

Fluazinam

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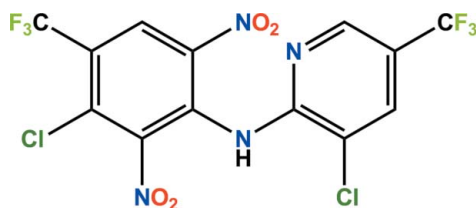
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.032; wR factor = 0.085; data-to-parameter ratio = 15.6.

In the asymmetric unit of the title compound {systematic name: 3-chloro-*N*-[3-chloro-5-(trifluoromethyl)pyridin-2-yl]-2,6-dinitro-4-(trifluoromethyl)aniline}, $\text{C}_{13}\text{H}_4\text{Cl}_2\text{F}_6\text{N}_4\text{O}_4$, which is the fungicide fluazinam, the dihedral angle between the pyridine and benzene ring planes is $42.20(4)^\circ$. In the crystal, pairs of $\text{N}-\text{H}\cdots\text{F}$ hydrogen bonds link the molecules into inversion dimers which are linked by $\text{C}-\text{Cl}\cdots\pi$ [$\text{Cl}\cdots\text{ring centroid} = 3.3618(4)$ Å] and $\text{N}-\text{O}\cdots\pi$ [$\text{O}\cdots\text{ring centroid} = 3.1885(16)$ Å] interactions into chains along [100]. In addition, short $\text{Cl}\cdots\text{Cl}$, $\text{O}\cdots\text{Cl}$, and $\text{F}\cdots\text{F}$ contacts [3.4676 (7), 3.2371 (13) and 2.7910 (15) Å] are present which connect the chains, yielding a three-dimensional network.

Related literature

For information on the toxicity and fungicidal properties of the title compound, see: Yoshida & Yukimoto (1993); Draper *et al.* (2003). For a related structure, see: McCullough *et al.* (1972).



Experimental

Crystal data

$\text{C}_{13}\text{H}_4\text{Cl}_2\text{F}_6\text{N}_4\text{O}_4$

$M_r = 465.10$

Triclinic, $P\bar{1}$

$a = 8.9546(1)$ Å

$b = 9.0724(1)$ Å

$c = 10.6818(2)$ Å

$\alpha = 79.556(1)^\circ$

$\beta = 75.420(1)^\circ$

$\gamma = 83.451(1)^\circ$

$V = 823.79(2)$ Å³

$Z = 2$

Mo $K\alpha$ radiation

$\mu = 0.49$ mm⁻¹

$T = 173$ K

$0.30 \times 0.18 \times 0.15$ mm

Data collection

Bruker APEXII CCD detector
diffractometer

Absorption correction: multi-scan
(*SADABS*; Bruker, 2006)

$T_{\min} = 0.867$, $T_{\max} = 0.930$

15193 measured reflections

4096 independent reflections

3662 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.022$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.032$

$wR(F^2) = 0.085$

$S = 1.04$

4096 reflections

262 parameters

H-atom parameters constrained

$\Delta\rho_{\text{max}} = 0.47$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.45$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N3}-\text{H3}\cdots\text{F3}^i$	0.88	2.52	3.0690 (15)	121

Symmetry code: (i) $-x + 2, -y + 1, -z + 1$.

Data collection: *APEX2* (Bruker, 2006); cell refinement: *SAINT* (Bruker, 2006); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008) and *DIAMOND* (Brandenburg, 1998); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: KJ2230).

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supporting information

Acta Cryst. (2013). E69, o1467 [doi:10.1107/S1600536813023210]

Fluazinam

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S1. Comment

The title compound fluazinam, $C_{13}H_4Cl_2F_6N_4O_4$, is a broad spectrum contact fungicide that can be applied as a foliar spray or soil treatment (Yoshida *et al.*, 1993; Draper *et al.*, 2003) and its crystal structure is reported herein. In this compound (Fig. 1), the dihedral angle between the pyridyl ring and the phenyl ring is $42.20(4)^\circ$. All bond lengths and bond angles are normal and comparable to those observed in the crystal structures of a similar compound (McCullough *et al.*, 1972). In the crystal structure (Fig. 2), an intermolecular N—H \cdots F hydrogen bond is observed (Table 1), giving a dimer structure. In this structure there are both a C3—C11 $\cdots\pi$ interaction with the pyridyl ring [C11 \cdots Cg1ⁱⁱ = 3.3618 (4) Å] and a N1—O1 $\cdots\pi$ interaction with the phenyl ring [O1 \cdots Cg2ⁱⁱ = 3.1885 (16) Å]. In addition, short C1 \cdots Cl, O \cdots Cl, and F \cdots F contacts [C12 \cdots Cl2ⁱⁱⁱ, 3.4676 (7) Å, O3 \cdots Cl1^{iv}, 3.2371 (13) Å, and F3 \cdots F6^v, 2.7910 (15) Å] are present [for symmetry codes: (ii), $-x+1, -y+1, -z+1$, (iii), $-x+1, -y+1, -z+2$, (iv), $x, y-1, z$, and (v), $x+1, y+1, z+1$]. A three-dimensional network is formed by the hydrogen bond and these interactions.

S2. Experimental

The title compound was purchased from the Dr. Ehrenstorfer GmbH Company. Slow evaporation of a solution in CH_2Cl_2 gave single crystals suitable for X-ray analysis.

S3. Refinement

All H-atoms were positioned geometrically and refined using a riding model with $d(N-H) = 0.88$ Å, $U_{iso} = 1.2U_{eq}(C)$ for amine and $d(C-H) = 0.95$ Å, $U_{iso} = 1.2U_{eq}(C)$ for Csp^2-H .

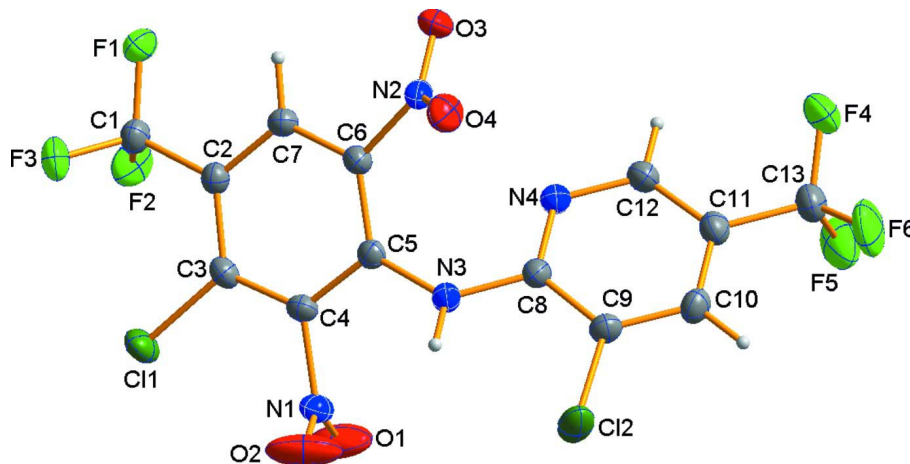
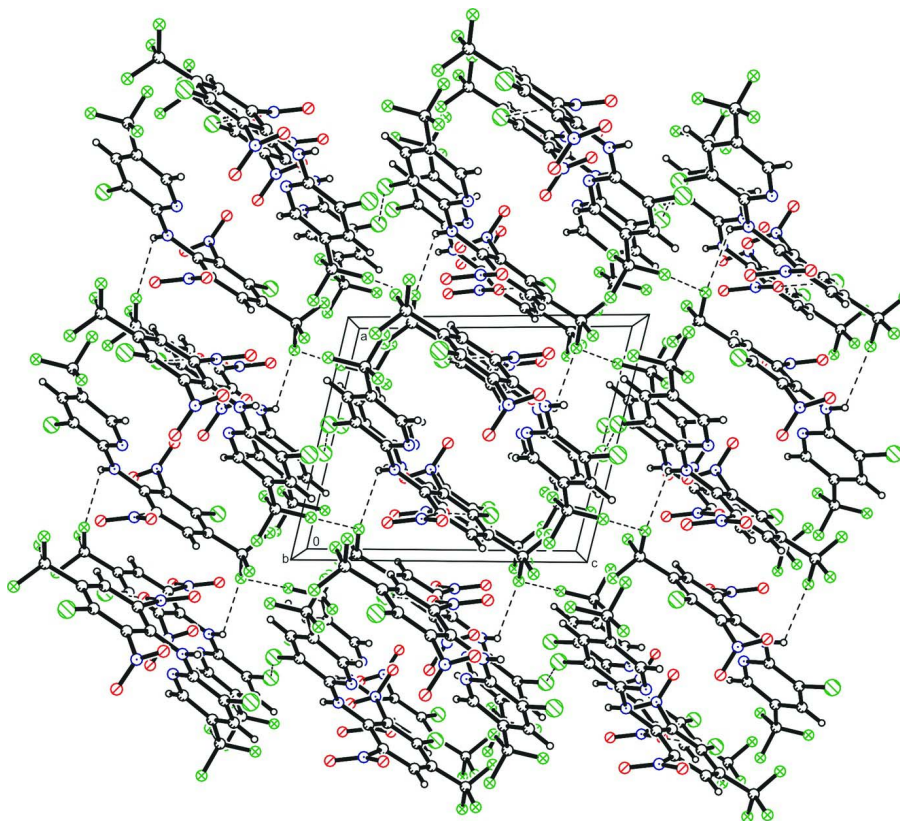


Figure 1

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radius.

**Figure 2**

Crystal packing of the title compound with N—H...F hydrogen bonds and weak intermolecular Cl...Cl and F...F interactions shown as dashed lines.

3-Chloro-*N*-[3-chloro-5-(trifluoromethyl)pyridin-2-yl]-2,6-dinitro-4-(trifluoromethyl)aniline

Crystal data

$C_{13}H_4Cl_2F_6N_4O_4$

$M_r = 465.10$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 8.9546\ (1)\ \text{\AA}$

$b = 9.0724\ (1)\ \text{\AA}$

$c = 10.6818\ (2)\ \text{\AA}$

$\alpha = 79.556\ (1)^\circ$

$\beta = 75.420\ (1)^\circ$

$\gamma = 83.451\ (1)^\circ$

$V = 823.79\ (2)\ \text{\AA}^3$

$Z = 2$

$F(000) = 460$

$D_x = 1.875\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 8261 reflections

$\theta = 2.4\text{--}28.3^\circ$

$\mu = 0.49\ \text{mm}^{-1}$

$T = 173\ \text{K}$

Block, yellow

$0.30 \times 0.18 \times 0.15\ \text{mm}$

Data collection

Bruker APEXII CCD detector

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2006)

$T_{\min} = 0.867$, $T_{\max} = 0.930$

15193 measured reflections

4096 independent reflections

3662 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.022$

$\theta_{\max} = 28.4^\circ$, $\theta_{\min} = 2.0^\circ$

$h = -11 \rightarrow 11$

$k = -12 \rightarrow 10$

$l = -14 \rightarrow 14$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.032$
 $wR(F^2) = 0.085$
 $S = 1.04$
 4096 reflections
 262 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0402P)^2 + 0.3959P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.47 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.45 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.83432 (5)	0.72367 (4)	0.34806 (4)	0.03069 (10)
Cl2	0.43809 (5)	0.32745 (5)	0.99480 (4)	0.03760 (11)
F1	1.06852 (11)	0.34123 (11)	0.15585 (9)	0.0342 (2)
F2	0.90981 (11)	0.53398 (12)	0.12664 (9)	0.0369 (2)
F3	1.11164 (11)	0.55470 (11)	0.19484 (10)	0.0357 (2)
F4	0.25749 (13)	-0.25110 (12)	0.84106 (12)	0.0484 (3)
F5	0.07528 (12)	-0.07681 (13)	0.86228 (14)	0.0542 (3)
F6	0.17049 (16)	-0.17401 (14)	1.02625 (11)	0.0572 (3)
N1	0.63148 (15)	0.61200 (13)	0.60382 (12)	0.0257 (3)
N2	0.82927 (13)	0.07922 (13)	0.58834 (12)	0.0239 (2)
N3	0.62868 (14)	0.31709 (14)	0.72658 (11)	0.0234 (2)
H3	0.6329	0.3764	0.7821	0.028*
N4	0.52522 (14)	0.10739 (14)	0.69279 (12)	0.0240 (2)
O1	0.51649 (16)	0.65800 (18)	0.56501 (17)	0.0595 (4)
O2	0.6693 (2)	0.65407 (19)	0.68990 (16)	0.0653 (5)
O3	0.83309 (13)	-0.01445 (12)	0.51781 (11)	0.0321 (2)
O4	0.84483 (13)	0.05111 (13)	0.70083 (11)	0.0321 (2)
C1	1.00024 (17)	0.46373 (17)	0.20469 (14)	0.0257 (3)
C2	0.90782 (15)	0.42479 (15)	0.34387 (13)	0.0210 (3)
C3	0.82543 (16)	0.53636 (15)	0.41309 (13)	0.0210 (3)
C4	0.73336 (15)	0.49449 (15)	0.53729 (13)	0.0209 (3)
C5	0.72421 (15)	0.34551 (15)	0.60148 (13)	0.0197 (3)
C6	0.81363 (15)	0.23826 (15)	0.53006 (13)	0.0204 (3)
C7	0.90021 (16)	0.27574 (15)	0.40317 (13)	0.0215 (3)
H7	0.9548	0.1987	0.3564	0.026*

C8	0.52749 (15)	0.20410 (15)	0.77180 (13)	0.0208 (3)
C9	0.42873 (16)	0.19730 (16)	0.89770 (14)	0.0238 (3)
C10	0.32506 (17)	0.08814 (17)	0.94103 (14)	0.0267 (3)
H10	0.2564	0.0818	1.0253	0.032*
C11	0.32343 (17)	-0.01298 (16)	0.85797 (15)	0.0257 (3)
C12	0.42492 (17)	0.00019 (16)	0.73578 (15)	0.0261 (3)
H12	0.4234	-0.0698	0.6800	0.031*
C13	0.20746 (19)	-0.12907 (18)	0.89734 (17)	0.0332 (3)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0402 (2)	0.01712 (16)	0.03235 (19)	-0.00298 (13)	-0.00704 (15)	0.00063 (13)
C12	0.0392 (2)	0.0455 (2)	0.02904 (19)	-0.01359 (17)	0.00518 (15)	-0.02051 (17)
F1	0.0373 (5)	0.0320 (5)	0.0270 (4)	-0.0033 (4)	0.0067 (4)	-0.0076 (4)
F2	0.0418 (5)	0.0452 (6)	0.0211 (4)	0.0003 (4)	-0.0090 (4)	0.0013 (4)
F3	0.0328 (5)	0.0331 (5)	0.0372 (5)	-0.0146 (4)	-0.0007 (4)	0.0018 (4)
F4	0.0441 (6)	0.0303 (5)	0.0704 (8)	-0.0134 (4)	-0.0029 (5)	-0.0154 (5)
F5	0.0303 (5)	0.0468 (7)	0.0869 (9)	-0.0104 (5)	-0.0176 (6)	-0.0038 (6)
F6	0.0738 (8)	0.0510 (7)	0.0409 (6)	-0.0357 (6)	0.0012 (6)	0.0069 (5)
N1	0.0302 (6)	0.0191 (6)	0.0264 (6)	0.0011 (5)	-0.0048 (5)	-0.0047 (5)
N2	0.0218 (6)	0.0194 (6)	0.0264 (6)	-0.0010 (4)	0.0003 (4)	-0.0018 (5)
N3	0.0264 (6)	0.0231 (6)	0.0203 (5)	-0.0068 (4)	0.0004 (4)	-0.0074 (5)
N4	0.0246 (6)	0.0230 (6)	0.0237 (6)	-0.0033 (4)	-0.0018 (4)	-0.0067 (5)
O1	0.0400 (7)	0.0658 (10)	0.0874 (11)	0.0247 (7)	-0.0296 (8)	-0.0471 (9)
O2	0.0887 (12)	0.0629 (10)	0.0628 (10)	0.0380 (9)	-0.0454 (9)	-0.0464 (8)
O3	0.0366 (6)	0.0195 (5)	0.0370 (6)	-0.0018 (4)	0.0002 (5)	-0.0092 (4)
O4	0.0354 (6)	0.0287 (6)	0.0283 (6)	0.0002 (4)	-0.0075 (4)	0.0039 (4)
C1	0.0263 (7)	0.0261 (7)	0.0224 (7)	-0.0049 (5)	-0.0024 (5)	-0.0013 (5)
C2	0.0211 (6)	0.0221 (6)	0.0195 (6)	-0.0036 (5)	-0.0044 (5)	-0.0021 (5)
C3	0.0245 (6)	0.0168 (6)	0.0227 (6)	-0.0032 (5)	-0.0081 (5)	-0.0009 (5)
C4	0.0232 (6)	0.0184 (6)	0.0223 (6)	-0.0007 (5)	-0.0057 (5)	-0.0064 (5)
C5	0.0199 (6)	0.0205 (6)	0.0191 (6)	-0.0034 (5)	-0.0040 (5)	-0.0044 (5)
C6	0.0213 (6)	0.0175 (6)	0.0218 (6)	-0.0019 (5)	-0.0042 (5)	-0.0025 (5)
C7	0.0225 (6)	0.0199 (6)	0.0218 (6)	-0.0011 (5)	-0.0032 (5)	-0.0052 (5)
C8	0.0205 (6)	0.0200 (6)	0.0209 (6)	-0.0011 (5)	-0.0033 (5)	-0.0031 (5)
C9	0.0254 (7)	0.0250 (7)	0.0209 (6)	-0.0016 (5)	-0.0027 (5)	-0.0072 (5)
C10	0.0248 (7)	0.0299 (8)	0.0220 (7)	-0.0044 (6)	0.0000 (5)	-0.0020 (6)
C11	0.0246 (7)	0.0215 (7)	0.0296 (7)	-0.0035 (5)	-0.0054 (5)	-0.0005 (6)
C12	0.0268 (7)	0.0219 (7)	0.0298 (7)	-0.0030 (5)	-0.0046 (6)	-0.0070 (6)
C13	0.0321 (8)	0.0272 (8)	0.0377 (8)	-0.0090 (6)	-0.0032 (6)	-0.0014 (6)

Geometric parameters (Å, °)

C11—C3	1.7180 (13)	N4—C8	1.3280 (18)
C12—C9	1.7283 (14)	N4—C12	1.3391 (18)
F1—C1	1.3326 (17)	C1—C2	1.5090 (19)
F2—C1	1.3366 (17)	C2—C7	1.3868 (19)

F3—C1	1.3382 (17)	C2—C3	1.3936 (19)
F4—C13	1.334 (2)	C3—C4	1.3864 (19)
F5—C13	1.340 (2)	C4—C5	1.4017 (19)
F6—C13	1.330 (2)	C5—C6	1.3978 (18)
N1—O2	1.1911 (18)	C6—C7	1.3849 (18)
N1—O1	1.2084 (18)	C7—H7	0.9500
N1—C4	1.4834 (17)	C8—C9	1.4065 (19)
N2—O4	1.2220 (16)	C9—C10	1.373 (2)
N2—O3	1.2263 (16)	C10—C11	1.391 (2)
N2—C6	1.4701 (17)	C10—H10	0.9500
N3—C8	1.3846 (17)	C11—C12	1.384 (2)
N3—C5	1.3906 (17)	C11—C13	1.495 (2)
N3—H3	0.8800	C12—H12	0.9500
O2—N1—O1	125.30 (14)	C7—C6—C5	122.46 (12)
O2—N1—C4	117.99 (13)	C7—C6—N2	115.65 (12)
O1—N1—C4	116.71 (12)	C5—C6—N2	121.78 (12)
O4—N2—O3	125.24 (12)	C6—C7—C2	120.52 (13)
O4—N2—C6	117.51 (12)	C6—C7—H7	119.7
O3—N2—C6	117.15 (12)	C2—C7—H7	119.7
C8—N3—C5	126.00 (12)	N4—C8—N3	118.35 (12)
C8—N3—H3	117.0	N4—C8—C9	122.28 (13)
C5—N3—H3	117.0	N3—C8—C9	119.35 (12)
C8—N4—C12	118.20 (12)	C10—C9—C8	119.38 (13)
F1—C1—F2	107.50 (12)	C10—C9—C12	120.99 (11)
F1—C1—F3	107.14 (12)	C8—C9—C12	119.63 (11)
F2—C1—F3	106.99 (12)	C9—C10—C11	118.07 (13)
F1—C1—C2	111.51 (12)	C9—C10—H10	121.0
F2—C1—C2	111.11 (12)	C11—C10—H10	121.0
F3—C1—C2	112.33 (12)	C12—C11—C10	119.23 (13)
C7—C2—C3	119.09 (12)	C12—C11—C13	120.38 (14)
C7—C2—C1	119.92 (12)	C10—C11—C13	120.30 (14)
C3—C2—C1	120.96 (12)	N4—C12—C11	122.83 (14)
C4—C3—C2	118.91 (12)	N4—C12—H12	118.6
C4—C3—C11	119.39 (11)	C11—C12—H12	118.6
C2—C3—C11	121.70 (10)	F6—C13—F4	107.38 (14)
C3—C4—C5	123.74 (12)	F6—C13—F5	106.72 (14)
C3—C4—N1	118.60 (12)	F4—C13—F5	106.11 (14)
C5—C4—N1	117.56 (12)	F6—C13—C11	112.20 (14)
N3—C5—C6	126.22 (12)	F4—C13—C11	112.41 (13)
N3—C5—C4	118.65 (12)	F5—C13—C11	111.63 (13)
C6—C5—C4	115.13 (12)		
F1—C1—C2—C7	-1.24 (18)	O3—N2—C6—C7	-41.24 (17)
F2—C1—C2—C7	118.67 (14)	O4—N2—C6—C5	-40.98 (18)
F3—C1—C2—C7	-121.52 (14)	O3—N2—C6—C5	142.43 (13)
F1—C1—C2—C3	-179.22 (12)	C5—C6—C7—C2	3.5 (2)
F2—C1—C2—C3	-59.31 (17)	N2—C6—C7—C2	-172.80 (12)

F3—C1—C2—C3	60.50 (17)	C3—C2—C7—C6	-0.9 (2)
C7—C2—C3—C4	-2.56 (19)	C1—C2—C7—C6	-178.88 (12)
C1—C2—C3—C4	175.44 (12)	C12—N4—C8—N3	-178.61 (13)
C7—C2—C3—C11	177.84 (10)	C12—N4—C8—C9	-0.1 (2)
C1—C2—C3—C11	-4.16 (18)	C5—N3—C8—N4	3.9 (2)
C2—C3—C4—C5	3.7 (2)	C5—N3—C8—C9	-174.67 (13)
C11—C3—C4—C5	-176.73 (10)	N4—C8—C9—C10	-0.5 (2)
C2—C3—C4—N1	-172.68 (12)	N3—C8—C9—C10	178.00 (13)
C11—C3—C4—N1	6.93 (17)	N4—C8—C9—C12	-179.84 (11)
O2—N1—C4—C3	-104.70 (18)	N3—C8—C9—C12	-1.30 (19)
O1—N1—C4—C3	74.84 (19)	C8—C9—C10—C11	0.6 (2)
O2—N1—C4—C5	78.73 (19)	C12—C9—C10—C11	179.93 (11)
O1—N1—C4—C5	-101.72 (17)	C9—C10—C11—C12	-0.2 (2)
C8—N3—C5—C6	-43.9 (2)	C9—C10—C11—C13	-176.82 (14)
C8—N3—C5—C4	135.41 (14)	C8—N4—C12—C11	0.5 (2)
C3—C4—C5—N3	179.48 (12)	C10—C11—C12—N4	-0.4 (2)
N1—C4—C5—N3	-4.14 (18)	C13—C11—C12—N4	176.21 (14)
C3—C4—C5—C6	-1.2 (2)	C12—C11—C13—F6	151.47 (15)
N1—C4—C5—C6	175.22 (12)	C10—C11—C13—F6	-31.9 (2)
N3—C5—C6—C7	176.86 (13)	C12—C11—C13—F4	30.3 (2)
C4—C5—C6—C7	-2.45 (19)	C10—C11—C13—F4	-153.10 (14)
N3—C5—C6—N2	-7.1 (2)	C12—C11—C13—F5	-88.78 (18)
C4—C5—C6—N2	173.63 (12)	C10—C11—C13—F5	87.80 (19)
O4—N2—C6—C7	135.34 (13)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N3—H3...F3 ⁱ	0.88	2.52	3.0690 (15)	121

Symmetry code: (i) $-x+2, -y+1, -z+1$.