

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## 4-Chloro-2-[(2,6-diisopropylphenyl)-iminomethyl]phenol

 P. Balamurugan,<sup>a</sup> K. Kanmani Raja,<sup>b</sup> D. Easwaramoorthy,<sup>c</sup> G. Chakkaravarthi<sup>d\*</sup> and G. Rajagopal<sup>e\*</sup>

<sup>a</sup>Department of Chemistry, Government Arts College (Men), Nandanam, Chennai 600 035, India, <sup>b</sup>Department of Chemistry, Government Thirumagal Mills College, Gudiyattam 632 604, India, <sup>c</sup>Department of Chemistry, B.S. Abdur Rahman University, Vandalur, Chennai 600 049, India, <sup>d</sup>Department of Physics, CPCL Polytechnic College, Chennai 600 068, India, and <sup>e</sup>Department of Chemistry, Government Arts College, Melur 625 106, India

Correspondence e-mail: chakkaravarthi\_2005@yahoo.com, rajagopal18@yahoo.com

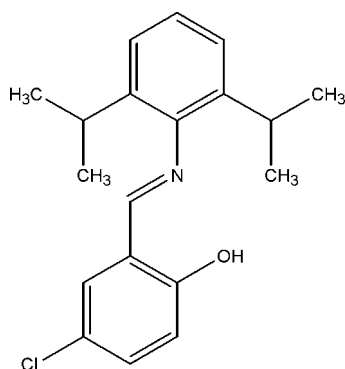
Received 5 May 2012; accepted 7 May 2012

Key indicators: single-crystal X-ray study;  $T = 295$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.049;  $wR$  factor = 0.153; data-to-parameter ratio = 21.9.

The asymmetric unit of the title compound,  $\text{C}_{19}\text{H}_{22}\text{ClNO}$ , contains two independent molecules in which the dihedral angles between the aromatic rings are  $76.45$  (9) and  $74.69$  (9)°. An intramolecular  $\text{O}-\text{H}\cdots\text{N}$  hydrogen bond occurs in each molecule. The crystal structure features weak  $\text{C}-\text{H}\cdots\pi$  interactions.

### Related literature

For the biological activity of Schiff base ligands, see: Santos *et al.* (2001). For related structures, see: Raja *et al.* (2008); Lin *et al.* (2005).



### Experimental

#### Crystal data

 $\text{C}_{19}\text{H}_{22}\text{ClNO}$ 
 $M_r = 315.83$ 

Triclinic,  $P\bar{1}$   
 $a = 11.276$  (2) Å  
 $b = 11.776$  (2) Å  
 $c = 14.189$  (3) Å  
 $\alpha = 73.01$  (3)°  
 $\beta = 88.42$  (2)°  
 $\gamma = 85.57$  (3)°

$V = 1796.5$  (6) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.21$  mm<sup>-1</sup>  
 $T = 295$  K  
 $0.30 \times 0.20 \times 0.16$  mm

#### Data collection

Bruker Kappa APEXII diffractometer  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.939$ ,  $T_{\max} = 0.967$

40436 measured reflections  
 8907 independent reflections  
 5518 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.049$   
 $wR(F^2) = 0.153$   
 $S = 1.02$   
 8907 reflections

407 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.30$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.36$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$\text{Cg1}$  and  $\text{Cg3}$  are the centroids of the  $\text{C1}-\text{C6}$  and  $\text{C20}-\text{C25}$  rings, respectively.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1}\cdots\text{N1}$	0.82	1.92	2.646 (2)	147
$\text{O2}-\text{H2}\cdots\text{N2}$	0.82	1.90	2.630 (2)	147
$\text{C31}-\text{H31c}\cdots\text{Cg1}$	0.96	2.90	3.743 (3)	147
$\text{C12}-\text{H12A}\cdots\text{Cg3}^i$	0.96	2.98	3.833 (3)	149

Symmetry code: (i)  $x - 1, y, z$ .

Data collection: APEX2 (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009); software used to prepare material for publication: SHELXL97.

The authors wish to acknowledge the SAIF, IIT Madras, for the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: VM2174).

### References

- Bruker (2004). APEX2. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Lin, J., Cui, G.-H., Li, J.-R. & Xu, S.-S. (2005). *Acta Cryst.* E61, o627–o628.  
 Raja, K. K., Bilal, I. M., Thambidurai, S., Rajagopal, G. & SubbiahPandi, A. (2008). *Acta Cryst.* E64, o2265.  
 Santos, M. L. P., Bagatin, I. A., Pereira, E. M. & Ferreira, A. M. D. C. (2001). *J. Chem. Soc. Dalton Trans.* pp. 838–844.  
 Sheldrick, G. M. (1996). SADABS, University of Göttingen, Germany.  
 Sheldrick, G. M. (2008). *Acta Cryst.* A64, 112–122.  
 Spek, A. L. (2009). *Acta Cryst.* D65, 148–155.

## supporting information

*Acta Cryst.* (2012). E68, o1721 [doi:10.1107/S1600536812020612]

## 4-Chloro-2-[(2,6-diisopropylphenyl)iminomethyl]phenol

P. Balamurugan, K. Kanmani Raja, D. Easwaramoorthy, G. Chakkaravarthi and G. Rajagopal

### S1. Comment

Schiff base derivatives are found to exhibit important pharmacological properties, such as antibacterial, antitumor and antitoxic activities (Santos *et al.*, 2001).

The asymmetric unit of the title compound (I) (Fig. 1) contains two molecules. The geometric parameters in (I) are comparable with the similar reported structures (Raja *et al.*, 2008; Lin *et al.*, 2005). The dihedral angles between the aromatic rings (C1-C6) and (C14-C19) & (C21-C26) and (C33-C38) are 76.45 (9) and 74.69 (9)°, respectively. In the asymmetric unit of (I), both molecules adopt an anti-periplanar conformation [C14-C13-N1-C1 = 176.31 (15)° and C33-C32-N2-C20 = -176.22 (15)°] about the C=N bond.

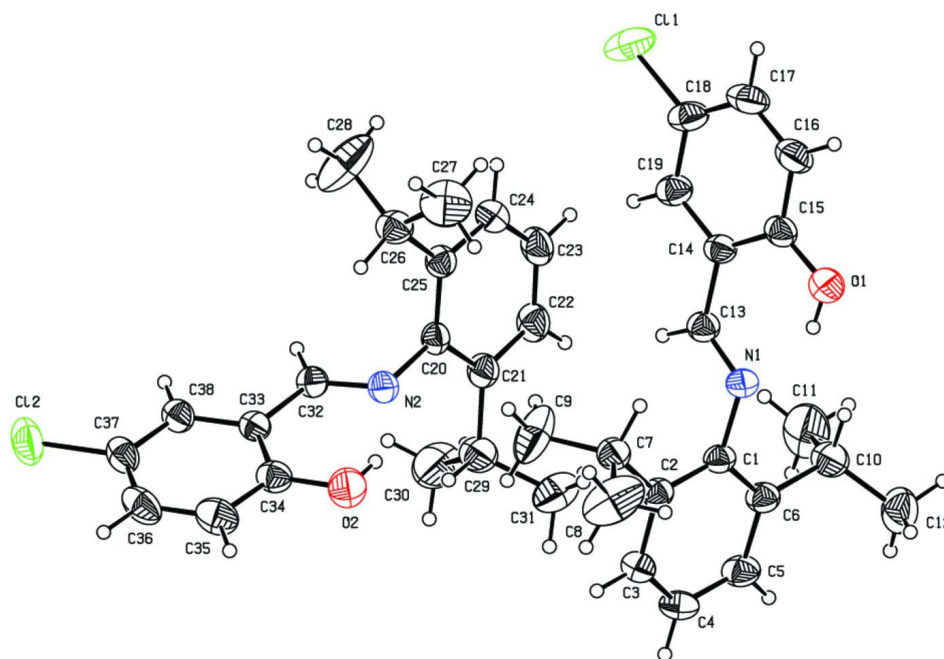
The molecular structure is stabilized by weak intramolecular O-H...N hydrogen bonds and C-H... $\pi$  interactions (Table 1). The crystal structure is stabilized by a weak intermolecular C-H... $\pi$  interaction (Table 1).

### S2. Experimental

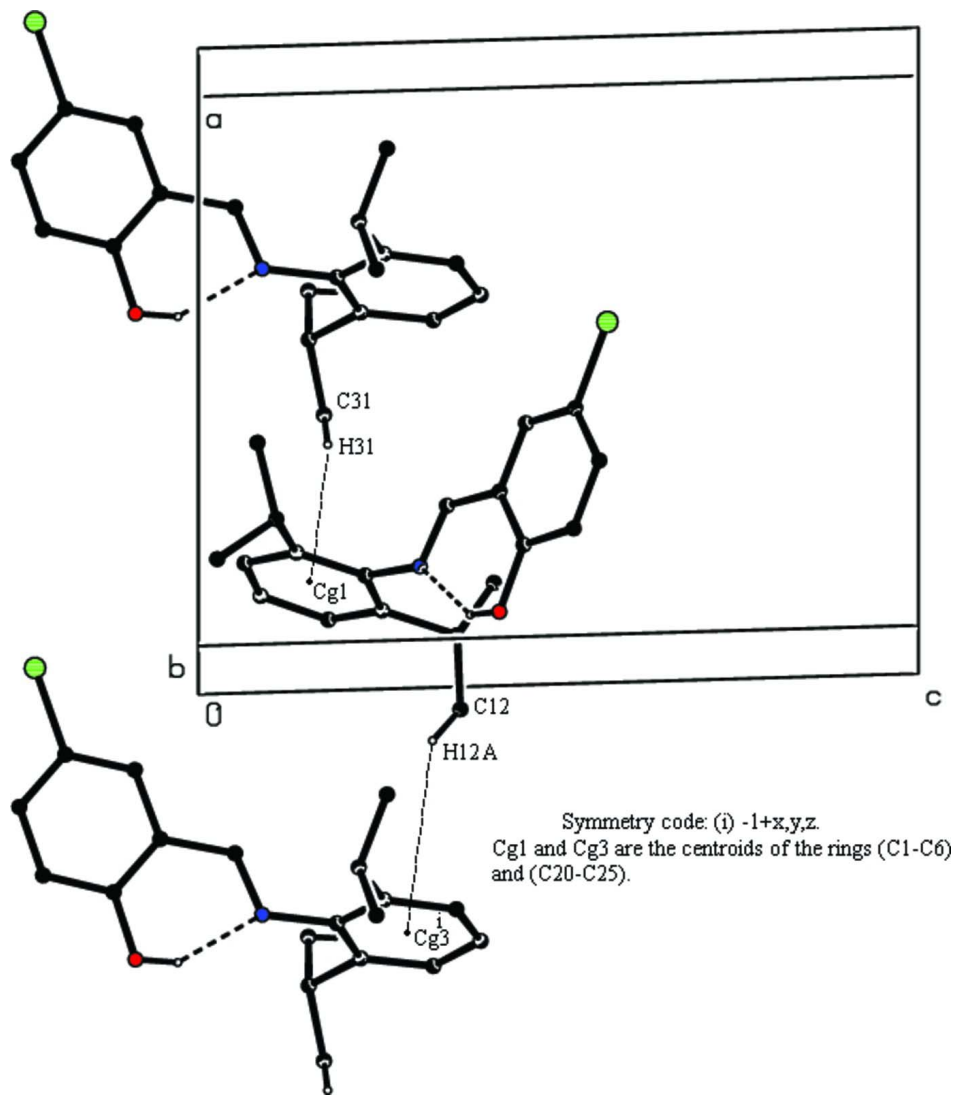
An ethanolic solution (10 ml) of 2,6-diisopropylaniline [2 mmol] was magnetically stirred in a round bottom flask followed by drop wise addition of ethanolic solution (10 ml) of 5-chlorosalicylaldehyde [2 mmol]. The reaction mixture was then refluxed for 3 h and upon cooling to 273 K, a yellow solid precipitated from the reaction mixture. The solid which separated out was filtered, washed with ice cold ethanol and dried over anhydrous CaCl<sub>2</sub>. Single crystals of good diffraction quality were obtained by recrystallization of the compound from an ethanol solution by the slow evaporation method. Yield: 70 %.

### S3. Refinement

H atoms were positioned geometrically with C—H = 0.93–0.97 Å and O—H = 0.82 Å and allowed to ride on their parent atoms, with  $U_{iso}(H) = 1.5 U_{eq}(O)$  or  $1.2 U_{eq}(C)$  or  $1.5 U_{eq}(C_{methyl})$ .

**Figure 1**

The molecular structure of (I), with atom labels and 30% probability displacement ellipsoids for non-H atoms.

**Figure 2**

The packing of (I), viewed down *b* axis. Intramolecular hydrogen bond and C-H $\cdots$  $\pi$  interactions are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted. [Symmetry code (i)  $-1 + x, y, z$ . Cg1 and Cg3 are the centroids of the rings (C1-C6) and (C20-C25)].

#### 4-Chloro-2-[(2,6-diisopropylphenyl)iminomethyl]phenol

##### Crystal data

$C_{19}H_{22}ClNO$

$M_r = 315.83$

Triclinic,  $P\bar{1}$

Hall symbol:  $-P\ 1$

$a = 11.276$  (2) Å

$b = 11.776$  (2) Å

$c = 14.189$  (3) Å

$\alpha = 73.01$  (3)°

$\beta = 88.42$  (2)°

$\gamma = 85.57$  (3)°

$V = 1796.5$  (6) Å<sup>3</sup>

$Z = 4$

$F(000) = 672$

$D_x = 1.168$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 8910 reflections

$\theta = 1.5$ – $28.4$ °

$\mu = 0.21$  mm<sup>-1</sup>

$T = 295$  K

Prism, light yellow

$0.30 \times 0.20 \times 0.16$  mm

*Data collection*

Bruker Kappa APEXII  
diffractometer  
Radiation source: fine-focus sealed tube  
Graphite monochromator  
 $\omega$  and  $\varphi$  scans  
Absorption correction: multi-scan  
(*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.939$ ,  $T_{\max} = 0.967$

40436 measured reflections  
8907 independent reflections  
5518 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$   
 $\theta_{\max} = 28.4^\circ$ ,  $\theta_{\min} = 2.0^\circ$   
 $h = -15 \rightarrow 15$   
 $k = -15 \rightarrow 15$   
 $l = -18 \rightarrow 18$

*Refinement*

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.049$   
 $wR(F^2) = 0.153$   
 $S = 1.02$   
8907 reflections  
407 parameters  
0 restraints  
Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0607P)^2 + 0.4833P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.30 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.36 \text{ e } \text{\AA}^{-3}$

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.13739 (14)	0.66107 (15)	0.23173 (12)	0.0498 (4)
C2	0.17448 (14)	0.71891 (15)	0.13594 (13)	0.0509 (4)
C3	0.16389 (16)	0.66098 (16)	0.06416 (14)	0.0584 (4)
H3	0.1890	0.6969	0.0001	0.070*
C4	0.11715 (18)	0.55202 (17)	0.08624 (15)	0.0652 (5)
H4	0.1117	0.5143	0.0375	0.078*
C5	0.07837 (18)	0.49844 (17)	0.18045 (15)	0.0658 (5)
H5	0.0457	0.4251	0.1944	0.079*
C6	0.08694 (16)	0.55152 (16)	0.25501 (13)	0.0572 (4)
C7	0.22149 (17)	0.84165 (16)	0.10807 (14)	0.0602 (5)
H7	0.2122	0.8713	0.1658	0.072*
C8	0.1505 (3)	0.9267 (2)	0.0261 (3)	0.1324 (13)
H8A	0.1773	1.0052	0.0141	0.199*
H8B	0.0679	0.9276	0.0444	0.199*
H8C	0.1608	0.9020	-0.0325	0.199*
C9	0.3505 (2)	0.8389 (3)	0.0819 (3)	0.1291 (12)
H9A	0.3952	0.7883	0.1370	0.194*
H9B	0.3767	0.9180	0.0657	0.194*
H9C	0.3627	0.8085	0.0263	0.194*
C10	0.0474 (2)	0.48956 (19)	0.35955 (15)	0.0753 (6)
H10	0.0405	0.5493	0.3955	0.090*
C11	0.1404 (3)	0.3948 (3)	0.4105 (2)	0.1328 (12)
H11A	0.1499	0.3350	0.3766	0.199*
H11B	0.1161	0.3591	0.4773	0.199*
H11C	0.2146	0.4297	0.4101	0.199*

---

C12	-0.0733 (2)	0.4395 (2)	0.3639 (2)	0.0987 (8)
H12A	-0.1291	0.4996	0.3252	0.148*
H12B	-0.1005	0.4148	0.4310	0.148*
H12C	-0.0668	0.3723	0.3382	0.148*
C13	0.24570 (16)	0.71952 (15)	0.34528 (13)	0.0543 (4)
H13	0.3115	0.6794	0.3255	0.065*
C14	0.26162 (16)	0.78161 (15)	0.41832 (12)	0.0517 (4)
C15	0.16653 (17)	0.84406 (16)	0.45179 (13)	0.0577 (4)
C16	0.1866 (2)	0.9024 (2)	0.52119 (15)	0.0752 (6)
H16	0.1238	0.9438	0.5436	0.090*
C17	0.2994 (2)	0.8991 (2)	0.55676 (15)	0.0798 (7)
H17	0.3126	0.9380	0.6035	0.096*
C18	0.3930 (2)	0.83857 (18)	0.52351 (14)	0.0691 (6)
C19	0.37495 (18)	0.78031 (16)	0.45505 (14)	0.0613 (5)
H19	0.4388	0.7397	0.4331	0.074*
C38	0.89501 (17)	0.76937 (15)	-0.08837 (13)	0.0574 (4)
H38	0.9565	0.7309	-0.0459	0.069*
C20	0.63784 (14)	0.65812 (15)	0.19237 (12)	0.0500 (4)
C21	0.58671 (16)	0.54870 (16)	0.22519 (14)	0.0590 (4)
C22	0.57451 (18)	0.49846 (18)	0.32620 (15)	0.0683 (5)
H22	0.5417	0.4253	0.3498	0.082*
C23	0.60990 (18)	0.55449 (19)	0.39185 (15)	0.0693 (5)
H23	0.6033	0.5181	0.4592	0.083*
C24	0.65504 (17)	0.66414 (18)	0.35815 (14)	0.0637 (5)
H24	0.6762	0.7026	0.4032	0.076*
C25	0.66973 (15)	0.71898 (16)	0.25773 (13)	0.0541 (4)
C26	0.71529 (19)	0.84265 (17)	0.22278 (15)	0.0676 (5)
H26	0.7221	0.8639	0.1508	0.081*
C27	0.6271 (3)	0.9335 (2)	0.2472 (2)	0.1123 (9)
H27A	0.6172	0.9144	0.3174	0.169*
H27B	0.5520	0.9325	0.2173	0.169*
H27C	0.6562	1.0112	0.2224	0.169*
C28	0.8359 (3)	0.8457 (3)	0.2619 (3)	0.1506 (16)
H28A	0.8322	0.8263	0.3324	0.226*
H28B	0.8633	0.9239	0.2352	0.226*
H28C	0.8899	0.7887	0.2432	0.226*
C29	0.5491 (2)	0.4852 (2)	0.15336 (17)	0.0784 (6)
H29	0.5484	0.5422	0.0873	0.094*
C30	0.6386 (3)	0.3834 (3)	0.1526 (3)	0.1459 (15)
H30A	0.7165	0.4123	0.1399	0.219*
H30B	0.6184	0.3493	0.1020	0.219*
H30C	0.6378	0.3239	0.2154	0.219*
C31	0.4257 (2)	0.4417 (2)	0.1744 (2)	0.0997 (8)
H31A	0.4262	0.3787	0.2353	0.150*
H31B	0.4019	0.4125	0.1219	0.150*
H31C	0.3708	0.5061	0.1795	0.150*
C32	0.75544 (15)	0.71162 (14)	0.05083 (12)	0.0505 (4)
H32	0.8191	0.6740	0.0912	0.061*

C33	0.77830 (15)	0.76946 (14)	-0.05304 (12)	0.0496 (4)
C34	0.68686 (18)	0.82811 (16)	-0.11783 (13)	0.0607 (5)
C35	0.7139 (2)	0.8834 (2)	-0.21581 (16)	0.0809 (6)
H35	0.6533	0.9218	-0.2594	0.097*
C36	0.8297 (2)	0.88172 (19)	-0.24875 (15)	0.0791 (6)
H36	0.8472	0.9190	-0.3145	0.095*
C37	0.91965 (19)	0.82525 (17)	-0.18488 (15)	0.0647 (5)
Cl1	0.53583 (7)	0.83560 (6)	0.56803 (5)	0.1042 (3)
Cl2	1.06620 (6)	0.82311 (6)	-0.22699 (5)	0.0990 (2)
N1	0.14538 (13)	0.71810 (13)	0.30767 (11)	0.0546 (3)
N2	0.65162 (12)	0.71064 (12)	0.08848 (10)	0.0528 (3)
O1	0.05522 (12)	0.85034 (14)	0.41805 (11)	0.0753 (4)
H1	0.0543	0.8131	0.3775	0.113*
O2	0.57289 (13)	0.83301 (16)	-0.08738 (11)	0.0832 (4)
H2	0.5687	0.7966	-0.0287	0.125*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0504 (9)	0.0495 (9)	0.0544 (10)	-0.0034 (7)	-0.0043 (7)	-0.0226 (8)
C2	0.0494 (9)	0.0491 (9)	0.0567 (10)	-0.0026 (7)	-0.0023 (7)	-0.0196 (8)
C3	0.0671 (11)	0.0584 (11)	0.0526 (10)	-0.0031 (8)	-0.0008 (8)	-0.0210 (9)
C4	0.0814 (13)	0.0597 (11)	0.0642 (12)	-0.0053 (9)	-0.0102 (10)	-0.0325 (10)
C5	0.0825 (13)	0.0528 (10)	0.0684 (13)	-0.0161 (9)	-0.0052 (10)	-0.0241 (10)
C6	0.0649 (10)	0.0529 (10)	0.0575 (11)	-0.0108 (8)	-0.0024 (8)	-0.0199 (8)
C7	0.0686 (11)	0.0555 (10)	0.0603 (11)	-0.0140 (8)	0.0046 (9)	-0.0209 (9)
C8	0.147 (3)	0.0542 (14)	0.179 (3)	-0.0117 (15)	-0.063 (2)	-0.0002 (17)
C9	0.0758 (17)	0.096 (2)	0.216 (4)	-0.0333 (14)	0.026 (2)	-0.042 (2)
C10	0.0984 (16)	0.0677 (13)	0.0634 (12)	-0.0279 (11)	0.0082 (11)	-0.0199 (10)
C11	0.120 (2)	0.154 (3)	0.085 (2)	-0.005 (2)	-0.0113 (17)	0.0267 (19)
C12	0.0974 (18)	0.0964 (18)	0.1012 (19)	-0.0333 (14)	0.0262 (15)	-0.0225 (15)
C13	0.0599 (10)	0.0517 (10)	0.0543 (10)	-0.0074 (8)	-0.0005 (8)	-0.0192 (8)
C14	0.0655 (10)	0.0471 (9)	0.0436 (9)	-0.0151 (8)	-0.0012 (7)	-0.0117 (7)
C15	0.0745 (12)	0.0549 (10)	0.0449 (9)	-0.0179 (9)	0.0066 (8)	-0.0135 (8)
C16	0.0992 (16)	0.0766 (14)	0.0590 (12)	-0.0182 (12)	0.0127 (11)	-0.0321 (11)
C17	0.123 (2)	0.0747 (14)	0.0514 (12)	-0.0334 (14)	-0.0014 (12)	-0.0264 (11)
C18	0.0927 (15)	0.0597 (11)	0.0539 (11)	-0.0296 (11)	-0.0161 (10)	-0.0074 (9)
C19	0.0732 (12)	0.0546 (10)	0.0566 (11)	-0.0150 (9)	-0.0072 (9)	-0.0137 (9)
C38	0.0677 (11)	0.0507 (10)	0.0536 (10)	-0.0117 (8)	0.0064 (8)	-0.0135 (8)
C20	0.0473 (8)	0.0520 (9)	0.0467 (9)	-0.0048 (7)	0.0039 (7)	-0.0083 (7)
C21	0.0625 (11)	0.0558 (10)	0.0569 (11)	-0.0118 (8)	0.0061 (8)	-0.0122 (8)
C22	0.0752 (13)	0.0607 (12)	0.0636 (12)	-0.0174 (9)	0.0121 (10)	-0.0075 (10)
C23	0.0752 (13)	0.0746 (13)	0.0493 (11)	-0.0099 (10)	0.0096 (9)	-0.0039 (10)
C24	0.0684 (11)	0.0714 (12)	0.0529 (11)	-0.0073 (9)	0.0014 (8)	-0.0202 (9)
C25	0.0543 (9)	0.0543 (10)	0.0519 (10)	-0.0061 (7)	0.0014 (7)	-0.0125 (8)
C26	0.0854 (14)	0.0578 (11)	0.0624 (12)	-0.0177 (10)	0.0006 (10)	-0.0189 (9)
C27	0.140 (3)	0.0686 (16)	0.131 (2)	-0.0010 (16)	-0.008 (2)	-0.0328 (16)
C28	0.100 (2)	0.100 (2)	0.230 (4)	-0.0460 (17)	-0.046 (2)	0.000 (2)

C29	0.1006 (16)	0.0690 (13)	0.0709 (13)	-0.0313 (12)	0.0087 (12)	-0.0230 (11)
C30	0.098 (2)	0.178 (3)	0.218 (4)	-0.007 (2)	0.016 (2)	-0.148 (3)
C31	0.0865 (17)	0.0907 (17)	0.130 (2)	-0.0159 (13)	-0.0193 (15)	-0.0411 (16)
C32	0.0564 (9)	0.0456 (9)	0.0465 (9)	-0.0060 (7)	-0.0013 (7)	-0.0081 (7)
C33	0.0615 (10)	0.0412 (8)	0.0466 (9)	-0.0098 (7)	0.0019 (7)	-0.0122 (7)
C34	0.0724 (12)	0.0570 (11)	0.0516 (10)	-0.0075 (9)	-0.0044 (9)	-0.0130 (8)
C35	0.1028 (17)	0.0781 (15)	0.0524 (12)	0.0004 (12)	-0.0126 (11)	-0.0053 (10)
C36	0.122 (2)	0.0659 (13)	0.0458 (11)	-0.0168 (13)	0.0152 (12)	-0.0093 (10)
C37	0.0876 (13)	0.0505 (10)	0.0587 (11)	-0.0206 (9)	0.0217 (10)	-0.0183 (9)
C11	0.1159 (5)	0.1021 (5)	0.0971 (5)	-0.0390 (4)	-0.0434 (4)	-0.0211 (4)
C12	0.1058 (5)	0.1024 (5)	0.0922 (4)	-0.0357 (4)	0.0498 (4)	-0.0310 (4)
N1	0.0585 (8)	0.0561 (8)	0.0556 (9)	-0.0110 (6)	-0.0003 (6)	-0.0244 (7)
N2	0.0563 (8)	0.0521 (8)	0.0480 (8)	-0.0085 (6)	0.0024 (6)	-0.0107 (6)
O1	0.0675 (8)	0.0914 (11)	0.0773 (10)	-0.0064 (7)	0.0050 (7)	-0.0413 (8)
O2	0.0667 (9)	0.1046 (12)	0.0678 (9)	0.0040 (8)	-0.0107 (7)	-0.0104 (9)

*Geometric parameters (Å, °)*

C1—C6	1.397 (2)	C38—C33	1.395 (2)
C1—C2	1.399 (2)	C38—H38	0.9300
C1—N1	1.435 (2)	C20—C25	1.396 (2)
C2—C3	1.394 (2)	C20—C21	1.400 (2)
C2—C7	1.516 (2)	C20—N2	1.431 (2)
C3—C4	1.372 (3)	C21—C22	1.389 (3)
C3—H3	0.9300	C21—C29	1.515 (3)
C4—C5	1.376 (3)	C22—C23	1.372 (3)
C4—H4	0.9300	C22—H22	0.9300
C5—C6	1.386 (3)	C23—C24	1.372 (3)
C5—H5	0.9300	C23—H23	0.9300
C6—C10	1.521 (3)	C24—C25	1.393 (3)
C7—C9	1.492 (3)	C24—H24	0.9300
C7—C8	1.498 (3)	C25—C26	1.520 (3)
C7—H7	0.9800	C26—C28	1.490 (3)
C8—H8A	0.9600	C26—C27	1.512 (3)
C8—H8B	0.9600	C26—H26	0.9800
C8—H8C	0.9600	C27—H27A	0.9600
C9—H9A	0.9600	C27—H27B	0.9600
C9—H9B	0.9600	C27—H27C	0.9600
C9—H9C	0.9600	C28—H28A	0.9600
C10—C11	1.502 (4)	C28—H28B	0.9600
C10—C12	1.517 (3)	C28—H28C	0.9600
C10—H10	0.9800	C29—C30	1.508 (4)
C11—H11A	0.9600	C29—C31	1.511 (3)
C11—H11B	0.9600	C29—H29	0.9800
C11—H11C	0.9600	C30—H30A	0.9600
C12—H12A	0.9600	C30—H30B	0.9600
C12—H12B	0.9600	C30—H30C	0.9600
C12—H12C	0.9600	C31—H31A	0.9600



C13—N1	1.267 (2)	C31—H31B	0.9600
C13—C14	1.454 (2)	C31—H31C	0.9600
C13—H13	0.9300	C32—N2	1.273 (2)
C14—C19	1.391 (2)	C32—C33	1.457 (2)
C14—C15	1.404 (3)	C32—H32	0.9300
C15—O1	1.345 (2)	C33—C34	1.399 (3)
C15—C16	1.388 (3)	C34—O2	1.347 (2)
C16—C17	1.375 (3)	C34—C35	1.388 (3)
C16—H16	0.9300	C35—C36	1.376 (3)
C17—C18	1.377 (3)	C35—H35	0.9300
C17—H17	0.9300	C36—C37	1.373 (3)
C18—C19	1.370 (3)	C36—H36	0.9300
C18—C11	1.740 (2)	C37—C12	1.743 (2)
C19—H19	0.9300	O1—H1	0.8200
C38—C37	1.366 (3)	O2—H2	0.8200
C6—C1—C2	121.94 (15)	C33—C38—H38	119.7
C6—C1—N1	118.55 (15)	C25—C20—C21	121.75 (16)
C2—C1—N1	119.43 (14)	C25—C20—N2	119.82 (14)
C3—C2—C1	117.57 (15)	C21—C20—N2	118.35 (16)
C3—C2—C7	119.87 (16)	C22—C21—C20	117.72 (17)
C1—C2—C7	122.54 (15)	C22—C21—C29	120.96 (17)
C4—C3—C2	121.24 (17)	C20—C21—C29	121.30 (17)
C4—C3—H3	119.4	C23—C22—C21	121.37 (18)
C2—C3—H3	119.4	C23—C22—H22	119.3
C3—C4—C5	120.05 (17)	C21—C22—H22	119.3
C3—C4—H4	120.0	C22—C23—C24	120.05 (18)
C5—C4—H4	120.0	C22—C23—H23	120.0
C4—C5—C6	121.33 (17)	C24—C23—H23	120.0
C4—C5—H5	119.3	C23—C24—C25	121.25 (18)
C6—C5—H5	119.3	C23—C24—H24	119.4
C5—C6—C1	117.82 (17)	C25—C24—H24	119.4
C5—C6—C10	120.80 (16)	C24—C25—C20	117.73 (16)
C1—C6—C10	121.33 (16)	C24—C25—C26	119.86 (17)
C9—C7—C8	110.6 (2)	C20—C25—C26	122.38 (16)
C9—C7—C2	112.29 (18)	C28—C26—C27	111.8 (2)
C8—C7—C2	111.16 (17)	C28—C26—C25	112.04 (19)
C9—C7—H7	107.5	C27—C26—C25	110.81 (19)
C8—C7—H7	107.5	C28—C26—H26	107.3
C2—C7—H7	107.5	C27—C26—H26	107.3
C7—C8—H8A	109.5	C25—C26—H26	107.3
C7—C8—H8B	109.5	C26—C27—H27A	109.5
H8A—C8—H8B	109.5	C26—C27—H27B	109.5
C7—C8—H8C	109.5	H27A—C27—H27B	109.5
H8A—C8—H8C	109.5	C26—C27—H27C	109.5
H8B—C8—H8C	109.5	H27A—C27—H27C	109.5
C7—C9—H9A	109.5	H27B—C27—H27C	109.5
C7—C9—H9B	109.5	C26—C28—H28A	109.5

H9A—C9—H9B	109.5	C26—C28—H28B	109.5
C7—C9—H9C	109.5	H28A—C28—H28B	109.5
H9A—C9—H9C	109.5	C26—C28—H28C	109.5
H9B—C9—H9C	109.5	H28A—C28—H28C	109.5
C11—C10—C12	111.2 (2)	H28B—C28—H28C	109.5
C11—C10—C6	110.4 (2)	C30—C29—C31	110.5 (2)
C12—C10—C6	112.97 (19)	C30—C29—C21	110.2 (2)
C11—C10—H10	107.3	C31—C29—C21	112.84 (19)
C12—C10—H10	107.3	C30—C29—H29	107.7
C6—C10—H10	107.3	C31—C29—H29	107.7
C10—C11—H11A	109.5	C21—C29—H29	107.7
C10—C11—H11B	109.5	C29—C30—H30A	109.5
H11A—C11—H11B	109.5	C29—C30—H30B	109.5
C10—C11—H11C	109.5	H30A—C30—H30B	109.5
H11A—C11—H11C	109.5	C29—C30—H30C	109.5
H11B—C11—H11C	109.5	H30A—C30—H30C	109.5
C10—C12—H12A	109.5	H30B—C30—H30C	109.5
C10—C12—H12B	109.5	C29—C31—H31A	109.5
H12A—C12—H12B	109.5	C29—C31—H31B	109.5
C10—C12—H12C	109.5	H31A—C31—H31B	109.5
H12A—C12—H12C	109.5	C29—C31—H31C	109.5
H12B—C12—H12C	109.5	H31A—C31—H31C	109.5
N1—C13—C14	122.37 (17)	H31B—C31—H31C	109.5
N1—C13—H13	118.8	N2—C32—C33	122.50 (16)
C14—C13—H13	118.8	N2—C32—H32	118.7
C19—C14—C15	119.09 (16)	C33—C32—H32	118.7
C19—C14—C13	118.86 (17)	C38—C33—C34	119.01 (16)
C15—C14—C13	122.04 (16)	C38—C33—C32	119.15 (16)
O1—C15—C16	118.21 (19)	C34—C33—C32	121.81 (16)
O1—C15—C14	122.01 (16)	O2—C34—C35	118.98 (19)
C16—C15—C14	119.79 (19)	O2—C34—C33	121.58 (16)
C17—C16—C15	119.9 (2)	C35—C34—C33	119.44 (19)
C17—C16—H16	120.0	C36—C35—C34	120.3 (2)
C15—C16—H16	120.0	C36—C35—H35	119.8
C16—C17—C18	120.36 (19)	C34—C35—H35	119.8
C16—C17—H17	119.8	C37—C36—C35	120.21 (19)
C18—C17—H17	119.8	C37—C36—H36	119.9
C19—C18—C17	120.6 (2)	C35—C36—H36	119.9
C19—C18—C11	119.25 (19)	C38—C37—C36	120.4 (2)
C17—C18—C11	120.18 (16)	C38—C37—C12	119.51 (18)
C18—C19—C14	120.3 (2)	C36—C37—C12	120.04 (16)
C18—C19—H19	119.9	C13—N1—C1	119.38 (15)
C14—C19—H19	119.9	C32—N2—C20	118.93 (15)
C37—C38—C33	120.55 (19)	C15—O1—H1	109.5
C37—C38—H38	119.7	C34—O2—H2	109.5
C6—C1—C2—C3	-2.6 (2)	N2—C20—C21—C29	1.4 (3)
N1—C1—C2—C3	-179.26 (15)	C20—C21—C22—C23	0.8 (3)

C6—C1—C2—C7	175.68 (16)	C29—C21—C22—C23	179.1 (2)
N1—C1—C2—C7	-1.0 (2)	C21—C22—C23—C24	2.1 (3)
C1—C2—C3—C4	1.0 (3)	C22—C23—C24—C25	-2.2 (3)
C7—C2—C3—C4	-177.32 (17)	C23—C24—C25—C20	-0.5 (3)
C2—C3—C4—C5	0.8 (3)	C23—C24—C25—C26	177.53 (18)
C3—C4—C5—C6	-1.0 (3)	C21—C20—C25—C24	3.4 (3)
C4—C5—C6—C1	-0.6 (3)	N2—C20—C25—C24	-179.81 (16)
C4—C5—C6—C10	-178.0 (2)	C21—C20—C25—C26	-174.56 (17)
C2—C1—C6—C5	2.4 (3)	N2—C20—C25—C26	2.2 (3)
N1—C1—C6—C5	179.10 (16)	C24—C25—C26—C28	60.4 (3)
C2—C1—C6—C10	179.81 (18)	C20—C25—C26—C28	-121.6 (3)
N1—C1—C6—C10	-3.5 (3)	C24—C25—C26—C27	-65.2 (3)
C3—C2—C7—C9	-69.4 (3)	C20—C25—C26—C27	112.7 (2)
C1—C2—C7—C9	112.4 (2)	C22—C21—C29—C30	-75.1 (3)
C3—C2—C7—C8	55.2 (3)	C20—C21—C29—C30	103.2 (3)
C1—C2—C7—C8	-123.0 (2)	C22—C21—C29—C31	49.0 (3)
C5—C6—C10—C11	78.3 (3)	C20—C21—C29—C31	-132.8 (2)
C1—C6—C10—C11	-99.1 (3)	C37—C38—C33—C34	-0.3 (3)
C5—C6—C10—C12	-46.9 (3)	C37—C38—C33—C32	-178.46 (16)
C1—C6—C10—C12	135.7 (2)	N2—C32—C33—C38	178.38 (16)
N1—C13—C14—C19	-179.12 (17)	N2—C32—C33—C34	0.3 (3)
N1—C13—C14—C15	0.0 (3)	C38—C33—C34—O2	-178.73 (17)
C19—C14—C15—O1	178.93 (16)	C32—C33—C34—O2	-0.6 (3)
C13—C14—C15—O1	-0.2 (3)	C38—C33—C34—C35	0.9 (3)
C19—C14—C15—C16	-0.5 (3)	C32—C33—C34—C35	178.98 (17)
C13—C14—C15—C16	-179.62 (17)	O2—C34—C35—C36	178.9 (2)
O1—C15—C16—C17	-179.35 (19)	C33—C34—C35—C36	-0.7 (3)
C14—C15—C16—C17	0.1 (3)	C34—C35—C36—C37	0.0 (3)
C15—C16—C17—C18	0.3 (3)	C33—C38—C37—C36	-0.4 (3)
C16—C17—C18—C19	-0.4 (3)	C33—C38—C37—C12	-179.74 (13)
C16—C17—C18—C11	179.50 (16)	C35—C36—C37—C38	0.6 (3)
C17—C18—C19—C14	0.0 (3)	C35—C36—C37—C12	179.87 (17)
C11—C18—C19—C14	-179.91 (14)	C14—C13—N1—C1	176.31 (15)
C15—C14—C19—C18	0.5 (3)	C6—C1—N1—C13	106.50 (19)
C13—C14—C19—C18	179.61 (16)	C2—C1—N1—C13	-76.7 (2)
C25—C20—C21—C22	-3.6 (3)	C33—C32—N2—C20	-176.22 (15)
N2—C20—C21—C22	179.64 (16)	C25—C20—N2—C32	75.3 (2)
C25—C20—C21—C29	178.15 (18)	C21—C20—N2—C32	-107.84 (19)

Hydrogen-bond geometry (Å, °)

Cg1 and Cg3 are the centroids of the C1—C6 and C20—C25 rings, respectively.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1—H1...N1	0.82	1.92	2.646 (2)	147
O2—H2...N2	0.82	1.90	2.630 (2)	147

C31—H31c···Cg1	0.96	2.90	3.743 (3)	147
C12—H12A···Cg3 <sup>i</sup>	0.96	2.98	3.833 (3)	149

---

Symmetry code: (i)  $x-1, y, z$ .