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5-Methyl-1H-indole-3-carbaldehyde

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Key indicators: single-crystal X-ray study; T = 296 K; mean σ (C–C) = 0.004 Å; R factor = 0.038; wR factor = 0.107; data-to-parameter ratio = 10.2.

The title molecule, $C_{10}H_9NO$, is almost planar with an r.m.s. deviation for all non-H atoms of 0.0115 Å. In the crystal, molecules are connected through N-H···O hydrogen bonds into chains running along [021]. The chains are further connected *via* C-H··· π interactions, forming layers in the *bc* plane.

Related literature

For the structure of 1*H*-indole-3-carbaldehyde, see: Ng (2007) and for the structure of 6-bromo-1*H*-indole-3-carbaldehyde, see: Johnson *et al.* (2009).



Experimental

Crystal data

C ₁₀ H ₉ NO
$M_r = 159.18$
Orthorhombic, Pca2 ₁
a = 16.9456 (19) Å

b = 5.	7029 (6)
c = 8.	6333 (9) Å
V = 8	34.31 (15)
Z = 4	

Mo $K\alpha$ radiation $\mu = 0.08 \text{ mm}^{-1}$

Data collection

Bruker APEXII CCD diffractometer Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\rm min} = 0.962, T_{\rm max} = 0.996$

Refinement

$$\begin{split} R[F^2 > 2\sigma(F^2)] &= 0.038 & \text{H atoms treated by a mixture of} \\ wR(F^2) &= 0.107 & \text{independent and constrained} \\ S &= 0.98 & \text{refinement} \\ 1147 \text{ reflections} & \Delta\rho_{\text{max}} &= 0.12 \text{ e } \text{ Å}^{-3} \\ 113 \text{ parameters} & \Delta\rho_{\text{min}} &= -0.14 \text{ e } \text{ Å}^{-3} \end{split}$$

Table 1

Hydrogen-bond geometry (Å, °).

Cg is the centroid of the N1/C1/C2/C3/C8 ring.

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$	
$N1 - H1N \cdots O1^{i}$ $C9 - H9 \cdots Cg^{ii}$	0.93 (3) 0.93	1.90 (3) 2.91	2.818 (3) 3.312 (3)	169 (3) 107	
Symmetry codes: (i) $-x + \frac{1}{2}$, $y - 1$, $z - \frac{1}{2}$; (ii) $-x + \frac{1}{2}$, $y, z + \frac{1}{2}$.					

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *SHELXL97* and *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2498).

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 $0.47 \times 0.15 \times 0.05 \text{ mm}$

5499 measured reflections

1147 independent reflections

717 reflections with $I > 2\sigma(I)$

T = 296 K

 $R_{\rm int} = 0.039$

supporting information

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5-Methyl-1H-indole-3-carbaldehyde

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S1. Comment

The structure of the title compound is isomorphous with that of 1*H*-indole-3-carbaldehyde (Ng, 2007). The planar molecules are connected *via* N—H···O hydrogen bonds (Table 1) into chains in the [021] direction. The chains are further linked through C—H··· π interactions (Table 1) to form layers in the *bc* plane. The structure of 6-bromo-1*H*-indole-3-carbaldehyde (Johnson *et al.*, 2009) exhibits similar N—H···O bonded chains, however, further supramolecular aggregation by Br-involved interactions is observed.

S2. Experimental

The title crystals were obtained by slow evaporation of an ethanolic solution of the commercially available 5-methylindole-3-carboxaldehyde at room temperature.

S3. Refinement

The C-bound hydrogen atoms were located in calculated positions and refined in a riding mode with C—H distances of 0.93 (C_{sp2}) and 0.96 (C_{methyl}) Å. The N-bound H atom was found in a difference Fourier map and refined freely. For all hydrogen atoms, U_{iso} were set to 1.2–1.5 U_{eq} (carrier atom). In the absence of significant anomalous scattering effects Friedel pairs were merged.



Figure 1

Molecular structure of the title compound showing thermal ellipsoids at the 30% prbability level. Hydrogen spheres are drawn with an arbitrary radius.



Figure 2

Crystal packing view looking down the *a* axis, thus showing the two-dimensional-supramolecular structure formed by N -H \cdots O and C-H \cdots π interactions (dashed lines).

5-Methyl-1*H*-indole-3-carbaldehyde

Crystal data	
C ₁₀ H ₉ NO	F(000) = 336
$M_r = 159.18$	$D_{\rm x} = 1.267 {\rm Mg} {\rm m}^{-3}$
Orthorhombic, <i>Pca</i> 2 ₁	Mo <i>K</i> α radiation, $\lambda = 0.71073$ Å
Hall symbol: P 2c -2ac	Cell parameters from 1172 reflections
a = 16.9456 (19) Å	$\theta = 2.4 - 22.1^{\circ}$
b = 5.7029 (6) Å	$\mu=0.08~\mathrm{mm^{-1}}$
c = 8.6333 (9) Å	T = 296 K
$V = 834.31 (15) Å^3$	Lath, yellow
Z = 4	$0.47 \times 0.15 \times 0.05 \text{ mm}$
Data collection	
Bruker APEXII CCD	5499 measured reflections
diffractometer	1147 independent reflections
Radiation source: fine-focus sealed tube	717 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.039$
φ and ω scans	$\theta_{\rm max} = 28.8^\circ, \ \theta_{\rm min} = 2.4^\circ$
Absorption correction: multi-scan	$h = -22 \rightarrow 13$
(SADABS; Sheldrick, 1996)	$k = -7 \rightarrow 7$
$T_{\min} = 0.962, \ T_{\max} = 0.996$	$l = -11 \rightarrow 11$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.038$	Hydrogen site location: inferred from
$wR(F^2) = 0.107$	neighbouring sites
S = 0.98	H atoms treated by a mixture of independent
1147 reflections	and constrained refinement
113 parameters	$w = 1/[\sigma^2(F_o^2) + (0.0596P)^2]$
1 restraint	where $P = (F_{o}^{2} + 2F_{c}^{2})/3$
Primary atom site location: structure-invariant	$(\Delta/\sigma)_{\rm max} < 0.001$
direct methods	$\Delta ho_{ m max} = 0.12 \ m e \ m \AA^{-3}$
	$\Delta \rho_{\rm min} = -0.14 \text{ e} \text{ Å}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes. **Refinement**. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (A^2)

are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

	x	У	<i>Z</i>	$U_{ m iso}$ */ $U_{ m eq}$	
01	0.27204 (11)	1.0610 (3)	0.6573 (2)	0.0742 (6)	
N1	0.29282 (16)	0.4126 (4)	0.3481 (3)	0.0760 (7)	
H1N	0.2766 (15)	0.285 (5)	0.290 (5)	0.091*	
C1	0.2435 (2)	0.5303 (4)	0.4384 (3)	0.0717 (8)	
H1	0.1913	0.4879	0.4563	0.086*	
C2	0.27969 (14)	0.7233 (4)	0.5018 (3)	0.0587 (6)	
C3	0.35916 (14)	0.7227 (4)	0.4414 (3)	0.0535 (6)	
C4	0.42432 (15)	0.8710 (4)	0.4561 (3)	0.0554 (6)	
H4	0.4212	1.0036	0.5185	0.066*	
C5	0.49316 (16)	0.8208 (4)	0.3782 (3)	0.0638 (7)	
C6	0.49673 (18)	0.6225 (5)	0.2839 (4)	0.0785 (8)	
H6	0.5434	0.5904	0.2311	0.094*	
C7	0.4341 (2)	0.4736 (5)	0.2661 (3)	0.0766 (8)	
H7	0.4377	0.3422	0.2026	0.092*	
C8	0.36502 (18)	0.5241 (4)	0.3453 (3)	0.0626 (7)	
С9	0.24192 (16)	0.8851 (4)	0.6026 (3)	0.0640 (6)	
H9	0.1897	0.8544	0.6292	0.077*	
C10	0.56347 (18)	0.9773 (6)	0.3966 (4)	0.0860 (10)	
H10A	0.5654	1.0866	0.3121	0.129*	
H10B	0.6107	0.8842	0.3969	0.129*	
H10C	0.5595	1.0615	0.4926	0.129*	

U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
0.0843 (15)	0.0640 (9)	0.0743 (12)	0.0106 (9)	0.0086 (10)	-0.0101 (9)
0.109 (2)	0.0565 (10)	0.0622 (13)	-0.0129 (12)	-0.0146 (15)	-0.0053 (11)
0.0820 (19)	0.0660 (13)	0.0671 (17)	-0.0143 (15)	-0.0109 (17)	0.0081 (14)
0.0722 (18)	0.0533 (11)	0.0508 (12)	-0.0012 (11)	-0.0030 (12)	0.0033 (10)
0.0668 (17)	0.0499 (10)	0.0440 (11)	0.0056 (10)	-0.0070 (11)	-0.0013 (10)
0.0639 (16)	0.0528 (10)	0.0495 (12)	0.0026 (11)	-0.0052 (12)	-0.0003 (10)
0.0640 (18)	0.0699 (14)	0.0575 (14)	0.0141 (12)	-0.0022 (13)	0.0073 (13)
0.081 (2)	0.0869 (18)	0.0678 (16)	0.0295 (15)	0.0074 (16)	0.0032 (15)
0.105 (2)	0.0657 (14)	0.0590 (16)	0.0237 (16)	0.0008 (17)	-0.0133 (12)
0.089 (2)	0.0488 (10)	0.0502 (13)	0.0051 (12)	-0.0112 (14)	-0.0021 (11)
0.0662 (17)	0.0683 (13)	0.0574 (14)	0.0083 (14)	0.0013 (13)	0.0124 (13)
0.068 (2)	0.102 (2)	0.088 (2)	0.0008 (17)	0.0032 (16)	0.0136 (17)
	0.0843 (15) 0.109 (2) 0.0820 (19) 0.0722 (18) 0.0668 (17) 0.0639 (16) 0.0640 (18) 0.081 (2) 0.105 (2) 0.089 (2) 0.0662 (17) 0.068 (2)	$\begin{array}{c ccccc} 0 & 0 & 0 \\ \hline 0.0843 (15) & 0.0640 (9) \\ 0.109 (2) & 0.0565 (10) \\ 0.0820 (19) & 0.0660 (13) \\ 0.0722 (18) & 0.0533 (11) \\ 0.0668 (17) & 0.0499 (10) \\ 0.0639 (16) & 0.0528 (10) \\ 0.0640 (18) & 0.0699 (14) \\ 0.081 (2) & 0.0869 (18) \\ 0.105 (2) & 0.0657 (14) \\ 0.089 (2) & 0.0488 (10) \\ 0.0662 (17) & 0.0683 (13) \\ 0.068 (2) & 0.102 (2) \end{array}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

Atomic displacement parameters $(Å^2)$

Geometric parameters (Å, °)

O1—C9	1.221 (3)	C4—H4	0.9300
N1—C1	1.326 (4)	C5—C6	1.395 (4)
N1—C8	1.379 (4)	C5—C10	1.497 (4)
N1—H1N	0.93 (3)	C6—C7	1.368 (4)
C1—C2	1.374 (3)	С6—Н6	0.9300
C1—H1	0.9300	C7—C8	1.385 (4)
С2—С9	1.421 (3)	С7—Н7	0.9300
C2—C3	1.444 (3)	С9—Н9	0.9300
C3—C4	1.397 (3)	C10—H10A	0.9600
C3—C8	1.408 (3)	C10—H10B	0.9600
C4—C5	1.377 (3)	C10—H10C	0.9600
C1—N1—C8	109 7 (2)	C7—C6—C5	122 3 (3)
C1—N1—H1N	121.9(18)	C7—C6—H6	118.8
C8—N1—H1N	128.1 (18)	C5-C6-H6	118.8
N1—C1—C2	111.0 (3)	C6—C7—C8	118.1 (2)
N1—C1—H1	124.5	С6—С7—Н7	121.0
C2—C1—H1	124.5	С8—С7—Н7	121.0
C1—C2—C9	124.3 (3)	N1—C8—C7	131.4 (2)
C1—C2—C3	105.7 (2)	N1—C8—C3	107.3 (2)
C9—C2—C3	130.0 (2)	C7—C8—C3	121.2 (3)
C4—C3—C8	119.0 (2)	O1—C9—C2	125.6 (3)
C4—C3—C2	134.7 (2)	O1—C9—H9	117.2
C8—C3—C2	106.3 (2)	С2—С9—Н9	117.2
C5—C4—C3	120.0 (2)	C5C10H10A	109.5
C5—C4—H4	120.0	C5-C10-H10B	109.5
C3—C4—H4	120.0	H10A—C10—H10B	109.5
C4—C5—C6	119.4 (3)	C5-C10-H10C	109.5
C4—C5—C10	119.9 (2)	H10A-C10-H10C	109.5
C6—C5—C10	120.7 (3)	H10B—C10—H10C	109.5

Hydrogen-bond geometry (Å, °)

Cg is the centroid of the N1/C1/C2/C3/C8 ring.

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	D—H···A
$\overline{\begin{array}{c} N1 \longrightarrow H1N \cdots O1^{i} \\ C9 \longrightarrow Cg^{ii} \end{array}}$	0.93 (3)	1.90 (3)	2.818 (3)	169 (3)
	0.93	2.91	3.312 (3)	107

Symmetry codes: (i) -*x*+1/2, *y*-1, *z*-1/2; (ii) -*x*+1/2, *y*, *z*+1/2.