metal-organic compounds

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Rubidium pentaaqua(L-serine)cobalt(II) hexahydrogenhexamolybdocobaltate(III) L-serine monosolvate decahydrate

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Received 4 September 2013; accepted 15 October 2013

Key indicators: single-crystal X-ray study; T = 183 K; mean σ (C–C) = 0.008 Å; Hatom completeness 29%; R factor = 0.036; wR factor = 0.099; data-to-parameter ratio = 17.9.

The Co^{2+} ion in the title compound, $\text{Rb}[\text{Co}(\text{C}_3\text{H}_7\text{NO}_3)-(\text{H}_2\text{O})_5][\text{H}_6\text{Co}\text{Mo}_6\text{O}_{24}]\cdot\text{C}_3\text{H}_7\text{NO}_3\cdot10\text{H}_2\text{O}$, is coordinated by five water molecules and one *O*-monodentate L-serine ligand in a slightly distorted octahedral geometry. The Rb^+ ion is irregularly coordinated by nine O atoms. In the crystal, the $[\text{H}_6\text{Co}^{III}\text{Mo}_6\text{O}_{24}]^{3-}$ polyanions are stacked along the *b*-axis direction, mediated by bridging Rb-O bonds. $\text{N}-\text{H}\cdots$ O and $\text{O}-\text{H}\cdots$ O hydrogen bonds are observed involving the L-serine molecules.

Related literature

For background to polyoxidometallates (POMs), see: Hasenknopf *et al.* (2008); Du *et al.* (2013); Fang *et al.* (2005); Kortz *et al.* (2002); Sadakane *et al.* (2001); Tan *et al.* (2007); Inoue & Yamase (1995). For C–O bond lengths in carboxylates, see: Lide (2007). For bond-valence sums, see: Brown (1980). For protonation of POMs, see: Perloff (1970); Honda *et al.* (2007); Yang *et al.* (2013). For chiral POMs constructed from an Anderson-type POM as a building block, see: An *et al.* (2008).



Experimental

Crystal data Rb[Co(C₃H₇NO₃)(H₂O)₅]-[H₆CoMo₆O₂₄]·C₃H₇NO₃·10H₂O

 $M_r = 1649.42$ Orthorhombic, $P2_12_12_1$ a = 10.8411 (5) Å b = 11.5923 (4) Å c = 34.8078 (12) Å $V = 4374.4 (3) \text{ Å}^{3}$ Z = 4

Data collection

Rigaku R-AXIS RAPID diffractometer Absorption correction: numerical (NUMABS; Rigaku, 1999) $T_{\rm min} = 0.123, T_{\rm max} = 0.500$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$	
$wR(F^2) = 0.099$	
S = 1.04	
10010 reflections	
559 parameters	
H-atom parameters constrained	

 Table 1

 Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O27−H4···O41	0.84	2.24	2.859 (6)	131
O27−H4···N1	0.84	2.62	2.940 (6)	104
$O30-H11\cdots O42^{i}$	0.84	2.24	2.928 (6)	139
N1-H5···O2	0.91	2.11	2.944 (6)	151
$N1 - H6 \cdots O39^{i}$	0.91	2.14	3.041 (6)	170
$N1 - H7 \cdots O17^{ii}$	0.91	2.09	2.905 (6)	149
N2-H13···O43	0.91	2.11	3.013 (7)	170
$N2-H12\cdots O8^{iii}$	0.91	2.31	2.912 (6)	123
$N2-H14\cdots O13^{iv}$	0.91	1.93	2.827 (6)	167
Symmetry codes:	(i) $r + 1$	1 - 7 (ii) $-r + 2 = n - 1$	$-\pi \perp 1$ (iii)

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{1}{2}, -z$; (ii) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$; (iii) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$; (iv) x, y - 1, z.

Data collection: *RAPID-AUTO* (Rigaku/MSC, 2002); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2010); software used to prepare material for publication: *CrystalStructure*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB7134).

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Mo $K\alpha$ radiation

 $0.41 \times 0.36 \times 0.19 \text{ mm}$

70065 measured reflections

10010 independent reflections

9728 reflections with $F^2 > 2\sigma(F^2)$

Absolute structure: Flack (1983),

Absolute structure parameter:

 $\mu = 3.63 \text{ mm}^{-1}$

T = 183 K

 $R_{\rm int}=0.062$

 $\Delta \rho_{\text{max}} = 3.61 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\text{min}} = -1.10 \text{ e } \text{\AA}^{-3}$

4426 Friedel pairs

0.025 (7)

Perloff, A. (1970). Inorg. Chem. 9, 2228-2239.

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supporting information

Acta Cryst. (2013). E69, m612-m613 [doi:10.1107/S1600536813028304]

Rubidium pentaaqua(L-serine)cobalt(II) hexahydrogenhexamolybdocobaltate(III) L-serine monosolvate decahydrate

Jun Iijima, Haruo Naruke and Hiroshi Takiyama

S1. Comment

Continuous interests in chiral polyoxometalates (POMs) are endowed owing to their prominent molecular applications such as nonlinear optics, magneto optical effect, and circular polarized luminescence (Hasenknopf et al., 2008; Du et al., 2013). Two synthetic procedures for chiral POMs have been delivered to date. One is the way based on the connection of chiral organic ligands and achiral inorganic POMs possible to propagate the local chirality to POM framework, which have been extensively studied by the groups of Hill, Kortz, Pope, Wang, and Yamase for a long time (Fang et al., 2005; Kortz et al., 2002; Sadakane et al., 2001; Tan et al., 2007, Inoue & Yamase, 1995). Although Anderson type POM is one of the typical achiral structures along with Keggin and Wells-Dawson types, there are little chiral POMs constructed from Anderson type POM as building block (An et al., 2008). Herein, the structural characterization of the chiral crystal containing Anderson type $[H_6CoMo_6O_{24}]^3$ as building block and amino acid L-Serine (L-Ser) as chiral organic ligand is reported. The crystallization of enantiomeric unit (Figure 1) consisting of the CoMo6 polyanion, $[Co(H_2O)_5(L-Ser)]^{2+}$ (Co(II) complex), Rb^+ , one water molecule, and free L-Ser, and the other water molecules with noncentrosymmetric space group $P2_12_12_1$ provided the title compound, of which the absolute configuration was supported by Flack parameter 0.025 (7). The p K_a (2.2 and 9.2) of L-Ser and the acidic condition (pH 3.2) in the synthetic procedure suggest that all L-Ser molecules are zwitterionic. In fact, the carboxylate group in L-Ser coordinating to Co^{2+} and that in free L-Ser have similar C—O distances (1.250 (7), 1.264 (7) Å and 1.247 (7), 1.257 (7) Å) due to their resonance state (Lide, 2007). The BVS values calculated from the observed bond lengths are 2.0 and 3.2 for Co in Co(II) complex and the CoMo6 polyanion, 5.9–6.0 for Mo, indicating that the original valences of Co^{2+} , Co^{3+} , and Mo^{6+} are retained in the title compound. Additionally, the BVS values for six O atoms forming CoO_6 octahedra in the CoMo6 polyanion are 1.1–1.2, suggesting that they are protonated (Brown, 1980). Similar protonation in Anderson type POMs have been frequently observed for [H₆CrMo₆O₂₄]³⁻, [H₂IMo₆O₂₄]³⁻, and [H₆NiMo₆O₂₄]⁴⁻ (Perloff, 1970; Honda *et al.*, 2007; Yang *et al.*, 2013). Figure 2 shows the crystal structure viewed along the *a* axis. The two kinds of polyanion (A-type and B-type) with different orientation in the *ac* plane are alternately connected along the crystallographic b axis by Rb⁺-bridging of terminal O atoms in the CoMo6 polyanion, forming the b-axially stacked polyanion unit. The polyanion units are aligned along the c axis with a help of hydrogen bonding ($O \cdot O < 3.3$ Å) relating with some waters of crystallization and aqua ligands in Co(II) complex, of which the unit distance is the half of c axis value. The both distance of A-type...A-type and B-type \cdot B-type are 11.6 Å; (Co···Co distance), which corresponds to the b axis value. Figure 3 is the representation of the b-axially polyanion stacking mentioned above along the c axis. It is clearly demonstrated that A-type polyanion and B-type polyanion are altenately stacked with a cross angle of 46.1°, which is calculated from the dihedral angle of the least square plane defined by Co, Mo1, Mo2, Mo3, Mo4, Mo5, and Mo6 in the CoMo6 polyanion.

S2. Experimental

An aqueous solution containing the Na salt of CoMo6, *L*-Ser, and CoCl₂·6H₂O with a molar ratio of 1:4:2 was acidified to pH = 3.2 by conc HCl and then boiled for 1 h. After cooling to room temperature, 5 eq. of RbClO₄ to CoMo6 was added into the solution. Viridian thin platelets of the title compound were obtained at 4 °C being stood for several days.

S3. Refinement

Fragile thin platelet crystals of the title compound gave a medium absorption corrections (transmission factor ranging from 12.3 to 50.0%), resulting in a somewhat high residual electron densities around Mo1, Mo2, Mo4, and Mo6. Generally, a complete convergence of structure refinement with residual density less than 1 e Å⁻³ in difference Fourier synthesis indicates the success of structural analysis. However, for the compounds containing many heavy atoms, since the anisotropy of electron cloud around the heavy atoms makes difficult to conduct the ellipse approximation, the relative large residual densities cannot help being remaining. In the structural analysis of polyoxometalate consisting of a variety of heavy atoms such as Mo, V, and W, the relatively high residual densities have been frequently observed for the crystals with poor absorption corrections and crystallinity. In fact, after some elements were allocated to the pointed out densities, the refinements resulted in the divergence. Therefore, the pointed out high residual densities for Mo1, Mo2, Mo4, and Mo6 are not responsible for the incompleteness of structural analysis but the intrinsic crystal qualities such as morphology and crystallinity. All the atoms except for H atoms were refined anisotropically, and the H atoms were isotropically. The H atoms in L-Ser molecules were located in the calculated position, and the remaining H atoms were not included in the refinements. The residual density around O31 may be H atom. Due to the undesirable absorption correction, it is conceivable that high residual density corresponding to H atom is remained. In the polyoxometalate bearing many heavy atoms, the electron density for H atom is hard to be found. In order to escape from the divergence and lowering the accuracy of refinement caused by intentional assignment of H atom to remarkable low electron density, for polyoxometalate compounds, the H atoms in water crystallization are not included in refinements. Therefore, the water of crystallization O atoms (O37 to O45) are refined without H atoms.



Figure 1

The asymmetric unit of the title compound with displacement ellipsoids drawn at the 50% probability level. White, red, blue, green, moss green and yellowish brown ellipsoids denote Mo, O, Co, Rb, C, and N atoms, respectively. The H atoms protonated to the CoMo6 polyanion are omitted for clarity.



Figure 2

Packing diagram viewed along the *a* axis. The rectangle traced with standard line indicates the unit cell. Color codes; white octahedron: MoO_6 , blue octahedron CoO_6 , and Co(II) complex, green sphere: Rb, red sphere O. The hydrogen bondings are denoted by sky blue dashed lines. The *L*-Ser molecule and waters of crystallization O atoms unrelated to hydrogen bonding are omitted for clarity.



Figure 3

The alternate polyanion stacking of A-type and B-type viewed along the c axis. The square shows the unit cell. The MoO₆ octahedron is denoted in open faced manner, and the O atoms in the CoMo6 polyanion are denoted by red spheres. The definition of the least square plane consisting of Co, Mo1, Mo2, Mo3, Mo4, Mo5, and Mo6 in the CoMo6 polyanion is also represented.

Rubidium pentaaqua(*L*-serine)cobalt(II) hexahydrogenhexamolybdocobaltate(III) *L*-serine monosolvate decahydrate

Crystal data	
$Rb[Co(C_3H_7NO_3)(H_2O)_5]$	V = 4374.4 (3) Å ³
$[H_6CoMo_6O_{24}] \cdot C_3H_7NO_3 \cdot 10H_2O$	Z = 4
$M_r = 1649.42$	F(000) = 3212.00
Orthorhombic, $P2_12_12_1$	$D_{\rm x} = 2.504 {\rm ~Mg} {\rm ~m}^{-3}$
Hall symbol: P 2ac 2ab	Mo <i>K</i> α radiation, $\lambda = 0.71075$ Å
a = 10.8411 (5) Å	Cell parameters from 59700 reflections
b = 11.5923 (4) Å	$\theta = 3.0-27.5^{\circ}$
c = 34.8078 (12) Å	$\mu = 3.63 \text{ mm}^{-1}$

T = 183 K Platelet, green	$0.41 \times 0.36 \times 0.19 \text{ mm}$
Data collection Rigaku R-AXIS RAPID diffractometer Detector resolution: 10.000 pixels mm ⁻¹ ω scans Absorption correction: numerical (<i>NUMABS</i> ; Rigaku, 1999) $T_{min} = 0.123, T_{max} = 0.500$ 70065 measured reflections	10010 independent reflections 9728 reflections with $F^2 > 2\sigma(F^2)$ $R_{int} = 0.062$ $\theta_{max} = 27.5^{\circ}$ $h = -14 \rightarrow 14$ $k = -15 \rightarrow 14$ $l = -45 \rightarrow 45$
RefinementRefinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.036$ $wR(F^2) = 0.099$ $S = 1.04$ 10010 reflections559 parameters0 restraintsPrimary atom site location: structure-invariant direct methodsSecondary atom site location: difference Fourier map	Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0641P)^2 + 9.0772P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 3.61$ e Å ⁻³ $\Delta\rho_{min} = -1.10$ e Å ⁻³ Absolute structure: Flack (1983), 4426 Friedel pairs Absolute structure parameter: 0.025 (7)

Special details

Geometry. ENTER SPECIAL DETAILS OF THE MOLECULAR GEOMETRY

Refinement. Refinement was performed using all reflections. The weighted *R*-factor (*wR*) and goodness of fit (*S*) are based on F^2 . *R*-factor (gt) are based on *F*. The threshold expression of $F^2 > 2.0 \sigma(F^2)$ is used only for calculating *R*-factor (gt).

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

	x	у	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
Mo1	0.98964 (4)	0.94684 (3)	0.207793 (11)	0.01204 (9)	
Mo2	0.74633 (3)	0.84989 (3)	0.159456 (12)	0.01213 (9)	
Mo3	0.49911 (3)	0.75710(3)	0.205914 (11)	0.01245 (9)	
Mo4	0.50288 (4)	0.74903 (3)	0.301413 (11)	0.01297 (9)	
Mo5	0.74159 (3)	0.84414 (3)	0.349594 (12)	0.01364 (9)	
Mo6	0.98220 (4)	0.94607 (3)	0.303320 (11)	0.01208 (9)	
Rb1	0.73248 (5)	0.52247 (4)	0.145507 (15)	0.02399 (11)	
Col	0.74451 (5)	0.84817 (4)	0.254347 (19)	0.00969 (12)	
Co2	0.67086 (6)	0.46834 (6)	0.043914 (18)	0.01672 (14)	
01	1.0518 (4)	1.0822 (4)	0.20918 (11)	0.0217 (8)	
O2	1.0846 (4)	0.8663 (4)	0.17903 (10)	0.0212 (8)	
03	0.8426 (4)	0.7691 (3)	0.13123 (10)	0.0196 (7)	
O4	0.6534 (4)	0.9266 (4)	0.12866 (10)	0.0203 (8)	
05	0.4363 (4)	0.6206 (3)	0.20593 (11)	0.0205 (8)	
06	0.4087 (4)	0.8391 (4)	0.17621 (10)	0.0205 (8)	
O7	0.4401 (4)	0.6137 (3)	0.29822 (11)	0.0211 (8)	
08	0.4089 (4)	0.8259 (3)	0.33182 (10)	0.0181 (8)	

O9	0.6422 (4)	0.9134 (4)	0.37969 (11)	0.0233 (8)
O10	0.8398 (4)	0.7659 (4)	0.37805 (10)	0.0229 (8)
011	1.0744 (3)	0.8673 (3)	0.33354 (10)	0.0174 (7)
O12	1.0443 (4)	1.0825 (3)	0.30257 (11)	0.0200 (8)
013	0.8528 (3)	0.9780 (3)	0.17385 (9)	0.0140 (7)
O14	0.6360 (3)	0.7246 (3)	0.17290 (10)	0.0146 (7)
015	0.4347 (3)	0.8152 (3)	0.25403 (10)	0.0156 (7)
O16	0.6406 (4)	0.7154 (3)	0.33395 (10)	0.0172 (7)
017	0.8416 (3)	0.9759 (3)	0.33605 (10)	0.0166 (7)
O18	1.0544 (3)	0.8863 (3)	0.25617 (10)	0.0143 (7)
019	0.8487 (3)	0.7967 (3)	0.29521 (9)	0.0122 (7)
020	0.8484(3)	0.9818 (3)	0.25489(9)	0.0120 (6)
021	0.8514(3)	0.7968(3)	0.21394(9)	0.0120(0)
022	0.6311(3) 0.6412(3)	0.9013(3)	0.21391(9) 0.21348(9)	0.0105(7)
023	0.6387(3)	0.7160(3)	0.25320(10)	0.0130(6)
024	0.6384(3)	0.8976 (3)	0.29320(10) 0.29495(9)	0.0124(7)
025	1.0441(4)	0.5975(3)	0.29495(9) 0.04817(12)	0.0124(1)
025	1.0441(4) 1 2444(4)	0.5955(4)	0.04817(12) 0.06457(11)	0.0314(10)
020	1.2444(4) 1 1876(4)	0.3605(4)	0.06784(11)	0.0250(8)
027	1.1070(4)	0.3081(4) 0.3482(3)	0.00784(11) 0.08204(10)	0.0234(3)
020	0.7330(3) 0.9125(4)	0.3482(3) 0.2885(4)	0.08204(10) 0.04575(11)	0.0134(7) 0.0243(8)
029	0.9125(4) 0.7725(4)	0.2005(4)	0.04373(11) 0.05742(11)	0.0243(0)
031	0.7723(4) 0.7102(3)	0.0400(4) 0.5053(3)	0.03742(11) 0.23535(11)	0.0334(10)
031	0.7102(3) 0.5472(4)	0.5055(5)	0.23333(11) 0.08612(11)	0.0208(8)
032	0.5473(4) 0.5412(4)	0.3390(4) 0.3362(4)	0.08012(11) 0.03548(11)	0.0229(8) 0.0247(8)
033	0.3412(4) 0.7017(4)	0.3302(4)	-0.00155(10)	0.0247(8)
034	0.7917(4) 0.7058(4)	0.4190(4)	-0.00133(10)	0.0210(8)
035	0.7938(4)	0.3901(4)	0.00412(11)	0.0243(8)
030	0.3870(4)	0.3030(4)	0.00108(11)	0.0310(10)
037	0.0383(4)	0.2721(4)	-0.04401(11)	0.0273(9)
038	0.2030(4)	0.0949(4)	0.2307(2)	0.0303(18)
039	0.9018(4)	-0.2655(4)	-0.09497(11)	0.0265(9)
040	0.8044 (4)	0.0929 (4)	-0.01818(11)	0.0230(9)
041	1.5218(5)	1.0091(4)	0.12009(12)	0.0361(10)
042	0.2127(4)	0.0982 (4)	-0.04824(14)	0.0353 (10)
043	0.5376(5)	0.1397 (4)	0.08536(14)	0.0382 (11)
044	0.3835(4)	0.0982 (4)	0.01085(12)	0.0326 (10)
045	0.4649 (5)	0.9210 (4)	0.02937 (13)	0.0389 (11)
NI	1.2151 (5)	0.6988 (4)	0.13015 (12)	0.0201 (9)
N2	0.7690 (4)	0.1/19(4)	0.13211(14)	0.0199 (9)
	1.1344 (5)	0.6185 (5)	0.068/5(15)	0.0201 (10)
C2	1.1089 (5)	0.7005 (5)	0.10294 (14)	0.01/8 (10)
C3	1.0849 (6)	0.8225 (5)	0.08854 (16)	0.0240 (11)
C4	0.8428 (5)	0.2819 (5)	0.07428 (13)	0.0170 (10)
CS	0.8689 (5)	0.1813 (5)	0.10152 (15)	0.0193 (10)
C6	0.8809 (6)	0.0683 (5)	0.07916 (16)	0.0253 (12)
H1	1.0337	0.6727	0.1168	0.0214*
H2	1.0115	0.8221	0.0716	0.0288*
H13	0.6941	0.1642	0.1206	0.0238*

H12	0.7693	0.2367	0.1469	0.0238*
H14	0.7839	0.1092	0.1472	0.0238*
H3	1.0668	0.8732	0.1107	0.0288*
H4	1.2497	0.8700	0.0823	0.0305*
H5	1.1981	0.7449	0.1506	0.0241*
H6	1.2839	0.7251	0.1180	0.0241*
H7	1.2282	0.6254	0.1384	0.0241*
H8	0.9492	0.1966	0.1147	0.0231*
H9	0.9528	0.0728	0.0617	0.0304*
H10	0.8950	0.0039	0.0973	0.0304*
H11	0.7805	-0.0157	0.0453	0.0401*

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Mo1	0.00991 (17)	0.01256 (18)	0.01365 (17)	-0.00216 (14)	0.00079 (14)	0.00013 (14)
Mo2	0.01080 (18)	0.01307 (19)	0.01252 (18)	-0.00066 (15)	-0.00029 (13)	0.00023 (13)
Mo3	0.00977 (18)	0.01192 (18)	0.01566 (18)	-0.00179 (15)	-0.00060 (14)	-0.00069 (14)
Mo4	0.01106 (18)	0.01205 (18)	0.01579 (18)	-0.00148 (15)	0.00241 (15)	0.00050 (14)
Mo5	0.01245 (19)	0.01610 (19)	0.01237 (18)	0.00063 (16)	0.00120 (13)	-0.00002 (14)
Mo6	0.00996 (17)	0.01245 (18)	0.01384 (18)	-0.00125 (14)	-0.00103 (14)	-0.00076 (14)
Rb1	0.0239 (3)	0.0221 (3)	0.0259 (3)	0.0034 (2)	-0.00233 (19)	-0.0013 (2)
Co1	0.0077 (3)	0.0092 (3)	0.0121 (3)	0.0002 (2)	0.0003 (2)	-0.0001 (2)
Co2	0.0147 (3)	0.0177 (3)	0.0177 (3)	0.0013 (3)	0.0006 (3)	0.0017 (3)
O1	0.0194 (18)	0.0199 (18)	0.0257 (19)	-0.0014 (15)	0.0031 (15)	0.0068 (16)
O2	0.0160 (17)	0.029 (2)	0.0191 (18)	0.0003 (15)	0.0028 (14)	-0.0047 (16)
O3	0.0191 (18)	0.0188 (17)	0.0210 (17)	0.0008 (15)	-0.0011 (14)	-0.0049 (15)
O4	0.0196 (18)	0.0209 (17)	0.0204 (17)	-0.0006 (15)	-0.0066 (14)	0.0056 (15)
O5	0.0197 (18)	0.0162 (16)	0.0257 (19)	-0.0058 (14)	0.0034 (15)	-0.0026 (15)
O6	0.0161 (17)	0.027 (2)	0.0188 (17)	0.0060 (16)	-0.0053 (14)	0.0008 (16)
O7	0.0223 (19)	0.0164 (17)	0.0246 (19)	-0.0067 (15)	-0.0016 (16)	0.0025 (15)
08	0.0188 (18)	0.0186 (18)	0.0170 (17)	0.0014 (14)	0.0040 (14)	0.0021 (15)
09	0.0195 (18)	0.030 (2)	0.0208 (18)	-0.0012 (15)	0.0054 (15)	-0.0027 (16)
O10	0.0213 (18)	0.030 (2)	0.0174 (18)	0.0017 (16)	-0.0018 (15)	0.0075 (16)
O11	0.0159 (16)	0.0191 (18)	0.0174 (17)	0.0009 (14)	-0.0011 (13)	0.0013 (15)
O12	0.0177 (18)	0.0194 (17)	0.0228 (18)	0.0024 (14)	0.0017 (15)	0.0010 (15)
O13	0.0140 (15)	0.0110 (15)	0.0169 (16)	-0.0037 (13)	-0.0031 (13)	0.0025 (13)
O14	0.0139 (16)	0.0130 (15)	0.0168 (16)	-0.0049 (13)	0.0029 (13)	-0.0019 (14)
O15	0.0102 (14)	0.0157 (15)	0.0209 (17)	0.0005 (11)	0.0018 (14)	0.0014 (14)
O16	0.0189 (18)	0.0165 (17)	0.0164 (17)	0.0017 (14)	0.0010 (13)	0.0024 (14)
O17	0.0148 (16)	0.0171 (17)	0.0179 (16)	0.0012 (14)	0.0000 (13)	-0.0055 (14)
O18	0.0099 (14)	0.0153 (15)	0.0178 (16)	0.0010 (11)	-0.0008 (13)	-0.0011 (14)
O19	0.0120 (15)	0.0107 (15)	0.0139 (16)	0.0004 (12)	0.0012 (12)	0.0006 (13)
O20	0.0105 (14)	0.0095 (13)	0.0161 (15)	0.0002 (11)	0.0004 (12)	-0.0011 (13)
O21	0.0072 (14)	0.0104 (14)	0.0152 (16)	-0.0005 (11)	0.0016 (12)	-0.0014 (13)
O22	0.0095 (15)	0.0098 (15)	0.0159 (16)	-0.0007 (12)	-0.0020 (12)	0.0007 (13)
O23	0.0108 (14)	0.0098 (13)	0.0184 (16)	-0.0018 (11)	0.0014 (13)	-0.0002 (14)
O24	0.0108 (15)	0.0109 (15)	0.0154 (16)	-0.0013 (12)	0.0018 (13)	0.0016 (13)

O25	0.022 (2)	0.041 (3)	0.031 (3)	-0.0027 (17)	-0.0023 (17)	-0.0115 (19)
O26	0.0184 (18)	0.0302 (19)	0.028 (2)	0.0011 (16)	0.0004 (16)	-0.0045 (17)
O27	0.0224 (19)	0.029 (2)	0.0243 (19)	-0.0013 (16)	-0.0044 (16)	0.0040 (17)
O28	0.0178 (17)	0.0171 (17)	0.0203 (17)	0.0046 (15)	-0.0001 (13)	-0.0011 (13)
O29	0.0192 (18)	0.030 (2)	0.0237 (19)	0.0063 (15)	0.0047 (15)	0.0074 (17)
O30	0.044 (3)	0.028 (2)	0.028 (2)	-0.014 (2)	0.0039 (18)	-0.0053 (18)
O31	0.0145 (16)	0.0127 (16)	0.035 (2)	0.0019 (13)	-0.0039 (14)	-0.0001 (15)
O32	0.0158 (16)	0.0253 (19)	0.0276 (19)	0.0024 (15)	0.0028 (14)	-0.0021 (16)
O33	0.0200 (18)	0.0232 (18)	0.031 (2)	-0.0027 (15)	-0.0017 (15)	0.0007 (17)
O34	0.0203 (18)	0.0229 (17)	0.0217 (18)	0.0000 (15)	0.0020 (14)	-0.0005 (15)
O35	0.0225 (19)	0.0197 (18)	0.031 (2)	-0.0033 (14)	0.0032 (16)	-0.0035 (16)
O36	0.026 (2)	0.042 (3)	0.028 (2)	0.0136 (19)	0.0042 (16)	0.0127 (19)
O37	0.0190 (18)	0.040 (3)	0.0231 (19)	-0.0027 (17)	-0.0008 (15)	-0.0048 (18)
O38	0.0155 (19)	0.0092 (17)	0.145 (6)	0.0025 (14)	0.003 (3)	-0.008 (3)
O39	0.0219 (19)	0.029 (2)	0.0283 (19)	0.0050 (17)	0.0029 (15)	-0.0050 (17)
O40	0.023 (2)	0.027 (2)	0.025 (2)	-0.0047 (15)	-0.0025 (15)	-0.0065 (16)
O41	0.050 (3)	0.027 (2)	0.031 (3)	0.010 (2)	-0.007 (2)	-0.0032 (19)
O42	0.034 (3)	0.027 (2)	0.045 (3)	-0.0032 (18)	0.005 (2)	-0.0022 (19)
O43	0.034 (3)	0.033 (3)	0.047 (3)	0.0042 (19)	0.000 (2)	0.010 (2)
O44	0.028 (3)	0.037 (3)	0.032 (3)	0.0086 (18)	0.0041 (18)	0.0043 (19)
O45	0.034 (3)	0.044 (3)	0.039 (3)	0.015 (2)	0.003 (2)	0.006 (2)
N1	0.028 (3)	0.017 (2)	0.016 (2)	0.0029 (17)	-0.0006 (17)	0.0017 (17)
N2	0.0127 (19)	0.0080 (16)	0.039 (3)	0.0054 (15)	0.0093 (18)	0.0129 (18)
C1	0.019 (3)	0.021 (3)	0.020 (3)	0.003 (2)	0.0010 (19)	-0.003 (2)
C2	0.016 (3)	0.019 (3)	0.018 (3)	0.0013 (18)	0.0034 (19)	0.001 (2)
C3	0.026 (3)	0.022 (3)	0.024 (3)	0.010 (2)	0.003 (2)	0.007 (3)
C4	0.017 (3)	0.019 (3)	0.015 (2)	-0.0000 (19)	-0.0012 (18)	-0.0021 (19)
C5	0.015 (3)	0.020 (3)	0.023 (3)	0.0019 (19)	0.0035 (19)	0.006 (2)
C6	0.028 (3)	0.020 (3)	0.028 (3)	0.009 (3)	0.008 (3)	0.001 (3)

Geometric parameters (Å, °)

Mo1-01	1.709 (4)	Rb1—O28	3.002 (4)
Mo1	1.713 (4)	Rb1—O31	3.143 (4)
Mo1-013	1.931 (4)	Rb1—O32	2.889 (4)
Mo1-018	1.955 (4)	Rb1—O35	3.018 (4)
Mo1-O20	2.280 (3)	Co1—O19	1.912 (4)
Mo1-021	2.306 (3)	Co1—O20	1.915 (3)
Mo2—O3	1.712 (4)	Co1—O21	1.917 (4)
Mo2—O4	1.719 (4)	Co1—O22	1.913 (4)
Mo2—O13	1.946 (4)	Co1—O23	1.915 (3)
Mo2—O14	1.939 (4)	Co1—O24	1.910 (4)
Mo2-O21	2.296 (4)	Co2—O28	2.123 (4)
Mo2—O22	2.278 (4)	Co2—O32	2.154 (4)
Mo3—O5	1.722 (4)	Co2—O33	2.099 (4)
Mo3—O6	1.713 (4)	Co2—O34	2.132 (4)
Mo3—O14	1.914 (4)	Co2—O35	2.078 (4)
Mo3—O15	1.936 (4)	Co2—O36	2.060 (5)

Mo3—O22	2.288 (3)	O25—C1	1.248 (7)
Mo3—O23	2.286 (4)	O26—C1	1.257 (7)
Mo4—Mo5	3.2749 (6)	O27—C3	1.428 (7)
Mo4—O7	1.713 (4)	O28—C4	1.264 (6)
Mo4—O8	1.718 (4)	O29—C4	1.250 (6)
Mo4—O15	1.964 (4)	O30—C6	1.420 (7)
Mo4—O16	1.914 (4)	N1—C2	1.491 (7)
Mo4—O23	2.265 (4)	N2—C5	1.523 (7)
Mo4	2.275 (4)	C1—C2	1.548 (8)
Mo5	1.704 (4)	C2-C3	1.523 (8)
Mo5-010	1 713 (4)	C4-C5	1 529 (7)
Mo5-016	1 929 (4)	C5-C6	1.529(7) 1 530(8)
Mo5-017	1.929(1) 1.931(4)	027—H4	0.840
M05-019	2288(4)	O30—H11	0.840
Mo5024	2.200(4)	N1H5	0.040
Mo5_024	2.292(4)	N1 H6	0.910
Mo6_012	1.714(4)	N1 H7	0.910
Mo6_017	1.719(4) 1.024(4)	N1	0.910
Mo6_018	1.934 (4)	N2 H12	0.910
M00-018	1.940(4)	N2—H12	0.910
M00-019	2.274(4)	N2—H14	0.910
M00-020	2.202(3)	C2—H1	1.000
RDI	5.158 (4) 2.858 (4)	C3—H2	0.990
Rb1	2.858 (4)	C3—H3	0.990
	2.855 (4)	C5—H8	1.000
Rb1—012 ⁿ	3.099 (4)	С6—Н9	0.990
Rb1—014	2.737 (4)	С6—Н10	0.990
Co1013	3.390 (4)	O7…H12 ^v	3.2919
Co1O14	3.387 (4)	O7…H14 ^v	3.0848
Co1O15	3.380 (3)	O8…H13 ^v	2.7395
Co1O16	3.364 (4)	08…H12 ^v	2.3132
Co1O17	3.374 (4)	08…H14 ^v	3.3485
Co1…O18	3.389 (3)	O10…H8 ⁱⁱⁱ	2.4384
Co2…O29	3.349 (4)	O11H12 ⁱⁱⁱ	2.3729
01012	3.252 (6)	011···H14 ⁱⁱⁱ	3.4299
$01\cdots019^{iii}$	2.714 (5)	011····H8 ⁱⁱⁱ	2.6886
01031	3.344 (6)	013…H13 ^{vi}	3.3249
0203	3 305 (5)	013···H12 ^{vi}	3 2702
02^{-00}	3 528 (5)	$013 \cdots H14^{vi}$	1 9329
0406	3.287 (5)	O13···H8 ^{vi}	3.4279
$04\cdots07^{v}$	3 494 (6)	$013 \cdots H10^{vi}$	2 7195
$04\cdots 030^{vi}$	3 122 (6)	025···H1	2.5619
$04 \cdots N2^{v_i}$	3 110 (6)	025····H2	2.7950
$04\cdots C6^{v_i}$	3 427 (7)	026···H1	3 0852
0507	3 214 (6)	026···H4	3 3433
$05 \cdots 024^{i}$	2 710 (5)	O26H5	3 5492
05031	3414(5)	026 HG	2 4944
07022 ⁱ	2 647 (5)	O26…H7	2.6156

O7…N2 ^v	3.387 (6)	O27…H1	3.2887
08…09	3.194 (6)	O27…H5	3.2180
08…N2 ^v	2.912 (6)	O27…H6	2.6237
O9…O32 ^v	2.789 (6)	O28…H13	2.6015
O10····O11	3.202 (5)	O28…H12	2.6064
O10…C5 ⁱⁱⁱ	3.383 (7)	O28…H14	3.5942
O11····N2 ⁱⁱⁱ	3.073 (6)	O28…H8	2.9795
O11····C4 ⁱⁱⁱ	3.476 (6)	O29…H8	2.6561
011…C5 ⁱⁱⁱ	3.184 (7)	O29…H9	2.5982
O12…O21 ⁱⁱⁱ	2.790 (5)	O30…H13	2.7231
O13····N2 ^{vi}	2.827 (6)	O30…H14	3.2090
013…C5 ^{vi}	3.453 (6)	O30…H8	3.2665
O13…C6 ^{vi}	3.472 (7)	O36…H9 ^{ix}	3.0878
O15…O31 ^v	2.731 (5)	N1…H2	3.3264
O18…O31 ⁱⁱⁱ	2.915 (5)	N1…H3	2.6702
023…031	2.637 (5)	N1…H4	2.6186
O25…C3	3.036 (7)	N2…H9	3.3613
026…027	3.323 (6)	N2…H10	2.6694
026…N1	2.647 (6)	C1H2	2.7114
026···C3	3.342(7)	C1…H3	3.3743
027…N1	2.940 (6)	C1···H4	3.2060
027···C1	2.950 (7)	C1H5	3.2783
028N2	2.691 (6)	C1H6	2.6629
028····C6	3.528(7)	C1H7	2.6307
029030	3.215 (6)	C2…H4	2.5901
029033 ^{vii}	3.469 (6)	C3···H5	2.6429
029034	2.592 (6)	С3…Н6	2.6418
029····C6	2.826(7)	C3···H7	3.2631
030N2	2.978 (6)	C4···H13	2.6571
O30····C4	2.893(7)	C4…H12	2 7012
033····C4	3 593 (6)	C4…H14	3 2938
034····C4	3 130 (6)	C4…H9	2,7369
01···038 ^v	3 283 (6)	C4…H10	3 3691
$02\cdots 038^{iv}$	3.205(0) 3.594(7)	C5…H11	3 1 5 5 8
02N1	2 944 (6)	C6H13	2 7233
$02 \cdots C2$	3 283 (6)	C6···H12	3 2912
02C3	3.190 (7)	C6H14	2 6329
02 °C3 03…042 ^{viii}	3 236 (6)	H1H2	2.052)
03	3.152 (7)	H1H3	2.5511
03	3.132(7) 3.081(7)	H1H4	3 4866
03 °C3	3,030 (6)	H1H5	2 2942
$04 \cdot 03^{\text{viii}}$	3 216 (6)	H1H6	2.2942
$04 \cdots 042$ $04 \cdots 043^{vi}$	3.210(0)	H1H7	2.7790
05038	3.134(0) 3.108(7)	нт н <i>і</i>	2.3043
05 050 06030 ^{ix}	3,170 (7)	H2H5	2.0075
06041×	2.009 (0) 2.016 (6)	112 113 ЦЭЦА	3.3292 2.5401
06N1x	2.910(0)	112 П0 Ц12ЦQ	2.2401 2.7002
00111	3.101(0)	П15 Пð Ц12Ц10	2.7903
07030	3.091(/)	n13n10	2.9/31

O9…O26 ⁱⁱⁱ	3.050 (6)	H13…H11	3.4765
O10…O37 ^{xi}	2.721 (6)	H12…H8	2.2966
O10…O41 ⁱⁱ	3.455 (6)	H12…H10	3.4808
O13…O38 ^v	3.541 (7)	H14…H8	2.3477
O14…O39 ^{ix}	3.339 (6)	H14…H9	3.5184
015038	2.869 (5)	H14…H10	2.4393
O16…O41 ⁱⁱ	2.894 (6)	H3…H4	2.2173
Q16Q43 ^v	3.520 (6)	H3…H5	2.4834
017…N1 ⁱⁱⁱ	2.905 (6)	Н3…Н6	2.9242
017···C2 ⁱⁱⁱ	3.403 (6)	H3…H7	3.4992
018038 ^{iv}	2.746 (5)	H4…H5	2.8423
020.038^{v}	2.566 (5)	H4…H6	2.1224
025035	2,749 (6)	H4···H7	3 4521
$025 \cdot 035$	2,837 (7)	H8···H9	2 3377
$\Omega_{26}^{26} \cdots \Omega_{9}^{16}$	3 050 (6)	H8···H10	2.3876
$\Omega_{26}^{\circ} \Omega_{32}^{\circ}$	3 411 (6)	H9H11	2.2061
$\Omega_{26}^{-0.022}$	3 313 (6)	H10····H11	2.2001
$\Omega_{26}^{-0.00} \Omega_{26}^{-0.00}$	2 713 (6)	Mo1····H3	3 5841
$\Omega_{26}^{-10} \Omega_{4iv}^{-10}$	2 729 (6)	Mo5…H7 ⁱⁱⁱ	3 3029
$027 \cdots 034^{\text{viii}}$	3 562 (6)	O2…H1	3 1673
$027 \cdots 036^{\text{viii}}$	2,766 (6)	02····H3	2 3869
027039^{vii}	3 337 (6)	02	2.1144
027041	2.859 (6)	O2…H6	3.4440
027044^{iv}	3 511 (6)	O2…H7	3 4959
$027 \cdots 045^{iv}$	3.347 (6)	O3…H1	2.4073
028043	3.367 (6)	O3…H2	2.8346
$029 \cdots 037^{\text{vii}}$	2.758 (6)	O3…H3	2.8059
029040	3,386 (6)	05H5 ^x	3.5281
030040	2.708 (6)	O5…H7 ^x	3.2581
030042 ^{vii}	2.928 (6)	O6…H5 [×]	2.6821
030043	2,931 (7)	O6···H6 ^x	2,7716
032026 ^x	3.411 (6)	O6…H7 ^x	3.4202
032039 ^{ix}	2.793 (6)	09H7 ⁱⁱⁱ	2.9001
033037	3,159 (6)	O12…H1 ⁱⁱⁱ	3.1126
033…040 ^{ix}	2.762 (6)	O13···H3	3.4190
033043	2.864 (6)	017…H1 ⁱⁱⁱ	3.1189
034…027 ^{xii}	3.562 (6)	017···H5 ⁱⁱⁱ	3.1826
034037	2.693 (6)	017···H6 ⁱⁱⁱ	3.5719
034…045 ^{viii}	2.812 (6)	017···H7 ⁱⁱⁱ	2.0891
035025	2.749 (6)	O27····H9 ^{vi}	3.4869
035…042 ^{viii}	2.672 (6)	O35…H1	3.3071
O36…O27 ^{xii}	2.766 (6)	O35…H2	3.5739
036…042 ^{viii}	3.467 (6)	O36···H2 ^{xii}	2.9821
O36…O44	2.711 (6)	O36…H4 ^{xii}	3.4952
O36…C3 ^{xii}	3.401 (7)	O37…H8 ^{ix}	3.3473
O37···O10 ^{xiii}	2.721 (6)	O37····H9 ^{ix}	2.9248
O37…O29 ^{ix}	2.758 (6)	O38…H12 ^v	3.4037
O37…O33	3.159 (6)	O38…H14 ^v	3.4937

O37…O34	2.693 (6)	O39····H4 ^{ix}	2.6354
O37…O40	2.770 (6)	O39…H5 ^{ix}	3.4609
O38…O1 ⁱ	3.283 (6)	O39…H6 ^{ix}	2.1396
O38…O2 ^x	3.594 (7)	O39…H7 ^{ix}	3.3673
038…05	3.198 (7)	O40…H9	3.2207
O38…O7	3.091 (7)	O40…H11	2.5578
O38…O13 ⁱ	3.541 (7)	O41…H3	3.2008
038…015	2.869 (5)	O41…H4	2.2367
038…018 ^x	2.746 (5)	041	3.5017
O38…O20 ⁱ	2.566 (5)	O41…H6	3.3191
039…04 ^{vii}	3.030 (6)	O42···H2 ^{xii}	3.3487
03906 ^{vii}	3 009 (6)	$042\cdots H10^{ix}$	3 5090
039…014 ^{vii}	3 339 (6)	042···H11 ^{ix}	2 2422
039026 ^{ix}	3 313 (6)	043	2.1120
$039\cdots027^{ix}$	3 337 (6)	043H12	3 4863
039····032 ^{vii}	2 793 (6)	043	3 4463
039041^{ix}	3 328 (6)	$043 \cdots H11$	3 4813
039042^{xiv}	3.197 (6)	$044 \cdots H2^{xii}$	3 1971
039044 ^{vii}	3.197 (6)	$044 \cdots H4^{x}$	3 4995
039····045 ^{vii}	2 909 (6)	$044\cdots$ H11 ^{ix}	3 0896
039N1 ^{ix}	3.041 (6)	$O45 \cdots H4^{x}$	3 0294
$040\cdots026^{ix}$	2 713 (6)	$O45 \cdots H11^{vi}$	3 5431
040.020	3 386 (6)	C3···H9 ^{vi}	3 3681
$040\cdots030$	2 708 (6)	$C3\cdots H10^{vi}$	2 9592
040···033 ^{vii}	2.768 (6)	C6···H2 ^{xvii}	3 1970
040.033	2.702 (0)	C6···H3 ^{xvii}	3 2224
$040\cdots044^{vii}$	3 491 (6)	$H1\cdots O2$	3 1673
$O40\cdots C1^{ix}$	3,536 (7)	H1····O3	2 4073
$O40 \cdots C6$	3,500 (7)	H1···O12 ⁱⁱ	3 1126
$041\cdots06^{iv}$	2 916 (6)	H1017 ⁱⁱ	3 1189
041010^{iii}	2.910 (0) 3 455 (6)	H1···O35	3 3071
041…016 ⁱⁱⁱ	2 894 (6)	H2···O3	2.8346
041027	2.859 (6)	H2···O35	3 5739
041039 ^{vii}	3328(6)	$H2\cdots O36^{viii}$	2 9821
$041\cdots043^{xv}$	3.026(0) 3.046(7)	$H2\cdots O42^{viii}$	3 3487
041	3,539 (8)	$H2\cdots O44^{viii}$	3 1971
$042\cdots03^{xii}$	3 236 (6)	$H2\cdots C6^{vi}$	3 1970
$042 \cdots 04^{xii}$	3,216 (6)	$H2\cdots H9^{vi}$	2 9953
$042 \cdots 030^{ix}$	2 928 (6)	$H2\cdots H10^{vi}$	2.6152
042···035 ^{xii}	2.528 (6)	$H2\cdots H11^{vi}$	3 2630
042036 ^{xii}	3.467 (6)	H13····O43	2.1120
042···039 ^{xvi}	3.197 (6)	H12O38 ⁱ	3 4037
042044	2 768 (7)	H12····043	3 4863
042045 ^{xii}	3.092 (7)	H14O38 ⁱ	3 4937
$043\cdots04^{xvii}$	3 154 (6)	H14…O43	3 4463
043016^{i}	3 520 (6)	H3…Mo1	3 5841
043028	3,367 (6)	H3…O2	2 3869
043030	2.931 (7)	H3···O3	2.8059
	=		

O43…O33	2.864 (6)	H3…O13	3.4190
O43····O41 ^{xviii}	3.046 (7)	H3…O41	3.2008
O43…O45 ^{xvii}	3.294 (7)	H3····C6 ^{vi}	3.2224
O43…N2	3.013 (7)	H3····H9 ^{vi}	3.1293
O44…O26 ^x	2.729 (6)	H3…H10 ^{vi}	2.4462
O44…O27 ^x	3.511 (6)	H4…O36 ^{viiii}	3.4952
O44…O36	2.711 (6)	H4…O39 ^{vii}	2.6354
O44…O39 ^{ix}	3.147 (6)	H4…O41	2.2367
O44…O40 ^{ix}	3.491 (6)	H4…O44 ^{iv}	3.4995
O44…O42	2.768 (7)	H4…O45 ^{iv}	3.0294
044…045	2.805 (7)	H5…O2	2.1144
044…C1 ^x	3.494 (7)	H5…O5 ^{iv}	3.5281
045…025 ^{xii}	2.837 (7)	H5…O6 ^{iv}	2.6821
045…027 ^x	3.347 (6)	H5…O17 ⁱⁱ	3.1826
Q45…Q34 ^{xii}	2.812 (6)	H5…O39 ^{vii}	3.4609
045039 ^{ix}	2.909 (6)	H5…O41	3.5017
045…042 ^{viii}	3.092 (7)	H6…O2	3.4440
045043 ^{vi}	3.294 (7)	H6···O6 ^{iv}	2.7716
045044	2.805 (7)	H6…O17 ⁱⁱ	3.5719
N1…O2	2.944 (6)	H6···O39 ^{vii}	2.1396
N1…O6 ^{iv}	3.101 (6)	H6…O41	3.3191
N1017 ⁱⁱ	2,905 (6)	H7···Mo5 ⁱⁱ	3 3029
N1…O39 ^{vii}	3 041 (6)	Η7…Ω2	3 4959
N2…O43	3,013(7)	H7···O5 ^{iv}	3 2581
$C1\cdots O40^{vii}$	3.536(7)	H7···O6 ^{iv}	3.4202
$C1 \cdots O44^{iv}$	3,494(7)	H7O9 ⁱⁱ	2,9001
C2O2	3 283 (6)	H7…017 ⁱⁱ	2.0891
C2O3	3152(7)	H7···O39 ^{vii}	3 3673
C2O17 ⁱⁱ	3.403 (6)	H8···O37 ^{vii}	3.3473
C3····O2	3 190 (7)	H9····O27 ^{xvii}	3 4869
C3···O3	3.081(7)	H9037 ^{vii}	2.9248
$C3\cdots O36^{viii}$	3.401(7)	H9···O40	3.2207
C3…O41	3 539 (8)	H9····C3 ^{xvii}	3 3681
C6…O40	3 500 (7)	H9····H2 ^{xvii}	2,9953
$M_0 2 \cdots H_1 4^{v_i}$	3.0638	H9H3 ^{xvii}	3 1293
Mo2…H10 ^{vi}	3 2343	$H10\cdots O42^{vii}$	3 5090
Mo4···H13 ^v	3 5913	H10····C3 ^{xvii}	2 9592
Mo4…H12 ^v	3 4589	H10···H2 ^{xvii}	2.6152
Rb1H12	3 3372	H10···H3 ^{xvii}	2.0152
O3…H10 ^{vi}	3.0200	H11O40	2.5578
$04 \cdots H13^{vi}$	2 8030	H11····O42 ^{vii}	2.2270
$04 \cdots H14^{vi}$	2.6050	H11O43	3 4813
O4…H10 ^{vi}	2.0200	H11····O44 ^{vii}	3 0896
04···H11 ^{vi}	3 2797	$H11\cdots O45^{xvii}$	3 5431
07···H13 ^v	3 2320	H11···H2 ^{xvii}	3 2630
0, III <i>3</i>	5.2320	1111 112	5.2050
$01 - M_01 - 02$	106 26 (19)	$O12^{ii}$	121 37 (10)
01 - Mo1 - 013	98 55 (16)	$O12^{ii}$ Rb1 $O20$	59 59 (10)
	20.22 (10)	012 101 001	27.27 (10)

O1—Mo1—O18	99.41 (16)	O12 ⁱⁱ —Rb1—O32	160.80 (11)
O1—Mo1—O20	94.66 (15)	O12 ⁱⁱ —Rb1—O35	108.16 (10)
O1—Mo1—O21	161.17 (15)	O14—Rb1—O28	149.48 (10)
O2—Mo1—O13	101.91 (16)	O14—Rb1—O31	71.23 (10)
O2—Mo1—O18	95.28 (16)	O14—Rb1—O32	85.65 (11)
O2—Mo1—O20	157.08 (15)	O14—Rb1—O35	101.05 (11)
O2—Mo1—O21	91.95 (15)	O28—Rb1—O31	134.06 (9)
013—Mo1—018	150.50 (14)	Q28—Rb1—Q32	64.74 (10)
013 - Mo1 - 020	83 71 (13)	0.28 Rb1 0.35	57 77 (10)
013 - Mo1 - 021	72,44 (13)	0.31 Rb1 0.32	13153(10)
018 - Mo1 - 020	71 68 (12)	031 Rb1 032	165.42(10)
018 - Mo1 - 021	83 28 (12)	032 _Rb1_035	57 89 (11)
010 - M01 - 021	68.32(11)	0.00000000000000000000000000000000000	<i>S1</i> . <i>39</i> (11) <i>S1</i> .3 <i>(</i> 1 <i>4</i>)
020 - 1001 - 021	106.32(11)	019 - 010 - 020	04.13(14)
03 - M02 - 04	100.40(17) 101.74(10)	019 - 01 - 021	93.27(14)
03 - M02 - 013	101.74(10)	019 - 001 - 022	1/9.3/(14)
03—Mo2—014	96.02 (16)	019 - 001 - 023	96.87 (14)
03—Mo2—021	91.44 (14)	019—Co1—024	84.19 (14)
O3—Mo2—O22	157.18 (15)	O20—Co1—O21	84.43 (13)
O4—Mo2—O13	96.50 (16)	O20—Co1—O22	95.24 (13)
O4—Mo2—O14	100.17 (16)	O20—Co1—O23	178.98 (14)
O4—Mo2—O21	160.79 (15)	O20—Co1—O24	95.97 (14)
O4—Mo2—O22	94.94 (15)	O21—Co1—O22	84.74 (14)
O13—Mo2—O14	151.07 (14)	O21—Co1—O23	95.64 (14)
O13—Mo2—O21	72.40 (12)	O21—Co1—O24	179.29 (14)
O13—Mo2—O22	83.38 (13)	O22—Co1—O23	83.75 (14)
O14—Mo2—O21	84.61 (13)	O22—Co1—O24	95.80 (14)
O14—Mo2—O22	71.82 (13)	O23—Co1—O24	83.96 (14)
O21—Mo2—O22	68.69 (11)	O28—Co2—O32	95.09 (14)
O5—Mo3—O6	106.47 (18)	O28—Co2—O33	83.78 (14)
O5—Mo3—O14	97.22 (16)	O28—Co2—O34	91.63 (14)
05—Mo3—015	100.19 (16)	$0.28 - C_0 2 - 0.35$	87.62 (14)
05—Mo3—022	159.57 (15)	$028 - C_0 2 - 036$	171.73 (16)
$05 - M_0 3 - 023$	94.01 (15)	032 - 002 - 033	87 67 (15)
$06 - M_0 3 - 014$	100.99 (16)	032 - 002 - 033	172.49(15)
$06 M_{03} 015$	97.04 (16)	032 - 002 - 034	85.04 (15)
$06 M_{0}^{2} 0^{2}$	97.04(10) 92.83(15)	032 - 002 - 035	90.17(16)
$06 M_{02} 022$	32.03(13)	032 - 022 - 030	90.17 (10)
00 - M03 - 023	150.51(15)	033 - 02 - 034	90.41(13)
014 M03 013	150.19 (14)	033-02-035	108.19(10)
014—Mo3—022	/2.02 (13)	033 - 02 - 036	90.08 (16)
014—Mo3—023	82.99 (13)	034-02-035	91.89 (15)
015—Mo3—022	83.62 (12)	034—Co2—O36	83.53 (15)
O15—Mo3—O23	71.85 (13)	O35—Co2—O36	99.20 (17)
O22—Mo3—O23	67.91 (11)	Mo2—O3—Rb1	100.10 (15)
Mo5—Mo4—O7	130.93 (13)	Mo4—O8—Rb1 ^v	154.38 (19)
Mo5—Mo4—O8	88.76 (12)	Mo6—O11—Rb1 ⁱⁱⁱ	104.41 (16)
Mo5—Mo4—O15	126.59 (10)	Mo6—O12—Rb1 ⁱⁱⁱ	95.20 (15)
Mo5—Mo4—O16	31.70 (11)	Mo1-013-Mo2	118.07 (17)
Mo5—Mo4—O23	85.56 (8)	Mo2-014-Mo3	118.38 (18)

Mo5-Mo4-O24	44.39 (8)	Mo2-014-Rb1	108.75 (14)
O7—Mo4—O8	106.22 (18)	Mo3—O14—Rb1	132.38 (16)
O7—Mo4—O15	98.85 (16)	Mo3—O15—Mo4	117.05 (16)
O7—Mo4—O16	99.30 (17)	Mo4	116.88 (18)
O7—Mo4—O23	93.16 (15)	Mo5—O17—Mo6	116.42 (18)
O7—Mo4—O24	160.57 (16)	Mo1—O18—Mo6	117.03 (16)
O8—Mo4—O15	95.26 (16)	Mo5—O19—Mo6	92.14 (12)
O8—Mo4—O16	101.74 (16)	Mo5—O19—Co1	103.93 (15)
O8—Mo4—O23	158.36 (15)	Mo6—O19—Co1	103.34 (14)
O8—Mo4—O24	92.94 (15)	Mo1—O20—Mo6	94.15 (11)
O15—Mo4—O16	150.49 (14)	Mo1—O20—Co1	104.13 (14)
Q15—Mo4—Q23	71.86 (13)	Mo6—O20—Co1	103.67 (14)
Q15—Mo4—Q24	82.20 (12)	Mo1-021-Mo2	92.51 (12)
$016 - M_0 4 - 023$	84.08 (14)	Mol-O21-Col	103.11 (14)
Q16—Mo4—Q24	73.07 (14)	Mo2-O21-Co1	102.89 (14)
0^{23} -Mo4-024	68 58 (11)	$M_0^2 = 0^{22} = M_0^3$	92.91 (12)
Mo4—Mo5—O9	88 49 (13)	$M_0^2 = 0^{22} = C_0^1$	103 68 (14)
Mo4 - Mo5 - O10	127 50 (14)	Mo2 = 022 = Co1 Mo3 = 022 = Co1	103.00(11) 104.17(14)
Mo4 Mo5 O16	31.42(11)	Mo3-022 CO1	93 89 (11)
Mo4 - Mo5 - 017	125 79 (11)	Mo3-023-Col	104 17 (14)
Mo4 - Mo5 - 019	84 06 (9)	Mo4 - 023 - Co1	101.17(11) 103.81(15)
Mo4—Mo5—024	43.98 (8)	Mo4 - O24 - Mo5	91 63 (12)
$09 - M_0 5 - 010$	106 67 (18)	Mo4 - O24 - Co1	103 59 (14)
$09 - M_0 = 016$	100.31(17)	$M_{05} = 0.024 = C_{01}$	103.89(14) 103.84(15)
$09 - M_0 5 - 017$	97 62 (17)	Rb1 - 0.28 - 0.27	89 21 (11)
$09 - M_0 = 019$	160 55 (16)	Rb1 = 0.28 = 0.02	128 6 (3)
$09 - M_0 5 - 0.24$	94 26 (16)	$C_{0}^{2} = 0.028 = 0.4$	126.0(3)
$010 - M_05 - 016$	96 11 (17)	$Rb1 = 0.032 = Co^2$	91 66 (12)
$010 - M_05 - 017$	102 17 (17)	Rb1 = 0.035 = Co2	89.61 (13)
$010 - M_05 - 019$	92.05(15)	025-026	127.5(5)
$010 - M_{0}5 - 024$	157.81 (15)	025 - C1 - C2	127.5(5) 1164(5)
$016 - M_{05} - 017$	149 46 (15)	025 - C1 - C2	116.4(5)
$016 - M_05 - 019$	82 45 (14)	N1 - C2 - C1	110.1(3) 110.0(4)
$016 M_{05} 024$	72 42 (13)	N1 = C2 = C1 N1 = C2 = C3	110.0(4)
$017 M_{0}5 019$	72.72(13)	$C_1 = C_2 = C_3$	110.7(4)
$017 - M_{0}5 - 024$	81 81 (13)	027 - 03 - 02	110.3(5)
017 - M05 - 024	68.03(12)	027 - 029 - 029	112.1(5) 126.5(5)
019 - M05 - 024	105.03(12)	028 C4 C5	120.3(3)
$011 M_{0}6 017$	103.70(16)	028 - C4 - C5	110.2(4)
011 Mo6 018	95.36(15)	$N_{2} = C_{1} = C_{3}$	110.9(3)
$011 M_{0}6 019$	93.30 (15)	$N_2 = C_5 = C_4$ $N_2 = C_5 = C_6$	110.9(4)
$011 M_{0}6 020$	52.55(15)	C_{4} C_{5} C_{6}	110.8(4)
012 Mo6 017	138.22(13) 08.83(16)	$C_{4} C_{5} C_{6} C_{5}$	110.7(5)
012 - M00 - 017	90.03(10)	$C_{3}^{2} = C_{3}^{2} = C_{3}^{2}$	110.0(3)
012 - 100 - 010	99.02 (10) 161 42 (15)	$C_{5} = 0_{27} = \pi_{4}$	107.4/3
012 - 100 - 019	101.42(13) 04.11(15)	$C_0 = 0.00 = 0.01$	109.400
012 - 100 - 020	77.11(13) 151 24 (14)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	109.4/1
$017 M_{0}6 010$	131.34(14) 72.00(12)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	107.433
01/-10100-019	/ 5.00 (15)	U_2 —INI— Π /	107.407

O17—Mo6—O20	84.30 (13)	H5—N1—H6	109.477
O18—Mo6—O19	83.13 (13)	H5—N1—H7	109.484
O18—Mo6—O20	72.22 (13)	H6—N1—H7	109.471
O19—Mo6—O20	68.83 (11)	C5—N2—H13	109.480
O3—Rb1—O8 ⁱ	167.01 (10)	C5—N2—H12	109.476
$03-Rb1-011^{ii}$	109 60 (10)	C5—N2—H14	109 470
O_3 —Rb1— O_12^{ii}	65.85 (10)	H13—N2—H12	109.468
03 - Rb1 - 014	54 60 (10)	H13_N2_H14	109 467
03 - Rb1 - 028	117 88 (9)	H12 - N2 - H14	109 467
03 - Rb1 - 031	104 15 (9)	N1—C2—H1	108 575
03 - Rb1 - 032	95.04 (10)	C1 - C2 - H1	108.597
03 - Rb1 - 035	61 86 (10)	$C_3 - C_2 - H_1$	108 589
O^{gi} Rb1 $O^{11^{\text{ii}}}$	79.66 (10)	$O_{27} C_{3} H_{2}$	100.505
$O8^{i}$ Rb1 $O12^{ii}$	115.00(10)	027 - 027 - 03 - 112	109.170
O_{1}^{i} Pb1 O14	113.90(10) 112.41(11)	$C_2 C_3 H_2$	109.174
$O_{8} = R_{01} = O_{14}$	112.41(11)	$C_2 = C_3 = H_2$	109.171
O_8^{o} R01 $-O_{28}^{\text{o}}$	72.99 (10) (8.50 (10)		109.182
$O8^{\circ}$ Rb1 $O31$	68.50 (10) 92.11 (11)	H2-C3-H3	107.896
08	83.11 (11)	N2-C5-H8	108.109
08'	126.03 (11)	C4—C5—H8	108.111
Olli-Rbl-Ol2	54.54 (10)	C6—C5—H8	108.105
Oll ⁿ —Rbl—Ol4	136.93 (11)	O30—C6—H9	109.516
Ollin—Rb1—O28	73.03 (10)	O30—C6—H10	109.526
O11 ⁿ —Rb1—O31	76.24 (10)	С5—С6—Н9	109.517
O11 ⁱⁱ —Rb1—O32	137.42 (11)	C5—C6—H10	109.524
O11 ⁱⁱ —Rb1—O35	103.69 (10)	H9—C6—H10	108.087
012 ⁱⁱ —Rb1—014	84.44 (10)		
O1—Mo1—O13—Mo2	177.9 (2)	O17—Mo5—O19—Co1	-88.44 (16)
01—Mo1—O18—Mo6	72.4 (2)	019—Mo5—017—Mo6	-21.03(16)
01—Mo1—020—Mo6	-83.55(16)	017—Mo5—024—Mo4	179.52 (14)
01 - Mo1 - 020 - Co1	171 23 (17)	$017 - M_05 - 024 - C_01$	75.09(16)
02 - Mo1 - 013 - Mo2	69 2 (2)	024—Mo5—017—Mo6	-90.48(19)
Ω^2 —Mo1— Ω 18—Mo6	179.83(19)	019 - Mo5 - 024 - Mo4	$104\ 90\ (13)$
Ω^2 —Mo1— Ω^2 0—Mo6	72 5 (4)	019 - Mo5 - 024 - Co1	0.47(12)
$\Omega^2 - Mo1 - \Omega^2 0 - Co1$	-32.7(5)	024—Mo5— 019 —Mo6	103 84 (13)
02 - Mo1 - 020 - C01	-87.70(15)	024 - M05 - 019 - Co1	-0.47(12)
02 - Mo1 - 021 - Co1	168 45 (17)	$011 - M_06 - 012 - Bh1^{iii}$	334(17)
$O_2 = MO_1 = O_2 = CO_1$	-54.4(4)	O12 Mo6 $O11$ Bb1 ⁱⁱⁱ	-3.73(10)
O_{13} Mol O_{13} Mol O_{23}	-55.1 (4)	$011 M_{0}6 017 M_{0}5$	-680(2)
013 Mol - 013 Mol	178 34 (12)	$017 \text{ Mo6} 011 \text{ Ph}^{111}$	-106.31(15)
013 - M01 - 020 - M00	170.34(13)	011 Mo6 018 Mo1	-178.75(10)
013 - M01 - 020 - C01	75.12(13)	018 Mof 011 Ph1ii	-1/8.73(19)
020 - W101 - 013 - W102	-00.30(10)	$010 - 100 - 011 - KD1^{}$	97.23 (14) 85.10 (15)
013 - W01 - 021 - W02	14.1/(12)	011 - W100 - 019 - W103	03.10(13)
013 - M01 - 021 - 001	-69.08(13)	$\begin{array}{c} 011 \\ 010 \\ 010 \\ 011 \\$	-1/0.05(1/)
021 - M01 - 013 - M02	-19.09(15)	$U19 - M100 - U11 - Kb1^{HI}$	-1/9.45(13)
U18—M01—U20—M06	14.84 (12)	O11 - Mob - O20 - Mol	-12.3(4)
018—Mo1—020—Col	-90.39 (15)	011—Mo6—020—Col	33.3 (5)
020-Mo1-018-Mo6	-19.47 (15)	020—Mo6—011—Rb1 ^m	151.0(3)

O18—Mo1—O21—Mo2	177.21 (13)	O12—Mo6—O17—Mo5	-176.0(2)
O18—Mo1—O21—Co1	73.36 (15)	O17—Mo6—O12—Rb1 ⁱⁱⁱ	107.63 (13)
O21—Mo1—O18—Mo6	-88.82(17)	O12—Mo6—O18—Mo1	-71.9(2)
O20—Mo1—O21—Mo2	104.29 (13)	O18—Mo6—O12—Rb1 ⁱⁱⁱ	-94.88(13)
020 - Mo1 - 021 - Co1	0.44(12)	012—Mo6—020—Mo1	83 31 (15)
021 - Mo1 - 020 - Mo6	104.78(13)	012 - Mo6 - 020 - Co1	-171.05(17)
021 - Mo1 - 020 - Co1	-0.45(12)	$O20 - Mo6 - O12 - Bh1^{iii}$	-16751(11)
$04 - M_0^2 - 03 - Rh^1$	-10837(16)	$017 - M_06 - 018 - M_01$	56.0 (4)
$03 - M_0 2 = 013 - M_0 1$	-68.6(2)	$018 - M_06 - 017 - M_05$	56.0 (4)
013 - Mo2 = 013 - Mo1	151.16(12)	$017 - M_06 - 019 - M_05$	-15.82(13)
$O_3 M_{O2} O_1 M_{O3}$	131.10(12) 180.0(2)	O17 Mo6 $O19$ Mo5	80.03 (16)
$O_3 M_{O2} O_1 4 R_{D1}$	7 03 (17)	019 Mo6 017 Mo5	21.13 (16)
014 Mo2 03 Pb1	-5.80(15)	017 Mo6 020 Mo1	$-178\ 21\ (10)$
$O_{14} = MO_{2} = O_{3} = KO_{1}$	5.89 (15) 87 80 (15)	017 Mo6 020 Col	-7257(15)
$O_{2} M_{02} O_{21} M_{01}$	-168 14 (17)	017 - M00 - 020 - 001	72.37(13)
$O_{21} M_{22} O_{21} - C_{01}$	-100.14(17)	020 - M00 - 017 - M03	90.70 (18)
$O_2 = MO_2 = O_3 = RO1$	/8.84 (12)	018 - M06 - 019 - M05	-1/9.78(14)
03 - M02 - 022 - M03	-/4./(4)	018 - M06 - 019 - Col	-74.93(15)
$03 - M_{02} - 022 - C_{01}$	30.6 (5)	019—M06—018—M01	89.48 (17)
022—Mo2—O3—Rb1	50.2 (4)	018—Mo6—020—Mo1	-14.86 (12)
04—Mo2—O13—Mo1	-176.89 (19)	018—Mo6—020—Col	90.78 (15)
O4—Mo2—O14—Mo3	-/2.1 (3)	020—Mo6—018—Mo1	19.57 (15)
O4—Mo2—O14—Rb1	114.93 (16)	O19—Mo6—O20—Mo1	-104.26 (13)
O4—Mo2—O22—Mo3	84.66 (15)	O19—Mo6—O20—Co1	1.38 (12)
O4—Mo2—O22—Co1	-170.00 (17)	O20—Mo6—O19—Mo5	-106.23 (13)
O13—Mo2—O14—Mo3	52.1 (4)	O20—Mo6—O19—Co1	-1.38 (12)
O13—Mo2—O14—Rb1	-120.8 (3)	O3—Rb1—O11 ⁱⁱ —Mo6 ⁱⁱ	40.54 (18)
O14—Mo2—O13—Mo1	58.1 (4)	O11 ⁱⁱ —Rb1—O3—Mo2	-129.20 (14)
O13—Mo2—O21—Mo1	-14.06 (12)	O3—Rb1—O12 ⁱⁱ —Mo6 ⁱⁱ	-142.79 (18)
O13—Mo2—O21—Co1	90.00 (15)	O12 ⁱⁱ —Rb1—O3—Mo2	-95.77 (16)
O21—Mo2—O13—Mo1	19.17 (15)	O3—Rb1—O14—Mo2	-4.67 (12)
O13—Mo2—O22—Mo3	-179.35 (13)	O3—Rb1—O14—Mo3	-176.3 (3)
O13—Mo2—O22—Co1	-74.01 (15)	O14—Rb1—O3—Mo2	5.09 (13)
O22—Mo2—O13—Mo1	88.87 (18)	O3—Rb1—O28—Co2	-53.19 (14)
O14—Mo2—O21—Mo1	-176.28 (14)	O3—Rb1—O28—C4	84.8 (3)
O14—Mo2—O21—Co1	-72.22 (15)	O28—Rb1—O3—Mo2	150.14 (13)
O21—Mo2—O14—Mo3	89.09 (19)	O31—Rb1—O3—Mo2	-49.02 (16)
O21—Mo2—O14—Rb1	-83.85 (14)	O3—Rb1—O32—Co2	90.18 (13)
O14—Mo2—O22—Mo3	-14.44 (12)	O32—Rb1—O3—Mo2	86.11 (15)
O14—Mo2—O22—Co1	90.90 (16)	O3—Rb1—O35—Co2	-156.67 (16)
O22—Mo2—O14—Mo3	19.78 (17)	O35—Rb1—O3—Mo2	135.39 (18)
O22—Mo2—O14—Rb1	-153.17 (17)	O8 ⁱ —Rb1—O11 ⁱⁱ —Mo6 ⁱⁱ	-130.02 (17)
O21—Mo2—O22—Mo3	-105.70 (13)	O11 ⁱⁱ —Rb1—O8 ⁱ —Mo4 ⁱ	105.3 (5)
O21—Mo2—O22—Co1	-0.35 (12)	O8 ⁱ —Rb1—O12 ⁱⁱ —Mo6 ⁱⁱ	51.42 (18)
O22—Mo2—O21—Mo1	-103.71 (13)	O12 ⁱⁱ —Rb1—O8 ⁱ —Mo4 ⁱ	63.4 (5)
O22—Mo2—O21—Co1	0.35 (12)	O8 ⁱ —Rb1—O14—Mo2	175.29 (13)
O5—Mo3—O14—Mo2	178.2 (2)	O8 ⁱ —Rb1—O14—Mo3	3.7 (3)
O5—Mo3—O14—Rb1	-10.9 (3)	O14—Rb1—O8 ⁱ —Mo4 ⁱ	-31.4 (5)
O5—Mo3—O15—Mo4	70.6 (2)	O8 ⁱ —Rb1—O28—Co2	119.11 (13)
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O5—Mo3—O22—Mo2	75.1 (5)	O8 ⁱ —Rb1—O28—C4	-102.9 (3)
O5—Mo3—O22—Co1	-29.8 (5)	O28—Rb1—O8 ⁱ —Mo4 ⁱ	-179.4 (5)
O5—Mo3—O23—Mo4	-83.85 (15)	O31—Rb1—O8 ⁱ —Mo4 ⁱ	26.2 (4)
O5—Mo3—O23—Co1	170.86 (17)	O8 ⁱ —Rb1—O32—Co2	-102.75 (13)
O6—Mo3—O14—Mo2	69.7 (3)	$O32$ —Rb1— $O8^{i}$ —Mo4 ⁱ	-113.8 (5)
O6—Mo3—O14—Rb1	-119.3 (3)	O8 ⁱ —Rb1—O35—Co2	11.12 (18)
O6—Mo3—O15—Mo4	178.83 (19)	O35—Rb1—O8 ⁱ —Mo4 ⁱ	-155.2 (4)
O6—Mo3—O22—Mo2	-86.02 (15)	O11 ⁱⁱ —Rb1—O12 ⁱⁱ —Mo6 ⁱⁱ	-2.37(12)
O6—Mo3—O22—Co1	169.08 (17)	O12 ⁱⁱ —Rb1—O11 ⁱⁱ —Mo6 ⁱⁱ	2.44 (13)
O6—Mo3—O23—Mo4	77.2 (4)	O11 ⁱⁱ —Rb1—O14—Mo2	76.3 (2)
O6—Mo3—O23—Co1	-28.1(5)	O11 ⁱⁱ —Rb1—O14—Mo3	-95.3 (3)
014—Mo3—015—Mo4	-54.2 (4)	O14—Rb1—O11 ⁱⁱ —Mo6 ⁱⁱ	-18.2(3)
$015 - M_03 - 014 - M_02$	-56.4(4)	$O11^{ii}$ Rb1 $O28$ $Co2$	-156.80(13)
015 - Mo3 - 014 - Rb1	114.5 (3)	$O11^{ii}$ Rb1 $O28$ C4	-18.9(3)
$014 - M_03 - 022 - M_02$	14.62 (13)	$O28$ —Rb1— $O11^{ii}$ —Mo6 ⁱⁱ	154.77 (17)
$014 - M_03 - 022 - C_01$	-90.28(16)	O_{31} Rb1 O_{11}^{ii} Mo6 ⁱⁱ	-59.83(15)
022 - Mo3 - 014 - Mo2	-1967(17)	$O11^{ii}$ _Rb1_ $O32$ _Co2	-362(2)
022 - Mo3 - 014 - Rb1	151 3 (3)	O_{32} Rb1 O_{11i} Mo6 ⁱⁱ	162.24(13)
014 - Mo3 - 023 - Mo4	179 36 (14)	$O11^{ii}$ _Rb1_O35_Co2	98 18 (13)
014 - Mo3 - 023 - Co1	74.06 (16)	$O35$ —Rb1— $O11^{ii}$ —Mo6 ⁱⁱ	105 17 (16)
$023 - M_03 - 014 - M_02$	-88.66(19)	$O12^{ii}$ Rb1 $O14$ Mo2	59 53 (15)
023 - Mo3 - 014 - Rb1	82.3 (2)	$O12^{ii}$ Rb1 $O14$ Mo2	-1121(3)
015 - Mo3 - 022 - Mo2	$177\ 20\ (14)$	014 Rb1 012^{ii} Mo6 ⁱⁱ	163 66 (16)
015 - Mo3 - 022 - Co1	72 30 (15)	$O12^{ii}$ Rb1 $O28$ $O2$	-13050(10)
$022 - M_0 3 - 015 - M_0 4$	-89.09(17)	$O12^{ii}$ Rb1 $O28$ C02	74(3)
015 - Mo3 - 023 - Mo4	15 57 (12)	0.12 Rol 0.20 0.1	-3372(19)
015 - Mo3 - 023 - Co1	-89.73(16)	O_{31} Rb1 O_{12}^{ii} Mo6 ⁱⁱ	92 19 (16)
0^{23} Mo ³ 0^{15} Mo ⁴	-20.29(15)	$O12^{ii}$ Rb1 $O35$ Co2	154.91(11)
022 - Mo3 - 023 - Mo4	105 96 (13)	O_{12} Rol O_{23} O_{23}	-9650(15)
022 - Mo3 - 023 - Co1	0.66 (12)	014—Rb1— 028 —Co2	137(3)
022 - M03 - 022 - M02	$104\ 23\ (13)$	014 Rb1 020 002	151.6(2)
023 - Mo3 - 022 - Co1	-0.67(13)	O28—Rb1—O14—Mo2	-903(3)
0.25 Mos 0.22 cor 07-Mo4-Mo5-09	-108 32 (17)	O_{28} Rb1 O_{14} Mo2	98.1.(3)
$07 - M_04 - M_05 - 010$	1 64 (18)	O_{31} Rb1 O_{14} Mo2	119 27 (16)
07 - M04 - M05 - 016	4 43 (17)	O_{31} Rb1 O_{14} Mo2	-523(2)
$07 - M_04 - M_05 - 017$	153 37 (17)	014—Rb1— 032 — $Co2$	$144\ 03\ (13)$
$07 - M_04 - M_05 - 019$	89 66 (17)	O_{32} Rb1 O_{14} Mo2	$-104\ 00\ (16)$
$07 - M_04 - M_05 - 024$	153 96 (18)	O_{32} Rb1 O_{14} Mo3	84 4 (3)
M_{05} Mo4 M_{05} M_{05} M_{04} M_{05} M_{05} M_{04} M_{05}	94 4 (5)	014 —Rb1— 035 — Co^2	-117 37 (12)
08-M04-M05-09	2 23 (12)	O_{35} Rb1 O_{14} Mo2	-47.92(16)
$08 - M_04 - M_05 - 010$	112, 20(12)	O_{35} Rb1 O_{14} Mo3	140 5 (2)
$08 - M_04 - M_05 - 016$	112.20 (12)	O_{31} Rb1 O_{28} C_{02}	153.09(10)
$08 - M_04 - M_05 - 017$	-96.07(12)	O_{31} Rb1 O_{28} C_{4}	-690(3)
$08 - M_04 - M_05 - 019$	-15978(12)	O28—Rb1— $O32$ — $Co2$	-28.34(11)
08-M04-M05-024	-95.48 (12)	O_{32} Rb1 O_{28} C_{02}	28.78 (11)
Mo5—Mo4—O15—Mo3	90.33 (17)	O_{32} Rb1 O_{28} C4	166.7 (3)
015—Mo4—Mo5—O9	97.86 (12)	O28—Rb1—O35—Co2	38.75 (11)
O15—Mo4—Mo5—O10	-152.17 (12)	O35—Rb1—O28—Co2	-37.80 (11)
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O15—Mo4—Mo5—O16	-149.39 (12)	O35—Rb1—O28—C4	100.1 (3)
O15—Mo4—Mo5—O17	-0.44 (12)	O31—Rb1—O32—Co2	-155.87 (10)
O15—Mo4—Mo5—O19	-64.15 (12)	O32—Rb1—O35—Co2	-39.72 (12)
O15—Mo4—Mo5—O24	0.15 (12)	O35—Rb1—O32—Co2	38.08 (11)
Mo5—Mo4—O16—Mo5	0.000 (13)	O19—Co1—O20—Mo1	96.40 (14)
O16—Mo4—Mo5—O9	-112.8(2)	O19—Co1—O20—Mo6	-1.54 (14)
O16—Mo4—Mo5—O10	-2.8 (2)	O20—Co1—O19—Mo5	97.16 (14)
O16—Mo4—Mo5—O16	0.0 (2)	O20—Co1—O19—Mo6	1.53 (14)
O16—Mo4—Mo5—O17	148.9 (2)	O19—Co1—O21—Mo1	-84.05 (14)
O16—Mo4—Mo5—O19	85.2 (2)	O19—Co1—O21—Mo2	-179.76 (14)
O16—Mo4—Mo5—O24	149.5 (2)	O21—Co1—O19—Mo5	-179.00 (14)
Mo5—Mo4—O23—Mo3	-146.22 (9)	O21—Co1—O19—Mo6	85.36 (14)
Mo5—Mo4—O23—Co1	-40.60 (12)	O19—Co1—O23—Mo3	179.18 (14)
O23—Mo4—Mo5—O9	161.34 (8)	O19—Co1—O23—Mo4	81.48 (15)
O23—Mo4—Mo5—O10	-88.70 (9)	O23—Co1—O19—Mo5	-82.65 (15)
O23—Mo4—Mo5—O16	-85.91 (8)	O23—Co1—O19—Mo6	-178.29 (14)
O23—Mo4—Mo5—O17	63.03 (8)	O19—Co1—O24—Mo4	-95.67 (15)
O23—Mo4—Mo5—O19	-0.68(8)	O19—Co1—O24—Mo5	-0.53 (14)
O23—Mo4—Mo5—O24	63.62 (8)	O24—Co1—O19—Mo5	0.53 (14)
Mo5—Mo4—O24—Mo5	0.0	O24—Co1—O19—Mo6	-95.11 (15)
Mo5—Mo4—O24—Co1	104.66 (16)	O20—Co1—O21—Mo1	-0.49(13)
O24—Mo4—Mo5—O9	97.71 (12)	O20—Co1—O21—Mo2	-96.20 (14)
O24—Mo4—Mo5—O10	-152.32(12)	O21—Co1—O20—Mo1	0.50 (14)
024—Mo4—Mo5—O16	-149.54(12)	021 - Co1 - 020 - Mo6	-97.44(14)
024 - Mo4 - Mo5 - 017	-0.59(12)	020—Co1— 022 —Mo2	84.29 (14)
024—Mo4—Mo5—O19	-64.30(12)	020—Co1— 022 —Mo3	-179.08(14)
$024 - M_04 - M_05 - 024$	0.00(12)	022—Co1—O20—Mo1	-83.68(14)
$07-Mo4-08-Rb1^{v}$	-1330(5)	022 - Col - 020 - Moc	178 39 (14)
07 - Mo4 - 015 - Mo3	-700(2)	020—Co1— 024 —Mo4	-179 13 (14)
07 - Mo4 - 016 - Mo5	-1766(2)	020 Col 021 Mot	-83.98(14)
07 - Mo4 - 023 - Mo3	82 94 (16)	020 - 021 - 021 - 020 - 021	179 91 (14)
$07 - M_0 4 - 023 - C_0 1$	-17143(17)	024—Co1— 020 —Mo6	81 97 (14)
08 - Mo4 - 015 - Mo3	-177 30 (19)	021 - Co1 - 022 - Mo2	0.39(13)
$015 - M_0 4 - 08 - Rh^{1}$	-322(5)	021 - Co1 - 022 - Mo2	97.02 (14)
$08 - M_0 4 - 016 - M_0 5$	-67.8(3)	022 - Co1 - 021 - Mo1	95 31 (14)
$016 - Mo4 - 08 - Rh1^{v}$	123 6 (5)	022 - Col - 021 - Mo1	-0.39(13)
$08 - M_0 4 - 023 - M_0 3$	-710(4)	022 - Co1 - 021 - Mo2	-84.80(15)
$O_8 M_0 4 O_{23} C_{01}$	34.7(5)	O_{21}^{-1} Col O_{23}^{-1} Mod	17750(14)
$O_{23} M_{04} O_{8} R_{1}^{v}$	10.8 (8)	021 - 001 - 023 - 004	177.30(14) 178.48(14)
$O_2 Mo_4 O_2 Mo_5$	19.0 (0) 85 21 (15)	$O_{23}^{23} = Co_{1}^{23} = O_{21}^{23} = Mo_{1}^{23}$	170.40(14) 82.78(14)
$08 M_04 024 C_01$	-17013(17)	$O_{23}^{23} = Co_{1}^{23} = O_{23}^{23} = Mo_{23}^{23}$	-0.74(14)
$O_{24} M_{24} O_{24} D_{11}$	1/0.13(17)	$O_{22} = C_{01} = O_{23} = M_{03}$	-08.44(14)
024 M04 0.6 Mo5	50.2(3)	022 - 01 - 023 - 004	-96.44(13)
015 - M04 - 010 - M03	50.1(4)	$O_{23} = C_{01} = O_{22} = M_{02}$	95.89(15)
$015 M_0 4 022 M_0^2$	-15 34 (12)	022 - 01 - 022 - 003	8/96(15)
015 Mod 023 Col	13.34(12)	$O_{22} = O_{1} = O_{24} = W_{104}$	$-170\ 80\ (13)$
013 - 1004 - 023 - 001	20.20(10)	$O_{22} = O_{1} = O_{24} = W_{103}$	1/9.09(14) -170 14(14)
023 - W104 - 013 - W103	20.49(13)	$O_{24} = O_{1} = O_{22} = M_{02}$	1/9.14(14) -92.51(15)
013—W04—024—W105	-1/9.88 (14)	024—001—022—M03	-82.31 (13)

O15—Mo4—O24—Co1	-75.22 (15)	O23—Co1—O24—Mo4	1.89 (14)
O24—Mo4—O15—Mo3	90.43 (17)	O23—Co1—O24—Mo5	97.04 (15)
O16—Mo4—O23—Mo3	-178.02 (14)	O24—Co1—O23—Mo3	95.80 (15)
O16—Mo4—O23—Co1	-72.40 (16)	O24—Co1—O23—Mo4	-1.90 (14)
O23—Mo4—O16—Mo5	91.14 (19)	O28—Co2—O32—Rb1	37.56 (13)
O16—Mo4—O24—Mo5	-16.17 (13)	O32—Co2—O28—Rb1	-35.90 (12)
O16—Mo4—O24—Co1	88.49 (17)	O32—Co2—O28—C4	-175.6 (3)
O24—Mo4—O16—Mo5	21.76 (17)	O33—Co2—O28—Rb1	-122.99 (13)
O23—Mo4—O24—Mo5	-106.37 (13)	O33—Co2—O28—C4	97.3 (3)
O23—Mo4—O24—Co1	-1.71 (13)	O34—Co2—O28—Rb1	140.73 (12)
O24—Mo4—O23—Mo3	-103.91 (13)	O34—Co2—O28—C4	1.0 (3)
O24—Mo4—O23—Co1	1.71 (13)	O28—Co2—O35—Rb1	-48.56 (12)
Mo4—Mo5—O16—Mo4	0.000 (14)	O35—Co2—O28—Rb1	48.91 (13)
Mo4—Mo5—O17—Mo6	-90.06 (18)	O35—Co2—O28—C4	-90.8 (3)
Mo4—Mo5—O19—Mo6	146.27 (9)	O33—Co2—O32—Rb1	121.10 (14)
Mo4—Mo5—O19—Co1	41.96 (12)	O32—Co2—O35—Rb1	46.77 (13)
Mo4—Mo5—O24—Mo4	0.0	O35—Co2—O32—Rb1	-49.60 (13)
Mo4—Mo5—O24—Co1	-104.43 (16)	O36—Co2—O32—Rb1	-148.82 (15)
O9—Mo5—O16—Mo4	69.6 (3)	O34—Co2—O35—Rb1	-140.10 (12)
O9—Mo5—O17—Mo6	176.3 (2)	O36—Co2—O35—Rb1	136.14 (13)
O9—Mo5—O24—Mo4	-83.39 (16)	Rb1—O28—C4—O29	-110.3 (5)
O9—Mo5—O24—Co1	172.19 (18)	Rb1—O28—C4—C5	71.0 (5)
O10-Mo5-O16-Mo4	177.8 (2)	Co2—O28—C4—O29	13.8 (7)
O10—Mo5—O17—Mo6	67.3 (3)	Co2—O28—C4—C5	-164.9 (3)
O10—Mo5—O19—Mo6	-86.24 (16)	O25—C1—C2—N1	165.7 (4)
O10-Mo5-O19-Co1	169.46 (18)	O25—C1—C2—C3	-71.9 (6)
O10-Mo5-O24-Mo4	77.3 (5)	O26—C1—C2—N1	-15.5 (6)
O10-Mo5-O24-Co1	-27.1 (5)	O26—C1—C2—C3	106.9 (5)
O16—Mo5—O17—Mo6	-58.1 (4)	N1—C2—C3—O27	61.8 (6)
O17—Mo5—O16—Mo4	-55.4 (4)	C1—C2—C3—O27	-60.2 (6)
O16—Mo5—O19—Mo6	177.87 (14)	O28—C4—C5—N2	6.4 (6)
O16—Mo5—O19—Co1	73.57 (16)	O28—C4—C5—C6	129.8 (4)
O19—Mo5—O16—Mo4	-90.95 (19)	O29—C4—C5—N2	-172.4 (4)
O16—Mo5—O24—Mo4	16.10 (13)	O29—C4—C5—C6	-49.0 (6)
O16—Mo5—O24—Co1	-88.33 (17)	N2-C5-C6-O30	66.2 (5)
O24—Mo5—O16—Mo4	-21.67 (17)	C4—C5—C6—O30	-57.3 (6)
O17—Mo5—O19—Mo6	15.87 (12)		

Symmetry codes: (i) -x+1, y-1/2, -z+1/2; (ii) -x+2, y-1/2, -z+1/2; (iii) -x+2, y+1/2, -z+1/2; (iv) x+1, y, z; (v) -x+1, y+1/2, -z+1/2; (vi) x, y+1, z; (vii) x+1/2, -y+1/2, -z; (viii) x+1/2, -y+3/2, -z; (ix) x-1/2, -y+1/2, -z; (x) x-1, y, z; (x) -x+3/2, -y+1, z+1/2; (xii) x-1/2, -y+3/2, -z; (xiii) -x+3/2, -y+1, z-1/2; (xiv) x+1, y-1, z; (xvi) x-1, y+1, z; (xvi) x-1, y+1, z; (xvi) x-1, y-1, z; (xvi) x-1, y+1, z; (xvi) x-1, y+1, z; (xvi) x-1, y-1, z.

Hydrogen-bond geometry (Å, °))
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D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	<i>D</i> —H··· <i>A</i>
027—H4…O41	0.84	2.24	2.859 (6)	131
O27—H4…N1	0.84	2.62	2.940 (6)	104
O30—H11…O40	0.84	2.56	2.708 (6)	91
O30—H11…O42 ^{vii}	0.84	2.24	2.928 (6)	139

supporting information

N1—H5····O2	0.91	2.11	2.944 (6)	151	
N1—H6…O26	0.91	2.49	2.647 (6)	89	
N1—H6…O39 ^{vii}	0.91	2.14	3.041 (6)	170	
N1—H7…O17 ⁱⁱ	0.91	2.09	2.905 (6)	149	
N2—H13…O28	0.91	2.60	2.691 (6)	86	
N2—H13…O43	0.91	2.11	3.013 (7)	170	
N2—H12…O8 ⁱ	0.91	2.31	2.912 (6)	123	
N2—H12…O28	0.91	2.61	2.691 (6)	85	
N2—H14····O13 ^{xvii}	0.91	1.93	2.827 (6)	167	

Symmetry codes: (i) -*x*+1, *y*-1/2, -*z*+1/2; (ii) -*x*+2, *y*-1/2, -*z*+1/2; (vii) *x*+1/2, -*y*+1/2, -*z*; (xvii) *x*, *y*-1, *z*.