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Tetraethylammonium toluene-4-sulfonate

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Key indicators: single-crystal X-ray study; T = 100 K; mean σ (C–C) = 0.005 Å; disorder in main residue; R factor = 0.076; wR factor = 0.163; data-to-parameter ratio = 15.5.

There are two tetraethylammonium cations and two toluene-4-sulfate anions in the asymmetric unit of the title salt, $C_8H_{20}N^+ \cdot C_7H_7O_3S^-$. One of the anions is disordered over two positions, with refined occupancies of 0.447 (3) and 0.553 (3). In the crystal, the cations and anions are linked by $C-H\cdots O$ hydrogen bonds, forming ribbons along [101]. The ribbons are linked *via* $C-H\cdots O$ hydrogen bonds, forming a twodimensional network lying parallel to (101).

Related literature

For the preparation of tetraethylammonium toluene-4-sulfonate from ethyl 4-toluenesulfonate and triethylamine, see: Baizer (1964). For its application as a phase-transfer catalyst, see: Cerveau *et al.* (2002) or as the supporting electrolyte, see: Adachi *et al.* (1979); Wynne & Street (1985); Yoshida *et al.* (1986); Wong & Moeller (1993); Ben *et al.* (2011).

Experimental

Crystal data $C_8H_{20}N^+ \cdot C_7H_7O_3S^ M_r = 301.21$ Monoclinic, $P2_1/n$ a = 16.8771 (3) Å

b = 7.53713 (16) Å c = 26.2404 (6) Å $\beta = 97.2938 (18)^{\circ}$ $V = 3310.90 (12) \text{ Å}^{3}$ Z = 8Mo $K\alpha$ radiation $\mu = 0.20 \text{ mm}^{-1}$

Data collection

Agilent SuperNova (Dual, Cu at zero, Atlas) diffractometer Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2011) $T_{\rm min} = 0.771, T_{\rm max} = 1.000$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.076$ $wR(F^2) = 0.163$ S = 1.196276 reflections 406 parameters T = 100 K $0.8 \times 0.6 \times 0.3 \text{ mm}$

6276 measured reflections 6276 independent reflections 5477 reflections with $I > 2\sigma(I)$ $R_{int} = 0.050$

82 restraints H-atom parameters constrained $\Delta \rho_{max} = 0.46$ e Å⁻³ $\Delta \rho_{min} = -0.50$ e Å⁻³

Table 1		_	
Hydrogen-bond	geometry	(Å,	°).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C6B - H6B \cdots O23$	0.95	2.57	3.351 (6)	140
$C31 - H31B \cdot \cdot \cdot O3B^{i}$	0.99	2.49	3.344 (4)	145
C33-H33A···O2B	0.99	2.47	3.354 (4)	148
$C35-H35A\cdots O22^{ii}$	0.99	2.42	3.228 (4)	138
C36−H36C···O3B ⁱⁱⁱ	0.98	2.58	3.544 (4)	169
C43−H43B···O22	0.99	2.44	3.269 (4)	141
$C45 - H45A \cdots O2B$	0.99	2.53	3.367 (4)	142
$C47 - H47A \cdots O3B^{i}$	0.99	2.57	3.440 (4)	147
$C48-H48B\cdots O22^{iv}$	0.98	2.58	3.562 (4)	175

Symmetry codes: (i) x, y + 1, z; (ii) $x + \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$; (iii) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{3}{2}$; (iv) -x, -y + 1, -z + 1.

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXD* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *OLEX2* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: KP2445).

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Tetraethylammonium toluene-4-sulfonate

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S1. Comment

Tetraethylammonium toluene-4-sulfonate is applied as the phase-transfer catalyst in the preparation of bis-silanetriols (Cerveau *et al.*, 2002). The compound is also widely used in electrochemistry as the supporting electrolyte (Wynne *et al.*, 1985; Yoshida *et al.*, 1986; Wong *et al.*, 1993; Ben *et al.*, 2011), because it could be easly removed from the reaction by the extraction with water (Adachi *et al.*, 1979).

The asymmetric unit contains two tetraethylammonium cations and two toluene-4-sulfate anions (Fig. 1). One of the toluene-4-sulfate ions is disordered and is modeled in the two locations. The occupancy of two major positions in the final model is refined to 0.447 (3) and 0.553 (3). Within the crystal lattice the columns of cations and anions are formed along b and ac directions (Figs. 2 and 3, respectively).

S2. Experimental

The title compound was prepared according to the procedure described by Baizer (1964). Briefly, ethyl toluene-4-sulfonate (200 g, 1.0 mole) was dissolved in 100 mL of anhydrous ethanol and triethylamine was added (101 g, 1.0 mole). The reaction mixture was stirred and heated under reflux for 6 h. The excess of triethylamine and ethanol was removed *in vacuo*. The crude product was washed several times with a dry ethylether and recrystallized from ethanol.

S3. Refinement

The disordered toluene-4-sulfate anion is modeled at the two locations with geometric (FLAT instruction) and dispacement parameter (SIMU instruction) restraints and with AFIX 66, EADP and EXYZ constraints. Seven reflections for which I(obs) and I(calc) differed more then 10 times SigmaW were ommited from the refinement. All H atoms were initially located in electron density difference maps. Hydrogen atoms were constrained to idealised positions with C—H distances fixed at 0.95–0.99 Å and $1.5U_{eq}(C)$ for methyl hydrogen atoms and $1.2U_{eq}(C)$ for others.



Figure 1

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level. For clarity, only more populated location of the disordered anion (B) is shown.



Figure 2

Crystal packing viewed along b direction. For clarity, hydrogen atoms are ommited.



Figure 3

Crystal packing viewed along ac direction. For clarity, hydrogen atoms are ommited.

Tetraethylammonium toluene-4-sulfonate

Crystal data

C₈H₂₀N⁺·C₇H₇O₃S⁻ $M_r = 301.21$ Monoclinic, $P2_1/n$ Hall symbol: -P 2yn a = 16.8771 (3) Å b = 7.53713 (16) Å c = 26.2404 (6) Å $\beta = 97.2938$ (18)° V = 3310.90 (12) Å³ Z = 8

Data collection

Agilent SuperNova (Dual, Cu at zero, Atlas) diffractometer Radiation source: SuperNova (Mo) X-ray Source Mirror monochromator Detector resolution: 10.4052 pixels mm⁻¹ ω scans Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2011)

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.076$ $wR(F^2) = 0.163$ S = 1.196276 reflections 406 parameters 82 restraints Primary atom site location: structure-invariant direct methods F(000) = 1312 $D_x = 1.209 \text{ Mg m}^{-3}$ Mo K\alpha radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 9780 reflections $\theta = 2.6-25.6^{\circ}$ $\mu = 0.20 \text{ mm}^{-1}$ T = 100 KPlate, colourless $0.8 \times 0.6 \times 0.3 \text{ mm}$

 $T_{\min} = 0.771, T_{\max} = 1.000$ 6276 measured reflections 6276 independent reflections 5477 reflections with $I > 2\sigma(I)$ $R_{int} = 0.050$ $\theta_{\max} = 25.7^{\circ}, \theta_{\min} = 2.7^{\circ}$ $h = -20 \rightarrow 20$ $k = 0 \rightarrow 9$ $l = 0 \rightarrow 31$

Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0161P)^2 + 13.1727P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.46$ e Å⁻³ $\Delta\rho_{min} = -0.50$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

	x	у	Z	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
C1A	0.4305 (4)	0.3064 (10)	0.48865 (18)	0.0178 (18)	0.447 (3)
C2A	0.4856 (3)	0.3393 (9)	0.5316 (2)	0.0227 (19)	0.447 (3)
H2A	0.5390	0.3704	0.5276	0.027*	0.447 (3)
C3A	0.4626 (4)	0.3266 (9)	0.58052 (18)	0.0182 (18)	0.447 (3)
H3A	0.5002	0.3491	0.6099	0.022*	0.447 (3)
C4A	0.3845 (4)	0.2811 (15)	0.5864 (3)	0.019 (4)	0.447 (3)
C5A	0.3293 (3)	0.2482 (18)	0.5434 (4)	0.0163 (8)	0.447 (3)
H5A	0.2760	0.2171	0.5475	0.020*	0.447 (3)
C6A	0.3524 (3)	0.2609 (15)	0.4946 (3)	0.017 (4)	0.447 (3)
H6A	0.3147	0.2384	0.4652	0.021*	0.447 (3)
C7A	0.4550 (6)	0.3225 (15)	0.4349 (4)	0.032 (2)	0.447 (3)
H7AA	0.4883	0.2208	0.4281	0.048*	0.447 (3)
H7AB	0.4853	0.4324	0.4324	0.048*	0.447 (3)
H7AC	0.4071	0.3248	0.4095	0.048*	0.447 (3)
O1A	0.41953 (15)	0.2857 (4)	0.68544 (10)	0.0208 (6)	0.447 (3)
O2A	0.29813 (16)	0.4298 (3)	0.64542 (10)	0.0208 (6)	0.447 (3)
O3A	0.30363 (16)	0.1087 (3)	0.65189 (10)	0.0208 (6)	0.447 (3)
S1A	0.34859 (5)	0.27202 (12)	0.64777 (3)	0.0152 (2)	0.447 (3)
C1B	0.4338 (4)	0.1902 (9)	0.49086 (17)	0.0330 (19)	0.553 (3)
C2B	0.4878 (3)	0.1764 (9)	0.5353 (2)	0.039 (2)	0.553 (3)
H2B	0.5421	0.1482	0.5330	0.047*	0.553 (3)
C3B	0.4624 (3)	0.2038 (10)	0.58303 (17)	0.033 (2)	0.553 (3)
H3B	0.4993	0.1943	0.6134	0.039*	0.553 (3)
C4B	0.3829 (4)	0.2450 (13)	0.5863 (2)	0.016 (3)	0.553 (3)
C5B	0.3289 (3)	0.2588 (15)	0.5419 (3)	0.0163 (8)	0.55
H5B	0.2746	0.2870	0.5442	0.020*	0.553 (3)
C6B	0.3543 (3)	0.2314 (13)	0.4942 (2)	0.027 (4)	0.553 (3)
H6B	0.3174	0.2409	0.4638	0.033*	0.553 (3)
C7B	0.4610 (6)	0.1629 (15)	0.4382 (3)	0.046 (2)	0.553 (3)
H7BA	0.4996	0.0651	0.4399	0.069*	0.553 (3)
H7BB	0.4863	0.2717	0.4277	0.069*	0.553 (3)
H7BC	0.4148	0.1342	0.4130	0.069*	0.553 (3)
O1B	0.41953 (15)	0.2857 (4)	0.68544 (10)	0.0208 (6)	0.55
O2B	0.29813 (16)	0.4298 (3)	0.64542 (10)	0.0208 (6)	0.55
O3B	0.30363 (16)	0.1087 (3)	0.65189 (10)	0.0208 (6)	0.55

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\hat{A}^2)

S1B	0.34859 (5)	0.27202 (12)	0.64777 (3)	0.0152 (2)	0.55
C41	0.1861 (2)	0.9052 (5)	0.48107 (14)	0.0197 (8)	
H41A	0.2323	0.9814	0.4935	0.024*	
H41B	0.1387	0.9830	0.4741	0.024*	
C42	0.2016 (3)	0.8186 (6)	0.43110 (15)	0.0294 (10)	
H42A	0.2035	0.9099	0.4047	0.044*	
H42B	0.1586	0.7344	0.4200	0.044*	
H42C	0.2528	0.7555	0.4363	0.044*	
C43	0.0994 (2)	0.6613 (5)	0.50777 (14)	0.0198 (8)	
H43A	0.0908	0.5841	0.5371	0.024*	
H43B	0.1105	0.5834	0.4791	0.024*	
C44	0.0231 (2)	0.7642 (5)	0.49120 (15)	0.0233 (8)	
H44A	0.0298	0.8362	0.4609	0.035*	
H44B	0.0115	0.8419	0 5193	0.035*	
H44C	-0.0212	0.6810	0.4827	0.035*	
C45	0.0212 0.2424(2)	0.6505 (5)	0.53511(14)	0.0196 (8)	
U45 H45Δ	0.2424(2) 0.2318	0.5717	0.5637	0.024*	
114JA 11450	0.2318	0.5740	0.5037	0.024	
1145D C46	0.2437 0.3226 (2)	0.3749	0.5040	0.024	
U40	0.3220 (2)	0.7390 (0)	0.54957 (17)	0.0284 (9)	
П40А 1146D	0.3301	0.8090	0.5204	0.043	
	0.3030	0.0490	0.5385	0.043*	
H40C	0.5197	0.81/8	0.5791	0.043^{+}	
C4/	0.1613 (2)	0.8873 (5)	0.57032(13)	0.01/5 (8)	
H47A	0.2102	0.9586	0.5/98	0.021*	
H4/B	0.1166	0.9710	0.5610	0.021*	
C48	0.1444 (2)	0.7823 (5)	0.61699 (14)	0.0194 (8)	
H48A	0.1860	0.6922	0.6250	0.029*	
H48B	0.0922	0.7243	0.6098	0.029*	
H48C	0.1441	0.8626	0.6463	0.029*	
N41	0.17227 (17)	0.7763 (4)	0.52337 (11)	0.0143 (6)	
C31	0.4385 (2)	0.8762 (5)	0.72905 (13)	0.0171 (8)	
H31A	0.4868	0.9464	0.7413	0.021*	
H31B	0.3936	0.9605	0.7214	0.021*	
C32	0.4520 (3)	0.7833 (6)	0.67967 (15)	0.0260 (9)	
H32A	0.5004	0.7105	0.6857	0.039*	
H32B	0.4583	0.8720	0.6532	0.039*	
H32C	0.4061	0.7074	0.6682	0.039*	
C33	0.3440 (2)	0.6490 (5)	0.75647 (13)	0.0148 (7)	
H33A	0.3527	0.5692	0.7277	0.018*	
H33B	0.3333	0.5737	0.7858	0.018*	
C34	0.2712 (2)	0.7617 (5)	0.74017 (14)	0.0178 (8)	
H34A	0.2599	0.8362	0.7690	0.027*	
H34B	0.2252	0.6848	0.7296	0.027*	
H34C	0.2811	0.8374	0.7113	0.027*	
C35	0.4117 (2)	0.8715 (5)	0.81795 (13)	0.0174 (8)	
H35A	0.4618	0.9397	0.8264	0.021*	
H35B	0.3684	0.9580	0.8080	0.021*	
C36	0.3941 (2)	0.7765 (5)	0.86586 (14)	0.0196 (8)	
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H36A	0.3916	0.8629	0.8935	0.029*
H36B	0.4364	0.6901	0.8763	0.029*
H36C	0.3427	0.7149	0.8589	0.029*
C37	0.4863 (2)	0.6190 (5)	0.78468 (15)	0.0181 (8)
H37A	0.4884	0.5427	0.7542	0.022*
H37B	0.4723	0.5424	0.8129	0.022*
C38	0.5688 (2)	0.6967 (6)	0.80046 (17)	0.0284 (9)
H38A	0.5670	0.7768	0.8297	0.043*
H38B	0.5858	0.7628	0.7716	0.043*
H38C	0.6068	0.6006	0.8103	0.043*
N31	0.42024 (17)	0.7550 (4)	0.77214 (11)	0.0140 (6)
C21	0.1825 (2)	0.3831 (5)	0.24019 (15)	0.0222 (8)
C22	0.1070 (2)	0.3169 (5)	0.24493 (14)	0.0173 (8)
H22	0.0717	0.2895	0.2148	0.021*
C23	0.0824 (2)	0.2902 (5)	0.29249 (14)	0.0156 (7)
H23	0.0303	0.2460	0.2947	0.019*
C24	0.1334 (2)	0.3274 (4)	0.33731 (14)	0.0140 (7)
C25	0.2094 (2)	0.3924 (5)	0.33320 (15)	0.0204 (8)
H25	0.2450	0.4174	0.3633	0.025*
C26	0.2332 (2)	0.4210 (6)	0.28522 (16)	0.0264 (9)
H26	0.2849	0.4672	0.2829	0.032*
C27	0.2089 (3)	0.4133 (6)	0.18794 (16)	0.0331 (10)
H27A	0.1626	0.4449	0.1633	0.050*
H27B	0.2333	0.3046	0.1766	0.050*
H27C	0.2481	0.5100	0.1901	0.050*
O21	0.05965 (16)	0.1235 (3)	0.39491 (10)	0.0215 (6)
O22	0.04246 (15)	0.4433 (3)	0.40172 (9)	0.0194 (6)
O23	0.16880 (16)	0.3076 (4)	0.43637 (10)	0.0235 (6)
S21	0.09844 (5)	0.29743 (12)	0.39789 (3)	0.0154 (2)

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1A	0.019 (4)	0.016 (4)	0.019 (4)	0.009 (4)	0.005 (3)	0.004 (4)
C2A	0.015 (4)	0.033 (5)	0.021 (4)	0.002 (4)	0.005 (3)	-0.003 (4)
C3A	0.016 (4)	0.010 (4)	0.027 (5)	0.004 (3)	-0.005 (3)	-0.002 (4)
C4A	0.022 (7)	0.019 (6)	0.015 (7)	-0.003 (4)	-0.003 (5)	-0.005 (4)
C5A	0.0181 (18)	0.012 (2)	0.019 (2)	-0.0004 (14)	0.0032 (15)	0.0023 (15)
C6A	0.023 (8)	0.014 (5)	0.011 (7)	0.008 (5)	-0.008 (6)	-0.001 (4)
C7A	0.030 (5)	0.051 (7)	0.018 (5)	0.003 (5)	0.014 (4)	0.006 (4)
O1A	0.0218 (14)	0.0225 (14)	0.0168 (13)	0.0000 (11)	-0.0033 (11)	0.0001 (11)
O2A	0.0254 (14)	0.0151 (13)	0.0217 (14)	0.0038 (11)	0.0027 (11)	0.0016 (11)
O3A	0.0260 (14)	0.0123 (13)	0.0239 (14)	-0.0019 (11)	0.0020 (11)	-0.0002 (11)
S1A	0.0177 (5)	0.0139 (4)	0.0137 (5)	0.0012 (3)	0.0005 (3)	0.0003 (3)
C1B	0.038 (5)	0.035 (5)	0.027 (4)	-0.012 (4)	0.009 (3)	-0.008 (4)
C2B	0.025 (4)	0.060 (6)	0.034 (5)	-0.002 (4)	0.006 (3)	-0.005 (4)
C3B	0.018 (4)	0.059 (6)	0.020 (4)	0.002 (4)	0.000 (3)	0.003 (4)
C4B	0.019 (6)	0.010 (4)	0.019 (6)	0.000 (3)	0.003 (4)	0.008 (3)

C5B	0.0181 (18)	0.012 (2)	0.019 (2)	-0.0004 (14)	0.0032 (15)	0.0023 (15)
C6B	0.034 (8)	0.024 (5)	0.025 (8)	-0.013 (5)	0.008 (6)	0.000 (4)
C7B	0.042 (5)	0.070 (7)	0.029 (5)	-0.009 (5)	0.012 (4)	-0.009 (5)
O1B	0.0218 (14)	0.0225 (14)	0.0168 (13)	0.0000 (11)	-0.0033 (11)	0.0001 (11)
O2B	0.0254 (14)	0.0151 (13)	0.0217 (14)	0.0038 (11)	0.0027 (11)	0.0016 (11)
O3B	0.0260 (14)	0.0123 (13)	0.0239 (14)	-0.0019 (11)	0.0020 (11)	-0.0002 (11)
S1B	0.0177 (5)	0.0139 (4)	0.0137 (5)	0.0012 (3)	0.0005 (3)	0.0003 (3)
C41	0.0232 (19)	0.0195 (19)	0.0164 (19)	0.0010 (16)	0.0024 (15)	0.0068 (15)
C42	0.039 (2)	0.032 (2)	0.019 (2)	0.0084 (19)	0.0101 (18)	0.0072 (17)
C43	0.0232 (19)	0.0184 (19)	0.0166 (19)	-0.0037 (16)	-0.0019 (15)	-0.0031 (15)
C44	0.0212 (19)	0.027 (2)	0.020 (2)	-0.0043 (16)	-0.0015 (15)	-0.0027 (16)
C45	0.0230 (19)	0.0193 (19)	0.0165 (19)	0.0055 (16)	0.0029 (15)	0.0014 (15)
C46	0.019 (2)	0.030 (2)	0.036 (2)	0.0045 (17)	-0.0011 (17)	0.0095 (19)
C47	0.0221 (19)	0.0155 (18)	0.0141 (18)	0.0006 (15)	-0.0005 (14)	-0.0023 (14)
C48	0.026 (2)	0.0190 (19)	0.0135 (18)	0.0012 (16)	0.0025 (15)	-0.0004 (15)
N41	0.0164 (15)	0.0150 (15)	0.0112 (15)	0.0019 (12)	0.0000 (12)	0.0002 (12)
C31	0.0194 (18)	0.0154 (18)	0.0161 (18)	-0.0038 (14)	0.0008 (14)	0.0052 (14)
C32	0.034 (2)	0.027 (2)	0.019 (2)	-0.0003 (18)	0.0076 (17)	0.0053 (17)
C33	0.0171 (17)	0.0138 (17)	0.0131 (17)	-0.0049 (14)	0.0003 (14)	-0.0016 (14)
C34	0.0153 (17)	0.0177 (19)	0.0197 (19)	-0.0031 (14)	-0.0003 (14)	0.0022 (15)
C35	0.0247 (19)	0.0124 (17)	0.0138 (18)	-0.0006 (15)	-0.0022 (14)	-0.0035 (14)
C36	0.025 (2)	0.0194 (19)	0.0134 (18)	-0.0016 (16)	0.0002 (15)	-0.0023 (15)
C37	0.0175 (18)	0.0132 (18)	0.023 (2)	0.0022 (14)	-0.0009 (15)	0.0030 (15)
C38	0.020 (2)	0.024 (2)	0.039 (3)	-0.0002 (17)	-0.0049 (18)	0.0075 (19)
N31	0.0175 (15)	0.0100 (14)	0.0139 (15)	-0.0013 (12)	-0.0013 (12)	-0.0005 (12)
C21	0.024 (2)	0.0189 (19)	0.026 (2)	0.0070 (16)	0.0124 (16)	0.0057 (16)
C22	0.0203 (18)	0.0142 (18)	0.0168 (18)	0.0034 (15)	0.0004 (14)	-0.0006 (14)
C23	0.0152 (17)	0.0109 (17)	0.0205 (19)	-0.0004 (14)	0.0020 (14)	-0.0003 (14)
C24	0.0150 (17)	0.0085 (16)	0.0189 (18)	0.0041 (13)	0.0034 (14)	-0.0017 (13)
C25	0.0157 (18)	0.022 (2)	0.023 (2)	0.0026 (15)	-0.0009 (15)	-0.0019 (16)
C26	0.0158 (18)	0.031 (2)	0.034 (2)	-0.0025 (16)	0.0065 (16)	0.0013 (18)
C27	0.033 (2)	0.041 (3)	0.028 (2)	0.004 (2)	0.0165 (19)	0.007 (2)
O21	0.0262 (14)	0.0200 (14)	0.0192 (14)	-0.0025 (11)	0.0066 (11)	0.0029 (11)
O22	0.0215 (13)	0.0209 (14)	0.0155 (13)	0.0042 (11)	0.0016 (10)	-0.0017 (11)
O23	0.0229 (14)	0.0278 (15)	0.0179 (14)	0.0058 (12)	-0.0043 (11)	0.0020 (11)
S21	0.0169 (4)	0.0156 (4)	0.0132 (4)	0.0035 (3)	-0.0005 (3)	0.0005 (3)

Geometric parameters (Å, °)

C1A—C2A	1.3900	C47—C48	1.515 (5)	
C1AC6A	1.3900	C47—N41	1.520 (4)	
C1A—C7A	1.525 (10)	C47—H47A	0.9900	
C2A—C3A	1.3900	C47—H47B	0.9900	
C2A—H2A	0.9500	C48—H48A	0.9800	
C3A—C4A	1.3900	C48—H48B	0.9800	
СЗА—НЗА	0.9500	C48—H48C	0.9800	
C4A—C5A	1.3900	C31—N31	1.515 (4)	
C4A—S1A	1.792 (5)	C31—C32	1.516 (5)	

C5A—C6A	1.3900	C31—H31A	0.9900
С5А—Н5А	0.9500	C31—H31B	0.9900
C6A—H6A	0.9500	C32—H32A	0.9800
C7A—H7AA	0 9800	C32—H32B	0 9800
C7A—H7AB	0.9800	$C_{32}$ H32D	0.9800
C7A - H7AC	0.9800	$C_{33}$ $C_{34}$	1.511(5)
014 $11/100$	1 456 (3)	$C_{33}$ N31	1.511(3) 1.526(4)
$O_{2A} = S_{1A}$	1.450 (3)	C33 H33A	0.0000
$O_2A = S_1A$	1.400(3) 1.457(3)	C33 H33P	0.9900
C1P C2P	1.457 (5)	C24 H24A	0.9900
C1B - C2B	1.3900	$C_{24}$ $H_{24}$	0.9800
	1.5900	C24 H24C	0.9800
	1.526 (9)	C34—H34C	0.9800
C2B—C3B	1.3900	C35—C36	1.509 (5)
C2B—H2B	0.9500	C35—N31	1.510 (4)
C3B—C4B	1.3900	С35—Н35А	0.9900
СЗВ—НЗВ	0.9500	C35—H35B	0.9900
C4B—C5B	1.3900	С36—Н36А	0.9800
C5B—C6B	1.3900	С36—Н36В	0.9800
C5B—H5B	0.9500	C36—H36C	0.9800
C6B—H6B	0.9500	C37—N31	1.519 (4)
С7В—Н7ВА	0.9800	C37—C38	1.518 (5)
C7B—H7BB	0.9800	С37—Н37А	0.9900
C7B—H7BC	0.9800	С37—Н37В	0.9900
C41—N41	1.516 (4)	C38—H38A	0.9800
C41—C42	1.517 (5)	C38—H38B	0.9800
C41—H41A	0.9900	C38—H38C	0.9800
C41—H41B	0.9900	C21—C22	1.389 (5)
C42—H42A	0.9800	C21—C26	1.397 (6)
C42—H42B	0.9800	C21—C27	1.512 (5)
C42—H42C	0.9800	C22—C23	1.379 (5)
C43—N41	1.517 (5)	C22—H22	0.9500
C43—C44	1.519 (5)	C23—C24	1.395 (5)
C43—H43A	0.9900	C23—H23	0.9500
C43—H43B	0.9900	$C_{24}$ $C_{25}$	1 389 (5)
C44—H44A	0.9800	$C_{24}$ $S_{21}$	1.369(3) 1 779(4)
C44—H44B	0.9800	$C_{24} = 521$	1.775 (4)
C44—H44C	0.9800	C25 C20	0.9500
$C_{45}$ $C_{46}$	1 515 (5)	C26 H26	0.9500
C45 N41	1.515(3) 1.517(4)	C27 H27A	0.9500
$C_{45}$ $H_{45A}$	0.0000	$C_2 / - H_2 / A$	0.9800
C45 = H45R	0.9900	$C_2/-n_2/B$	0.9800
С45—Н45В	0.9900	$C_2/-H_2/C$	0.9800
С40—П40А С4(Ц4(D	0.9800	022 521	1.403(3)
	0.9800	022	1.401 (3)
C46—H46C	0.9800	023—S21	1.460 (3)
C2A—C1A—C6A	120.0	H48A—C48—H48C	109.5
C2A—C1A—C7A	120.3 (6)	H48B—C48—H48C	109.5
C6A—C1A—C7A	119.7 (6)	C41—N41—C43	111.3 (3)

C3A—C2A—C1A	120.0	C41—N41—C45	111.1 (3)
C3A—C2A—H2A	120.0	C43—N41—C45	106.4 (3)
C1A—C2A—H2A	120.0	C41—N41—C47	106.7 (3)
C2A—C3A—C4A	120.0	C43—N41—C47	110.8 (3)
С2А—С3А—НЗА	120.0	C45—N41—C47	110.6 (3)
С4А—С3А—НЗА	120.0	N31—C31—C32	115.2 (3)
C5A—C4A—C3A	120.0	N31—C31—H31A	108.5
C5A—C4A—S1A	117.0 (5)	С32—С31—Н31А	108.5
C3A—C4A—S1A	122.9 (5)	N31—C31—H31B	108.5
C4A—C5A—C6A	120.0	С32—С31—Н31В	108.5
С4А—С5А—Н5А	120.0	H31A—C31—H31B	107.5
С6А—С5А—Н5А	120.0	C31—C32—H32A	109.5
C5A - C6A - C1A	120.0	C31—C32—H32B	109.5
C5A—C6A—H6A	120.0	H32A—C32—H32B	109.5
C1A - C6A - H6A	120.0	$C_{31} - C_{32} - H_{32}C_{32}$	109.5
01A = S1A = 03A	113 61 (16)	$H_{32A} - C_{32} - H_{32C}$	109.5
014 - S14 - 024	113 33 (16)	$H_{32B} = C_{32} = H_{32C}$	109.5
034 - S14 - 024	112 64 (15)	$C_{34}$ $C_{33}$ $N_{31}$	107.3 114 2 (3)
014 $14$ $-C44$	105.4(3)	$C_{34}$ $C_{33}$ $H_{33}$	108 7
$O_{3A}$ S1A C4A	109.7(3)	N31 C33 H33A	108.7
03A = 51A = C4A	109.7 (4)	$C_{34}$ $C_{33}$ $H_{33B}$	108.7
$C_{2R} = C_{1R} = C_{4R}$	120.0	N31 C33 H33B	108.7
$C_{2B}$ $C_{1B}$ $C_{7B}$	120.6 (5)	H22A C22 H22P	107.6
$C_{2B}$ $C_{1B}$ $C_{7B}$	120.0(5)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	107.0
$C_{0B}$ $C_{1B}$ $C_{1B}$ $C_{1B}$	119.4 (3)	$C_{33} = C_{34} = H_{34} R_{34}$	109.5
$C_{2B}$ $C_{2B}$ $C_{2B}$ $C_{2B}$ $C_{2B}$	120.0	$C_{33}$ — $C_{34}$ — $H_{34B}$	109.5
$C_{3}B = C_{2}B = H_{2}B$	120.0	$H_{34A} - C_{34} - H_{34B}$	109.5
CIB - C2B - H2B	120.0	U24A C24 H24C	109.5
C4B = C3B = C2B	120.0	H34A - C34 - H34C	109.5
C4B - C3B - H3B	120.0	H34B - C34 - H34C	109.5
$C_2B = C_3B = H_3B$	120.0	$C_{36} = C_{35} = N_{31}$	115.8 (3)
C3B—C4B—C5B	120.0	C36—C35—H35A	108.3
C6B—C5B—C4B	120.0	N31—C35—H35A	108.3
С6В—С5В—Н5В	120.0	С36—С35—Н35В	108.3
C4B—C5B—H5B	120.0	N31—C35—H35B	108.3
C5B—C6B—C1B	120.0	H35A—C35—H35B	107.4
C5B—C6B—H6B	120.0	С35—С36—Н36А	109.5
C1B—C6B—H6B	120.0	С35—С36—Н36В	109.5
C1B—C7B—H7BA	109.5	H36A—C36—H36B	109.5
C1B—C7B—H7BB	109.5	С35—С36—Н36С	109.5
H7BA—C7B—H7BB	109.5	H36A—C36—H36C	109.5
C1B—C7B—H7BC	109.5	H36B—C36—H36C	109.5
Н7ВА—С7В—Н7ВС	109.5	N31—C37—C38	114.9 (3)
H7BB—C7B—H7BC	109.5	N31—C37—H37A	108.5
N41—C41—C42	114.6 (3)	С38—С37—Н37А	108.5
N41—C41—H41A	108.6	N31—C37—H37B	108.5
C42—C41—H41A	108.6	С38—С37—Н37В	108.5
N41—C41—H41B	108.6	Н37А—С37—Н37В	107.5
C42—C41—H41B	108.6	C37—C38—H38A	109.5

H41A—C41—H41B	107.6	С37—С38—Н38В	109.5
C41—C42—H42A	109.5	H38A—C38—H38B	109.5
C41—C42—H42B	109.5	С37—С38—Н38С	109.5
H42A—C42—H42B	109.5	H38A—C38—H38C	109.5
C41—C42—H42C	109.5	H38B—C38—H38C	109.5
H42A—C42—H42C	109.5	C35—N31—C31	107.0 (3)
H42B—C42—H42C	109.5	C35—N31—C37	111.2 (3)
N41—C43—C44	114.4 (3)	C31—N31—C37	110.8 (3)
N41—C43—H43A	108.6	C35—N31—C33	110.9 (3)
C44—C43—H43A	108.6	C31—N31—C33	111.0 (3)
N41—C43—H43B	108.6	C37—N31—C33	106.0 (3)
C44—C43—H43B	108.6	C22—C21—C26	117.9 (3)
H43A—C43—H43B	107.6	C22—C21—C27	121.0 (4)
C43—C44—H44A	109.5	C26—C21—C27	121.2 (4)
C43—C44—H44B	109.5	C23—C22—C21	121.2 (3)
H44A—C44—H44B	109.5	C23—C22—H22	119.4
C43—C44—H44C	109.5	C21—C22—H22	119.4
H44A—C44—H44C	109.5	C22—C23—C24	120.6 (3)
H44B—C44—H44C	109.5	С22—С23—Н23	119.7
C46—C45—N41	115.0 (3)	С24—С23—Н23	119.7
C46—C45—H45A	108.5	C25—C24—C23	118.8 (3)
N41—C45—H45A	108.5	C25—C24—S21	121.9 (3)
C46—C45—H45B	108.5	C23—C24—S21	119.3 (3)
N41—C45—H45B	108.5	C26—C25—C24	120.1 (3)
H45A—C45—H45B	107.5	С26—С25—Н25	119.9
C45—C46—H46A	109.5	С24—С25—Н25	119.9
C45—C46—H46B	109.5	C25—C26—C21	121.3 (4)
H46A—C46—H46B	109.5	С25—С26—Н26	119.3
C45—C46—H46C	109.5	C21—C26—H26	119.3
H46A—C46—H46C	109.5	С21—С27—Н27А	109.5
H46B—C46—H46C	109.5	С21—С27—Н27В	109.5
C48—C47—N41	115.0 (3)	H27A—C27—H27B	109.5
C48—C47—H47A	108.5	С21—С27—Н27С	109.5
N41—C47—H47A	108.5	H27A—C27—H27C	109.5
C48—C47—H47B	108.5	H27B—C27—H27C	109.5
N41—C47—H47B	108.5	O23—S21—O22	112.92 (16)
H47A—C47—H47B	107.5	O23—S21—O21	113.78 (16)
C47—C48—H48A	109.5	O22—S21—O21	112.88 (16)
C47—C48—H48B	109.5	O23—S21—C24	106.16 (16)
H48A—C48—H48B	109.5	O22—S21—C24	104.88 (15)
C47—C48—H48C	109.5	O21—S21—C24	105.23 (16)

## Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D····A	<i>D</i> —H··· <i>A</i>
С6В—Н6В…О23	0.95	2.57	3.351 (6)	140
C31—H31 $B$ ···O3 $B^{i}$	0.99	2.49	3.344 (4)	145
C33—H33 <i>A</i> ···O2 <i>B</i>	0.99	2.47	3.354 (4)	148

# supporting information

C35—H35A···O22 ⁱⁱ	0.99	2.42	3.228 (4)	138
C36—H36 <i>C</i> ···O3 <i>B</i> ⁱⁱⁱ	0.98	2.58	3.544 (4)	169
C43—H43 <i>B</i> ···O22	0.99	2.44	3.269 (4)	141
C45—H45 <i>A</i> ···O2 <i>B</i>	0.99	2.53	3.367 (4)	142
C47—H47 $A$ ···O3 $B^{i}$	0.99	2.57	3.440 (4)	147
C48—H48 <i>B</i> ···O22 ^{iv}	0.98	2.58	3.562 (4)	175

Symmetry codes: (i) x, y+1, z; (ii) x+1/2, -y+3/2, z+1/2; (iii) -x+1/2, y+1/2, -z+3/2; (iv) -x, -y+1, -z+1.