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(1*S*,3*R*,8*R*,9*S*,11*R*)-2,2-Dibromo-10,10-dichloro-3,7,7,11-tetramethyltetracyclo-[6.5.0.0^{1,3}.0^{9,11}]tridecane

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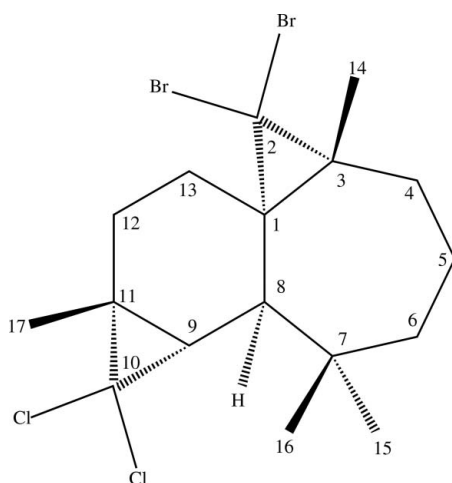
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.038; wR factor = 0.092; data-to-parameter ratio = 19.6.

The title compound, $\text{C}_{17}\text{H}_{24}\text{Br}_2\text{Cl}_2$, was synthesized from β -himachalene (3,5,5,9-tetramethyl-2,4a,5,6,7,8-hexahydro-1*H*-benzocycloheptene), which was isolated from the essential oil of the Atlas cedar (*Cedrus Atlantica*). The molecule contains fused six-, seven- and two three-membered rings. The six-membered ring has a half-chair conformation, while the seven-membered ring displays a boat conformation. The absolute structure was unambiguously established from anomalous dispersion effects. The crystal packing exhibits no short intermolecular contacts.

Related literature

For the crystal structures of related compounds, see: Ourhiss *et al.* (2013); Oukhrib *et al.* (2013*a,b*). For puckering parameters, see: Cremer & Pople (1975).



Experimental

Crystal data

$\text{C}_{17}\text{H}_{24}\text{Br}_2\text{Cl}_2$
 $M_r = 459.08$
Monoclinic, $P2_1$
 $a = 9.0112$ (6) Å
 $b = 11.6772$ (8) Å
 $c = 9.0849$ (6) Å
 $\beta = 108.045$ (5)°
 $V = 908.94$ (11) Å³
 $Z = 2$
Mo $K\alpha$ radiation
 $\mu = 4.75$ mm⁻¹
 $T = 296$ K
 $0.45 \times 0.34 \times 0.29$ mm

Data collection

Bruker X8 APEX diffractometer
Absorption correction: multi-scan (SADABS; Sheldrick, 2008)
 $T_{\min} = 0.739$, $T_{\max} = 0.867$
6676 measured reflections
3720 independent reflections
2910 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.092$
 $S = 1.03$
3720 reflections
190 parameters
1 restraint
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.64$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.79$ e Å⁻³
Absolute structure: Flack & Bernardinelli (2000), 1096 Friedel pairs
Flack parameter: 0.039 (10)

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012); software used to prepare material for publication: PLATON (Spek, 2009) and publCIF (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5397).

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supporting information

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(1*S*,3*R*,8*R*,9*S*,11*R*)-2,2-Dibromo-10,10-dichloro-3,7,7,11-tetramethyltetracyclo-[6.5.0.0^{1,3}.0^{9,11}]tridecane

Najia Ourhriss, Ahmed Benharref, Mohamed Saadi, Moha Berraho and Lahcen El Ammari

S1. Comment

As a continuation of our structural study of β -himachalene derivatives isolated from the essential oil of the Atlas cedar (*Cedrus Atlantica*) (Ourhriss *et al.*, 2013; Oukhrib *et al.*, 2013*a,b*), we present here the crystal structure of the title compound, (1*S*,3*R*,8*R*,9*S*,11*R*)-2,2-dibromo-10,10-dichloro-3,7,7,11-tetramethyltetracyclo[6.5.0.0^{1,3}.0^{9,11}]tridecane, (I).

In (I) (Fig. 1), the molecule contains a fused six- and seven-membered rings, which are fused to two three-membered rings. The six-membered ring has a half chair conformation as indicated by the total puckering amplitude QT = 0.446 (5) Å and spherical polar angle $\theta = 140.8$ (6)° and $\varphi_2 = 143$ (1)°, whereas the seven-membered ring displays a boat conformation with QT = 1.122 (5) Å, $\theta_2 = 87.4$ (3)°, $\varphi_2 = -48.2$ (3)° and $\varphi_3 = -118$ (5)° (Cremer & Pople, 1975). The three-membered rings (C1/C2/C3) and (C9/C10/C11) are nearly perpendicular to the six and seven-membered rings (C1/C8–C13) and (C1/C3–C8), with a dihedral angles of 84.1 (4) and 80.6 (4), respectively. Owing to the presence of Br and Cl atoms, the absolute configuration could be fully confirmed from anomalous dispersion effects, by refining the Flack parameter as C1(*S*), C3(*R*), C8(*R*), C9(*S*), and C11(*R*). The crystal packing exhibits no short intermolecular contacts.

S2. Experimental

A solution containing 4 g (10 mmol) of (1*S*,3*R*,8*S*)-2,2-dibromo-3,7,7,10-tetramethyltricyclo[6.4.0.01,3]dodec-9-ene and 1 ml (12.4 mmol) of CHCl₃ in 40 ml of dichloromethane was added dropwise at 273 K over 30 min to 1 g (25 mmol) of pulverized sodium hydroxide and 40 mg of *N*-benzyltriethylammonium chloride placed in a 100 ml three-necked flask. After stirring at room temperature for 2 h, the mixture was filtered on celite and concentrated in vacuum. The residue obtained was chromatographed on silicagel column impregnated with silver nitrate (10%) with a mixture of hexane - ethyl acetate (96–4) used as eluent. The two diastereoisomers (1*S*,3*R*,8*S*,9*S*,11*R*)-2,2-Dibromo-10,10-dichloro-3,7,7,11-tetramethyltetracyclo[6.5.0.0^{1,3}.0^{9,11}]tridecane(*X*) and its isomer (1*S*,3*R*,8*S*,9*R*,11*S*)-2,2-Dibromo-10,10-dichloro-3,7,7,11-tetramethyltetracyclo[6.5.0.0^{1,3}.0^{9,11}]tridecane(*Y*) were obtained by this procedure in a 80/20 ratio and a combined yield of 65% (3 g; 6.5 mmol). The title compound (isomer *X*) was recrystallized from hexane.

S3. Refinement

All H atoms were fixed geometrically and treated as riding with C—H = 0.96 Å (methyl), 0.97 Å (methylene), 0.98 Å (methine) with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{methylene, methine})$ and $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{methyl})$. The space group is non-centrosymmetric and the polar axis restraint is generated automatically by *SHELXL* program. The 1096 Friedel opposites reflections are not merged.

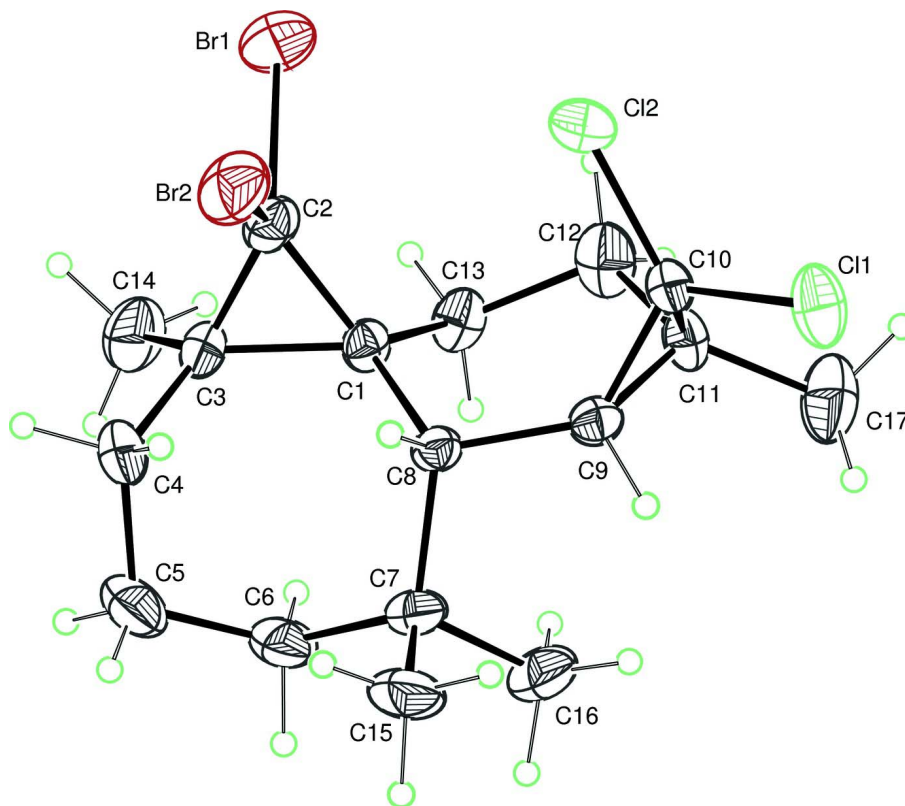


Figure 1

Molecular structure of the title compound with the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are represented as small spheres of arbitrary radii.

(1*S*,3*R*,8*R*,9*S*,11*R*)-2,2-Dibromo-10,10-dichloro-3,7,7,11-tetramethyltetracyclo[6.5.0.0^{1,3}.0^{9,11}]tridecane

Crystal data

$C_{17}H_{24}Br_2Cl_2$

$M_r = 459.08$

Monoclinic, $P2_1$

Hall symbol: $P\ 2y_b$

$a = 9.0112\ (6)\ \text{\AA}$

$b = 11.6772\ (8)\ \text{\AA}$

$c = 9.0849\ (6)\ \text{\AA}$

$\beta = 108.045\ (5)^\circ$

$V = 908.94\ (11)\ \text{\AA}^3$

$Z = 2$

$F(000) = 460$

$D_x = 1.677\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 3720 reflections

$\theta = 2.8\text{--}29.6^\circ$

$\mu = 4.75\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Block, colourless

$0.45 \times 0.34 \times 0.29\ \text{mm}$

Data collection

Bruker X8 APEX

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan

(*SADABS*; Sheldrick, 2008)

$T_{\min} = 0.739$, $T_{\max} = 0.867$

6676 measured reflections

3720 independent reflections

2910 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.034$

$\theta_{\max} = 29.6^\circ$, $\theta_{\min} = 2.8^\circ$

$h = -12 \rightarrow 12$

$k = -16 \rightarrow 15$

$l = -12 \rightarrow 12$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.038$ $wR(F^2) = 0.092$ $S = 1.03$

3720 reflections

190 parameters

1 restraint

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0333P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.64 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.79 \text{ e } \text{\AA}^{-3}$ Absolute structure: Flack & Bernardinelli
(2000), 1096 Friedel pairs

Absolute structure parameter: 0.039 (10)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against all reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on all data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.8734 (5)	0.4080 (3)	0.7265 (5)	0.0211 (8)
C2	0.9989 (5)	0.3302 (4)	0.6990 (6)	0.0288 (10)
C3	1.0234 (4)	0.3775 (4)	0.8568 (5)	0.0275 (9)
C4	1.0081 (6)	0.2965 (4)	0.9812 (5)	0.0357 (11)
H4A	1.1102	0.2657	1.0363	0.043*
H4B	0.9413	0.2330	0.9329	0.043*
C5	0.9408 (6)	0.3542 (5)	1.0962 (6)	0.0442 (13)
H5A	1.0232	0.3954	1.1721	0.053*
H5B	0.9014	0.2960	1.1507	0.053*
C6	0.8095 (6)	0.4372 (5)	1.0187 (5)	0.0387 (12)
H6A	0.8567	0.5027	0.9849	0.046*
H6B	0.7670	0.4645	1.0981	0.046*
C7	0.6702 (5)	0.3992 (4)	0.8798 (5)	0.0301 (9)
C8	0.7181 (4)	0.3573 (3)	0.7345 (5)	0.0206 (8)
H8	0.7319	0.2741	0.7439	0.025*
C9	0.5812 (4)	0.3798 (4)	0.5883 (4)	0.0224 (8)
H9	0.4807	0.3749	0.6086	0.027*
C10	0.5657 (5)	0.3530 (4)	0.4239 (5)	0.0261 (9)
C11	0.5826 (5)	0.4755 (4)	0.4748 (5)	0.0278 (9)
C12	0.7378 (6)	0.5328 (4)	0.4939 (6)	0.0373 (11)
H12A	0.7190	0.6125	0.4639	0.045*
H12B	0.7856	0.4970	0.4234	0.045*
C13	0.8528 (5)	0.5277 (3)	0.6570 (5)	0.0277 (9)

H13A	0.9534	0.5556	0.6546	0.033*
H13B	0.8168	0.5784	0.7235	0.033*
C14	1.1485 (6)	0.4682 (5)	0.9207 (7)	0.0432 (13)
H14A	1.1495	0.4896	1.0230	0.065*
H14B	1.1262	0.5343	0.8546	0.065*
H14C	1.2486	0.4377	0.9247	0.065*
C15	0.5843 (6)	0.2992 (5)	0.9266 (6)	0.0453 (13)
H15A	0.4981	0.2763	0.8394	0.068*
H15B	0.5463	0.3228	1.0096	0.068*
H15C	0.6547	0.2360	0.9601	0.068*
C16	0.5593 (6)	0.5035 (5)	0.8432 (6)	0.0419 (12)
H16A	0.6113	0.5673	0.8140	0.063*
H16B	0.5305	0.5232	0.9333	0.063*
H16C	0.4673	0.4849	0.7596	0.063*
C17	0.4417 (7)	0.5546 (4)	0.4175 (7)	0.0479 (14)
H17A	0.4693	0.6299	0.4593	0.072*
H17B	0.3576	0.5255	0.4506	0.072*
H17C	0.4095	0.5581	0.3065	0.072*
Cl1	0.37818 (14)	0.30587 (10)	0.30591 (14)	0.0415 (3)
Cl2	0.70977 (16)	0.28463 (12)	0.36234 (14)	0.0429 (3)
Br1	1.11977 (6)	0.38295 (6)	0.57096 (7)	0.05382 (18)
Br2	0.97449 (6)	0.16666 (4)	0.66883 (6)	0.04537 (16)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0167 (18)	0.022 (2)	0.023 (2)	-0.0038 (15)	0.0039 (16)	-0.0013 (15)
C2	0.019 (2)	0.034 (2)	0.033 (3)	-0.0019 (17)	0.0085 (19)	-0.0011 (18)
C3	0.0185 (18)	0.035 (2)	0.028 (2)	0.005 (2)	0.0054 (17)	0.004 (2)
C4	0.028 (2)	0.049 (3)	0.024 (2)	0.003 (2)	-0.0001 (19)	0.003 (2)
C5	0.044 (3)	0.066 (4)	0.021 (3)	-0.004 (3)	0.007 (2)	-0.002 (2)
C6	0.038 (3)	0.057 (3)	0.023 (3)	-0.001 (2)	0.013 (2)	-0.010 (2)
C7	0.029 (2)	0.038 (2)	0.028 (2)	0.0027 (19)	0.0153 (19)	-0.004 (2)
C8	0.0199 (18)	0.0217 (19)	0.022 (2)	0.0009 (15)	0.0096 (17)	0.0021 (15)
C9	0.0205 (18)	0.0241 (18)	0.024 (2)	-0.0013 (18)	0.0093 (16)	0.0001 (18)
C10	0.022 (2)	0.035 (2)	0.019 (2)	0.0020 (17)	0.0027 (17)	0.0016 (16)
C11	0.029 (2)	0.028 (2)	0.023 (2)	0.0031 (18)	0.0021 (19)	0.0076 (17)
C12	0.035 (3)	0.030 (2)	0.043 (3)	-0.006 (2)	0.007 (2)	0.011 (2)
C13	0.022 (2)	0.022 (2)	0.037 (3)	-0.0034 (17)	0.0067 (19)	0.0028 (18)
C14	0.023 (2)	0.054 (3)	0.047 (3)	-0.008 (2)	0.004 (2)	-0.005 (2)
C15	0.043 (3)	0.068 (3)	0.033 (3)	-0.005 (3)	0.024 (3)	0.004 (3)
C16	0.036 (3)	0.051 (3)	0.044 (3)	0.011 (2)	0.019 (3)	-0.008 (2)
C17	0.042 (3)	0.038 (3)	0.053 (3)	0.012 (2)	-0.001 (3)	0.003 (2)
Cl1	0.0330 (6)	0.0449 (7)	0.0341 (7)	-0.0066 (5)	-0.0077 (5)	-0.0011 (5)
Cl2	0.0438 (7)	0.0613 (8)	0.0262 (6)	0.0125 (6)	0.0144 (5)	-0.0040 (5)
Br1	0.0336 (3)	0.0840 (4)	0.0516 (3)	-0.0014 (3)	0.0244 (3)	0.0083 (3)
Br2	0.0452 (3)	0.0386 (2)	0.0525 (3)	0.0134 (3)	0.0156 (3)	-0.0033 (3)

Geometric parameters (Å, °)

C1—C13	1.522 (6)	C9—C11	1.524 (6)
C1—C2	1.530 (6)	C9—H9	0.9800
C1—C3	1.537 (5)	C10—C11	1.496 (6)
C1—C8	1.542 (5)	C10—C12	1.756 (4)
C2—C3	1.488 (6)	C10—C11	1.785 (4)
C2—Br1	1.924 (4)	C11—C12	1.511 (6)
C2—Br2	1.932 (4)	C11—C17	1.525 (7)
C3—C4	1.513 (6)	C12—C13	1.524 (7)
C3—C14	1.524 (7)	C12—H12A	0.9700
C4—C5	1.519 (7)	C12—H12B	0.9700
C4—H4A	0.9700	C13—H13A	0.9700
C4—H4B	0.9700	C13—H13B	0.9700
C5—C6	1.523 (7)	C14—H14A	0.9600
C5—H5A	0.9700	C14—H14B	0.9600
C5—H5B	0.9700	C14—H14C	0.9600
C6—C7	1.544 (6)	C15—H15A	0.9600
C6—H6A	0.9700	C15—H15B	0.9600
C6—H6B	0.9700	C15—H15C	0.9600
C7—C15	1.531 (7)	C16—H16A	0.9600
C7—C16	1.545 (7)	C16—H16B	0.9600
C7—C8	1.587 (5)	C16—H16C	0.9600
C8—C9	1.529 (6)	C17—H17A	0.9600
C8—H8	0.9800	C17—H17B	0.9600
C9—C10	1.490 (6)	C17—H17C	0.9600
C13—C1—C2	118.6 (3)	C10—C9—H9	111.8
C13—C1—C3	119.8 (3)	C11—C9—H9	111.8
C2—C1—C3	58.1 (3)	C8—C9—H9	111.8
C13—C1—C8	112.1 (3)	C9—C10—C11	61.4 (3)
C2—C1—C8	120.5 (3)	C9—C10—C12	124.4 (3)
C3—C1—C8	118.0 (3)	C11—C10—C12	121.3 (3)
C3—C2—C1	61.2 (3)	C9—C10—C11	116.4 (3)
C3—C2—Br1	121.4 (3)	C11—C10—C11	117.8 (3)
C1—C2—Br1	119.6 (3)	C12—C10—C11	108.9 (2)
C3—C2—Br2	118.7 (3)	C10—C11—C12	117.4 (4)
C1—C2—Br2	123.5 (3)	C10—C11—C9	59.1 (3)
Br1—C2—Br2	106.9 (2)	C12—C11—C9	116.6 (4)
C2—C3—C4	117.9 (4)	C10—C11—C17	118.6 (4)
C2—C3—C14	119.7 (4)	C12—C11—C17	114.5 (4)
C4—C3—C14	112.2 (4)	C9—C11—C17	119.7 (4)
C2—C3—C1	60.7 (3)	C11—C12—C13	114.9 (4)
C4—C3—C1	117.3 (3)	C11—C12—H12A	108.5
C14—C3—C1	120.2 (4)	C13—C12—H12A	108.5
C3—C4—C5	112.6 (4)	C11—C12—H12B	108.5
C3—C4—H4A	109.1	C13—C12—H12B	108.5
C5—C4—H4A	109.1	H12A—C12—H12B	107.5

C3—C4—H4B	109.1	C1—C13—C12	113.5 (4)
C5—C4—H4B	109.1	C1—C13—H13A	108.9
H4A—C4—H4B	107.8	C12—C13—H13A	108.9
C4—C5—C6	112.5 (4)	C1—C13—H13B	108.9
C4—C5—H5A	109.1	C12—C13—H13B	108.9
C6—C5—H5A	109.1	H13A—C13—H13B	107.7
C4—C5—H5B	109.1	C3—C14—H14A	109.5
C6—C5—H5B	109.1	C3—C14—H14B	109.5
H5A—C5—H5B	107.8	H14A—C14—H14B	109.5
C5—C6—C7	120.9 (4)	C3—C14—H14C	109.5
C5—C6—H6A	107.1	H14A—C14—H14C	109.5
C7—C6—H6A	107.1	H14B—C14—H14C	109.5
C5—C6—H6B	107.1	C7—C15—H15A	109.5
C7—C6—H6B	107.1	C7—C15—H15B	109.5
H6A—C6—H6B	106.8	H15A—C15—H15B	109.5
C15—C7—C6	110.2 (4)	C7—C15—H15C	109.5
C15—C7—C16	108.2 (4)	H15A—C15—H15C	109.5
C6—C7—C16	104.9 (4)	H15B—C15—H15C	109.5
C15—C7—C8	106.7 (4)	C7—C16—H16A	109.5
C6—C7—C8	114.0 (3)	C7—C16—H16B	109.5
C16—C7—C8	112.8 (4)	H16A—C16—H16B	109.5
C9—C8—C1	113.2 (3)	C7—C16—H16C	109.5
C9—C8—C7	108.4 (3)	H16A—C16—H16C	109.5
C1—C8—C7	113.9 (3)	H16B—C16—H16C	109.5
C9—C8—H8	107.0	C11—C17—H17A	109.5
C1—C8—H8	107.0	C11—C17—H17B	109.5
C7—C8—H8	107.0	H17A—C17—H17B	109.5
C10—C9—C11	59.5 (3)	C11—C17—H17C	109.5
C10—C9—C8	129.4 (3)	H17A—C17—H17C	109.5
C11—C9—C8	122.7 (3)	H17B—C17—H17C	109.5
