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# *N*-Benzoyl-*N*-(1,4-dioxonaphthalen-2-yl)benzamide

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Key indicators: single-crystal X-ray study; T = 123 K; mean  $\sigma$ (C–C) = 0.002 Å; R factor = 0.070; wR factor = 0.221; data-to-parameter ratio = 50.7.

The title molecule,  $C_{24}H_{15}NO_4$ , crystallizes with two molecules in the asymmetric unit (Z' = 2). For both molecules, the two amide groups are not coplanar, as the dihedral angles of the respective NCO groups are similar at 50.37 (14) and 51.22 (13)°. However, the orientations of the substituent phenyl rings with the central naphthalene system are significantly different for the two molecules; for one molecule, these dihedral angles are 80.29 (3) and 80.95 (4)°, while for the second molecule they are 86.63 (3) and 72.82 (4)°. The crystal packing shows the molecules to be linked by weak C–H···O interactions.

#### **Related literature**

For related structures, see: Akinboye, Butcher, Brandy *et al.* (2009); Akinboye, Butcher, Wright *et al.* (2009). For pharmacological properties of related compounds, see: Bakare *et al.* (2003); Khraiwesh *et al.* (2011); Berhe *et al.* (2008).



## Experimental

Crystal data

	0
$C_{24}H_{15}NO_4$	c = 15.6472 (14) A
$M_r = 381.37$	$\alpha = 90.735 \ (7)^{\circ}$
Triclinic, P1	$\beta = 98.213 \ (9)^{\circ}$
a = 9.8704 (11)  Å	$\gamma = 108.187 \ (9)^{\circ}$
b = 12.6776 (11)  Å	V = 1837.6 (3) Å <sup>3</sup>

#### Z = 4Mo $K\alpha$ radiation $\mu = 0.10 \text{ mm}^{-1}$

#### Data collection

Oxford Diffraction Xcalibur Ruby	Diffraction, 2007)
Gemini diffractometer	$T_{\min} = 0.751, T_{\max} = 1.000$
Absorption correction: multi-scan	26547 measured reflections
(CrysAlis PRO; Oxford	26547 independent reflections
	17970 reflections with $I > 2\sigma(I)$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.070$  $wR(F^2) = 0.221$ S = 1.0326547 reflections

 Table 1

 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C21A - H21A \cdots O2B^{i}$	0.95	2.41	3.1270 (19)	132
$C14B - H14B \cdots O3A^{ii}$	0.95	2.50	3.4345 (17)	168
$C15B - H15B \cdots O4A^{iii}$	0.95	2.62	3.4638 (17)	148
$C22B - H22B \cdot \cdot \cdot O2A$	0.95	2.54	3.2054 (17)	127
$C14A - H14A \cdot \cdot \cdot O3B$	0.95	2.44	3.2303 (16)	140

Symmetry codes: (i) x, y + 1, z; (ii) x, y - 1, z; (iii) -x + 1, -y + 1, -z + 2.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2007); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: JJ2146).

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 $0.52 \times 0.28 \times 0.15 \text{ mm}$ 

T = 123 K

524 parameters

 $\Delta \rho_{\rm max} = 0.70 \ {\rm e} \ {\rm \AA}^-$ 

 $\Delta \rho_{\rm min} = -0.43 \text{ e} \text{ Å}^{-3}$ 

H-atom parameters constrained

# supporting information

Acta Cryst. (2012). E68, o2379 [https://doi.org/10.1107/S1600536812030231] N-Benzoyl-N-(1,4-dioxonaphthalen-2-yl)benzamide Yakini Brandy, Ray J. Butcher and Oladapo Bakare

#### S1. Comment

We have previously synthesized some cyclic and acyclic imido-3-chloro-naphthoquinone analogs and studied their anticancer activities against some prostate cancer cell lines (Bakare *et al.* (2003); Berhe *et al.* (2008)). In addition, some aryl imido -2-chloro and -2-bromo-1,4-naphthoquinone analogs were synthesized and some crystal structures were already reported for the 2-*N*-bis(2-chlorobenzoyl)amino and 2-*N*-bis(4-fluorobenzoyl) amino-3-bromo-1,4-naphtho-quinone derivatives (Akinboye, Butcher, Brandy *et al.* (2009); Akinboye, Butcher, Wright *et al.* (2009)). Recently, we have reported some antiparasitic studies of some of the aryl imidonaphthoquinones (Khraiwesh, *et al.*, (2011)) and are currently studying their anticancer properties against PC3 prostate cancer cells. In continuation of our work, *N*-benzoyl-*N*-(1,4-dioxonaphthalen-2-yl)benzamide was synthesized as a potential antiprostate cancer agent and its structure is reported here.

In the structure of the title compound there are two molecules in the asymmetric unit (Z' = 2). For both molecules the two amide moieties are not coplanar as the dihedral angles of the respective NCO groups are similar at 50.37 (14)° and 51.22 (13)°). However, the orientations of the substituent phenyl rings with the central naphthyl ring are significantly different for the two molecules. For molecule A these dihedral angles are 80.29 (3)° and 80.95 (4)° between the naphthyl ring and rings C12 - C17 and C19 - C24, while for molecule B these values are 86.63 (3)° and 72.82 (4)°.

In the two molecules the imide C=O's are at dihedral angles of  $42.95 (17)^{\circ}$  and  $-121.12 (13)^{\circ}$  for molecule A and  $-44.89 (17)^{\circ}$  and  $121.05 (13)^{\circ}$  for molecule B and are thus anti to each other. The anti conformation of these imide C=O's and the dihedral angles between the phenyl rings and imide C=O groups oriented the phenyl groups away from each other (values of  $(34.2 (2)^{\circ}$  and  $25.29 (18)^{\circ}$ ) for A and -24.03 (17) and  $-42.44 (19)^{\circ}$  for B). The crystal structure packing pattern shows the molecules were linked by weak intermolecular C—H…O interactions.

#### **S2. Experimental**

2-Amino-1,4-naphthoquinone (318 mg, 1.83 mmol) was dissolved in freshly distilled THF (15 ml). NaH (115 mg, 4.78 mmol) was added and the mixture was stirred at room temperature for 15 min. The appropriate benzoyl chloride (0.55 ml, 4.74 mmol) was added, drop wise, and the mixture was stirred for 24 h. THF was evaporated under vacuum and the mixture was washed with ice-water (10 g ice in 10 ml water). The aqueous mixture was extracted with  $CH_2Cl_2$  (30 ml, 20 ml consecutively) and the combined organic phase washed with water (3 *x* 20 ml), saturated NaCl solution (20 ml), then dried over anhydrous MgSO<sub>4</sub>. The crude was purified *via* tirturating in ethanol (2 ml) and column chromatography with an eluent mixture of ethyl acetate and hexane to furnish the imide(70 mg, 10%).

#### **S3. Refinement**

H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms with a C—H distance of 0.95 and  $U_{iso}(H) = 1.2U_{eq}(C)$ . The structure was a non-merohedral twin and was refined with a BASF (batch



scale factor) value of 0.4084 (7) and a twin law for the two components of -1 0 0, 0 - 1 0, 0 0 1.

Figure 1

Diagram of  $C_{24}H_{15}NO_4$  showing the atom labeling for the two molecules in the asymmetric unit. The C—H···O interactions are shown by dashed lines. Thermal ellipsoids are at the 30% probability level.



### Figure 2

The molecular packing for C<sub>24</sub>H<sub>15</sub>NO<sub>4</sub> viewed along the *a* axis. The C—H…O interactions are shown by dashed lines.

N-Benzoyl-N-(1,4-dioxonaphthalen-2-yl)benzamide

Crystal data C<sub>24</sub>H<sub>15</sub>NO<sub>4</sub>  $M_r = 381.37$ Triclinic,  $P\overline{1}$ Hall symbol: -P 1 a = 9.8704 (11) Å b = 12.6776 (11) Å c = 15.6472 (14) Å a = 90.735 (7)°  $\beta = 98.213$  (9)°  $\gamma = 108.187$  (9)° V = 1837.6 (3) Å<sup>3</sup>

Z = 4 F(000) = 792  $D_x = 1.378 \text{ Mg m}^{-3}$ Mo K $\alpha$  radiation,  $\lambda = 0.71073 \text{ Å}$ Cell parameters from 3142 reflections  $\theta = 3.0-34.9^{\circ}$   $\mu = 0.10 \text{ mm}^{-1}$ T = 123 K Block, colorless  $0.52 \times 0.28 \times 0.15 \text{ mm}$  Data collection

Oxford Diffraction Xcalibur Ruby Gemini diffractometer Radiation source: Enhance (Mo) X-ray Source Graphite monochromator Detector resolution: 10.5081 pixels mm <sup>-1</sup> $\omega$ scans Absorption correction: multi-scan ( <i>CrysAlis PRO</i> ; Oxford Diffraction, 2007) $T_{\min} = 0.751, T_{\max} = 1.000$	26547 measured reflections 26547 independent reflections 17970 reflections with $I > 2\sigma(I)$ $R_{int} = 0.0000$ $\theta_{max} = 35.2^{\circ}, \theta_{min} = 3.1^{\circ}$ $h = -15 \rightarrow 15$ $k = -20 \rightarrow 20$ $l = -24 \rightarrow 21$
Refinement	
Refinement on $F^2$ Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.070$ $wR(F^2) = 0.221$ S = 1.03 26547 reflections 524 parameters 0 restraints Primary atom site location: structure-invariant direct methods	Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.1412P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.70$ e Å <sup>-3</sup> $\Delta\rho_{min} = -0.43$ e Å <sup>-3</sup>

### Special details

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(Å^2)$ 

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
01A	0.61934 (11)	0.65075 (8)	1.11669 (6)	0.0244 (2)	
O2A	0.15230 (10)	0.64038 (8)	0.89442 (6)	0.02104 (19)	
O3A	0.50651 (10)	1.01393 (8)	0.90297 (6)	0.02088 (19)	
O4A	0.18605 (11)	0.84042 (8)	1.00964 (6)	0.02152 (19)	
N1A	0.36272 (12)	0.84143 (9)	0.93028 (7)	0.0166 (2)	
C1A	0.37834 (13)	0.74566 (10)	0.97147 (7)	0.0153 (2)	
C2A	0.49758 (14)	0.74697 (10)	1.02443 (8)	0.0173 (2)	
H2AA	0.5767	0.8139	1.0342	0.021*	
C3A	0.50990 (14)	0.64715 (10)	1.06827 (8)	0.0173 (2)	
C4A	0.38274 (13)	0.54455 (10)	1.05355 (7)	0.0160 (2)	
C5A	0.38512 (15)	0.45054 (11)	1.09716 (9)	0.0209 (2)	
H5AA	0.4686	0.4515	1.1362	0.025*	
C6A	0.26582 (16)	0.35463 (11)	1.08401 (9)	0.0240 (3)	
H6AA	0.2685	0.2902	1.1137	0.029*	
C7A	0.14371 (16)	0.35298 (11)	1.02791 (9)	0.0230 (3)	
H7AA	0.0620	0.2877	1.0195	0.028*	

C8A	0.14043 (15)	0.44648 (11)	0.98398 (8)	0.0198 (2)
H8AA	0.0567	0.4450	0.9449	0.024*
C9A	0.25916 (14)	0.54244 (10)	0.99677 (7)	0.0158 (2)
C10A	0.25331 (13)	0.64191 (10)	0.94990 (7)	0.0155 (2)
C11A	0.47714 (14)	0.91402 (10)	0.89268 (7)	0.0165 (2)
C12A	0.55113 (14)	0.86068 (10)	0.83735 (8)	0.0166 (2)
C13A	0.47877 (15)	0.75844 (11)	0.79278 (8)	0.0187 (2)
H13A	0.3835	0.7184	0.8017	0.022*
C14A	0.54511 (16)	0.71467 (12)	0.73529 (8)	0.0226 (3)
H14A	0.4952	0.6450	0.7046	0.027*
C15A	0.68457 (17)	0.77295 (13)	0.72280 (8)	0.0251 (3)
H15A	0.7307	0.7429	0.6840	0.030*
C16A	0.75694 (16)	0.87551 (13)	0.76714 (9)	0.0253(3)
H16A	0.8525	0.9152	0.7585	0.030*
C17A	0.69028 (15)	0.92022 (11)	0.82395 (8)	0.0207(2)
H17A	0.7392	0.9908	0.8534	0.025*
C18A	0.24559(14)	0.87524 (10)	0.94867 (8)	0.0169(2)
C19A	0.19812(14)	0.95011(11)	0 88779 (9)	0.0206(2)
C20A	0.20504(17)	0.94166 (13)	0 79946 (9)	0.0200(2) 0.0271(3)
H20A	0.2413	0.8876	0.7771	0.033*
C21A	0.15885 (19)	1.01249(15)	0.74495(11)	0.035 0.0367(4)
H21A	0.1623	1.0066	0.6848	0.0307 (1)
C22A	0.1023	1.0000	0.77739(13)	0.0410(4)
H22A	0.0791	1 1415	0.7397	0.0410 (4)
C23A	0.0770 (10)	1.1415 1 00017 (14)	0.86423 (13)	0.049
H23A	0.05770 (15)	1.05517 (14)	0.8860	0.0389 (4)
C24A	0.14178 (16)	1.1332 1.02722 (12)	0.0000	0.047 0.0268 (3)
H24A	0.1333	1.02/22 (12)	0.9793	0.0208 (5)
01R	0.1333	0.13107(0)	0.37332 (6)	0.032
O1D O2B	0.00121(11) 0.18800(11)	0.13107(9) 0.14122(8)	0.59532 (0)	0.0239(2)
02B	0.18809(11) 0.53144(11)	0.14122(8) 0.51504(8)	0.57590(0) 0.50071(6)	0.02112(19)
030	0.33144(11) 0.19979(11)	0.31394(8) 0.32712(8)	0.39971(0)	0.01901(18)
U4D N1D	0.18878(11) 0.28724(11)	0.32712(6) 0.22871(0)	0.43919(0) 0.55724(6)	0.0252(2)
	0.38724(11) 0.20502(12)	0.33871(9)	0.55754(0) 0.51797(7)	0.01398(19)
CIB	0.39302(13)	0.23977(10) 0.22682(11)	0.51/8/(7)	0.0149(2)
	0.50258 (14)	0.23682 (11)	0.47347 (8)	0.0181 (2)
H2BA C2D	0.5809	0.3029	0.4738	$0.022^{*}$
C3B	0.50210 (14)	0.13280 (11)	0.43110 (8)	0.0176(2)
C4B	0.3/39/(14)	0.03230(10)	0.4314/(/)	0.0170(2)
COB	0.36221 (16)	-0.06502 (11)	0.38440 (8)	0.0216 (3)
НЭВА	0.4353	-0.06/0	0.3515	0.026*
C6B	0.24392 (16)	-0.15902 (12)	0.38559 (9)	0.0248 (3)
H6BA	0.2356	-0.2249	0.3528	0.030*
C7B	0.13726 (15)	-0.15763 (11)	0.43448 (9)	0.0242 (3)
H'/BA	0.0570	-0.2226	0.4354	0.029*
C8B	0.14839 (15)	-0.06103 (11)	0.48195 (9)	0.0215 (2)
H8BA	0.0764	-0.0600	0.5160	0.026*
C9B	0.26546 (13)	0.03420 (10)	0.47945 (8)	0.0163 (2)
C10B	0.27458 (13)	0.13758 (10)	0.52801 (7)	0.0157 (2)

C11B	0.50716 (13)	0.41715 (10)	0.60839 (7)	0.0155 (2)
C12B	0.59360 (13)	0.37346 (10)	0.67570 (7)	0.0164 (2)
C13B	0.53263 (14)	0.27461 (11)	0.71402 (8)	0.0179 (2)
H13B	0.4363	0.2297	0.6932	0.021*
C14B	0.61293 (16)	0.24183 (12)	0.78280 (8)	0.0226 (3)
H14B	0.5723	0.1742	0.8088	0.027*
C15B	0.75295 (17)	0.30876 (13)	0.81313 (8)	0.0258 (3)
H15B	0.8081	0.2866	0.8601	0.031*
C16B	0.81343 (16)	0.40773 (13)	0.77565 (8)	0.0247 (3)
H16B	0.9095	0.4529	0.7968	0.030*
C17B	0.73313 (14)	0.44047 (11)	0.70715 (8)	0.0199 (2)
H17B	0.7736	0.5087	0.6818	0.024*
C18B	0.26274 (14)	0.36802 (10)	0.52716 (8)	0.0174 (2)
C19B	0.22434 (13)	0.44303 (10)	0.58654 (8)	0.0176 (2)
C20B	0.23316 (15)	0.42713 (11)	0.67508 (8)	0.0211 (2)
H20B	0.2701	0.3713	0.6990	0.025*
C21B	0.18735 (16)	0.49364 (13)	0.72766 (9)	0.0251 (3)
H21B	0.1918	0.4826	0.7878	0.030*
C22B	0.13520 (16)	0.57591 (13)	0.69353 (10)	0.0266 (3)
H22B	0.1049	0.6215	0.7303	0.032*
C23B	0.12693 (16)	0.59214 (12)	0.60551 (10)	0.0257 (3)
H23B	0.0923	0.6493	0.5821	0.031*
C24B	0.16968 (15)	0.52406 (11)	0.55198 (9)	0.0215 (3)
H24B	0.1613	0.5332	0.4915	0.026*

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1A	0.0214 (5)	0.0244 (5)	0.0260 (5)	0.0075 (4)	-0.0014 (4)	0.0038 (4)
O2A	0.0221 (5)	0.0188 (4)	0.0199 (4)	0.0053 (4)	-0.0014 (3)	0.0022 (3)
O3A	0.0260 (5)	0.0144 (4)	0.0220 (4)	0.0050 (4)	0.0060 (4)	0.0017 (3)
O4A	0.0227 (5)	0.0232 (5)	0.0195 (4)	0.0068 (4)	0.0074 (3)	0.0027 (4)
N1A	0.0217 (5)	0.0134 (4)	0.0162 (4)	0.0065 (4)	0.0057 (4)	0.0035 (4)
C1A	0.0192 (6)	0.0133 (5)	0.0144 (5)	0.0052 (4)	0.0049 (4)	0.0027 (4)
C2A	0.0181 (6)	0.0146 (5)	0.0184 (5)	0.0036 (4)	0.0038 (4)	0.0004 (4)
C3A	0.0185 (6)	0.0174 (5)	0.0163 (5)	0.0056 (4)	0.0034 (4)	0.0007 (4)
C4A	0.0178 (5)	0.0156 (5)	0.0154 (5)	0.0060 (4)	0.0041 (4)	0.0022 (4)
C5A	0.0237 (6)	0.0189 (6)	0.0217 (6)	0.0089 (5)	0.0033 (5)	0.0047 (5)
C6A	0.0289 (7)	0.0163 (6)	0.0272 (6)	0.0067 (5)	0.0062 (5)	0.0085 (5)
C7A	0.0245 (7)	0.0149 (5)	0.0287 (7)	0.0036 (5)	0.0074 (5)	0.0034 (5)
C8A	0.0197 (6)	0.0171 (5)	0.0219 (6)	0.0054 (5)	0.0025 (4)	0.0007 (4)
C9A	0.0191 (6)	0.0141 (5)	0.0151 (5)	0.0060 (4)	0.0040 (4)	0.0011 (4)
C10A	0.0193 (6)	0.0139 (5)	0.0140 (5)	0.0060 (4)	0.0035 (4)	0.0007 (4)
C11A	0.0196 (6)	0.0161 (5)	0.0134 (5)	0.0052 (4)	0.0027 (4)	0.0022 (4)
C12A	0.0211 (6)	0.0157 (5)	0.0142 (5)	0.0069 (4)	0.0042 (4)	0.0027 (4)
C13A	0.0243 (6)	0.0175 (5)	0.0153 (5)	0.0073 (5)	0.0044 (4)	0.0017 (4)
C14A	0.0323 (7)	0.0227 (6)	0.0156 (5)	0.0127 (5)	0.0037 (5)	-0.0005 (5)
C15A	0.0330 (8)	0.0327 (7)	0.0180 (6)	0.0197 (6)	0.0094 (5)	0.0056 (5)

# supporting information

C16A	0.0239 (7)	0.0294 (7)	0.0258 (6)	0.0100 (6)	0.0106 (5)	0.0083 (5)
C17A	0.0222 (6)	0.0197 (6)	0.0205 (6)	0.0054 (5)	0.0064 (5)	0.0051 (5)
C18A	0.0198 (6)	0.0138 (5)	0.0170 (5)	0.0054 (4)	0.0025 (4)	0.0001 (4)
C19A	0.0194 (6)	0.0186 (6)	0.0239 (6)	0.0067 (5)	0.0023 (5)	0.0040 (5)
C20A	0.0285 (7)	0.0295 (7)	0.0232 (6)	0.0092 (6)	0.0032 (5)	0.0074 (5)
C21A	0.0333 (9)	0.0418 (9)	0.0315 (8)	0.0086 (7)	0.0001 (6)	0.0172 (7)
C22A	0.0337 (9)	0.0283 (8)	0.0546 (11)	0.0075 (7)	-0.0095 (8)	0.0182 (8)
C23A	0.0348 (9)	0.0237 (7)	0.0571 (11)	0.0149 (7)	-0.0085 (8)	0.0021 (7)
C24A	0.0244 (7)	0.0212 (6)	0.0355 (7)	0.0105 (5)	-0.0001 (6)	-0.0019 (6)
O1B	0.0253 (5)	0.0267 (5)	0.0251 (5)	0.0130 (4)	0.0103 (4)	0.0033 (4)
O2B	0.0216 (5)	0.0201 (4)	0.0223 (4)	0.0056 (4)	0.0079 (4)	-0.0004 (4)
O3B	0.0238 (5)	0.0150 (4)	0.0200 (4)	0.0059 (3)	0.0039 (3)	0.0006 (3)
O4B	0.0240 (5)	0.0264 (5)	0.0197 (4)	0.0112 (4)	-0.0021 (4)	-0.0034 (4)
N1B	0.0171 (5)	0.0149 (5)	0.0164 (4)	0.0062 (4)	0.0018 (4)	-0.0007 (4)
C1B	0.0176 (5)	0.0134 (5)	0.0146 (5)	0.0066 (4)	0.0015 (4)	0.0004 (4)
C2B	0.0198 (6)	0.0179 (5)	0.0178 (5)	0.0070 (5)	0.0047 (4)	0.0020 (4)
C3B	0.0201 (6)	0.0198 (6)	0.0161 (5)	0.0104 (5)	0.0038 (4)	0.0029 (4)
C4B	0.0213 (6)	0.0180 (5)	0.0144 (5)	0.0105 (5)	0.0012 (4)	0.0011 (4)
C5B	0.0271 (7)	0.0208 (6)	0.0199 (6)	0.0125 (5)	0.0025 (5)	-0.0024 (5)
C6B	0.0307 (7)	0.0189 (6)	0.0258 (6)	0.0116 (5)	-0.0004 (5)	-0.0050 (5)
C7B	0.0218 (6)	0.0157 (6)	0.0317 (7)	0.0040 (5)	-0.0021 (5)	-0.0029 (5)
C8B	0.0191 (6)	0.0183 (6)	0.0269 (6)	0.0066 (5)	0.0017 (5)	-0.0015 (5)
C9B	0.0177 (6)	0.0153 (5)	0.0169 (5)	0.0073 (4)	0.0007 (4)	-0.0002 (4)
C10B	0.0166 (5)	0.0154 (5)	0.0157 (5)	0.0061 (4)	0.0017 (4)	0.0004 (4)
C11B	0.0172 (5)	0.0167 (5)	0.0137 (5)	0.0063 (4)	0.0035 (4)	0.0001 (4)
C12B	0.0183 (6)	0.0184 (5)	0.0137 (5)	0.0073 (4)	0.0029 (4)	-0.0001 (4)
C13B	0.0212 (6)	0.0170 (5)	0.0157 (5)	0.0057 (4)	0.0042 (4)	0.0007 (4)
C14B	0.0344 (7)	0.0233 (6)	0.0149 (5)	0.0148 (6)	0.0066 (5)	0.0034 (5)
C15B	0.0326 (8)	0.0350 (8)	0.0153 (5)	0.0194 (6)	0.0017 (5)	0.0035 (5)
C16B	0.0233 (7)	0.0324 (7)	0.0184 (6)	0.0104 (6)	0.0000 (5)	0.0013 (5)
C17B	0.0203 (6)	0.0216 (6)	0.0171 (5)	0.0063 (5)	0.0018 (4)	0.0000 (4)
C18B	0.0172 (6)	0.0171 (5)	0.0192 (5)	0.0073 (4)	0.0025 (4)	0.0017 (4)
C19B	0.0159 (5)	0.0179 (5)	0.0200 (5)	0.0065 (4)	0.0029 (4)	-0.0016 (4)
C20B	0.0207 (6)	0.0225 (6)	0.0212 (6)	0.0079 (5)	0.0047 (5)	0.0007 (5)
C21B	0.0239 (7)	0.0299 (7)	0.0228 (6)	0.0088 (6)	0.0071 (5)	-0.0020 (5)
C22B	0.0228 (7)	0.0274 (7)	0.0308 (7)	0.0103 (5)	0.0045 (5)	-0.0098 (6)
C23B	0.0243 (7)	0.0239 (7)	0.0314 (7)	0.0134 (5)	-0.0001 (5)	-0.0030 (5)
C24B	0.0198 (6)	0.0228 (6)	0.0229 (6)	0.0097 (5)	-0.0001 (5)	-0.0003 (5)

Geometric parameters (Å, °)

O1A—C3A	1.2154 (16)	O1B—C3B	1.2188 (15)
O2A-C10A	1.2199 (15)	O2B—C10B	1.2243 (15)
O3A—C11A	1.2109 (15)	O3B—C11B	1.2127 (15)
O4A-C18A	1.2080 (15)	O4B—C18B	1.2086 (15)
N1A—C18A	1.4158 (16)	N1B—C18B	1.4125 (16)
N1A—C1A	1.4218 (15)	N1B—C11B	1.4152 (16)
N1A—C11A	1.4226 (16)	N1B—C1B	1.4192 (15)

C1A—C2A	1 3345 (17)	C1B—C2B	1 3395 (17)
C1A - C10A	1.4911 (18)	C1B— $C10B$	1.4882 (17)
$C^2A - C^3A$	1 4769 (18)	$C^{2}B-C^{3}B$	14802(17)
$C_{2A}$ H2AA	0.9500	C2B—H2BA	0.9500
$C_{3A}$ $C_{4A}$	1 4869 (18)	C3B - C4B	14897(19)
$C_{4}$	1.1009(10) 1.3858(17)	C4B-C5B	1.1037(17) 1.3033(17)
C4A - C9A	1.3055 (17)	C4B-C9B	1.3933(17) 1 3988 (17)
$C_{1}^{5}$	1.3935(17) 1.392(2)	$C_{5B}$ $C_{6B}$	1.3960(17)
$C_{5A} = C_{6A}$	0.9500	C5B + C5B	0.9500
	1.320(2)	C6P C7P	0.9500
$C_{A}$	1.560 (2)		1.391 (2)
	0.9300		0.9300
C/A = C8A	1.3846 (18)		1.3903 (18)
C/A - H/AA	0.9500	C/B - H/BA	0.9500
C8A—C9A	1.3893 (18)	C8B—C9B	1.3920 (18)
C8A—H8AA	0.9500	C8B—H8BA	0.9500
C9A—C10A	1.4793 (17)	C9B—C10B	1.4780 (16)
C11A—C12A	1.4880 (17)	C11B—C12B	1.4824 (17)
C12A—C13A	1.3904 (18)	C12B—C17B	1.3877 (18)
C12A—C17A	1.3926 (18)	C12B—C13B	1.3942 (18)
C13A—C14A	1.3886 (18)	C13B—C14B	1.3901 (18)
C13A—H13A	0.9500	C13B—H13B	0.9500
C14A—C15A	1.387 (2)	C14B—C15B	1.388 (2)
C14A—H14A	0.9500	C14B—H14B	0.9500
C15A—C16A	1.392 (2)	C15B—C16B	1.388 (2)
C15A—H15A	0.9500	C15B—H15B	0.9500
C16A—C17A	1.3897 (18)	C16B—C17B	1.3869 (18)
C16A—H16A	0.9500	C16B—H16B	0.9500
C17A—H17A	0.9500	C17B—H17B	0.9500
C18A—C19A	1.4836 (18)	C18B—C19B	1.4881 (17)
C19A—C24A	1.3860 (19)	C19B—C24B	1.3841 (18)
C19A—C20A	1.398 (2)	C19B—C20B	1.3967 (18)
C20A—C21A	1.380 (2)	C20B—C21B	1.3852 (18)
C20A—H20A	0.9500	C20B—H20B	0.9500
$C_{21A}$ $C_{22A}$	1 376 (3)	C21B—C22B	1.382(2)
$C_{21A}$ H21A	0.9500	C21B—H21B	0.9500
$C^{22}\Delta - C^{23}\Delta$	1.381(3)	$C_{22}B - C_{23}B$	1.390(2)
$C_{22}A = H_{22}A$	0.9500	C22B H22B	0.9500
$C_{22}A = C_{24}A$	1.301(2)	C22B C24B	1 3007 (10)
$C_{23A} = C_{24A}$	0.0500	C23B H23B	0.0500
C24A H24A	0.9500	C24P H24P	0.9500
С24А—п24А	0.9300	С24Б—п24Б	0.9300
C18A—N1A—C1A	115.64 (10)	C18B—N1B—C11B	119.88 (10)
C18A—N1A—C11A	121.20 (10)	C18B—N1B—C1B	115.62 (10)
C1A—N1A—C11A	121.16 (11)	C11B—N1B—C1B	123.00 (10)
C2A— $C1A$ — $N1A$	122.76 (11)	C2B-C1B-N1B	123 34 (11)
C2A— $C1A$ — $C10A$	121.76 (11)	C2B— $C1B$ — $C10B$	121.83(11)
NIA—CIA—CI0A	115 46 (11)	N1B— $C1B$ — $C10B$	114 83 (10)
C1A - C2A - C3A	121.80 (11)	C1B— $C2B$ — $C3B$	121.46 (12)

С1А—С2А—Н2АА	119.1	C1B—C2B—H2BA	119.3
СЗА—С2А—Н2АА	119.1	C3B—C2B—H2BA	119.3
O1A—C3A—C2A	120.50 (12)	O1B—C3B—C2B	120.78 (12)
O1A—C3A—C4A	121.68 (12)	O1B—C3B—C4B	121.54 (12)
C2A—C3A—C4A	117.79 (11)	C2B—C3B—C4B	117.63 (11)
С5А—С4А—С9А	119.36 (12)	C5B—C4B—C9B	119.45 (12)
C5A—C4A—C3A	120.22 (11)	C5B—C4B—C3B	119.95 (12)
C9A—C4A—C3A	120.41 (11)	C9B—C4B—C3B	120.60 (11)
C4A—C5A—C6A	120.35 (13)	C6B—C5B—C4B	120.00 (13)
С4А—С5А—Н5АА	119.8	C6B—C5B—H5BA	120.0
С6А—С5А—Н5АА	119.8	C4B—C5B—H5BA	120.0
C7A—C6A—C5A	120.09 (12)	C5B—C6B—C7B	120.52 (12)
С7А—С6А—Н6АА	120.0	C5B—C6B—H6BA	119.7
С5А—С6А—Н6АА	120.0	C7B—C6B—H6BA	119.7
C6A—C7A—C8A	119.95 (13)	C8B—C7B—C6B	119.88 (13)
С6А—С7А—Н7АА	120.0	C8B—C7B—H7BA	120.1
С8А—С7А—Н7АА	120.0	C6B—C7B—H7BA	120.1
C7A—C8A—C9A	120.24 (13)	C7B—C8B—C9B	119.75 (13)
С7А—С8А—Н8АА	119.9	C7B—C8B—H8BA	120.1
С9А—С8А—Н8АА	119.9	C9B—C8B—H8BA	120.1
C8A—C9A—C4A	120.00 (11)	C8B—C9B—C4B	120.37 (11)
C8A—C9A—C10A	119.53 (11)	C8B—C9B—C10B	119.53 (11)
C4A—C9A—C10A	120.48 (11)	C4B—C9B—C10B	120.10 (11)
O2A—C10A—C9A	122.29 (11)	O2B—C10B—C9B	122.45 (11)
O2A—C10A—C1A	120.24 (11)	O2B—C10B—C1B	119.78 (11)
C9A—C10A—C1A	117.45 (11)	C9B—C10B—C1B	117.77 (11)
O3A—C11A—N1A	120.52 (11)	O3B—C11B—N1B	120.39 (11)
O3A—C11A—C12A	122.84 (12)	O3B—C11B—C12B	122.35 (11)
N1A—C11A—C12A	116.56 (11)	N1B-C11B-C12B	117.14 (11)
C13A—C12A—C17A	120.08 (12)	C17B—C12B—C13B	120.25 (12)
C13A—C12A—C11A	121.06 (12)	C17B—C12B—C11B	117.80 (12)
C17A—C12A—C11A	118.55 (12)	C13B—C12B—C11B	121.52 (11)
C14A—C13A—C12A	120.29 (13)	C14B—C13B—C12B	119.90 (13)
C14A—C13A—H13A	119.9	C14B—C13B—H13B	120.1
C12A—C13A—H13A	119.9	C12B—C13B—H13B	120.1
C15A—C14A—C13A	119.78 (13)	C15B—C14B—C13B	119.42 (13)
C15A—C14A—H14A	120.1	C15B—C14B—H14B	120.3
C13A—C14A—H14A	120.1	C13B—C14B—H14B	120.3
C14A—C15A—C16A	119.99 (12)	C14B—C15B—C16B	120.80 (13)
C14A—C15A—H15A	120.0	C14B—C15B—H15B	119.6
C16A—C15A—H15A	120.0	C16B—C15B—H15B	119.6
C17A—C16A—C15A	120.40 (13)	C17B—C16B—C15B	119.74 (14)
C17A—C16A—H16A	119.8	C17B—C16B—H16B	120.1
C15A—C16A—H16A	119.8	C15B—C16B—H16B	120.1
C16A—C17A—C12A	119.45 (13)	C16B—C17B—C12B	119.88 (13)
C16A—C17A—H17A	120.3	C16B—C17B—H17B	120.1
C12A—C17A—H17A	120.3	C12B—C17B—H17B	120.1
O4A—C18A—N1A	120.27 (11)	O4B—C18B—N1B	120.19 (11)

O4A—C18A—C19A	123.00 (12)	O4B—C18B—C19B	123.43 (12)
N1A—C18A—C19A	116.68 (11)	N1B—C18B—C19B	116.20 (11)
C24A—C19A—C20A	120.03 (13)	C24B—C19B—C20B	120.16 (12)
C24A—C19A—C18A	118.57 (12)	C24B—C19B—C18B	118.96 (11)
C20A-C19A-C18A	121.36 (12)	C20B-C19B-C18B	120.72 (11)
C21A—C20A—C19A	119.50 (15)	C21B—C20B—C19B	119.25 (13)
C21A—C20A—H20A	120.3	C21B—C20B—H20B	120.4
C19A—C20A—H20A	120.3	C19B—C20B—H20B	120.4
C22A—C21A—C20A	120.30 (16)	C22B—C21B—C20B	120.66 (13)
C22A—C21A—H21A	119.8	C22B—C21B—H21B	119.7
C20A—C21A—H21A	119.8	C20B—C21B—H21B	119.7
C21A—C22A—C23A	120.65 (15)	C21B—C22B—C23B	120.15 (13)
C21A—C22A—H22A	119.7	C21B—C22B—H22B	119.9
C23A—C22A—H22A	119.7	C23B—C22B—H22B	119.9
C22A—C23A—C24A	119.68 (16)	C22B—C23B—C24B	119.52 (13)
C22A—C23A—H23A	120.2	C22B—C23B—H23B	120.2
C24A—C23A—H23A	120.2	C24B—C23B—H23B	120.2
C19A—C24A—C23A	119.77 (15)	C19B—C24B—C23B	120.22 (13)
C19A—C24A—H24A	120.1	C19B—C24B—H24B	119.9
C23A—C24A—H24A	120.1	C23B—C24B—H24B	119.9
C18A—N1A—C1A—C2A	-121.12 (13)	C18B—N1B—C1B—C2B	121.05 (13)
C11A—N1A—C1A—C2A	42.95 (17)	C11B—N1B—C1B—C2B	-44.89 (17)
C18A—N1A—C1A—C10A	60.33 (14)	C18B—N1B—C1B—C10B	-59.77 (14)
C11A—N1A—C1A—C10A	-135.60 (11)	C11B—N1B—C1B—C10B	134.29 (12)
N1A—C1A—C2A—C3A	178.38 (11)	N1B—C1B—C2B—C3B	-176.55 (11)
C10A—C1A—C2A—C3A	-3.15 (18)	C10B—C1B—C2B—C3B	4.32 (18)
C1A—C2A—C3A—O1A	-179.48 (12)	C1B—C2B—C3B—O1B	-179.43 (12)
C1A—C2A—C3A—C4A	-1.66 (18)	C1B—C2B—C3B—C4B	2.96 (17)
O1A—C3A—C4A—C5A	1.75 (19)	O1B—C3B—C4B—C5B	-3.17 (18)
C2A—C3A—C4A—C5A	-176.04 (11)	C2B—C3B—C4B—C5B	174.43 (11)
O1A—C3A—C4A—C9A	-179.21 (12)	O1B—C3B—C4B—C9B	176.20 (12)
C2A—C3A—C4A—C9A	2.99 (17)	C2B—C3B—C4B—C9B	-6.21 (17)
C9A—C4A—C5A—C6A	0.46 (19)	C9B—C4B—C5B—C6B	-0.20(19)
C3A—C4A—C5A—C6A	179.50 (12)	C3B—C4B—C5B—C6B	179.17 (12)
C4A—C5A—C6A—C7A	-0.6(2)	C4B—C5B—C6B—C7B	-0.9(2)
C5A—C6A—C7A—C8A	0.7 (2)	C5B—C6B—C7B—C8B	0.6 (2)
C6A—C7A—C8A—C9A	-0.6(2)	C6B—C7B—C8B—C9B	0.8 (2)
C7A—C8A—C9A—C4A	0.51 (19)	C7B—C8B—C9B—C4B	-1.84(19)
C7A—C8A—C9A—C10A	-179.45(12)	C7B—C8B—C9B—C10B	178.24 (12)
C5A—C4A—C9A—C8A	-0.42(18)	C5B—C4B—C9B—C8B	1.57 (18)
C3A—C4A—C9A—C8A	-179.46 (11)	C3B—C4B—C9B—C8B	-177.80(12)
C5A—C4A—C9A—C10A	179.53 (11)	C5B—C4B—C9B—C10B	-178.51 (11)
C3A—C4A—C9A—C10A	0.49 (17)	C3B—C4B—C9B—C10B	2.11 (17)
C8A—C9A—C10A—O2A	-6.40(18)	C8B—C9B—C10B—O2B	5.41 (19)
C4A - C9A - C10A - O2A	173.65 (11)	C4B-C9B-C10B-O2B	-174.51 (11)
C8A—C9A—C10A—C1A	174.88 (11)	C8B-C9B-C10B-C1B	-175.12(11)
C4A - C9A - C10A - C1A	-5.08 (17)	C4B—C9B—C10B—C1B	4.96 (17)
	••••• ( ± / )		

C2A-C1A-C10A-O2A	-172.23 (12)	C2B-C1B-C10B-O2B	171.11 (12)
N1A—C1A—C10A—O2A	6.34 (17)	N1B-C1B-C10B-O2B	-8.09 (16)
C2A-C1A-C10A-C9A	6.52 (17)	C2B-C1B-C10B-C9B	-8.38 (17)
N1A—C1A—C10A—C9A	-174.91 (10)	N1B-C1B-C10B-C9B	172.42 (10)
C18A—N1A—C11A—O3A	26.32 (17)	C18B—N1B—C11B—O3B	-29.07 (17)
C1A—N1A—C11A—O3A	-136.86 (12)	C1B—N1B—C11B—O3B	136.30 (12)
C18A—N1A—C11A—C12A	-150.51 (11)	C18B—N1B—C11B—C12B	146.89 (11)
C1A—N1A—C11A—C12A	46.31 (15)	C1B—N1B—C11B—C12B	-47.74 (15)
O3A—C11A—C12A—C13A	-148.30 (13)	O3B—C11B—C12B—C17B	-24.03 (17)
N1A—C11A—C12A—C13A	28.45 (17)	N1B—C11B—C12B—C17B	160.09 (11)
O3A—C11A—C12A—C17A	25.29 (18)	O3B-C11B-C12B-C13B	148.41 (12)
N1A—C11A—C12A—C17A	-157.96 (11)	N1B-C11B-C12B-C13B	-27.47 (16)
C17A—C12A—C13A—C14A	0.56 (19)	C17B—C12B—C13B—C14B	-1.38 (18)
C11A—C12A—C13A—C14A	174.05 (11)	C11B—C12B—C13B—C14B	-173.64 (11)
C12A—C13A—C14A—C15A	0.46 (19)	C12B—C13B—C14B—C15B	0.62 (19)
C13A—C14A—C15A—C16A	-0.7 (2)	C13B—C14B—C15B—C16B	0.0 (2)
C14A—C15A—C16A—C17A	-0.1 (2)	C14B—C15B—C16B—C17B	0.1 (2)
C15A—C16A—C17A—C12A	1.1 (2)	C15B—C16B—C17B—C12B	-0.9 (2)
C13A—C12A—C17A—C16A	-1.33 (19)	C13B—C12B—C17B—C16B	1.51 (19)
C11A—C12A—C17A—C16A	-174.98 (11)	C11B—C12B—C17B—C16B	174.05 (11)
C1A—N1A—C18A—O4A	16.11 (17)	C11B—N1B—C18B—O4B	149.16 (12)
C11A—N1A—C18A—O4A	-147.96 (12)	C1B—N1B—C18B—O4B	-17.25 (17)
C1A—N1A—C18A—C19A	-161.39 (11)	C11B—N1B—C18B—C19B	-35.53 (16)
C11A—N1A—C18A—C19A	34.55 (16)	C1B—N1B—C18B—C19B	158.06 (11)
O4A—C18A—C19A—C24A	34.2 (2)	O4B-C18B-C19B-C24B	-42.44 (19)
N1A—C18A—C19A—C24A	-148.34 (13)	N1B-C18B-C19B-C24B	142.42 (12)
O4A—C18A—C19A—C20A	-143.71 (14)	O4B-C18B-C19B-C20B	132.94 (14)
N1A—C18A—C19A—C20A	33.72 (18)	N1B-C18B-C19B-C20B	-42.21 (17)
C24A—C19A—C20A—C21A	1.6 (2)	C24B—C19B—C20B—C21B	-0.3 (2)
C18A—C19A—C20A—C21A	179.56 (14)	C18B—C19B—C20B—C21B	-175.65 (13)
C19A—C20A—C21A—C22A	0.7 (3)	C19B—C20B—C21B—C22B	-0.9 (2)
C20A—C21A—C22A—C23A	-2.0 (3)	C20B—C21B—C22B—C23B	0.6 (2)
C21A—C22A—C23A—C24A	1.0 (3)	C21B—C22B—C23B—C24B	0.9 (2)
C20A—C19A—C24A—C23A	-2.6 (2)	C20B—C19B—C24B—C23B	1.8 (2)
C18A—C19A—C24A—C23A	179.40 (14)	C18B—C19B—C24B—C23B	177.18 (13)
C22A—C23A—C24A—C19A	1.3 (3)	C22B—C23B—C24B—C19B	-2.0 (2)

## Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	D—H···A
$C21A$ — $H21A$ ···O2 $B^{i}$	0.95	2.41	3.1270 (19)	132
C14 <i>B</i> —H14 <i>B</i> ···O3 <i>A</i> <sup>ii</sup>	0.95	2.50	3.4345 (17)	168
C15 <i>B</i> —H15 <i>B</i> ····O4 <i>A</i> <sup>iii</sup>	0.95	2.62	3.4638 (17)	148
C22 <i>B</i> —H22 <i>B</i> ····O2 <i>A</i>	0.95	2.54	3.2054 (17)	127
C14 <i>A</i> —H14 <i>A</i> ···O3 <i>B</i>	0.95	2.44	3.2303 (16)	140

Symmetry codes: (i) *x*, *y*+1, *z*; (ii) *x*, *y*-1, *z*; (iii) –*x*+1, –*y*+1, –*z*+2.