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## Structure Reports

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## 2-Methyl-4-(4-methylpiperazin-1-yl)-10H-thieno[2,3-b][1,5]benzodiazepine (olanzapine) propan-2-ol disolvate

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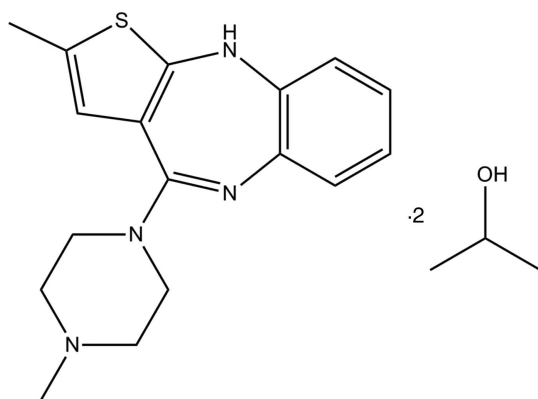
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Key indicators: single-crystal X-ray study;  $T = 123$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.038;  $wR$  factor = 0.106; data-to-parameter ratio = 16.9.

In the title solvate,  $\text{C}_{17}\text{H}_{20}\text{N}_4\text{S}\cdot 2\text{C}_3\text{H}_8\text{O}$ , pairs of olanzapine molecules related by a centre of inversion stack along the  $a$  axis, forming columns, which are packed parallel to each other along the  $b$  axis, forming a sheet arrangement. The columns within these sheets are hydrogen bonded to each other through the propan-2-ol solvent molecules. The diazepine ring of the olanzapine exists in a puckered conformation with the thiophene and phenyl rings making a dihedral angle of  $57.66$  (7)° and the piperazine ring adopts a chair conformation with the methyl group in an equatorial position.

### Related literature

For literature on olanzapine and related structural studies, see: Fulton & Goa (1997); Sanger *et al.* (2001); Tollefson *et al.* (1997); Reutzel-Edens *et al.* (2003); Bhardwaj *et al.* (2013). For details of experimental methods used, see: Florence *et al.* (2003). For details of *XPac*, see: Gelbrich & Hursthouse (2005).



### Experimental

#### Crystal data

$\text{C}_{17}\text{H}_{20}\text{N}_4\text{S}\cdot 2\text{C}_3\text{H}_8\text{O}$   
 $M_r = 432.62$   
 Triclinic,  $P\bar{1}$   
 $a = 9.9621$  (5) Å  
 $b = 10.8702$  (6) Å  
 $c = 12.2298$  (7) Å  
 $\alpha = 70.421$  (2)°  
 $\beta = 74.560$  (2)°

$\gamma = 77.296$  (2)°  
 $V = 1190.24$  (11) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.16$  mm<sup>-1</sup>  
 $T = 123$  K  
 $0.41 \times 0.34 \times 0.11$  mm

#### Data collection

Bruker APEXII CCD diffractometer  
 Absorption correction: multi-scan (*SADABS*; Bruker, 2007)  
 $T_{\min} = 0.581$ ,  $T_{\max} = 0.745$

16365 measured reflections  
 4880 independent reflections  
 4073 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.025$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$   
 $wR(F^2) = 0.106$   
 $S = 1.04$   
 4880 reflections  
 289 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.33$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.29$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1N}\cdots\text{O2S}^{\text{i}}$	0.90 (2)	2.04 (2)	2.933 (2)	177 (2)
$\text{O1S}-\text{H3S}\cdots\text{N4}^{\text{ii}}$	0.86 (2)	1.94 (2)	2.778 (2)	167 (2)

Symmetry codes: (i)  $-x, -y + 2, -z + 1$ ; (ii)  $-x + 1, -y + 1, -z + 1$ .

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008) and *WinGX* (Farrugia, 2012); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008) and *WinGX* (Farrugia, 2012); molecular graphics: *Mercury* (Macrae *et al.*, 2008) and *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *enCIFer* (Allen *et al.*, 2004) and *pubCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FJ2624).

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## supporting information

*Acta Cryst.* (2013). E69, o752–o753 [https://doi.org/10.1107/S1600536813009811]

## 2-Methyl-4-(4-methylpiperazin-1-yl)-10*H*-thieno[2,3-*b*][1,5]benzodiazepine (olanzapine) propan-2-ol disolvate

**Rajni M. Bhardwaj and Alastair J. Florence**

### S1. Comment

2-methyl-4-(4-methylpiperazin-1-yl)-10*H*-thieno[2,3-*b*][1,5]benzodiazepine (Olanzapine, OZPN) is used in the treatment of schizophrenia and related psychoses (Fulton *et al.*, 1997; Tollefson *et al.*, 1997; Sanger *et al.*, 2001). The compound is known to exist in three anhydrous polymorphic forms and 56 solvates including four hydrates have been reported. The crystal structures of two polymorphs and 33 solvates have been reported and all are based on a centrosymmetric dimer motif, which is considered to be the structural building block (Reutzel-Edens *et al.*, 2003; Bhardwaj *et al.*, 2013). The sample of OZPN propan-2-ol solvate was isolated during an experimental physical form screen. The sample was identified as a novel form using multi-sample foil transmission X-ray powder diffraction analysis (Florence *et al.*, 2003).

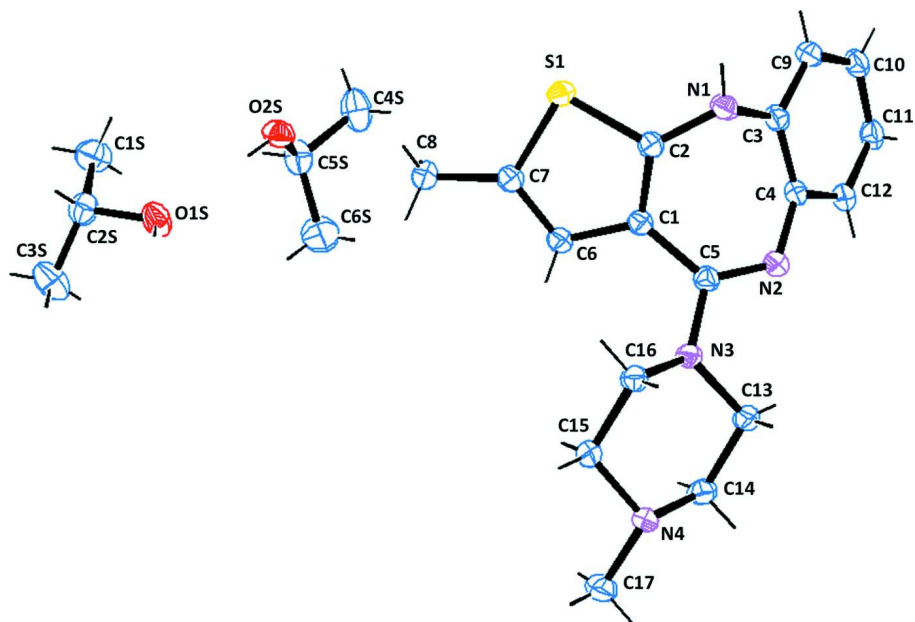
A suitable sample for single-crystal X-ray diffraction analysis was obtained from slow evaporation of saturated propan-2-ol solution at room temperature. The title compound crystallizes in space group *P*-1 with one molecule of OZPN and two molecules of propan-2-ol in the asymmetric unit (Fig. 1). OZPN molecules form centrosymmetric dimers, which stack along the *a*-direction to form columns. These columns further stack along the *b*-direction to form sheets and are H-bonded to each other through propan-2-ol molecules, which are present between the sheets. The solvent separated sheets stack along the *c*-direction to form three-dimensional structure (Fig. 2). XPac (Gelbrich *et al.*, 2005) analysis revealed that this solvate shares 2-D similarity with form I (CSD refcode: UNOGIN01) and dihydrate D (CSD refcode: AQOMAU). The major difference between the structures arises from the relative orientation of the sheets due to incorporation of solvent molecules between them.

### S2. Experimental

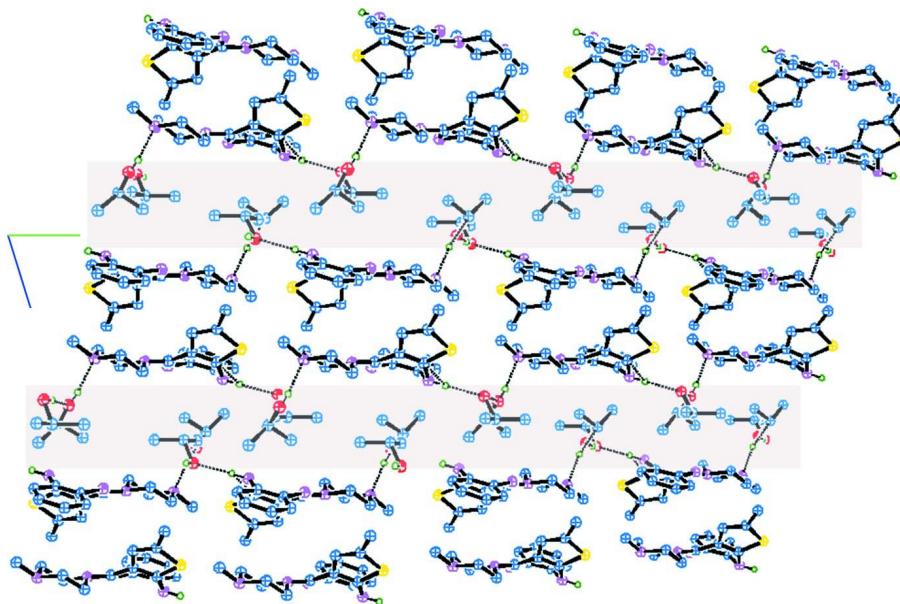
A single plate shaped crystal was grown from the saturated solution of OZPN in propan-2-ol by isothermal solvent evaporation at 298 K.

### S3. Refinement

The positions of the nitrogen and oxygen-bound H atoms were refined freely. All other H atoms were placed in calculated positions and refined in riding modes with C—H = 0.95, 0.98 and 0.99 Å for the aromatic CH, CH<sub>3</sub> and CH<sub>2</sub> groups respectively. The  $U_{\text{iso}}(\text{H})$  values were set to 1.2 and 1.5 times  $U_{\text{eq}}$  of their parent C atoms for the aromatic CH, CH<sub>2</sub> and CH<sub>3</sub> groups respectively.

**Figure 1**

The asymmetric unit of olanzapine propan-2-ol solvate. Displacement ellipsoids are drawn at 50% probability level.

**Figure 2**

The crystal packing in olanzapine propan-2-ol solvate, viewed down the *a*-axis. H-bonds are shown by black dotted line. Carbon, nitrogen, oxygen, sulfur and hydrogen atoms are shown in blue, violet, red, yellow and green colour respectively. Other hydrogen atoms are omitted for clarity.

## 2-Methyl-4-(4-methylpiperazin-1-yl)-10H-thieno[2,3-b][1,5]benzodiazepine propan-2-ol disolvate

## Crystal data

 $C_{17}H_{20}N_4S \cdot 2C_3H_8O$  $M_r = 432.62$ Triclinic,  $P\bar{1}$ 

Hall symbol: -P 1

 $a = 9.9621$  (5) Å $b = 10.8702$  (6) Å $c = 12.2298$  (7) Å $\alpha = 70.421$  (2)° $\beta = 74.560$  (2)° $\gamma = 77.296$  (2)° $V = 1190.24$  (11) Å<sup>3</sup> $Z = 2$  $F(000) = 468$  $D_x = 1.207$  Mg m<sup>-3</sup>Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 8504 reflections

 $\theta = 2.3$ – $26.5$ ° $\mu = 0.16$  mm<sup>-1</sup> $T = 123$  K

Plate, yellow

 $0.41 \times 0.34 \times 0.11$  mm

## Data collection

Bruker APEXII CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 $\varphi$  and  $\omega$  scans

Absorption correction: multi-scan

(SADABS; Bruker, 2007)

 $T_{\min} = 0.581$ ,  $T_{\max} = 0.745$ 

16365 measured reflections

4880 independent reflections

4073 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.025$  $\theta_{\text{max}} = 26.6$ °,  $\theta_{\text{min}} = 1.8$ ° $h = -12 \rightarrow 10$  $k = -11 \rightarrow 13$  $l = -15 \rightarrow 15$ 

## Refinement

Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.038$  $wR(F^2) = 0.106$  $S = 1.04$ 

4880 reflections

289 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

 $w = 1/[\sigma^2(F_o^2) + (0.055P)^2 + 0.4568P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\text{max}} < 0.001$  $\Delta\rho_{\text{max}} = 0.33$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.29$  e Å<sup>-3</sup>

## Special details

**Geometry.** All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
H1N	-0.1193 (19)	0.9466 (19)	0.1993 (16)	0.032 (5)*
H3S	0.514 (2)	0.758 (2)	0.825 (2)	0.050 (6)*
H1S	0.260 (3)	0.830 (2)	0.854 (2)	0.058 (7)*

S1	0.04115 (4)	0.91878 (4)	0.36298 (3)	0.02407 (11)
N1	-0.09642 (13)	0.85854 (13)	0.22328 (11)	0.0230 (3)
N4	0.36303 (13)	0.26018 (12)	0.32261 (11)	0.0219 (3)
N3	0.14149 (12)	0.47811 (12)	0.30511 (11)	0.0217 (3)
N2	-0.08278 (13)	0.56928 (12)	0.28169 (11)	0.0219 (3)
C14	0.21368 (16)	0.24248 (14)	0.35573 (14)	0.0244 (3)
H14A	0.1787	0.2325	0.4415	0.029*
H14B	0.2046	0.1606	0.3414	0.029*
C16	0.28887 (15)	0.49962 (14)	0.26562 (14)	0.0223 (3)
H16A	0.2985	0.5831	0.2769	0.027*
H16B	0.3210	0.5068	0.1801	0.027*
C5	0.03110 (15)	0.57973 (14)	0.30703 (12)	0.0200 (3)
C6	0.13109 (15)	0.67258 (14)	0.43396 (13)	0.0213 (3)
H6	0.1755	0.5898	0.4756	0.026*
C7	0.13712 (15)	0.78773 (15)	0.45145 (13)	0.0228 (3)
C13	0.12395 (16)	0.35797 (14)	0.28589 (14)	0.0229 (3)
H13A	0.1534	0.3649	0.2003	0.027*
H13B	0.0238	0.3448	0.3126	0.027*
C12	-0.33300 (16)	0.60148 (15)	0.33644 (13)	0.0246 (3)
H12	-0.3275	0.5082	0.3590	0.030*
C3	-0.21953 (15)	0.79562 (14)	0.26365 (12)	0.0211 (3)
C15	0.37834 (16)	0.38554 (15)	0.33649 (14)	0.0243 (3)
H15A	0.4783	0.3993	0.3090	0.029*
H15B	0.3493	0.3817	0.4214	0.029*
C2	-0.00519 (15)	0.81452 (14)	0.30203 (13)	0.0208 (3)
C10	-0.47383 (16)	0.81431 (16)	0.30737 (14)	0.0259 (3)
H10	-0.5633	0.8678	0.3091	0.031*
C4	-0.20831 (15)	0.65708 (14)	0.29640 (13)	0.0213 (3)
C8	0.20862 (17)	0.81074 (16)	0.53509 (15)	0.0273 (3)
H8A	0.1404	0.8144	0.6087	0.041*
H8B	0.2467	0.8944	0.4981	0.041*
H8C	0.2854	0.7384	0.5530	0.041*
C17	0.44267 (18)	0.14983 (16)	0.39751 (15)	0.0301 (4)
H17A	0.5423	0.1610	0.3746	0.045*
H17B	0.4325	0.0669	0.3873	0.045*
H17C	0.4066	0.1476	0.4808	0.045*
C11	-0.46419 (16)	0.67801 (16)	0.34425 (14)	0.0259 (3)
H11	-0.5471	0.6373	0.3747	0.031*
C1	0.05185 (15)	0.68643 (14)	0.34722 (13)	0.0206 (3)
C9	-0.35177 (16)	0.87206 (15)	0.26790 (13)	0.0241 (3)
H9	-0.3585	0.9655	0.2433	0.029*
O1S	0.44594 (13)	0.77562 (13)	0.87986 (11)	0.0348 (3)
C2S	0.50035 (18)	0.80746 (17)	0.96257 (14)	0.0318 (4)
H2S	0.5619	0.8773	0.9179	0.038*
C3S	0.58641 (19)	0.6878 (2)	1.02927 (16)	0.0425 (5)
H3S1	0.5278	0.6176	1.0712	0.064*
H3S2	0.6211	0.7107	1.0868	0.064*
H3S3	0.6664	0.6572	0.9733	0.064*

C1S	0.3754 (2)	0.8628 (2)	1.04315 (16)	0.0420 (5)
H1S1	0.3228	0.9397	0.9952	0.063*
H1S2	0.4087	0.8892	1.0994	0.063*
H1S3	0.3140	0.7952	1.0869	0.063*
O2S	0.17158 (13)	0.85357 (11)	0.84879 (10)	0.0308 (3)
C5S	0.08774 (19)	0.77005 (18)	0.94873 (15)	0.0345 (4)
H6S	0.1012	0.7803	1.0232	0.041*
C4S	-0.0636 (2)	0.8177 (2)	0.93962 (18)	0.0464 (5)
H5S1	-0.0883	0.9095	0.9413	0.070*
H5S2	-0.1242	0.7625	1.0066	0.070*
H5S3	-0.0771	0.8120	0.8650	0.070*
C6S	0.1325 (3)	0.62811 (19)	0.9508 (2)	0.0577 (6)
H7S1	0.1192	0.6169	0.8783	0.087*
H7S2	0.0755	0.5723	1.0202	0.087*
H7S3	0.2320	0.6026	0.9552	0.087*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0235 (2)	0.01788 (19)	0.0328 (2)	-0.00162 (14)	-0.00854 (15)	-0.00866 (15)
N1	0.0219 (6)	0.0188 (6)	0.0268 (6)	-0.0008 (5)	-0.0093 (5)	-0.0029 (5)
N4	0.0211 (6)	0.0199 (6)	0.0267 (6)	0.0027 (5)	-0.0098 (5)	-0.0092 (5)
N3	0.0173 (6)	0.0179 (6)	0.0316 (7)	-0.0013 (5)	-0.0065 (5)	-0.0092 (5)
N2	0.0208 (6)	0.0206 (6)	0.0259 (6)	-0.0006 (5)	-0.0085 (5)	-0.0074 (5)
C14	0.0241 (8)	0.0187 (7)	0.0306 (8)	-0.0019 (6)	-0.0050 (6)	-0.0088 (6)
C16	0.0181 (7)	0.0198 (7)	0.0306 (8)	-0.0027 (6)	-0.0051 (6)	-0.0096 (6)
C5	0.0203 (7)	0.0181 (7)	0.0208 (7)	-0.0029 (6)	-0.0051 (6)	-0.0039 (5)
C6	0.0182 (7)	0.0204 (7)	0.0258 (7)	-0.0008 (6)	-0.0063 (6)	-0.0072 (6)
C7	0.0188 (7)	0.0232 (7)	0.0272 (7)	-0.0017 (6)	-0.0054 (6)	-0.0088 (6)
C13	0.0202 (7)	0.0201 (7)	0.0310 (8)	-0.0023 (6)	-0.0068 (6)	-0.0103 (6)
C12	0.0247 (8)	0.0246 (8)	0.0283 (8)	-0.0023 (6)	-0.0121 (6)	-0.0082 (6)
C3	0.0210 (7)	0.0236 (7)	0.0201 (7)	-0.0008 (6)	-0.0074 (6)	-0.0071 (6)
C15	0.0210 (7)	0.0249 (8)	0.0315 (8)	0.0009 (6)	-0.0098 (6)	-0.0135 (6)
C2	0.0171 (7)	0.0209 (7)	0.0253 (7)	-0.0022 (6)	-0.0044 (6)	-0.0084 (6)
C10	0.0201 (7)	0.0337 (9)	0.0267 (8)	0.0042 (6)	-0.0096 (6)	-0.0140 (7)
C4	0.0211 (7)	0.0231 (7)	0.0223 (7)	0.0005 (6)	-0.0099 (6)	-0.0082 (6)
C8	0.0261 (8)	0.0267 (8)	0.0346 (8)	-0.0025 (6)	-0.0105 (7)	-0.0138 (7)
C17	0.0330 (9)	0.0266 (8)	0.0309 (8)	0.0065 (7)	-0.0143 (7)	-0.0094 (7)
C11	0.0201 (7)	0.0337 (8)	0.0279 (8)	-0.0037 (6)	-0.0083 (6)	-0.0119 (6)
C1	0.0167 (7)	0.0201 (7)	0.0256 (7)	-0.0008 (6)	-0.0050 (6)	-0.0084 (6)
C9	0.0259 (8)	0.0225 (7)	0.0244 (7)	0.0027 (6)	-0.0100 (6)	-0.0079 (6)
O1S	0.0261 (6)	0.0516 (8)	0.0322 (6)	0.0042 (5)	-0.0095 (5)	-0.0228 (6)
C2S	0.0355 (9)	0.0347 (9)	0.0283 (8)	-0.0080 (7)	-0.0084 (7)	-0.0104 (7)
C3S	0.0288 (9)	0.0608 (13)	0.0309 (9)	0.0016 (9)	-0.0060 (7)	-0.0098 (9)
C1S	0.0504 (12)	0.0442 (11)	0.0350 (10)	0.0068 (9)	-0.0151 (9)	-0.0201 (8)
O2S	0.0258 (6)	0.0242 (6)	0.0362 (6)	-0.0032 (5)	-0.0058 (5)	-0.0019 (5)
C5S	0.0340 (9)	0.0412 (10)	0.0281 (8)	-0.0136 (8)	-0.0074 (7)	-0.0042 (7)
C4S	0.0314 (10)	0.0684 (14)	0.0401 (10)	-0.0170 (9)	-0.0035 (8)	-0.0139 (10)

C6S    0.0651 (15)    0.0318 (10)    0.0726 (15)    -0.0172 (10)    -0.0297 (12)    0.0074 (10)

*Geometric parameters (Å, °)*

S1—C7	1.7415 (15)	C10—C11	1.386 (2)
S1—C2	1.7415 (15)	C10—C9	1.388 (2)
N1—C2	1.3959 (19)	C10—H10	0.9500
N1—C3	1.4268 (19)	C8—H8A	0.9800
N1—H1N	0.897 (19)	C8—H8B	0.9800
N4—C17	1.4627 (19)	C8—H8C	0.9800
N4—C14	1.4723 (19)	C17—H17A	0.9800
N4—C15	1.4728 (18)	C17—H17B	0.9800
N3—C5	1.3782 (18)	C17—H17C	0.9800
N3—C13	1.4558 (18)	C11—H11	0.9500
N3—C16	1.4633 (18)	C9—H9	0.9500
N2—C5	1.2897 (19)	O1S—C2S	1.430 (2)
N2—C4	1.4061 (18)	O1S—H3S	0.86 (2)
C14—C13	1.516 (2)	C2S—C3S	1.504 (2)
C14—H14A	0.9900	C2S—C1S	1.515 (2)
C14—H14B	0.9900	C2S—H2S	1.0000
C16—C15	1.513 (2)	C3S—H3S1	0.9800
C16—H16A	0.9900	C3S—H3S2	0.9800
C16—H16B	0.9900	C3S—H3S3	0.9800
C5—C1	1.474 (2)	C1S—H1S1	0.9800
C6—C7	1.355 (2)	C1S—H1S2	0.9800
C6—C1	1.434 (2)	C1S—H1S3	0.9800
C6—H6	0.9500	O2S—C5S	1.436 (2)
C7—C8	1.501 (2)	O2S—H1S	0.87 (2)
C13—H13A	0.9900	C5S—C6S	1.502 (3)
C13—H13B	0.9900	C5S—C4S	1.504 (3)
C12—C11	1.385 (2)	C5S—H6S	1.0000
C12—C4	1.399 (2)	C4S—H5S1	0.9800
C12—H12	0.9500	C4S—H5S2	0.9800
C3—C9	1.393 (2)	C4S—H5S3	0.9800
C3—C4	1.409 (2)	C6S—H7S1	0.9800
C15—H15A	0.9900	C6S—H7S2	0.9800
C15—H15B	0.9900	C6S—H7S3	0.9800
C2—C1	1.368 (2)		
C7—S1—C2	91.98 (7)	C7—C8—H8A	109.5
C2—N1—C3	114.13 (12)	C7—C8—H8B	109.5
C2—N1—H1N	113.8 (12)	H8A—C8—H8B	109.5
C3—N1—H1N	110.7 (12)	C7—C8—H8C	109.5
C17—N4—C14	109.50 (12)	H8A—C8—H8C	109.5
C17—N4—C15	110.02 (12)	H8B—C8—H8C	109.5
C14—N4—C15	110.22 (11)	N4—C17—H17A	109.5
C5—N3—C13	120.29 (12)	N4—C17—H17B	109.5
C5—N3—C16	122.85 (12)	H17A—C17—H17B	109.5



C13—N3—C16	110.77 (11)	N4—C17—H17C	109.5
C5—N2—C4	123.11 (13)	H17A—C17—H17C	109.5
N4—C14—C13	112.02 (12)	H17B—C17—H17C	109.5
N4—C14—H14A	109.2	C12—C11—C10	119.60 (14)
C13—C14—H14A	109.2	C12—C11—H11	120.2
N4—C14—H14B	109.2	C10—C11—H11	120.2
C13—C14—H14B	109.2	C2—C1—C6	112.09 (13)
H14A—C14—H14B	107.9	C2—C1—C5	121.52 (13)
N3—C16—C15	109.49 (12)	C6—C1—C5	126.38 (13)
N3—C16—H16A	109.8	C10—C9—C3	121.24 (14)
C15—C16—H16A	109.8	C10—C9—H9	119.4
N3—C16—H16B	109.8	C3—C9—H9	119.4
C15—C16—H16B	109.8	C2S—O1S—H3S	109.0 (15)
H16A—C16—H16B	108.2	O1S—C2S—C3S	110.46 (14)
N2—C5—N3	118.01 (13)	O1S—C2S—C1S	106.97 (14)
N2—C5—C1	126.03 (13)	C3S—C2S—C1S	112.88 (15)
N3—C5—C1	115.78 (12)	O1S—C2S—H2S	108.8
C7—C6—C1	114.22 (13)	C3S—C2S—H2S	108.8
C7—C6—H6	122.9	C1S—C2S—H2S	108.8
C1—C6—H6	122.9	C2S—C3S—H3S1	109.5
C6—C7—C8	128.92 (14)	C2S—C3S—H3S2	109.5
C6—C7—S1	110.50 (11)	H3S1—C3S—H3S2	109.5
C8—C7—S1	120.55 (11)	C2S—C3S—H3S3	109.5
N3—C13—C14	108.59 (12)	H3S1—C3S—H3S3	109.5
N3—C13—H13A	110.0	H3S2—C3S—H3S3	109.5
C14—C13—H13A	110.0	C2S—C1S—H1S1	109.5
N3—C13—H13B	110.0	C2S—C1S—H1S2	109.5
C14—C13—H13B	110.0	H1S1—C1S—H1S2	109.5
H13A—C13—H13B	108.4	C2S—C1S—H1S3	109.5
C11—C12—C4	122.09 (14)	H1S1—C1S—H1S3	109.5
C11—C12—H12	119.0	H1S2—C1S—H1S3	109.5
C4—C12—H12	119.0	C5S—O2S—H1S	108.9 (15)
C9—C3—C4	119.78 (14)	O2S—C5S—C6S	110.43 (16)
C9—C3—N1	119.63 (13)	O2S—C5S—C4S	107.23 (14)
C4—C3—N1	120.57 (13)	C6S—C5S—C4S	112.72 (17)
N4—C15—C16	110.50 (12)	O2S—C5S—H6S	108.8
N4—C15—H15A	109.5	C6S—C5S—H6S	108.8
C16—C15—H15A	109.5	C4S—C5S—H6S	108.8
N4—C15—H15B	109.5	C5S—C4S—H5S1	109.5
C16—C15—H15B	109.5	C5S—C4S—H5S2	109.5
H15A—C15—H15B	108.1	H5S1—C4S—H5S2	109.5
C1—C2—N1	125.87 (13)	C5S—C4S—H5S3	109.5
C1—C2—S1	111.18 (11)	H5S1—C4S—H5S3	109.5
N1—C2—S1	122.83 (11)	H5S2—C4S—H5S3	109.5
C11—C10—C9	119.46 (14)	C5S—C6S—H7S1	109.5
C11—C10—H10	120.3	C5S—C6S—H7S2	109.5
C9—C10—H10	120.3	H7S1—C6S—H7S2	109.5
C12—C4—N2	116.18 (13)	C5S—C6S—H7S3	109.5

C12—C4—C3	117.75 (13)	H7S1—C6S—H7S3	109.5
N2—C4—C3	125.76 (13)	H7S2—C6S—H7S3	109.5
C17—N4—C14—C13	176.70 (12)	C7—S1—C2—N1	175.64 (13)
C15—N4—C14—C13	55.54 (16)	C11—C12—C4—N2	-173.92 (13)
C5—N3—C16—C15	146.27 (13)	C11—C12—C4—C3	0.1 (2)
C13—N3—C16—C15	-61.22 (15)	C5—N2—C4—C12	-142.24 (15)
C4—N2—C5—N3	171.00 (13)	C5—N2—C4—C3	44.3 (2)
C4—N2—C5—C1	-4.0 (2)	C9—C3—C4—C12	-2.3 (2)
C13—N3—C5—N2	-5.6 (2)	N1—C3—C4—C12	179.32 (13)
C16—N3—C5—N2	144.38 (14)	C9—C3—C4—N2	171.08 (13)
C13—N3—C5—C1	169.89 (13)	N1—C3—C4—N2	-7.3 (2)
C16—N3—C5—C1	-40.10 (19)	C4—C12—C11—C10	2.5 (2)
C1—C6—C7—C8	179.38 (15)	C9—C10—C11—C12	-2.8 (2)
C1—C6—C7—S1	1.28 (16)	N1—C2—C1—C6	-174.71 (13)
C2—S1—C7—C6	-0.43 (12)	S1—C2—C1—C6	1.34 (16)
C2—S1—C7—C8	-178.71 (13)	N1—C2—C1—C5	5.7 (2)
C5—N3—C13—C14	-146.81 (13)	S1—C2—C1—C5	-178.28 (11)
C16—N3—C13—C14	59.88 (15)	C7—C6—C1—C2	-1.73 (19)
N4—C14—C13—N3	-57.24 (16)	C7—C6—C1—C5	177.87 (13)
C2—N1—C3—C9	126.29 (14)	N2—C5—C1—C2	-38.1 (2)
C2—N1—C3—C4	-55.31 (18)	N3—C5—C1—C2	146.84 (14)
C17—N4—C15—C16	-176.32 (13)	N2—C5—C1—C6	142.37 (16)
C14—N4—C15—C16	-55.47 (16)	N3—C5—C1—C6	-32.7 (2)
N3—C16—C15—N4	58.31 (16)	C11—C10—C9—C3	0.6 (2)
C3—N1—C2—C1	55.2 (2)	C4—C3—C9—C10	2.0 (2)
C3—N1—C2—S1	-120.37 (13)	N1—C3—C9—C10	-179.59 (13)
C7—S1—C2—C1	-0.55 (12)		

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1N...O2S <sup>i</sup>	0.90 (2)	2.04 (2)	2.933 (2)	177 (2)
O1S—H3S...N4 <sup>ii</sup>	0.86 (2)	1.94 (2)	2.778 (2)	167 (2)

Symmetry codes: (i)  $-x, -y+2, -z+1$ ; (ii)  $-x+1, -y+1, -z+1$ .