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catena-Poly[[acetato- κ O][4-(1H-pyrazol-3-yl)pyridine- κ N¹]zinc]- μ -acetato- κ^2 O:O']

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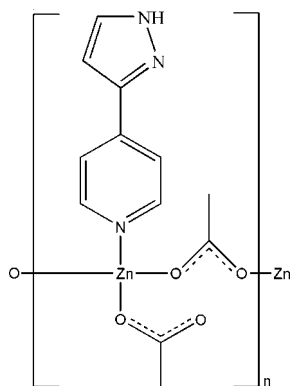
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.056; wR factor = 0.131; data-to-parameter ratio = 13.7.

In the title compound, $[\text{Zn}(\text{CH}_3\text{CO}_2)_2(\text{C}_8\text{H}_7\text{N}_3)]_n$, the Zn^{II} atom is coordinated by one N atom from a 4-(1H-pyrazol-3-yl)pyridine ligand and three O atoms from two bridging and one terminal acetate ligands, forming a distorted tetrahedral geometry. The bridging acetate ligands link the Zn atoms into a chain along [001]. N—H \cdots O hydrogen bonds and π – π interactions between the pyridine and pyrazole rings [centroid–centroid distance = 3.927 (3) Å] connect the chains into a layer parallel to (011).

Related literature

For background to complexes of 4-(1H-pyrazol-3-yl)pyridine, see: Davies *et al.* (2005). For the synthesis of the ligand, see: Davies *et al.* (2003).



Experimental

Crystal data

 $[\text{Zn}(\text{C}_2\text{H}_3\text{O}_2)_2(\text{C}_8\text{H}_7\text{N}_3)]$ $M_r = 322.58$

Monoclinic, $P2_1/c$
 $a = 16.371$ (3) Å
 $b = 8.8526$ (18) Å
 $c = 9.5041$ (19) Å
 $\beta = 94.18$ (3)°
 $V = 1373.7$ (5) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 1.80$ mm⁻¹
 $T = 293$ K
 $0.28 \times 0.23 \times 0.19$ mm

Data collection

Rigaku SCXmini CCD diffractometer
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)
 $T_{\text{min}} = 0.632$, $T_{\text{max}} = 0.726$

11822 measured reflections
2479 independent reflections
1957 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.068$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.056$
 $wR(F^2) = 0.131$
 $S = 1.06$
2479 reflections

181 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.51$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.54$ e Å⁻³

Table 1

Selected bond lengths (Å).

Zn1—N1	2.026 (4)	Zn1—O3	1.958 (3)
Zn1—O1	1.942 (4)	Zn1—O4 ⁱ	1.984 (3)

Symmetry code: (i) $x, -y + \frac{1}{2}, z + \frac{1}{2}$.

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N3—H3 \cdots O2 ⁱⁱ	0.86	1.93	2.769 (6)	163

Symmetry code: (ii) $-x, -y + 1, -z + 2$.

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEPII* (Johnson, 1976) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HY2465).

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supporting information

Acta Cryst. (2011). E67, m1512 [doi:10.1107/S1600536811040190]

***catena*-Poly[[*(acetato-κO)*[4-(1*H*-pyrazol-3-yl)pyridine-*κN*¹]*zinc*]-*μ*-acetato-*κ*²*O:O'*]**

Zheng-De Tan, Feng-Jiao Tan, Bo Tan and Zhi-Wen Yi

S1. Comment

Pyridine derivatives are an important class of ligands for constructing metal–organic frameworks. 4-(1*H*-Pyrazol-3-yl)pyridine can be used as pyridine ligand in building coordination compounds (Davies *et al.*, 2005). In the present paper, we report the synthesis and structure of the title compound.

As shown in Fig. 1, the Zn^{II} atom exhibits a distorted tetrahedral coordination geometry, defined by one N atom from a 4-(1*H*-pyrazol-3-yl)pyridine ligand and three O atoms from two types of acetate ligands (Table 1). One acetate anion coordinates the Zn atom as a monodentate terminal ligand. The other acetate anion links the Zn atoms *via* two O atoms, forming a one-dimensional chain along [0 0 1] (Fig. 2). N—H⋯O hydrogen bonds (Table 2) and π – π interactions between the pyridine and pyrazole rings [centroid–centroid distance = 3.927 (3) Å] connect the chains into a layer parallel to (0 1 1) (Fig. 3).

S2. Experimental

4-(1*H*-Pyrazol-3-yl)pyridine was prepared according to the published method of Davies *et al.* (2003). An aqueous solution (20 ml) containing zinc acetate (0.1 mmol, 22 mg) and 4-(1*H*-pyrazol-3-yl)pyridine (0.2 mmol, 29 mg) was stirred for a few minutes in air. Colorless crystals were obtained by allowing the solution to stand at room temperature for a few weeks.

S3. Refinement

H atoms were placed at calculated positions and refined as riding atoms, with C—H = 0.93 and N—H = 0.86 Å and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$.

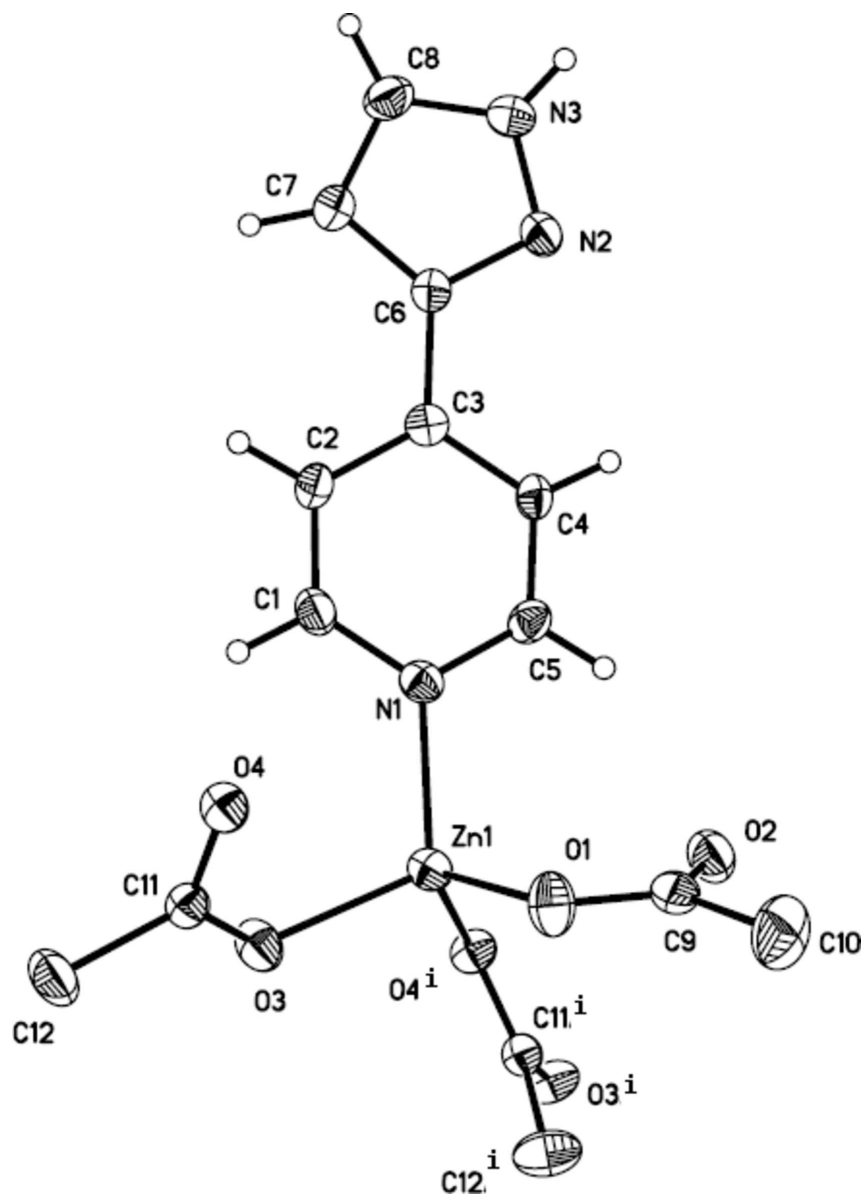
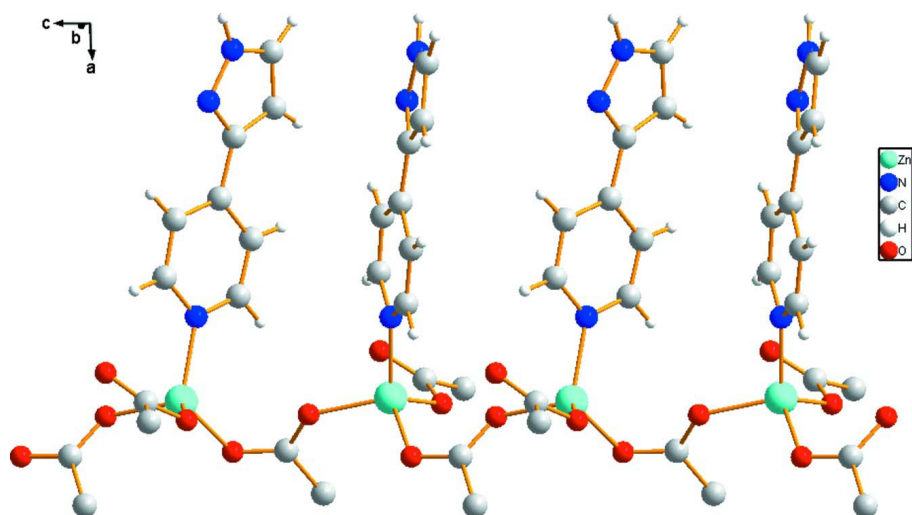
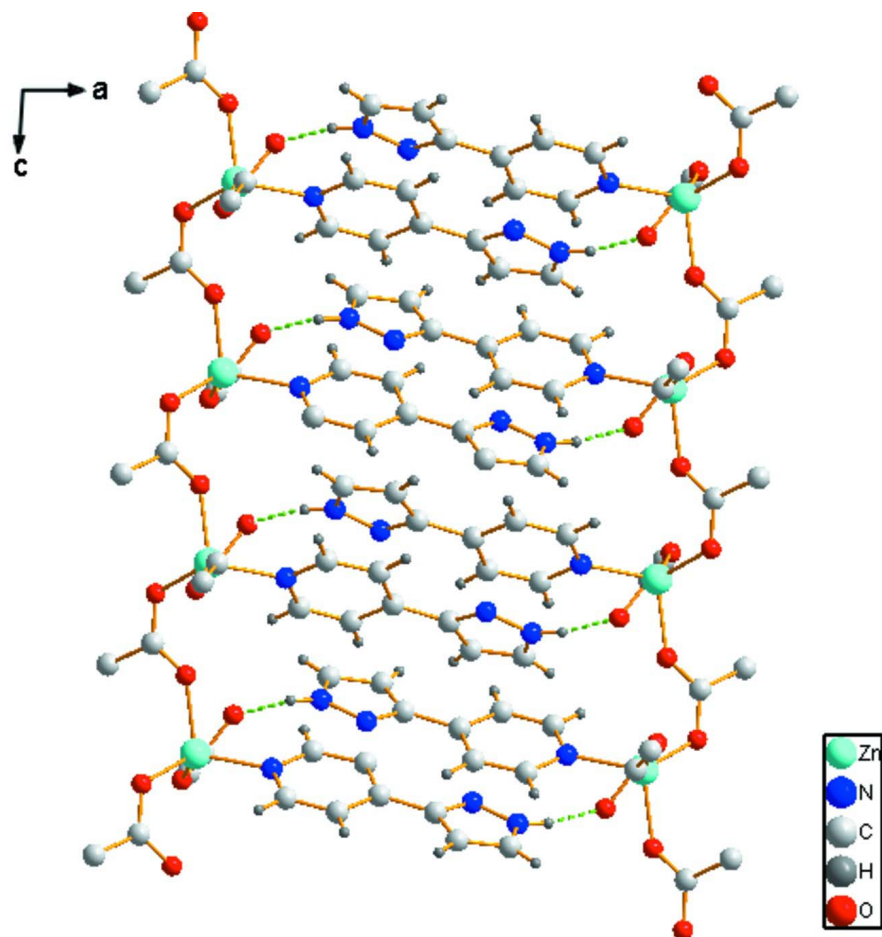


Figure 1

The asymmetric unit of the title compound. Displacement ellipsoids are drawn at the 30% probability level. [Symmetry code: (i) $x, 3/2-y, 1/2+z$.]

**Figure 2**

A view of the one-dimensional structure of the title compound along $[0\ 0\ 1]$.

**Figure 3**

A view of the layer network. Hydrogen bonds are shown as dashed lines.

catena-Poly[[acetato-κO][4-(1H-pyrazol-3-yl)pyridine-κN¹]zinc]-μ-acetato-κ²O:O']

Crystal data

[Zn(C₂H₃O₂)₂(C₈H₇N₃)]
M_r = 322.58
 Monoclinic, *P*2₁/*c*
 Hall symbol: -*P* 2ybc
a = 16.371 (3) Å
b = 8.8526 (18) Å
c = 9.5041 (19) Å
 β = 94.18 (3)°
V = 1373.7 (5) Å³
Z = 4

F(000) = 648
D_x = 1.560 Mg m⁻³
 Mo *Kα* radiation, λ = 0.71073 Å
 Cell parameters from 11614 reflections
 θ = 3.2–27.6°
 μ = 1.80 mm⁻¹
T = 293 K
 Block, colourless
 0.28 × 0.23 × 0.19 mm

Data collection

Rigaku SCXmini CCD
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 ω scans
 Absorption correction: multi-scan
 (*CrystalClear*; Rigaku, 2005)
T_{min} = 0.632, *T_{max}* = 0.726

11822 measured reflections
 2479 independent reflections
 1957 reflections with *I* > 2σ(*I*)
R_{int} = 0.068
 θ_{max} = 25.2°, θ_{min} = 3.2°
h = -19→19
k = -10→10
l = -11→11

Refinement

Refinement on *F*²
 Least-squares matrix: full
R[*F*² > 2σ(*F*²)] = 0.056
wR(*F*²) = 0.131
S = 1.06
 2479 reflections
 181 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
w = 1/[σ²(*F_o*²) + (0.063*P*)² + 1.9274*P*]
 where *P* = (*F_o*² + 2*F_c*²)/3
 (Δ/σ)_{max} < 0.001
 Δρ_{max} = 0.51 e Å⁻³
 Δρ_{min} = -0.54 e Å⁻³

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U_{iso}</i> [*] / <i>U_{eq}</i>
Zn1	0.34131 (3)	0.68665 (6)	1.02084 (6)	0.0343 (2)
N1	0.2233 (2)	0.7526 (5)	0.9820 (4)	0.0348 (9)
C3	0.0573 (3)	0.8142 (5)	0.9123 (5)	0.0325 (11)
C4	0.0820 (3)	0.6924 (5)	0.9969 (5)	0.0362 (11)
H4	0.0430	0.6294	1.0325	0.043*
C5	0.1635 (3)	0.6647 (6)	1.0280 (5)	0.0372 (12)
H5	0.1784	0.5813	1.0836	0.045*
C1	0.1992 (3)	0.8734 (6)	0.9045 (5)	0.0403 (12)
H1	0.2393	0.9371	0.8736	0.048*
C2	0.1192 (3)	0.9082 (5)	0.8682 (5)	0.0365 (12)
H2	0.1059	0.9938	0.8146	0.044*
N3	-0.1561 (3)	0.7919 (5)	0.8372 (5)	0.0491 (12)
H3	-0.2025	0.7466	0.8393	0.059*

C6	-0.0297 (3)	0.8401 (5)	0.8676 (5)	0.0331 (11)
N2	-0.0849 (3)	0.7342 (5)	0.8937 (5)	0.0454 (11)
C7	-0.0660 (3)	0.9619 (6)	0.7937 (5)	0.0438 (13)
H7	-0.0402	1.0480	0.7624	0.053*
C8	-0.1469 (3)	0.9277 (7)	0.7774 (6)	0.0495 (14)
H8	-0.1880	0.9870	0.7333	0.059*
O3	0.4195 (2)	0.8246 (4)	0.9418 (3)	0.0453 (9)
O1	0.3590 (2)	0.4837 (4)	0.9518 (4)	0.0500 (10)
O2	0.2878 (2)	0.3956 (4)	1.1226 (4)	0.0501 (10)
C9	0.3292 (3)	0.3761 (6)	1.0208 (6)	0.0405 (12)
C10	0.3494 (5)	0.2163 (7)	0.9709 (8)	0.072 (2)
C11	0.4183 (3)	0.8446 (5)	0.8100 (5)	0.0334 (11)
C12	0.4840 (3)	0.9438 (7)	0.7551 (6)	0.0578 (17)
O4	0.3643 (2)	0.7841 (4)	0.7269 (3)	0.0373 (8)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Zn1	0.0330 (3)	0.0415 (4)	0.0283 (3)	-0.0030 (3)	0.0026 (2)	-0.0012 (3)
N1	0.035 (2)	0.039 (2)	0.030 (2)	-0.004 (2)	0.0011 (18)	-0.0030 (19)
C3	0.034 (3)	0.034 (3)	0.030 (3)	0.003 (2)	0.005 (2)	-0.005 (2)
C4	0.039 (3)	0.034 (3)	0.036 (3)	-0.004 (2)	0.007 (2)	0.008 (2)
C5	0.040 (3)	0.041 (3)	0.030 (3)	0.003 (2)	0.003 (2)	0.006 (2)
C1	0.040 (3)	0.041 (3)	0.040 (3)	-0.010 (2)	0.004 (2)	0.003 (2)
C2	0.042 (3)	0.032 (3)	0.036 (3)	0.001 (2)	0.005 (2)	0.007 (2)
N3	0.032 (2)	0.053 (3)	0.061 (3)	-0.001 (2)	-0.002 (2)	0.007 (2)
C6	0.037 (3)	0.031 (3)	0.031 (3)	0.000 (2)	0.004 (2)	0.000 (2)
N2	0.034 (3)	0.039 (3)	0.063 (3)	-0.004 (2)	0.003 (2)	0.008 (2)
C7	0.045 (3)	0.043 (3)	0.043 (3)	-0.002 (3)	0.000 (2)	0.010 (3)
C8	0.042 (3)	0.054 (4)	0.052 (3)	0.009 (3)	-0.001 (3)	0.010 (3)
O3	0.048 (2)	0.061 (2)	0.0267 (19)	-0.0171 (18)	0.0011 (15)	0.0038 (17)
O1	0.068 (3)	0.038 (2)	0.046 (2)	-0.0065 (19)	0.0162 (19)	-0.0039 (17)
O2	0.044 (2)	0.046 (2)	0.062 (3)	-0.0055 (18)	0.0122 (19)	0.0023 (19)
C9	0.036 (3)	0.042 (3)	0.042 (3)	-0.004 (2)	-0.008 (2)	-0.001 (3)
C10	0.094 (5)	0.037 (4)	0.086 (5)	0.005 (3)	0.006 (4)	-0.013 (3)
C11	0.036 (3)	0.035 (3)	0.030 (3)	0.002 (2)	0.004 (2)	0.001 (2)
C12	0.047 (3)	0.075 (4)	0.052 (4)	-0.023 (3)	0.005 (3)	0.015 (3)
O4	0.042 (2)	0.044 (2)	0.0258 (17)	-0.0044 (16)	0.0040 (15)	-0.0009 (15)

Geometric parameters (Å, °)

Zn1—N1	2.026 (4)	N3—C8	1.344 (7)
Zn1—O1	1.942 (4)	N3—N2	1.348 (6)
Zn1—O3	1.958 (3)	N3—H3	0.8600
Zn1—O4 ⁱ	1.984 (3)	C6—N2	1.338 (6)
N1—C1	1.341 (6)	C6—C7	1.395 (7)
N1—C5	1.348 (6)	C7—C8	1.356 (7)
C3—C4	1.387 (6)	C7—H7	0.9300

C3—C2	1.398 (7)	C8—H8	0.9300
C3—C6	1.475 (7)	O3—C11	1.264 (6)
C4—C5	1.368 (7)	O1—C9	1.274 (6)
C4—H4	0.9300	O2—C9	1.233 (6)
C5—H5	0.9300	C9—C10	1.536 (8)
C1—C2	1.365 (7)	C11—O4	1.261 (5)
C1—H1	0.9300	C11—C12	1.510 (7)
C2—H2	0.9300		
O1—Zn1—O3	109.25 (16)	C3—C2—H2	120.2
O1—Zn1—O4 ⁱ	115.59 (15)	C8—N3—N2	112.7 (4)
O3—Zn1—O4 ⁱ	102.40 (14)	C8—N3—H3	123.6
O1—Zn1—N1	111.65 (17)	N2—N3—H3	123.6
O3—Zn1—N1	113.05 (16)	N2—C6—C7	111.5 (4)
O4 ⁱ —Zn1—N1	104.60 (15)	N2—C6—C3	119.3 (4)
C1—N1—C5	116.5 (4)	C7—C6—C3	129.2 (4)
C1—N1—Zn1	124.6 (3)	C6—N2—N3	103.7 (4)
C5—N1—Zn1	118.7 (3)	C8—C7—C6	105.1 (5)
C4—C3—C2	116.7 (4)	C8—C7—H7	127.4
C4—C3—C6	121.4 (4)	C6—C7—H7	127.4
C2—C3—C6	121.8 (4)	N3—C8—C7	106.9 (5)
C5—C4—C3	120.2 (4)	N3—C8—H8	126.5
C5—C4—H4	119.9	C7—C8—H8	126.5
C3—C4—H4	119.9	C11—O3—Zn1	120.3 (3)
N1—C5—C4	123.1 (5)	C9—O1—Zn1	116.5 (3)
N1—C5—H5	118.4	O2—C9—O1	123.5 (5)
C4—C5—H5	118.4	O2—C9—C10	121.0 (5)
N1—C1—C2	123.9 (5)	O1—C9—C10	115.5 (5)
N1—C1—H1	118.1	O4—C11—O3	121.4 (4)
C2—C1—H1	118.1	O4—C11—C12	121.0 (4)
C1—C2—C3	119.5 (5)	O3—C11—C12	117.6 (4)
C1—C2—H2	120.2	C11—O4—Zn1 ⁱⁱ	129.6 (3)

Symmetry codes: (i) $x, -y+3/2, z+1/2$; (ii) $x, -y+3/2, z-1/2$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N3—H3 ⁱⁱⁱ —O2 ⁱⁱⁱ	0.86	1.93	2.769 (6)	163

Symmetry code: (iii) $-x, -y+1, -z+2$.