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4-Methyl-N-(4-nitrobenzylidene)-piperazin-1-amine

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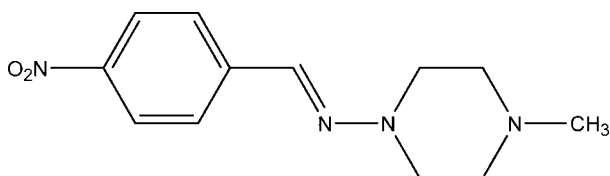
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 Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.041; wR factor = 0.119; data-to-parameter ratio = 14.8.

In the title compound, $\text{C}_{12}\text{H}_{16}\text{N}_4\text{O}_2$, the piperazine ring is in a slightly distorted chair conformation. In the molecule, the mean plane of the nitro group is twisted by $8.0(3)^\circ$ from that of the benzene ring. Also, the mean plane of the 2-nitrobenzyl ring is twisted slightly from that of the piperazine ring, with an $\text{N}-\text{N}=\text{C}-\text{C}$ torsion angle of $-176.24(11)^\circ$. In the crystal, pairs of weak $\text{C}-\text{H}\cdots\text{O}$ interactions link the molecules into dimers approximately along $[010]$.

Related literature

For the biological activity of Schiff base piperzine derivatives, see: Kharb *et al.* (2012); Savaliya *et al.* (2010); Xu *et al.* (2009); Zhou *et al.* (2011). For therapeutic areas related to piperazines as drug molecules, see: Bogatcheva *et al.* (2006); Brockunier *et al.* (2004); Cai *et al.* (2009); Choudhary *et al.* (2006); Upadhayaya *et al.* (2004). For a review of current pharmacological and toxicological information for piperazine derivatives, see: Elliott (2011). For the synthesis of related piperazine compounds and their medicinal and pharmaceutical activity, see: Capuano *et al.* (2002); Contreras *et al.* (2001). For related structures, see: Guo (2007); Ming-Lin *et al.* (2007); Xu *et al.* (2012); Zhou *et al.* (2011). For puckering parameters, see: Cremer & Pople (1975). For standard bond lengths, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{12}\text{H}_{16}\text{N}_4\text{O}_2$	$V = 2497.2(2) \text{ \AA}^3$
$M_r = 248.29$	$Z = 8$
Monoclinic, $C2/c$	Cu $K\alpha$ radiation
$a = 27.9353(14) \text{ \AA}$	$\mu = 0.77 \text{ mm}^{-1}$
$b = 5.9247(3) \text{ \AA}$	$T = 173 \text{ K}$
$c = 18.7763(7) \text{ \AA}$	$0.38 \times 0.32 \times 0.22 \text{ mm}$
$\beta = 126.527(3)^\circ$	

Data collection

Agilent Xcalibur (Eos, Gemini) diffractometer	7200 measured reflections
Absorption correction: multi-scan (<i>CrysAlis PRO</i> and <i>CrysAlis RED</i> ; Agilent, 2012)	2439 independent reflections
$T_{\min} = 0.868$, $T_{\max} = 1.000$	2022 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.031$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$	165 parameters
$wR(F^2) = 0.119$	H-atom parameters constrained
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$
2439 reflections	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C2}-\text{H2B}\cdots\text{O1}^i$	0.99	2.47	3.4052(19)	158

 Symmetry code: (i) $-x + \frac{3}{2}, -y + \frac{3}{2}, -z + 2$.

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis RED* (Agilent, 2012); program(s) used to solve structure: *SUPERFLIP* (Palatinus & Chapuis, 2007); program(s) used to refine structure: *SHELXL2012* (Sheldrick, 2008); molecular graphics: *OLEX2* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *OLEX2*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2568).

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supporting information

Acta Cryst. (2013). E69, o1687–o1688 [doi:10.1107/S1600536813028493]

4-Methyl-N-(4-nitrobenzylidene)piperazin-1-amine

Channappa N. Kavitha, Jerry P. Jasinski, Brian J. Anderson, H.S. Yathirajan and Manpreet Kaur

S1. Comment

Schiff base ligands derived from 1-amino-4-methylpiperazine have attracted interest due to diverse biological activities associated with the piperazine moiety. Schiff base piperazine derivatives have been designed to study their antimicrobial (Savaliya *et al.*, 2010; Kharb *et al.*, 2012) and antibacterial activity (Xu *et al.*, 2012). In addition, many drugs contain a piperazine ring as part of their molecular structure (Cai *et al.*, 2009). Piperazines are among the most important building blocks in today's drug discovery and are found in biologically active compounds across a number of different therapeutic areas (Brockunier *et al.*, 2004; Bogatcheva *et al.*, 2006) such as antifungal (Upadhayaya *et al.*, 2004), anti-bacterial, antimalarial activity and as in antipsychotic agents (Choudhary *et al.*, 2006). A review on the current pharmacological and toxicological information for piperazine derivatives has been recently presented (Elliott, 2011). The synthesis of related piperazine compounds and their medicinal and pharmaceutical activity have also been reported (Contreras *et al.*, 2001; Capuano *et al.*, 2002). The crystal structures of some related compounds, viz., 2-[(4-methylpiperazin-1-yl)iminomethyl]-phenol (Guo, 2007), 1,4-bis{3-[4-(dimethylamino)benzylideneamino] propyl}piperazine (Xu *et al.*, 2009), 2-methoxy-4-[(4-methylpiperazin-1-yl)-iminomethyl]phenol (Zhou *et al.*, 2011) and 2,4-dibromo-6-[(4-methylpiperazin-1-yl)iminomethyl]phenol (Ming-Lin *et al.*, 2007) have been reported. In view of the above importance of N-piperazinyl Schiff bases, the title compound, (I), C₁₂H₁₆N₄O₂ has been synthesized and the crystal structure is reported herein.

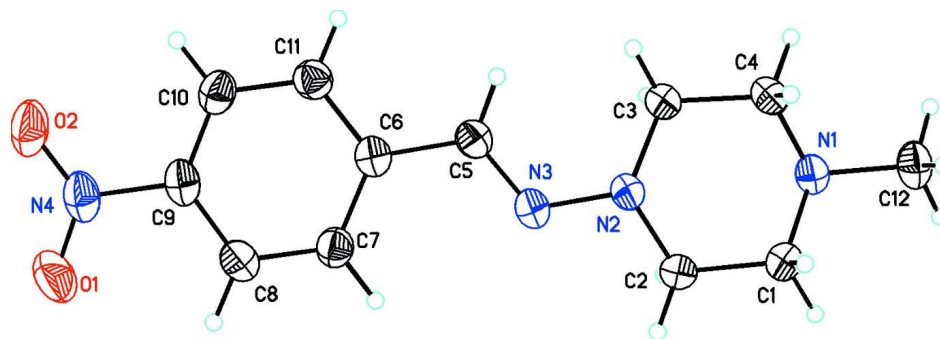
In the title compound, (I), the piperazine ring is in a slightly distorted chair conformation with puckering parameters Q , θ , and $\varphi = 0.5646\text{\AA}$, $170.8(5)^\circ$ and $187.961(8)^\circ$ (Cremer & Pople, 1975) (Fig. 1). In the molecule, the mean plane of the nitro group is twisted by $8.0(3)^\circ$ from that of the phenyl ring. Also, the mean plane of the 2-nitrobenzyl ring is twisted slightly from that of the piperazine ring with an N1/N2/C5/C6 torsion angle of $-176.24(11)^\circ$. Bond lengths are in normal ranges (Allen *et al.*, 1987). Weak C—H \cdots O intermolecular interactions are observed which lead to formation of dimers approximately along [010] and influence crystal packing (Fig. 2).

S2. Experimental

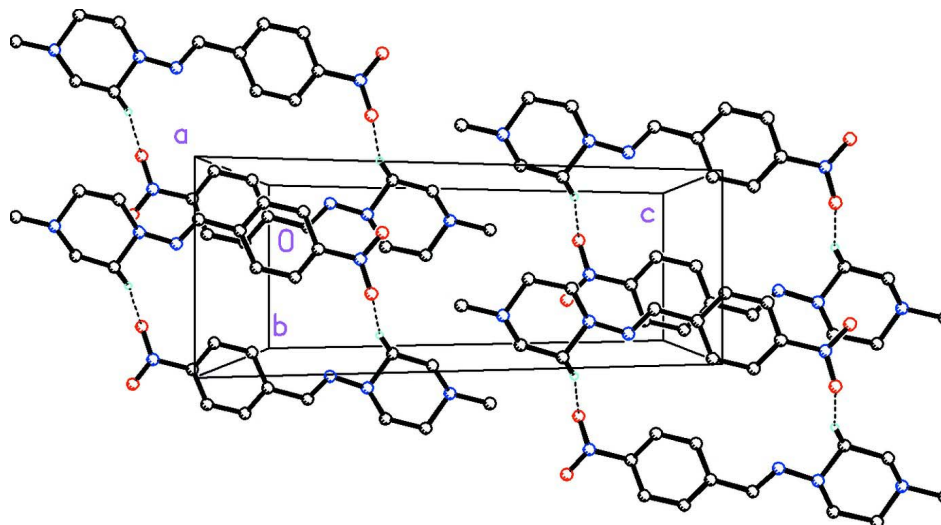
To a solution of o-nitrobenzaldehyde (0.75 g, 0.005 mol) in 10 ml of methanol, an equimolar amount of (1-amino-4-methyl)piperazine (0.57 g, 0.005 mol) is added dropwise with constant stirring. The mixture was refluxed for 8 hours to obtain an orange solution. The solution was evaporated to a small volume at room temperature and allowed to stand. Yellow crystals were formed in one day (m.p.: 358–360 K) and were used as such for x-ray diffraction studies.

S3. Refinement

All of the H atoms were placed in their calculated positions and then refined using the riding model with Atom—H lengths of 0.95 Å (CH), 0.99 Å (CH₂) or 0.98 Å (CH₃). Isotropic displacement parameters for these atoms were set to 1.2 (CH, CH₂) or 1.5 (CH₃) times U_{eq} of the parent atom. Idealised Me were refined as rotating groups.

**Figure 1**

ORTEP drawing of (I) ($C_{12}H_{16}N_4O_2$) showing the labeling scheme with 50% probability displacement ellipsoids.

**Figure 2**

Molecular packing for (I) viewed along the a axis. Dashed lines indicate weak C—H \cdots O intermolecular interactions linking the molecules into dimers along $[0\ 1\ 0]$. H atoms not involved in hydrogen bonding have been removed for clarity.

4-Methyl-*N*-(4-nitrobenzylidene)piperazin-1-amine

Crystal data

$C_{12}H_{16}N_4O_2$

$M_r = 248.29$

Monoclinic, $C2/c$

$a = 27.9353$ (14) Å

$b = 5.9247$ (3) Å

$c = 18.7763$ (7) Å

$\beta = 126.527$ (3)°

$V = 2497.2$ (2) Å³

$Z = 8$

$F(000) = 1056$

$D_x = 1.321$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 2934 reflections

$\theta = 3.2$ – 72.3 °

$\mu = 0.77$ mm⁻¹

$T = 173$ K

Irregular, yellow

$0.38 \times 0.32 \times 0.22$ mm

Data collection

Agilent Xcalibur (Eos, Gemini) diffractometer
 Radiation source: Enhance (Cu) X-ray Source
 Detector resolution: 16.0416 pixels mm⁻¹
 ω scans
 Absorption correction: multi-scan
 (*CrysAlis PRO* and *CrysAlis RED*; Agilent, 2012)
 $T_{\min} = 0.868$, $T_{\max} = 1.000$

7200 measured reflections
 2439 independent reflections
 2022 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.031$
 $\theta_{\max} = 72.3^\circ$, $\theta_{\min} = 3.9^\circ$
 $h = -34 \rightarrow 32$
 $k = -7 \rightarrow 7$
 $l = -15 \rightarrow 22$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.119$
 $S = 1.02$
 2439 reflections
 165 parameters
 0 restraints
 Primary atom site location: structure-invariant direct methods

Hydrogen site location: inferred from neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0636P)^2 + 0.9105P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.22 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL2012* (Sheldrick, 2008), $F_c^* = kFc[1 + 0.001x \text{Fc}^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
 Extinction coefficient: 0.00056 (10)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.93562 (6)	1.1504 (2)	1.26970 (8)	0.0636 (4)
O2	0.96924 (5)	0.8095 (2)	1.29508 (7)	0.0509 (3)
N1	0.60437 (5)	0.7685 (2)	0.55865 (7)	0.0301 (3)
N2	0.66959 (5)	0.80238 (19)	0.74619 (7)	0.0282 (3)
N3	0.71366 (5)	0.8675 (2)	0.83118 (7)	0.0290 (3)
N4	0.93437 (5)	0.9567 (3)	1.24576 (8)	0.0403 (3)
C1	0.58784 (6)	0.9349 (2)	0.59774 (9)	0.0328 (3)
H1A	0.5570	0.8709	0.6023	0.039*
H1B	0.5710	1.0702	0.5594	0.039*
C2	0.64208 (6)	1.0007 (2)	0.68894 (9)	0.0329 (3)
H2A	0.6714	1.0761	0.6836	0.039*
H2B	0.6302	1.1090	0.7160	0.039*
C3	0.68086 (6)	0.6183 (2)	0.70644 (9)	0.0311 (3)
H3A	0.6928	0.4816	0.7438	0.037*
H3B	0.7140	0.6601	0.7037	0.037*
C4	0.62582 (6)	0.5677 (2)	0.61410 (9)	0.0317 (3)
H4A	0.6352	0.4494	0.5868	0.038*
H4B	0.5941	0.5090	0.6177	0.038*

C5	0.75644 (6)	0.7311 (2)	0.88472 (8)	0.0290 (3)
H5	0.7586	0.5880	0.8640	0.035*
C6	0.80153 (6)	0.7948 (2)	0.97695 (9)	0.0282 (3)
C7	0.80100 (6)	1.0039 (2)	1.01127 (9)	0.0317 (3)
H7	0.7706	1.1101	0.9737	0.038*
C8	0.84416 (6)	1.0572 (3)	1.09893 (9)	0.0336 (3)
H8	0.8438	1.1991	1.1221	0.040*
C9	0.88821 (6)	0.8997 (3)	1.15268 (9)	0.0326 (3)
C10	0.89029 (6)	0.6922 (3)	1.12129 (9)	0.0344 (3)
H10	0.9208	0.5867	1.1594	0.041*
C11	0.84702 (6)	0.6411 (2)	1.03298 (9)	0.0330 (3)
H11	0.8482	0.4998	1.0101	0.040*
C12	0.55353 (7)	0.7136 (3)	0.46826 (9)	0.0387 (4)
H12A	0.5211	0.6537	0.4689	0.058*
H12B	0.5654	0.6002	0.4436	0.058*
H12C	0.5400	0.8502	0.4317	0.058*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0527 (8)	0.0698 (9)	0.0412 (7)	0.0008 (6)	0.0132 (6)	-0.0240 (6)
O2	0.0326 (6)	0.0792 (9)	0.0295 (6)	0.0096 (6)	0.0123 (5)	0.0014 (6)
N1	0.0278 (6)	0.0347 (6)	0.0256 (6)	-0.0026 (5)	0.0146 (5)	-0.0031 (5)
N2	0.0279 (6)	0.0289 (6)	0.0245 (6)	-0.0001 (4)	0.0138 (5)	-0.0035 (4)
N3	0.0278 (6)	0.0326 (6)	0.0257 (6)	-0.0038 (5)	0.0154 (5)	-0.0042 (4)
N4	0.0280 (6)	0.0624 (9)	0.0287 (6)	-0.0011 (6)	0.0159 (6)	-0.0066 (6)
C1	0.0290 (7)	0.0327 (7)	0.0301 (7)	0.0031 (5)	0.0140 (6)	-0.0005 (5)
C2	0.0335 (7)	0.0274 (7)	0.0310 (7)	0.0027 (5)	0.0156 (6)	-0.0022 (6)
C3	0.0312 (7)	0.0286 (7)	0.0284 (7)	0.0021 (5)	0.0150 (6)	-0.0031 (5)
C4	0.0334 (7)	0.0291 (7)	0.0302 (7)	-0.0029 (5)	0.0177 (6)	-0.0061 (5)
C5	0.0291 (7)	0.0304 (7)	0.0290 (7)	-0.0023 (5)	0.0181 (6)	-0.0026 (5)
C6	0.0271 (7)	0.0331 (7)	0.0278 (7)	-0.0036 (5)	0.0183 (6)	-0.0010 (5)
C7	0.0291 (7)	0.0354 (7)	0.0289 (7)	0.0009 (6)	0.0163 (6)	-0.0002 (6)
C8	0.0325 (7)	0.0365 (8)	0.0330 (7)	-0.0035 (6)	0.0202 (6)	-0.0065 (6)
C9	0.0261 (7)	0.0467 (8)	0.0264 (7)	-0.0050 (6)	0.0165 (6)	-0.0041 (6)
C10	0.0282 (7)	0.0437 (8)	0.0295 (7)	0.0037 (6)	0.0161 (6)	0.0037 (6)
C11	0.0325 (7)	0.0341 (7)	0.0334 (7)	0.0004 (6)	0.0201 (6)	-0.0018 (6)
C12	0.0333 (8)	0.0498 (9)	0.0265 (7)	-0.0039 (6)	0.0142 (6)	-0.0052 (6)

Geometric parameters (Å, °)

O1—N4	1.2253 (19)	C4—H4A	0.9900
O2—N4	1.2219 (18)	C4—H4B	0.9900
N1—C1	1.4586 (17)	C5—H5	0.9500
N1—C4	1.4546 (18)	C5—C6	1.4598 (19)
N1—C12	1.4608 (17)	C6—C7	1.401 (2)
N2—N3	1.3682 (15)	C6—C11	1.400 (2)
N2—C2	1.4639 (17)	C7—H7	0.9500

N2—C3	1.4580 (16)	C7—C8	1.379 (2)
N3—C5	1.2889 (18)	C8—H8	0.9500
N4—C9	1.4657 (18)	C8—C9	1.388 (2)
C1—H1A	0.9900	C9—C10	1.379 (2)
C1—H1B	0.9900	C10—H10	0.9500
C1—C2	1.5137 (19)	C10—C11	1.384 (2)
C2—H2A	0.9900	C11—H11	0.9500
C2—H2B	0.9900	C12—H12A	0.9800
C3—H3A	0.9900	C12—H12B	0.9800
C3—H3B	0.9900	C12—H12C	0.9800
C3—C4	1.5122 (18)		
C1—N1—C12	110.78 (11)	C3—C4—H4A	109.4
C4—N1—C1	108.16 (10)	C3—C4—H4B	109.4
C4—N1—C12	110.55 (11)	H4A—C4—H4B	108.0
N3—N2—C2	110.17 (10)	N3—C5—H5	119.7
N3—N2—C3	119.30 (10)	N3—C5—C6	120.54 (13)
C3—N2—C2	113.80 (10)	C6—C5—H5	119.7
C5—N3—N2	120.39 (12)	C7—C6—C5	122.44 (13)
O1—N4—C9	117.78 (14)	C11—C6—C5	118.67 (13)
O2—N4—O1	123.67 (13)	C11—C6—C7	118.89 (13)
O2—N4—C9	118.55 (14)	C6—C7—H7	119.6
N1—C1—H1A	109.7	C8—C7—C6	120.72 (14)
N1—C1—H1B	109.7	C8—C7—H7	119.6
N1—C1—C2	109.86 (11)	C7—C8—H8	120.6
H1A—C1—H1B	108.2	C7—C8—C9	118.76 (13)
C2—C1—H1A	109.7	C9—C8—H8	120.6
C2—C1—H1B	109.7	C8—C9—N4	118.83 (13)
N2—C2—C1	111.02 (11)	C10—C9—N4	118.94 (13)
N2—C2—H2A	109.4	C10—C9—C8	122.22 (13)
N2—C2—H2B	109.4	C9—C10—H10	120.7
C1—C2—H2A	109.4	C9—C10—C11	118.56 (13)
C1—C2—H2B	109.4	C11—C10—H10	120.7
H2A—C2—H2B	108.0	C6—C11—H11	119.6
N2—C3—H3A	109.5	C10—C11—C6	120.84 (13)
N2—C3—H3B	109.5	C10—C11—H11	119.6
N2—C3—C4	110.69 (11)	N1—C12—H12A	109.5
H3A—C3—H3B	108.1	N1—C12—H12B	109.5
C4—C3—H3A	109.5	N1—C12—H12C	109.5
C4—C3—H3B	109.5	H12A—C12—H12B	109.5
N1—C4—C3	111.29 (11)	H12A—C12—H12C	109.5
N1—C4—H4A	109.4	H12B—C12—H12C	109.5
N1—C4—H4B	109.4		
O1—N4—C9—C8	-7.5 (2)	C3—N2—N3—C5	-21.56 (18)
O1—N4—C9—C10	171.92 (14)	C3—N2—C2—C1	50.72 (15)
O2—N4—C9—C8	172.30 (13)	C4—N1—C1—C2	62.44 (14)
O2—N4—C9—C10	-8.2 (2)	C5—C6—C7—C8	179.56 (12)

N1—C1—C2—N2	-56.86 (15)	C5—C6—C11—C10	-179.02 (12)
N2—N3—C5—C6	-176.24 (11)	C6—C7—C8—C9	0.0 (2)
N2—C3—C4—N1	55.30 (15)	C7—C6—C11—C10	1.1 (2)
N3—N2—C2—C1	-172.24 (11)	C7—C8—C9—N4	179.59 (12)
N3—N2—C3—C4	177.74 (11)	C7—C8—C9—C10	0.1 (2)
N3—C5—C6—C7	-0.5 (2)	C8—C9—C10—C11	0.4 (2)
N3—C5—C6—C11	179.62 (12)	C9—C10—C11—C6	-1.0 (2)
N4—C9—C10—C11	-179.07 (12)	C11—C6—C7—C8	-0.6 (2)
C1—N1—C4—C3	-62.15 (14)	C12—N1—C1—C2	-176.25 (11)
C2—N2—N3—C5	-155.91 (12)	C12—N1—C4—C3	176.40 (11)
C2—N2—C3—C4	-49.44 (15)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C2—H2B \cdots O1 ⁱ	0.99	2.47	3.4052 (19)	158

Symmetry code: (i) $-x+3/2, -y+5/2, -z+2$.