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(E)-1-(4-Decyloxyphenyl)-3-(2-hydroxyphenyl)prop-2-en-1-one

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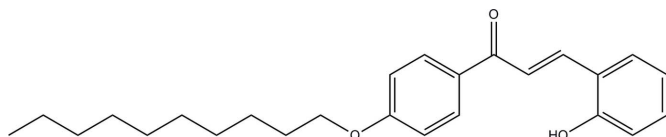
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 Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.001$ Å; R factor = 0.050; wR factor = 0.160; data-to-parameter ratio = 32.7.

In the title compound, $\text{C}_{25}\text{H}_{32}\text{O}_3$, the enone group adopts an *s-cis* conformation. The alkoxy chain is in an all-*trans* conformation. The dihedral angle between the benzene rings is $7.86(5)^\circ$. In the crystal, molecules are connected by pairs of $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds, forming inversion dimers and giving $R_2^2(10)$ rings. Within these dimers, weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds form two $R_2^2(7)$ rings. In the crystal, the approximately planar molecules [largest deviation for an atom being $0.4737(12)$ Å for the terminal C atom of the alkoxy chain] are arranged in sheets parallel to $(20\bar{1})$. Weak $\text{C}-\text{H}\cdots\pi$ interactions are also observed.

Related literature

For the biological properties of chalcone derivatives, see: Bhat *et al.* (2005); Xue *et al.* (2004); Won *et al.* (2005); Zhao *et al.* (2005); Satyanarayana *et al.* (2004). For related structures, see: Razak *et al.* (2009); Ngaini *et al.* (2010, 2011). For graph-set theory, see: Bernstein *et al.* (1995). For the stability of the temperature controller used in the data collection, see: Cosier & Glazer (1986). For standard bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{25}\text{H}_{32}\text{O}_3$
 $M_r = 380.51$
 Triclinic, $P\bar{1}$
 $a = 8.6674(13)$ Å
 $b = 10.9865(17)$ Å
 $c = 12.1352(19)$ Å
 $\alpha = 74.405(3)^\circ$
 $\beta = 72.891(3)^\circ$
 $\gamma = 85.981(4)^\circ$
 $V = 1063.7(3)$ Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 100$ K
 $0.62 \times 0.15 \times 0.14$ mm

Data collection

Bruker APEXII DUO CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2009)
 $T_{\min} = 0.955$, $T_{\max} = 0.990$
 30192 measured reflections
 8434 independent reflections
 6213 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.050$
 $wR(F^2) = 0.160$
 $S = 1.02$
 8434 reflections
 258 parameters
 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.63$ e Å⁻³
 $\Delta\rho_{\min} = -0.20$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C10–C15 ring.

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|---|--------------|--------------------|-------------|----------------------|
| $\text{O1}-\text{H1O1}\cdots\text{O2}^i$ | 0.939 (18) | 1.789 (18) | 2.6867 (12) | 159.0 (16) |
| $\text{C7}-\text{H7A}\cdots\text{O1}^i$ | 0.93 | 2.31 | 3.2169 (13) | 164 |
| $\text{C22}-\text{H22A}\cdots\text{Cg1}^{ii}$ | 0.97 | 2.85 | 3.6887 (12) | 146 |

 Symmetry codes: (i) $-x + 2, -y + 1, -z + 1$; (ii) $x, y, z - 1$.

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5525).

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supporting information

Acta Cryst. (2012). E68, o2935–o2936 [https://doi.org/10.1107/S160053681203872X]

(E)-1-(4-Decyloxyphenyl)-3-(2-hydroxyphenyl)prop-2-en-1-one

Zainab Ngaini, Siti Muhaini Haris Fadzillah, Hasnain Hussain, Ibrahim Abdul Razak and Safra Izuani Jama Asik

S1. Comment

Chalcones have displayed an impressive array of biological activities and are extensively reported (Xue *et al.*, 2004; Bhat *et al.*, 2005; Won *et al.*, 2005; Zhao *et al.*, 2005 and Satyanarayana *et al.*, 2004). Here in, we report the crystal structure of the title compound (I).

In (I), Fig. 1, the bond lengths observed are comparable with standard reported values (Allen *et al.*, 1987). The enone (O2/C7–C9) moiety adopts a *s-cis* conformation with a torsion angle of -6.41 (15)°. The mean plane through the enone (O2/C7–C9) moiety form dihedral angles of 2.80 (6) and 7.26 (6)° respectively with the benzene (C1–C6 and C10–C15) rings. The dihedral angle between the two benzene rings is 7.86 (5)°.

An observed widening of the C1–C6–C7 and C6–C7–C8 angles of 122.85 (8) and 126.23 (8)° respectively, may be the consequence of the short contact between H1A and H8A (2.19 Å). Similarly, the slight distortion of the O3–C13–C14 angle to 125.18 (9)° may be the result of a close H14A and H16B (2.37 Å) contact. The geometric parameters are consistent to those observed in closely related structures (Razak *et al.*, 2009; Ngaini *et al.*, 2010; Ngaini *et al.*, 2011).

The alkoxy chain adopts an all *trans* conformation with the difference from the ideal values of the torsion angles ranging from 0.98 (7)° to 7.98 (9)°. The C16–O3–C13–C14 torsion angle of 4.73 (13)° indicate that atoms C16 and O3 and the attached benzene ring are approximately co-planar. The alkoxy chain appears to deviate from co-planarity with the ring for atoms further away i.e. C24 and C25. The dihedral angle between the least-square plane through atoms O3/C16–C25 [maximum deviation = 0.2507 (10)Å at C25] and the attached benzene ring is 15.20 (5)°.

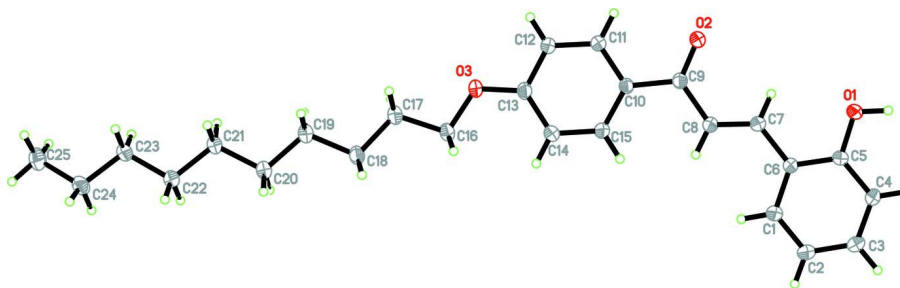
In the crystal (Fig. 2), molecules are connected by a pair of O1—H1O1 \cdots O2ⁱ and a pair of weak C7—H7A \cdots O1ⁱ hydrogen bonds to form a $R^2_2(10)$ ring and two $R^2_2(7)$ rings within inversion dimers. These dimers are arranged into sheets parallel to (20 $\bar{1}$). Weak C—H \cdots π interactions (Table 1) are also observed.

S2. Experimental

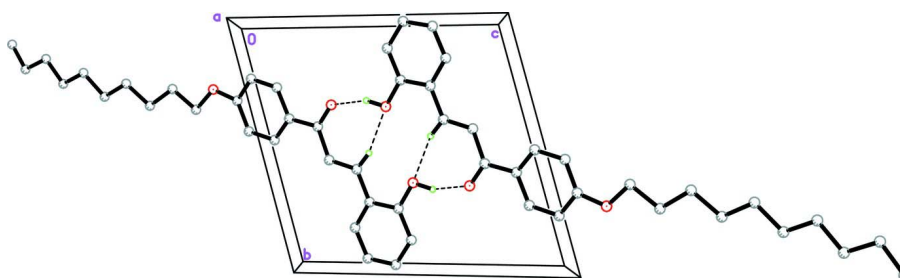
A mixture of 2-hydroxybenzaldehyde (2.44 ml, 20 mmol), 4-decyloxyacetophenone (6.65 g, 20 mmol) and KOH (4.04 g, 72 mmol) in methanol (60 ml) was heated at reflux for 12 h. The reaction was cooled to room temperature and acidified with cold diluted HCl (2N). After redissolving in hexane followed by few days of slow evaporation, crystals were collected.

S3. Refinement

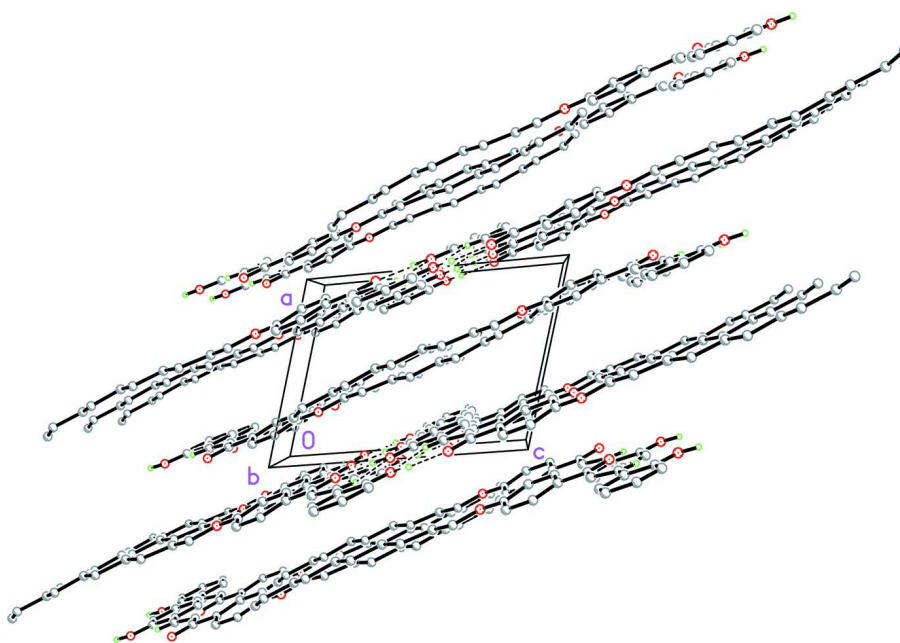
The O-bound H atom was located in a difference Fourier map and refined freely with O–H = 0.943 (17) Å. The remaining H atoms were placed in calculated positions with C–H = 0.93 – 0.97 Å. The U_{iso} values were constrained to be $1.5U_{\text{eq}}$ (methyl-H atom) and $1.2U_{\text{eq}}$ (other H atoms). The rotating model group was applied for the methyl group.

**Figure 1**

The molecular structure of the title compound, showing 50% probability displacement ellipsoids.

**Figure 2**

A pair inversion related molecules connected by hydrogen bonds (dashed lines).

**Figure 3**

The crystal packing, viewed along the *b*-axis, showing the molecules in pairs, arranged into sheets parallel to $(20\bar{1})$ plane. Hydrogen bonds are shown as dashed lines. Only H atoms involved in hydrogen bonds are shown.

(E)-1-(4-Decyloxyphenyl)-3-(2-hydroxyphenyl)prop-2-en-1-one*Crystal data*

| | |
|--------------------------------|---|
| $C_{25}H_{32}O_3$ | $Z = 2$ |
| $M_r = 380.51$ | $F(000) = 412$ |
| Triclinic, $P\bar{1}$ | $D_x = 1.188 \text{ Mg m}^{-3}$ |
| Hall symbol: $-P\ 1$ | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| $a = 8.6674 (13) \text{ \AA}$ | Cell parameters from 7111 reflections |
| $b = 10.9865 (17) \text{ \AA}$ | $\theta = 2.5\text{--}33.7^\circ$ |
| $c = 12.1352 (19) \text{ \AA}$ | $\mu = 0.08 \text{ mm}^{-1}$ |
| $\alpha = 74.405 (3)^\circ$ | $T = 100 \text{ K}$ |
| $\beta = 72.891 (3)^\circ$ | Block, yellow |
| $\gamma = 85.981 (4)^\circ$ | $0.62 \times 0.15 \times 0.14 \text{ mm}$ |
| $V = 1063.7 (3) \text{ \AA}^3$ | |

Data collection

| | |
|---|--|
| Bruker APEXII DUO CCD area-detector diffractometer | 30192 measured reflections |
| Radiation source: fine-focus sealed tube | 8434 independent reflections |
| Graphite monochromator | 6213 reflections with $I > 2\sigma(I)$ |
| φ and ω scans | $R_{\text{int}} = 0.034$ |
| Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2009) | $\theta_{\text{max}} = 33.9^\circ$, $\theta_{\text{min}} = 2.5^\circ$ |
| $T_{\text{min}} = 0.955$, $T_{\text{max}} = 0.990$ | $h = -13 \rightarrow 13$ |
| | $k = -17 \rightarrow 17$ |
| | $l = -18 \rightarrow 18$ |

Refinement

| | |
|--|--|
| Refinement on F^2 | Secondary atom site location: difference Fourier map |
| Least-squares matrix: full | Hydrogen site location: inferred from neighbouring sites |
| $R[F^2 > 2\sigma(F^2)] = 0.050$ | H atoms treated by a mixture of independent and constrained refinement |
| $wR(F^2) = 0.160$ | $w = 1/[\sigma^2(F_o^2) + (0.0957P)^2 + 0.086P]$ |
| $S = 1.02$ | where $P = (F_o^2 + 2F_c^2)/3$ |
| 8434 reflections | $(\Delta/\sigma)_{\text{max}} < 0.001$ |
| 258 parameters | $\Delta\rho_{\text{max}} = 0.63 \text{ e \AA}^{-3}$ |
| 0 restraints | $\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$ |
| Primary atom site location: structure-invariant direct methods | |

Special details

Experimental. The crystal was placed in the cold stream of an Oxford Cryosystems Cobra open-flow nitrogen cryostat (Cosier & Glazer, 1986) operating at 100.0 (1) K.

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|--------------|-------------|-------------|----------------------------------|
| O1 | 0.97173 (10) | 0.64655 (7) | 0.51189 (7) | 0.03236 (17) |

| | | | | |
|------|--------------|-------------|--------------|--------------|
| O2 | 0.94280 (8) | 0.34433 (6) | 0.29484 (6) | 0.02528 (14) |
| O3 | 0.72670 (9) | 0.27611 (6) | −0.13495 (6) | 0.02605 (15) |
| C1 | 0.83637 (12) | 0.82610 (8) | 0.26041 (8) | 0.02406 (17) |
| H1A | 0.8087 | 0.8127 | 0.1961 | 0.029* |
| C2 | 0.82611 (13) | 0.94641 (9) | 0.27668 (9) | 0.0290 (2) |
| H2A | 0.7910 | 1.0128 | 0.2241 | 0.035* |
| C3 | 0.86846 (12) | 0.96808 (9) | 0.37210 (9) | 0.02815 (19) |
| H3A | 0.8630 | 1.0493 | 0.3825 | 0.034* |
| C4 | 0.91859 (11) | 0.86906 (8) | 0.45150 (9) | 0.02482 (17) |
| H4A | 0.9466 | 0.8836 | 0.5152 | 0.030* |
| C5 | 0.92699 (11) | 0.74707 (8) | 0.43560 (8) | 0.02130 (16) |
| C6 | 0.88752 (10) | 0.72389 (8) | 0.33845 (7) | 0.01911 (15) |
| C7 | 0.89920 (10) | 0.59575 (8) | 0.32398 (8) | 0.02062 (16) |
| H7A | 0.9260 | 0.5326 | 0.3836 | 0.025* |
| C8 | 0.87499 (11) | 0.56017 (8) | 0.23288 (8) | 0.02137 (16) |
| H8A | 0.8464 | 0.6213 | 0.1727 | 0.026* |
| C9 | 0.89166 (10) | 0.42876 (8) | 0.22394 (7) | 0.01902 (15) |
| C10 | 0.84626 (10) | 0.39669 (8) | 0.12615 (7) | 0.01838 (15) |
| C11 | 0.87712 (10) | 0.27355 (8) | 0.11183 (8) | 0.02067 (16) |
| H11A | 0.9258 | 0.2158 | 0.1626 | 0.025* |
| C12 | 0.83636 (11) | 0.23708 (8) | 0.02362 (8) | 0.02239 (16) |
| H12A | 0.8581 | 0.1555 | 0.0151 | 0.027* |
| C13 | 0.76235 (10) | 0.32268 (8) | −0.05314 (7) | 0.02090 (16) |
| C14 | 0.73151 (11) | 0.44598 (8) | −0.04146 (8) | 0.02183 (16) |
| H14A | 0.6834 | 0.5036 | −0.0927 | 0.026* |
| C15 | 0.77373 (10) | 0.48127 (8) | 0.04772 (7) | 0.02078 (16) |
| H15A | 0.7532 | 0.5633 | 0.0554 | 0.025* |
| C16 | 0.66227 (11) | 0.35667 (9) | −0.22515 (8) | 0.02424 (17) |
| H16A | 0.7379 | 0.4242 | −0.2752 | 0.029* |
| H16B | 0.5613 | 0.3935 | −0.1892 | 0.029* |
| C17 | 0.63573 (12) | 0.27105 (9) | −0.29722 (8) | 0.02506 (17) |
| H17A | 0.5556 | 0.2074 | −0.2455 | 0.030* |
| H17B | 0.7358 | 0.2277 | −0.3235 | 0.030* |
| C18 | 0.58040 (11) | 0.33829 (9) | −0.40639 (8) | 0.02417 (17) |
| H18A | 0.4746 | 0.3743 | −0.3809 | 0.029* |
| H18B | 0.6551 | 0.4066 | −0.4560 | 0.029* |
| C19 | 0.57271 (12) | 0.24580 (9) | −0.47838 (8) | 0.02537 (18) |
| H19A | 0.5075 | 0.1735 | −0.4251 | 0.030* |
| H19B | 0.6809 | 0.2157 | −0.5082 | 0.030* |
| C20 | 0.50411 (11) | 0.29821 (9) | −0.58370 (8) | 0.02291 (16) |
| H20A | 0.3978 | 0.3323 | −0.5557 | 0.027* |
| H20B | 0.5730 | 0.3666 | −0.6408 | 0.027* |
| C21 | 0.49168 (11) | 0.19692 (9) | −0.64525 (8) | 0.02409 (17) |
| H21A | 0.4292 | 0.1263 | −0.5861 | 0.029* |
| H21B | 0.5993 | 0.1665 | −0.6763 | 0.029* |
| C22 | 0.41473 (11) | 0.24020 (8) | −0.74698 (8) | 0.02339 (17) |
| H22A | 0.4848 | 0.3026 | −0.8115 | 0.028* |
| H22B | 0.3129 | 0.2806 | −0.7190 | 0.028* |

| | | | | |
|------|--------------|--------------|---------------|--------------|
| C23 | 0.38450 (12) | 0.13207 (9) | -0.79502 (8) | 0.02553 (18) |
| H23A | 0.4858 | 0.0898 | -0.8200 | 0.031* |
| H23B | 0.3116 | 0.0712 | -0.7310 | 0.031* |
| C24 | 0.31312 (12) | 0.17379 (10) | -0.89953 (9) | 0.02908 (19) |
| H24A | 0.3923 | 0.2250 | -0.9676 | 0.035* |
| H24B | 0.2199 | 0.2261 | -0.8785 | 0.035* |
| C25 | 0.26215 (14) | 0.06325 (11) | -0.93513 (10) | 0.0364 (2) |
| H25A | 0.2134 | 0.0949 | -0.9987 | 0.055* |
| H25B | 0.1857 | 0.0109 | -0.8676 | 0.055* |
| H25C | 0.3552 | 0.0143 | -0.9615 | 0.055* |
| H1O1 | 0.9937 (19) | 0.6709 (15) | 0.5747 (15) | 0.054 (4)* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|------------|------------|------------|-------------|-------------|-------------|
| O1 | 0.0556 (5) | 0.0212 (3) | 0.0321 (4) | 0.0096 (3) | -0.0303 (3) | -0.0091 (3) |
| O2 | 0.0347 (3) | 0.0214 (3) | 0.0252 (3) | 0.0039 (2) | -0.0166 (3) | -0.0073 (2) |
| O3 | 0.0373 (4) | 0.0246 (3) | 0.0235 (3) | 0.0037 (3) | -0.0170 (3) | -0.0099 (2) |
| C1 | 0.0322 (4) | 0.0187 (4) | 0.0236 (4) | 0.0014 (3) | -0.0130 (3) | -0.0042 (3) |
| C2 | 0.0410 (5) | 0.0178 (4) | 0.0316 (5) | 0.0032 (3) | -0.0178 (4) | -0.0049 (3) |
| C3 | 0.0368 (5) | 0.0178 (4) | 0.0343 (5) | 0.0028 (3) | -0.0164 (4) | -0.0082 (3) |
| C4 | 0.0310 (4) | 0.0204 (4) | 0.0287 (4) | 0.0026 (3) | -0.0143 (3) | -0.0100 (3) |
| C5 | 0.0247 (4) | 0.0178 (3) | 0.0240 (4) | 0.0031 (3) | -0.0112 (3) | -0.0058 (3) |
| C6 | 0.0220 (4) | 0.0171 (3) | 0.0198 (3) | 0.0004 (3) | -0.0084 (3) | -0.0047 (3) |
| C7 | 0.0249 (4) | 0.0171 (3) | 0.0216 (4) | 0.0016 (3) | -0.0092 (3) | -0.0055 (3) |
| C8 | 0.0273 (4) | 0.0179 (3) | 0.0209 (4) | -0.0003 (3) | -0.0104 (3) | -0.0045 (3) |
| C9 | 0.0203 (3) | 0.0191 (3) | 0.0188 (3) | -0.0006 (3) | -0.0068 (3) | -0.0052 (3) |
| C10 | 0.0205 (3) | 0.0181 (3) | 0.0172 (3) | -0.0011 (3) | -0.0063 (3) | -0.0044 (3) |
| C11 | 0.0247 (4) | 0.0188 (3) | 0.0207 (4) | 0.0014 (3) | -0.0100 (3) | -0.0054 (3) |
| C12 | 0.0291 (4) | 0.0196 (4) | 0.0220 (4) | 0.0021 (3) | -0.0114 (3) | -0.0073 (3) |
| C13 | 0.0240 (4) | 0.0225 (4) | 0.0182 (3) | -0.0006 (3) | -0.0076 (3) | -0.0066 (3) |
| C14 | 0.0259 (4) | 0.0211 (4) | 0.0206 (4) | 0.0019 (3) | -0.0102 (3) | -0.0055 (3) |
| C15 | 0.0252 (4) | 0.0187 (3) | 0.0197 (3) | 0.0007 (3) | -0.0083 (3) | -0.0052 (3) |
| C16 | 0.0285 (4) | 0.0271 (4) | 0.0210 (4) | 0.0023 (3) | -0.0114 (3) | -0.0085 (3) |
| C17 | 0.0295 (4) | 0.0281 (4) | 0.0221 (4) | -0.0002 (3) | -0.0114 (3) | -0.0093 (3) |
| C18 | 0.0263 (4) | 0.0281 (4) | 0.0218 (4) | 0.0012 (3) | -0.0100 (3) | -0.0093 (3) |
| C19 | 0.0302 (4) | 0.0279 (4) | 0.0216 (4) | 0.0011 (3) | -0.0103 (3) | -0.0092 (3) |
| C20 | 0.0246 (4) | 0.0263 (4) | 0.0197 (4) | -0.0002 (3) | -0.0079 (3) | -0.0073 (3) |
| C21 | 0.0289 (4) | 0.0260 (4) | 0.0207 (4) | 0.0016 (3) | -0.0110 (3) | -0.0077 (3) |
| C22 | 0.0264 (4) | 0.0243 (4) | 0.0219 (4) | 0.0015 (3) | -0.0102 (3) | -0.0068 (3) |
| C23 | 0.0316 (4) | 0.0254 (4) | 0.0239 (4) | 0.0023 (3) | -0.0144 (3) | -0.0070 (3) |
| C24 | 0.0330 (5) | 0.0309 (5) | 0.0279 (4) | 0.0006 (4) | -0.0176 (4) | -0.0056 (4) |
| C25 | 0.0427 (6) | 0.0421 (6) | 0.0349 (5) | 0.0011 (5) | -0.0217 (4) | -0.0156 (5) |

Geometric parameters (Å, °)

| | | | |
|---------|-------------|----------|-------------|
| O1—C5 | 1.3523 (10) | C16—C17 | 1.5146 (12) |
| O1—H1O1 | 0.943 (17) | C16—H16A | 0.9700 |

| | | | |
|------------|-------------|---------------|-------------|
| O2—C9 | 1.2389 (10) | C16—H16B | 0.9700 |
| O3—C13 | 1.3485 (10) | C17—C18 | 1.5260 (13) |
| O3—C16 | 1.4363 (11) | C17—H17A | 0.9700 |
| C1—C2 | 1.3811 (13) | C17—H17B | 0.9700 |
| C1—C6 | 1.4000 (12) | C18—C19 | 1.5236 (12) |
| C1—H1A | 0.9300 | C18—H18A | 0.9700 |
| C2—C3 | 1.3938 (14) | C18—H18B | 0.9700 |
| C2—H2A | 0.9300 | C19—C20 | 1.5239 (13) |
| C3—C4 | 1.3846 (13) | C19—H19A | 0.9700 |
| C3—H3A | 0.9300 | C19—H19B | 0.9700 |
| C4—C5 | 1.3985 (12) | C20—C21 | 1.5228 (12) |
| C4—H4A | 0.9300 | C20—H20A | 0.9700 |
| C5—C6 | 1.4080 (12) | C20—H20B | 0.9700 |
| C6—C7 | 1.4579 (11) | C21—C22 | 1.5229 (12) |
| C7—C8 | 1.3414 (12) | C21—H21A | 0.9700 |
| C7—H7A | 0.9300 | C21—H21B | 0.9700 |
| C8—C9 | 1.4706 (12) | C22—C23 | 1.5250 (13) |
| C8—H8A | 0.9300 | C22—H22A | 0.9700 |
| C9—C10 | 1.4854 (11) | C22—H22B | 0.9700 |
| C10—C15 | 1.3998 (11) | C23—C24 | 1.5201 (13) |
| C10—C11 | 1.4072 (12) | C23—H23A | 0.9700 |
| C11—C12 | 1.3800 (12) | C23—H23B | 0.9700 |
| C11—H11A | 0.9300 | C24—C25 | 1.5252 (15) |
| C12—C13 | 1.3999 (12) | C24—H24A | 0.9700 |
| C12—H12A | 0.9300 | C24—H24B | 0.9700 |
| C13—C14 | 1.3988 (12) | C25—H25A | 0.9600 |
| C14—C15 | 1.3906 (12) | C25—H25B | 0.9600 |
| C14—H14A | 0.9300 | C25—H25C | 0.9600 |
| C15—H15A | 0.9300 | | |
| | | | |
| C5—O1—H1O1 | 110.9 (10) | C16—C17—H17A | 108.6 |
| C13—O3—C16 | 120.53 (7) | C18—C17—H17A | 108.6 |
| C2—C1—C6 | 121.51 (9) | C16—C17—H17B | 108.6 |
| C2—C1—H1A | 119.2 | C18—C17—H17B | 108.6 |
| C6—C1—H1A | 119.2 | H17A—C17—H17B | 107.6 |
| C1—C2—C3 | 119.86 (9) | C19—C18—C17 | 110.31 (8) |
| C1—C2—H2A | 120.1 | C19—C18—H18A | 109.6 |
| C3—C2—H2A | 120.1 | C17—C18—H18A | 109.6 |
| C4—C3—C2 | 120.20 (9) | C19—C18—H18B | 109.6 |
| C4—C3—H3A | 119.9 | C17—C18—H18B | 109.6 |
| C2—C3—H3A | 119.9 | H18A—C18—H18B | 108.1 |
| C3—C4—C5 | 119.80 (8) | C18—C19—C20 | 115.59 (8) |
| C3—C4—H4A | 120.1 | C18—C19—H19A | 108.4 |
| C5—C4—H4A | 120.1 | C20—C19—H19A | 108.4 |
| O1—C5—C4 | 122.25 (8) | C18—C19—H19B | 108.4 |
| O1—C5—C6 | 116.99 (8) | C20—C19—H19B | 108.4 |
| C4—C5—C6 | 120.76 (8) | H19A—C19—H19B | 107.4 |
| C1—C6—C5 | 117.86 (8) | C21—C20—C19 | 111.70 (8) |

| | | | |
|---------------|-------------|-----------------|-------------|
| C1—C6—C7 | 122.86 (8) | C21—C20—H20A | 109.3 |
| C5—C6—C7 | 119.27 (7) | C19—C20—H20A | 109.3 |
| C8—C7—C6 | 126.24 (8) | C21—C20—H20B | 109.3 |
| C8—C7—H7A | 116.9 | C19—C20—H20B | 109.3 |
| C6—C7—H7A | 116.9 | H20A—C20—H20B | 107.9 |
| C7—C8—C9 | 122.90 (8) | C20—C21—C22 | 114.80 (8) |
| C7—C8—H8A | 118.6 | C20—C21—H21A | 108.6 |
| C9—C8—H8A | 118.6 | C22—C21—H21A | 108.6 |
| O2—C9—C8 | 122.21 (8) | C20—C21—H21B | 108.6 |
| O2—C9—C10 | 119.06 (7) | C22—C21—H21B | 108.6 |
| C8—C9—C10 | 118.72 (7) | H21A—C21—H21B | 107.5 |
| C15—C10—C11 | 117.94 (8) | C21—C22—C23 | 112.95 (7) |
| C15—C10—C9 | 124.00 (7) | C21—C22—H22A | 109.0 |
| C11—C10—C9 | 118.06 (7) | C23—C22—H22A | 109.0 |
| C12—C11—C10 | 121.04 (8) | C21—C22—H22B | 109.0 |
| C12—C11—H11A | 119.5 | C23—C22—H22B | 109.0 |
| C10—C11—H11A | 119.5 | H22A—C22—H22B | 107.8 |
| C11—C12—C13 | 120.15 (8) | C24—C23—C22 | 113.85 (8) |
| C11—C12—H12A | 119.9 | C24—C23—H23A | 108.8 |
| C13—C12—H12A | 119.9 | C22—C23—H23A | 108.8 |
| O3—C13—C14 | 125.19 (8) | C24—C23—H23B | 108.8 |
| O3—C13—C12 | 114.85 (8) | C22—C23—H23B | 108.8 |
| C14—C13—C12 | 119.96 (8) | H23A—C23—H23B | 107.7 |
| C15—C14—C13 | 119.15 (8) | C23—C24—C25 | 113.04 (8) |
| C15—C14—H14A | 120.4 | C23—C24—H24A | 109.0 |
| C13—C14—H14A | 120.4 | C25—C24—H24A | 109.0 |
| C14—C15—C10 | 121.76 (8) | C23—C24—H24B | 109.0 |
| C14—C15—H15A | 119.1 | C25—C24—H24B | 109.0 |
| C10—C15—H15A | 119.1 | H24A—C24—H24B | 107.8 |
| O3—C16—C17 | 104.98 (7) | C24—C25—H25A | 109.5 |
| O3—C16—H16A | 110.8 | C24—C25—H25B | 109.5 |
| C17—C16—H16A | 110.8 | H25A—C25—H25B | 109.5 |
| O3—C16—H16B | 110.8 | C24—C25—H25C | 109.5 |
| C17—C16—H16B | 110.8 | H25A—C25—H25C | 109.5 |
| H16A—C16—H16B | 108.8 | H25B—C25—H25C | 109.5 |
| C16—C17—C18 | 114.77 (8) | | |
| C6—C1—C2—C3 | 0.50 (15) | C9—C10—C11—C12 | 179.29 (7) |
| C1—C2—C3—C4 | -0.91 (16) | C10—C11—C12—C13 | -0.32 (13) |
| C2—C3—C4—C5 | 0.14 (15) | C16—O3—C13—C14 | 4.73 (13) |
| C3—C4—C5—O1 | -178.52 (9) | C16—O3—C13—C12 | -175.05 (8) |
| C3—C4—C5—C6 | 1.05 (14) | C11—C12—C13—O3 | -179.37 (8) |
| C2—C1—C6—C5 | 0.65 (14) | C11—C12—C13—C14 | 0.84 (13) |
| C2—C1—C6—C7 | 179.81 (9) | O3—C13—C14—C15 | 179.56 (8) |
| O1—C5—C6—C1 | 178.17 (8) | C12—C13—C14—C15 | -0.67 (13) |
| C4—C5—C6—C1 | -1.42 (13) | C13—C14—C15—C10 | -0.01 (13) |
| O1—C5—C6—C7 | -1.02 (12) | C11—C10—C15—C14 | 0.51 (13) |
| C4—C5—C6—C7 | 179.39 (8) | C9—C10—C15—C14 | -179.10 (8) |

| | | | |
|-----------------|-------------|-----------------|-------------|
| C1—C6—C7—C8 | 4.99 (14) | C13—O3—C16—C17 | -179.02 (7) |
| C5—C6—C7—C8 | -175.86 (9) | O3—C16—C17—C18 | -175.00 (7) |
| C6—C7—C8—C9 | 179.01 (8) | C16—C17—C18—C19 | 174.72 (8) |
| C7—C8—C9—O2 | -6.40 (14) | C17—C18—C19—C20 | 174.57 (8) |
| C7—C8—C9—C10 | 173.90 (8) | C18—C19—C20—C21 | -176.48 (7) |
| O2—C9—C10—C15 | 173.99 (8) | C19—C20—C21—C22 | 176.53 (8) |
| C8—C9—C10—C15 | -6.30 (12) | C20—C21—C22—C23 | -172.41 (8) |
| O2—C9—C10—C11 | -5.62 (12) | C21—C22—C23—C24 | -177.98 (8) |
| C8—C9—C10—C11 | 174.09 (7) | C22—C23—C24—C25 | -172.02 (9) |
| C15—C10—C11—C12 | -0.34 (13) | | |

Hydrogen-bond geometry (Å, °)

Cg1 is the centroid of the C10—C15 ring.

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|-------------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| O1—H1O1 \cdots O2 ⁱ | 0.939 (18) | 1.789 (18) | 2.6867 (12) | 159.0 (16) |
| C7—H7A \cdots O1 ⁱ | 0.93 | 2.31 | 3.2169 (13) | 164 |
| C22—H22A \cdots Cg1 ⁱⁱ | 0.97 | 2.85 | 3.6887 (12) | 146 |

Symmetry codes: (i) $-x+2, -y+1, -z+1$; (ii) $x, y, z-1$.