

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

Ethyl 5-methyl-3-phenylisoxazole-4-carboxylate

 Chandra,^a K. Raghu,^b S. Jeyaseelan,^c K. B. Umesh^a and M. Mahendra^{a*}
^aDepartment of Studies in Physics, Manasagangothri, University of Mysore, Mysore 570 006, India, ^bDepartment of Chemistry, Yuvaraja's College, University of Mysore, Mysore 570 005, India, and ^cDepartment of Physics, St Philomena's College, Mysore, India

Correspondence e-mail: mahendra@physics.uni-mysore.ac.in

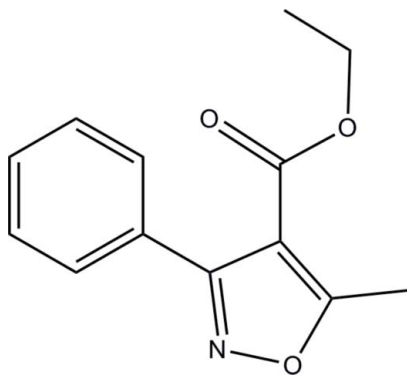
Received 5 April 2013; accepted 23 May 2013

 Key indicators: single-crystal X-ray study; $T = 273$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; R factor = 0.054; wR factor = 0.186; data-to-parameter ratio = 13.2.

In the title compound, $\text{C}_{13}\text{H}_{13}\text{NO}_3$, the dihedral angle between the phenyl and isoxazole rings is $43.40(13)^\circ$. The ethoxy-carbonyl group is rotated out of the plane of the isoxazole ring by $16.2(13)^\circ$.

Related literature

For the biological and pharmacological importance of isoxazoles, see: Lin *et al.* (1997). For the synthesis of isoxazole derivatives and a related structure, see: Chandra *et al.* (2013).



Experimental

Crystal data

$\text{C}_{13}\text{H}_{13}\text{NO}_3$	$V = 1192.3(18) \text{ \AA}^3$
$M_r = 231.24$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 9.750(8) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$b = 14.589(13) \text{ \AA}$	$T = 273 \text{ K}$
$c = 9.397(8) \text{ \AA}$	$0.30 \times 0.25 \times 0.20 \text{ mm}$
$\beta = 116.872(13)^\circ$	

Data collection

Bruker APEXII CCD area-detector diffractometer	2060 independent reflections
10036 measured reflections	1340 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.039$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.054$	156 parameters
$wR(F^2) = 0.186$	H-atom parameters constrained
$S = 0.99$	$\Delta\rho_{\text{max}} = 0.16 \text{ e \AA}^{-3}$
2060 reflections	$\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

Chandra would like to thank the University of Mysore for awarding an RFSMS fellowship under the head DV5/Physics/389/RFSMS/2009–2010/10.07.2012.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NC2309).

References

- Bruker (2009). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Chandra, Raghu, K., Srikantamurthy, N., Umesh, K. B., Palani, K. & Mahendra, M. (2013). *Acta Cryst.* **E69**, o388.
- Lin, S. T., Kuo, S. H. & Yang, F. M. (1997). *J. Org. Chem.* **62**, 5229–5231.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

supporting information

Acta Cryst. (2013). E69, o987 [doi:10.1107/S160053681301427X]

Ethyl 5-methyl-3-phenylisoxazole-4-carboxylate

Chandra, K. Raghu, S. Jeyaseelan, K. B. Umesha and M. Mahendra

S1. Comment

Isoxazole and its derivatives are of well known heterocyclic compounds, which have a variety of biologically activities; such as anti-convulsant, antibacterial, antiasthmatic, and other pharmacological activities (Lin *et al.*, 1997). In view of their importance we have special interest in the synthesis and structural studies of isoxazole derivatives (Chandra *et al.*, 2013). Within this project, the title compound was prepared and characterized by single-crystal X-ray diffraction.

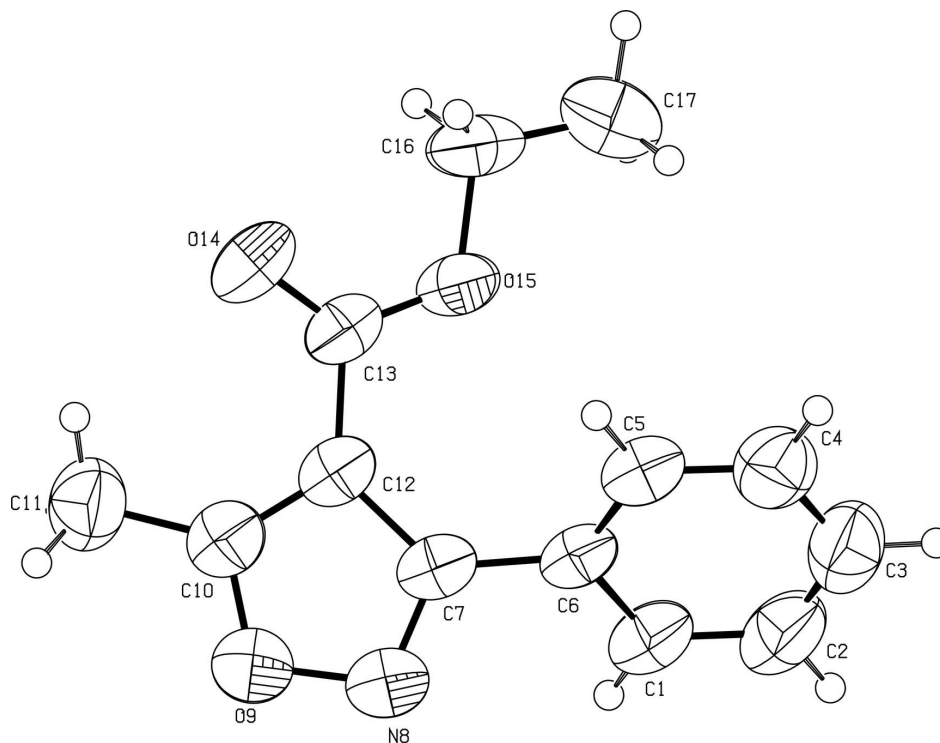
In the molecular structure of the title compound (Fig. 1), the dihedral angle between the phenyl ring (C1/C2/C3/C4/C5/C6) and the isoxazole ring (C7/N8/O9/C10/C12) amount to 43.40 (13)°. The ethoxycarbonyl unit is not in same plane with the isoxazole ring, as indicated by the torsion angle of 16.2 (13)°.

S2. Experimental

A mixture of benzaldehyde oxime (1 g, 8.33 mmol), chloramine-T (2.33 g, 8.33 mmol) and freshly distilled ethyl acetoacetate (2.16 g, 16.6 mmol) in ethyl alcohol (20 ml) were stirred at 10°C about 6 h. The progress of the reaction was monitored by TLC. After the completion of the reaction the solvent was evaporated in vacuum. The solids thus obtained were recrystallized from hot ethanol to get single crystals of the title compound.

S3. Refinement

The H atoms were placed in idealized positions and allowed to ride on their parent atoms with C–H distances in the range of 0.93 to 0.97 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ (1.5 for methyl H atoms).

**Figure 1**

Perspective diagram of the title molecule with labeling and displacement ellipsoids drawn at the 50% probability level.

Ethyl 5-methyl-3-phenylisoxazole-4-carboxylate

Crystal data

$C_{13}H_{13}NO_3$

$M_r = 231.24$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2ybc$

$a = 9.750\ (8)\ \text{\AA}$

$b = 14.589\ (13)\ \text{\AA}$

$c = 9.397\ (8)\ \text{\AA}$

$\beta = 116.872\ (13)^\circ$

$V = 1192.3\ (18)\ \text{\AA}^3$

$Z = 4$

$F(000) = 488$

$D_x = 1.288\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2060 reflections

$\theta = 2.3\text{--}25.0^\circ$

$\mu = 0.09\ \text{mm}^{-1}$

$T = 273\ \text{K}$

Block, yellow

$0.30 \times 0.25 \times 0.20\ \text{mm}$

Data collection

Bruker APEXII CCD area-detector
diffractometer

ω and φ scans

10036 measured reflections

2060 independent reflections

1340 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.039$

$\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 2.3^\circ$

$h = -11 \rightarrow 11$

$k = -17 \rightarrow 17$

$l = -11 \rightarrow 11$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.054$	H-atom parameters constrained
$wR(F^2) = 0.186$	$w = 1/[\sigma^2(F_o^2) + (0.1145P)^2]$
$S = 0.99$	where $P = (F_o^2 + 2F_c^2)/3$
2060 reflections	$(\Delta/\sigma)_{\max} < 0.001$
156 parameters	$\Delta\rho_{\max} = 0.16 \text{ e } \text{\AA}^{-3}$
0 restraints	$\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O9	0.6154 (2)	0.13026 (13)	-0.1802 (3)	0.0780 (8)
O14	0.6825 (2)	0.12137 (19)	0.2907 (3)	0.1153 (12)
O15	0.4317 (2)	0.09643 (11)	0.17563 (19)	0.0667 (6)
N8	0.4554 (3)	0.14345 (15)	-0.2522 (3)	0.0747 (9)
C1	0.1443 (3)	0.10417 (18)	-0.3147 (3)	0.0721 (10)
C2	-0.0093 (3)	0.1141 (2)	-0.3603 (4)	0.0840 (11)
C3	-0.0594 (3)	0.1676 (2)	-0.2736 (4)	0.0861 (11)
C4	0.0469 (3)	0.21272 (19)	-0.1411 (4)	0.0798 (11)
C5	0.2008 (3)	0.20332 (16)	-0.0937 (3)	0.0649 (9)
C6	0.2520 (3)	0.14949 (15)	-0.1799 (3)	0.0571 (8)
C7	0.4161 (3)	0.13941 (14)	-0.1372 (3)	0.0577 (8)
C10	0.6666 (3)	0.11924 (16)	-0.0234 (3)	0.0646 (9)
C11	0.8327 (3)	0.0997 (2)	0.0691 (4)	0.0865 (11)
C12	0.5479 (3)	0.12501 (14)	0.0123 (3)	0.0592 (8)
C13	0.5628 (3)	0.11456 (17)	0.1739 (3)	0.0654 (9)
C16	0.4341 (4)	0.0912 (2)	0.3301 (3)	0.0824 (11)
C17	0.2752 (4)	0.0730 (2)	0.3043 (4)	0.0957 (14)
H1	0.17710	0.06700	-0.37380	0.0870*
H2	-0.08040	0.08420	-0.45110	0.1010*
H3	-0.16400	0.17340	-0.30390	0.1030*
H4	0.01350	0.25020	-0.08290	0.0950*
H5	0.27120	0.23350	-0.00280	0.0780*
H11A	0.85120	0.03600	0.05850	0.1300*
H11B	0.86540	0.11380	0.17950	0.1300*
H11C	0.88940	0.13650	0.02920	0.1300*

H16A	0.47100	0.14850	0.38720	0.0990*
H16B	0.50210	0.04240	0.39260	0.0990*
H17A	0.20880	0.12140	0.24150	0.1430*
H17B	0.27330	0.06990	0.40550	0.1430*
H17C	0.24040	0.01570	0.24930	0.1430*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O9	0.0756 (13)	0.0937 (13)	0.0679 (14)	0.0024 (9)	0.0352 (11)	0.0109 (10)
O14	0.0684 (14)	0.196 (3)	0.0521 (14)	−0.0009 (13)	0.0015 (11)	−0.0013 (13)
O15	0.0746 (12)	0.0772 (11)	0.0415 (10)	−0.0015 (8)	0.0203 (9)	0.0048 (7)
N8	0.0768 (16)	0.0898 (15)	0.0555 (15)	0.0044 (11)	0.0281 (12)	0.0099 (11)
C1	0.0728 (18)	0.0765 (16)	0.0501 (16)	−0.0049 (12)	0.0129 (13)	−0.0015 (12)
C2	0.072 (2)	0.091 (2)	0.0633 (19)	−0.0137 (15)	0.0079 (16)	0.0049 (15)
C3	0.0632 (18)	0.091 (2)	0.092 (2)	0.0039 (15)	0.0245 (18)	0.0233 (18)
C4	0.081 (2)	0.0796 (17)	0.076 (2)	0.0125 (14)	0.0329 (16)	0.0104 (15)
C5	0.0744 (17)	0.0609 (14)	0.0497 (15)	0.0009 (11)	0.0195 (13)	0.0038 (11)
C6	0.0634 (15)	0.0558 (12)	0.0411 (14)	−0.0012 (10)	0.0139 (12)	0.0063 (10)
C7	0.0676 (16)	0.0533 (13)	0.0445 (15)	−0.0043 (10)	0.0186 (12)	0.0011 (10)
C10	0.0661 (16)	0.0592 (13)	0.0621 (18)	−0.0047 (11)	0.0234 (14)	0.0014 (11)
C11	0.0643 (18)	0.096 (2)	0.091 (2)	−0.0043 (14)	0.0279 (16)	0.0033 (16)
C12	0.0619 (15)	0.0555 (12)	0.0494 (15)	−0.0033 (10)	0.0157 (12)	0.0027 (10)
C13	0.0633 (17)	0.0723 (15)	0.0466 (16)	0.0026 (12)	0.0126 (13)	0.0014 (11)
C16	0.110 (2)	0.090 (2)	0.0457 (16)	0.0163 (16)	0.0338 (16)	0.0099 (13)
C17	0.121 (3)	0.098 (2)	0.090 (2)	0.0003 (18)	0.067 (2)	−0.0064 (18)

Geometric parameters (Å, °)

O9—N8	1.405 (4)	C12—C13	1.467 (4)
O9—C10	1.335 (4)	C16—C17	1.480 (6)
O14—C13	1.191 (4)	C1—H1	0.9300
O15—C13	1.313 (4)	C2—H2	0.9300
O15—C16	1.443 (4)	C3—H3	0.9300
N8—C7	1.301 (4)	C4—H4	0.9300
C1—C2	1.367 (5)	C5—H5	0.9300
C1—C6	1.394 (4)	C11—H11A	0.9600
C2—C3	1.368 (5)	C11—H11B	0.9600
C3—C4	1.375 (5)	C11—H11C	0.9600
C4—C5	1.366 (5)	C16—H16A	0.9700
C5—C6	1.374 (4)	C16—H16B	0.9700
C6—C7	1.470 (5)	C17—H17A	0.9600
C7—C12	1.427 (4)	C17—H17B	0.9600
C10—C11	1.479 (5)	C17—H17C	0.9600
C10—C12	1.345 (5)		
O14...C2 ⁱ	3.300 (5)	C5...H17A ⁱⁱⁱ	3.0100
O14...C11	3.056 (5)	C12...H5	3.0800

O15...C5	2.959 (4)	C13...H11B	2.9300
O15...C10 ⁱⁱ	3.410 (4)	C13...H5	3.1000
O15...C6	3.089 (4)	C16...H16B ^{vii}	3.0800
O9...H17C ⁱⁱ	2.7900	H1...N8	2.6700
O14...H16B	2.6200	H2...O14 ^{iv}	2.5400
O14...H11B	2.4400	H4...H11C ^{vi}	2.5500
O14...H16A	2.6300	H4...C1 ^v	3.1000
O14...H2 ⁱ	2.5400	H4...C2 ^v	2.9600
O15...H5	2.6200	H5...O15	2.6200
N8...H1	2.6700	H5...C12	3.0800
N8...H5 ⁱⁱⁱ	2.8600	H5...C13	3.1000
C2...O14 ^{iv}	3.300 (5)	H5...N8 ^v	2.8600
C5...C17 ⁱⁱⁱ	3.567 (5)	H11B...O14	2.4400
C5...C13	3.526 (5)	H11B...C13	2.9300
C5...O15	2.959 (4)	H11C...C4 ^{viii}	2.8900
C6...O15	3.089 (4)	H11C...H4 ^{viii}	2.5500
C10...O15 ⁱⁱ	3.410 (4)	H16A...O14	2.6300
C11...O14	3.056 (5)	H16B...O14	2.6200
C13...C5	3.526 (5)	H16B...C16 ^{vii}	3.0800
C17...C5 ^v	3.567 (5)	H16B...H16B ^{vii}	2.3800
C1...H4 ⁱⁱⁱ	3.1000	H17A...C5 ^v	3.0100
C2...H4 ⁱⁱⁱ	2.9600	H17C...O9 ⁱⁱ	2.7900
C4...H11C ^{vi}	2.8900		
N8—O9—C10	109.1 (2)	C1—C2—H2	120.00
C13—O15—C16	116.7 (2)	C3—C2—H2	120.00
O9—N8—C7	105.9 (2)	C2—C3—H3	120.00
C2—C1—C6	120.2 (3)	C4—C3—H3	120.00
C1—C2—C3	120.6 (3)	C3—C4—H4	120.00
C2—C3—C4	119.1 (3)	C5—C4—H4	119.00
C3—C4—C5	121.0 (3)	C4—C5—H5	120.00
C4—C5—C6	120.2 (3)	C6—C5—H5	120.00
C1—C6—C5	118.8 (3)	C10—C11—H11A	109.00
C1—C6—C7	118.7 (2)	C10—C11—H11B	110.00
C5—C6—C7	122.5 (2)	C10—C11—H11C	109.00
N8—C7—C6	117.5 (2)	H11A—C11—H11B	109.00
N8—C7—C12	110.6 (3)	H11A—C11—H11C	109.00
C6—C7—C12	131.9 (3)	H11B—C11—H11C	109.00
O9—C10—C11	115.8 (3)	O15—C16—H16A	110.00
O9—C10—C12	109.5 (3)	O15—C16—H16B	110.00
C11—C10—C12	134.6 (3)	C17—C16—H16A	110.00
C7—C12—C10	104.9 (2)	C17—C16—H16B	110.00
C7—C12—C13	131.2 (3)	H16A—C16—H16B	108.00
C10—C12—C13	123.9 (3)	C16—C17—H17A	110.00
O14—C13—O15	124.0 (3)	C16—C17—H17B	109.00
O14—C13—C12	122.9 (3)	C16—C17—H17C	109.00
O15—C13—C12	113.1 (2)	H17A—C17—H17B	110.00
O15—C16—C17	107.8 (2)	H17A—C17—H17C	109.00

C2—C1—H1	120.00	H17B—C17—H17C	109.00
C6—C1—H1	120.00		
C10—O9—N8—C7	-0.4 (3)	C1—C6—C7—N8	42.7 (3)
N8—O9—C10—C11	176.9 (2)	C1—C6—C7—C12	-137.0 (3)
N8—O9—C10—C12	-0.3 (3)	C5—C6—C7—N8	-136.0 (3)
C16—O15—C13—O14	4.6 (4)	C5—C6—C7—C12	44.3 (4)
C16—O15—C13—C12	-175.9 (2)	N8—C7—C12—C10	-1.1 (3)
C13—O15—C16—C17	178.4 (2)	N8—C7—C12—C13	-179.1 (2)
O9—N8—C7—C6	-178.90 (18)	C6—C7—C12—C10	178.7 (2)
O9—N8—C7—C12	0.9 (2)	C6—C7—C12—C13	0.7 (4)
C6—C1—C2—C3	-0.8 (4)	O9—C10—C12—C7	0.8 (3)
C2—C1—C6—C5	0.6 (4)	O9—C10—C12—C13	179.0 (2)
C2—C1—C6—C7	-178.2 (2)	C11—C10—C12—C7	-175.7 (3)
C1—C2—C3—C4	1.1 (5)	C11—C10—C12—C13	2.4 (4)
C2—C3—C4—C5	-1.3 (5)	C7—C12—C13—O14	-164.3 (3)
C3—C4—C5—C6	1.1 (4)	C7—C12—C13—O15	16.2 (3)
C4—C5—C6—C1	-0.7 (4)	C10—C12—C13—O14	18.0 (4)
C4—C5—C6—C7	178.0 (2)	C10—C12—C13—O15	-161.5 (2)

Symmetry codes: (i) $x+1, y, z+1$; (ii) $-x+1, -y, -z$; (iii) $x, -y+1/2, z-1/2$; (iv) $x-1, y, z-1$; (v) $x, -y+1/2, z+1/2$; (vi) $x-1, y, z$; (vii) $-x+1, -y, -z+1$; (viii) $x+1, y, z$.