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Ethyl 2-amino-6-(4-bromophenyl)-4-(4-fluorophenyl)cyclohexa-1,3-diene-1-carboxylate

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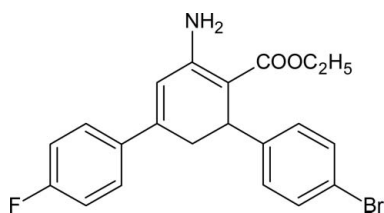
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 Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; R factor = 0.051; wR factor = 0.148; data-to-parameter ratio = 15.3.

In the title compound, $\text{C}_{21}\text{H}_{19}\text{BrFNO}_2$, two independent molecules crystallize in the asymmetric unit. The cyclohexa-1,3-diene ring is in a slightly distorted screw-boat conformation. The dihedral angles between the mean planes of the 4-bromophenyl and 4-fluorophenyl rings are 81.0 (3) and 76.4 (2)° in the two independent molecules. In the crystal, $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into [100] chains.

Related literature

For cyclohexenones as precursors for functionalized derivatives, see: Samshuddin *et al.* (2013); For 4-bromo-4'-fluorochalcone derivatives, see: Fun *et al.* (2012*a,b,c*). For related structures, see: Jasinski *et al.* (2012); Kant *et al.* (2012). For puckering parameters, see: Cremer & Pople (1975). For standard bond lengths, see Allen *et al.* (1987).



Experimental

Crystal data

 $\text{C}_{21}\text{H}_{19}\text{BrFNO}_2$
 $M_r = 416.28$

 Monoclinic, $P2_1/n$
 $a = 9.4599$ (2) Å

 $b = 23.3634$ (5) Å

 $c = 17.2312$ (4) Å

 $\beta = 96.001$ (2)°

 $V = 3787.51$ (15) Å³
 $Z = 8$

 Cu $K\alpha$ radiation

 $\mu = 3.16$ mm⁻¹
 $T = 173$ K

 $0.32 \times 0.28 \times 0.22$ mm

Data collection

Agilent Xcalibur (Eos, Gemini) diffractometer

 Absorption correction: multi-scan (*CrysAlis PRO* and *CrysAlis RED*; Agilent, 2012)

 $T_{\min} = 0.587$, $T_{\max} = 1.000$

25170 measured reflections

7427 independent reflections

 5973 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.040$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.051$
 $wR(F^2) = 0.148$
 $S = 1.03$

7427 reflections

487 parameters

H atoms treated by a mixture of independent and constrained refinement

 $\Delta\rho_{\text{max}} = 1.13$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.53$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1A}-\text{H1AA}\cdots\text{O2A}$	0.77 (3)	2.09 (3)	2.698 (4)	136 (3)
$\text{N1A}-\text{H1AB}\cdots\text{O2B}^i$	0.77 (4)	2.17 (4)	2.915 (4)	162 (4)
$\text{N1B}-\text{H1BA}\cdots\text{O2A}^{ii}$	0.87 (3)	2.20 (4)	3.044 (3)	163 (3)
$\text{N1B}-\text{H1BB}\cdots\text{O2B}$	0.85 (4)	2.07 (4)	2.714 (3)	132 (3)

 Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{1}{2}, z - \frac{1}{2}$; (ii) $x - \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$.

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis RED* (Agilent, 2012); program(s) used to solve structure: *SUPERFLIP* (Palatinus & Chapuis, 2007); program(s) used to refine structure: *SHELXL2012* (Sheldrick, 2008); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *OLEX2* (Dolomanov *et al.*, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG5341).

References

- Agilent (2012). *CrysAlis PRO* and *CrysAlis RED*. Agilent Technologies, Yarnton, England.
- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Dolomanov, O. V., Bourhis, L. J., Gildea, R. J., Howard, J. A. K. & Puschmann, H. (2009). *J. Appl. Cryst.* **42**, 339–341.
- Fun, H.-K., Chia, T. S., Sapnakumari, M., Narayana, B. & Sarojini, B. K. (2012*a*). *Acta Cryst. E* **68**, o2680.
- Fun, H.-K., Loh, W.-S., Sapnakumari, M., Narayana, B. & Sarojini, B. K. (2012*b*). *Acta Cryst. E* **68**, o2655–o2656.
- Fun, H.-K., Loh, W.-S., Sapnakumari, M., Narayana, B. & Sarojini, B. K. (2012*c*). *Acta Cryst. E* **68**, o2586.
- Jasinski, J. P., Golen, J. A., Samshuddin, S., Narayana, B. & Yathirajan, H. S. (2012). *Acta Cryst. E* **68**, o585.
- Kant, R., Gupta, V. K., Kapoor, K., Sapnakumari, M., Narayana, B. & Sarojini, B. K. (2012). *Acta Cryst. E* **68**, o2917–o2918.
- Palatinus, L. & Chapuis, G. (2007). *J. Appl. Cryst.* **40**, 786–790.
- Samshuddin, S., Narayana, B., Sarojini, B. K. & Madhu, L. N. (2013). *Med. Chem. Res.* **22**, 3002–3011.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.

supporting information

Acta Cryst. (2013). E69, o1473 [doi:10.1107/S1600536813023325]

Ethyl 2-amino-6-(4-bromophenyl)-4-(4-fluorophenyl)cyclohexa-1,3-diene-1-carboxylate

B. Narayana, M. Sapnakumari, Jerry P. Jasinski, Peter M. Fraiser and H. S. Yathirajan

S1. Comment

Cyclohexenones are well known precursors for various functionalized derivatives such as indazoles, dibenzodiazepines, terphenyls, aminated derivatives etc. (Samshuddin *et al.*, 2013). The title compound, (I), was prepared by the amination of cyclohexenone derivative of 4-bromo-4'-fluorochalcone with ammonium acetate. The crystal structure of a similar aminated product viz., ethyl 2-amino-4,6-bis(4-fluorophenyl)cyclohexa-1,3-diene-1-carboxylate has been reported (Jasinski *et al.*, 2012). Also, the crystal structure of precursor of the title compound (I) viz., ethyl 6-(4-bromophenyl)-4-(4-fluorophenyl)-2-oxocyclohex-3-ene-1-carboxylate has been reported (Kant *et al.*, 2012). In continuation of our work on the synthesis of 4-bromo-4'-fluorochalcone derivatives (Fun *et al.* 2012a, 2012b, 2012c), the title compound (I), C₂₁H₁₉NO₂FBr, was prepared and its crystal structure is reported.

In the title compound, C₂₁H₁₉NO₂FBr two independent molecules [A & B] crystallize in the asymmetric unit (Fig. 1). The cyclohexa-1,3-diene ring is in a slightly distorted screw-boat conformation (puckering parameters Q, θ , and φ = 0.434 (3) Å, 64.7 (4)°, 271.3 (4)° [A] and 0.355 (3) Å, 63.3 (5)°, 269.7 (6)° [B], respectively; (Cremer & Pople, 1975). Bond lengths are in normal ranges (Allen *et al.*, 1987)). The dihedral angle between the mean planes of the 4-bromophenyl and 4-fluorophenyl rings is 81.0 (3)° [A] and 76.4 (2)° [B]. In the crystal, N—H...O hydrogen bonds form chains along [010] and contribute to packing stability (Fig. 2).

S2. Experimental

A mixture of ethyl 6-(4-bromophenyl)-4-(4-fluorophenyl)-2-oxocyclohex-3-ene-1-carboxylate (4.17 g, 0.01 mol) and ammonium acetate (1.54 g, 0.02 mol) in 30 ml glacial acetic acid was refluxed for 6 hours. The reaction mixture was cooled and poured into 50 ml ice-cold water. The precipitate was collected by filtration and purified by recrystallization from ethanol. Single crystals were grown from ethanol by the slow evaporation method. (m.p. 418–420 K).

S3. Refinement

H1AA, H1AB, H1BA and H1BB were located by a difference map and refined isotropically. All of the remaining H atoms were placed in their calculated positions and then refined using the riding model with Atom—H lengths of 0.95 Å (CH) or 0.99° (CH₂) or 0.98° (CH₃). Isotropic displacement parameters for these atoms were set to 1.2 (CH) or 1.5 (OH) times U_{eq} of the parent atom. Idealised Me refined as rotating group: C1A(H1AC,H1AD,H1AE), C1B(H1BC,H1BD,H1BE).

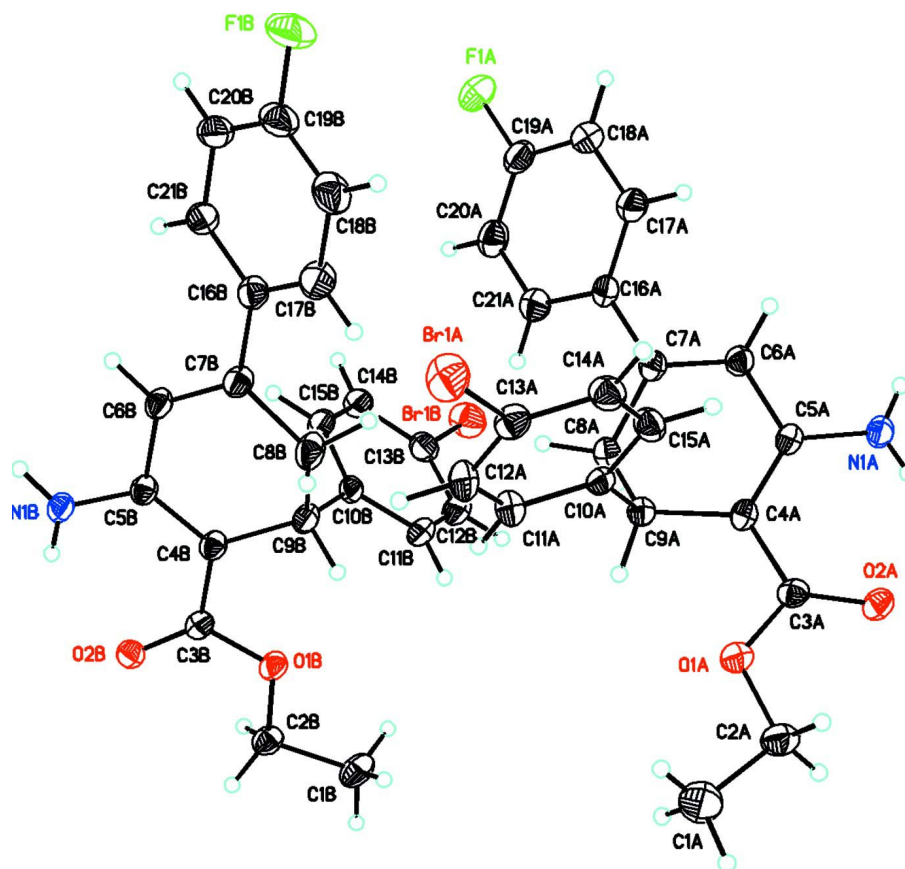
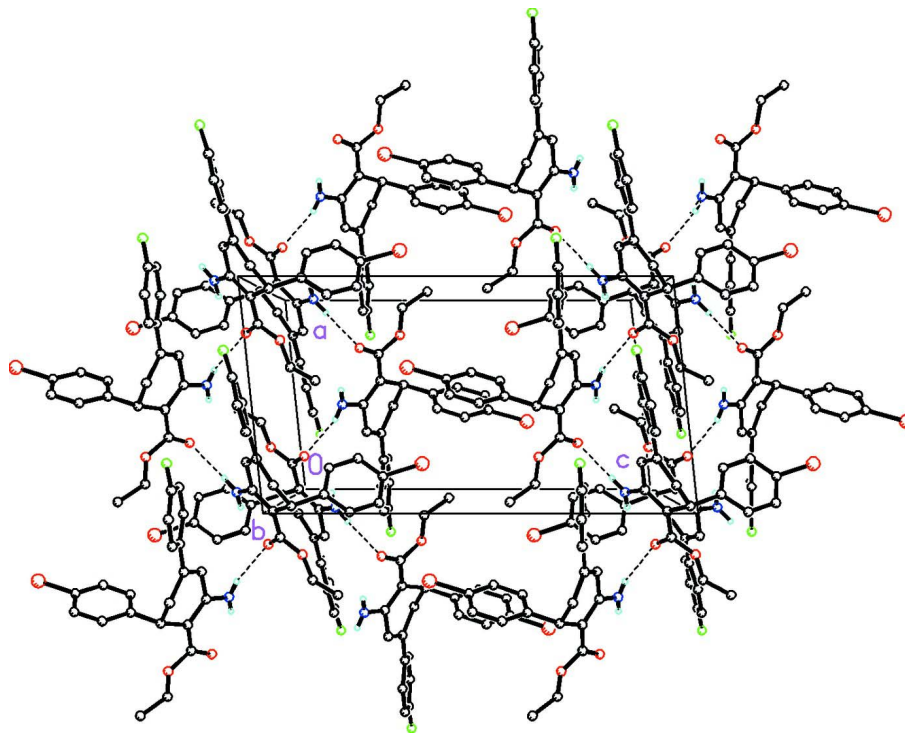
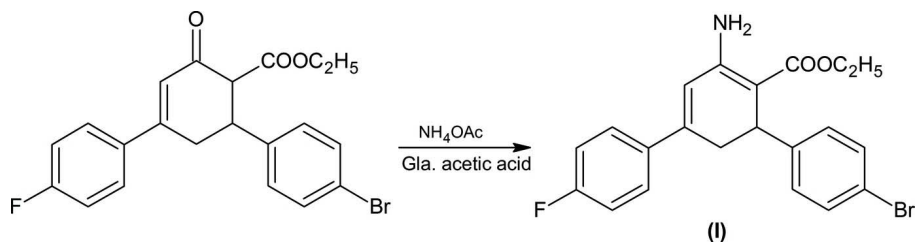


Figure 1

Molecular structure of the title compound showing the atom labeling scheme and 30% probability displacement ellipsoids.

**Figure 2**

Packing diagram of the title compound viewed along the *b* axis. Dashed lines indicate weak N—H···O hydrogen bonds which form chains along [100]. H atoms not involved in these hydrogen bonds have been deleted for clarity.

**Figure 3**

Synthesis of (I).

Ethyl 2-amino-6-(4-bromophenyl)-4-(4-fluorophenyl)cyclohexa-1,3-diene-1-carboxylate

Crystal data

$C_{21}H_{19}BrFNO_2$
 $M_r = 416.28$
 Monoclinic, $P2_1/n$
 $a = 9.4599$ (2) Å
 $b = 23.3634$ (5) Å
 $c = 17.2312$ (4) Å
 $\beta = 96.001$ (2)°
 $V = 3787.51$ (15) Å³
 $Z = 8$

$F(000) = 1696$
 $D_x = 1.460$ Mg m⁻³
 Cu $K\alpha$ radiation, $\lambda = 1.5418$ Å
 Cell parameters from 7321 reflections
 $\theta = 3.2$ – 72.3 °
 $\mu = 3.16$ mm⁻¹
 $T = 173$ K
 Irregular, yellow
 $0.32 \times 0.28 \times 0.22$ mm

Data collection

Agilent Xcalibur (Eos, Gemini)
diffractometer

Radiation source: Enhance (Cu) X-ray Source
Graphite monochromator

Detector resolution: 16.0416 pixels mm⁻¹

ω scans

Absorption correction: multi-scan
(*CrysAlis PRO* and *CrysAlis RED*; Agilent,
2012)

$T_{\min} = 0.587$, $T_{\max} = 1.000$

25170 measured reflections

7427 independent reflections

5973 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.040$

$\theta_{\max} = 72.5^\circ$, $\theta_{\min} = 3.2^\circ$

$h = -11 \rightarrow 10$

$k = -28 \rightarrow 25$

$l = -21 \rightarrow 20$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.051$

$wR(F^2) = 0.148$

$S = 1.03$

7427 reflections

487 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Hydrogen site location: mixed

H atoms treated by a mixture of independent
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0771P)^2 + 2.9658P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 1.13 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.53 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Br1A	0.37157 (5)	0.43234 (2)	0.64815 (2)	0.05829 (15)
F1A	-0.1959 (2)	0.20568 (10)	0.24677 (16)	0.0692 (7)
O1A	0.7906 (2)	0.41473 (10)	0.33655 (13)	0.0414 (5)
O2A	0.7261 (2)	0.46881 (9)	0.23026 (12)	0.0364 (5)
N1A	0.4536 (3)	0.45651 (12)	0.16870 (16)	0.0377 (6)
H1AA	0.523 (4)	0.4738 (13)	0.1701 (18)	0.022 (8)*
H1AB	0.382 (4)	0.4657 (16)	0.147 (2)	0.039 (10)*
C1A	1.0010 (5)	0.4226 (2)	0.4216 (3)	0.0819 (16)
H1AC	1.0218	0.3825	0.4110	0.123*
H1AD	0.9448	0.4249	0.4662	0.123*
H1AE	1.0903	0.4436	0.4337	0.123*
C2A	0.9191 (4)	0.44839 (17)	0.3518 (2)	0.0478 (8)
H2AA	0.8955	0.4888	0.3620	0.057*
H2AB	0.9752	0.4471	0.3064	0.057*
C3A	0.6981 (3)	0.43038 (12)	0.27564 (17)	0.0317 (6)
C4A	0.5687 (3)	0.39797 (12)	0.27159 (16)	0.0315 (6)
C5A	0.4532 (3)	0.41309 (12)	0.22021 (16)	0.0315 (6)
C6A	0.3182 (3)	0.38225 (13)	0.22013 (17)	0.0342 (6)
H6A	0.2340	0.3988	0.1947	0.041*
C7A	0.3106 (3)	0.33092 (12)	0.25538 (16)	0.0325 (6)

C8A	0.4461 (3)	0.30516 (12)	0.29415 (17)	0.0345 (6)
H8AA	0.4924	0.2827	0.2552	0.041*
H8AB	0.4227	0.2786	0.3357	0.041*
C9A	0.5506 (3)	0.35068 (12)	0.32998 (16)	0.0320 (6)
H9A	0.6451	0.3316	0.3416	0.038*
C10A	0.5058 (3)	0.37256 (12)	0.40751 (16)	0.0303 (6)
C11A	0.5381 (4)	0.33980 (13)	0.47408 (18)	0.0385 (7)
H11A	0.5885	0.3049	0.4704	0.046*
C12A	0.4988 (4)	0.35662 (14)	0.54587 (18)	0.0436 (7)
H12A	0.5218	0.3337	0.5910	0.052*
C13A	0.4255 (3)	0.40728 (14)	0.55037 (18)	0.0388 (7)
C14A	0.3931 (4)	0.44123 (15)	0.4856 (2)	0.0418 (7)
H14A	0.3431	0.4762	0.4897	0.050*
C15A	0.4342 (3)	0.42383 (14)	0.41475 (18)	0.0391 (7)
H15A	0.4130	0.4474	0.3701	0.047*
C16A	0.1765 (3)	0.29819 (13)	0.25295 (17)	0.0348 (6)
C17A	0.0446 (4)	0.32440 (14)	0.2399 (2)	0.0452 (8)
H17A	0.0400	0.3646	0.2319	0.054*
C18A	-0.0808 (4)	0.29345 (16)	0.2383 (2)	0.0500 (8)
H18A	-0.1703	0.3122	0.2308	0.060*
C19A	-0.0729 (4)	0.23585 (16)	0.2475 (2)	0.0472 (8)
C20A	0.0542 (4)	0.20668 (15)	0.2570 (2)	0.0518 (9)
H20A	0.0565	0.1661	0.2612	0.062*
C21A	0.1786 (4)	0.23819 (14)	0.2602 (2)	0.0443 (7)
H21A	0.2674	0.2189	0.2674	0.053*
Br1B	0.32238 (5)	0.06632 (2)	0.13555 (2)	0.06171 (15)
F1B	-0.1976 (3)	0.29436 (12)	0.58030 (17)	0.0815 (8)
O1B	0.7450 (2)	0.08834 (9)	0.46357 (12)	0.0365 (5)
O2B	0.7067 (2)	0.02371 (9)	0.55597 (12)	0.0341 (4)
N1B	0.4602 (3)	0.03596 (11)	0.62187 (15)	0.0357 (5)
H1BA	0.394 (4)	0.0269 (14)	0.6509 (19)	0.030 (8)*
H1BB	0.532 (4)	0.0150 (15)	0.618 (2)	0.036 (9)*
C1B	0.9329 (4)	0.09062 (15)	0.3842 (2)	0.0476 (8)
H1BC	0.8671	0.0900	0.3364	0.071*
H1BD	1.0235	0.0733	0.3740	0.071*
H1BE	0.9490	0.1303	0.4014	0.071*
C2B	0.8696 (3)	0.05690 (13)	0.4476 (2)	0.0387 (7)
H2BA	0.8437	0.0177	0.4292	0.046*
H2BB	0.9381	0.0542	0.4950	0.046*
C3B	0.6680 (3)	0.06649 (12)	0.51815 (17)	0.0311 (6)
C4B	0.5413 (3)	0.09959 (12)	0.52506 (17)	0.0322 (6)
C5B	0.4440 (3)	0.08185 (12)	0.57450 (17)	0.0315 (6)
C6B	0.3117 (3)	0.11419 (12)	0.57836 (17)	0.0353 (6)
H6B	0.2381	0.0980	0.6048	0.042*
C7B	0.2919 (3)	0.16682 (12)	0.54492 (18)	0.0367 (6)
C8B	0.4108 (4)	0.19339 (14)	0.5065 (2)	0.0478 (8)
H8BA	0.4680	0.2174	0.5454	0.057*
H8BB	0.3694	0.2190	0.4643	0.057*

C9B	0.5104 (3)	0.15061 (12)	0.47171 (18)	0.0344 (6)
H9B	0.6029	0.1709	0.4692	0.041*
C10B	0.4599 (3)	0.13196 (12)	0.38868 (18)	0.0338 (6)
C11B	0.5379 (3)	0.14645 (15)	0.32743 (19)	0.0439 (8)
H11B	0.6200	0.1698	0.3375	0.053*
C12B	0.4985 (4)	0.12771 (17)	0.2523 (2)	0.0492 (8)
H12B	0.5529	0.1379	0.2110	0.059*
C13B	0.3789 (4)	0.09399 (14)	0.23795 (19)	0.0425 (7)
C14B	0.2974 (4)	0.07961 (14)	0.2971 (2)	0.0427 (7)
H14B	0.2139	0.0571	0.2864	0.051*
C15B	0.3393 (3)	0.09842 (13)	0.3717 (2)	0.0408 (7)
H15B	0.2843	0.0882	0.4126	0.049*
C16B	0.1623 (3)	0.20068 (13)	0.55263 (18)	0.0383 (7)
C17B	0.1648 (4)	0.26074 (15)	0.5510 (2)	0.0506 (9)
H17B	0.2508	0.2799	0.5432	0.061*
C18B	0.0451 (5)	0.29259 (17)	0.5605 (2)	0.0583 (10)
H18B	0.0479	0.3332	0.5602	0.070*
C19B	-0.0789 (4)	0.26373 (18)	0.5706 (2)	0.0545 (9)
C20B	-0.0863 (4)	0.20618 (17)	0.5734 (2)	0.0496 (8)
H20B	-0.1734	0.1877	0.5806	0.060*
C21B	0.0355 (3)	0.17440 (14)	0.56566 (19)	0.0412 (7)
H21B	0.0319	0.1339	0.5693	0.049*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Br1A	0.0728 (3)	0.0676 (3)	0.0371 (2)	0.0088 (2)	0.01827 (18)	-0.00322 (16)
F1A	0.0491 (12)	0.0690 (15)	0.0906 (18)	-0.0199 (11)	0.0129 (12)	0.0164 (13)
O1A	0.0323 (11)	0.0490 (12)	0.0426 (12)	-0.0040 (9)	0.0018 (9)	0.0067 (10)
O2A	0.0354 (11)	0.0427 (11)	0.0327 (10)	-0.0033 (9)	0.0117 (8)	0.0009 (9)
N1A	0.0364 (15)	0.0392 (14)	0.0367 (14)	-0.0076 (13)	-0.0002 (12)	0.0096 (11)
C1A	0.062 (3)	0.101 (4)	0.076 (3)	-0.025 (3)	-0.023 (2)	0.025 (3)
C2A	0.0348 (17)	0.057 (2)	0.050 (2)	-0.0069 (15)	-0.0012 (14)	0.0017 (16)
C3A	0.0294 (14)	0.0369 (15)	0.0303 (14)	0.0033 (11)	0.0090 (11)	-0.0031 (11)
C4A	0.0355 (15)	0.0318 (14)	0.0280 (13)	-0.0030 (12)	0.0068 (11)	0.0012 (11)
C5A	0.0356 (15)	0.0307 (14)	0.0290 (14)	-0.0035 (11)	0.0076 (11)	0.0007 (11)
C6A	0.0345 (15)	0.0361 (15)	0.0316 (14)	-0.0031 (12)	0.0016 (11)	0.0028 (11)
C7A	0.0360 (15)	0.0329 (14)	0.0287 (14)	-0.0028 (12)	0.0044 (11)	0.0001 (11)
C8A	0.0404 (16)	0.0305 (14)	0.0320 (14)	-0.0013 (12)	0.0015 (12)	0.0031 (11)
C9A	0.0315 (14)	0.0330 (14)	0.0313 (14)	-0.0011 (11)	0.0027 (11)	0.0031 (11)
C10A	0.0251 (13)	0.0332 (14)	0.0325 (14)	-0.0048 (11)	0.0025 (11)	0.0024 (11)
C11A	0.0459 (17)	0.0357 (15)	0.0339 (15)	0.0017 (13)	0.0038 (13)	0.0040 (12)
C12A	0.056 (2)	0.0441 (17)	0.0304 (15)	0.0007 (15)	0.0056 (14)	0.0080 (13)
C13A	0.0393 (16)	0.0478 (17)	0.0303 (15)	-0.0025 (14)	0.0080 (12)	-0.0032 (13)
C14A	0.0397 (17)	0.0468 (18)	0.0393 (17)	0.0067 (14)	0.0061 (13)	-0.0005 (13)
C15A	0.0404 (17)	0.0451 (17)	0.0315 (15)	0.0060 (13)	0.0018 (12)	0.0059 (12)
C16A	0.0393 (16)	0.0341 (15)	0.0314 (14)	-0.0049 (12)	0.0050 (12)	0.0039 (11)
C17A	0.0445 (18)	0.0381 (16)	0.054 (2)	-0.0030 (14)	0.0088 (15)	0.0052 (14)

C18A	0.0397 (18)	0.056 (2)	0.056 (2)	0.0000 (15)	0.0092 (15)	0.0112 (17)
C19A	0.0426 (18)	0.053 (2)	0.0471 (19)	-0.0129 (15)	0.0082 (15)	0.0070 (15)
C20A	0.058 (2)	0.0392 (18)	0.059 (2)	-0.0102 (16)	0.0086 (17)	0.0055 (15)
C21A	0.0453 (18)	0.0407 (17)	0.0469 (18)	-0.0027 (14)	0.0053 (14)	0.0029 (14)
Br1B	0.0622 (3)	0.0741 (3)	0.0478 (2)	0.0055 (2)	0.00066 (19)	-0.00554 (18)
F1B	0.0622 (15)	0.0926 (19)	0.0917 (19)	0.0427 (14)	0.0173 (14)	0.0110 (15)
O1B	0.0350 (11)	0.0394 (11)	0.0377 (11)	0.0048 (9)	0.0154 (9)	0.0039 (9)
O2B	0.0336 (10)	0.0355 (10)	0.0332 (10)	0.0024 (8)	0.0039 (8)	0.0014 (8)
N1B	0.0362 (13)	0.0371 (13)	0.0358 (13)	0.0046 (11)	0.0140 (11)	0.0068 (10)
C1B	0.0456 (18)	0.0488 (19)	0.052 (2)	-0.0019 (15)	0.0248 (16)	-0.0068 (16)
C2B	0.0348 (15)	0.0390 (16)	0.0440 (17)	0.0022 (13)	0.0127 (13)	-0.0027 (13)
C3B	0.0321 (14)	0.0334 (14)	0.0283 (14)	-0.0037 (11)	0.0057 (11)	-0.0051 (11)
C4B	0.0351 (15)	0.0297 (14)	0.0332 (14)	0.0009 (12)	0.0106 (11)	-0.0002 (11)
C5B	0.0356 (15)	0.0300 (13)	0.0299 (14)	-0.0001 (11)	0.0081 (11)	-0.0007 (11)
C6B	0.0390 (16)	0.0334 (14)	0.0359 (15)	0.0025 (12)	0.0150 (12)	0.0038 (12)
C7B	0.0390 (16)	0.0303 (14)	0.0430 (16)	0.0029 (12)	0.0144 (13)	0.0004 (12)
C8B	0.058 (2)	0.0322 (16)	0.058 (2)	0.0094 (15)	0.0280 (17)	0.0110 (14)
C9B	0.0356 (15)	0.0287 (14)	0.0415 (16)	0.0016 (12)	0.0160 (12)	0.0051 (12)
C10B	0.0326 (14)	0.0303 (14)	0.0403 (16)	0.0085 (11)	0.0120 (12)	0.0120 (12)
C11B	0.0329 (15)	0.057 (2)	0.0435 (18)	-0.0055 (14)	0.0102 (13)	0.0132 (15)
C12B	0.0404 (18)	0.068 (2)	0.0408 (18)	-0.0018 (16)	0.0120 (14)	0.0133 (16)
C13B	0.0422 (17)	0.0464 (17)	0.0391 (17)	0.0081 (14)	0.0045 (13)	0.0049 (14)
C14B	0.0372 (16)	0.0399 (16)	0.0515 (19)	-0.0001 (13)	0.0075 (14)	0.0062 (14)
C15B	0.0378 (16)	0.0385 (16)	0.0482 (18)	0.0005 (13)	0.0148 (14)	0.0095 (13)
C16B	0.0437 (17)	0.0360 (15)	0.0367 (16)	0.0048 (13)	0.0112 (13)	0.0015 (12)
C17B	0.055 (2)	0.0398 (17)	0.059 (2)	0.0078 (16)	0.0197 (17)	0.0024 (15)
C18B	0.071 (3)	0.046 (2)	0.060 (2)	0.0206 (19)	0.015 (2)	0.0042 (17)
C19B	0.051 (2)	0.067 (2)	0.047 (2)	0.0254 (19)	0.0099 (16)	0.0043 (17)
C20B	0.0385 (17)	0.070 (2)	0.0410 (18)	0.0084 (16)	0.0059 (14)	0.0041 (16)
C21B	0.0407 (17)	0.0443 (17)	0.0391 (16)	0.0046 (14)	0.0073 (13)	0.0021 (13)

Geometric parameters (Å, °)

Br1A—C13A	1.903 (3)	Br1B—C13B	1.903 (3)
F1A—C19A	1.359 (4)	F1B—C19B	1.357 (4)
O1A—C2A	1.449 (4)	O1B—C2B	1.439 (4)
O1A—C3A	1.345 (4)	O1B—C3B	1.348 (3)
O2A—C3A	1.237 (4)	O2B—C3B	1.228 (4)
N1A—H1AA	0.77 (3)	N1B—H1BA	0.87 (3)
N1A—H1AB	0.77 (4)	N1B—H1BB	0.85 (4)
N1A—C5A	1.348 (4)	N1B—C5B	1.346 (4)
C1A—H1AC	0.9800	C1B—H1BC	0.9800
C1A—H1AD	0.9800	C1B—H1BD	0.9800
C1A—H1AE	0.9800	C1B—H1BE	0.9800
C1A—C2A	1.489 (6)	C1B—C2B	1.519 (4)
C2A—H2AA	0.9900	C2B—H2BA	0.9900
C2A—H2AB	0.9900	C2B—H2BB	0.9900
C3A—C4A	1.434 (4)	C3B—C4B	1.442 (4)

C4A—C5A	1.379 (4)	C4B—C5B	1.382 (4)
C4A—C9A	1.516 (4)	C4B—C9B	1.515 (4)
C5A—C6A	1.466 (4)	C5B—C6B	1.469 (4)
C6A—H6A	0.9500	C6B—H6B	0.9500
C6A—C7A	1.350 (4)	C6B—C7B	1.363 (4)
C7A—C8A	1.507 (4)	C7B—C8B	1.498 (4)
C7A—C16A	1.478 (4)	C7B—C16B	1.477 (4)
C8A—H8AA	0.9900	C8B—H8BA	0.9900
C8A—H8AB	0.9900	C8B—H8BB	0.9900
C8A—C9A	1.537 (4)	C8B—C9B	1.537 (4)
C9A—H9A	1.0000	C9B—H9B	1.0000
C9A—C10A	1.531 (4)	C9B—C10B	1.524 (4)
C10A—C11A	1.386 (4)	C10B—C11B	1.391 (4)
C10A—C15A	1.388 (4)	C10B—C15B	1.389 (4)
C11A—H11A	0.9500	C11B—H11B	0.9500
C11A—C12A	1.385 (4)	C11B—C12B	1.381 (5)
C12A—H12A	0.9500	C12B—H12B	0.9500
C12A—C13A	1.378 (5)	C12B—C13B	1.379 (5)
C13A—C14A	1.377 (5)	C13B—C14B	1.382 (5)
C14A—H14A	0.9500	C14B—H14B	0.9500
C14A—C15A	1.381 (5)	C14B—C15B	1.377 (5)
C15A—H15A	0.9500	C15B—H15B	0.9500
C16A—C17A	1.388 (5)	C16B—C17B	1.404 (5)
C16A—C21A	1.407 (4)	C16B—C21B	1.387 (5)
C17A—H17A	0.9500	C17B—H17B	0.9500
C17A—C18A	1.387 (5)	C17B—C18B	1.379 (5)
C18A—H18A	0.9500	C18B—H18B	0.9500
C18A—C19A	1.356 (5)	C18B—C19B	1.379 (6)
C19A—C20A	1.378 (5)	C19B—C20B	1.348 (6)
C20A—H20A	0.9500	C20B—H20B	0.9500
C20A—C21A	1.384 (5)	C20B—C21B	1.389 (5)
C21A—H21A	0.9500	C21B—H21B	0.9500
C3A—O1A—C2A	117.3 (2)	C3B—O1B—C2B	116.9 (2)
H1AA—N1A—H1AB	125 (4)	H1BA—N1B—H1BB	122 (3)
C5A—N1A—H1AA	116 (2)	C5B—N1B—H1BA	120 (2)
C5A—N1A—H1AB	118 (3)	C5B—N1B—H1BB	117 (2)
H1AC—C1A—H1AD	109.5	H1BC—C1B—H1BD	109.5
H1AC—C1A—H1AE	109.5	H1BC—C1B—H1BE	109.5
H1AD—C1A—H1AE	109.5	H1BD—C1B—H1BE	109.5
C2A—C1A—H1AC	109.5	C2B—C1B—H1BC	109.5
C2A—C1A—H1AD	109.5	C2B—C1B—H1BD	109.5
C2A—C1A—H1AE	109.5	C2B—C1B—H1BE	109.5
O1A—C2A—C1A	106.4 (3)	O1B—C2B—C1B	105.4 (3)
O1A—C2A—H2AA	110.4	O1B—C2B—H2BA	110.7
O1A—C2A—H2AB	110.4	O1B—C2B—H2BB	110.7
C1A—C2A—H2AA	110.4	C1B—C2B—H2BA	110.7
C1A—C2A—H2AB	110.4	C1B—C2B—H2BB	110.7

H2AA—C2A—H2AB	108.6	H2BA—C2B—H2BB	108.8
O1A—C3A—C4A	112.1 (2)	O1B—C3B—C4B	111.6 (2)
O2A—C3A—O1A	121.7 (3)	O2B—C3B—O1B	121.7 (3)
O2A—C3A—C4A	126.2 (3)	O2B—C3B—C4B	126.7 (3)
C3A—C4A—C9A	120.5 (3)	C3B—C4B—C9B	118.8 (2)
C5A—C4A—C3A	120.7 (3)	C5B—C4B—C3B	120.2 (3)
C5A—C4A—C9A	118.4 (3)	C5B—C4B—C9B	120.7 (3)
N1A—C5A—C4A	124.0 (3)	N1B—C5B—C4B	124.9 (3)
N1A—C5A—C6A	115.6 (3)	N1B—C5B—C6B	115.2 (3)
C4A—C5A—C6A	120.4 (3)	C4B—C5B—C6B	119.9 (3)
C5A—C6A—H6A	119.2	C5B—C6B—H6B	119.2
C7A—C6A—C5A	121.5 (3)	C7B—C6B—C5B	121.7 (3)
C7A—C6A—H6A	119.2	C7B—C6B—H6B	119.2
C6A—C7A—C8A	118.2 (3)	C6B—C7B—C8B	119.0 (3)
C6A—C7A—C16A	122.2 (3)	C6B—C7B—C16B	121.6 (3)
C16A—C7A—C8A	119.6 (3)	C16B—C7B—C8B	119.1 (3)
C7A—C8A—H8AA	109.1	C7B—C8B—H8BA	108.5
C7A—C8A—H8AB	109.1	C7B—C8B—H8BB	108.5
C7A—C8A—C9A	112.5 (2)	C7B—C8B—C9B	115.0 (3)
H8AA—C8A—H8AB	107.8	H8BA—C8B—H8BB	107.5
C9A—C8A—H8AA	109.1	C9B—C8B—H8BA	108.5
C9A—C8A—H8AB	109.1	C9B—C8B—H8BB	108.5
C4A—C9A—C8A	110.5 (2)	C4B—C9B—C8B	111.3 (2)
C4A—C9A—H9A	107.3	C4B—C9B—H9B	106.5
C4A—C9A—C10A	113.4 (2)	C4B—C9B—C10B	111.5 (2)
C8A—C9A—H9A	107.3	C8B—C9B—H9B	106.5
C10A—C9A—C8A	110.9 (2)	C10B—C9B—C8B	114.0 (3)
C10A—C9A—H9A	107.3	C10B—C9B—H9B	106.5
C11A—C10A—C9A	118.7 (3)	C11B—C10B—C9B	120.2 (3)
C11A—C10A—C15A	118.0 (3)	C15B—C10B—C9B	121.9 (3)
C15A—C10A—C9A	123.3 (3)	C15B—C10B—C11B	117.8 (3)
C10A—C11A—H11A	119.1	C10B—C11B—H11B	119.3
C12A—C11A—C10A	121.7 (3)	C12B—C11B—C10B	121.5 (3)
C12A—C11A—H11A	119.1	C12B—C11B—H11B	119.3
C11A—C12A—H12A	120.7	C11B—C12B—H12B	120.5
C13A—C12A—C11A	118.5 (3)	C13B—C12B—C11B	119.0 (3)
C13A—C12A—H12A	120.7	C13B—C12B—H12B	120.5
C12A—C13A—Br1A	119.8 (2)	C12B—C13B—Br1B	120.4 (3)
C14A—C13A—Br1A	118.8 (2)	C12B—C13B—C14B	121.2 (3)
C14A—C13A—C12A	121.3 (3)	C14B—C13B—Br1B	118.5 (3)
C13A—C14A—H14A	120.4	C13B—C14B—H14B	120.6
C13A—C14A—C15A	119.2 (3)	C15B—C14B—C13B	118.8 (3)
C15A—C14A—H14A	120.4	C15B—C14B—H14B	120.6
C10A—C15A—H15A	119.4	C10B—C15B—H15B	119.1
C14A—C15A—C10A	121.2 (3)	C14B—C15B—C10B	121.8 (3)
C14A—C15A—H15A	119.4	C14B—C15B—H15B	119.1
C17A—C16A—C7A	122.1 (3)	C17B—C16B—C7B	121.2 (3)
C17A—C16A—C21A	117.3 (3)	C21B—C16B—C7B	121.2 (3)

C21A—C16A—C7A	120.6 (3)	C21B—C16B—C17B	117.5 (3)
C16A—C17A—H17A	119.1	C16B—C17B—H17B	119.3
C18A—C17A—C16A	121.8 (3)	C18B—C17B—C16B	121.4 (4)
C18A—C17A—H17A	119.1	C18B—C17B—H17B	119.3
C17A—C18A—H18A	120.7	C17B—C18B—H18B	121.0
C19A—C18A—C17A	118.6 (3)	C17B—C18B—C19B	118.1 (4)
C19A—C18A—H18A	120.7	C19B—C18B—H18B	121.0
F1A—C19A—C20A	118.7 (3)	F1B—C19B—C18B	118.9 (4)
C18A—C19A—F1A	118.5 (3)	C20B—C19B—F1B	118.3 (4)
C18A—C19A—C20A	122.8 (3)	C20B—C19B—C18B	122.7 (3)
C19A—C20A—H20A	121.0	C19B—C20B—H20B	120.6
C19A—C20A—C21A	118.0 (3)	C19B—C20B—C21B	118.9 (3)
C21A—C20A—H20A	121.0	C21B—C20B—H20B	120.6
C16A—C21A—H21A	119.3	C16B—C21B—C20B	121.3 (3)
C20A—C21A—C16A	121.5 (3)	C16B—C21B—H21B	119.3
C20A—C21A—H21A	119.3	C20B—C21B—H21B	119.3
Br1A—C13A—C14A—C15A	178.8 (3)	Br1B—C13B—C14B—C15B	178.6 (2)
F1A—C19A—C20A—C21A	-177.9 (3)	F1B—C19B—C20B—C21B	-178.3 (3)
O1A—C3A—C4A—C5A	170.7 (3)	O1B—C3B—C4B—C5B	175.9 (3)
O1A—C3A—C4A—C9A	-2.3 (4)	O1B—C3B—C4B—C9B	1.7 (4)
O2A—C3A—C4A—C5A	-8.9 (5)	O2B—C3B—C4B—C5B	-4.6 (5)
O2A—C3A—C4A—C9A	178.1 (3)	O2B—C3B—C4B—C9B	-178.8 (3)
N1A—C5A—C6A—C7A	165.1 (3)	N1B—C5B—C6B—C7B	168.0 (3)
C2A—O1A—C3A—O2A	6.0 (4)	C2B—O1B—C3B—O2B	4.4 (4)
C2A—O1A—C3A—C4A	-173.7 (3)	C2B—O1B—C3B—C4B	-176.1 (2)
C3A—O1A—C2A—C1A	179.5 (3)	C3B—O1B—C2B—C1B	179.4 (3)
C3A—C4A—C5A—N1A	3.0 (5)	C3B—C4B—C5B—N1B	3.5 (5)
C3A—C4A—C5A—C6A	-175.7 (3)	C3B—C4B—C5B—C6B	-176.9 (3)
C3A—C4A—C9A—C8A	-152.6 (3)	C3B—C4B—C9B—C8B	-157.6 (3)
C3A—C4A—C9A—C10A	82.2 (3)	C3B—C4B—C9B—C10B	73.9 (3)
C4A—C5A—C6A—C7A	-16.0 (4)	C4B—C5B—C6B—C7B	-11.6 (5)
C4A—C9A—C10A—C11A	-155.2 (3)	C4B—C9B—C10B—C11B	-118.5 (3)
C4A—C9A—C10A—C15A	24.8 (4)	C4B—C9B—C10B—C15B	58.9 (4)
C5A—C4A—C9A—C8A	34.3 (4)	C5B—C4B—C9B—C8B	28.3 (4)
C5A—C4A—C9A—C10A	-90.9 (3)	C5B—C4B—C9B—C10B	-100.2 (3)
C5A—C6A—C7A—C8A	-1.0 (4)	C5B—C6B—C7B—C8B	-2.8 (5)
C5A—C6A—C7A—C16A	-177.4 (3)	C5B—C6B—C7B—C16B	-176.5 (3)
C6A—C7A—C8A—C9A	33.8 (4)	C6B—C7B—C8B—C9B	29.7 (5)
C6A—C7A—C16A—C17A	-23.2 (5)	C6B—C7B—C16B—C17B	151.0 (3)
C6A—C7A—C16A—C21A	153.7 (3)	C6B—C7B—C16B—C21B	-25.7 (5)
C7A—C8A—C9A—C4A	-48.8 (3)	C7B—C8B—C9B—C4B	-40.7 (4)
C7A—C8A—C9A—C10A	77.8 (3)	C7B—C8B—C9B—C10B	86.4 (4)
C7A—C16A—C17A—C18A	-179.5 (3)	C7B—C16B—C17B—C18B	-178.0 (3)
C7A—C16A—C21A—C20A	-179.1 (3)	C7B—C16B—C21B—C20B	179.6 (3)
C8A—C7A—C16A—C17A	160.5 (3)	C8B—C7B—C16B—C17B	-22.7 (5)
C8A—C7A—C16A—C21A	-22.6 (4)	C8B—C7B—C16B—C21B	160.6 (3)
C8A—C9A—C10A—C11A	79.9 (3)	C8B—C9B—C10B—C11B	114.5 (3)

C8A—C9A—C10A—C15A	-100.2 (3)	C8B—C9B—C10B—C15B	-68.1 (4)
C9A—C4A—C5A—N1A	176.1 (3)	C9B—C4B—C5B—N1B	177.6 (3)
C9A—C4A—C5A—C6A	-2.7 (4)	C9B—C4B—C5B—C6B	-2.9 (4)
C9A—C10A—C11A—C12A	-179.0 (3)	C9B—C10B—C11B—C12B	176.7 (3)
C9A—C10A—C15A—C14A	178.5 (3)	C9B—C10B—C15B—C14B	-177.2 (3)
C10A—C11A—C12A—C13A	0.1 (5)	C10B—C11B—C12B—C13B	0.1 (5)
C11A—C10A—C15A—C14A	-1.5 (5)	C11B—C10B—C15B—C14B	0.3 (5)
C11A—C12A—C13A—Br1A	-179.3 (3)	C11B—C12B—C13B—Br1B	-179.1 (3)
C11A—C12A—C13A—C14A	-0.9 (5)	C11B—C12B—C13B—C14B	1.0 (5)
C12A—C13A—C14A—C15A	0.5 (5)	C12B—C13B—C14B—C15B	-1.5 (5)
C13A—C14A—C15A—C10A	0.8 (5)	C13B—C14B—C15B—C10B	0.8 (5)
C15A—C10A—C11A—C12A	1.0 (5)	C15B—C10B—C11B—C12B	-0.8 (5)
C16A—C7A—C8A—C9A	-149.8 (3)	C16B—C7B—C8B—C9B	-156.5 (3)
C16A—C17A—C18A—C19A	-1.9 (6)	C16B—C17B—C18B—C19B	-0.9 (6)
C17A—C16A—C21A—C20A	-2.0 (5)	C17B—C16B—C21B—C20B	2.8 (5)
C17A—C18A—C19A—F1A	179.2 (3)	C17B—C18B—C19B—F1B	179.8 (4)
C17A—C18A—C19A—C20A	-1.2 (6)	C17B—C18B—C19B—C20B	1.7 (6)
C18A—C19A—C20A—C21A	2.6 (6)	C18B—C19B—C20B—C21B	-0.2 (6)
C19A—C20A—C21A—C16A	-0.9 (5)	C19B—C20B—C21B—C16B	-2.1 (5)
C21A—C16A—C17A—C18A	3.5 (5)	C21B—C16B—C17B—C18B	-1.2 (6)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1A—H1AA \cdots O2A	0.77 (3)	2.09 (3)	2.698 (4)	136 (3)
N1A—H1AB \cdots O2B ⁱ	0.77 (4)	2.17 (4)	2.915 (4)	162 (4)
N1B—H1BA \cdots O2A ⁱⁱ	0.87 (3)	2.20 (4)	3.044 (3)	163 (3)
N1B—H1BB \cdots O2B	0.85 (4)	2.07 (4)	2.714 (3)	132 (3)

Symmetry codes: (i) $x-1/2, -y+1/2, z-1/2$; (ii) $x-1/2, -y+1/2, z+1/2$.