

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

Benzyl 3-[(E)-(furan-2-yl)methylidene]-2-methyldithiocarbazate

Benu K. Dey,^a Sebastian Suarez,^b Biplab Ganguly,^a Fabio Doctorovich^b and Tapashi G. Roy^{a*}^aUniversity of Chittagong, Chittagong 4331, Bangladesh, and ^bDepartamento de Química Inorgánica Analítica y Química Física, INQUIMAE-CONICET, Facultad de Ciencias Exactas y Naturales, Universidad de Buenos Aires, Argentina

Correspondence e-mail: tapashir57@gmail.com

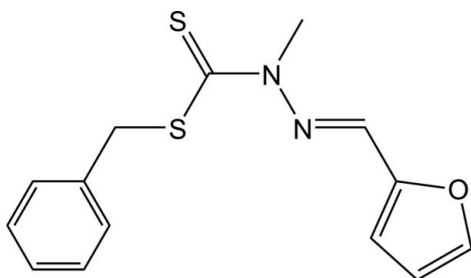
Received 19 July 2012; accepted 12 August 2012

Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; disorder in main residue; R factor = 0.040; wR factor = 0.090; data-to-parameter ratio = 15.3.

In the title compound, $\text{C}_{14}\text{H}_{14}\text{N}_2\text{OS}_2$, the furan ring exhibits rotational disorder over two orientations, with an occupancy ratio of 0.508 (7):0.492 (7). The furan and phenyl rings form dihedral angles of 8.2 (6) (major occupancy component), 14.8 (6) (minor occupancy component) and 73.65 (9)°, respectively, with the central residue ($\text{C}_4\text{N}_2\text{S}_2$), indicating a twisted conformation for the molecule. The methyl group and the thione S atom are *syn* and the conformation about the imine bond is *E*. In the crystal, $\text{C}-\text{H}\cdots\pi$ interactions involving the phenyl ring are observed.

Related literature

For background to the biological activity of S-containing ligands, see: Hazari *et al.* (2012). For related structures, see: Shan *et al.* (2008); Ganguly *et al.* (2011). For a similar compound with a thiophene instead of a furan ring, see: Hazari *et al.* (2012).



Experimental

Crystal data

 $\text{C}_{14}\text{H}_{14}\text{N}_2\text{OS}_2$
 $M_r = 290.39$ Monoclinic, $P2_1/n$
 $a = 6.0415$ (3) Å $b = 20.4840$ (11) Å
 $c = 11.8959$ (7) Å
 $\beta = 101.601$ (5)°
 $V = 1442.09$ (14) Å³
 $Z = 4$ Mo $K\alpha$ radiation
 $\mu = 0.36$ mm⁻¹
 $T = 298$ K
 $0.5 \times 0.5 \times 0.3$ mm

Data collection

Oxford Diffraction Gemini CCD S
Ultra diffractometer
Absorption correction: multi-scan
(*CrysAlis PRO*; Oxford
Diffraction, 2009)
 $T_{\min} = 0.850$, $T_{\max} = 0.897$ 21725 measured reflections
3361 independent reflections
2393 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.090$
 $S = 1.06$
3361 reflections
219 parameters12 restraints
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.16$ e Å⁻³
 $\Delta\rho_{\min} = -0.17$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Table 1. C_g is the centroid of the phenyl ring

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C7-H7B\cdots Cg^i$	0.97	2.88	3.560 (2)	128
$C13-H13A\cdots Cg^{ii}$	0.93	2.80	3.62 (2)	149

Symmetry codes: (i) $-x - 1, -y, -z + 1$; (ii) $-x + 1, -y, -z + 2$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

The authors acknowledge the University Grants Commission (UGC), Bangladesh, for the award of a fellowship to BG and thank the Third World Academy of Sciences (TWAS), Trieste, Italy, for awarding a TWAS-UNESCO Associateship to TGR. They are also grateful to ANPCyT for a grant (PME-2006-01113) and especially to R. Baggio for his helpful suggestions.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2449).

References

- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Ganguly, B., Foi, A., Doctorovich, F., K. Dey, B. & Roy, T. G. (2011). *Acta Cryst.* **E67**, o2777.
Hazari, S. K. S., Dey, B. K., Roy, T. G., Ganguly, B., Ng, S. W. & Tiekink, E. R. T. (2012). *Acta Cryst.* **E68**, o1216.
Oxford Diffraction (2009). *CrysAlis PRO*. Oxford Diffraction Ltd, Yarnton, England.
Shan, S., Tian, Y.-L., Wang, S.-H., Wang, W.-L. & Xu, Y.-L. (2008). *Acta Cryst.* **E64**, o1024.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

supporting information

Acta Cryst. (2012). E68, o2752 [doi:10.1107/S1600536812035520]

Benzyl 3-[(*E*)-(furan-2-yl)methylidene]-2-methyldithiocarbazate

Benu K. Dey, Sebastian Suarez, Biplab Ganguly, Fabio Doctorovich and Tapashi G. Roy

S1. Comment

As a continuation of systematic studies into the synthesis, characterization and biological activities of substituted Schiff base ligands and their metal complexes (Ganguly *et al.*, 2011; Hazari *et al.*, 2012), the present investigation is an attempt to prepare complexes of vanadium(IV) and molybdenum(VI) with the title Schiff base ligand, benzyl 2-methyl-3-[(*E*)-(furan-2-yl)-methylidene]dithiocarbazate. Crystals of the title compound were isolated (see *Experimental*) and characterized crystallographically.

In the title compound (Fig. 1), C₁₄H₁₄N₂OS₂, the furan ring exhibits rotational disorder over two orientations, with an occupancy of 0.5 for each orientation. The thione S atom and methyl group are *syn* and the conformation about the imine N2=C10 bond [1.281 (2) Å] is *E*, in agreement with similar structures (Hazari *et al.*, 2012).

The eight atoms of the central residue (S1, S2, N1, N2, C7, C8, C9 and C10) are co-planar having a r.m.s. deviation for the fitted atoms of 0.002 Å. The maximum deviations from this plane are 0.043 (2) Å for the N2 atom and -0.033 (3) Å for the N1 atom. The molecule is twisted, the dihedral angles between the C₄N₂S₂ residue and the pendent 2-furanyl and phenyl rings being 14.8 (6) [or 8.22 (6) for the disordered part of the furanyl] and 73.65 (9)° respectively, as found in a similar compound (Shan *et al.*, 2008).

In the crystal, molecules assemble into a three-dimensional architecture by $\pi \cdots \pi$ stacking between 2-furanyl rings [Cg₁...Cg₁ⁱⁱⁱ = 4.467 (7) Å, Cg₁ is the centroid of ring O1, C11, C12, C13, C14; symmetry code: iii 4-x, -y, 2-z], and C-H $\cdots \pi$ interactions, involving the phenyl ring as acceptor (see Table 1 and Fig. 2).

S2. Experimental

Single crystals of the title compound were prepared by following three steps.

Step 1 (Hazari *et al.*, 2012). Synthesis of *N*-methyl-*S*-benzyldithiocarbazate. Potassium hydroxide (11.5 g) was dissolved in 60 ml of 90% ethanol and the mixture was cooled down to 273 K in an ice bath. Methyl hydrazine (11.1 ml) was added slowly with mechanical stirring. A solution of CS₂ (12 ml) was added dropwise from a burette with constant stirring over a period of 1 h. During the addition of CS₂, the temperature of the reaction mixture was not allowed to rise above 279 K. A yellow colour was obtained. After adding carbon disulfide, benzyl chloride (25 ml) was added from a burette dropwise with vigorous mechanical stirring. After complete addition, the mixture was stirred for further 15 min, whereupon shining crystals appeared. The product was separated by filtration, washed with water, recrystallized from ethanol and dried in a vacuum desiccator over silica gel. Yield: 14.20 g. m.p. 373–374 K.

Step 2. Synthesis of the title molecule. A hot solution of furan-2-carbaldehyde (10 mmol) in absolute ethanol (40 ml) was mixed with a hot solution of *N*-methyl-*S*-benzyldithiocarbazate (10 mmol) in 40 ml of the same solvent. The mixture was refluxed for 6 h on a water bath. After reducing the volume, an off white product appeared which was filtered off. This product was washed with ethanol several times and dried in a vacuum desiccator over silica gel. Yield: 1.65 g. m.p. 432–434 K.

Step 3. Crystallization. The product was dissolved in ethanol to which half volume of petroleum ether was added (2:1 v/v, 10 ml ethanol and 5 ml petroleum ether). The solution was left for several days after which crystals of the title compound deposited.

S3. Refinement

All H atoms were placed in idealized positions and allowed to ride on their parent C atoms, with C—H bond lengths fixed to 0.93 (aromatic CH), 0.97 (methylene CH₂) or 0.96 Å (methyl CH₃). Displacement parameters were taken as $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C9})$ for the methyl group and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier C})$ otherwise. The furan ring exhibits rotational disorder over two orientations. The occupancies for all sites were fixed to 0.5, since the refined occupancy for each part was very close to that distribution. In order to approximate the expected geometry for both furan groups, their bond lengths were restrained to be identical, with an effective standard deviation of 0.01 Å (command *SAME* in *SHELXL97*; Sheldrick, 2008).

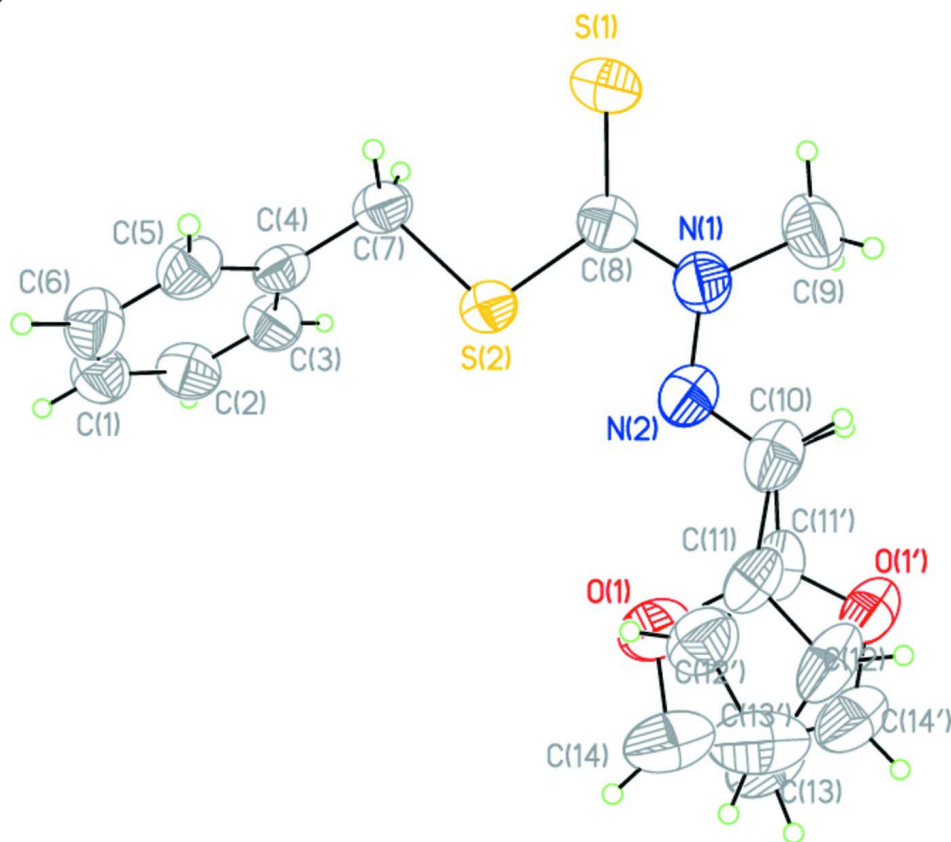


Figure 1

The molecular structure of the title molecule, showing displacement ellipsoids at the 50% probability level for non-H atoms.

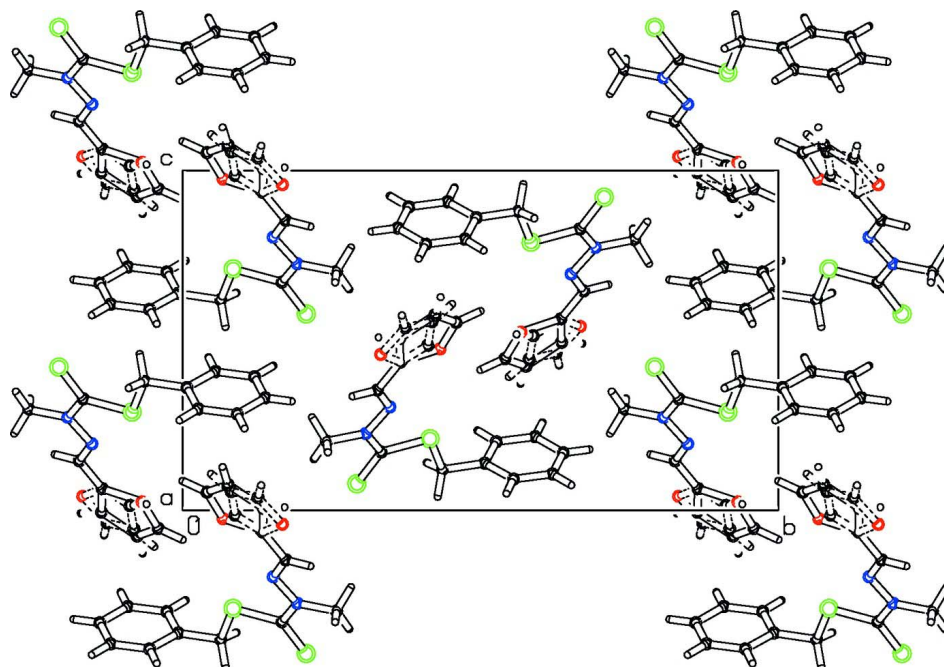


Figure 2

Crystal packing for the title compound viewed along *a*.

Benzyl 3-[(*E*)-(furan-2-yl)methylidene]-2-methyldithiocarbazate

Crystal data

$C_{14}H_{14}N_2OS_2$

$M_r = 290.39$

Monoclinic, $P2_1/n$

Hall symbol: $-P\ 2_1/n$

$a = 6.0415\ (3)\ \text{\AA}$

$b = 20.4840\ (11)\ \text{\AA}$

$c = 11.8959\ (7)\ \text{\AA}$

$\beta = 101.601\ (5)^\circ$

$V = 1442.09\ (14)\ \text{\AA}^3$

$Z = 4$

$F(000) = 608$

$D_x = 1.338\ \text{Mg m}^{-3}$

Melting point: 432 K

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 4212 reflections

$\theta = 4.0\text{--}28.9^\circ$

$\mu = 0.36\ \text{mm}^{-1}$

$T = 298\ \text{K}$

Prism, green

$0.5 \times 0.5 \times 0.3\ \text{mm}$

Data collection

Oxford Diffraction Gemini CCD S Ultra diffractometer

Graphite monochromator

Detector resolution: $16.1158\ \text{pixels mm}^{-1}$

ω scans

Absorption correction: multi-scan

(*CrysAlis PRO*; Oxford Diffraction, 2009)

$T_{\min} = 0.850$, $T_{\max} = 0.897$

21725 measured reflections

3361 independent reflections

2393 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.033$

$\theta_{\max} = 27.9^\circ$, $\theta_{\min} = 4.0^\circ$

$h = -7 \rightarrow 7$

$k = 0 \rightarrow 26$

$l = 0 \rightarrow 15$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.040$

$wR(F^2) = 0.090$

$S = 1.06$

3361 reflections

219 parameters

12 restraints

0 constraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0275P)^2 + 0.3918P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.16 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.84126 (9)	0.20754 (3)	0.57831 (5)	0.06926 (18)	
S2	0.99621 (8)	0.08488 (2)	0.71050 (4)	0.05239 (14)	
N1	1.2300 (3)	0.19263 (7)	0.72613 (14)	0.0561 (4)	
N2	1.3649 (2)	0.15222 (8)	0.80423 (13)	0.0554 (4)	
C1	0.5997 (3)	-0.13887 (10)	0.66897 (17)	0.0637 (5)	
H1A	0.5707	-0.1828	0.6791	0.076*	
C2	0.7562 (4)	-0.12057 (10)	0.60675 (18)	0.0656 (5)	
H2A	0.8344	-0.1522	0.5744	0.079*	
C3	0.7987 (3)	-0.05537 (10)	0.59175 (17)	0.0605 (5)	
H3A	0.9062	-0.0436	0.5494	0.073*	
C4	0.6844 (3)	-0.00718 (9)	0.63849 (14)	0.0471 (4)	
C5	0.5281 (3)	-0.02642 (10)	0.70175 (16)	0.0580 (5)	
H5A	0.4504	0.005	0.735	0.07*	
C6	0.4856 (3)	-0.09170 (11)	0.71639 (19)	0.0676 (6)	
H6A	0.3786	-0.1039	0.7588	0.081*	
C7	0.7263 (3)	0.06390 (9)	0.61993 (16)	0.0554 (5)	
H7A	0.6067	0.0902	0.6403	0.066*	
H7B	0.7302	0.0719	0.54	0.066*	
C8	1.0304 (3)	0.16650 (9)	0.67204 (15)	0.0496 (4)	
C9	1.3012 (4)	0.25833 (10)	0.7034 (2)	0.0813 (7)	
H9A	1.1825	0.2797	0.6503	0.122*	
H9B	1.4343	0.2561	0.671	0.122*	
H9C	1.3339	0.2826	0.7738	0.122*	
C10	1.5598 (3)	0.17332 (11)	0.85386 (18)	0.0663 (6)	
H10A	1.6166	0.2134	0.8364	0.08*	0.508 (7)
H10B	1.5948	0.2156	0.8347	0.08*	0.492 (7)
O1	1.6236 (10)	0.0685 (3)	0.9740 (6)	0.0728 (16)	0.508 (7)
C11	1.691 (2)	0.1271 (8)	0.9441 (14)	0.055 (3)	0.508 (7)
C12	1.9117 (12)	0.1360 (5)	1.0053 (7)	0.081 (2)	0.508 (7)
H12A	2.0061	0.1716	1.003	0.097*	0.508 (7)
C13	1.960 (4)	0.0774 (10)	1.0732 (16)	0.083 (4)	0.508 (7)
H13A	2.095	0.0684	1.124	0.1*	0.508 (7)
C14	1.7950 (12)	0.0413 (4)	1.0535 (6)	0.088 (2)	0.508 (7)

H14A	1.7869	0.0008	1.0878	0.106*	0.508 (7)
O1'	1.9236 (7)	0.1675 (2)	0.9587 (4)	0.0668 (14)	0.492 (7)
C11'	1.720 (2)	0.1418 (9)	0.9281 (16)	0.054 (3)	0.492 (7)
C12'	1.6978 (18)	0.0874 (6)	0.9882 (9)	0.075 (2)	0.492 (7)
H12B	1.5713	0.0611	0.9855	0.09*	0.492 (7)
C13'	1.922 (5)	0.0804 (15)	1.058 (2)	0.112 (9)	0.492 (7)
H13B	1.969	0.0467	1.1097	0.134*	0.492 (7)
C14'	2.0407 (11)	0.1267 (4)	1.0379 (6)	0.078 (2)	0.492 (7)
H14B	2.1915	0.1324	1.073	0.093*	0.492 (7)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0677 (3)	0.0587 (3)	0.0773 (4)	0.0155 (2)	0.0048 (3)	0.0102 (3)
S2	0.0496 (3)	0.0497 (3)	0.0529 (3)	0.0008 (2)	-0.00161 (19)	0.0016 (2)
N1	0.0523 (9)	0.0497 (9)	0.0666 (10)	-0.0012 (7)	0.0124 (8)	-0.0024 (8)
N2	0.0479 (9)	0.0599 (9)	0.0573 (9)	0.0000 (7)	0.0082 (7)	-0.0103 (8)
C1	0.0673 (13)	0.0523 (12)	0.0684 (13)	-0.0010 (10)	0.0061 (10)	0.0024 (10)
C2	0.0726 (13)	0.0577 (12)	0.0690 (13)	0.0165 (10)	0.0201 (11)	-0.0040 (10)
C3	0.0589 (12)	0.0627 (13)	0.0638 (12)	0.0075 (9)	0.0218 (10)	0.0016 (10)
C4	0.0400 (9)	0.0520 (10)	0.0451 (9)	0.0031 (7)	-0.0014 (7)	-0.0034 (8)
C5	0.0487 (10)	0.0617 (12)	0.0637 (12)	0.0049 (9)	0.0116 (9)	-0.0107 (10)
C6	0.0597 (12)	0.0703 (14)	0.0767 (14)	-0.0065 (10)	0.0228 (11)	-0.0010 (11)
C7	0.0479 (10)	0.0534 (11)	0.0592 (11)	0.0034 (8)	-0.0027 (8)	0.0001 (9)
C8	0.0513 (10)	0.0489 (10)	0.0509 (10)	0.0061 (8)	0.0156 (8)	-0.0048 (8)
C9	0.0736 (15)	0.0534 (13)	0.120 (2)	-0.0069 (11)	0.0259 (14)	0.0048 (13)
C10	0.0540 (12)	0.0694 (13)	0.0745 (14)	-0.0052 (10)	0.0108 (10)	-0.0231 (11)
O1	0.064 (3)	0.081 (4)	0.066 (3)	0.006 (2)	-0.004 (2)	-0.003 (2)
C11	0.037 (3)	0.078 (10)	0.051 (6)	-0.010 (5)	0.008 (4)	-0.014 (4)
C12	0.053 (3)	0.100 (7)	0.083 (5)	-0.011 (4)	-0.002 (4)	-0.034 (4)
C13	0.070 (5)	0.102 (10)	0.068 (6)	0.018 (5)	-0.006 (4)	-0.022 (5)
C14	0.080 (5)	0.114 (6)	0.064 (3)	0.031 (4)	-0.003 (3)	-0.002 (4)
O1'	0.052 (2)	0.074 (3)	0.068 (3)	-0.0043 (19)	-0.0011 (17)	-0.0178 (19)
C11'	0.053 (5)	0.056 (5)	0.056 (5)	-0.015 (4)	0.019 (4)	-0.015 (3)
C12'	0.074 (6)	0.090 (7)	0.058 (4)	-0.007 (5)	0.007 (5)	-0.003 (4)
C13'	0.13 (2)	0.129 (14)	0.070 (7)	0.048 (13)	0.003 (9)	0.010 (9)
C14'	0.060 (3)	0.094 (5)	0.070 (4)	0.013 (3)	-0.007 (3)	-0.018 (3)

Geometric parameters (Å, °)

S1—C8	1.6562 (18)	C9—H9B	0.96
S2—C8	1.7566 (19)	C9—H9C	0.96
S2—C7	1.8161 (18)	C10—C11	1.529 (10)
N1—C8	1.357 (2)	C10—C11'	1.338 (11)
N1—N2	1.381 (2)	C10—H10A	0.93
N1—C9	1.455 (2)	C10—H10B	0.9299
N2—C10	1.281 (2)	O1—C11	1.338 (14)
C1—C2	1.365 (3)	O1—C14	1.373 (9)

C1—C6	1.372 (3)	C11—C12	1.396 (13)
C1—H1A	0.93	C12—C13	1.444 (18)
C2—C3	1.378 (3)	C12—H12A	0.93
C2—H2A	0.93	C13—C14	1.23 (3)
C3—C4	1.384 (2)	C13—H13A	0.93
C3—H3A	0.93	C14—H14A	0.93
C4—C5	1.378 (2)	O1'—C11'	1.320 (15)
C4—C7	1.502 (2)	O1'—C14'	1.349 (8)
C5—C6	1.379 (3)	C11'—C12'	1.347 (14)
C5—H5A	0.93	C12'—C13'	1.45 (2)
C6—H6A	0.93	C12'—H12B	0.93
C7—H7A	0.97	C13'—C14'	1.24 (3)
C7—H7B	0.97	C13'—H13B	0.93
C9—H9A	0.96	C14'—H14B	0.93
C8—S2—C7	102.08 (8)	H9A—C9—H9C	109.5
C8—N1—N2	115.50 (15)	H9B—C9—H9C	109.5
C8—N1—C9	123.00 (17)	N2—C10—C11'	128.2 (7)
N2—N1—C9	121.50 (16)	N2—C10—C11	114.4 (6)
C10—N2—N1	118.11 (17)	N2—C10—H10A	122.8
C2—C1—C6	119.30 (19)	C11'—C10—H10A	108.8
C2—C1—H1A	120.3	C11—C10—H10A	122.8
C6—C1—H1A	120.3	N2—C10—H10B	115.7
C1—C2—C3	120.25 (18)	C11'—C10—H10B	116.1
C1—C2—H2A	119.9	C11—C10—H10B	129.6
C3—C2—H2A	119.9	C11—O1—C14	108.6 (6)
C2—C3—C4	121.21 (18)	O1—C11—C12	106.8 (7)
C2—C3—H3A	119.4	O1—C11—C10	126.8 (9)
C4—C3—H3A	119.4	C12—C11—C10	126.3 (11)
C5—C4—C3	117.87 (18)	C11—C12—C13	104.3 (13)
C5—C4—C7	120.78 (17)	C11—C12—H12A	127.9
C3—C4—C7	121.35 (17)	C13—C12—H12A	127.9
C4—C5—C6	120.78 (18)	C14—C13—C12	109.3 (13)
C4—C5—H5A	119.6	C14—C13—H13A	125.4
C6—C5—H5A	119.6	C12—C13—H13A	125.4
C1—C6—C5	120.59 (19)	C13—C14—O1	111.0 (9)
C1—C6—H6A	119.7	C13—C14—H14A	124.5
C5—C6—H6A	119.7	O1—C14—H14A	124.5
C4—C7—S2	107.45 (12)	C11'—O1'—C14'	105.9 (6)
C4—C7—H7A	110.2	O1'—C11'—C10	120.0 (10)
S2—C7—H7A	110.2	O1'—C11'—C12'	111.8 (9)
C4—C7—H7B	110.2	C10—C11'—C12'	127.8 (13)
S2—C7—H7B	110.2	C11'—C12'—C13'	101.9 (15)
H7A—C7—H7B	108.5	C11'—C12'—H12B	129.1
N1—C8—S1	123.08 (14)	C13'—C12'—H12B	129.1
N1—C8—S2	113.05 (13)	C14'—C13'—C12'	108.9 (12)
S1—C8—S2	123.87 (11)	C14'—C13'—H13B	125.5
N1—C9—H9A	109.5	C12'—C13'—H13B	125.5

N1—C9—H9B	109.5	C13'—C14'—O1'	111.5 (9)
H9A—C9—H9B	109.5	C13'—C14'—H14B	124.3
N1—C9—H9C	109.5	O1'—C14'—H14B	124.3

Hydrogen-bond geometry (Å, °)

Table 1. Cg is the centroid of the phenyl ring

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C7—H7B...Cg ⁱ	0.97	2.88	3.560 (2)	128
C13—H13A...Cg ⁱⁱ	0.93	2.80	3.62 (2)	149

Symmetry codes: (i) $-x-1, -y, -z+1$; (ii) $-x+1, -y, -z+2$.