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## Structure Reports

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# Ethyl 2-benzoyl-6-methylindolizine-7-carboxylate

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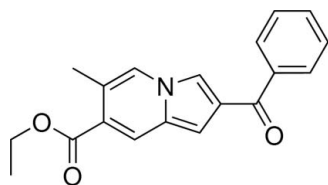
Received 30 March 2012; accepted 14 April 2012

 Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.056;  $wR$  factor = 0.166; data-to-parameter ratio = 14.9.

The title compound,  $\text{C}_{19}\text{H}_{17}\text{NO}_3$ , was synthesized using a tandem annulation reaction between 4-benzoyl-1*H*-pyrrole-2-carbaldehyde and (*E*)-ethyl 4-bromobut-2-enoate under mild conditions. The dihedral angle between the benzene ring and the indolizine ring system is  $41.73(4)^\circ$ .

## Related literature

For background to indolizines, see: Ge *et al.* (2009*a*, 2011). For bond lengths and angles in related structures, see: Ge *et al.* (2009*b*). For the synthesis of imidazo[1,2-*a*]pyridines *via* a tandem reaction, see: Jia *et al.* (2010).



## Experimental

### Crystal data

$\text{C}_{19}\text{H}_{17}\text{NO}_3$	$V = 1543.0(13)$ Å <sup>3</sup>
$M_r = 307.34$	$Z = 4$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 8.177(5)$ Å	$\mu = 0.09$ mm <sup>-1</sup>
$b = 17.243(5)$ Å	$T = 293$ K
$c = 11.191(5)$ Å	$0.18 \times 0.15 \times 0.14$ mm
$\beta = 102.070(5)^\circ$	

### Data collection

Bruker SMART APEX CCD area-detector diffractometer	8583 measured reflections
Absorption correction: multi-scan (SADABS; Bruker, 2005)	3150 independent reflections
$T_{\min} = 0.983$ , $T_{\max} = 0.990$	2434 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.124$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.056$	211 parameters
$wR(F^2) = 0.166$	H-atom parameters constrained
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.30$ e Å <sup>-3</sup>
3150 reflections	$\Delta\rho_{\text{min}} = -0.22$ e Å <sup>-3</sup>

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: XP in SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

The author thank Dr Qing Feng Wang, Taishan University, for the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG5204).

## References

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## supporting information

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**Ethyl 2-benzoyl-6-methylindolizine-7-carboxylate****Shang-Tie Liao****S1. Comment**

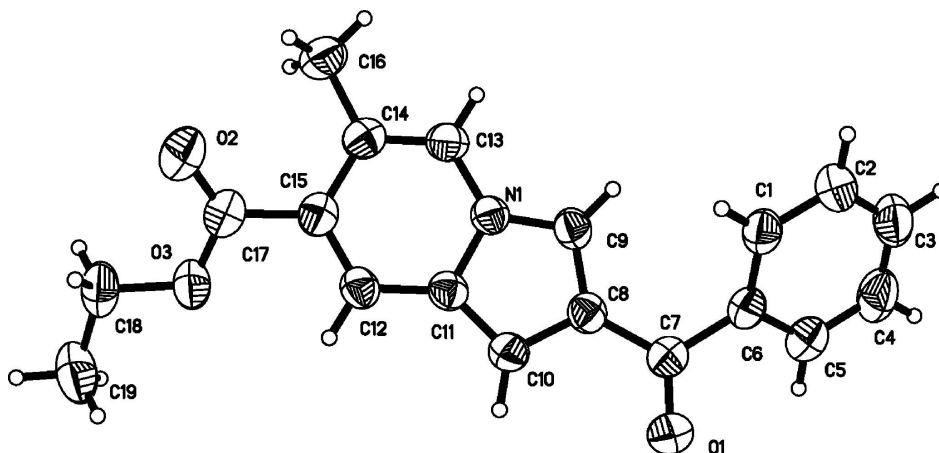
Indolizines have attracted considerable attention from medicinal and organic chemists because of the interesting similarities and diversions in structure to indole (Ge *et al.*; 2009a, 2011). Synthetic indolizines play important roles as calcium entry blockers, potential central nervous system depressants, 5-HT<sub>3</sub> receptor antagonist, histamine H<sub>3</sub> receptor antagonists, cardiovascular agents, and PLA<sub>2</sub> inhibitors. They have also drawn much attention owing to their possible usage as dyes and chemosensors. The title indolizine (I) (Fig. 1) was synthesized in order to study its biological properties. (I) was screened for anticancer activities and found to be inactive. We report here the crystal structure of the title compound. In the title compound, C<sub>19</sub>H<sub>17</sub>NO<sub>3</sub>, all bond lengths and angles show normal values (Ge *et al.*, 2009b). The dihedral angle between the benzene and indolizine rings is 41.73 (4)°.

**S2. Experimental**

To a 50 ml round-bottomed flask were added 4-benzoyl-1*H*-pyrrole-2-carbaldehyde (1.00 mmol), (*E*)-ethyl 4-bromobut-2-enoate (2.00 mmol), potassium carbonate (0.28 g, 2.05 mmol) and dry DMF (10 ml). The mixture was stirred at room temperature for 8 h. The solvent was removed under reduced pressure and a product was isolated by column chromatography on silica gel (yield 76%). Crystals of (I) suitable for X-ray diffraction were obtained by allowing a refluxed solution of the product in ethyl acetate to cool slowly to room temperature (without temperature control) and allowing the solvent to evaporate for 10 d.

**S3. Refinement**

All H atoms were placed in geometrically calculated positions and refined using a riding model with C—H = 0.97 Å (for CH<sub>2</sub> groups), 0.96 Å (for CH<sub>3</sub> groups) and 0.93 Å (for aromatic protons), their isotropic displacement parameters were set to 1.2 times (1.5 times for CH<sub>3</sub> groups) the equivalent displacement parameter of their parent atoms.

**Figure 1**

The molecular structure of (I), showing displacement ellipsoids drawn at the 50% probability level.

### Ethyl 2-benzoyl-6-methylindolizine-7-carboxylate

#### Crystal data

$C_{19}H_{17}NO_3$

$M_r = 307.34$

Monoclinic,  $P2_1/n$

Hall symbol:  $-P\ 2_1n$

$a = 8.177\ (5)\ \text{\AA}$

$b = 17.243\ (5)\ \text{\AA}$

$c = 11.191\ (5)\ \text{\AA}$

$\beta = 102.070\ (5)^\circ$

$V = 1543.0\ (13)\ \text{\AA}^3$

$Z = 4$

$F(000) = 648$

$D_x = 1.323\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71069\ \text{\AA}$

Cell parameters from 4026 reflections

$\theta = 2.4\text{--}28.4^\circ$

$\mu = 0.09\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Block, yellow

$0.18 \times 0.15 \times 0.14\ \text{mm}$

#### Data collection

Bruker SMART APEX CCD area-detector  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\phi$  and  $\omega$  scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2005)

$T_{\min} = 0.983$ ,  $T_{\max} = 0.990$

8583 measured reflections

3150 independent reflections

2434 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.124$

$\theta_{\max} = 26.4^\circ$ ,  $\theta_{\min} = 2.2^\circ$

$h = -10 \rightarrow 10$

$k = -17 \rightarrow 21$

$l = -13 \rightarrow 9$

#### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.056$

$wR(F^2) = 0.166$

$S = 1.04$

3150 reflections

211 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0781P)^2 + 0.2178P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.30\ \text{e \AA}^{-3}$

$\Delta\rho_{\min} = -0.22\ \text{e \AA}^{-3}$

Extinction correction: *SHELXL97* (Sheldrick,  
2008),  $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.023 (4)

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.5745 (2)	0.16484 (10)	1.05807 (18)	0.0509 (5)
H1	0.6122	0.1859	0.9922	0.061*
C2	0.4798 (3)	0.09802 (12)	1.0431 (2)	0.0657 (6)
H2	0.4536	0.0743	0.9668	0.079*
C3	0.4235 (3)	0.06598 (12)	1.1396 (2)	0.0694 (7)
H3	0.3572	0.0217	1.1282	0.083*
C4	0.4657 (3)	0.09960 (13)	1.2523 (2)	0.0649 (6)
H4	0.4298	0.0773	1.3180	0.078*
C5	0.5610 (2)	0.16633 (11)	1.26986 (19)	0.0522 (5)
H5	0.5901	0.1884	1.3472	0.063*
C6	0.61349 (19)	0.20050 (9)	1.17163 (16)	0.0419 (4)
C7	0.7090 (2)	0.27484 (10)	1.19416 (16)	0.0430 (4)
C8	0.6858 (2)	0.33474 (9)	1.09863 (16)	0.0411 (4)
C9	0.5640 (2)	0.33689 (9)	0.99243 (17)	0.0430 (4)
H9	0.4831	0.2992	0.9663	0.052*
C10	0.7812 (2)	0.40322 (9)	1.10421 (16)	0.0432 (4)
H10	0.8712	0.4169	1.1662	0.052*
C11	0.71730 (19)	0.44609 (9)	1.00163 (15)	0.0397 (4)
C12	0.7536 (2)	0.51834 (9)	0.95286 (17)	0.0426 (4)
H12	0.8394	0.5488	0.9967	0.051*
C13	0.4925 (2)	0.43088 (10)	0.82315 (17)	0.0459 (4)
H13	0.4036	0.4015	0.7811	0.055*
C14	0.5288 (2)	0.49875 (10)	0.77489 (16)	0.0454 (4)
C15	0.6659 (2)	0.54404 (9)	0.84358 (16)	0.0432 (4)
C16	0.4222 (3)	0.52534 (12)	0.6562 (2)	0.0625 (6)
H16A	0.3335	0.4888	0.6298	0.094*
H16B	0.4894	0.5290	0.5955	0.094*
H16C	0.3756	0.5753	0.6672	0.094*
C17	0.7185 (2)	0.61723 (11)	0.79141 (19)	0.0533 (5)
C18	0.8515 (2)	0.73858 (11)	0.8309 (2)	0.0616 (6)
H18A	0.7597	0.7674	0.7816	0.074*
H18B	0.9315	0.7264	0.7807	0.074*
C19	0.9329 (3)	0.78540 (12)	0.9384 (3)	0.0786 (7)
H19A	0.8522	0.7980	0.9865	0.118*
H19B	0.9765	0.8323	0.9110	0.118*

H19C	1.0225	0.7561	0.9870	0.118*
N1	0.58336 (16)	0.40372 (7)	0.93311 (12)	0.0394 (4)
O1	0.80406 (17)	0.28578 (8)	1.29271 (13)	0.0604 (4)
O2	0.7043 (3)	0.62853 (11)	0.68326 (16)	0.0966 (7)
O3	0.78952 (16)	0.66744 (7)	0.87619 (13)	0.0533 (4)

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0701 (11)	0.0331 (8)	0.0482 (11)	0.0025 (8)	0.0096 (8)	0.0030 (8)
C2	0.0881 (14)	0.0369 (9)	0.0660 (14)	-0.0047 (9)	0.0019 (11)	-0.0014 (9)
C3	0.0727 (13)	0.0413 (10)	0.0886 (18)	-0.0103 (9)	0.0041 (12)	0.0131 (11)
C4	0.0664 (12)	0.0542 (11)	0.0764 (15)	-0.0043 (9)	0.0202 (11)	0.0244 (11)
C5	0.0586 (10)	0.0493 (10)	0.0496 (11)	0.0033 (8)	0.0135 (8)	0.0081 (9)
C6	0.0465 (8)	0.0341 (8)	0.0451 (10)	0.0052 (6)	0.0094 (7)	0.0063 (7)
C7	0.0478 (8)	0.0384 (8)	0.0428 (10)	0.0027 (7)	0.0097 (7)	0.0009 (7)
C8	0.0484 (8)	0.0303 (8)	0.0447 (10)	0.0021 (6)	0.0097 (7)	-0.0028 (7)
C9	0.0485 (9)	0.0307 (8)	0.0493 (10)	-0.0017 (6)	0.0094 (7)	-0.0007 (7)
C10	0.0511 (9)	0.0346 (8)	0.0419 (10)	-0.0017 (7)	0.0049 (7)	-0.0040 (7)
C11	0.0457 (8)	0.0308 (8)	0.0424 (9)	0.0003 (6)	0.0088 (7)	-0.0056 (7)
C12	0.0506 (9)	0.0307 (8)	0.0469 (10)	-0.0021 (7)	0.0115 (7)	-0.0046 (7)
C13	0.0497 (9)	0.0399 (9)	0.0450 (10)	0.0019 (7)	0.0026 (7)	-0.0038 (8)
C14	0.0544 (9)	0.0392 (9)	0.0429 (10)	0.0078 (7)	0.0105 (7)	-0.0032 (7)
C15	0.0534 (9)	0.0329 (8)	0.0460 (10)	0.0048 (7)	0.0163 (7)	-0.0016 (7)
C16	0.0757 (13)	0.0548 (11)	0.0511 (12)	0.0037 (10)	-0.0005 (10)	0.0053 (10)
C17	0.0625 (11)	0.0446 (10)	0.0538 (12)	0.0022 (8)	0.0146 (9)	0.0094 (9)
C18	0.0614 (11)	0.0425 (10)	0.0830 (16)	-0.0018 (8)	0.0202 (10)	0.0214 (10)
C19	0.0987 (16)	0.0453 (11)	0.0980 (19)	-0.0186 (12)	0.0344 (14)	0.0007 (12)
N1	0.0462 (7)	0.0303 (7)	0.0412 (8)	0.0017 (5)	0.0077 (6)	-0.0027 (6)
O1	0.0713 (8)	0.0561 (8)	0.0481 (8)	-0.0085 (6)	-0.0003 (6)	0.0048 (6)
O2	0.1448 (16)	0.0845 (13)	0.0577 (11)	-0.0378 (12)	0.0149 (10)	0.0161 (10)
O3	0.0681 (8)	0.0339 (6)	0.0612 (9)	-0.0057 (5)	0.0208 (6)	0.0044 (6)

*Geometric parameters (Å, °)*

C1—C2	1.379 (3)	C11—C12	1.416 (2)
C1—C6	1.388 (3)	C12—C15	1.356 (2)
C1—H1	0.9300	C12—H12	0.9300
C2—C3	1.375 (3)	C13—C14	1.348 (3)
C2—H2	0.9300	C13—N1	1.380 (2)
C3—C4	1.365 (4)	C13—H13	0.9300
C3—H3	0.9300	C14—C15	1.448 (2)
C4—C5	1.381 (3)	C14—C16	1.500 (3)
C4—H4	0.9300	C15—C17	1.491 (2)
C5—C6	1.392 (3)	C16—H16A	0.9600
C5—H5	0.9300	C16—H16B	0.9600
C6—C7	1.495 (2)	C16—H16C	0.9600
C7—O1	1.224 (2)	C17—O2	1.207 (3)

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C7—C8	1.470 (2)	C17—O3	1.326 (2)
C8—C9	1.382 (2)	C18—O3	1.458 (2)
C8—C10	1.409 (2)	C18—C19	1.485 (3)
C9—N1	1.356 (2)	C18—H18A	0.9700
C9—H9	0.9300	C18—H18B	0.9700
C10—C11	1.373 (2)	C19—H19A	0.9600
C10—H10	0.9300	C19—H19B	0.9600
C11—N1	1.404 (2)	C19—H19C	0.9600
C2—C1—C6	119.8 (2)	C11—C12—H12	119.3
C2—C1—H1	120.1	C14—C13—N1	121.98 (15)
C6—C1—H1	120.1	C14—C13—H13	119.0
C3—C2—C1	120.8 (2)	N1—C13—H13	119.0
C3—C2—H2	119.6	C13—C14—C15	117.81 (16)
C1—C2—H2	119.6	C13—C14—C16	118.96 (16)
C4—C3—C2	119.6 (2)	C15—C14—C16	123.17 (16)
C4—C3—H3	120.2	C12—C15—C14	120.47 (16)
C2—C3—H3	120.2	C12—C15—C17	119.22 (16)
C3—C4—C5	120.8 (2)	C14—C15—C17	120.21 (16)
C3—C4—H4	119.6	C14—C16—H16A	109.5
C5—C4—H4	119.6	C14—C16—H16B	109.5
C4—C5—C6	119.9 (2)	H16A—C16—H16B	109.5
C4—C5—H5	120.1	C14—C16—H16C	109.5
C6—C5—H5	120.1	H16A—C16—H16C	109.5
C1—C6—C5	119.08 (17)	H16B—C16—H16C	109.5
C1—C6—C7	123.12 (16)	O2—C17—O3	123.24 (18)
C5—C6—C7	117.79 (16)	O2—C17—C15	123.64 (19)
O1—C7—C8	120.54 (16)	O3—C17—C15	113.05 (17)
O1—C7—C6	119.69 (16)	O3—C18—C19	107.76 (18)
C8—C7—C6	119.76 (14)	O3—C18—H18A	110.2
C9—C8—C10	107.95 (15)	C19—C18—H18A	110.2
C9—C8—C7	127.19 (15)	O3—C18—H18B	110.2
C10—C8—C7	124.77 (15)	C19—C18—H18B	110.2
N1—C9—C8	107.87 (14)	H18A—C18—H18B	108.5
N1—C9—H9	126.1	C18—C19—H19A	109.5
C8—C9—H9	126.1	C18—C19—H19B	109.5
C11—C10—C8	107.68 (14)	H19A—C19—H19B	109.5
C11—C10—H10	126.2	C18—C19—H19C	109.5
C8—C10—H10	126.2	H19A—C19—H19C	109.5
C10—C11—N1	107.06 (14)	H19B—C19—H19C	109.5
C10—C11—C12	136.21 (15)	C9—N1—C13	128.99 (14)
N1—C11—C12	116.74 (14)	C9—N1—C11	109.44 (14)
C15—C12—C11	121.40 (15)	C13—N1—C11	121.57 (14)
C15—C12—H12	119.3	C17—O3—C18	115.64 (17)

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