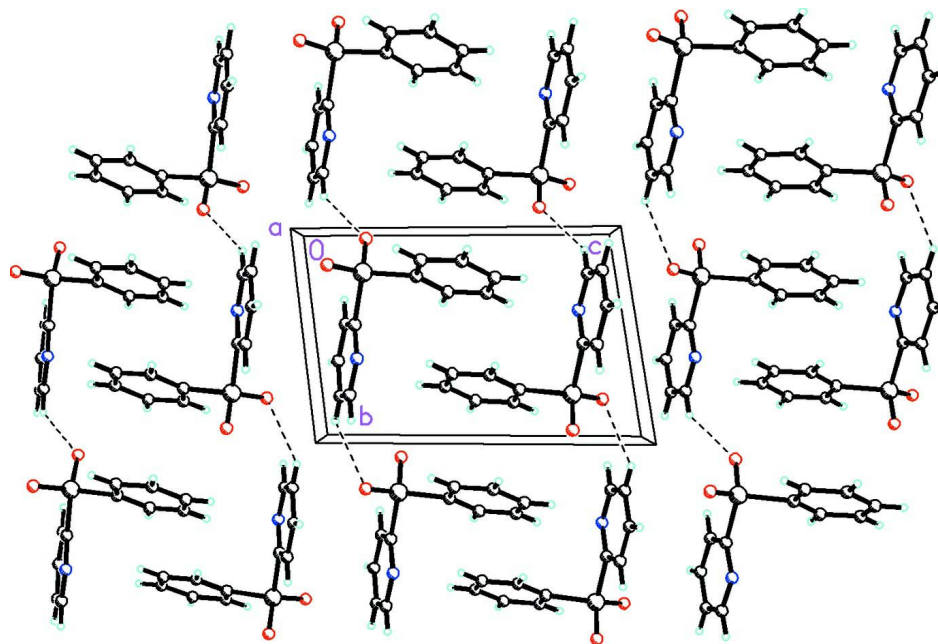
**Figure 2**

Fig. 2. Molecular packing for (I) viewed along the *c* axis. Dashed lines indicate the weak C—H...O intermolecular interactions forming ruffled sheets in the (1 1 0) plane.

**Figure 3**

Molecular packing for (I) showing the  $\pi$ - $\pi$  interactions between the mean planes of both the phenyl and pyridine rings.

## 2-(Phenylselenonyl)pyridine

### Crystal data

$C_{11}H_9NO_2Se$

$M_r = 266.15$

Triclinic,  $P\bar{1}$

Hall symbol: -P 1

$a = 6.1598$  (5) Å

$b = 7.7223$  (6) Å

$c = 11.4952$  (7) Å

$\alpha = 80.683$  (6)°

$\beta = 83.494$  (6)°

$\gamma = 74.614$  (7)°

$V = 518.83$  (7) Å<sup>3</sup>

$Z = 2$

$F(000) = 264$

$D_x = 1.704$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 2941 reflections

$\theta = 3.1$ – $37.5$ °

$\mu = 3.60$  mm<sup>-1</sup>

$T = 123$  K

Triangular plate, colorless

$0.50 \times 0.26 \times 0.16$  mm

### Data collection

Agilent Xcalibur (Ruby, Gemini)

diffractometer

Radiation source: Enhance (Mo) X-ray Source

Graphite monochromator

Detector resolution: 10.5081 pixels mm<sup>-1</sup>

$\omega$  scans

Absorption correction: analytical

(*CrysAlis PRO* and *CrysAlis RED*; Agilent, 2012)

$T_{\min} = 0.383$ ,  $T_{\max} = 0.613$

8688 measured reflections

5196 independent reflections

3965 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.041$

$\theta_{\max} = 37.6$ °,  $\theta_{\min} = 3.1$ °

$h = -10 \rightarrow 10$

$k = -12 \rightarrow 13$

$l = -18 \rightarrow 19$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.090$   
 $S = 1.01$   
 5196 reflections  
 136 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0241P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.64 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.76 \text{ e } \text{\AA}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|      | $x$         | $y$         | $z$           | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|-------------|-------------|---------------|----------------------------------|
| Se   | 0.80434 (4) | 0.23043 (3) | 0.224458 (18) | 0.02645 (6)                      |
| O1   | 1.0283 (3)  | 0.2030 (2)  | 0.13351 (14)  | 0.0370 (4)                       |
| O2   | 0.6629 (3)  | 0.0770 (2)  | 0.24217 (15)  | 0.0380 (4)                       |
| N1A  | 0.6710 (4)  | 0.6075 (3)  | 0.17338 (18)  | 0.0397 (5)                       |
| C1A  | 0.5932 (3)  | 0.4583 (3)  | 0.17700 (17)  | 0.0247 (4)                       |
| C2A  | 0.3857 (3)  | 0.4556 (3)  | 0.14449 (17)  | 0.0277 (4)                       |
| H2AA | 0.3426      | 0.3453      | 0.1487        | 0.033*                           |
| C3A  | 0.2427 (4)  | 0.6216 (4)  | 0.10521 (19)  | 0.0358 (5)                       |
| H3AA | 0.0978      | 0.6274      | 0.0816        | 0.043*                           |
| C4A  | 0.3131 (5)  | 0.7786 (3)  | 0.1008 (2)    | 0.0427 (6)                       |
| H4AA | 0.2166      | 0.8929      | 0.0741        | 0.051*                           |
| C5A  | 0.5245 (5)  | 0.7685 (3)  | 0.1354 (2)    | 0.0453 (7)                       |
| H5AA | 0.5696      | 0.8777      | 0.1327        | 0.054*                           |
| C1B  | 0.8984 (4)  | 0.2491 (3)  | 0.37495 (17)  | 0.0256 (4)                       |
| C2B  | 0.7332 (4)  | 0.3166 (3)  | 0.45649 (18)  | 0.0298 (4)                       |
| H2BA | 0.5799      | 0.3593      | 0.4382        | 0.036*                           |
| C3B  | 0.7969 (4)  | 0.3207 (4)  | 0.5661 (2)    | 0.0399 (6)                       |
| H3BA | 0.6853      | 0.3666      | 0.6251        | 0.048*                           |
| C4B  | 1.0186 (4)  | 0.2598 (3)  | 0.5925 (2)    | 0.0369 (5)                       |
| H4BA | 1.0587      | 0.2646      | 0.6691        | 0.044*                           |
| C5B  | 1.1842 (4)  | 0.1913 (3)  | 0.50800 (19)  | 0.0339 (5)                       |
| H5BA | 1.3373      | 0.1479      | 0.5267        | 0.041*                           |
| C6B  | 1.1244 (4)  | 0.1864 (3)  | 0.39527 (18)  | 0.0290 (4)                       |
| H6BA | 1.2344      | 0.1419      | 0.3351        | 0.035*                           |

Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$     | $U^{22}$     | $U^{33}$     | $U^{12}$     | $U^{13}$     | $U^{23}$     |
|-----|--------------|--------------|--------------|--------------|--------------|--------------|
| Se  | 0.02928 (11) | 0.02457 (10) | 0.02608 (11) | -0.00452 (8) | -0.00550 (7) | -0.00652 (8) |
| O1  | 0.0355 (9)   | 0.0416 (10)  | 0.0296 (8)   | 0.0001 (7)   | 0.0011 (6)   | -0.0110 (7)  |
| O2  | 0.0452 (10)  | 0.0285 (8)   | 0.0461 (10)  | -0.0152 (7)  | -0.0171 (8)  | -0.0027 (7)  |
| N1A | 0.0512 (13)  | 0.0364 (11)  | 0.0364 (11)  | -0.0169 (10) | -0.0067 (9)  | -0.0069 (9)  |
| C1A | 0.0273 (9)   | 0.0254 (9)   | 0.0207 (8)   | -0.0040 (7)  | -0.0021 (7)  | -0.0053 (7)  |
| C2A | 0.0274 (10)  | 0.0341 (11)  | 0.0219 (9)   | -0.0085 (8)  | 0.0002 (7)   | -0.0044 (8)  |
| C3A | 0.0293 (11)  | 0.0469 (14)  | 0.0254 (10)  | -0.0010 (10) | 0.0005 (8)   | -0.0044 (10) |
| C4A | 0.0578 (16)  | 0.0332 (12)  | 0.0253 (11)  | 0.0083 (11)  | 0.0003 (10)  | -0.0052 (10) |
| C5A | 0.078 (2)    | 0.0304 (12)  | 0.0326 (12)  | -0.0202 (13) | -0.0063 (12) | -0.0063 (10) |
| C1B | 0.0283 (10)  | 0.0251 (9)   | 0.0233 (9)   | -0.0062 (8)  | -0.0023 (7)  | -0.0033 (8)  |
| C2B | 0.0265 (10)  | 0.0337 (11)  | 0.0262 (10)  | -0.0013 (8)  | -0.0007 (7)  | -0.0070 (9)  |
| C3B | 0.0440 (14)  | 0.0417 (14)  | 0.0281 (11)  | 0.0002 (11)  | 0.0024 (9)   | -0.0092 (10) |
| C4B | 0.0481 (14)  | 0.0371 (12)  | 0.0248 (10)  | -0.0079 (11) | -0.0055 (9)  | -0.0051 (9)  |
| C5B | 0.0323 (11)  | 0.0383 (12)  | 0.0318 (11)  | -0.0101 (10) | -0.0096 (9)  | 0.0003 (10)  |
| C6B | 0.0255 (10)  | 0.0334 (11)  | 0.0271 (10)  | -0.0071 (8)  | -0.0004 (7)  | -0.0032 (9)  |

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

|              |             |              |             |
|--------------|-------------|--------------|-------------|
| Se—O1        | 1.6218 (16) | C5A—H5AA     | 0.9500      |
| Se—O2        | 1.6234 (16) | C1B—C2B      | 1.359 (3)   |
| Se—C1B       | 1.9240 (19) | C1B—C6B      | 1.381 (3)   |
| Se—C1A       | 1.929 (2)   | C2B—C3B      | 1.368 (3)   |
| N1A—C1A      | 1.354 (3)   | C2B—H2BA     | 0.9500      |
| N1A—C5A      | 1.367 (3)   | C3B—C4B      | 1.373 (3)   |
| C1A—C2A      | 1.378 (3)   | C3B—H3BA     | 0.9500      |
| C2A—C3A      | 1.388 (3)   | C4B—C5B      | 1.385 (3)   |
| C2A—H2AA     | 0.9500      | C4B—H4BA     | 0.9500      |
| C3A—C4A      | 1.383 (4)   | C5B—C6B      | 1.395 (3)   |
| C3A—H3AA     | 0.9500      | C5B—H5BA     | 0.9500      |
| C4A—C5A      | 1.383 (4)   | C6B—H6BA     | 0.9500      |
| C4A—H4AA     | 0.9500      |              |             |
| O1—Se—O2     | 117.59 (9)  | N1A—C5A—H5AA | 118.8       |
| O1—Se—C1B    | 106.80 (8)  | C4A—C5A—H5AA | 118.8       |
| O2—Se—C1B    | 109.14 (8)  | C2B—C1B—C6B  | 124.39 (19) |
| O1—Se—C1A    | 110.40 (9)  | C2B—C1B—Se   | 116.78 (16) |
| O2—Se—C1A    | 106.21 (9)  | C6B—C1B—Se   | 118.75 (15) |
| C1B—Se—C1A   | 106.17 (8)  | C1B—C2B—C3B  | 117.2 (2)   |
| C1A—N1A—C5A  | 115.3 (2)   | C1B—C2B—H2BA | 121.4       |
| N1A—C1A—C2A  | 126.1 (2)   | C3B—C2B—H2BA | 121.4       |
| N1A—C1A—Se   | 115.30 (16) | C2B—C3B—C4B  | 121.4 (2)   |
| C2A—C1A—Se   | 118.53 (16) | C2B—C3B—H3BA | 119.3       |
| C1A—C2A—C3A  | 116.9 (2)   | C4B—C3B—H3BA | 119.3       |
| C1A—C2A—H2AA | 121.6       | C3B—C4B—C5B  | 120.3 (2)   |
| C3A—C2A—H2AA | 121.6       | C3B—C4B—H4BA | 119.8       |

|                 |              |                 |              |
|-----------------|--------------|-----------------|--------------|
| C4A—C3A—C2A     | 119.4 (2)    | C5B—C4B—H4BA    | 119.8        |
| C4A—C3A—H3AA    | 120.3        | C4B—C5B—C6B     | 119.5 (2)    |
| C2A—C3A—H3AA    | 120.3        | C4B—C5B—H5BA    | 120.3        |
| C3A—C4A—C5A     | 119.8 (2)    | C6B—C5B—H5BA    | 120.3        |
| C3A—C4A—H4AA    | 120.1        | C1B—C6B—C5B     | 117.1 (2)    |
| C5A—C4A—H4AA    | 120.1        | C1B—C6B—H6BA    | 121.4        |
| N1A—C5A—C4A     | 122.5 (2)    | C5B—C6B—H6BA    | 121.4        |
|                 |              |                 |              |
| C5A—N1A—C1A—C2A | 1.2 (3)      | O1—Se—C1B—C2B   | 164.23 (17)  |
| C5A—N1A—C1A—Se  | 177.60 (16)  | O2—Se—C1B—C2B   | -67.68 (19)  |
| O1—Se—C1A—N1A   | -60.19 (17)  | C1A—Se—C1B—C2B  | 46.41 (19)   |
| O2—Se—C1A—N1A   | 171.30 (15)  | O1—Se—C1B—C6B   | -18.7 (2)    |
| C1B—Se—C1A—N1A  | 55.21 (17)   | O2—Se—C1B—C6B   | 109.38 (18)  |
| O1—Se—C1A—C2A   | 116.53 (16)  | C1A—Se—C1B—C6B  | -136.53 (18) |
| O2—Se—C1A—C2A   | -11.98 (18)  | C6B—C1B—C2B—C3B | -0.6 (4)     |
| C1B—Se—C1A—C2A  | -128.08 (16) | Se—C1B—C2B—C3B  | 176.31 (18)  |
| N1A—C1A—C2A—C3A | -0.5 (3)     | C1B—C2B—C3B—C4B | 0.3 (4)      |
| Se—C1A—C2A—C3A  | -176.84 (14) | C2B—C3B—C4B—C5B | -0.3 (4)     |
| C1A—C2A—C3A—C4A | -0.1 (3)     | C3B—C4B—C5B—C6B | 0.7 (4)      |
| C2A—C3A—C4A—C5A | 0.0 (3)      | C2B—C1B—C6B—C5B | 0.9 (3)      |
| C1A—N1A—C5A—C4A | -1.2 (3)     | Se—C1B—C6B—C5B  | -175.90 (16) |
| C3A—C4A—C5A—N1A | 0.7 (4)      | C4B—C5B—C6B—C1B | -0.9 (3)     |

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

| $D-H\cdots A$                       | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-------------------------------------|-------|-------------|-------------|---------------|
| C2A—H2AA $\cdots$ O1 <sup>i</sup>   | 0.95  | 2.50        | 3.331 (3)   | 146           |
| C4A—H4AA $\cdots$ O1 <sup>ii</sup>  | 0.95  | 2.53        | 3.341 (3)   | 143           |
| C5A—H5AA $\cdots$ O2 <sup>iii</sup> | 0.95  | 2.35        | 3.188 (3)   | 146           |

Symmetry codes: (i)  $x-1, y, z$ ; (ii)  $x-1, y+1, z$ ; (iii)  $x, y+1, z$ .