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## Structure Reports

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# N'-(4-Fluorobenzylidene)-2-(4-fluorophenyl)acetohydrazide

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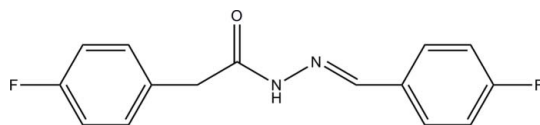
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 Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.045;  $wR$  factor = 0.153; data-to-parameter ratio = 19.3.

In the title compound,  $\text{C}_{15}\text{H}_{12}\text{F}_2\text{N}_2\text{O}$ , the dihedral angle between the two benzene rings is  $48.73$  ( $8^\circ$ ). The hydrazine group is twisted slightly, with a  $\text{C}-\text{N}-\text{N}-\text{C}$  torsion angle of  $172.48$  ( $12^\circ$ ). In the crystal, molecules are connected by strong  $\text{N}-\text{H}\cdots\text{O}$  and weak  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds, forming supramolecular chains along the  $c$  axis. The structure is consolidated by  $\pi-\pi$  [centroid-centroid separation =  $3.6579$  ( $10$ ) Å] and  $\text{C}-\text{H}\cdots\pi$  interactions.

## Related literature

For further details of aroylhydrozones, see: Li & Qu (2011); Zhang (2011); Fan *et al.* (2008). Ajani *et al.* (2010); Avaji *et al.* (2009); Rasras *et al.* (2010).



## Experimental

## Crystal data

 $\text{C}_{15}\text{H}_{12}\text{F}_2\text{N}_2\text{O}$   
 $M_r = 274.27$   
 Monoclinic,  $P2_1/c$   
 $a = 13.8754$  ( $15$ ) Å  
 $b = 12.5349$  ( $13$ ) Å  
 $c = 7.7093$  ( $8$ ) Å  
 $\beta = 93.566$  ( $2$ )°

 $V = 1338.3$  ( $2$ ) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.11$  mm<sup>-1</sup>  
 $T = 296$  K  
 $0.85 \times 0.26 \times 0.12$  mm

## Data collection

 Bruker APEXII DUO CCD diffractometer  
 Absorption correction: multi-scan (SADABS; Bruker, 2009)  
 $T_{\min} = 0.915$ ,  $T_{\max} = 0.988$   
 17153 measured reflections  
 4415 independent reflections  
 2586 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.026$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.153$   
 $S = 1.02$   
 4415 reflections  
 229 parameters  
 All H-atom parameters refined  
 $\Delta\rho_{\text{max}} = 0.21$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.23$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C1–C6 ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H1N1}\cdots\text{O1}^{\text{i}}$	0.892 (15)	2.013 (15)	2.8841 (14)	165.2 (14)
$\text{C4}-\text{H4}\cdots\text{O1}^{\text{ii}}$	0.92 (2)	2.47 (2)	3.370 (2)	168 (2)
$\text{C1}-\text{H1}\cdots\text{Cg1}^{\text{iii}}$	0.98 (2)	2.92 (2)	3.7025 (18)	138.0 (15)

 Symmetry codes: (i)  $x, -y + \frac{3}{2}, z - \frac{1}{2}$ ; (ii)  $-x + 1, -y + 1, -z + 1$ ; (iii)  $x, -y + \frac{1}{2}, z - \frac{3}{2}$ .

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6416).

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## supporting information

*Acta Cryst.* (2011). E67, o2835 [doi:10.1107/S1600536811039845]

***N'*-(4-Fluorobenzylidene)-2-(4-fluorophenyl)acetohydrazide**

Hoong-Kun Fun, Madhukar Hemamalini, V. Sumangala, G. K. Nagaraja and Boja Poojary

**S1. Comment**

Large number of aroylhydrazones have been synthesized in the recent years (Li & Qu, 2011; Zhang, 2011; Fan *et al.*, 2008) which can serve as intermediates in synthesizing biologically active compounds (Ajani *et al.*, 2010; Avaji *et al.*, 2009; Rasras *et al.*, 2010).

The asymmetric unit of the title compound is shown in Fig. 1. The dihedral angle between the two benzene rings (C1–C6/C10–C15) is 48.73 (8)°. The hydrazine group is twisted slightly with C9–N1–N2–C8, N1–N2–C8–C7 and N2–N1–C9–C10 torsion angles of 172.48 (12)°, 169.41 (12)° and 174.13 (11)°, respectively.

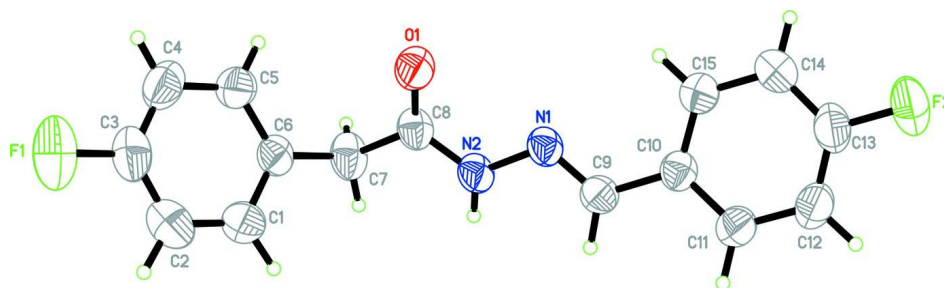
In the crystal structure, (Fig. 2), the molecules are connected *via* intermolecular strong N—H···O and weak C—H···O (Table 1) hydrogen bonds forming one-dimensional supramolecular chains along the *c*-axis. The crystal structure is further stabilized by  $\pi$ – $\pi$  interactions between the benzene (Cg2; C10–C15) rings [Cg2···Cg2 = 3.6579 (10) Å; -*x*, 2-*y*, 1-*z*] and C—H··· $\pi$  interaction involving the centroid of the C1–C6 (Cg1) ring.

**S2. Experimental**

An equimolar mixture of 2-(4-fluorophenyl)acetohydrazide and 4-fluorobenzaldehyde was refluxed for four hours in the presence of few drops of acid catalyst and ethanol as solvent. The compound obtained was filtered, washed, dried and recrystallised from ethanol to yield colourless needles.

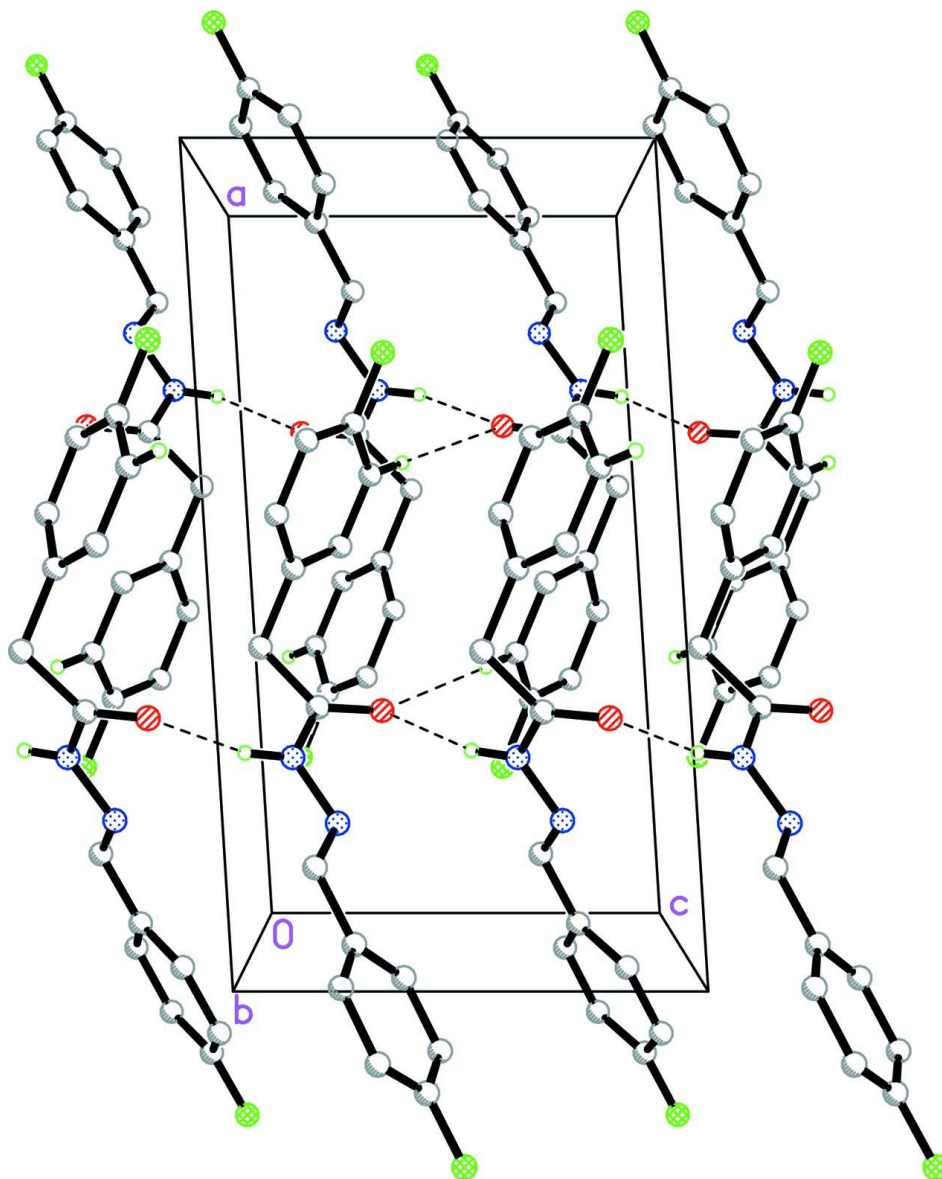
**S3. Refinement**

All hydrogen atoms were located from a difference Fourier maps and refined freely [N–H = 0.890 (17) Å and C–H = 0.92 (2)–1.001 (18) Å].



**Figure 1**

The asymmetric unit of the title compound, showing 50% probability displacement ellipsoids.



**Figure 2**

The crystal packing of the title compound (I). H atoms not involved in hydrogen bonding are omitted.

***N'*-(4-Fluorobenzylidene)-2-(4-fluorophenyl)acetohydrazide**

*Crystal data*

$C_{15}H_{12}F_2N_2O$

$M_r = 274.27$

Monoclinic,  $P2_1/c$

Hall symbol:  $-P\ 2_1/c$

$a = 13.8754\ (15)\ \text{\AA}$

$b = 12.5349\ (13)\ \text{\AA}$

$c = 7.7093\ (8)\ \text{\AA}$

$\beta = 93.566\ (2)^\circ$

$V = 1338.3\ (2)\ \text{\AA}^3$

$Z = 4$

$F(000) = 568$

$D_x = 1.361\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 3867 reflections

$\theta = 3.1\text{--}28.1^\circ$

$\mu = 0.11\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Needle, colourless

$0.85 \times 0.26 \times 0.12\ \text{mm}$

*Data collection*

Bruker APEXII DUO CCD  
diffractometer  
Radiation source: fine-focus sealed tube  
Graphite monochromator  
 $\varphi$  and  $\omega$  scans  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2009)  
 $T_{\min} = 0.915$ ,  $T_{\max} = 0.988$

17153 measured reflections  
4415 independent reflections  
2586 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.026$   
 $\theta_{\max} = 31.4^\circ$ ,  $\theta_{\min} = 2.2^\circ$   
 $h = -20 \rightarrow 18$   
 $k = -18 \rightarrow 17$   
 $l = -11 \rightarrow 11$

*Refinement*

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.153$   
 $S = 1.02$   
4415 reflections  
229 parameters  
0 restraints  
Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites  
All H-atom parameters refined  
 $w = 1/[\sigma^2(F_o^2) + (0.0752P)^2 + 0.0834P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.21 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$

*Special details*

**Geometry.** All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional R-factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.76995 (8)	0.61553 (12)	0.39014 (18)	0.1100 (4)
F2	-0.21176 (8)	0.96296 (11)	0.46342 (17)	0.1026 (4)
O1	0.31477 (7)	0.63023 (8)	0.33790 (11)	0.0593 (3)
N1	0.18163 (7)	0.77410 (8)	0.22901 (12)	0.0457 (2)
N2	0.26129 (8)	0.75056 (9)	0.13713 (13)	0.0486 (3)
C1	0.56239 (12)	0.71146 (14)	0.1320 (2)	0.0681 (4)
C2	0.65526 (13)	0.70642 (17)	0.2100 (3)	0.0799 (5)
C3	0.67904 (11)	0.62143 (15)	0.3128 (2)	0.0695 (4)
C4	0.61612 (13)	0.54169 (15)	0.3421 (2)	0.0693 (4)
C5	0.52362 (12)	0.54705 (12)	0.2623 (2)	0.0598 (4)
C6	0.49585 (9)	0.63228 (11)	0.15654 (15)	0.0501 (3)
C7	0.39518 (11)	0.63844 (15)	0.07075 (17)	0.0603 (4)
C8	0.32068 (9)	0.67230 (11)	0.19528 (14)	0.0478 (3)
C9	0.13383 (10)	0.85672 (11)	0.17896 (16)	0.0482 (3)
C10	0.04314 (9)	0.88361 (10)	0.25544 (15)	0.0469 (3)
C11	0.00248 (11)	0.98366 (12)	0.22675 (18)	0.0572 (3)

C12	-0.08399 (11)	1.01067 (14)	0.2957 (2)	0.0647 (4)
C13	-0.12794 (11)	0.93654 (14)	0.3933 (2)	0.0655 (4)
C14	-0.09066 (12)	0.83664 (14)	0.4254 (2)	0.0663 (4)
C15	-0.00469 (11)	0.81012 (12)	0.35531 (19)	0.0571 (3)
H1	0.5397 (14)	0.7715 (16)	0.059 (3)	0.094 (6)*
H2	0.7039 (17)	0.7598 (17)	0.196 (3)	0.101 (7)*
H4	0.6331 (15)	0.4869 (19)	0.417 (3)	0.104 (7)*
H5	0.4770 (13)	0.4897 (15)	0.275 (2)	0.079 (5)*
H7A	0.3934 (12)	0.6871 (14)	-0.029 (2)	0.074 (5)*
H7B	0.3739 (14)	0.5682 (15)	0.026 (2)	0.080 (5)*
H9	0.1566 (11)	0.9042 (12)	0.0904 (19)	0.058 (4)*
H11	0.0356 (12)	1.0384 (14)	0.157 (2)	0.071 (5)*
H12	-0.1126 (15)	1.0787 (17)	0.276 (2)	0.088 (6)*
H14	-0.1233 (15)	0.7855 (16)	0.491 (3)	0.090 (6)*
H15	0.0238 (13)	0.7400 (14)	0.374 (2)	0.073 (5)*
H1N1	0.2685 (11)	0.7821 (12)	0.035 (2)	0.061 (4)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
F1	0.0595 (6)	0.1415 (12)	0.1263 (9)	0.0162 (6)	-0.0142 (6)	-0.0132 (8)
F2	0.0706 (7)	0.1242 (10)	0.1173 (9)	0.0344 (6)	0.0415 (6)	0.0197 (7)
O1	0.0625 (6)	0.0731 (7)	0.0435 (5)	0.0189 (5)	0.0134 (4)	0.0069 (4)
N1	0.0455 (5)	0.0527 (6)	0.0396 (5)	0.0043 (4)	0.0094 (4)	-0.0022 (4)
N2	0.0490 (6)	0.0616 (7)	0.0362 (5)	0.0069 (5)	0.0119 (4)	0.0018 (4)
C1	0.0654 (9)	0.0731 (10)	0.0675 (9)	0.0061 (8)	0.0180 (7)	0.0172 (8)
C2	0.0626 (10)	0.0866 (13)	0.0921 (13)	-0.0099 (9)	0.0175 (9)	0.0057 (10)
C3	0.0493 (8)	0.0895 (12)	0.0701 (9)	0.0114 (8)	0.0064 (6)	-0.0068 (8)
C4	0.0693 (10)	0.0696 (10)	0.0695 (9)	0.0247 (8)	0.0079 (7)	0.0071 (8)
C5	0.0606 (8)	0.0538 (8)	0.0664 (8)	0.0082 (7)	0.0148 (7)	0.0035 (6)
C6	0.0502 (7)	0.0589 (8)	0.0427 (6)	0.0099 (6)	0.0155 (5)	-0.0009 (5)
C7	0.0545 (8)	0.0864 (11)	0.0409 (6)	0.0146 (7)	0.0107 (5)	-0.0074 (7)
C8	0.0462 (6)	0.0611 (8)	0.0367 (5)	0.0050 (6)	0.0059 (4)	-0.0054 (5)
C9	0.0515 (7)	0.0508 (7)	0.0429 (6)	0.0015 (6)	0.0071 (5)	0.0005 (5)
C10	0.0490 (6)	0.0489 (7)	0.0429 (6)	0.0039 (5)	0.0035 (5)	-0.0028 (5)
C11	0.0612 (8)	0.0547 (8)	0.0563 (7)	0.0090 (6)	0.0092 (6)	0.0071 (6)
C12	0.0660 (9)	0.0607 (9)	0.0682 (9)	0.0205 (8)	0.0103 (7)	0.0060 (7)
C13	0.0510 (8)	0.0800 (11)	0.0669 (8)	0.0149 (7)	0.0136 (6)	0.0006 (7)
C14	0.0589 (9)	0.0680 (10)	0.0736 (9)	-0.0006 (7)	0.0185 (7)	0.0084 (8)
C15	0.0564 (8)	0.0511 (8)	0.0647 (8)	0.0042 (6)	0.0115 (6)	0.0022 (6)

*Geometric parameters (Å, °)*

F1—C3	1.3633 (18)	C6—C7	1.510 (2)
F2—C13	1.3538 (17)	C7—C8	1.5146 (17)
O1—C8	1.2268 (15)	C7—H7A	0.982 (18)
N1—C9	1.2765 (16)	C7—H7B	0.984 (19)
N1—N2	1.3812 (14)	C9—C10	1.4619 (18)

N2—C8	1.3402 (17)	C9—H9	0.973 (16)
N2—H1N1	0.890 (17)	C10—C11	1.3873 (19)
C1—C6	1.377 (2)	C10—C15	1.3947 (19)
C1—C2	1.389 (3)	C11—C12	1.384 (2)
C1—H1	0.98 (2)	C11—H11	1.001 (18)
C2—C3	1.356 (3)	C12—C13	1.363 (2)
C2—H2	0.96 (2)	C12—H12	0.95 (2)
C3—C4	1.355 (3)	C13—C14	1.372 (2)
C4—C5	1.390 (2)	C14—C15	1.380 (2)
C4—H4	0.92 (2)	C14—H14	0.95 (2)
C5—C6	1.3840 (19)	C15—H15	0.971 (18)
C5—H5	0.976 (19)		
C9—N1—N2	115.83 (10)	C8—C7—H7B	105.7 (11)
C8—N2—N1	118.67 (10)	H7A—C7—H7B	106.8 (15)
C8—N2—H1N1	121.3 (10)	O1—C8—N2	122.72 (11)
N1—N2—H1N1	119.7 (10)	O1—C8—C7	122.27 (12)
C6—C1—C2	121.31 (16)	N2—C8—C7	115.01 (11)
C6—C1—H1	116.0 (12)	N1—C9—C10	120.62 (12)
C2—C1—H1	122.6 (12)	N1—C9—H9	121.5 (9)
C3—C2—C1	118.30 (17)	C10—C9—H9	117.8 (9)
C3—C2—H2	117.8 (13)	C11—C10—C15	118.84 (12)
C1—C2—H2	123.9 (13)	C11—C10—C9	119.73 (12)
C4—C3—C2	122.74 (16)	C15—C10—C9	121.43 (12)
C4—C3—F1	118.37 (16)	C12—C11—C10	120.90 (14)
C2—C3—F1	118.89 (17)	C12—C11—H11	118.5 (10)
C3—C4—C5	118.51 (16)	C10—C11—H11	120.6 (10)
C3—C4—H4	120.9 (14)	C13—C12—C11	118.25 (14)
C5—C4—H4	120.5 (14)	C13—C12—H12	120.2 (12)
C6—C5—C4	120.90 (16)	C11—C12—H12	121.5 (12)
C6—C5—H5	117.8 (11)	F2—C13—C12	118.59 (15)
C4—C5—H5	121.2 (11)	F2—C13—C14	118.36 (15)
C1—C6—C5	118.24 (14)	C12—C13—C14	123.05 (14)
C1—C6—C7	120.86 (14)	C13—C14—C15	118.33 (15)
C5—C6—C7	120.90 (14)	C13—C14—H14	121.7 (12)
C6—C7—C8	112.71 (10)	C15—C14—H14	119.9 (12)
C6—C7—H7A	110.7 (10)	C14—C15—C10	120.63 (14)
C8—C7—H7A	109.8 (10)	C14—C15—H15	120.8 (10)
C6—C7—H7B	110.9 (11)	C10—C15—H15	118.5 (10)
C9—N1—N2—C8	172.48 (12)	C6—C7—C8—O1	-49.3 (2)
C6—C1—C2—C3	-0.4 (3)	C6—C7—C8—N2	131.47 (14)
C1—C2—C3—C4	0.0 (3)	N2—N1—C9—C10	174.13 (11)
C1—C2—C3—F1	-179.98 (16)	N1—C9—C10—C11	166.80 (12)
C2—C3—C4—C5	0.5 (3)	N1—C9—C10—C15	-14.1 (2)
F1—C3—C4—C5	-179.50 (14)	C15—C10—C11—C12	0.0 (2)
C3—C4—C5—C6	-0.6 (2)	C9—C10—C11—C12	179.11 (13)
C2—C1—C6—C5	0.3 (2)	C10—C11—C12—C13	0.3 (2)

C2—C1—C6—C7	-179.85 (14)	C11—C12—C13—F2	178.96 (15)
C4—C5—C6—C1	0.2 (2)	C11—C12—C13—C14	-0.3 (3)
C4—C5—C6—C7	-179.62 (13)	F2—C13—C14—C15	-179.36 (15)
C1—C6—C7—C8	-102.93 (16)	C12—C13—C14—C15	-0.1 (3)
C5—C6—C7—C8	76.90 (18)	C13—C14—C15—C10	0.5 (2)
N1—N2—C8—O1	-9.9 (2)	C11—C10—C15—C14	-0.4 (2)
N1—N2—C8—C7	169.41 (12)	C9—C10—C15—C14	-179.50 (14)

*Hydrogen-bond geometry (Å, °)*

Cg1 is the centroid of the C1—C6 ring.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H1N1...O1 <sup>i</sup>	0.892 (15)	2.013 (15)	2.8841 (14)	165.2 (14)
C4—H4...O1 <sup>ii</sup>	0.92 (2)	2.47 (2)	3.370 (2)	168 (2)
C1—H1...Cg1 <sup>iii</sup>	0.98 (2)	2.92 (2)	3.7025 (18)	138.0 (15)

Symmetry codes: (i)  $x, -y+3/2, z-1/2$ ; (ii)  $-x+1, -y+1, -z+1$ ; (iii)  $x, -y+1/2, z-3/2$ .