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## Structure Reports

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**Pentacarbonyl-1 $\kappa^2$ C,2 $\kappa^3$ C-[(diphenylphosphoryl)diphenylphosphane-1 $\kappa$ P]- $\mu$ -ethane-1,2-dithiolato-1:2 $\kappa^4$ S,S':S,S'-diiron(I)(Fe—Fe)**

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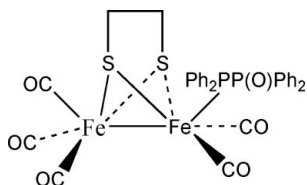
Received 7 October 2011; accepted 12 October 2011

Key indicators: single-crystal X-ray study;  $T = 113$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.026;  $wR$  factor = 0.059; data-to-parameter ratio = 18.8.

The dinuclear title compound,  $[\text{Fe}_2(\text{C}_2\text{H}_4\text{S}_2)(\text{C}_{24}\text{H}_{20}\text{OP}_2)(\text{CO})_5]$  or  $(\mu\text{-SCH}_2\text{CH}_2\text{S}-\mu)\text{Fe}_2(\text{CO})_5[\text{Ph}_2\text{PP}(\text{O})\text{Ph}_2]$ , contains a butterfly-shaped  $\text{Fe}_2\text{S}_2$  core in which the  $\text{Fe}\cdots\text{Fe}$  separation is 2.5275 (6) Å. One of the Fe atoms is also coordinated to three carbonyl ligands and the other to two carbonyl ligands and one phosphane ligand  $[\text{Ph}_2\text{PP}(\text{O})\text{Ph}_2]$ . Both Fe-atom geometries could be described as grossly distorted octahedral and the  $\text{Ph}_2\text{PP}(\text{O})\text{Ph}_2$  ligand lies *trans* to the  $\text{Fe}\cdots\text{Fe}$  link.

## Related literature

For more details about diiron dithiolate complexes, see: Song *et al.* (2005); Wang *et al.* (2009); Yin *et al.* (2011).



## Experimental

## Crystal data

 $[\text{Fe}_2(\text{C}_2\text{H}_4\text{S}_2)(\text{C}_{24}\text{H}_{20}\text{OP}_2)(\text{CO})_5]$   $M_r = 730.26$ 

Monoclinic,  $P2_1/n$   
 $a = 13.865$  (4) Å  
 $b = 15.398$  (4) Å  
 $c = 14.459$  (5) Å  
 $\beta = 98.357$  (4)°  
 $V = 3054.1$  (16) Å<sup>3</sup>

$Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 1.24$  mm<sup>-1</sup>  
 $T = 113$  K  
 $0.20 \times 0.18 \times 0.10$  mm

## Data collection

Rigaku Saturn724 CCD diffractometer  
 Absorption correction: multi-scan (*CrystalClear*; Rigaku/MS, 2005)  
 $T_{\min} = 0.790$ ,  $T_{\max} = 0.886$

31351 measured reflections  
 7281 independent reflections  
 6102 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.039$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.026$   
 $wR(F^2) = 0.059$   
 $S = 1.06$   
 7281 reflections

388 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.36$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.36$  e Å<sup>-3</sup>

Table 1

Selected bond lengths (Å).

Fe1—C2	1.7855 (18)	Fe2—C4	1.7733 (17)
Fe1—C1	1.7981 (17)	Fe2—C5	1.7742 (17)
Fe1—C3	1.8006 (18)	Fe2—P1	2.2426 (7)
Fe1—S1	2.2484 (6)	Fe2—S1	2.2495 (7)
Fe1—S2	2.2495 (8)	Fe2—S2	2.2530 (7)

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *CrystalStructure* (Rigaku/MS, 2005).

This work was supported by the Ningbo petrochemical application-oriented base projects (Jd080109 and Jd090110).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6442).

## References

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 Wang, N., Wang, M., Liu, J., Jin, K., Chen, L. & Sun, L. (2009). *Inorg. Chem.* **48**, 11551–11558.  
 Yin, B. S., Li, T. B. & Yang, M. S. (2011). *J. Coord. Chem.* **64**, 2066–2074.

## supporting information

*Acta Cryst.* (2011). E67, m1552 [doi:10.1107/S1600536811042139]

**Pentacarbonyl- $1\kappa^2\text{C}, 2\kappa^3\text{C}$ -[(diphenylphosphoryl)diphenylphosphane- $1\kappa\text{P}$ ]- $\mu$ -ethane-1,2-dithiolato- $1:2\kappa^4\text{S}, \text{S}':\text{S}, \text{S}'$ -diiron(I) (Fe—Fe)**

**Xu-Feng Liu and Xiao-Yong Yu**

### S1. Comment

Diiron dithiolate complexes have received much attention in recent years due to their structures close to the active site of [FeFe]-hydrogenases (Song *et al.* (2005), Wang *et al.* (2009), Yin *et al.* (2011)). In continuation of our work in this area, the title complex, (I), was synthesized and its structure was determined by X-ray crystallography.

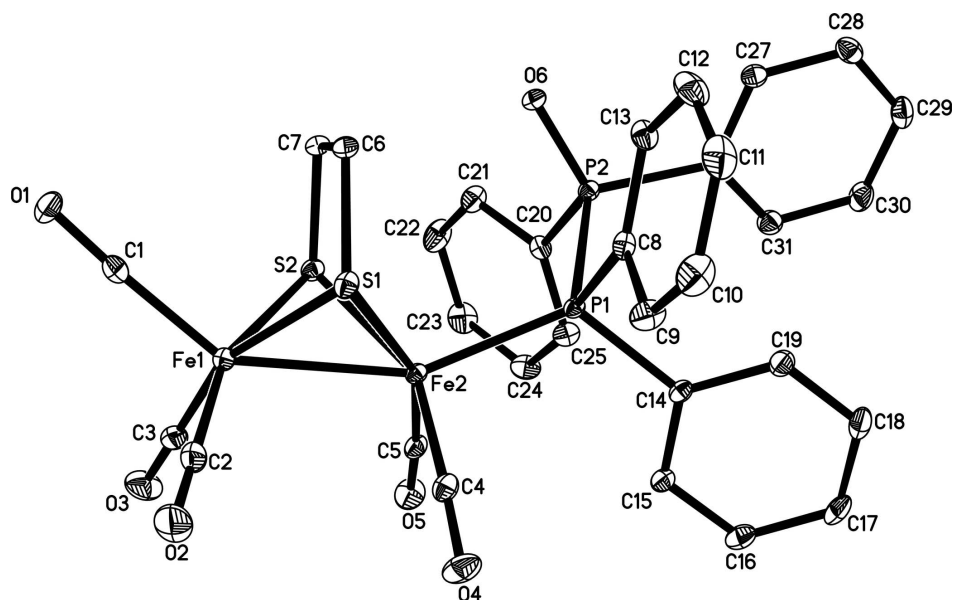
As shown in Fig. 1, the title complex contains five carbonyls and one  $\text{Ph}_2\text{PP}(\text{O})\text{Ph}_2$  ligands. The diiron ethanedithiolate cluster consists of two fused five-membered rings. The  $\text{Ph}_2\text{PP}(\text{O})\text{Ph}_2$  ligands occupies an axial position of the square-pyramidal geometry of the Fe atom.

### S2. Experimental

The title complex was prepared from  $(\mu\text{-SCH}_2\text{CH}_2\text{S-}\mu)\text{Fe}_2(\text{CO})_6$  and 1,2-bis(diphenylphosphino)cyclopentane in the presence of  $\text{Me}_3\text{NO}$ . Colourless prisms were grown from slow evaporation of dichloromethane and hexane solution at room temperature.

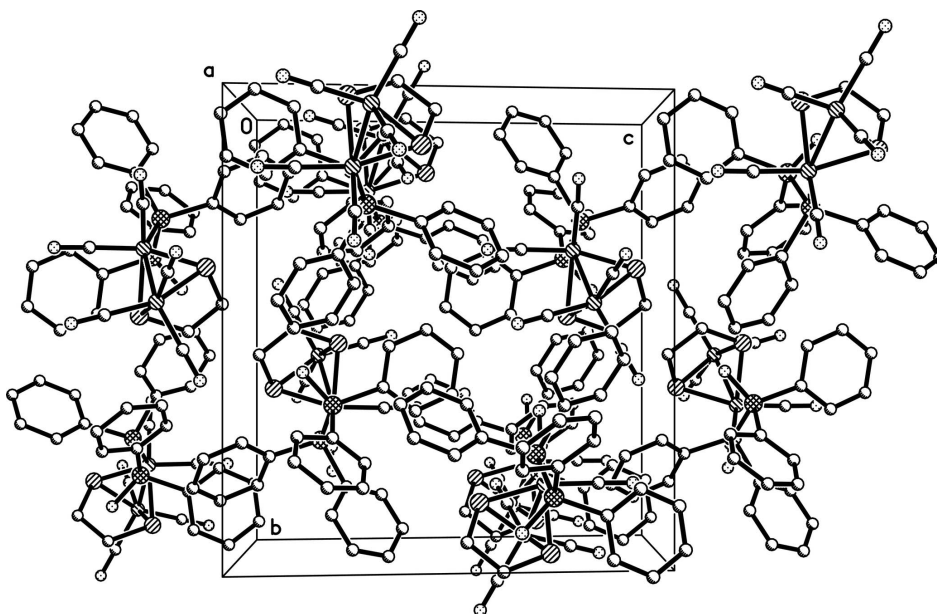
### S3. Refinement

All the H atoms were positioned geometrically (C—H = 0.93–0.97 Å) and refined as riding with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $1.5U_{\text{eq}}(\text{methyl C})$ .



**Figure 1**

The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.



**Figure 2**

The crystal packing for (I).

**Pentacarbonyl- $1\kappa^2C,2\kappa^3C$ - [(diphenylphosphoryl)diphenylphosphane- $1\kappa P$ ]- $\mu$ -ethane-1,2-dithiolato- $1:2\kappa^4S,S':S,S'$ -diiron(I) (Fe—Fe)**

*Crystal data*

$[\text{Fe}_2(\text{C}_2\text{H}_4\text{S}_2)(\text{C}_{24}\text{H}_{20}\text{OP}_2)(\text{CO})_5]$   
 $M_r = 730.26$

Monoclinic,  $P2_1/n$   
Hall symbol: -P 2yn

$a = 13.865$  (4) Å  
 $b = 15.398$  (4) Å  
 $c = 14.459$  (5) Å  
 $\beta = 98.357$  (4)°  
 $V = 3054.1$  (16) Å<sup>3</sup>  
 $Z = 4$   
 $F(000) = 1488$   
 $D_x = 1.588$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å  
 Cell parameters from 11006 reflections  
 $\theta = 1.4$ – $27.9$ °  
 $\mu = 1.24$  mm<sup>-1</sup>  
 $T = 113$  K  
 Prism, colorless  
 $0.20 \times 0.18 \times 0.10$  mm

*Data collection*

Rigaku Saturn724 CCD  
 diffractometer  
 Radiation source: rotating anode  
 Multilayer monochromator  
 Detector resolution: 14.22 pixels mm<sup>-1</sup>  
 $\omega$  and  $\phi$  scans  
 Absorption correction: multi-scan  
 (*CrystalClear*; Rigaku/MSC, 2005)  
 $T_{\min} = 0.790$ ,  $T_{\max} = 0.886$

31351 measured reflections  
 7281 independent reflections  
 6102 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.039$   
 $\theta_{\max} = 27.9$ °,  $\theta_{\min} = 1.9$ °  
 $h = -18 \rightarrow 18$   
 $k = -17 \rightarrow 20$   
 $l = -19 \rightarrow 19$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.026$   
 $wR(F^2) = 0.059$   
 $S = 1.06$   
 7281 reflections  
 388 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.024P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.002$   
 $\Delta\rho_{\max} = 0.36$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.36$  e Å<sup>-3</sup>

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
Fe1	1.261528 (16)	0.058771 (14)	0.331352 (16)	0.01557 (6)
Fe2	1.133418 (15)	0.174298 (14)	0.288051 (14)	0.01275 (6)
P1	0.99707 (3)	0.24364 (3)	0.31226 (3)	0.01291 (9)
P2	0.86096 (3)	0.16331 (3)	0.27704 (3)	0.01404 (9)
S1	1.18853 (3)	0.13327 (3)	0.43559 (3)	0.01697 (9)
S2	1.10474 (3)	0.03046 (2)	0.27533 (3)	0.01596 (9)
O1	1.30120 (9)	-0.10606 (7)	0.43282 (8)	0.0300 (3)
O2	1.43873 (9)	0.15892 (8)	0.39375 (10)	0.0411 (4)

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O3	1.32473 (9)	0.01325 (8)	0.15188 (9)	0.0366 (3)
O4	1.26144 (9)	0.32544 (8)	0.29575 (9)	0.0321 (3)
O5	1.12828 (9)	0.17617 (8)	0.08534 (8)	0.0298 (3)
O6	0.86136 (7)	0.09157 (6)	0.34670 (7)	0.0177 (2)
C1	1.28842 (11)	-0.04173 (11)	0.39306 (11)	0.0209 (4)
C2	1.36981 (12)	0.11973 (11)	0.36834 (13)	0.0250 (4)
C3	1.29927 (12)	0.03076 (10)	0.22113 (12)	0.0232 (4)
C4	1.20985 (12)	0.26713 (10)	0.29551 (11)	0.0193 (3)
C5	1.12641 (11)	0.17650 (10)	0.16457 (11)	0.0185 (3)
C6	1.09760 (11)	0.05588 (10)	0.46439 (11)	0.0209 (4)
H6A	1.1266	0.0200	0.5182	0.025*
H6B	1.0418	0.0881	0.4833	0.025*
C7	1.06081 (12)	-0.00340 (10)	0.38252 (11)	0.0195 (3)
H7A	0.9886	-0.0031	0.3724	0.023*
H7B	1.0827	-0.0636	0.3979	0.023*
C8	0.99500 (11)	0.27904 (10)	0.43317 (10)	0.0167 (3)
C9	1.05975 (13)	0.34539 (11)	0.46564 (12)	0.0259 (4)
H9	1.1029	0.3676	0.4260	0.031*
C10	1.06217 (14)	0.37937 (12)	0.55452 (12)	0.0328 (4)
H10	1.1068	0.4244	0.5756	0.039*
C11	0.99951 (13)	0.34765 (12)	0.61251 (12)	0.0310 (4)
H11	1.0001	0.3716	0.6731	0.037*
C12	0.93619 (13)	0.28118 (12)	0.58215 (12)	0.0303 (4)
H12	0.8940	0.2589	0.6226	0.036*
C13	0.93332 (11)	0.24635 (11)	0.49297 (11)	0.0209 (4)
H13	0.8895	0.2004	0.4729	0.025*
C14	0.96328 (11)	0.34554 (9)	0.24976 (11)	0.0151 (3)
C15	1.00002 (11)	0.36534 (10)	0.16773 (11)	0.0186 (3)
H15	1.0464	0.3279	0.1462	0.022*
C16	0.96935 (12)	0.43952 (10)	0.11709 (12)	0.0247 (4)
H16	0.9938	0.4517	0.0604	0.030*
C17	0.90373 (12)	0.49556 (11)	0.14854 (12)	0.0258 (4)
H17	0.8830	0.5462	0.1137	0.031*
C18	0.86815 (12)	0.47762 (11)	0.23126 (12)	0.0254 (4)
H18	0.8233	0.5163	0.2535	0.031*
C19	0.89793 (11)	0.40335 (10)	0.28154 (11)	0.0210 (4)
H19	0.8735	0.3916	0.3384	0.025*
C20	0.86646 (11)	0.12448 (10)	0.16014 (10)	0.0167 (3)
C21	0.85652 (12)	0.03529 (10)	0.14617 (11)	0.0235 (4)
H21	0.8466	-0.0013	0.1968	0.028*
C22	0.86095 (14)	-0.00041 (12)	0.05905 (12)	0.0327 (4)
H22	0.8525	-0.0612	0.0498	0.039*
C23	0.87764 (13)	0.05197 (12)	-0.01452 (12)	0.0308 (4)
H23	0.8817	0.0272	-0.0740	0.037*
C24	0.88851 (12)	0.14079 (12)	-0.00128 (12)	0.0257 (4)
H24	0.9003	0.1769	-0.0517	0.031*
C25	0.88221 (11)	0.17707 (11)	0.08506 (11)	0.0214 (4)
H25	0.8886	0.2381	0.0934	0.026*

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C26	0.75306 (11)	0.22975 (10)	0.27313 (10)	0.0156 (3)
C27	0.70310 (11)	0.22562 (10)	0.35030 (11)	0.0185 (3)
H27	0.7270	0.1897	0.4020	0.022*
C28	0.61916 (11)	0.27358 (10)	0.35174 (12)	0.0224 (4)
H28	0.5855	0.2704	0.4043	0.027*
C29	0.58382 (12)	0.32645 (10)	0.27659 (12)	0.0252 (4)
H29	0.5263	0.3597	0.2777	0.030*
C30	0.63304 (12)	0.33036 (11)	0.20009 (12)	0.0265 (4)
H30	0.6088	0.3662	0.1485	0.032*
C31	0.71692 (11)	0.28270 (10)	0.19801 (11)	0.0216 (4)
H31	0.7502	0.2860	0.1451	0.026*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Fe1	0.01547 (12)	0.01379 (12)	0.01783 (12)	0.00070 (9)	0.00372 (9)	0.00033 (9)
Fe2	0.01568 (12)	0.01181 (12)	0.01079 (11)	-0.00046 (8)	0.00202 (8)	0.00040 (8)
P1	0.0163 (2)	0.0118 (2)	0.01058 (19)	-0.00003 (15)	0.00202 (15)	0.00057 (15)
P2	0.0160 (2)	0.0135 (2)	0.01251 (19)	-0.00015 (15)	0.00168 (15)	0.00074 (15)
S1	0.0216 (2)	0.0164 (2)	0.01245 (19)	0.00252 (16)	0.00080 (15)	0.00018 (15)
S2	0.0180 (2)	0.0133 (2)	0.0166 (2)	-0.00165 (15)	0.00262 (15)	-0.00067 (15)
O1	0.0367 (7)	0.0210 (7)	0.0332 (7)	0.0076 (5)	0.0084 (6)	0.0089 (6)
O2	0.0249 (7)	0.0301 (8)	0.0672 (11)	-0.0091 (6)	0.0033 (7)	-0.0059 (7)
O3	0.0429 (8)	0.0386 (8)	0.0332 (8)	-0.0044 (6)	0.0223 (7)	-0.0082 (6)
O4	0.0317 (7)	0.0245 (7)	0.0378 (8)	-0.0127 (6)	-0.0031 (6)	0.0043 (6)
O5	0.0450 (8)	0.0307 (7)	0.0149 (6)	0.0051 (6)	0.0088 (6)	0.0003 (5)
O6	0.0208 (6)	0.0153 (6)	0.0174 (6)	-0.0001 (4)	0.0044 (5)	0.0040 (5)
C1	0.0178 (9)	0.0233 (10)	0.0224 (9)	0.0015 (7)	0.0059 (7)	-0.0028 (7)
C2	0.0224 (10)	0.0193 (9)	0.0341 (10)	0.0032 (7)	0.0069 (8)	0.0000 (8)
C3	0.0235 (9)	0.0171 (9)	0.0303 (10)	-0.0024 (7)	0.0083 (8)	0.0009 (7)
C4	0.0229 (9)	0.0187 (9)	0.0152 (8)	0.0021 (7)	-0.0007 (6)	0.0022 (7)
C5	0.0231 (9)	0.0126 (8)	0.0201 (9)	0.0006 (6)	0.0038 (7)	0.0004 (7)
C6	0.0242 (9)	0.0216 (9)	0.0190 (9)	0.0043 (7)	0.0100 (7)	0.0081 (7)
C7	0.0198 (9)	0.0160 (9)	0.0241 (9)	-0.0004 (6)	0.0083 (7)	0.0062 (7)
C8	0.0216 (8)	0.0162 (8)	0.0117 (8)	0.0048 (6)	-0.0002 (6)	-0.0015 (6)
C9	0.0349 (10)	0.0230 (9)	0.0198 (9)	-0.0062 (7)	0.0040 (7)	-0.0029 (7)
C10	0.0465 (12)	0.0260 (10)	0.0241 (10)	-0.0026 (8)	-0.0007 (8)	-0.0094 (8)
C11	0.0419 (12)	0.0354 (11)	0.0147 (9)	0.0088 (9)	0.0010 (8)	-0.0093 (8)
C12	0.0296 (10)	0.0450 (12)	0.0175 (9)	0.0072 (8)	0.0078 (7)	-0.0010 (8)
C13	0.0216 (9)	0.0253 (10)	0.0158 (8)	0.0043 (7)	0.0023 (7)	-0.0009 (7)
C14	0.0170 (8)	0.0127 (8)	0.0148 (8)	-0.0018 (6)	-0.0008 (6)	0.0007 (6)
C15	0.0203 (9)	0.0150 (8)	0.0205 (8)	-0.0011 (6)	0.0032 (7)	0.0012 (7)
C16	0.0294 (10)	0.0226 (9)	0.0215 (9)	-0.0025 (7)	0.0014 (7)	0.0079 (7)
C17	0.0284 (10)	0.0150 (9)	0.0312 (10)	0.0007 (7)	-0.0050 (8)	0.0072 (7)
C18	0.0252 (10)	0.0165 (9)	0.0336 (10)	0.0052 (7)	0.0007 (8)	-0.0004 (7)
C19	0.0239 (9)	0.0174 (9)	0.0219 (9)	0.0002 (7)	0.0040 (7)	0.0006 (7)
C20	0.0152 (8)	0.0194 (9)	0.0150 (8)	0.0009 (6)	0.0005 (6)	-0.0022 (6)
C21	0.0318 (10)	0.0195 (9)	0.0186 (9)	0.0011 (7)	0.0021 (7)	-0.0014 (7)

C22	0.0524 (13)	0.0205 (10)	0.0243 (10)	0.0032 (8)	0.0018 (9)	-0.0067 (8)
C23	0.0390 (11)	0.0365 (11)	0.0164 (9)	0.0011 (8)	0.0022 (8)	-0.0093 (8)
C24	0.0270 (10)	0.0329 (11)	0.0170 (9)	-0.0052 (8)	0.0028 (7)	-0.0006 (7)
C25	0.0240 (9)	0.0211 (9)	0.0186 (8)	-0.0040 (7)	0.0012 (7)	-0.0015 (7)
C26	0.0156 (8)	0.0138 (8)	0.0171 (8)	-0.0016 (6)	0.0011 (6)	-0.0003 (6)
C27	0.0192 (8)	0.0180 (9)	0.0179 (8)	-0.0032 (6)	0.0018 (6)	0.0001 (7)
C28	0.0196 (9)	0.0245 (9)	0.0239 (9)	-0.0035 (7)	0.0057 (7)	-0.0054 (7)
C29	0.0183 (9)	0.0213 (9)	0.0348 (10)	0.0037 (7)	0.0002 (7)	-0.0053 (8)
C30	0.0247 (9)	0.0248 (10)	0.0282 (10)	0.0065 (7)	-0.0019 (8)	0.0062 (8)
C31	0.0220 (9)	0.0231 (9)	0.0197 (9)	0.0006 (7)	0.0031 (7)	0.0022 (7)

*Geometric parameters (Å, °)*

Fe1—C2	1.7855 (18)	C12—C13	1.392 (2)
Fe1—C1	1.7981 (17)	C12—H12	0.9500
Fe1—C3	1.8006 (18)	C13—H13	0.9500
Fe1—S1	2.2484 (6)	C14—C15	1.391 (2)
Fe1—S2	2.2495 (8)	C14—C19	1.395 (2)
Fe1—Fe2	2.5275 (6)	C15—C16	1.390 (2)
Fe2—C4	1.7733 (17)	C15—H15	0.9500
Fe2—C5	1.7742 (17)	C16—C17	1.378 (2)
Fe2—P1	2.2426 (7)	C16—H16	0.9500
Fe2—S1	2.2495 (7)	C17—C18	1.386 (2)
Fe2—S2	2.2530 (7)	C17—H17	0.9500
P1—C8	1.8353 (16)	C18—C19	1.386 (2)
P1—C14	1.8371 (16)	C18—H18	0.9500
P1—P2	2.2519 (8)	C19—H19	0.9500
P2—O6	1.4943 (11)	C20—C21	1.392 (2)
P2—C20	1.8049 (16)	C20—C25	1.397 (2)
P2—C26	1.8066 (16)	C21—C22	1.384 (2)
S1—C6	1.8264 (16)	C21—H21	0.9500
S2—C7	1.8219 (16)	C22—C23	1.381 (2)
O1—C1	1.1461 (19)	C22—H22	0.9500
O2—C2	1.145 (2)	C23—C24	1.386 (2)
O3—C3	1.1414 (19)	C23—H23	0.9500
O4—C4	1.1476 (18)	C24—C25	1.382 (2)
O5—C5	1.1496 (19)	C24—H24	0.9500
C6—C7	1.523 (2)	C25—H25	0.9500
C6—H6A	0.9900	C26—C31	1.392 (2)
C6—H6B	0.9900	C26—C27	1.398 (2)
C7—H7A	0.9900	C27—C28	1.381 (2)
C7—H7B	0.9900	C27—H27	0.9500
C8—C9	1.396 (2)	C28—C29	1.389 (2)
C8—C13	1.396 (2)	C28—H28	0.9500
C9—C10	1.383 (2)	C29—C30	1.383 (2)
C9—H9	0.9500	C29—H29	0.9500
C10—C11	1.381 (2)	C30—C31	1.379 (2)
C10—H10	0.9500	C30—H30	0.9500

C11—C12	1.378 (3)	C31—H31	0.9500
C11—H11	0.9500		
C2—Fe1—C1	101.48 (8)	C10—C9—H9	119.4
C2—Fe1—C3	92.93 (8)	C8—C9—H9	119.4
C1—Fe1—C3	99.62 (7)	C11—C10—C9	119.85 (17)
C2—Fe1—S1	88.39 (6)	C11—C10—H10	120.1
C1—Fe1—S1	100.81 (5)	C9—C10—H10	120.1
C3—Fe1—S1	158.84 (6)	C12—C11—C10	119.80 (16)
C2—Fe1—S2	159.44 (5)	C12—C11—H11	120.1
C1—Fe1—S2	97.57 (5)	C10—C11—H11	120.1
C3—Fe1—S2	91.43 (6)	C11—C12—C13	120.83 (16)
S1—Fe1—S2	80.48 (2)	C11—C12—H12	119.6
C2—Fe1—Fe2	103.55 (6)	C13—C12—H12	119.6
C1—Fe1—Fe2	144.71 (5)	C12—C13—C8	119.83 (16)
C3—Fe1—Fe2	103.51 (6)	C12—C13—H13	120.1
S1—Fe1—Fe2	55.831 (19)	C8—C13—H13	120.1
S2—Fe1—Fe2	55.91 (2)	C15—C14—C19	118.50 (14)
C4—Fe2—C5	89.47 (7)	C15—C14—P1	120.51 (11)
C4—Fe2—P1	96.70 (6)	C19—C14—P1	120.96 (12)
C5—Fe2—P1	102.94 (5)	C16—C15—C14	120.49 (15)
C4—Fe2—S1	92.92 (5)	C16—C15—H15	119.8
C5—Fe2—S1	157.08 (5)	C14—C15—H15	119.8
P1—Fe2—S1	99.41 (2)	C17—C16—C15	120.46 (16)
C4—Fe2—S2	153.40 (5)	C17—C16—H16	119.8
C5—Fe2—S2	87.33 (5)	C15—C16—H16	119.8
P1—Fe2—S2	109.75 (2)	C16—C17—C18	119.65 (15)
S1—Fe2—S2	80.385 (18)	C16—C17—H17	120.2
C4—Fe2—Fe1	99.20 (6)	C18—C17—H17	120.2
C5—Fe2—Fe1	101.33 (5)	C19—C18—C17	120.06 (15)
P1—Fe2—Fe1	150.983 (16)	C19—C18—H18	120.0
S1—Fe2—Fe1	55.791 (13)	C17—C18—H18	120.0
S2—Fe2—Fe1	55.79 (2)	C18—C19—C14	120.81 (15)
C8—P1—C14	100.18 (7)	C18—C19—H19	119.6
C8—P1—Fe2	114.85 (5)	C14—C19—H19	119.6
C14—P1—Fe2	119.36 (5)	C21—C20—C25	118.86 (14)
C8—P1—P2	104.48 (5)	C21—C20—P2	116.48 (12)
C14—P1—P2	102.27 (5)	C25—C20—P2	124.64 (12)
Fe2—P1—P2	113.64 (3)	C22—C21—C20	120.49 (16)
O6—P2—C20	112.94 (7)	C22—C21—H21	119.8
O6—P2—C26	111.20 (6)	C20—C21—H21	119.8
C20—P2—C26	107.80 (7)	C23—C22—C21	120.24 (17)
O6—P2—P1	109.43 (5)	C23—C22—H22	119.9
C20—P2—P1	104.22 (5)	C21—C22—H22	119.9
C26—P2—P1	111.04 (6)	C22—C23—C24	119.80 (16)
C6—S1—Fe1	102.29 (6)	C22—C23—H23	120.1
C6—S1—Fe2	104.41 (6)	C24—C23—H23	120.1
Fe1—S1—Fe2	68.38 (2)	C25—C24—C23	120.25 (16)



C7—S2—Fe1	100.03 (6)	C25—C24—H24	119.9
C7—S2—Fe2	106.78 (5)	C23—C24—H24	119.9
Fe1—S2—Fe2	68.299 (13)	C24—C25—C20	120.35 (16)
O1—C1—Fe1	176.91 (14)	C24—C25—H25	119.8
O2—C2—Fe1	178.68 (16)	C20—C25—H25	119.8
O3—C3—Fe1	178.89 (16)	C31—C26—C27	119.05 (14)
O4—C4—Fe2	175.95 (15)	C31—C26—P2	123.97 (12)
O5—C5—Fe2	175.41 (15)	C27—C26—P2	116.97 (12)
C7—C6—S1	112.22 (10)	C28—C27—C26	120.28 (15)
C7—C6—H6A	109.2	C28—C27—H27	119.9
S1—C6—H6A	109.2	C26—C27—H27	119.9
C7—C6—H6B	109.2	C27—C28—C29	120.26 (15)
S1—C6—H6B	109.2	C27—C28—H28	119.9
H6A—C6—H6B	107.9	C29—C28—H28	119.9
C6—C7—S2	111.97 (10)	C30—C29—C28	119.49 (15)
C6—C7—H7A	109.2	C30—C29—H29	120.3
S2—C7—H7A	109.2	C28—C29—H29	120.3
C6—C7—H7B	109.2	C31—C30—C29	120.67 (16)
S2—C7—H7B	109.2	C31—C30—H30	119.7
H7A—C7—H7B	107.9	C29—C30—H30	119.7
C9—C8—C13	118.54 (14)	C30—C31—C26	120.25 (15)
C9—C8—P1	116.08 (12)	C30—C31—H31	119.9
C13—C8—P1	125.37 (12)	C26—C31—H31	119.9
C10—C9—C8	121.13 (16)		
C2—Fe1—Fe2—C4	8.72 (8)	C5—Fe2—S2—C7	-160.02 (8)
C1—Fe1—Fe2—C4	142.75 (11)	P1—Fe2—S2—C7	-57.23 (6)
C3—Fe1—Fe2—C4	-87.70 (7)	S1—Fe2—S2—C7	39.40 (6)
S1—Fe1—Fe2—C4	87.27 (5)	Fe1—Fe2—S2—C7	94.35 (6)
S2—Fe1—Fe2—C4	-170.14 (5)	C4—Fe2—S2—Fe1	22.18 (11)
C2—Fe1—Fe2—C5	100.03 (8)	C5—Fe2—S2—Fe1	105.64 (5)
C1—Fe1—Fe2—C5	-125.94 (11)	P1—Fe2—S2—Fe1	-151.57 (2)
C3—Fe1—Fe2—C5	3.60 (7)	S1—Fe2—S2—Fe1	-54.945 (14)
S1—Fe1—Fe2—C5	178.57 (5)	C4—Fe2—C5—O5	70.8 (18)
S2—Fe1—Fe2—C5	-78.83 (5)	P1—Fe2—C5—O5	167.5 (18)
C2—Fe1—Fe2—P1	-113.67 (7)	S1—Fe2—C5—O5	-25.5 (19)
C1—Fe1—Fe2—P1	20.35 (10)	S2—Fe2—C5—O5	-82.8 (18)
C3—Fe1—Fe2—P1	149.90 (6)	Fe1—Fe2—C5—O5	-28.5 (18)
S1—Fe1—Fe2—P1	-35.13 (3)	Fe1—S1—C6—C7	-31.06 (12)
S2—Fe1—Fe2—P1	67.46 (4)	Fe2—S1—C6—C7	39.46 (12)
C2—Fe1—Fe2—S1	-78.54 (6)	S1—C6—C7—S2	-8.36 (15)
C1—Fe1—Fe2—S1	55.48 (9)	Fe1—S2—C7—C6	43.88 (11)
C3—Fe1—Fe2—S1	-174.97 (6)	Fe2—S2—C7—C6	-26.31 (12)
S2—Fe1—Fe2—S1	102.59 (2)	C14—P1—C8—C9	-60.76 (14)
C2—Fe1—Fe2—S2	178.86 (6)	Fe2—P1—C8—C9	68.45 (13)
C1—Fe1—Fe2—S2	-47.11 (9)	P2—P1—C8—C9	-166.39 (12)
C3—Fe1—Fe2—S2	82.44 (5)	C14—P1—C8—C13	117.87 (14)
S1—Fe1—Fe2—S2	-102.59 (2)	Fe2—P1—C8—C13	-112.92 (13)

C4—Fe2—P1—C8	-73.60 (8)	P2—P1—C8—C13	12.24 (14)
C5—Fe2—P1—C8	-164.59 (8)	C13—C8—C9—C10	-1.0 (3)
S1—Fe2—P1—C8	20.50 (6)	P1—C8—C9—C10	177.70 (14)
S2—Fe2—P1—C8	103.59 (6)	C8—C9—C10—C11	-0.2 (3)
Fe1—Fe2—P1—C8	49.34 (7)	C9—C10—C11—C12	1.2 (3)
C4—Fe2—P1—C14	45.35 (8)	C10—C11—C12—C13	-1.1 (3)
C5—Fe2—P1—C14	-45.64 (8)	C11—C12—C13—C8	-0.2 (3)
S1—Fe2—P1—C14	139.45 (6)	C9—C8—C13—C12	1.2 (2)
S2—Fe2—P1—C14	-137.46 (6)	P1—C8—C13—C12	-177.41 (13)
Fe1—Fe2—P1—C14	168.29 (6)	C8—P1—C14—C15	147.58 (13)
C4—Fe2—P1—P2	166.17 (5)	Fe2—P1—C14—C15	21.35 (15)
C5—Fe2—P1—P2	75.18 (5)	P2—P1—C14—C15	-105.03 (13)
S1—Fe2—P1—P2	-99.72 (2)	C8—P1—C14—C19	-34.57 (14)
S2—Fe2—P1—P2	-16.64 (3)	Fe2—P1—C14—C19	-160.80 (11)
Fe1—Fe2—P1—P2	-70.88 (4)	P2—P1—C14—C19	72.82 (13)
C8—P1—P2—O6	-56.30 (7)	C19—C14—C15—C16	-2.3 (2)
C14—P1—P2—O6	-160.36 (6)	P1—C14—C15—C16	175.62 (12)
Fe2—P1—P2—O6	69.63 (5)	C14—C15—C16—C17	1.5 (2)
C8—P1—P2—C20	-177.36 (7)	C15—C16—C17—C18	0.0 (3)
C14—P1—P2—C20	78.58 (7)	C16—C17—C18—C19	-0.5 (3)
Fe2—P1—P2—C20	-51.43 (6)	C17—C18—C19—C14	-0.4 (3)
C8—P1—P2—C26	66.83 (7)	C15—C14—C19—C18	1.7 (2)
C14—P1—P2—C26	-37.23 (7)	P1—C14—C19—C18	-176.14 (13)
Fe2—P1—P2—C26	-167.24 (5)	O6—P2—C20—C21	8.73 (15)
C2—Fe1—S1—C6	-151.58 (8)	C26—P2—C20—C21	-114.53 (13)
C1—Fe1—S1—C6	-50.20 (8)	P1—P2—C20—C21	127.42 (12)
C3—Fe1—S1—C6	114.47 (16)	O6—P2—C20—C25	-169.59 (13)
S2—Fe1—S1—C6	45.77 (6)	C26—P2—C20—C25	67.16 (15)
Fe2—Fe1—S1—C6	100.81 (6)	P1—P2—C20—C25	-50.90 (14)
C2—Fe1—S1—Fe2	107.60 (6)	C25—C20—C21—C22	-0.8 (2)
C1—Fe1—S1—Fe2	-151.01 (5)	P2—C20—C21—C22	-179.23 (14)
C3—Fe1—S1—Fe2	13.65 (15)	C20—C21—C22—C23	1.6 (3)
S2—Fe1—S1—Fe2	-55.042 (18)	C21—C22—C23—C24	-1.0 (3)
C4—Fe2—S1—C6	163.12 (7)	C22—C23—C24—C25	-0.3 (3)
C5—Fe2—S1—C6	-101.33 (14)	C23—C24—C25—C20	1.1 (3)
P1—Fe2—S1—C6	65.83 (6)	C21—C20—C25—C24	-0.5 (2)
S2—Fe2—S1—C6	-42.79 (5)	P2—C20—C25—C24	177.80 (13)
Fe1—Fe2—S1—C6	-97.73 (6)	O6—P2—C26—C31	-159.14 (13)
C4—Fe2—S1—Fe1	-99.14 (6)	C20—P2—C26—C31	-34.83 (16)
C5—Fe2—S1—Fe1	-3.59 (13)	P1—P2—C26—C31	78.75 (14)
P1—Fe2—S1—Fe1	163.565 (15)	O6—P2—C26—C27	19.76 (14)
S2—Fe2—S1—Fe1	54.94 (2)	C20—P2—C26—C27	144.07 (12)
C2—Fe1—S2—C7	-107.34 (17)	P1—P2—C26—C27	-102.35 (12)
C1—Fe1—S2—C7	50.53 (7)	C31—C26—C27—C28	-0.1 (2)
C3—Fe1—S2—C7	150.43 (7)	P2—C26—C27—C28	-179.03 (12)
S1—Fe1—S2—C7	-49.23 (5)	C26—C27—C28—C29	-0.1 (2)
Fe2—Fe1—S2—C7	-104.19 (5)	C27—C28—C29—C30	0.3 (2)
C2—Fe1—S2—Fe2	-3.15 (16)	C28—C29—C30—C31	-0.3 (3)

## supporting information

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C1—Fe1—S2—Fe2	154.72 (5)	C29—C30—C31—C26	0.1 (3)
C3—Fe1—S2—Fe2	-105.38 (5)	C27—C26—C31—C30	0.1 (2)
S1—Fe1—S2—Fe2	54.960 (17)	P2—C26—C31—C30	178.96 (13)
C4—Fe2—S2—C7	116.53 (12)		

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